# Original paper Raman spectra of minerals containing interconnected As(Sb)O<sub>3</sub> pyramids: trippkeite and schafarzikite

## Sherif KHARBISH

Geology Department, Faculty of Science, Suez Canal University, Suez Branch, Suez Governate, El Salam City, 44634, Egypt; sherifkharbish@hotmail.com, Sherif\_abdallah@s-science.suez.edu.eg



Oriented single-crystals of trippkeite ( $CuAs_2O_4$ ) and an isostructural mineral schafarzikite ( $FeSb_2O_4$ ) were investigated by polarized Raman spectroscopy. The internal vibrations, i.e. the stretching and bending modes, of trippkeite occur between 250 and 850 cm<sup>-1</sup>, and those of schafarzikite between 200 and 750 cm<sup>-1</sup>. The higher mass and the longer bond distances of the SbO<sub>3</sub> groups readily explain the lower wavenumbers of the latter. Comparing the spectra of minerals containing XO<sub>3</sub> pyramids with those of the minerals from the present study, approximate similarities in the spectral features are evident. A clear distinction between Raman spectra of separated and interconnected (As,Sb)O<sub>3</sub> groups is not observed.

*Keywords: oriented single-crystals, pyramidal (As,Sb)O<sub>3</sub> groups, Raman spectroscopy, schafarzikite, trippkeite Received: 7 December 2011; accepted: 8 March 2012; handling editor: J. Sejkora* 

## 1. Introduction

Trippkeite (CuAs<sub>2</sub> $O_4$ ) and an isostructural schafarzikite  $(FeSb_{2}O_{4})$  are relatively rare in nature. Both minerals belong to the group of compounds with a general formula AB<sub>2</sub>O<sub>4</sub>, where A represents a divalent metal ion such as Pb, Cu, Sn, Ni, Zn, Mn, Fe and/or Mg, whereas the B represents trivalent metal ions like Pb, As, Sb Al, Cr and/or Mg (Hinrichsen et al. 2006). The trippkeite possesses AsO<sub>3</sub> chain polymer (Lee and Harrison 2004), which was first definitively characterized by Zemann (1951b). Pertlik (1988) also showed that the two synthetic lead chain-arsenite chlorides Pb(AsO<sub>2</sub>)Cl and Pb<sub>2</sub>(AsO<sub>2</sub>)<sub>3</sub>Cl contain the same AsO<sub>3</sub> chain polymer, as does the mineral leiteite, ZnAs<sub>2</sub>O<sub>4</sub> (Ghose et al. 1987). Schafarzikite is a meta-antimonite mineral containing polymeric SbO<sub>2</sub> pyramids (Krenner 1921; Zemann 1951a; Fischer and Pertlik 1975; Sejkora et al. 2007). Despite the work that has been done on the crystal structures of trippkeite and schafarzikite, Raman spectroscopic studies, at least of oriented trippkeite, have not been published so far. In general, there exist very few works dealing with Raman spectroscopy of "not oriented" trippkeite (Bahfenne 2011; Bahfenne et al. 2011) or "not oriented" (Sejkora et al. 2007) and "oriented" schafarzikite (Bahfenne 2011; Bahfenne et al. 2011).

The purpose of the present work is therefore (1) to investigate oriented crystals of synthetic trippkeite and naturally occurring schafarzikite by Raman spectroscopy, (2) to examine the spectral changes related to crystal structural features, such as bond distances and (As,Sb)O<sub>3</sub>

groups of reduced (non-trigonal) symmetry (all minerals of the present study) vs. isolated pyramidal (As,Sb)O<sub>3</sub> groups of ideal trigonal symmetry.

## 2. Crystallographic data

Trippkeite (CuAs<sub>2</sub>O<sub>4</sub>) and schafarzikite (FeSb<sub>2</sub>O<sub>4</sub>) crystallize in space group  $P4_2$ /mbc ( $D_{4h}^{13}$ ) with four formula units per unit cell (Z = 4). The structure of trippkeite was first solved by Zemann (1951b) and was refined by Pertlik (1975). Eby and Hawthorne (1993) mentioned that trippkeite has rutile-like chains. The trippkeite crystal structure is illustrated in Fig. 1a. There are two kinds of O atoms, O(1) and O(2). The Cu atoms are surrounded octahedrally by six O atoms [four O(2) and two O(1)] (Pertlik 1975). The As atoms are coordinated by three O atoms [two O(1) and one O(2)] to form a flat trigonal pyramid (Fig. 1a) (Zemann 1951b; Pertlik 1975). The As–O(1) and As–O(2) interatomic distances in trippkeite are 1.814 (2x) and 1.765 Å, respectively (Pertlik 1975). The O(1)-As-O(2) and the O(1)-As-O(1) angles are  $95.9^{\circ}$  (2x) and 100.3°, respectively (Pertlik 1975). The As polyhedra are connected by common corners, building chains parallel to the *c*-axis, into which the lone pairs of the As<sup>3+</sup> point (Pertlik 1975; Eby and Hawthorne 1993).

The structure of schafarzikite, containing small amounts of Mn<sup>2+</sup>, has been determined by Zemann (1951a) and refined by Fischer and Pertlik (1975). The crystal structure of schafarzikite is built up by columns, running along (001), of edge-sharing Fe<sup>2+</sup> octahedra; parallel to these are chains of corner-sharing Sb<sup>3+</sup>

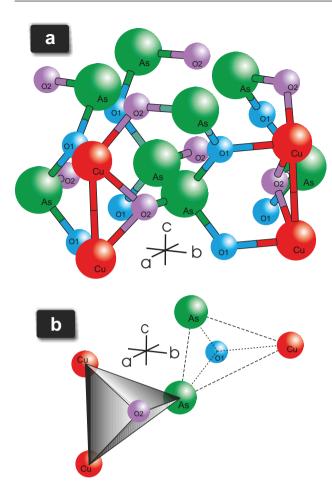


Fig. 1 Crystal structure of trippkeite redrawn after Pertlik (1975) (a), and the interconnection of the oxygen atoms to the surrounding ones (b).

