

## Biochemometrics for Natural Products Research: Comparison of Data Analysis Approaches and Application to Identification of Bioactive Compounds

By: [Joshua J. Kellogg](#), Daniel A. Todd, Joseph M. Egan, [Huzefa A. Raja](#), [Nicholas H. Oberlies](#), Olav M. Kvalheim, and [Nadja B. Cech](#)

This document is the Accepted Manuscript version of a Published Work that appeared in final form in *Journal of Natural Products*, copyright © American Chemical Society after peer review and technical editing by the publisher. To access the final edited and published work see:

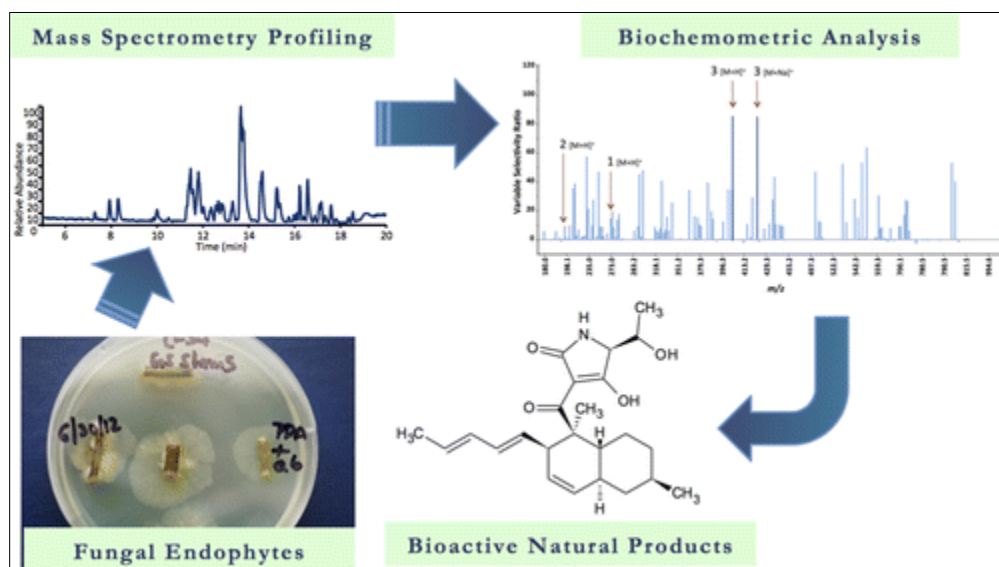
Kellogg, J.J., Todd, D.A., Egan, J.M., Raja, H.A., Oberlies, N.H., Kvalheim, O.M., Cech, N.B., (2016). Biochemometrics for natural products research: comparison of data analysis approaches and application to identification of bioactive compounds. *Journal of natural products*, 79(2), 376-386. doi: 10.1021/acs.jnatprod.5b01014

Made available courtesy of American Chemical Society:

<http://dx.doi.org/10.1021/acs.jnatprod.5b01014>

\*\*\*© American Chemical Society. Reprinted with permission. No further reproduction is authorized without written permission from American Chemical Society. This version of the document is not the version of record. Figures and/or pictures may be missing from this format of the document. \*\*\*

### Abstract:



A central challenge of natural products research is assigning bioactive compounds from complex mixtures. The gold standard approach to address this challenge, bioassay-guided fractionation, is often biased toward abundant, rather than bioactive, mixture components. This study evaluated the combination of bioassay-guided fractionation with untargeted metabolite profiling to improve active component identification early in the fractionation process. Key to this methodology was

statistical modeling of the integrated biological and chemical data sets (biochemometric analysis). Three data analysis approaches for biochemometric analysis were compared, namely, partial least-squares loading vectors, S-plots, and the selectivity ratio. Extracts from the endophytic fungi *Alternaria* sp. and *Pyrenochaeta* sp. with antimicrobial activity against *Staphylococcus aureus* served as test cases. Biochemometric analysis incorporating the selectivity ratio performed best in identifying bioactive ions from these extracts early in the fractionation process, yielding altersetin (**3**, MIC 0.23 µg/mL) and macrosphelide A (**4**, MIC 75 µg/mL) as antibacterial constituents from *Alternaria* sp. and *Pyrenochaeta* sp., respectively. This study demonstrates the potential of biochemometrics coupled with bioassay-guided fractionation to identify bioactive mixture components. A benefit of this approach is the ability to integrate multiple stages of fractionation and bioassay data into a single analysis.

**Keywords:** Biochemometrics | Fungi | *Alternaria* sp | *Pyrenochaeta* sp

### Article:

Natural products research has as its central goal the isolation and identification of bioactive constituent(s) from complex natural product mixtures. To achieve this, natural products chemists have developed a robust repertoire of techniques broadly termed “bioassay-guided fractionation”.(1) Bioassay-guided fractionation is an iterative methodology that alternates between chemical fractionation and bioassays. With each stage of fractionation, the complexity of the mixture is reduced, and eventually the compound(s) responsible for the observed biological effect can be isolated and characterized. This methodology has long been the gold standard in natural products research and has resulted in the discovery of critically important drugs, including camptothecin and Taxol (paclitaxel),(2, 3) artemisinin,(4) and vinblastine.(5) Weller reported in 2012 that over 1500 publications in *ISI Web of Science* employed bioassay-guided fractionation, with hundreds more citations using variants of the nomenclature.(6)

Despite the popularity and historical effectiveness of bioassay-guided fractionation, it has several limitations.(7) The process tends to be biased toward dominant peaks in each extract or fraction, and, as a result, bioactive constituents in low abundance can be overlooked.(8) Furthermore, isolation of all trace constituents can be difficult, given that each chemical separation step witnesses a decrease in material. Finally, there is the potential to lose activity due to irreversible binding of mixture components to chromatographic resins or degradation during the separation process.(9) In light of these limitations, new methods capable of focusing the isolation process on components most likely to be responsible for the desired biological effect are needed.

Recently, there has been a great deal of interest in the application of untargeted metabolomics to study biologically active natural product mixtures.(10-15) Metabolomics approaches are employed to profile multiple mixture components simultaneously, typically through the application of chromatographic analysis coupled to spectroscopic or spectrometric approaches (IR, UV, MS, or NMR detection). Such approaches can enable the detection of unstable compounds that would be lost upon purification and consider all compounds together rather than as distinct fractions in a series.(7, 16)

Metabolomic profiling results in the generation of large data sets that include both major and minor components.(11-13, 15) Data-driven methods are needed to extract meaning from these complex chemical data sets, and multivariate statisticians and chemometricians have developed a number of strategies toward this goal.(17, 18) The most commonly employed tool in metabolomics data analysis is principal component analysis (PCA), in which a data set is projected onto a series of latent variables, which are then mapped in two-dimensional space. Groupings of objects are discerned by their covariance, which is analyzed visually by the proximity of one object to another in the PCA scores plot.(18)

One limitation of metabolomics for studying natural product mixtures is the difficulty in tying identified metabolites to bioactive effects. If the end goal is determining which compounds are responsible for the biological activity of a mixture, comparing the chemical composition of different mixtures (the central goal of metabolomics) is not sufficient.(14, 19) There is a need to go beyond the metabolomics data sets and to use biological assay data to inform their interpretation. To address this need is an even greater data analysis challenge than that faced in classical metabolomics and requires the integration of both biological and chemical data sets. In 2006, chemometricians working to integrate chemical and biological data dubbed the field “biochemometrics”.(20) The present report is concerned primarily with the development of effective approaches for applying biochemometrics to natural products drug discovery.

Several approaches have been developed for correlating metabolite profiles with biological data sets (Table 1). Partial least-squares (PLS) decomposes the spectral data set (i.e., retention time and mass-to-charge ( $m/z$ ) pairings) into uncorrelated latent variables. PLS differs from PCA in that it seeks to maximize the covariance of independent variables (spectral data from IR, UV, or MS analysis) with a dependent variable (i.e., biological activity).(21) As an example of the effectiveness of this approach, Ali et al.(22) utilized PLS analysis of NMR signal data to identify bioactive metabolites from marine sponges against the adenosine A1 receptor. In some cases, however, data interpretation may be difficult with PLS, because variables possessing large variance yet small correlation may mask other variables with low variance and high correlation to the dependent (response) variable. In addition, multiple PLS components are often needed to optimize the discrimination between response groups.(23)

Recently, Wiklund et al.(24) described the S-plot as a means for interpreting orthogonal PLS (OPLS) predictive components. With an S-plot, the covariance and correlation loading variables are displayed graphically, which allows for visual identification of spectral variables that strongly correlate with a dependent biological activity variable.(24) S-plots have been utilized several times for natural product research, most recently to discover immunomodulatory components from *Phaleria nisidai*(25) and antidiabetic compounds of Cree medicinal plants.(26) A limitation of the S-plot approach is that the large number of spectral variables can make visualization and interpretation of the data difficult. In addition, the S-plot relies only on the correlation and covariance of independent variables to the dependent variable, which can lead to false positives.(27)

As another strategy for interpreting biochemometric data sets, Kvalheim and Karstang(28) developed the “target projection” component, wherein PLS components are transformed into a

univariate metric that facilitates analysis and interpretation of correlative data. The variance explained by the target projection component can be calculated for each independent (spectral) variable and compared against the residual variance. The ratio between explained and residual variance of the spectral variables of the target-projection component, termed the selectivity ratio, represents a quantitative measure of each variable's power to distinguish between different groups. Variables with a high selectivity ratio have an excellent ability to separate bioactive and nonbioactive groupings. This approach has been utilized to identify clinical biomarkers from human spinal fluid samples(27) as well as for chemical fingerprint analysis of the herbal medicine *Puerariae lobatae* (Radix Puerariae).(29) However, the selectivity ratio has not been applied to identify individual bioactive components of natural product mixtures.

