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TITLE: DISCOVERY AND DEVELOPMENT OF THERAPEUTIC DRUGS AGAINST LETHAL HUMAN RNA VIRUSES: A MULTIDISCIPLINARY ASSAULT

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19. ABSTRACT

A total of 5,799 samples were submitted for prescreen RNA-type antiviral evaluation over the grant period to USAMRIID. After confirmation of activity, a good number of high priority extracts of plant and animal origin (and synthetic compounds) were identified for further research (fractionation, isolation and characterization of new antiviral compounds). The continued fractionation of these leads is in progress.

In addition to the natural products research, further development of the scale-up isolation of pancratistatin, an active lead against Japanese Encephalitis from both plant sources by greenhouse cultivation and semi-synthetic transformation of another plant product, narciclasine, occurred during the grant period. For the semi-synthetic research, seven tons of *Narcissus incomparabilis* has been obtained and is at the initial stage of scale-up isolation.

In short, progress continues to be excellent and we have a promising number of new antiviral leads to pursue.



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FOREWORD

Opinions, interpretations, conclusions and recommendations are those of the author and are not necessarily endorsed by the US Army.

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 $\frac{1/4}{1000}$ In conducting research using animals, the investigator(s) adhered to the "Guide for the Care and Use of Laboratory Animals," prepared by the committee on Care and Use of Laboratory Animals of the Institute of Laboratory Resources, National Research Council (NIH Publication No. 86-23, Revised 1985).

N/A For the protection of human subjects, the investigator(s) adhered to policies of applicable Federal Law 45 CFR 46.

<u>A/A</u> In conducting research utilizing recombinant DNA technology, the investigator(s) adhered to current guidelines promulgated by the National Institutes of Health.

PI - Signature Date

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I. Introduction

A long-term USAMRIID research program directed at the isolation and structural elucidation of new and potentially useful antiviral drugs from marine animals and plants has been substantially advanced by the two years of USAMRIID financial assistance. The financial support provided by the USAMRIID program was used to isolate and characterize new antiviral chemotherapeutic drugs from confirmed active extracts of marine invertebrates and vertebrates as well as marine and terrestrial plants including fungi, algae and other microorganisms. The research was sharply directed at marine animal and plant species yielding extracts with an outstanding level of antiviral activity in the USAMRIID's programs (RNA viruses).

II. Current Advances

A. Fractionation of Active Leads from Natural Products

Over the grant period, a total of 5,799 samples were submitted for antiviral evaluation (see Appendix A). Of this total 1,612 were from crude plant extracts and 3,826 were from crude marine extracts with another 288 from microorganism mycellium extracts as listed in Appendix A, Table I. Prescreen activity was determined in 1,420 of the total number and are listed in Tables II and III. Full screen submissions of fractions from 26 plant and 21 marine species are listed in Table IV. Table V lists samples submitted either in response to a specific request or as possible actives from other sources.

The ten highest priority antiviral actives determined by November, 1990 and beyond are as follows:

Marine Animal Sources

AVS-709 Styela plicata B 705028 (D048)

AVS-7438 Unknown sponge (Papua New Guinea) B 723123 AVS-8374 & 9217 Unknown sponge (Antarctic) B 722902

Plant Sources

AVS-6976-6979 Cryptocarya multipaniculata B 611679 (FOO9-FO12) AVS-6986 & 6988 Virola oleifera B 619315 (FO12 & FO14) AVS-7032 Eucalyptus spathulata B 827298 (FO08) AVS-7067-7068 Ruprechtia tangarana B 836749 (FOO5-FOO6) AVS-7083-7087 Phyllanthus anisolobus B 848528 (FO11-FO15) AVS-7092 Notelaea ligustrina B 853791 (FO17) AVS-8259 Unknown Phaeophyta (Papua New Guinea) B 848990

The accompanying data sheets and flow sheets for fractionations of active extracts follow, and provide a status report at end of this report period.

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AVS-000709	B705028 D046	NHV 8	N BO	2/20/91	0.32	12.6	22.6	0	^	100	^	100	^	100	٨	4.43	۸	4.43	۸	27.98
AVS-000709	B705028 D046	3 HIV	MT2	2/20/91	0.32	2.77	13.7	0	۸	100	۸	100	۸	100	~	7.295		7 29		33.98
AVS-000709	B705028 D046	3 HIV	NBO	7/5/90	0.32	6.14	20.4	0	^	100	^	100		100		1.902		4		22.97
AVS-008433	B 705028-D04	0 HIV	NB 0	1/4/91	3.2	107	0	0		969		1000	^	1000		0			•	13.64
AVS-008433	B 705028-D04	0 HIV	NB0	12/11/90	0.32	2.26	14.7	0	۸	100	^	100	^	100	~	5.786	^	6.79		30.99
AVS-008433	B 705028-D04	0 HIV	MT2	12/11/90	0.32	4.19	59.6	0	^	100	۸	100	۸	100	^	1.677		1.68		24 19
AVS-008434	B 705028-F05	VIH 0	RBO	1/4/91	3.2	26.4	545	0	۸	1000	۸	1000		1000	~	1.833		1.83		19.9
AVS-008434	B 705028-F05		Ð	12/11/90	0.32	22.3	66.4	0	۸	100	^	100	۸	100	. ^	1.507		1.51	•	7.38
AVS-008434	B 705028-F05	VIH 0	MT2	12/11/90	0.32	-	2.11	0	۸	100	^	100		100	۷ 4	7.445		47.44		54.02
AVS-008435	B 705028-F05	VIH L	B	12/11/90	0.32	0	0	0		0.145		0.291	^	100		0			•	0
AVS-008435	B 705028-F05	NH L	MT2	12/11/90	0.32	0	0	0	۸	100	^	100	٨	100		0			٨	0.83
AVS-008436	B 705028-F05	2 HIV	NB0	12/11/90	0.32	8.5	0	0		2.74	۸	100		100		0				5.48
AVS-008436	B 705028-F05	2 HIV	MT2	12/11/90	0.32	1.38	14.1	0	۸	100	۸	100	۸	100	~	7.087	۸	7.09		38.71
AVS-008437	B 705028-F05	NH C	8	12/11/90	0.32	16.6	0	0		0.178	۸	100	^	100		0			۸	11.66
AVS-008437	B 705028-F05	NH E	MT2	12/11/90	0.32	1.66	37.7	•	۸	100	۸	100	^	100	~	2.656	۸	2.66	۸	34.06



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9443	B611679 F0	217	۴	NERO	4/24/91	0	0	0		70.9		145		303		0				0
9443	B611679 F0	717	አ	VERO	4/23/91	0	0	0		8.69		21.6		89.4		0				0
9443	B611679 F0	517	щ		4/24/91	0	0	0		62.2		100		298		0				0
9443	B611679 F0	317	ЪТ		4/23/91	49.8	84.5	0		88.7		163		304		1.928		1.05		4.38
9443	B611679 F0	017	Ĕ	VERO	4/26/91	0	ల	0		119		187		309		0			^	-
9444	B611679 F0	018	۴	VERO	4/24/91	0	0	0		45.2		63.4		96.3		0				0
9444	B611679 F0	318	አ	OHEN	4/23/91	2.71	0	0		13.7		28.9		92.4		0			۸	10.53
9444	B611679 F0	318	ų	VERO	4/24/91	0	0	0		44		62.6	-	96.3		0			^	0.03
9444	B611679 F0	318	РТ		4/23/91	17.2	0	0		49		66	-	96.6		0				7.25
9444	B611679 F0	018	Ë	VEHO	4/26/91	70.4	0	0		59		86		290		0			^	1.14
9445	B611679 F0	019	۶		4/24/91	11.7	30.8	0		30.4		54.7		66		1.776		0.99		7.88
9445	B611679 F0	019		N	4/23/91	-	0	0		2.61		8.12		29.4		0			^	3.64
9445	B611679 F0	019	щ		4/24/91	0	0	0		18.6		28.7		265		0				0
9445	B611679 F0	019	ΡŢ	OH H	4/23/91	4.15	6.38	0		17.5		25.1		89.5		3.927		2.75		14.12
9445	B611679 F0	019	Ë	NBN	4/26/91	57.5	0	0		22.5		68.6		309		0			^	0.3
9446	B611679 F0	320	щ	VERO	4/24/91	46.1	66.4	0		137		199		310		2.994		2.06		8.88