w-tetrahedra (Mellini and Merlino 1979), namely trigonal pyramids with a lone pair as fourth ligand (Cotton and Wilkinson 1966). All the O atoms of each tetrahedral chain lie on one plane, parallel to (110); successive Sb atoms of the chain are located on opposite sides of this plane (Fischer and Pertlik 1975). The connection between chains and octahedral columns builds up a framework with open channels parallel to (001) (Mellini and Merlino 1979). The diameter of the channels is related to the distance (4.1 Å) between antimony atoms lying on opposite sides of the channels and pointing their lone pairs one toward the other (Mellini and Merlino 1979). Similar to the trippkeite, each Sb in schafarzikite is connected to two O(1) and one O(2) and each Fe is connected to four O(2)and two O(1). The Sb-O(1) and Sb-O(2) interatomic distances in schafarzikite are 1.987 (2x) and 1.917 Å, respectively, whereas the O(1)-Sb-O(2) and O(1)-Sb-O(2)angles, 93.5° (2x) and 96.0°, respectively (Fischer and Pertlik 1975).

Inspection of the crystal structure of the trippkeite and schafarzikite reveals two kinds of O atoms [i.e. O(1) and O(2)]. O(1) atom is bonded to two As(Sb) and one Cu(Fe) atoms, forming a planar triangular group (Fig. 1b), whereas O(2) atom is connected to two Cu(Fe) and one As(Sb) atoms, forming a pyramidal group (Fig. 1b) (Fischer and Pertlik 1975; Pertlik 1975). Therefore, the As(Sb)O<sub>3</sub> pyramids in trippkeite and schafarzikite should be considered as interconnected group (i.e. all the O atoms are bridging atoms). However, in synthetic oxo-antimonates and -arsenates, the O atoms with longer interatomic distances are considered as bridging and those with shorter interatomic distances as terminal atoms (Hirschle and Röhr 2000; Emmerling and Röhr 2003).

#### 3. Samples and experimental

Raman spectroscopic measurements were done on synthetic trippkeite and naturally occurring schafarzikite samples. The synthetic trippkeite samples were kindly provided by Prof. Pertlik, Vienna. The natural schafarzikite samples (Pernek, Slovakia) are from the collection of the Institute of Mineralogy and Crystallography, University of Vienna. The trippkeite and schafarzikite samples are part of the original material that has been used for the crystal structure refinement by Zemann (1951a, b), Fischer and Pertlik (1975) and Pertlik (1975).

The unit-cell parameters of the trippkeite and schafarzikite were determined on a single-crystal X-ray Nonius Kappa four-circle diffractometer, equipped with a CCD detector and a 300 µm capillary optics collimator. Data were acquired using the program Collect, processed with the Nonius software suite DENZO-SMN (Otwinowski and Minor 1997) and corrected for Lorentz, polarization and background effects. The fullmatrix least-squares refinement was undertaken using the program SHELXTL 6.12 (Bruker Analytical X-ray Solutions), using neutral-atom scattering factors and corrections for anomalous dispersion. No change in the reflections was noticed.

Carefully separated single grains of trippkeite and schafarzikite were (i) oriented according to morphology, and then (ii) mounted on glass rods and oriented by using a Nonius Kappa CCD diffractometer.

The chemical compositions of trippkeite and schafarzikite samples were measured on a Jeol JSM-6400 scanning electron microscope equipped with a Link EDX unit. Cobalt was used for internal gain calibration. An acceleration voltage of 20 kV was applied, the channel width was set to 20 eV, matrix absorption and fluorescence effects were corrected for by the ZAF-4 algorithm (Link Analytical). The analyses were accurate to 1 %.

Raman spectra were collected with the laser polarization parallel to the *a*- and *c*-axes for the tetragonal trippkeite and schafarzikite minerals. The spectra of all minerals were acquired in the spectral range from 50 to 1200 cm<sup>-1</sup> by using a Renishaw RM1000 confocal notch filter-based micro-Raman system. The 785 nm excitation line of a 17 mW solid-state laser (attenuated to 4.25 mW for schafarzikite) with polarization extinction > 500:1 was focused with a  $50 \times /0.75$  objective on the sample surface. The laser polarization was parallel to the a- and c-axes. The back-scattered radiation (180° configuration) was analyzed with a 1200 lines/mm grating monochromator in the so-called "static grating scan" data collection mode. Raman intensities were collected with a thermo-electrically cooled CCD array detector. The resolution of the system ("apparatus function") was  $2 \text{ cm}^{-1}$  and the wavenumber accuracy was  $\pm 1 \text{ cm}^{-1}$  (both calibrated with the Rayleigh line and the 520.5 cm<sup>-1</sup> line of a Si standard). Instrument control and data acquisition were done with Grams/32 software (Galactic Ind. Corp.). Depending on the signal intensity, accumulations with 60-120 s for trippkeite and 60-90 s for schafarzikite per "spectral window" (i.e., exposure time of the detector) were measured. The peak centers of single bands were determined by fits of combined Gaussian/Lorentzian amplitude functions using the PeakFit 4.12 software (Jandel Scientific).

#### 4. Results

#### 4.1. Sample characterizations

Trippkeite crystals are short prismatic to equant in shape and greenish blue in color (pale bluish green in transmitted light); they show a vitreous luster. The unit-cell parameters for trippkeite are a = 8.5941(3) and c = 5.5728(4) Å. Schafarzikite crystals are prismatic, dark brown to black in color and have an adamantine to metallic luster with unit-cell parameters a = 8.6012(3) and c = 5.9133(4) Å. The newly refined unit-cell dimensions are in an excellent agreement with those given by Pertlik (1975) for trippkeite (a = 8.59 and c = 5.57 Å) as well as by Fischer and Pertlik (1975) and Sejkora et al. (2007) for schafarzikite (a = 8.59 and c = 5.91 Å). The structural formulae calculated from the chemical analyses on the basis of four oxygen atoms per formula unit are Cu<sub>1</sub>.05As<sub>1.95</sub>O<sub>4</sub> (trippkeite) and (Fe<sub>0.9</sub>Mn<sub>0.1</sub>)Sb<sub>2.0</sub>O<sub>4</sub> (schafarzikite).