**Table 1. Summary of Data Analysis Methods for Biochemometric Data Sets**

method	methodology	applications	analysis output
principal component analysis (PCA)	map objects and variables onto latent variables separately; identify correlations within groupings	outliers; quality control; object diversity	scores plot: summary of objects; loadings plot: summary of variables
partial least-squares (PLS)	incorporate objects and variables for predictive modeling	discriminating between groups; biomarker identification	scores plot: summary of objects; loadings plot: summary of variables
S-plot	combine modeled covariance and correlation from PLS in a scatter plot	same as PLS	low correlation/intensity variables are close to origin; highly correlated variables are distanced from origin
selectivity ratio	ratio of explained (predictive) and residual (uncorrelated) variance; variance developed from univariate "target projection"	variable discrimination; biomarker identification	x-axis: independent variables (spectral data, retention time, etc.); single variable for identifying highly correlated peaks

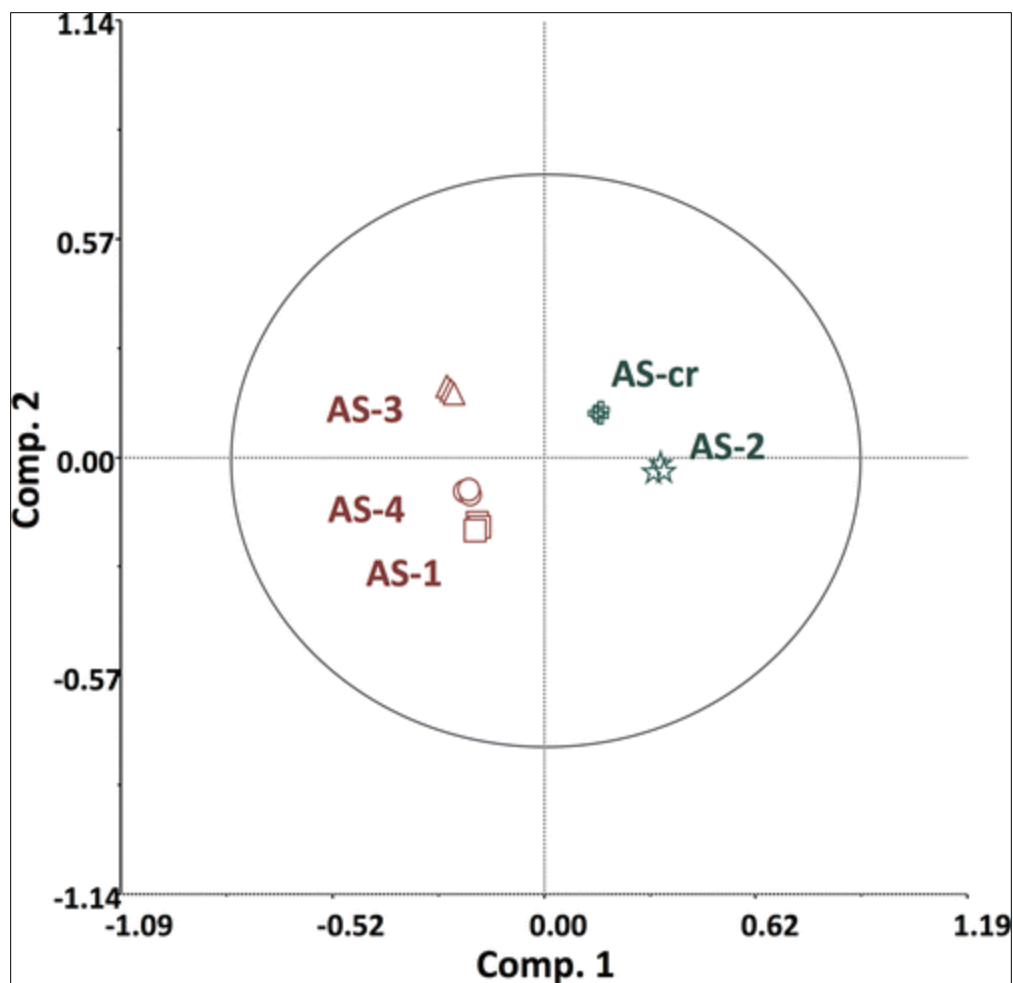
With this study, we compared three data analysis strategies (PLS, S-plot, and selectivity ratio) for integrating biological and chemical data sets from natural product mixtures. Our objective was to demonstrate which of these approaches would be most effective for distinguishing active and inactive compounds in the mixtures. As a case study, two endophytic fungi isolated from the

botanical goldenseal [*Hydrastis canadensis* L. (Ranunculaceae)] were selected, namely, *Alternaria* sp. and *Pyrenochaeta* sp. *Alternaria* sp. was chosen because the extract from this fungus demonstrated marked antimicrobial activity in screens performed in our laboratory (data not shown), but dereplication using a UPLC-HRMS-MS/MS protocol(30) identified only one primary chemical constituent, alternariol monomethyl ether (**1**), which is mildly active against Gram-positive bacteria. For *Pyrenochaeta* sp., the activity of the extract could not be correlated to known compounds in the dereplication library. Thus, these fungal extracts were a good example of mixtures containing unknown active compounds. The goal of our studies was to conduct biochemometric analysis on these fungi at an early stage of extraction and fractionation and subsequently to verify the predictions of the biochemometric analysis with follow-up isolation, structure elucidation, and biological evaluation.

## Results and Discussion

### Biochemometric Analysis of *Alternaria* sp

The first goal of these studies was to contrast various chemometric and biochemometric analysis techniques as applied to an extract from the fungus *Alternaria* sp. Toward this goal, chemometric profiling was first conducted on a crude extract from the *Alternaria* sp. fungus (AS-CR) and four fractions (AS-1 to AS-4). Untargeted metabolomic analysis of these fractions using ultraperformance liquid chromatography coupled to high-resolution mass spectrometry (UPLC-HRMS) yielded  $m/z$  472 total marker ions (unique retention time– $m/z$  pairs), which were compared using principal component analysis (Figure 1). The first three components of the PCA scores plot accounted for 93.09% of total variability of the model (component 1: 52.97%; component 2: 32.45%; component 3: 7.68%). The technical replicates (triplicate UPLC-HRMS analyses) of each sample are overlaid on the plot, indicating excellent repeatability of the chemical analysis (Figure 1). The AS-CR and AS-2 fractions group together, separated from the AS-1, AS-3, and AS-4 fractions (Figure 1), which indicates distinct chemical profiles between AS-2 and the fractions AS-1, -3, and -4.



**Figure 1.** Principal component analysis (PCA) scores plot of *Alternaria* sp. crude extract (AS-CR) and fractions AS1–AS4, drawn with Hotelling’s 95% confidence ellipse. All fractions were run in triplicate, and the resulting 472 marker ions were used to compute differences in mycochemical composition.

Bioactivity screening revealed complete inhibition of *S. aureus* strain SA1199 by the *Alternaria* sp. crude extract (AS-CR) as well as the second fraction (AS-2) (Table 2). At the 100 µg/mL level, the crude extract exhibited complete growth inhibition of *Staphylococcus aureus*, while fraction AS-2 evidenced near-complete inhibition ( $99.83 \pm 0.01\%$ ) (Table 2). The fractions AS-1, AS-3, and AS-4 demonstrated negligible growth inhibition of *S. aureus* (<1.0%).