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TAI	7.75 0 11.73 0.15
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S	1.59 2.13
T150	2.386 0 3.269 0
TC95	310 94.6 309 320
0	A
TC5(193 190
TC25	128 7.65 124 151
1095	0000
1050	81 0 58.2 0
IC25	47.8 0 41.1
DATE	4/24/91 4/23/91 4/23/91 4/26/91
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VIRUS	ょ≈⋷戦
CTR#	B611679 F020 B611679 F020 B611679 F020 B611679 F020 B611679 F020
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# SA	CTR #	VIRUS	ମ୍ମ	DATE	INIT. CONC.	IC25	1050	1C95		TC25		rc50		TC95		T150	s		TAI
6984	B619315-F010	ЧV	MT2	8/8/90	0.32	1.47	0	0	۸	100	۸	100	۸	100		0		۸	30.31
6984	B619315-F010	щ		6/21/90	-	0	0	0		144		203		308		0			0
6984	B619315-F010	ΡŢ	VERO	6/21/90	-	0	0	0		139		199		308		0		۸	1.1
6984	B619315-F010	አ		6/21/90	-	0	0	0		130		194		307		0			0.55
6984	B619315-F010	E E E E E		6/22/90	-	0	0	0		140		265	۸	320		0			0
6984	B619315-F010	₽		6/21/90	-	0	0	0		158		216		320		0			4.45
6985	B619315-F011	NH	MT2	8/16/90	0.32	0	0	0		49.3		66.7		97.9		•			1.06
06985	B619315-F011	Ч		6/21/90	-	0	0	0		40.1		60.3		96.8		0			0
5869 0	B619315-F011	Ы		6/21/90	-	0	0	0		28.9		52.8		95.3		0		٨	0.04
06985	B619315-F011	ዮ	VERV	6/21/90	-	0	0	0		38.6		59		95.9		0			0
06985	B619315-F011	VEE	VERO	6/22/90	-	0	0	0		8.06		70.9		286		0			0
06985	B619315-F011	>		7/19/90	-	0	0	0		13.5		27.9		247		0			0
06985	B619315-F011	₽		6/21/90	-	0	0	0		49		6 6		96.6		0		۸	0.1
06986	B619315-F012	NH N	MT2	8/16/90	0.32	0	0	0		70.6		100		100		0			2.01
06986	B619315-F012	щ	OHEN	8/9/90	3.2	134	179	302		492		663		973		3.709	2.75	٨	16,01
06986	B619315-F012	円		6/21/90	-	235	0	0		220	۸	320	۸	320		0			0
06986	B619315-F012	μ		6/21/90	-	21.3	0	0		135		224	۸	320		0		۸	14.36
06986	B619315-F012	Ŗ		8/7/90	3.2	15.2	28.3	88.1		452		634		963		22.437	15.97		41.73
06986	B619315-F012	ሦ		6/21/90	·	14.6	30.2	88.8		157		216	۸	320		7.151	5.19		25.88
06986	B619315-F012	Ë		8/10/90	3.2	129	227	0		283		520		970		2.288	1.25		8.26
06986	B619315-F012	¥		6/22/90	-	87.7	208	0		222	۸	320	۸	320	۸	1.54	1.07	۸	6.48
98690	B619315-F012	>		06/6/8	-	34.2	59.8	0		86.8		162		310		2.715	1.45		9.84
98690	B619315-F012	>		7/19/90	-	32	46.8	92.7		182		264	۸	320		5.647	3.89		21.69
06986	B619315-F012	₽		8/8/90	3.2	228	0	0		193		286		917		0			0.33
98690	B619315-F012	₽		6/21/90	-	45.1	63.6	0		174		255	۸	320		4.016	2.74		12.38
06987	B619315-F013	≥ 1	MT2	8/16/90	0.32	0	0	0		100		100		100		0			1.2
06987	B619315-F013	щ		6/21/90	-	0	0	0	۸	320	۸	320	۸	320		0			0
06987	B619315-F013	Ы		6/21/90	-	0	0	0	۸	320	۸	320	۸	320		0		۸	5.92
06987	B619315-F013	ዩ		6/21/90	÷	0	0	0	۸	320	۸	320	۸	320		0			3.73
06987	B619315-F013	斑		6/22/90	-	0	0	0		87.6	۸	320	۸	320		0			0
06987	B619315-F013	>>		7/19/90	-	184	0	0		200		300	۸	320		0			0.64
06987	B619315-F013	۶		6/21/90	-	0	0	0	۸	320	۸	320	۸	320		0			0
98690	B619315-F014	Ν	MT2	8/16/90	0.32	0	0	0		100		100		100		0			2.39
06988	B619315-F014	щ		8/9/90	3.2	0	0	0		527		735	۸	1000		0		۸	4.75
06988	B619315-F014	щ		6/21/90	-	0	0	0		142		259	۸	320		0			0
98690	B619315-F014	Ы		6/21/90	-	44.1	60.7	0		290	۸	320	۸	320	^	5.272	4.77	۸	23.2
96988	B619315-F014	Ŗ		8/7/90	3.2	140	220	0		497		674		663		3.068	2.26		12.59
969690	B619315-F014	አ		6/21/90	-	47.8	71.4	0		277	٨	320	۸	320	^	4.482	3.89	٨	17.8

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AVS-006988	B619315-F01	4 VEE	VERO	8/10/90	3.2	0	0	0		544	76	~ ~	10	00		0			2.3	9
AVS-006988	B619315-F01	4 1		6/22/90	*-	0	0	0	-	88	27	~ ~	e	20		0			0	
AVS-006988	B619315-F01	4 \	OHEN	7/19/90	***	0	0	0	-	95	29	^ 0	ð	20		0			3.3	S
AVS-006988	B619315-F01	₩	OHEN	8/8/90	3.2	166	0	٥		513	70	ہ 9	ž	00		0			8.7	60
AVS-006988	B619315-F01	4	VEHO	6/21/90	-	100	0	0	•	320	32	^ 0	e	20		0			-	9
AVS-009447	B619315 F01	5 E		4/24/91	-	0	0	0	G	0.3	27	^ +	e S	20		0			0	I
AVS-009447	B619315 F01	5		4/24/91	-	294	0	0	-	54	27	^ 9	m	20		0			0.3	4
AVS-009447	B619315 F01	5 Sr	VERO	4/23/91	-	53.1	100	0		66	15	2	e	10	,	.524	o.	66	-	6
AVS-009447	B619315 F01	5 PT		4/23/91	-	143	0	0	•	86	27	^ ~	е С	20		0			2.9	8
AVS-009447	B619315 F01	5 VEE		4/26/91	-	208	0	0	9	1.3	06	^ بو	e	20		0			0.2	ŝ
AVS-009448	B619315 F01	¥ 9	VENO	4/24/91	*	143	224	0		256	32	^ 0	ю	20	~	.43	-	4	5.6	5
AVS-009448	B619315 F01	6 E		4/24/91	-	148	219	0	8	18.7	• 32	Ô	ю	20	- -	458	0	4	2.3	6
AVS-009448	B619315 F01	ور م		4/23/91	-	20.4	0	0		40	ø	0	Ų,	96		0			4.0	6
AVS-009448	B619315 F01	6 PT		4/23/91	-	63	136	o	•	320	32	^ 0	ю	20	~ ^	356	~ ^	36	14.	5
AVS-009448	B619315 F01	6 VEE		4/26/91	-	156	265	0	•	320	33	ô	с С	20	~	.21	 - 	2	ë	5
AVS-009449	B619315 F01	۲ ۲		4/24/91	-	0	0	0	42)	1.2	~	5	N	81		0			0	
AVS-009449	B619315 F01	7 JL 7		4/24/91	-	44.9	63.1	0	8	19.8	16	6	e	Ŧ	N	.683	-	42	6.6	6
AVS-009449	B619315 F01	7 Sf		4/23/91	-	0	0	0	æ	.54	16	7	ĕ	0.5		0			0	
AVS-009449	B619315 F01	7 PT		4/23/91	-	0	0	0	4	9.9	67	80	-	00		0			1.3	n E
AVS-009449	B619315 F01	7 VEE	VENO	4/26/91	-	39.8	57.4	0	•	160	22	^ 0	с	20	с,	.825	તાં	78 、	16.	5
AVS-009450	B619315 F01	8		4/24/91	-	0	0	0		48	~	~	2	72		0			0	
AVS-009450	B619315 F01	8 5		4/24/91	-	0	0	0	6.5	1.3	76	2	2	79		0			0	
AVS-009450	B619315 F01	8 2		4/23/91	-	0	0	0	e)	.86	Ś	6	o,	59		0			0	
AVS-009450	B619315 F01	8 PT		4/23/91	-	0	0	0	LC)	5.9	79	6	2	82		0		Ŷ	4	ŝ
AVS-009450	B619315 F01	8 2 2 3		4/26/91	-	39.1	79.6	0	~	0.6	12	4	e	00	÷-	.552	Ö	68	6.1	S
AVS-009451	B619315 F01	9 ⊬		4/24/91	-	0	0	0	4	1.5	99	æ,	-	73		0			0.7	6
AVS-009451	B619315 F01	6 ۳		4/24/91	-	0	0	0	•	4.8	Ö	6	2	22		0			0	
AVS-009451	B619315 F01	е Гу		4/23/91	-	0	0	0	-	4.9	ġ.	9	o,	66		0		Ŷ		9
AVS-009451	B619315 F01	1d 6	VEHO	4/23/91	-	13.4	17.9	30.2	ι, L	0.7	69	4	-	8 6	e,	.877	N	83	15.	36
AVS-009451	B619315 F01	9 VEE		4/26/91	-	22.3	0	0		53	ř	-	2	71		0		Ŷ	6.2	N
AVS-009452	B619315 F02	₩		4/24/91	•	0	0	0	ч	8.5	74	Ņ	2	83		0			0.0	5
AVS-009452	B619315 F02	ы П О		4/24/91	-	0	0	0	4	2.9	73	ຄ	2	87		0			0	
AVS-009452	B619315 F02	о М		4/23/91		0	٥	0	(7)	.63	5.7	ŝ	ດ້	58		0			0.0	4
AVS-009452	B619315 F02	0 PT	OHEN	4/23/91	-	14.9	23	0	LO .	7.4	82	2	2	6 3	ຕ່	596	Ň	49	11.	2
AVS-009452	B619315 F02	٥ ۲		4/26/91	-	27.7	90.9	0	œ	6.7	2	ŝ	ო	15	-	.151	Ö	73 >	6.9	6