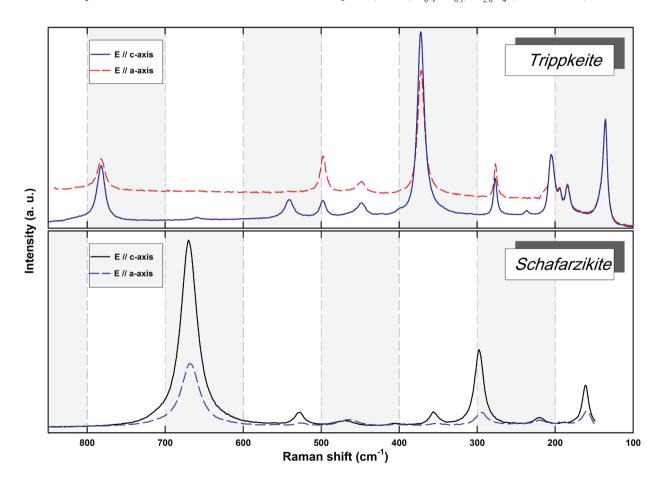


Fig. 2 Raman spectra and vibration modes of trippkeite and schafarzikite. E // c- axis, solid line; E // a-axis, dashed line.

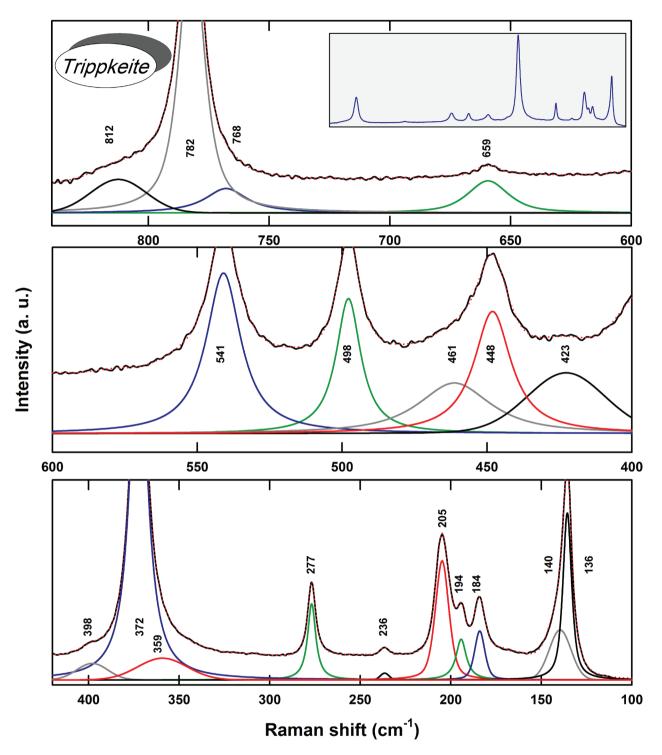


Fig. 3 The positions of the bands (cm<sup>-1</sup>) of trippkeite obtained by using the PeakFit program. Goodness of fit ( $r^2$ ) > 0.998.

### 4.2. Raman spectra of trippkeite

The bands in the trippkeite spectra parallel to the *a*and *c*-axes occur in the vibrational region between 100 and 850 cm<sup>-1</sup> (Fig. 2). No additional bands were observed in lower or higher wavenumber regions. There is nearly no difference in the band numbers and positions between the trippkeite spectra obtained parallel to the a- and caxes (Fig. 2). However, the band intensities measured parallel to the c-axis are significantly greater than those measured parallel to the a-axis. Raman spectra are characterized by a very strong band at 372 cm<sup>-1</sup> and strong band at 782 cm<sup>-1</sup>. In the spectra of trippkeite (parallel to the *a*- and *c*-axes), four vibration regions can be also easily distinguished (Fig. 2). The first occurs between 850 and 600 cm<sup>-1</sup>. Peak fitting reveals four medium, weak to very weak bands at 812, 782, 768 and 659 cm<sup>-1</sup> (Fig. 3). Five weak, medium to very strong bands and shoulders at 541, 498, 461, 448 and 423 cm<sup>-1</sup> are noticed in the second region (i.e. between 600 and 400 cm<sup>-1</sup>). The fitting of the third region between 400 and 250 cm<sup>-1</sup> reveals four very strong, medium to very weak vibrational modes at 398, 372, 359 and 277 cm<sup>-1</sup>. The last region appears below 250 cm<sup>-1</sup>.

#### 4.3. Raman spectra of schafarzikite

The spectra of schafarzikite were also collected parallel to the a- and c-axes (Fig. 2). The bands occur in the vibrational region between 150 and 750 cm<sup>-1</sup> (Fig. 2). The band numbers and positions measured parallel to the a- and c-axes are identical. Similar to the case of trippkeite, the intensities of the bands measured parallel to the *c*-axis are also significantly greater than those observed perpendicular to the c-axis. The spectra of schafarzikite are characterized by a very intense band at 668 cm<sup>-1</sup> and strong band at 295 cm<sup>-1</sup>. In the spectra measured in both directions, four vibration regions can be easily observed (Fig. 2). The first region, between 750 and 500 cm<sup>-1</sup>, contains a very strong sharp band at 668 cm<sup>-1</sup> and weak bands and/or shoulders at 707, 555 and 526 cm<sup>-1</sup> (Fig. 4). The second region occurs between 500 and 300 cm<sup>-1</sup>. Band fitting reveals four medium, weak to very weak bands at 464, 405, 353 and 347 cm<sup>-1</sup>. The third region (300–200 cm<sup>-1</sup>) contains five medium to very weak modes at 295, 275, 253, 229 and 218 cm<sup>-1</sup>. The last region ( $< 200 \text{ cm}^{-1}$ ) contains four medium to very weak modes.