**Table 2. Antimicrobial Activity of *Alternaria* sp. (AS) Crude Extract (CR) and Fractions AS-1–AS-4<sup>a</sup>**

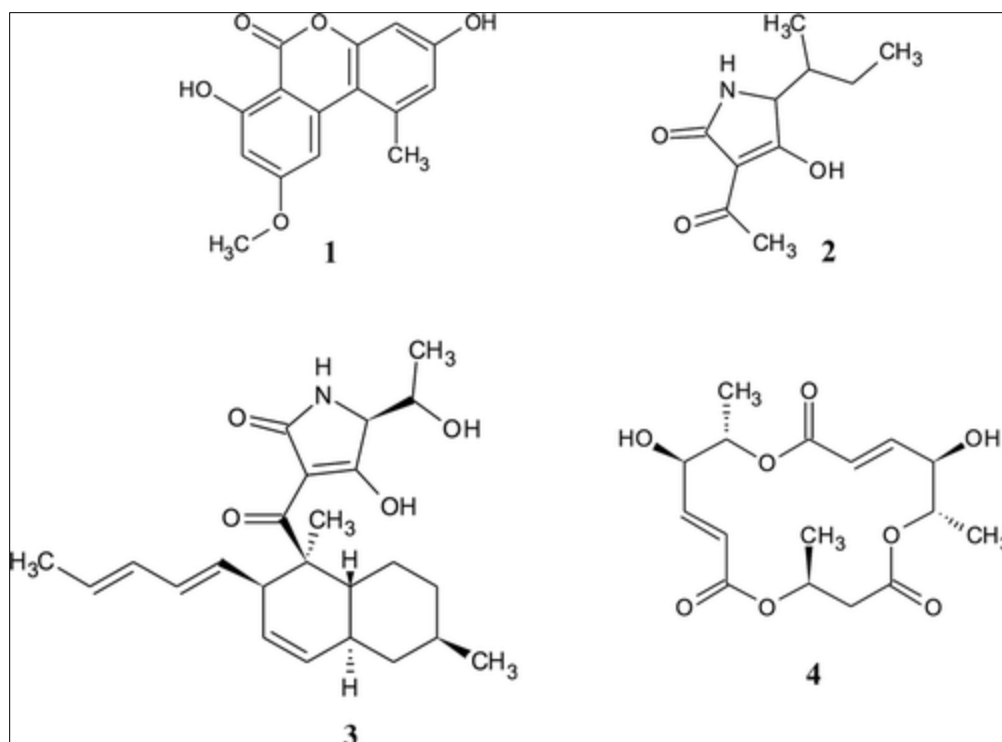
sample	<i>S. aureus</i> growth inhibition (%)
chloramphenicol <sup>b</sup>	$93.3 \pm 0.4$
AS-CR	$100 \pm 1$
AS-1	$0.00 \pm 0.03$
AS-2	$99.83 \pm 0.01$

AS-3	0.14 ± 0.02
AS-4	0.76 ± 0.01

<sup>a</sup> Growth inhibition of *S. aureus* strain SA1199 relative to the vehicle control as measured by OD600. AS samples measured at a concentration of 100 µg/mL. Data presented as mean of triplicate analyses ± SEM. <sup>b</sup>Chloramphenicol functioned as the positive control.

On pairing the antibacterial screening with high-resolution mass spectral data, the resulting biochemometric analytical matrix evidenced differences between the *Alternaria* sp. fractions based on their bioactivity. The internal cross-validated construction of the PLS model yielded four components, accounting for 100% of the independent (spectral) and dependent (bioactivity) block variation (component 1: 52.94% independent, 98.06% dependent; component 2: 10.60%, 1.79%; component 3: 29.08%, 0.16%; component 4: 7.38%, 0.00%). The PLS scores plot (Figure 2A) showed a similar clustering of fractions to the PCA analysis, with AS-CR and AS-2 separated graphically from the other, nonbioactive fractions. Also similar to the PCA analysis, the triplicate data points representing each sample are closely grouped together in the PLS scores plot, although the addition of biological variability causes a slight increase in the spread among replicates, as might be expected.

PLS scores plots do not provide information regarding which specific chemical species contribute to the observed antibacterial bioactivity. To obtain this information, three distinct analytical methods were contrasted: the loadings plot from the PLS model, the multivariate transformed S-plot, and the selectivity ratio. Examination of the PLS loadings plot (Figure 2B) yielded three major metabolites that were shifted in the same direction as the bioactive fractions from the PLS scores plot (Figure 2A). Identities of these compounds were proposed using literature and high-resolution mass spectrometry data (Table 3) as alternariol monomethyl ether (**1**), tenuazonic acid (**2**), and altersetin (**3**). The difference in location in the scores plot between **1**, **2**, and **3** was not sufficient to ascertain which of the three compounds was most responsible for the antibacterial activity of the fraction. The identifications of **1** and **3** were confirmed from the NMR spectra of the isolated compounds (Figures S2 and S4, Supporting Information, respectively), and NMR data were consistent with literature reports.<sup>(31, 32)</sup> Due to the low abundance and lack of bioactivity of fractions containing compound **2**, this compound was not pursued for isolation, and identification is only tentative, based on matching accurate mass spectrum with literature values.<sup>(29)</sup>



The S-plot graphically displays the covariance and correlation of loading variables against the dependent variable as a scatter plot (Figure 2C). In an S-plot, the further a marker ion is from the origin, the greater its contribution is to the variance between bioactivity levels. For the *Alternaria* sp. biochemometric model, the upper right quadrant of the S-plot contributed the most to the differentiation of biologically active versus inactive fractions, and compounds **1** and **2** were highlighted as possessing the greatest contribution to the observed bioactivity.



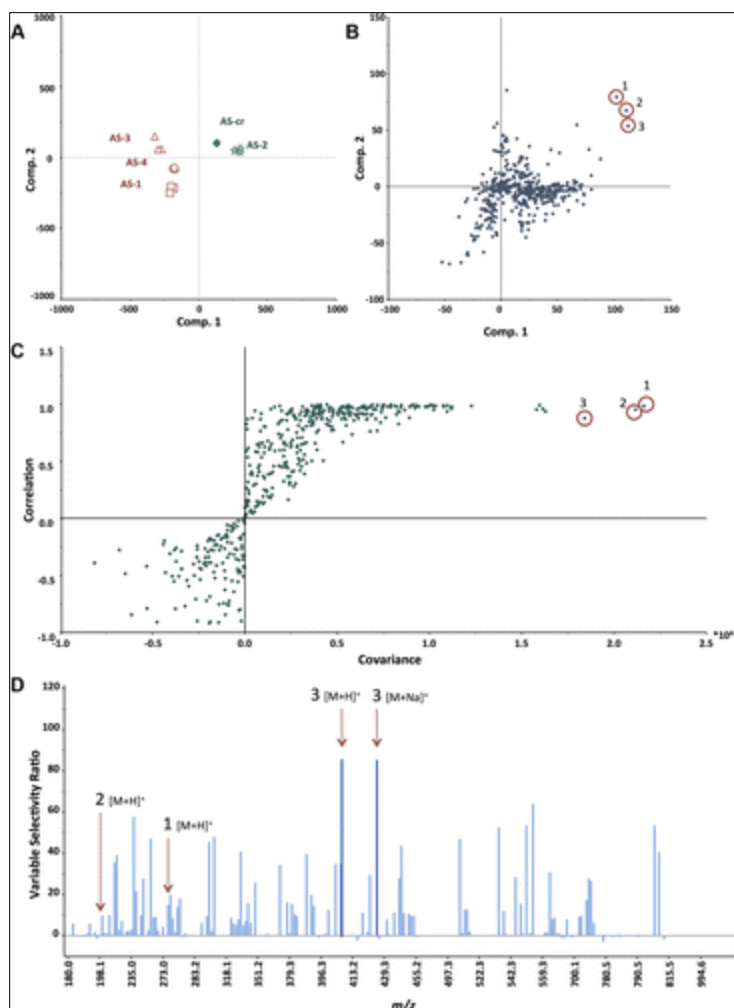


Figure 2. Marker ion selection from a biochemometric data set. The biochemometric data set was obtained from the mass spectral data coupled with bacterial growth inhibition data (against *S. aureus* SA1199) at a concentration of 100  $\mu\text{g}/\text{mL}$  (Table 2). (A) Partial least-squares (PLS) scores plot, showing the grouping of bioactive and nonbioactive fractions from *Alternaria* sp. (AS-CR and AS-1–AS-4). Each fraction was analyzed in triplicate via UPLC-MS and was subjected to triplicate biological assays. Thus, the replicate data points represent both biological and technical variability. (B) Loadings plot from the PLS analysis of biochemometric data. Variables located in the same region in the loadings plot (B) as the bioactive groups AS-CR and AS-2 in the scores plot (A) have the highest positive correlation with the dependent variable (bioactivity). Thus, three ions corresponding to alternariol monomethyl ether (**1**), tenuazonic acid (**2**), and altersetin (**3**) were identified from visual analysis of the loadings plot as potentially most bioactive. (C) S-plot from PLS model of antibacterial activity of *Alternaria* sp. extract and fractions. The upper right quadrant are the peaks with highest correlation to bioactivity, and ions **1**, **2**, and **3** were also identified from the S-plot. (D) Selectivity ratio analysis of the PLS model data. The ratio relates the explained variance of the variable to the residual variance. Higher values (taller lines) represent a more significant contribution to the observed bioactivity.

The selectivity ratio indicates compound **3** to have the highest activity and does not find strong correlation for compounds **1** and **2**.