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T150	0	0	0	0	0	0	0	0	0	0	0	0	0	1.303	4.456	3.929	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	٥	7.138	31.334	00 000
TC95	320	926	913	320	876	906	320	691	800	320	938	924	320	320	869	918	320	320	320	320	320	320	320	320	911	308	320	800	309	309	251	320	320	821	302	320	320	000
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1C50	100	276	97.8	149	115	96.2	173	87.3	98.4	249	382	283	220	130	81.8	94	320	320	320	320	251	320	154	320	259	196	279	218	180	138	58	320	320	206	137	155	192	010
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1C25	26.5	72.2	43	56.7	60.9	64.1	52	34.1	57.9	69.69	168	100	123	10	40.3	44	152	320	320	320	175	61.1	27.6	157	131	135	151	140	109	15.5	29	73.8	245	115	63.6	5.75	53.6	03 E
1085	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	•	0	0	0	0	0	0	0	0	0	0	a c
1050	0	0	0	0	0	0	0	0	0	0	0	0	0	100	18.3	23.9	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	21.7	6.12	010
IC25	0	0	0	0	0	0	241	0	0	0	0	0	0	41.1	9.45	13.7	0	215	0	0	0	٥	0	306	0	0	189	73.3	66.4	0	0	192	0	0	0	11.8	3.5	1 00
NIT. CONC.	-	3.2	3.2	-	3.2	3.2	-	3.2	3.2	-	3.2	3.2	-	-	3.2	3.2	-	-	-	-	-	-	-	-	3.2	•	-	3.2	-	-	3.2	-	-	3.2	-	-	-	•
DATE	8/31/90	10/25/90	10/3/90	8/30/90	10/25/90	10/2/90	8/30/90	10/25/90	10/2/90	8/28/90	10/26/90	10/5/90	10/4/90	8/31/90	10/25/90	10/3/90	8/31/90	8/30/90	8/30/90	8/28/90	10/4/90	8/31/90	8/31/90	10/25/90	10/3/90	8/30/90	10/25/90	10/2/90	8/30/90	10/25/90	10/2/90	8/28/90	10/26/90	10/5/90	10/4/90	8/31/90	10/25/90	1010100
CEL	VEHO	VEHO	VERO	VERO	VEHO	VEHO	VENO	VERO	VERO	VERO	VERO	VENO	VERO	VERO		VERO	VERO	OHEN	VEHO	VEHO	VERO	VERO	KERO	VERO	VEHO	KENO	VERO	VERO	VEHO	VERO	VERO	VERO	VERO	VERO	VERO	VERO	VERO	
VIRUS	щ	щ	ĥ	μ	ΡŢ	ЪТ	አ	ጽ	አ	ij	¥	¥	>	₽	⊭	⊭	щ	Ы	አ	Ĕ	>	₽	빙	щ	щ	Ы	ΡŢ	РТ	አ	አ	አ	ÿ	¥	ÿ	>>	₽	۴	5
CTR #	B836749-F003	B636749-F003	B836749-F003	B836749-F003	B836749-F003	B836749-F004	B8367 19-F004	BB36749-F004	B836749-F004	B836749-F004	B836749-F004	B836749-F005	B836749-F005	B836749-F005	B836749-F005	B836749-F005	B836749-F005	B836749-F005	B836749-F005	Desetto Cone																		
AVS #	AVS-007065	AVS-007066	AVS-007066	AVS-007066	AVS-007066	AVS-007066	AVS-007066	AVS-007067	AVS-007067	AVS-007067	AVS-007067	AVS-007067	AVS-007067	AVS-007067	AVC.007067																							

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IS							2.92							2.05	5.62	3.83																	0.94		-			
T150	0	0	0	0	0	0	4.006	0	0	Q	0	0	0	3.339	10.288	6.564	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	2.009	0	-	0	0	0
TC95	310	320	302	309	320	302	309	300	95.8	320	320	302	304	306	320	293	288	247	295	320	282	279	320	320	296	320	320	320	100	953	908	100	959	923	100	932	845	100
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TC50	165	261	140	210	194	140	208	124	48.7	205	320	143	161	181	189	87	77.8	99	92.6	79	78.4	62.8	100	121	94.7	320	320	100	100	534	259	100	495	296	100	320	222	100
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TC25	69.1	170	63.4	155	123	75.6	152	38.5	25.6	120	222	76.7	86.8	111	103	50.8	43.1	41.7	61.7	38.5	52.6	8.49	22.4	10	61.3	32	84.3	3.2	100	296	142	100	232	198	100	143	135	29.6
IC95	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	31	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
1050	0	0	0	0	0	0	51.9	0	0	0	0	0	0	54.1	18.4	13.2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	246	0	100	0	0	0
IC25	0	0	0	57.7	0	66.1	32	0	0	0	0	0	0	29.3	11.4	6.45	0	0	0	283	0	0	0	0	0	0	0	0	0	0	0	0	136	0	55.2	320	0	0
NIT. CONC.	-	-	-	-	-	-	-	**	+	-	-	-	-	-	-	-	~	-	-	-	-	-	-	•	-	-	-	-	0.3	3.2	3.2	0.3	3.2	3.2	0.3	3.2	3.2	0.3
DATE	8/31/90	10/25/90	10/3/90	8/30/90	10/25/90	10/2/90	8/30/90	10/25/90	10/2/90	8/28/90	10/26/90	10/5/90	10/4/90	8/31/90	10/25/90	10/3/90	8/31/90	8/30/90	8/30/90	8/28/90	10/4/90	8/31/90	8/31/90	8/30/90	8/30/90	8/28/90	10/4/90	8/31/90	8/31/90	10/25/90	10/3/90	8/30/90	10/25/90	10/2/90	8/30/90	10/25/90	10/2/90	8/28/90
G	VERO	VERO	KERO	VEHO	VERO	VERO		VERO			VEHO	VERO	VERO	VERO	VERO		VERO	VERO	VENO	VERO		VERO	VEHO	VERO	VERO	VERO	VERO	VERO	VEHO	VERO	VERO	VEHO	VERO	VERO	VERO	VERO	VERO	VERO
VIRUS	щ	щ	Ч	μ	μ	ΡŢ	አ	አ	አ	Ē	Ë	Æ	>	¥	⊭	۴	щ	Ы	ሦ	¥	>	ĥ	щ	Ы	አ	Ř	>	₽	щ	щ	щ	Ы	ΡŢ	Γd	ሦ	ሦ	ሦ	ÿ
CTR #	B836749-F006	B836749-F007	B836749-F007	B836749-F007	B836749-F007	B836749-F007	B836749-F007	B836749-F009	B836749-F009	B836749-F009	B836749-F009	B836749-F009	B836749-F009	B836749-F010																								
# SVA	AVS-007068	AVS-007069	AVS-007069	AVS-007069	AVS-007069	AVS-007069	AVS-007069	AVS-007070	AVS-007070	AVS-007070	AVS-007070	AVS-007070	AVS-007070	AVS-007071																								