#### 5. Discussion and conclusions

Raman bands of trippkeite as well as of schafarzikite samples were observed between 100 and 850 cm<sup>-1</sup> (Fig. 2, Tab. 1), with the low-energy limit determined only by the cut-off value of the employed Raman notch and edge filters. No additional bands were observed in lower or higher wave number regions. The band positions of both minerals agree with the characteristic spectral regions of oxides (700–300 cm<sup>-1</sup>) (Wang et al. 1994).

According to the "harmonic oscillator model" and Hooke's law, the frequency of vibrational modes in isostructural phases is shifted with the substitution of one element by another (or even one isotope by another). Thus, bands move to higher wave numbers with increasing bond strength (e.g. shorter bond lengths) and reduced mass. The shift of the vibrational regions between trippkeite and schafarzikite (Fig. 2) is attributed to the difference in the atomic masses between As and Sb, which simply explains the appearance of the trippkeite vibrational bands at wave numbers higher than in case of schafarzikite. On the other hand, band shifts can be related to differences in bonding behavior and, most evidently, to different bond lengths. Thus, the shorter As–O bonds in trippkeite (1.814 (2x), 1.765Å; as recorded by Pertlik 1975) should result in vibrational modes at higher wave numbers than the longer Sb–O bonds in schafarzikite (1.987 (2x), 1.917Å; as recorded by Fischer and Pertlik 1975).

Inspection of the spectra of the two studied minerals (Fig. 2) reveals that the band numbers and positions are not affected by the polarization directions of the incident laser beam. However, the band intensities are affected considerably and maximum intensities are observed parallel to the *c*-axis (Fig. 2). An apparent reason for this behavior is that the band intensities depend on the direction of the molecular axes of the (AsSb)O<sub>3</sub> pyramids, which are all aligned parallel to the *c*-axis, pointing only in one direction. In other words, the *c*-direction provides the strongest change in polarizability, i.e. the strongest component of the Raman tensor.

The trippkeite and schafarzikite minerals should exhibit the characteristic vibrational modes of (As,Sb) O<sub>3</sub> groups and further lattice modes of the mineral constituents. According to Loehr and Plane (1968) and Nakamoto (1997), an isolated ideal pyramidal (As,Sb)O<sub>2</sub> group, which belongs to the  $C_{3\nu}$  point group, is expected to show only four normal modes of vibration (i.e.  $2A_1 +$ 2E). All four modes of vibration of a pyramidal (As,Sb) O<sub>2</sub> group are IR- and Raman active (Nakamoto 1997). The symmetric (As,Sb)-O stretching mode is expected to lie at a higher frequency than the asymmetric mode (Loehr and Plane 1968). The two stretching frequencies symmetric  $v_1$  (A<sub>1</sub>) and antisymmetric  $v_3$  (E), as well as the two bending frequencies, symmetric  $v_2(A_1)$  and antisymmetric  $v_{4}(E)$ , overlap or are close in energy in most of similar compounds showing a sequence in wave number, i.e.  $v_1 > v_3 > v_2 > v_4$  (Nakamoto 1997).

Although only four normal modes of vibration (i.e.  $2A_1 + 2E$ ) should occur in the spectra of trippkeite and schafarzikite according to Loehr and Plane (1968) and Nakamoto (1997), both minerals show additional bands. The question why the number of bands increases can be explained if it is considered to be a function of the symmetry of the (As,Sb)O<sub>3</sub> pyramids. In the investigated minerals the symmetry of the pyramidal (As,Sb) O<sub>3</sub> groups is reduced to C<sub>s</sub>. On a C<sub>s</sub> symmetry A<sub>1</sub> modes translate into A' and E modes split to A' and A" (Nakamoto 1997; Bahfenne et al. 2011). Correlating this to a D<sub>4h</sub> crystal system splits each A' mode to A<sub>1g</sub>, A<sub>2g</sub>, B<sub>1g</sub>, B<sub>2g</sub> and 2E<sub>u</sub> and each A" mode to A<sub>1u</sub>, A<sub>2u</sub>, B<sub>1u</sub>, B<sub>2u</sub> and

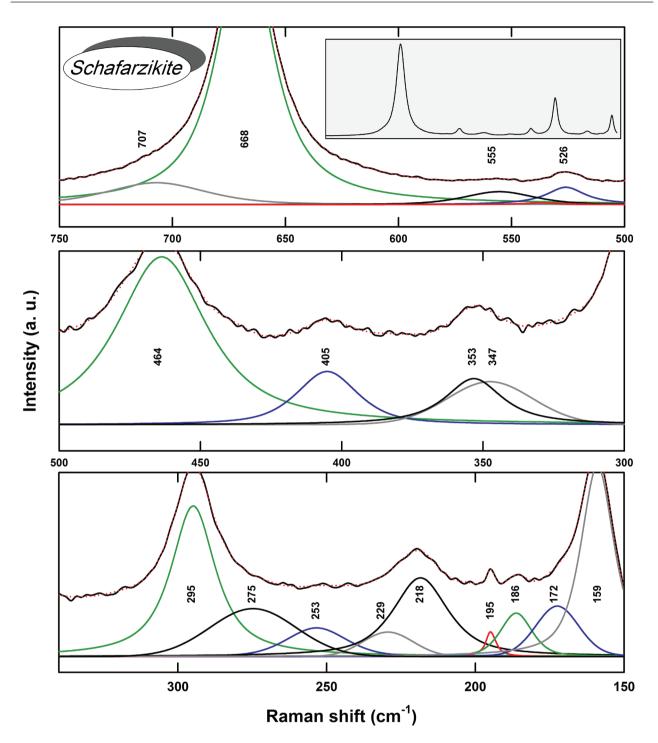


Fig. 4 The positions of the bands (cm<sup>-1</sup>) of schafarzikite obtained by using the PeakFit program. Goodness of fit ( $r^2$ ) > 0.997

 $2E_{\rm g}$ . Only the g vibrational modes (i.e.  $A_{\rm 1g}$ ,  $B_{\rm 1g}$ ,  $B_{\rm 2g}$  and  $2E_{\rm g}$ ) are Raman active, whereas the u modes are either inactive or IR active (Fig. 5 shows the splitting pattern of the A<sub>1</sub> and *E* modes).