The selectivity ratio produces a graphical representation in which the most abundant peaks correspond to marker ions that are most strongly associated with bioactivity (Figure 2D). From the selectivity ratio plot, the dominant marker ions were altersetin (**3**) and its sodium adduct, suggesting that this compound dominated the contributions to the antibacterial potency of the *Alternaria* sp. extract and fractions. In contrast to the PLS loadings plot (Figure 2B) and the S-plot (Figure 2C), alternariol monomethyl ether (**1**) and tenuazonic acid (**2**) were not among the most significant marker ions identified according to the selectivity ratio analysis (Figure 2D).

Selectivity ratio analysis has an additional advantage of enabling the application of multiple independent variables, which facilitates interpretation by natural product chemists consistent with the type of instrumentation being employed. Utilizing mass spectrometry data (signal versus  $m/z$ ) creates a plot similar to a mass spectrum (Figure 2D), where the  $x$ -axis is the  $m/z$  of the detected ion, and the  $y$ -axis represents how strongly associated that particular ion is with the biological activity.(23) The use of chromatographic data (detector signal versus retention time) yields a similar selectivity ratio plot to that generated with mass spectrometric data, except that the  $x$ -axis represents retention time rather than  $m/z$ .(33)

**Table 3. Identification of Bioactive Marker Ions from *Alternaria* Sp**

marker ion	ion (molecular formula, $\delta$ (ppm))	adducts and fragments (molecular formula, $\delta$ (ppm))	tentative identification
<b>1</b>	273.0756 [M + H] <sup>+</sup> (C <sub>15</sub> H <sub>13</sub> O <sub>5</sub> , 0.5)	255.0699 [M + H – H <sub>2</sub> O] <sup>+</sup> (C <sub>15</sub> H <sub>11</sub> O <sub>4</sub> , 4.2)	alternariol monomethyl ether <sup>a</sup>
<b>2</b>	198.1124 [M + H] <sup>+</sup> (C <sub>10</sub> H <sub>17</sub> NO <sub>3</sub> , 0.4)		tenuazonic acid <sup>a</sup>
<b>3</b>	400.2480 [M + H] <sup>+</sup> (C <sub>24</sub> H <sub>34</sub> NO <sub>4</sub> , 0.6)	422.2292 [M + Na] <sup>+</sup> (C <sub>24</sub> H <sub>33</sub> NO <sub>4</sub> Na, 1.6)	altersetin <sup>a</sup>
		382.2375 [M + H – H <sub>2</sub> O] <sup>+</sup> (C <sub>24</sub> H <sub>32</sub> NO <sub>3</sub> , 0.7)	

<sup>a</sup> Previously reported from cultured *Alternaria* spp.(31)

### Identification of Marker Compounds in *Alternaria* sp

Additional purification of the most active *Alternaria* sp. fraction (AS-2) was conducted to investigate the accuracy of the predictions provided by the biochemometric analysis. Subfractions of AS-2 (coded AS-2-1–AS-2-10) obtained with reversed-phase preparative-scale HPLC revealed marked differences in both chemical makeup and bioactivity. UPLC-HRMS analysis of individual subfractions showed that subfraction AS-2-3 was 94% enriched in tenuazonic acid (**2**) (Figure 3B), subfraction AS-2-7 was 94% alternariol monomethyl ether (**1**) (Figure 3C), and altersetin (**3**) was isolated in subfraction AS-2-9 at 85% purity (Figure 3D). The other subfractions contained insignificant quantities of these compounds. From the antibacterial screening protocol, subfraction AS-2-3 displayed only  $0.01 \pm 0.02\%$  inhibition of *S*.

*aureus* SA1199, AS-2-7 yielded  $0.08 \pm 0.01\%$  growth inhibition, and AS-2-9 inhibited bacterial growth by  $99.5 \pm 0.01\%$ . These data indicate that **3** was, indeed, the most active compound from the original extract. The application of the selectivity ratio made it possible to distinguish between active and inactive ions at an early stage of the analysis (with just the crude extract and four fractions). The selectivity ratio correctly predicted that ion **3** was the most active, while the S-plot and PLS loading vectors attributed activity to ions **1**, **2**, and **3**. Follow-up antimicrobial assays on pure compounds **1** and **3** supported the data in Figure 4, indicating that **1** is weakly active against *S. aureus*, with a minimum inhibitory concentration (MIC) of  $275 \mu\text{M}$  ( $75 \mu\text{g/mL}$ ), while **3** possesses pronounced activity (MIC  $0.59 \mu\text{M}$  ( $0.23 \mu\text{g/mL}$ )) (Table 4). Antimicrobial activity was also observed for both of these compounds against methicillin-resistant *Staphylococcus aureus* (MRSA) (Table 4).

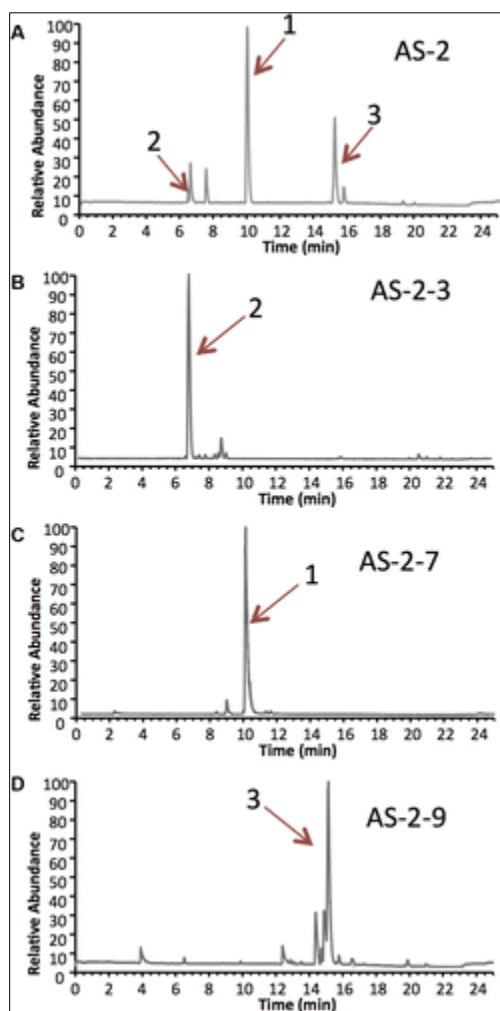


Figure 3. UPLS-HRMS chromatograms of fraction AS-2 (A), along with selected subfractions AS-2-3 (B), AS-2-7 (C), and AS-2-9 (D) representing the semipure fractions of tenuazonic acid (**2**), alternariol monomethyl ether (**1**), and altersetin (**3**), respectively.

**Table 4. Antimicrobial Activity of Isolated Compounds from *Alternaria* sp. and *Pyrenochaeta* sp.<sup>a</sup>**

sample	MIC <i>S. aureus</i>		MIC MRSA	
	μM	μg/mL	μM	μg/mL
berberine (+ control)	446	150	446	150
alternariol monomethyl ether ( <b>1</b> )	275	75	ND <sup>b</sup>	ND
altersetin ( <b>3</b> )	0.59	0.23	4.67	1.9
macrospheptide A ( <b>4</b> )	219	75	ND	ND

<sup>a</sup> Minimum inhibitory concentrations (MIC) against *S. aureus* strain SA1199 and a strain of MRSA (USA300 LAC strain AH1263) are presented as mean of triplicate analyses. Berberine functioned as the positive control. <sup>b</sup> ND, not detected.

### Refined Biochemometric Analysis

Multivariate statistical modeling increases in accuracy and precision as the sample size (number of objects) increases;(33) thus, it was hypothesized that further fractionation of the bioactive *Alternaria* sp. fraction AS-2 would provide enhanced separation between the active and inactive marker ions in the biochemometric analysis. The incorporation of subfractions (AS-2-1 through AS-2-10) of the original active fraction (AS-2) into the biochemometric matrix yielded a more refined statistical model and subsequent analysis. The PLS scores plot (Figure 4A) distinguished the active fractions and subfractions (AS-CR, AS-2, AS-2-9, and AS-2-10) from the nonactive fractions and subfractions. The increased analytical power of adding subfractions to the biochemometric matrix was reflected in changes in both the PLS loading plot (Figure 4B) and the S-plot (Figure 4C) compared to the initial matrix analysis (Figure 2B and C, respectively). In the expanded data matrix, both plots yielded altersetin (**3**) as the principal marker ion that contributed to the observed antibacterial activity, while the signals for compounds **1** and **2** were shifted downward toward a region of low covariance and correlation. The selectivity ratio (Figure 4D) maintained altersetin (**3**) as the principal bioactive marker ion. The inclusion of the subfractions in the selectivity ratio analysis effectively increased the abundance of the altersetin signals relative to those of other ions.

With the incorporation of 15 objects spanning three different degrees of chemical complexity—crude extract, fractions, and subfractions—the PLS loadings plot and S-plot could be employed to correctly identify **3** as the antibacterial compound from the mixture. However, the selectivity ratio analysis enabled identification of the active mixture components at an earlier stage of the isolation process.