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TAI	0.24	0			× 10.07	AE.1	× 13.51	0.05	> 1.36	0	1.27	0	0	•	0	0 0	000	× 26.00	× 26.42	× × 26.42	26.42 0.01 0.42 0.42 0.42
S					1.01		2.5											5.45	5.45	5.45	5.45
T150	0	0	0	0	2.193	0	3.649	0	0	0	0	0	0	¢	>	00	000	0 0 12.287	0 0 12.287 0	0 0 12.287 0	0 12.287 0 0
TC95	967	757	296	100	959	869	320	320	94.7	320	320	320	320	274	, i	320	320 320	320 320 320	320 320 320 320	320 320 320 320 137	320 320 320 320 320 320
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TC50	599	202	88.7	100	495	216	211	274	46.9	320	229	94.9	135	64.6		320	320 137	320 137 196	320 137 196 299	320 137 196 299 51	320 137 196 299 51 51 320
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TC25	394	120	13.7	24.7	227	109	145	185	23	238	165	52.4	59.2	28.9			73.5	73.5	73.5 87 128	73.5 87 128 23.2	73.5 87 128 23.2 320
95	0	0	Ö	0	0	0	0	0	0	0	0	0	0	0	0						^ \ 0 \ 0 0 0 0
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IC25	0	0	0	0	51.7	56.6	39.2	0	0	0	0	0	0	0	208		0	0 10.9	0 0.0	0 0 0 0	0 0 0 0 0 0 0 0 0 0
INIT. CONC.	3.2	3.2	-	0.3	3.2	3.2	-	-	÷	-	-	-	-	-	-			** **		* + * *	* - * - *
DATE	10/26/90	10/5/90	10/4/90	8/31/90	10/25/90	10/3/90	4/24/91	4/24/91	4/23/91	4/26/91	4/23/91	4/24/91	4/24/91	4/23/91	4/26/91		4/23/91	4/23/91 4/24/91	4/23/91 4/24/91 4/24/91	4/23/91 4/24/91 4/24/91 4/23/91	4/23/91 4/24/91 4/23/91 4/23/91
B	VERO	VERO	KERO	KERO	VERO	VERO	VERO	VERO	VERO	VERO	KERO	VERO	VERO	VERO							VERO VERO VERO
VIRUS	VEE	띬	>	⊭	¥	₽	⊭	щ	አ	Ŗ	ΡT	۶	щ	አ	¥	Ы		:⊭	∶⊭ሣ	F F S	┊╞╴╝╝
CTR #	B836749-F010	B836749-F010	B836749-F010	B836749-F010	B836749-F010	B836749-F010	B836749 F016	B836749 F016	B836749 F016	BB36749 F016	B836749 F016	B836749 F017		B836749 F018	B836749 F018 B836749 F018	B836749 F018 B836749 F018 B836749 F018 B836749 F018	B836749 F018 B836749 F018 B836749 F018 B836749 F018 B836749 F018				
AVS #	AVS-007071	AVS-007071	AVS-007071	AVS-007071	AVS-007071	AVS-007071	AVS-009458	AVS-009458	AVS-009458	AVS-009458	AVS-009458	AVS-009459	AVS-009459	AVS-009459	AVS-009459	AVS-009459		AVS-009460	AVS-009460 AVS-009460	AVS-009460 AVS-009460 AVS-009460	AVS-009460 AVS-009460 AVS-009460 AVS-009460

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TAI	0 87	7.55	00- F	46.0	29.69	12.67	35.95	12.33	16.41	2.04	9.89	4.52	0	17.13	7.35	15.83	0	5.58	0.04	6.82	42.4	23.51	38.97	17.48	16.6	0	0	0	0	13.21	11.84	5.56	5.24	7.29	5.72	6.23	21.13	0.77	30 82
		٨	•		^	•			۸		۸				۸			۸	•		۸	٨			۸						۸				۸		۸	۸	
S	0 63	1.51			6.21	1.28	10.22	2.99	3.15		2.06	1.09		2.37	0.53	2.34				0.92	12.88	1.47	12.21	1.65	2.3					1.86				1.3			3.34		6.44
TI50	1.513	2.287	0	0	8.727	2.261	18.897	6.706	4.905	0	3.422	2.276	0	6.694	1.865	4.133	0	0	0	2.981	24.006	2.656	21.78	22.348	4.162	0	0	0	0	3.709	0	0	0	2.652	0	0	5.229	0	12.113
TC95	309	307	320	300	315	305	304	299	297	320	320	314	290	309	313	305	100	98.9	98.9	100	95.8	97.1	96.3	93.3	91.9	100	100	100	85.8	100	100	96.4	100	100	308	93.4	100	262	100
0								-	~	_	•	_						_	_		•	_		_		_	^	^ _			^			^	_	_	^		
TCS	144	191	192	124	208	171	161	112	97.2	320	> 320	168	78.4	147	145	171	29	51.	50.3	27	47.9	53.4	40.7	33.5	28	100	34	27.4	21	27.6	44.6	41.7	44	65.3	84.3	25.6	64.6	57	45.4
TC25	50.9	126	88:7	52.1	148	97.4	86.8	50.1	62.5	320	192	80.6	37	52	41.1	96.9	14.2	27.2	21	8.3	25.7	29.5	22.8	2.47	15.5	100	17.3	12.9	10.4	13.9	20.2	21.3	21.8	32	40.7	16.6	41.3	7.11	24.1
1095	0	0	0	0	85.7	0	31.1	0	0	0	0	0	0	0	0	0	0	0	0	0	9.31	Ģ	8.32	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	9.06
1050	95.2	83.3	0	0	23.8	75.8	8.5	16.8	19.8	0	93.5	73.8	0	21.9	77.6	41.5	0	0	0	9.07	2	20.1	1.87	1.49	6.73	0	0	0	0	7.44	0	0	0	24.6	o	•	12.4	0	3.74
IC25	51.2	47.1	61.1	52.1	13.5	35.1	4.31	8.27	12.9	0	53.5	39.2	•	10	20.1	13.5	20.1	16.4	80 	4.02	0.856	1.5	0.769	0.58b	1.84	0	18.6	29	0	1.89	4.02	9.27	15.3	11.9	17.6	0	5.56	17.9	1.79
INIT. CONC.	-	*	-	***	-	-	-	-	-	-	-	-	-	-	-	-	0.32	0.32	0.32	0.32	0.32	0.32	0.32	0.32	0.32	0.32	0.32	0.32	~	0.32	0.32	0.32	0.32	0.32	-	0.32	0.32	-	0.32
DATE	8/31/90	10/3/90	10/25/90	8/30/90	10/25/90	10/2/90	8/30/90	10/25/90	10/2/90	8/28/90	10/26/90	10/5/90	10/11/90	8/31/90	10/25/90	10/3/90	8/31/90	10/25/90	10/3/90	8/30/90	10/25/90	10/2/90	8/30/90	10/25/90	10/2/90	8/28/90	10/26/90	10/5/90	10/11/90	8/31/90	10/25/90	10/3/90	8/31/90	10/25/90	10/3/90	8/30/90	10/25/90	10/2/90	8/30/90
CELL	VERO	VERO	VERO	KERO	VERO	VERO	VERO	VERO	KERO	VERO	VERO	KERO	VERO	VERO	KERO	VERO	VERO	VERO	KERO	KERO	VERO	VERO	VERO	VERO	VERO	KERO	KERO	KERO	S	KERO	KERO VERO	VERO	VERO	KERO	VERO	VERO	VERO	VERO	KERO
VIRUS	щ	щ	щ	ΡŢ	Ы	μ	ጽ	ጽ	ሦ	¥	¥	Å	>	₽	⊬	¥	щ	щ	щ	μ	ΡŢ	Γq	ጽ	ይ	ጽ	₩	¥	₩	>	⊭	₽	₽	щ	ч	띵	ЪТ	ЪТ	ΡŢ	ጽ
CTR#	B848528-F011	B848528-F012	B848528-F013																																				
* SVA	AVS-007083	AVS-007084	AVS-007085																																				