Band assignments (Tab. 1) are based on the sequence of band energies given by Nakamoto (1997) with  $v_1 > v_3$ 

 $> v_2 > v_4$  and are also facilitated by comparing the band positions with those given by Bahfenne (2011), Bahfenne et al. (2011) and Sejkora et al. (2007) for trippkeite and schafarzikite and those presented by Hirschle and Röhr (2000) and Emmerling and Röhr (2003) for oxo-arsenates and -antimonates.

			trippkeit	e		schafarzikite						
		Symmetry	Band cm <sup>-1</sup>	FWHM	Band cm <sup>-1</sup>		Symmetry	Band cm <sup>-1</sup>	FWHM	Band cm <sup>-1</sup>		
			Present study		$T^1$	$T^2$		Present study		$S^1$	$S^2$	$S^3$
As(Sb)–O symmetric stret- ching As(Sb)–O	<i>v</i> <sub>1</sub>	$\begin{array}{c} A_{1g} \\ A_{2g} \\ B_{1g} \end{array}$	812vw 782s 768vw <sub>s</sub>	25.05 11.95 21.00	810 780	810 780	$\begin{array}{c} A_{1g} \\ A_{2g} \end{array}$	707vw <sub>s</sub> 668vs	51.65 27.87	709 668 617	708 667 616	670 617
antisymmetric stretching	<i>v</i> <sub>3</sub>	A <sub>1g</sub> A <sub>1g</sub>	659w 541m	20.06 14.47	657 539	657 539	$\begin{array}{c} A_{_{1g}} \\ A_{_{2g}} \end{array}$	555vw 526w	36.20 20.86	558 526 478	557 525	524
As(Sb)–O–(Sb) As symmetric bending	<i>v</i> <sub>2</sub>	$\begin{array}{c} \mathbf{A}_{2g} \\ \mathbf{B}_{1g} \\ \mathbf{B}_{2g} \\ \mathbf{P}_{2g} \end{array}$	498m 461vw <sub>s</sub> 448w 423vw	10.32 29.95 15.46 30.09	496 421	496 421	$\begin{array}{c} A_{1g} \\ A_{2g} \\ B_{1g} \\ B_{2g} \end{array}$	464m 405w 353w 347yw	38.20 25.63 24.40 39.98	465 403 353 345	465 402 352 344	467 405 348
As(Sb)–O–(Sb) As antisymmetric bending	<i>v</i> <sub>4</sub>	$\begin{array}{c} A_{1g} \\ A_{2g} \\ B_{1g} \\ B_{2g} \end{array}$	398vw 372vs 359vw 227s	20.30 10.73 32.21 5.38	371	371	$\begin{array}{c} \overset{2_{2g}}{A_{1g}}\\ \overset{A_{2g}}{B_{1g}}\\ \overset{B_{2g}}{B_{2g}}\\ \overset{E_{g}}{E_{g}}\end{array}$	295m 275w <sub>s</sub> 253vw 229w 218w	17.81 34.10 23.60 19.99 21.99	295 249 219	295 249 219 208	295 217
Lattice modes			236vw 205s 194m 184m 140vw <sub>s</sub> 136s	5.69 8.86 6.38 3.66 12.20 5.70				195w 186w 172w <sub>s</sub> 159m	3.99 12.44 17.12 13.26	208 186 158 132 119	186 159 131	
Lattice modes			194m 184m	6.38 3.66				186w 172w <sub>s</sub>	12.44 17.12	158		159 131

Tab. 1 Band assignment, band positions (cm<sup>-1</sup>) and FWHM for the Raman spectra of trippkeite and schafarzikite in comparison with literature data

vs = very strong, s = strong, m = medium, w = weak, vw = very weak,  $w_s$  = weak shoulder,  $vw_s$  = very weak shoulder (average relative band intensities from all polarization directions).

T<sup>1</sup> trippkeite data and S<sup>1</sup> schafarzikite data after Bahfenne (2011)

T<sup>2</sup> trippkeite data and S<sup>2</sup> schafarzikite data after Bahfenne et al. (2011)

S<sup>3</sup> schafarzikite data after Sejkora et al. (2007)

In trippkeite (Figs 2–3) the strong band at 782 cm<sup>-1</sup> and two very low intensity shoulders at 812 and 768 cm<sup>-1</sup> were assigned to the symmetric  $v_1$  (A<sub>1</sub>) As–O stretching modes (Tab. 1). This band triplet was also assigned to  $A_{2g}$ ,  $A_{1g}$  and  $B_{1g}$ , respectively (Tab. 1). The very weak band at 659 cm<sup>-1</sup> was assigned to the antisymmetric  $v_3$ (*E*) stretching mode and also to  $A_{1g}$  (Tab. 1). The modes at 812, 782 and 659 cm<sup>-1</sup> were ascribed previously to stretching vibrations by Bahfenne (2011) and Bahfenne et al. (2011). The mode at 768 cm<sup>-1</sup> has never been recorded in trippkeite previously.

The medium and low intensity bands and shoulders at 541 ( $A_{1g}$ ), 498 ( $A_{2g}$ ), 461 ( $B_{1g}$ ), 448 ( $B_{2g}$ ) and 423(?) cm<sup>-1</sup> in trippkeite were assigned to the symmetric v<sub>2</sub> ( $A_1$ ) bending modes, whereas the very intense band and the strong band at 372 ( $A_2g$ ) and 277 ( $B_2g$ ) cm<sup>-1</sup>, respectively, as well as the very weak shoulders at 398 ( $A_{1g}$ ) and 359 ( $B_{1g}$ ) cm<sup>-1</sup>, ascribed to the antisymmetric v<sub>4</sub> (E) bending modes (Figs 2–3; Tab. 1).