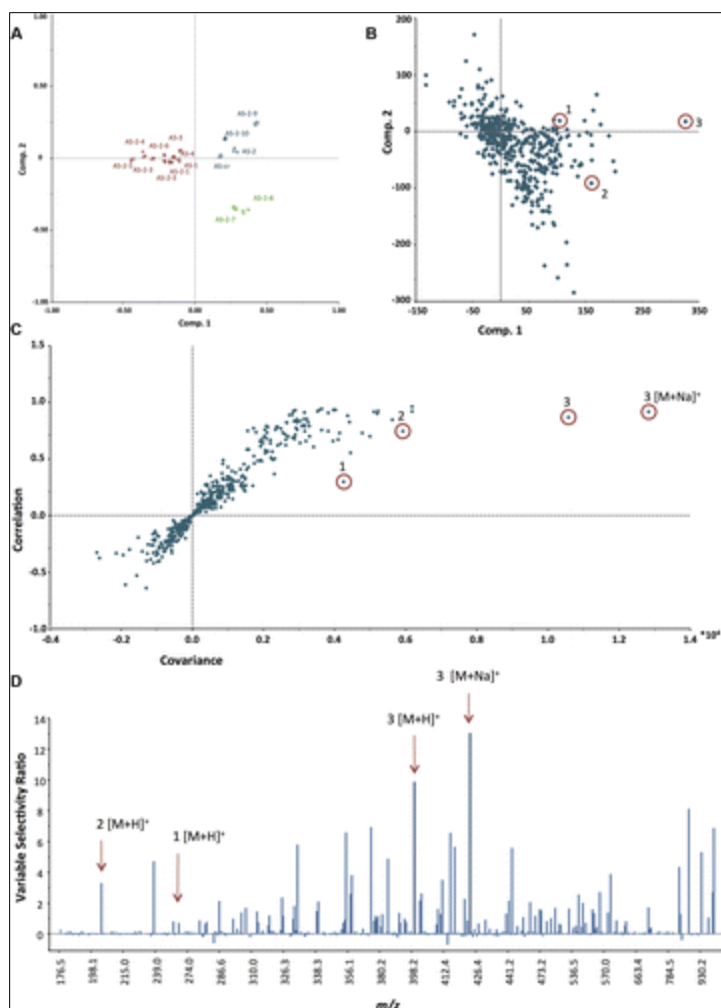


Figure 4. Marker ion selection from the postfractionation biochemometric data set of *Alternaria* sp. The biochemometric data set was obtained from the triplicate mass spectral data coupled with bacterial growth inhibition data (against *S. aureus* SA1199) at a concentration of 100  $\mu\text{g}/\text{mL}$ . (A) Partial least-squares (PLS) scores plot, showing the grouping of bioactive and inactive fractions from *Alternaria* sp. (AS-CR, AS-1–AS-4, and AS-2-1–AS-2-10). Each fraction was analyzed in triplicate, as shown in the scores plot. (B) Loadings plot from the PLS analysis of biochemometric data. Variable **3** was the most correlated to bioactivity, as implied by being shifted in the same direction as the bioactive samples in the scores plot. (C) S-plot from the larger PLS model of antibacterial activity of *Alternaria* sp. extract, fractions, and subfractions. The marker ion for **3** is distinctly separate from the others, indicating its greater contribution to the bioactivity. (D) Selectivity ratio analysis of the more comprehensive PLS model data. Similar to the initial selectivity ratio analysis (Figure 2D), ion **3** displays the highest selectivity ratio.

### Application of Biochemometrics to *Pyrenochaeta* sp

To confirm the analytical capability of the selectivity ratio, a second bioactive endophytic fungus (*Pyrenochaeta* sp.) was analyzed via the same methodology. Biochemometric profiling was conducted on the crude *Pyrenochaeta* sp. extract (PS-CR) and four fractions (PS-1 to PS-4).

UPLC-HRMS analysis yielded 659 marker ions, and bioactivity screening highlighted PS-4 as the most bioactive fraction (Table 5). The PLS scores plot (Figure 5A) separated the two active samples (PS-CR and PS-4) distinctly from the inactive samples. For *Pyrenochaeta* sp., the PLS loadings plot (Figure 5B) and the selectivity ratio (Figure 5C) revealed macrosphelide A (**4**) as the principal bioactive constituent. Subsequent isolation efforts confirmed the presence of macrosphelide A through high-resolution mass spectrometry and <sup>1</sup>H and <sup>13</sup>C NMR (Figure S6, Supporting Information),(34) and its MIC value against *S. aureus* was determined to be 219 μM (75 μg/mL) (Table 4).

**Table 5. Antimicrobial Activity of *Pyrenochaeta* sp. (PS) Crude Extract (CR) and Fractions PS-1–PS-4<sup>a</sup>**

<b>sample</b>	<b><i>S. aureus</i> growth inhibition (%)</b>
chloramphenicol <sup>b</sup>	99.2 ± 0.2
PS-CR	78.2 ± 2.4
PS-1	40.0 ± 1.2
PS-2	36.3 ± 1.5
PS-3	26.4 ± 4.7
PS-4	95.8 ± 1.7

<sup>a</sup> Growth inhibition of *S. aureus* strain SA1199 is displayed as percent growth inhibition normalized to the vehicle control as measured by OD600. *Pyrenochaeta* sp. samples were measured at a concentration of 100 μg/mL. Data presented as mean of triplicate analyses ± SEM.

<sup>b</sup> Chloramphenicol functioned as the positive control.

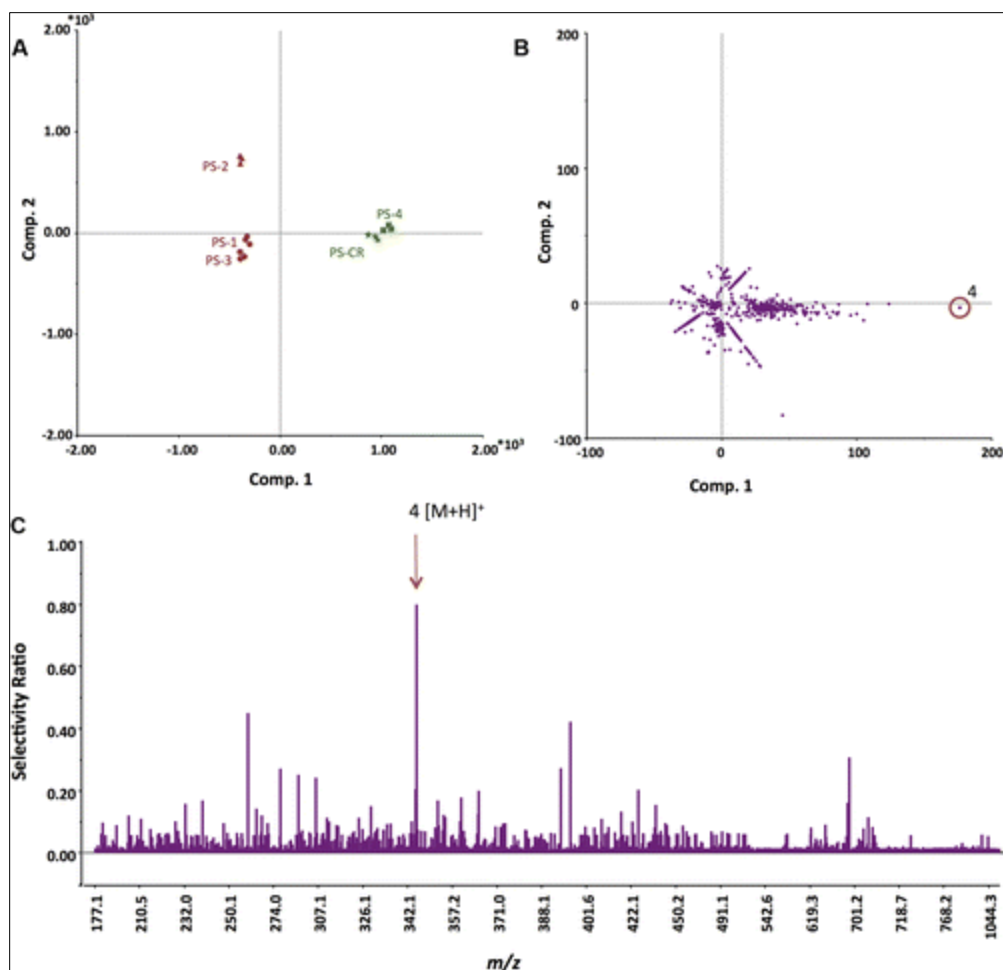


Figure 5. Identification of the bioactive principle from *Pyrenochaeta* sp. from the biochemometric data set. The biochemometric data set was obtained from the triplicate mass spectral data coupled with growth inhibition data against *S. aureus*(SA1199) at a concentration of 100  $\mu\text{g/mL}$ . (A) The partial least-squares (PLS) scores plot shows the grouping of bioactive and inactive fractions from *Pyrenochaeta* sp. (PS-CR, PS-1–PS-4). Each fraction was analyzed in triplicate, as shown in the scores plot. (B) Loadings plot from the PLS analysis of biochemometric data. The ion for macrosphelide A (**4**) was the most correlated with the bioactive samples in the scores plot. (C) Selectivity ratio analysis of the PLS model data.