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VIRI	US CI	표	DATE	INIT. CONC.	IC25	IC50	IC95	-	C25	¥	50	Ĕ	395	T150		s		TAI
	۲	2	10/25/90	0.32	1.89	3.69	0	0	0.83	4		-	00	11.417		0.22		14.8
	۳	8	10/2/90	-	4.28	7.85	0		2.2	4	7.0	2	73	5.189		2.83	^	14.5
U1	٣	<u>e</u>	8/28/90	0.32	0	0	0	·	00)	-	00	-	00	0				<u>-</u>
U1	۳	2	10/26/90	0.32	15.1	25.5	96.8	٨	001	~	00	-	^ 00	3.927	۸	3.93	۸	18.4
Ű.	Ψ.	ŝ	10/5/90	-	21.2	59.3	0		3.8	-	17	ີ ຕ	20	2.984		0.4		4.51
Ś	Ч.	0	11/1/90	0.32	20.1	0	0	-	8.01	õ	4.0	Ō	8.5	0				0
	5		10/25/90	0.32	3.02	21.1	0		33	ŝ		-	00	2.803		1.56	۸	16.0
-	5	2	10/3/90	-	10.9	0	0		23.1	ŝ	9.2	2	98	0				4.83
	5	ŝ	8/31/90	-	5.12	0	0	0,	0.07	2	 5. I	m	20	0				2.76
	٣		10/25/90	0.32	4.81	0	0	•	19.7	ē		-	00	0			۸	12.3
	3	0H	10/3/90	0.32	0	0	0	•	18.7	e	5.	-	00	0				2.06
	۲	0	8/30/90	-	7.59	0	0		.66	Ŧ	4.9	7	8.8	0				1.88
L.	۳	0 E	10/25/90	0.32	3.94	6.06	د د	•	19.2	Ñ	9.3 2	0,	33	4.674		3.16	۸	17.3
.	۳	2	10/2/90	0.32	0	0	0		18.7	N	8 N N	-	00	0			۸	2.17
	۲	Q	06/06/8	-	-	1.25	2.91	Ŭ	5.18	G	16	Ñ	9.9	7.317		4.94		21.6
••	۲	ŝ	10/25/90	0.32	1.06	1.74	0	•	13.7	Ñ	3.9	80	8.7	13.734		7.87	۸	26.91
••	۳	ŝ	10/2/90	0.32	4.68	7.71	0	•	15.2	Ñ	3.3	60	5.2	3.017		1.97		8.81
ш	۲	ê	8/28/90	-	0	0	0	Ű	58.3	-	30	ო	20	0				0.46
ш	5		10/26/90	0.32	6.04	21.7	0	•	9. 6	~	00	-	^ 00	4.605		0.64	۸	10.6
w	5	8	10/5/90	0.32	24.9	0	0	Ť	5.78	-	5.7	-	00	0				0
-	۲		10/11/90	-	0	0	0	•	14.1	Ñ	2.5	0	2.8	0				0.79
	5		10/25/90	0.32	5.31	0	0	•	15.8	Ñ	2.5 6.5	-	00	0			۸	8.12
	۲	2	10/3/90	0.32	6.84	0	•	•	16.4	Ň	6.9	+	00	0				4.42
	5	<u>0</u>	8/31/90	0.32	4.54	8.11	0	w	1.79	.,	0	-	00	3.698		1.08		8.05
	5	0 E	10/25/90	0.32	5.31	0	0		25.7	-	8	-	00	0				10.9
	5	ŝ	10/3/90	0.32	14.9	0	0		29	Ō		-	00	0			۸	8.08
_	5	Q	8/30/90	0.32	4.21	0	•		60.	Ñ	6.6	•••	00	0				2.91
-	۲	e E	10/25/90	0.32	1.76	3.62	9.03		24.1	ð	с. С.	-	00	26.091		6.67	۸	33.3
L.	5	O	10/2/90	0.32	5.01	0	0		2.5	ίΩ.	<u>.</u>	-	00	0				10.9
	۲	Q	8/30/90	0.32	0.477	0.97	0		9.5	č	Ξ	-	00	31.982		9.77		35.1
	۶	O Hi	10/25/90	0.32	0.702	1.41	0	•••	6.75	Ñ	 	-	00	16.531		4.07		22.8
	۲	OH	10/2/90	0.32	3.39	4.86	9.3	-	9.6	Ñ	9.3	-	00	6.026		4.04		23.41
ш	5	O	8/28/90	0.32	28.6	0	0	Ŭ	5.78	e	3.3	Ö,	3.3	0				0
ш	۲	O	10/28/90	0.32	11.3	20.7	0		40	~	80	-	^ 00	4.834		1.93	۸	13.4
ш	۲	2	10/5/90	0.32	18.1	0	0	-	1.8	4	9.3	-	00	0				0
~	۲	B	10/11/90	-	17.9	0	0	0,	.06		4	2	38	•				0
••	۲	0	8/31/90	0.32	1.35	3.88	0		18	e	5.1	-	00	8.099		4.65		24.2
	۲	B	10/25/90	0.32	3.66	0	0	-	9.3	4		-	00	0			۸	12.7
	۲	8	10/3/90	0.32	6.96	0	0		21.8	47	9	-	00	0				7.65

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TAI	0	2.18	9	• 0	0	0	0.06	0	1.16	4.1	2.94	0	0.01	0	3.79	0	0	1.7	0.31	0	0.8	1.6	36.77	31.69	32.11	22.55	1.3	0	3.62	20.15	22.36	0	1.84	1.55	13.64	7.45	0	0	3.77
										۸					۸						٨	۸	۸	۸	۸	۸	۸		۸	۸	۸				۸	۸			
S																							10.67	6.86	0.03	2.38				2.11	5.68				2.67				
																							۸	۸		۸				۸	۸								
T150	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	10.666	6.864	10.144	2.382	0	0	0	2.114	5.676	0	0	0	4.213	0	•	0	•
																							۸	۸	۸	۸				۸	۸								
TC95	320	240	320	310	272	320	320	1000	320	1000	320	903	320	1000	320	320	1000	100	320	320	1000	1000	1000	1000	1000	1000	1000	1000	320	1000	1000	97	955	845	924	800	830	811	1000
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TCSC	48.2	70.6	55.3	158	31.5	45.9	320	845	320	1000	320	228	320	1000	. 320	320	745	100	320	320	1000	1000	1000	1000	1000	1000	1000	1000	320	1000	1000	57.8	445	241	269	233	146	144	614
										^				۸	^				^	^	^	^	^	^	^	•	•	^	^	^	•								
TC25	19.8	51.3	24.9	58.2	20.1	23.4	218	459	320	505	320	87	320	830	320	320	391	100	320	520	1000	1000	1000	1000	2.67	1000	1000	90.9	320	1000	1000	36.3	214	171	171	166	39.2	67.9	361
															۸				۸	٨	۸	۸	۸	۸		۸	۸		۸	۸	۸								
1095	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	1030	0	1070	947	0	0	0	984	0	0	0	0	0	0	0	0	0
1050	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	93.8	146	98.6	420	0	0	0	473	176	0	0	0	63.8	0	0	0	0
IC25	0	0	0	•	0	0	0	0	264	402	163	0	0	0	247	0	0	0	0	0	0	0	24.8	63.4	26.3	147	0	927	0	290	120	0	132	0	41.5	88.1	0	0	179
INIT. CONC.	-	-	-	-	-	-	-	3.2		3.2	-	3.2		3.2		-	3.2	0.32	-	-	3.2	3.2	3.2	3.2	3.2	3.2	3.2	3.2	-	3.2	3.2	-	3.2	3.2	3.2	3.2	3.2	3.2	3.2
DATE	8/31/90	8/30/90	8/30/90	8/28/90	10/11/90	8/31/90	8/31/90	10/3/90	8/30/90	10/2/90	8/30/90	10/2/90	8/28/90	10/5/90	10/11/90	8/31/90	10/3/90	9/12/90	10/11/90	10/11/90	10/25/90	10/3/90	10/25/90	10/2/90	10/25/90	10/2/90	10/26/90	10/5/90	11/1/90	10/25/90	10/3/90	10/11/90	10/25/90	10/3/90	10/25/90	10/2/90	10/25/90	10/2/90	10/26/90
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<u>B. Scale-up Isolation of Narciclasine and Pancratistatin and Synthetic</u> <u>Modifications of Narciclasine</u>

We have pursued the pancratistatin family of antiviral leads as a top priority. The research results here have been very encouraging and pancratistatin, isonarciclasine, <u>cis</u>-dihydronarciclasine as well as <u>trans</u>dihydronarciclasine have proved to be quite promising. The most exciting antiviral result has been the discovery in USAMRIID's laboratories that pancratistatin will effectively treat the <u>in vivo</u> experimental version of Japanese Encephalitis.