Although the bands at 541, 498, 423 and 372 cm<sup>-1</sup> were recorded by Bahfenne (2011) and Bahfenne et al. (2011), only the latter two had been assigned to bending vibrations. The bands at 541 and 498 cm<sup>-1</sup> were considered as Cu–O stretching vibrations by Bahfenne

et al. (2011) or as As–O stretching vibrations by Bahfenne (2011).

In trippkeite, absorption bands which are not related to stretching or bending modes of  $AsO_3$  groups occur below ~ 250 cm<sup>-1</sup> (Tab. 1). The other bending and all lattice modes have never been reported in trippkeite previously, either.

The symmetric  $v_1(A_1)$  Sb–O stretching modes in schafarzikite consist of the very strong dominant Raman band at 668  $(A_{2\alpha})$  cm<sup>-1</sup> and the very weak shoulder at 707  $(A_{1\alpha})$  $cm^{-1}$  (Figs 2, 4; Tab. 1). The antisymmetric v<sub>2</sub> (E) Sb–O stretching vibrations occur as weak to very weak bands at 555  $(A_{1g})$  and 526  $(A_{2g})$  cm<sup>-1</sup> (Figs 2, 4; Tab. 1). The weak to very weak band triplet at 464  $(A_{1g})$ , 405  $(A_{2g})$  and 353  $(B_{10})$  cm<sup>-1</sup> with the weak satellite shoulder at 347  $(B_{20})$  $cm^{-1}$  were considered as the symmetric v<sub>2</sub> (A<sub>1</sub>) bending modes (Figs 2, 4; Tab. 1). The strong and medium bands at 295  $(A_{1g})$  and 218  $(E_g)$  as well as the weak shoulders at 275  $(A_{2g})$ , 253  $(B_{1g})$  and 229  $(B_{2g})$  cm<sup>-1</sup> were assigned to the antisymmetric  $v_4(E)$  bending modes. The bands occurring below  $\sim 200\ cm^{-1}$  are ascribed to lattice vibrational modes (Figs 2, 4; Tab. 1). The bands at 275, 229 cm<sup>-1</sup> and some lattice mode have never been recorded in schafarzikite spectrum previously. The other modes had

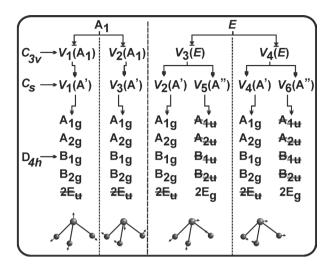


Fig. 5 Splitting pattern of the  $A_1$  and E modes. (Modes with strike-through are either IR active or inactive).

been reported by Sejkora et al. (2007), Bahfenne (2011) and Bahfenne et al. (2011).

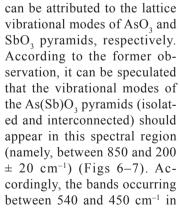
Comparison of the band positions and numbers of the present study with those given by Sejkora et al. (2007), Bahfenne (2011) and Bahfenne et al. (2011) shows a great similarity in the band positions and an increase in the band numbers, especially in the spectrum of trippkeite (Tab. 1). The increase in the bands numbers in trippkeite spectra may be attributed to the oriented measurements. Bahfenne et al. (2011) expected that other bands should be occurring in the spectra of trippkeite and schafarzikite.

In the present work, the assignment of some bands (Tab. 1) differs from that given by Bahfenne et al. (2011). Thus the bands at 541 and 498 cm<sup>-1</sup> in trippkeite and at 464 and 405 cm<sup>-1</sup> in schafarzikite were assigned

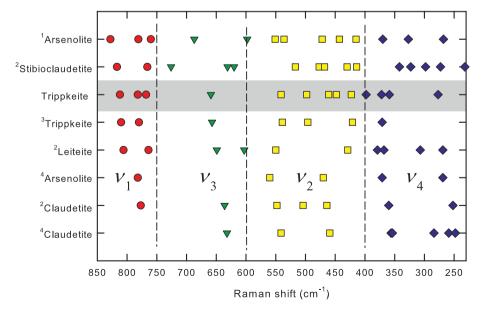
to Cu(Fe)–O stretch and the bands at 238 and 218 cm<sup>-1</sup> in schafarzikite to Fe–O deformation by Bahfenne et al. (2011). However, there was presented no evidence to support such a symmetry assignment. Furthermore, Bahfenne (2011) ascribed the same bands (i.e. 541 and 498 cm<sup>-1</sup> in trippkeite and 464 cm<sup>-1</sup> in schafarzikite) to As(Sb)–O stretching vibrations. Moreover, comparing the spectra of trippkeite with those of arsenolite and claudetite, unsurprisingly, the spectra of latter minerals show the same bands at 536 cm<sup>-1</sup> (arsenolite, Gilliam et al. 2003), 541 cm<sup>-1</sup> (claudetite, Flynn et al. 1976) and 504 cm<sup>-1</sup> (claudetite, Origlieri et al. 2009).

In addition, Loehr and Plane (1968) and Flynn et al. (1976) have collected the spectra of isolated  $C_{3\nu}$ pyramidal AsO<sub>3</sub> group and assigned the bands between ~ 750 and 300 cm<sup>-1</sup> to the vibrational modes of the isolated AsO<sub>3</sub> pyramids. The bands between ~ 850 and 250 cm<sup>-1</sup> were assigned also to the vibrational modes of the interconnected (C<sub>s</sub>) pyramidal AsO<sub>3</sub> group in claudetite (As<sub>2</sub>O<sub>3</sub>) (Flynn et al. 1976; Origlieri et al. 2009), arsenolite (As<sub>2</sub>O<sub>3</sub>) (Beattie et al. 1970; Brumbach and Rosenblatt 1972; Lucovsky and Galeener 1980; Grzechnik 1999; Gilliam et al. 2003; Jensen et al. 2003), leiteite (ZnAs<sub>2</sub>O<sub>4</sub>) and stibioclaudetite (AsSbO<sub>3</sub>) (Origlieri et al. 2009), oxo-arsenates (Emmerling and Röhr 2003) and oxo-antimonates (Hirschle and Röhr 2000).