In summary, the identification of bioactive compounds without the need for multiple bioactivity-guided isolation steps remains an important goal to improve the efficiency and productivity of natural product discovery programs. The study presented herein has utilized multivariate statistical modeling coupled to the selectivity ratio (a univariate metric) to reveal compounds from a complex chemical profile that were responsible for the observed antibacterial bioactivity. Application of this biochemometric approach has led to the identification of a minor compound (altersetin, **3**) from *Alternaria* sp. with potent antibacterial activity against both *S. aureus* and MRSA. The same approach was applied with a second endophytic fungus, *Pyrenochaeta* sp., which revealed the bioactive compound macrosphelide A. Although these are both known compounds, they were not identified in the crude extracts because they were not included in the

database of experimental UPLC-MS data used for dereplication.<sup>(30)</sup> However, once it was determined (based on biochemometric analysis) that these ions were likely responsible for the biological activity of the extracts, their structures could be rapidly predicted by comparison of UPLC-MS data with published literature. Such literature searches would have been inefficient and impractical had they been conducted for all of the unique features identified by UPLC-HRMS in the *Alternaria* sp. and *Pyrenochaeta* sp. extracts (472 and 659 ions, respectively).

Importantly, by using biochemometrics to integrate chemical and biological data sets, the bioactive extract compounds (**3** and **4**) were identified as active very early in the isolation process, after just one stage of fractionation. For the purpose of the study presented here, subsequent fractionation and isolation steps were then conducted to confirm the predictions of the biochemometric analysis. In future studies, such follow-up isolation efforts might not be pursued if it was determined that the putative active compounds were of known structure and biological activity. In this way, biochemometrics could serve as a useful tool in dereplication efforts. The inclusion of biochemometrics in the dereplication process could prevent the rejection of an extract for further study based on the presence of a known (but inactive) compound, a potential pitfall of dereplication approaches that rely exclusively on chemical data. Additionally, as was the case with *Alternaria* sp., biochemometric analysis can point to the biological importance of a seemingly minor extract component, enabling focused efforts to rapidly solve its structure. The data presented here suggest that selectivity ratio analysis, which made better predictions than other data analysis procedures early in the fractionation process, could be a particularly effective tool for integrating biological and chemical data sets as part of dereplication efforts.

Another potentially important application of the biochemometrics approach is for integrating the chemical and bioassay data obtained from multiple fractionation steps. In the process of bioassay-guided fractionation, each stage of separation and bioassay data is typically considered in isolation from previous isolation steps. Using biochemometrics, it was possible to develop a model in which the active mixture components were predicted based on the data from several stages of fractionation in combination. Indeed, the study described herein shows that adding successive stages of fractionation and bioassay to the biochemometric analysis (original extract plus fractions and subfractions) improves the quality of the resulting selectivity plot. A future goal of our work is to apply biochemometrics using selectivity ratios to identify active components from more complex mixtures such as botanical extracts. Toward this goal, it may be necessary to conduct additional stages of purification to obtain sufficient sample size (number of objects) to accurately predict the compounds responsible for the observed biological activity. A highly complex extract could be fractionated and subjected to biochemometric analysis repeatedly, with each successive step of the fractionation and bioassay added into the data set until a quality selectivity ratio plot could be obtained.

## **Experimental Section**

### **General Experimental Procedures**



NMR spectra were acquired with a JEOL ECA-400 spectrometer (400 MHz) using DMSO-*d*<sub>6</sub>. Optical rotations were obtained using a Rudolph Research Autopol III polarimeter (Rudolph Research Analytical, Hackettstown, NJ, USA). UPLC-HRESIMS data were acquired using a Q Extractive Plus quadrupole-orbitrap mass spectrometer (Thermo Scientific, Waltham, MA, USA) with an electrospray ionization source coupled to an Acquity UPLC system (Waters, Milford, MA, USA). To collect UPLC-HRESIMS data, each sample was resuspended in MeOH to a concentration of 1 mg/mL, and triplicate 3 µL injections of each sample were performed. The samples were eluted from the column (Acquity UPLC BEH C<sub>18</sub> 1.7 µm, 2.1 × 50 mm, Waters) at a flow rate of 0.3 mL/min using the following binary gradient with solvent A consisting of H<sub>2</sub>O (0.1% formic acid added) and solvent B consisting of CH<sub>3</sub>CN (0.1% formic acid added): initial isocratic composition of 95:5 (A:B) for 1.0 min, increasing linearly to 0:100 over 20 min, followed by an isocratic hold at 0:100 for 1 min, gradient returned to starting conditions of 95:5 for 2 min, and held isocratically again for 1 min. The mass spectrometer was operated in the positive ionization mode over a scan range of 150–2000 with the following settings: capillary voltage set at 5 V, capillary temperature set at 300 °C, tube lens offset set at 35 V, spray voltage set at 3.80 kV, sheath gas flow set at 35, and auxiliary gas flow set at 20.

Flash chromatography separations were accomplished using an automated CombiFlash RF system (Teledyne-Isco, Lincoln, NE, USA) and monitored with a PDA detector and an evaporative light scattering detector. HPLC separations were performed on a Varian HPLC system (Agilent Technologies, Santa Clara, CA, USA) with Galaxie Chromatography Workstation software (version 1.9.3.2, Agilent Technologies). Analytical and preparative-scale HPLC separations employed a Gemini-NX C<sub>18</sub> column (5 µm, 110 Å, 250 × 4.60 mm (analytical) or 250 × 21.20 mm (preparative); Phenomenex, Torrance, CA, USA). Unless otherwise noted, all chemicals were of spectroscopic or microbiological grade and obtained from Sigma-Aldrich (St. Louis, MO, USA).

### **Plant Collection and Fungal Isolation**

Individual, asymptomatic goldenseal (*Hydrastis canadensis* L.) plants were collected in July 2010 from William Burch in Hendersonville, North Carolina (N 35°24.2770, W 082°20.9930). A voucher specimen was deposited at the herbarium of the University of North Carolina at Chapel Hill (NCU583414) and authenticated by Dr. Alan S. Weakly. Isolation of fungal endophytes was performed using methods outlined previously.(35, 36) Two strains, G28 (isolated from seeds) and G41 (isolated from leaf segments), were used in the present study. Axenic fungal cultures are maintained at 9 °C at the University of North Carolina at Greensboro, Department of Chemistry and Biochemistry Fungal Culture Collection.

### **Identification of Fungal Isolates**

For molecular identification of fungal endophytes isolated from goldenseal, the internal transcribed spacer region of the 5.8S rRNA gene (ITS1-5.8S-ITS2) was sequenced using methods described previously.(35-38) Based on a BLAST search conducted with published ITS data in NCBI GenBank, strain G28 was identified as an *Alternaria* sp. (Pleosporales, Dothideomycetes), while strain G41 was identified as a *Pyrenochaeta* sp. (Pleosporales,

Dothideomycetes), using cutoff proxies for ITS sequence similarity outlined previously.(36) The sequences from strains utilized in the present study were deposited in GenBank under accession numbers KT825854 (strain G28) and KT825855 (strain G41).

### **Solid-State Fermentation of Fungal Cultures**

For chemical extraction, the fungal strains utilized in this study were grown on a rice medium.(39) Briefly, seed cultures were started on the liquid medium composed of 2% soy peptone, 2% dextrose, and 1% yeast extract (YESD). The seed culture was grown for 7 days at 22 °C with agitation and subsequently transferred to 10 g of rice autoclaved with 25 mL of water in a 250 mL Erlenmeyer flask for screener cultures. For large-scale production of fungal cultures, four 250 mL Erlenmeyer flasks were inoculated using one seed culture for each flask. All rice cultures were allowed to grow for approximately 14–21 days prior to extraction.

### **Extraction and Isolation**

Cultures of *Alternaria* sp. (AS) and *Pyrenochaeta* sp. (PS) were extracted following the established procedure.(40) Briefly, to each culture flask was added 60 mL of 1:1 MeOH–CHCl<sub>3</sub>, which was chopped and shaken overnight (~20 h) at ~100 rpm at room temperature. The sample was vacuum-filtered, 90 mL of CHCl<sub>3</sub> and 150 mL of H<sub>2</sub>O were added, and the mixture was stirred for 30 min. This mixture was then transferred into a separatory funnel, and the bottom (CHCl<sub>3</sub>) layer collected. The CHCl<sub>3</sub> layer was evaporated to dryness, then dissolved in 100 mL of 1:1 MeOH–CH<sub>3</sub>CN and 100 mL of hexanes. The biphasic solution was shaken in a separatory funnel, and the bottom layer drawn off and evaporated to yield the crude extract (CR).