We have completed reisolation of pancratistatin (1) from 1/2 ton of *Pancratium littorale* collected in the Republic of Seychelles. Enough pancratistatin is now available for the next series of <u>in vivo</u> experiments.

In order to ensure that pancratistatin is available for future clinical trials, we have been expanding our botanical research aimed at this objective. Presently we have grown 900 Pancratium littorale plants and are preparing to triple that amount in the coming year. In turn, that will ensure a steady clinical supply. In the botanical research, we have also been exploring tissue culture methods of cloning Pancratium littorale and this avenue is also becoming increasingly productive. The evaluation of other Pancratium and Hymenocallis species for pancratistatin and related compounds has continued and led to discovery of a new source for pancratistatin, namely, Hymenocallis calatay from a Singapore collection.

Related plants of the Amaryllidaceae were evaluated for antiviral constituents. We completed the isolation and structural elucidation of <u>trans</u>dihydronarciclasine (2) as the active (Japanese encephalitis <u>in vitro</u>) and anticancer (P388 lymphocytic leukemia) constituent of *Zephyranthes candida*.
Meanwhile we have undertaken a world-wide procurement of Amaryllidaceae plants in the Pancratium and Hymenocallis genera. Twenty such species obtained were evaluated for antiviral constituents related to pancratistatin. One of these collected in Singapore in the Hymenocallis genus (unpublished) yielded 7-deoxytrans-dihydronarciclasine (3) as the principle antiviral constituent. Interestingly this new natural product was just prepared by us using a semisynthetic route from 7-deoxy-narciclasine isolated from Pancratium littorale and represents one of those rare examples where the synthetic product preceded discovery of the natural product.

A major effort has been devoted to scaling-up the isolation of narciclasine from a Narcissus species.

We have isolated narciclasine (4) from a large scale collection (10,000 bulbs) of Narcissus in preparabilis. So far about 50 grams of narciclasine has been isolated by this procedure and another 100 grams is in progress. The narciclasine is being employed in a semi-synthetic approach to pancratistatin. For experimental details of the semi-synthetic approach to pancratistatin from narciclasine, please refer to Appendix B. Epoxide (5) has been prepared and converted to ketone 6. The ketone reduction step is being studied in detail to increase yields of the β -alcohol corresponding to pancratistatin. Additional quantities of narciclasine are being converted to isonarciclasine (7) another antiviral derivative of pancratistatin found by USAMRIID to have in vitro While total synthesis of pancratistatin was recently antiviral activity. achieved it involved some 30 steps and was not practical for scale-up. Thus, we plan to use the semi-synthesis from narciclasine for pancratistatin and related compounds to meet future clinical needs. Toward that end some 30,000 bulbs (about seven tons) of Narcissus incomparabilis was obtained in September, 1990,







PANCRATISTATIN 1 trans-Dihydronarciclasine 2

7-Deoxy-<u>trans</u>-dihydronarciclasine 3



Narciclasine 4





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Lycorine 8 and the scale-up isolation of additional narciclasine has begun.

Other structure/modification studies will be directed at the Amaryllidaceae antiviral constituent lycorine (δ). Here modification will involve various epoxidation and hydroxylation procedures along with conversion to glycoside derivatives. Fortunately we have already isolated 50 grams of lycorine for these investigations from Narcissus incomparabilis.

Detailed reports have been included in the manuscript section of this final summary of our research directed at evaluating structure/activity relationships in our bryostatin series of marine animal constituents. Bryostatins 1 and 2 have been found active against the HIV-1 <u>in vitro</u> syncytia screen. The contract renewal beginning June 28 will allow each of the most important RNA antiviral leads noted above to be vigorously pursued.

III. Near Term Plans

The discovery that pancratistatin will effectively retard USAMRIID's \underline{in} <u>vivo</u> Japanese Encephalitis has opened the way to a new generation of antiviral drugs. The combination of isolation and synthesis from narciclasine will ensure eventual large-scale production of pancratistatin and related compounds.

Very importantly for the future, we have now underway the fractionation research on a number of high priority USAMRIID RNA antiviral actives along with the HIV actives noted above.

In short the USAMRIID research program at the ASU-CRI has been making, as usual, excellent progress concerned with discovery and development of potentially useful antiviral drugs. Again, this was only made possible by the most necessary USAMRIID support and collaborative endeavors.

IV. Publications, Meeting Abstracts, Personnel

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"Potentiation of Ara-C Metabolism and Cytotoxicity in Leukemic Cells by Bryostatin-1, a Potent Activator of Protein Kinase C," <u>S. Grant</u>, G. R. Pettit, and C. McCrady

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"Modulation of the Response of Highly Purified Human Hematopoietic Progenitor Cells (MY-10⁺) to Hematopoietic Growth Factors by Bryostatin 1," C. McCrady, F. Lei, G. Pettit, and <u>S. Grant</u>

"The PK-C Activator Bryostatin 1 Potentiates the Radioprotective Effects of Recombinant Granulocyte-Macrophage Colony Stimulating Factor Toward Normal Human Hematopoietic Progenitor Cells," <u>S. Grant</u>, G. R. Pettit, and C. McCrady

"Tissue Culture of *Pancratium littorale* for Production of Pancratistatin, An Anticancer Drug," <u>R. A. Backhaus</u>, J. Ho, G. R. Pettit, III, D.-S. Huang, and G. R. Pettit, 23rd Int'l. Horticulture Congress, Italy, 8/27 - 9/1/90.

"Bryostatins Define the Role of Protein Kinase C in Pituitary Tumor Cell Proliferation," <u>E. A. Mackanos</u>, G. R. Pettit, and J. S. Ramsdell, The Endocrine Society, Bethesda, MD, June 1991.

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ANTINEOPLASTIC AGENTS, 162.1 ZEPHYRANTHES CANDIDA

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ABSTRACT.—The Chinese medicinal plant Zephyranthes candida was found to contain a cytostatic constituent. Separation of a *n*-BuOH extract directed by results of a bioassay employing the P-388 lymphocytic leukemia led to *trans*-dihydronarciclasine [2] as the principal cytostatic agent with ED_{50} 3.2 × 10⁻³ µg/ml.

Amaryllidaceous plants such as Narcissus poeticus were recorded in the Bible as well-established treatments for cancer (1), and others were in use by the Greek physicians of the fourth century BC (2). The first isolation, in 1877 (3), of a biologically active Amaryllidaceae constituent, the now well-known lycorine (4), was an early achievement of organic chemistry, and such studies have been intensifying (4-7). In 1984, we reported discovery and structural elucidation of a strongly antineoplastic phenanthridone designated pancratistatin [1] produced by plants of Pancratium littorale (2,8) and Zephyranthes grandiflora (5).

In 1964, extracts of the medicinal (5) Zephyranthes candida (Lindl.) Herb. (obtained in Hong Kong) had already proved active (KB cell line from a human epidermoid carcinoma of the nasopharynx) in the U.S. National Cancer Institute's exploratory research program, but we were unable to obtain a re-collection (People's Republic of China) until 1982. Earlier (1955) Boit and Ehmke (9) isolated four alkaloids from the Dutch Z. candida representing the pyrrolo[de]phenanthridine (lycorine), pretazettine (tazettine), and 5,10b-ethanophenanthridine (haemanthidine and nerinine) ring systems. The study was extended in 1964-65 (10,11) to isolation of dihydrolycorine and zephyranthine from a Japanese variety and in 1978 to a flavone glycoside (12). We now have found the principal cytostatic (murine P-388 lymphocytic

leukemia, PS system) (12, 13) constituent of Z. candida to be trans-dihydronarciclasine [2] previously (14) prepared by hydrogenation of narciclasine [6] and heretofore unknown as a biosynthetic product.



Ground bulbs of Z. candida were extracted with CH_2Cl_2 -MeOH (1:1) at ambient temperature. After addition of H_2O , the aqueous phase was concentrated and extracted with *n*-BuOH. The PS-active (cell line) *n*-BuOH extract was concentrated and triturated with MeOH to provide a fraction that was separated (guided by PS bioassay) by successive

¹For Part 161, see P.M. Blumberg and G.R. Pettit in BBA Reviews on Cancer, in preparation.