Comparison of band positions of the isolated  $C_{3\nu}$ and interconnected  $C_s$  pyramidal XO<sub>3</sub> group (Figs 6–7) shows that the stretching and bending vibrations of the isolated and interconnected AsO<sub>3</sub> and interconnected SbO<sub>3</sub> pyramids occur in the spectral region between 850–250 and 750–200 ± 20 cm<sup>-1</sup>, respectively, which is considered the characteristic spectral region of oxides (according to Wang et al. 1994). The vibrational modes that appear below 250 ± 20 and below 200 ± 20 cm<sup>-1</sup>



**Fig. 6** Positions of the  $v_1$ ,  $v_3$ ,  $v_2$  and  $v_4$  vibrational modes in trippkeite together with minerals containing AsO<sub>3</sub> pyramidal group data after 'Gilliam et al. (2003), <sup>2</sup>Origlieri et al. (2009), <sup>3</sup>Bahfenne et al. (2011), <sup>4</sup>Flynn et al. (1976).



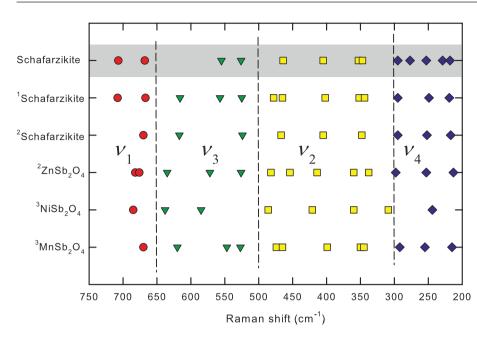


Fig. 7 Band positions in schafarzikite compared with the  $v_1$ ,  $v_3$ ,  $v_2$  and  $v_4$  vibrations in minerals containing SbO<sub>3</sub> pyramidal group, data after <sup>1</sup>Bahfenne et al. (2011), <sup>2</sup>Sejkora et al (2007), <sup>3</sup>Chater et al. (1986).

trippkeite and between 480 and 400 cm<sup>-1</sup> in schafarzikite can be attributed to the vibrational modes of the As(Sb)O<sub>2</sub> rather than the frequencies arising from the stretching of the Cu(Fe)-O as suggested by Bahfenne et al. (2011). The same observations have been noticed by Kharbish et al. (2007, 2009) and Kharbish (2011) in minerals containing isolated and interconnected As(Sb) S, pyramidal groups. Furthermore, it can be concluded from Figs 6 and 7 that the symmetric  $v_1$  (A<sub>1</sub>) As(Sb)–O stretching modes occur between 850-750 and between 750-650 cm<sup>-1</sup> for trippkeite and schafarzikite, respectively. The antisymmetric As(Sb)-O stretching vibrations appear in the region between 750-600 and between 650–500 cm<sup>-1</sup>, respectively (Figs 6–7). The symmetric  $v_{2}$  (A<sub>1</sub>) O-As(Sb)-O bending modes occur in the spectral regions 600–400 and 500–300 cm<sup>-1</sup>, respectively, and those for the antisymmetric  $v_{A}(E)$  modes between 400-250 and 300-200 cm<sup>-1</sup> (Figs 6-7).

The above discussion reveals that no striking differences in the spectral features are evident between minerals containing isolated and interconnected  $(As,Sb)O_3$ groups. The latter observation has been also noticed for isolated and interconnected  $(As,Sb)S_3$  groups by Kharbish et al. (2009) and Kharbish (2011).

Acknowledgements. The author is grateful to Prof. A. Beran and Prof. E. Libowitzky, for their help during the Raman measurements. Thanks are due to Prof. F. Pertlik for kindly providing samples and Prof. H. Effenberger for carrying out unit-cell parameter measurements. In addition to useful scientific comments, handling editor Dr. J. Sejkora as well as J. Plášil and an anonymous reviewer helped to improve the quality of the English text.

#### References

- BEATTIE IR, LIVINGSTON KMS, OZIN GA, REYNOLDS DJ (1970) Single-crystal Raman spectra of arsenolite  $(As_4O_6)$  and senarmonite  $(Sb_4O_6)$ . The gas-phase Raman spectra of  $P_4O_6$ ,  $P_4O_{10}$ , and  $As_4O_6$ . J Chem Soc A: 449–451
- BAHFENNE S (2011) Single Crystal Raman Spectroscopy of Selected Arsenite, Antimonite and Hydroxyantimonite Minerals. Unpublished Ph.D. thesis, Queensland University of Technology, pp 1–159
- BAHFENNE S, RINTOUL L, FROST, RL (2011) Single crystal Raman spectroscopy of natural schafarzikite FeSb<sub>2</sub>O<sub>4</sub> from Pernek, Slovak Republic. Am Mineral 96: 888–894
- BRUMBACH SB, ROSENBLATT GM (1972) In-cavity laser Raman spectroscopy of vapors at elevated temperatures. As<sub>4</sub> and As<sub>4</sub>O<sub>6</sub>. J Chem Phys 56: 3110-3118
- CHATER R, GAVARRI JR, GENET F (1986) Composes isomorphes  $MeX_2O_4E_2$ : I. Etude vibrationnelle de  $MnSb_2O_4$ entre 4 et 300 K: champ de force et tenseur élastique. J Solid State Chem 63: 295–307
- COTTON FA, WILKINSON G (1966) Advanced Inorganic Chemistry (6<sup>th</sup> edition). John Wiley & Sons, New Jersey, pp 1–1376
- EBY RK, HAWTHORNE FC (1993) Structural relations in copper oxysalt minerals. I. Structural hierarchy. Acta Cryst B49: 28–56
- EMMERLING F, RÖHR C (2003) Die neuen Oxoarsenate(III) AAsO<sub>2</sub> (A = Na, K, Rb) und  $Cs_3As_5O_9$ . Darstellung, Kristallstrukturen und Raman-Spektren. Z Naturforsch 58: 620–626
- FISCHER R, PERTLIK F (1975) Verfeinerung der Kristallstruktur des Schafarzikits, FeSb<sub>2</sub>O<sub>4</sub>. Tschermaks Min Petr Mitt 22: 236–241