First-stage separations of the crude extract were conducted with normal-phase flash chromatography on a CombiFlash RF system with a 4 g silica gel column at 18 mL/min flow rate with a 40 min hexane–CHCl<sub>3</sub>–MeOH gradient, which yielded four fractions (AS-1 to AS-4 and PS-1 to PS-4, respectively) pooled based on LC-UV chromatograms. Active fractions were subjected to a second stage of purification using a reversed-phase preparative HPLC with a Gemini NX C<sub>18</sub> column at a 21.20 mL/min flow rate. A linear CH<sub>3</sub>CN–H<sub>2</sub>O (both with 0.1% formic acid) gradient starting from 40:60 to 100:0 over 15 min was employed, with fractions collected every 0.5 min and pooled based on both UV and evaporative light scattering detector chromatograms. The *Alternaria* sp. (AS) extract yielded compounds **1** and **3**, while **4** was isolated from the *Pyrenochaeta* sp. (PS) extract. Compound **2**, tenuazonic acid, was putatively identified based upon its high-resolution mass ( $m/z$  198.1134 [M + H]<sup>+</sup>, calcd for C<sub>10</sub>H<sub>16</sub>NO<sub>3</sub><sup>+</sup>, 198.1130), but was not present in sufficient quantities for isolation and confirmation of identity.

#### **Alternariol monomethyl ether (1):**

white solid; HRESIMS  $m/z$  273.0756 [M + H]<sup>+</sup> (calcd for C<sub>15</sub>H<sub>13</sub>O<sub>5</sub><sup>+</sup>, 273.0763); <sup>1</sup>H NMR (400 MHz DMSO-*d*<sub>6</sub>) and <sup>13</sup>C NMR (100 MHz DMSO-*d*<sub>6</sub>), chemical shifts were in agreement with literature values(31) and are provided as Supporting Information (Table S1 and Figure S2).

#### **Altersetin (3):**

pale brown solid; HRESIMS  $m/z$  400.2480  $[M + H]^+$  (calcd for  $C_{24}H_{34}NO_4^+$ , 400.2488);  $^1H$  NMR (400 MHz DMSO- $d_6$ ) and  $^{13}C$  NMR (100 MHz DMSO- $d_6$ ), chemical shifts were in agreement with literature values(31) and are provided as Supporting Information (Table S2 and Figure S4).

#### **Macrosphelide A (4):**

yellow solid;  $[\alpha]_D^{20} = +84$  ( $c$  0.60, MeOH); HRESIMS  $m/z$  343.1383  $[M + H]^+$  (calcd for  $C_{16}H_{23}O_8^+$ , 343.1393);  $^1H$  NMR (400 MHz  $CDCl_3$ ) and  $^{13}C$  NMR (100 MHz  $CDCl_3$ ), chemical shifts were in agreement with literature values(34) and are provided as Supporting Information (Table S3 and Figure S6).

#### **Antibacterial Assay**

Antibacterial activity was assessed via growth inhibition of a laboratory strain of *Staphylococcus aureus* (strain SA1199)(41) and methicillin-resistant *S. aureus* (MRSA USA300 LAC strain AH1263).(42) Cultures were grown from a single colony isolate of each strain to log-phase in Müller Hinton broth (MHB) and plated at a final density of  $1.0 \times 10^6$  cfu/mL.

For screening, samples were assayed in triplicate at a concentration of 10 and 100  $\mu g/mL$ . Samples were dissolved in 1:1 EtOH–DMSO ( $v/v$ ) and diluted in MHB to achieve the appropriate concentration, with ethanol and DMSO concentrations of <2%. The positive control used for the screening procedure was chloramphenicol, at the same concentrations as the samples (10 and 100  $\mu g/mL$ ). Vehicle was 2% 1:1 EtOH–DMSO. Each well was inoculated with bacterial culture and incubated at 37 °C for 24 h.

Minimum inhibitory concentration was measured according to Clinical Laboratory Standards Institute (CLSI) standard procedures.(43) Briefly, extracts or purified berberine (positive control, a known antibacterial compound from the host plant of these endophytic species, *Hydrastis canadensis* L.)(44) was added to 96-well plates in triplicate at concentrations ranging from 2.3 to 300  $\mu g/mL$  in MHB. Vehicle (2% DMSO) served as the negative control, and DMSO content was fixed at 2% in all wells. Absorbance at 600 nm was measured after 24 h using a Synergy HI microplate reader (Biotek, Winooski, VT, USA). The minimum inhibitory concentration was defined as the concentration at which there was no statistically significant difference between the treatment and vehicle control. The absorbance for replicate wells containing all assay components except bacteria was subtracted from the absorbance of assay wells.

#### **Biochemometric Analysis**

Triplicate LC-MS data sets for each sample were individually analyzed, aligned, and filtered with MZmine 2.17 software (<http://mzmine.sourceforge.net/>).(45) Peak detection in MZmine was achieved as follows:  $m/z$  values were detected within each spectrum above a baseline, a chromatogram was constructed for each of the  $m/z$  values that spanned longer than 0.1 min, and finally, deconvolution algorithms were applied to each chromatogram to recognize the individual chromatographic peaks. The parameters were set as follows for peak detection: noise level (absolute value) at  $1 \times 10^7$ , minimum peak duration 0.5 s, tolerance for  $m/z$  variation 0.05, and tolerance for  $m/z$  intensity variation 20%. Deisotoping, peak list filtering, and retention time

alignment algorithm packages were employed to refine peak detection. Finally, the join align algorithm compiled a peak table according to the following parameters: the balance between  $m/z$  and retention time was set at 10.0 each,  $m/z$  tolerance was set at 0.05, and retention time tolerance size was defined as 2 min. The spectral data matrix (comprised of  $m/z$ , retention time, and peak area for each peak) was imported to Excel (Microsoft, Redmond, WA, USA) and merged with the bioactivity data set (at 100  $\mu\text{g/mL}$  concentration) to form a final biochemometric analytical matrix. Triplicate data sets were included in the analysis for each sample, which consisted of three separate bioassay measurements and three separate UPLC-HRMS analyses for the same extract or fraction.

Biochemometric analysis was performed using Sirius version 9.0 (Pattern Recognition Systems AS, Bergen, Norway).<sup>(28, 33)</sup> Initially, transformation from heteroscedastic to homoscedastic noise was carried out by a fourth root transform of the spectral variables. An internally cross-validated PLS model was constructed using 100 iterations, at a significance level of 0.05. Selectivity ratios from the final PLS model were calculated using algorithms internal to Sirius.

### Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: [10.1021/acs.jnatprod.5b01014](https://doi.org/10.1021/acs.jnatprod.5b01014).

The authors declare no competing financial interest.

### Acknowledgment

This work was supported by the National Center for Complementary and Integrative Health, National Institutes of Health (grant 1R01 AT006860), and by a Biotechnology Research Grant (2011-BRG-1206) from the North Carolina Biotechnology Center. Mass spectrometry data were collected in the Triad Mass Spectrometry Facility. We thank S. Anike and Dr. A. Kaur for technical assistance and V. Sica for assistance with manuscript editing.

### References

1. Kinghorn, A. D.; Fong, H. H. S.; Farnsworth, N. R.; Mehta, R. G.; Moon, R. C.; Moriarty, R. M.; Pezzuto, J. M. *Curr. Org. Chem.* **1998**, *2*, 597– 612
2. Wall, M. E.; Wani, M. C. *J. Ethnopharmacol.* **1996**, *51*, 239– 253 DOI: 10.1016/0378-8741(95)01367-9
3. Oberlies, N. H.; Kroll, D. J. *J. Nat. Prod.* **2004**, *67*, 129– 135 DOI: 10.1021/np030498t
4. Tu, Y. *Nat. Med.* **2011**, *17*, 1217– 1220 DOI: 10.1038/nm.2471
5. Noble, R. L. *Biochem. Cell Biol.* **1990**, *68*, 1344– 1351 DOI: 10.1139/o90-197
6. Weller, M. G. *Sensors* **2012**, *12*, 9181– 9209 DOI: 10.3390/s120709181
7. Prince, E. K.; Pohnert, G. *Anal. Bioanal. Chem.* **2010**, *396*, 193– 197 DOI: 10.1007/s00216-009-3162-5