Jan-Feb 1990]

Sephadex LH-20 and Si gel cc steps. The resulting enriched active (ED₅₀ 0.0034 µg/ml) fraction was devoid of acetate groups (ir), and the major component coeluted [tlc, CH₂Cl₂-MeOH-H₂O (90:10:0.1)] with an authentic synthetic specimen of trans-dihydronarciclasine. The fraction was acetylated and separated on a column of Si gel to yield trans-dihydronarciclasine peracetate [3] (PS ED₅₀ $3.2 \times 10^{-3} \,\mu g/ml)$ as the major component. The structure of the peracetate was established by detailed spectral analysis (2) and comparison with an authentic sample as well as with the product obtained by catalytic hydrogenation (Adam's catalyst in HOAc at 50 psi) of narciclasine, followed by acetylation. Hydrogenation afforded as the major product the expected cis-dihydronarciclasine accompanied by the trans isomer. Facile deacetylation of the phenolic acetoxy group was observed during chromatography and in MeOH solutions to give the 7-hydroxy-2,3,4-triacetoxy derivative 4. Trans-dihydronarciclasine (prepared from the acetate) was found to strongly inhibit the PS leukemia with ED_{50} 0.0032 µg/ml, while the synthetic cis-dihydro analogue 5 led to PS $ED_{50} 0.024 \ \mu g/ml.$

Isolation of *trans*-dihydronarciclasine [2] as the major antineoplastic constituent of Z. candida has revealed another interesting and potentially useful Amaryllidaceae biosynthetic product. Further study of this very productive plant family for anticancer and other medically useful components will doubtless prove rewarding and is in progress.

EXPERIMENTAL

GENERAL METHODS.—Details of general procedures and chromatographic techniques were provided in our earlier summaries (2,5).

PLANT MATERIAL. —Z. candida PR #55337, NSCB657832 was re-collected in China in 1981 (received February 1982) as part of the NCI-USDA collaborative program directed by Drs. J.L. Hartwell and M. Suffness. A voucher specimen is maintained at the USDA, Beltsville, MD, and in the ASU-CR1.

Extraction .--- Freshly ground bulbs (18 kg) were stored in MeOH-CH₃Cl₃(1:1)(32 liters) for 10 days. Addition of H₂O (15% by volume) caused separation of the CH₃Cl₃ phase. MeOH and CH₂Cl₂ were added to the aqueous phase to increase the original total volume by 50 and 25%, respectively. The plant was extracted with this mixture (2:1:0.5 ratio of MeOH-H2O to added MeOH and CH₂Cl₂) for a further 80 days. Addition of H₂O (25% by volume) allowed the CH₂Cl₂ phase to separate, which was combined with the first CH2Cl2 extract and concentrated to a 109-g residue (PS ED₅₀ 3.5 µg/ml). The aqueous phase was concentrated and partitioned between H_2O (6 liters) and *n*-BuOH (4 × 6 liters). Concentration of the n-BuOH extract to a small volume and addition of MeOH (2 liters) gave an active MeOH-soluble fraction (149 g, PS EDso 0.27 µg/ml). Upon further dilution with MeOH (600 ml) and CH₂Cl₂ (400 ml) the solution was filtered to yield 28 g of a solid (PS ED₅₀ 1.6 µg/ ml). The filtrate was chromatographed on a column of Sephadex LH-20 (2.5 kg) using MeOH-CH₂Cl₂ (3:2) as eluent.

ISOLATION OF TRANS-DIHYDRONARCICLASINE [2].-Elution (the preceding LH-20 column) between volumes 7215-16950 ml gave a 6.2-g fraction (PS ED₅₀<0.02 µg/ml). Trituration with Me₂CO (50 ml) provided a light orange solid (2.72 g, PS ED₅₀ 0.016 μ g/ml) and a soluble fraction (3.5 g, PS ED₅₀ 0.0043 μ g/ml). When the orange solid was triturated with MeOH- CH_3Cl_3 (1:1) (3 × 10 ml, 1 day), followed by MeOH (5 ml, 2 days), a soluble fraction (2.58 g) was obtained similar (by tlc) to the Me2CO-soluble fraction. An aliquot of the Me₂CO-soluble fraction (1.76 g) and the latter soluble fraction (2.58 g) were combined and the mixture subjected to rapid chromatography on a column of Si gel (200 g). Gradient elution with CH₂Cl₂ (1 liter) and CH₃Cl₃-MeOH (99:1 to 95:5 to 9:1) (2 liters) gave a fraction (1.05 g) which was triturated with MeOH (5 ml, 1 day) to give a buff-colored solid [0.20 g, PS ED₅₀ 0.0034 µg/ml, ir (KBr) 3350, 1660, 1460, 1340, 1280, 1225, 1060, 1025 cm⁻¹]. Half of the solid was acetylated [Ac2O-pyridine (1:1) (6 ml), 24 h, room temperature], and the product (0.12 g) was chromatographed on a column of Si gel (Lobar B column). Development with CH2Cl2 (200 ml) and CH₃Cl₃-MeOH (99:1) (400 ml) followed by CH2Cl2-MeOH (49:1) (all affording between 675 and 725 ml) total eluent volume, trans-dihydronarciclasine-2,3,4-triacetate [4] (16 mg) which recrystallized from MeOH-CH₃Cl₃ as small colorless needles: mp 309-311° [lit. (13) mp 293°]; $[\alpha]^{42}$ ti +81.94° (i = 0.72, CHCl₃); uv λ max MeOH (log €) 231 (4.04), 239 (4.01), 280 (3.75), 310 (3.33) nm; ¹H nmr (400 MHz, $CDCl_3$ 1.914.(1H, ddd, J = 14.0, 12.5, 3.0

Hz, H-1 β), 2.086(6H, s, 2 × Ac), 2.137(3H, s, Ac), 2.432(1H, ddd, J = 14.0, 3.5, 3.2 Hz, H-1 α), 3.134 (1H, ddd, J = 12.7, 12.5, 3.5 Hz, H-10b), 3.777 (1H, dd, J = 12.7, 11.8 Hz, H-4a), 5.175 (1H, dd, J = 11.8, 3.0 Hz, H-4), 5.189 (1H, m, H-3), 5.438 (1H, dd, J = 3.2, 3.0 Hz), 5.855 (1H, s, NH), 6.037, 6.049 (1H, each, d, J = 1.2 Hz, OCH₂O), 6.323 (1H, s, H-10), 9.704 (1H, s, ArOH). Acetylation [Ac₂Opyridine (1:1)] led to *trans*-dihydronarciclasine peracetate [**3**] identified by tlc and ir spectra (in CHCl₄) with an authentic specimen.

Continued elution between volumes 725–760 ml gave a mixture (14 mg) of the above triacetate and *trans*-dihydronarciclasine peracetate_and between volumes 760–810 ml *trans*-dihydronarciclasine peracetate (80 mg).

Recrystallization from MeOH/CH₂Cl₂ afforded a pure specimen of 3 as colorless needles: mp 181-182° [lit. (14) 188-189°]; $[\alpha]^{(1)}D + 123.9^{\circ}$ (c = 1.13, CHCl₃) [lit. (14) $[\alpha]^{20}D + 128.5^{\circ} (c = 0.82, CHCl_3)];$ uv λ max MeOH (log €) 231 (4.10), 239 (4.09), 280 (3.82), 310 (3.40) nm; ir (KBr) v max 3600, 3500, 3330 (sh), 3310, 1760, 1730 (sh), 1670, 1634, 1505, 1487, 1460, 1371, 1345, 1298, 1255, 1235, 1172, 1080, 1051, 1031, 930 cm⁻¹; ¹H nmr (400 MHz, CDCl₃) 1.906 (1H, ddd, J = 14.0, 12.7, 3.0 Hz, H-1), 2.054, 2.071, 2.139 (3H each, Ac), 2.364 (3H, ArOAc), 2.428 (1H, ddd, J = 14.0, 3.5, 3.2Hz, H-1 α), 3.140(1H, ddd, J = 12.7, 12.0, 3.5Hz, H-10b), 3.762(1H, dd, I = 12.0, 10.8 Hz,H-4a), 5.156 (1H, dd, J = 10.8, 3.0 Hz, H-4), 5.192 (1H, m, H-3), 5.416 (1H, dd, J = 3.2, 3.0 Hz, H-2), 5.810 (1H, s, NH), 6.065, 6.073 $(1H \text{ each}, d, J = 1.2 \text{ Hz}, -O-CH_3-O), 6.642$ (1H, s, H-10); ¹³C nmr (22.63 MHz, CDCl₃) 170.34, 169.35, 169.14 (4C, 4×OCOMe), 163.35 (C-6), 152.40 (C-9), 139.63 (C-7), 137.00 (C-10a), 134.33 (C-8), 116.04 (C-6a), 102.91 (OCH2O), 102.00 (C-10), 71.62, 68.60, 67.43 ($3 \times CHOAc$), 52.41 (C-4a), 35.61 (C-10b), 27.00 (C-1), 21.02, 20.86, 20.70 (4c, $4 \times \text{OCOCH}_3$) ppm; hreims m/z [M]⁴ 477.1258 (3.09%) (calcd 477.1271 for C25H33NO11), 435.1165 (100%) (calcd 435.1166 for C₂₀H₂₁NO₁₀).