- FLYNN EJ, SOLIN SA, PAPATHEODOROU GN (1976) Vibrational excitations of As<sub>2</sub>O<sub>3</sub>. II. Crystalline phases. Phys Rev B13: 1752–1758
- GHOSE S, SEN GUPTA PK, SCHLEMPER EO (1987) Leiteite, ZnAs<sub>2</sub>O<sub>4</sub>: a novel type of tetrahedral layer structure with arsenite chains. Am Mineral 72: 629–632
- GILLIAM SJ, MERROW CN, KIRKBY SJ, JENSEN JO, ZEROKA D, BANERJEEB A (2003) Raman spectroscopy of arsenolite: crystalline cubic As<sub>4</sub>O<sub>6</sub>. J Solid State Chem 173: 54–58
- GRZECHNIK A (1999) Compressibility and vibrational modes in solid As<sub>4</sub>O<sub>6</sub>. J Solid State Chem 144: 416–422
- HINRICHSEN B, DINNEBIER RE, RAJIV P, HANFLAND M, GRZECHNIK A, JANSEN M (2006) Advances in data reduction of high-pressure X-ray powder diffraction data from two-dimensional detectors: a case study of schafarzikite ( $FeSb_2O_4$ ). J Phys Condens Mat 18/25: S1021–S1037
- HIRSCHLE C, RÖHR C (2000) Alkalimetall-Oxoantimonate: Synthesen, Kristallstrukturen und Schwingungsspektren von ASbO<sub>2</sub> (A = K, Rb), A<sub>4</sub>Sb<sub>2</sub>O<sub>5</sub> (A = K, Rb, Cs) und Cs<sub>3</sub>SbO<sub>4</sub>. Z Anorg Allg Chem 626: 1305–1312
- JENSEN JO, GILLIAM SJ, BANERJEEB A, ZEROKA D, KIRKBY SJ, MERROW, CN (2003) A theoretical study of  $As_4O_6$ : vibrational analysis, infrared and Raman spectra. J Mol Struct (Theochem) 664–665: 145–156
- KHARBISH S (2011) Raman spectroscopic investigations of some Tl-sulfosalt minerals containing pyramidal (As,Sb) S<sub>3</sub> groups. Am Mineral 96: 609–616
- KHARBISH S, LIBOWITZKY E, BERAN A (2007) The effect of As–Sb substitution in the Raman spectra of tetrahedrite– tennantite and pyrargyrite–proustite solid solutions. Eur J Mineral 19: 567–574
- KHARBISH S, LIBOWITZKY E, BERAN A (2009) Raman spectra of isolated and interconnected pyramidal XS<sub>3</sub> groups (X = Sb,Bi) in stibnite, bismuthinite, kermesite, stephanite and bournonite. Eur J Mineral 21: 325–333
- KRENNER JA (1921) Schafarzikite, ein neues Mineral. Z Kristallogr 56: 198–200

- $\begin{array}{l} \text{Lee C, Harrison WTA (2004) The catena-arsenite chain an-}\\ \text{ion, } [AsO_2]^{n_{-}}: (H_3NCH_2CH_2NH_3)_{0.5}[AsO_2] \text{ and } NaAsO_2.\\ \text{Acta Cryst C60: } m215-m218 \end{array}$
- LOEHR TM, PLANE RA (1968) Raman spectra and structures of arsenious acid and arsenites in aqueous solution. Inorg Chem 7: 1708–1714
- LUCOVSKY G, GALEENER FL (1980) Local structure and vibrational spectra of vAs<sub>2</sub>O<sub>3</sub>. J Non-Cryst Solids 37: 53–70
- MELLINI M, MERLINO S (1979) Versiliaite and apuanite: derivative structures related to schafarzikite. Am Mineral 64: 1235–1242
- NAKAMOTO K (1997) Infrared and Raman Spectra of Inorganic and Coordination Compounds. Part A: Theory and Applications in Inorganic Chemistry (6<sup>th</sup> edition). John Wiley & Sons, New Jersey, pp 1–419
- ORIGLIERI MJ, DOWNS RT, PINCH WW, ZITO, GL (2009) Stibioclaudetite AsSbO<sub>3</sub>. A new mineral from Tsumeb, Namibia. Mineral Rec 40: 209–213
- OTWINOWSKI Z, MINOR W (1997) Processing of X-ray diffraction data collected in oscillation mode. In: CARTER CW, SWEET RM (eds) Methods in Enzymology. Academic Press, New York, pp 307–326
- PERTLIK F (1975) Verfeinerung der Kristallstruktur von synthetischem Trippkeit, CuAs<sub>2</sub>O<sub>4</sub>. Tschermaks Min Petr Mitt 22: 221–217
- PERTLIK F (1988) The single chain arsenites Pb(AsO<sub>2</sub>)Cl and Pb<sub>2</sub>(AsO<sub>2</sub>)<sub>3</sub>Cl. Preparation and structure investigation. Z Kristallogr 184: 191–201
- SEJKORA J, OZDÍN D, VITÁLOŠ J, TUČEK P, ČEJKA J, DUĎA R (2007) Schafarzikite from the type locality Pernek (Malé Karpaty Mountains, Slovak Republic). Eur J Mineral 19: 419–427
- WANG A, HAN J, GUO L, YU J, ZENG P (1994) Database of standard Raman spectra of minerals and related inorganic crystals. Appl Spectrosc 48: 959–968
- ZEMANN J (1951a) Formel und Kristallstruktur des Schafarzikits. Tschermaks Min Petr Mitt 2: 166–175
- Zemann J (1951b) Formel und Kristallstruktur des Trippkeits. Tschermaks Min Petr Mitt 2: 417–423