8. Inui, T.; Wang, Y.; Pro, S. M.; Franzblau, S. G.; Pauli, G.  
*F. Fitoterapia* **2012**, 83, 1218– 1225 DOI: 10.1016/j.fitote.2012.06.012
9. Qui, F.; Cai, G.; Jaki, B. U.; Lankin, D. C.; Franzblau, S. G.; Pauli, G. F. *J. Nat. Prod.* **2013**, 76, 413– 419 DOI: 10.1021/np3007809
10. Clendinen, C. S.; Pasquel, C.; Arjredini, R.; Edison, A. S. *Anal. Chem.* **2015**, 87, 5689– 5706 DOI: 10.1021/acs.analchem.5b00867
11. Kleigrew, K.; Almaliti, J.; Tian, I. Y.; Kinnel, R. B.; Korobeynikov, A.; Monroe, E. A.; Duggan, B. M.; Di Marzo, V.; Sherman, D. H.; Dorrestein, P. C.; Gerwick, L.; Gerwick, W. H. *J. Nat. Prod.* **2015**, 78, 1671–1682 DOI: 10.1021/acs.jnatprod.5b00301
12. Wu, C.; Du, C.; Gubbens, J.; Choi, Y. H.; van Wezel, G. P. *J. Nat. Prod.* **2015**, 78, 2355– 2363 DOI: 10.1021/acs.jnatprod.5b00276
13. Cortina, N. S.; Krug, D.; Plaza, A.; Revermann, O.; Müller, R. *Angew. Chem., Int. Ed.* **2012**, 51, 811– 816 DOI: 10.1002/anie.201106305
14. Sidebottom, A. M.; Johnson, A. R.; Karty, J. A.; Trader, D. J.; Carlson, E. E. *ACS Chem. Biol.* **2013**, 8, 2009– 2016 DOI: 10.1021/cb4002798
15. Hou, Y.; Braun, D. R.; Michel, C. R.; Klassen, J. L.; Adnani, N.; Wyche, T. P.; Bugni, T. S. *Anal. Chem.* **2012**, 84, 4277– 4283 DOI: 10.1021/ac202623g
16. Yuliana, N. D.; Khatib, A.; Choi, Y. H.; Verpoorte, R. *Phytother. Res.* **2010**, 25, 157– 169 DOI: 10.1002/ptr.3258
17. Okada, T.; Afendi, F. M.; Katoh, A.; Hirai, A.; Kanaya, S. In *Biotechnology for Medicinal Plants*; Chandra, S.; Lata, H.; Varma, A., Eds.; Springer-Verlag: Berlin, **2013**; pp 413– 438.
18. Rajalahti, T.; Kvalheim, O. M. *Int. J. Pharm.* **2011**, 417, 280– 290 DOI: 10.1016/j.ijpharm.2011.02.019
19. Choi, Y. H.; Kim, H. K.; Linthorst, H. J. M.; Hollander, J. G.; Lefeber, A. W. M.; Erkelens, C.; Nuzillard, J.; Verpoorte, R. *J. Nat. Prod.* **2006**, 69, 742– 748 DOI: 10.1021/np050535b
20. Martens, H.; Bruun, S. W.; Adt, I.; Sockalingum, G. D.; Kokhler, A. *J. Chemom.* **2006**, 20, 402– 417 DOI: 10.1002/cem.1015
21. Cox, D. G.; Oh, J.; Keasling, A.; Colson, K. L.; Hamann, M. T. *Biochim. Biophys. Acta, Gen. Subj.* **2014**, 1840, 3460– 3474 DOI: 10.1016/j.bbagen.2014.08.007
22. Ali, K.; Iqbal, M.; Yuliana, N. D.; Lee, Y. J.; Park, S.; Han, S.; Lee, J. W.; Lee, H. S.; Verpoorte, R.; Choi, Y. H. *Metabolomics* **2013**, 9, 778– 785 DOI: 10.1007/s11306-013-0498-9

23. Rajalahti, T.; Arneberg, R.; Kroksveen, A. C.; Berle, M.; Myhr, K.; Kvalheim, O. M. *Anal. Chem.* **2009**, 81,2581– 2590 DOI: 10.1021/ac802514y
24. Wiklund, S.; Johansson, E.; Sjöström, L.; Mellerowicz, E. J.; Edlund, U.; Shockcor, J. P.; Gottfries, J.;Moritz, T.; Trygg, J. *Anal. Chem.* **2008**, 80, 115– 122 DOI: 10.1021/ac0713510
25. Kulakowski, D. M.; Wu, S.-B.; Balick, M. J.; Kennelly, E. J. *J. Chromatogr. A* **2014**, 1364, 74– 82 DOI: 10.1016/j.chroma.2014.08.049
26. Shang, N.; Saleem, A.; Musallam, L.; Walshe-Roussel, B.; Badawi, A.; Cuerrier, A.; Arnason, J. T.;Haddad, P. S. *PLoS One* **2015**, 10, e0135721 DOI: 10.1371/journal.pone.0135721
27. Rajalahti, T.; Arneberg, R.; Berven, F. S.; Myhr, K.; Ulvik, R. J.; Kvalheim, O. M. *Chemom. Intell. Lab. Syst.***2009**, 95, 35– 48 DOI: 10.1016/j.chemolab.2008.08.004
28. Kvalheim, O. M.; Karstang, T. V. *Chemom. Intell. Lab. Syst.* **1989**, 7, 39– 51 DOI: 10.1016/0169-7439(89)80110-8
29. Chau, F.; Chan, H.; Cheung, C.; Xu, C.; Liang, Y.; Kvalheim, O. M. *Anal. Chem.* **2009**, 81, 7217– 7225 DOI: 10.1021/ac900731z
30. El-Elimat, T.; Figueroa, M.; Ehrmann, B. M.; Cech, N. B.; Pearce, C.; Oberlies, N. H. *J. Nat. Prod.* **2013**, 76,1709– 1716 DOI: 10.1021/np4004307
31. Hellwig, V.; Grothe, T.; Mayer-Bartschmid, A.; Endermann, R.; Geschke, F.; Henkel, T.; Stadler, M. *J. Antibiot.* **2002**, 55, 881– 892 DOI: 10.7164/antibiotics.55.881
32. Tan, N.; Tao, Y.; Pan, J.; Wang, S. Q.; Xu, F.; She, Z.; Lin, Y.-J.; Jones, E. B. G. *Chem. Nat. Compd.* **2008**,44, 296– 300 DOI: 10.1007/s10600-008-9046-7
33. Kvalheim, O. M.; Chan, H.; Benzie, I. F. F.; Szeto, Y.; Tzang, A. H.; Mok, D. K.; Chau, F. *Chemom. Intell. Lab. Syst.* **2011**, 107, 98– 105 DOI: 10.1016/j.chemolab.2011.02.002
34. Takamatsu, S.; Kim, Y.-P.; Hayashi, M.; Hiraoka, H.; Natori, M.; Komiyama, K.; Omura, S. *J. Antibiot.* **1995**,49, 95– 98 DOI: 10.7164/antibiotics.49.95
35. Figueroa, M.; Jarmusch, A. K.; Raja, H. A.; El-Elimat, T.; Kavanaugh, J. S.; Horswill, A. R.; Cooks, R. G.;Cech, N. B.; Oberlies, N. H. *J. Nat. Prod.* **2014**, 77, 1351– 1358 DOI: 10.1021/np5000704
36. Raja, H. A.; Kaur, A.; El-Elimat, T.; Figueroa, M.; Kumar, R.; Deep, G.; Agarwal, R.; Faeth, S. H.;Cech, N. B.; Oberlies, N. H. *Mycology* **2015**, 6, 8– 27 DOI: 10.1080/21501203.2015.1009186

37. El-Elimat, T.; Raja, H. A.; Day, C. S.; Chen, W.-L.; Swanson, S. M.; Oberlies, N. H. *J. Nat. Prod.* **2014**, *77*, 2088– 2098 DOI: 10.1021/np500497r
38. Raja, H. A.; El-Elimat, T.; Oberlies, N. H.; Shearer, C. A.; Miller, A. N.; Tanaka, K.; Hashimoto, A.; Fournier, J. *Mycologia* **2015**, *107*, 845– 862 DOI: 10.3852/15-013
39. VanderMolen, K. M.; Raja, H. A.; El-Elimat, T.; Oberlies, N. H. *AMB Express* **2013**, *3*, 71– 71 DOI: 10.1186/2191-0855-3-71
40. El-Elimat, T.; Raja, H. A.; Graf, T. N.; Faeth, S. H.; Cech, N. B.; Oberlies, N. H. *J. Nat. Prod.* **2014**, *77*, 193–199 DOI: 10.1021/np400955q
41. Kaatz, G. W.; Seo, S. M. *Antimicrob. Agents Chemother.* **1995**, *39*, 2650– 2655 DOI: 10.1128/AAC.39.12.2650
42. Junio, H. A.; Todd, D. A.; Ettefagh, K. A.; Ehrmann, B. M.; Kavanaugh, J. S.; Horswill, A. R.; Cech, N. B. *J. Chromatogr. B: Anal. Technol. Biomed. Life Sci.* **2013**, *930*, 7– 12 DOI: 10.1016/j.jchromb.2013.04.019
43. National Committee for Clinical Laboratory Standards. *Methods for Dilution Antimicrobial Susceptibility Tests for Bacteria That Grow Aerobically*, 3rd ed.; Villanova, PA, **1993**.
44. Junio, H. A.; Sy-Cordero, A. A.; Ettefagh, K. A.; Burns, J. T.; Micko, K. T.; Graf, T. N.; Richter, S. J.; Cannon, R. E.; Oberlies, N. H.; Cech, N. B. *J. Nat. Prod.* **2011**, *74*, 1621– 1629 DOI: 10.1021/np200336g
45. Pluskal, T.; Castillo, S.; Villar-Briones, A.; Orešič, M. *BMC Bioinf.* **2010**, *11*, 395 DOI: 10.1186/1471-2105-11-395