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We are very appreciative of financial assistance provided by the Arizona Disease Control Research Commission, Eleanor W. Libby, The Waddell Foundation (Donald Ware), the Fannie E. Rippel Foundation, Herbert K. and Dianne Cummings (the Nathan Cummings Foundation, Inc.), Virginia Piper, Lotte Flugel, Polly J. Trautman, Grant CA-30311-01-03 awarded by the National Cancer Institute and Contract NO1-CM-97262 with the Division of Cancer Treatment, NCI, NIH, and the U.S. Army Medical Research and Development Command under Grant No. DAMD17-89-Z-9021. For other assistance we thank Drs. J.M. Schmidt, V. Gaddamidi, M.I. Suffness, and Messrs P.J. Daschner and L. Williams. We thank Professor A. Mondon for samples of *trans*-dihydronarciclasine and its peracetate as well as the corresponding *cis*-dihydro-compounds (see Mondon and Krohn, 14).

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Antineoplastic Agents. 206. Structure of the Cytostatic Macrocyclic Lactone Combretastatin D-2¹

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The South African tree Combretum caffrum (Combretaceae) has been found to contain two new and cytostatic (P388 lymphocytic leukemia) macrocyclic lactones designated combretastatin D-1 (1, ED₅₀ 3.3 µg/mL) and D-2 (2, ED₉₉ 5.2 µg/mL). With the X-ray crystal structure of combretastatin D-1 (1) serving as an unequivocal reference point ¹³C NMR and high field (400 MHz) ¹H NMR spectral techniques were employed to assign structure 2 to combretastatin D-2.

The South African tree Combretum caffrum (Combretaceae) has been found to produce two cis-stilbenes, combretastatins A-1 and A-4, that strongly inhibit growth of the P-388 lymphocytic leukemia cell line (PS system) and tubulin polymerization.² Recently, we reported³ the iso-

 (3) Pettit, G. R.; Singh, S. B.; Niven, M. L. J. Am. Chem. Soc. 1988. 110, 8539.

lation and structure determination of an unexpected 17membered macrocyclic lactone designated combretastatin D-1 (1) from the same plant. We now summarize the



isolation and structural elucidation of another PS cell line inhibitory member of this unusual series of macrocyclic lactones named combretastatin D-2 (2) along with chemical

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⁽¹⁾ For the preceding paper, see: Pettit, G. R.; Singh, S. B.; Hogan, F.; Burkett, D. J. Med. Chem. In press. (2) (a) Pettit, G. R.; Singh, S. B.; Niven, M. L.; Hamel, E.; Schmidt, J. M. J. Nat. Prod. 1967, 50, 119. (b) Pettit, G. R.; Singh, S. B. Can. J. Chem. 1967, 65, 2390. (c) Pettit, G. R.; Singh, S. B.; Niven, M. L.; Schmidt, J. M. Can. J. Chem. 1968, 66, 406. (d) Pettit, G. R.; Singh, S. B.; Schmidt, J. M.; Niven, M. L.; Hamel, E.; Lin, C. M. J. Nat. Prod. 1968, 51, 517. (e) Lin, C. M.; Singh, S. B.; Chu, P. S.; Dempcy, R. O.; Schmidt, J. M.; Pettit, G. R.; Hamel, E. Mol. Pharmcol. 1968, 34, 200. (f) Pettit, G. R.; Singh, S. B.; Lin, C. M.; Hamel, E.; Alberts, D.; Garcia-Kendall, D. Experientia 1969, 45, 209.

Table I.	¹ H NMR Assignments for	Combretastatin D-2	: (2) and Der	ivatives 5a–c iı	a Deuteriochloroform	Solution in Solution
		(ppm) with Chlo	oroform as Ir	nternal Standa	rd	

position	2	5 a ª	5b	5e
2α	4.64, d, 6.8	4.64, d, 6.9	3.91, d, 11.4	4.05, d, 11.9
2β	4.64, d, 6.8	4.26, m	4.31, dd, 11.4, 7.0	4.56, dd, 12, 7.6
3	6.06, dt, 10.6, 6.8	2.09, 2.38, m	4.24, m	4.27, m
4	7.11, d, 10.6	3.72, m	2.81, dd, 12.9, 8.4	2.93, dd, 13.2, 11.4
		4.06, m	3.26, dd, 12.9, 5.1	3.60, dd, 13.2, 5.3
6	7.33, d, 8.4	7.33, br d, 9.7	7.35, dd, 8.3, 2.3	7.35, dd, 8.3, 2.1
7	7.09, d, 8.4	7.09, dd, 8.0, 1.7	7.05, dd, 8.4, 2.5	7.10, dd, 8.3, 2.5
12	6.85, d, 8.0	6.83, d, 8.2	6.84, d, 8.2	6.84, d, 8.2
13	6.63, ddd, 8, 1.8, 1.7	6.61, dd, 8.4, 1.7	6.61, dd, 8.3, 1.6	6.61, dd, 8.3, 1.8
15α	2.87, t, 5.0	2.82, m	2.72, br dd, 17.1, 8	2.61, br dd, 17.1, 7.2
15β	2.87, t, 5.0	2.85, m	2.96, br dd, 16.4, 9.9	3.05, br dd, 16.6, 10.6
16a	2.29, dt, 5.0, 1.7	2.25, m	2.23, ddd, 17, 10.5, 1.7	2.15, ddd, 15.2, 12, 1.4
16β	2.29, dt, 5.0, 1.7	2.30, m	2.35, ddd, 17, 8.2, 1.8	2.40, ddd, 16.8, 7.4, 1.4
18	7.33, d, 8.4	7.31, dd, 8.0, 1.9	7.31, dd, 8.0, 2.5	7.31, dd, 8.1, 2.2
19	7.09, d, 8.4	7.02, dd, 8.3, 2.0	7.02, dd, 8.2, 2.5	7.00, dd, 8.1, 2.5
20	5.07, d, 1.8	5.30, d, 2.0	5.23, d, 1.8	5.21, d, 1.8
11- OH	5.47, s	5.51, br s	5.50, br s	5.48, s
3-OH	-		2.07, d, 6.1	•

^a Two major conformers; the chemical shift of the major conformer is reported.

transformations of combretastatin D-1 undertaken as part of the original structural elucidation.³

A methylene chloride-methanol (1:1) extract of Combretum caffrum stem wood was initially fractionated and separated as described.^{2a,b} The fraction that previously yielded^{2b} combretastatin A-2 was subjected to a similar PS bioassay guided chromatographic separation sequence (a series of Sephadex LH-20 partition chromatograms using hexane-toluene-methanol, 3:1:1, and silica gel column chromatographic procedures employing various combinations of hexane-ethyl acetate as eluant) afforded combretastatin D-2 (2, 5.8 mg from 77 kg of wood), which exhibited PS ED₅₀ 5.2 μ g/mL.

As with combretastatin D-1 (1) mass spectral analysis of combretastatin D-2 indicated a molecular formula $(C_{18}H_{16}O_4)$ with 11 double-bond equivalents. The infrared spectrum of lactone 2 showed absorption due to a lactone or an ester carbonyl (at 1728 cm⁻¹), hydroxyl group (3436, 3429 cm⁻¹), and aromatic rings. The ¹H NMR spectrum contained signals corresponding to methylene adjacent to carbonyl, a benzylic methylene, an oxymethylene, eight olefinic and/or aromatic protons, and a shielded aromatic proton (Table I). The proton NMR spectrum was assigned on the basis of 2D ¹H NMR and ¹H-COSY techniques.⁴ The spin systems were (a) ArCH₂CH₂CO-; (b) -OCH2CH=CH-; (c) a para-substituted aromatic ring; and (d) an ortho, ortho, meta-substituted aromatic ring. On the assumption that combretastatin D-2 had a lactone ring. all the double-bond equivalents were thereby accounted for. The ¹³C NMR spectrum of olefin 2 was consistent with this deduction. The ¹H NMR spin systems were assembled³ on the basis of NOEDS experiments.

Comparison of the ¹³C NMR spectrum (Table II) of combretastatin D-2 with the spectrum of combretastatin D-1 (1, confirmed by X-ray crystal structure determination) provided unequivocal support for the proposed structure. The carbon-13 spectrum of lactone 2 was found to be essentially identical with that of combretastatin D-1, except for the olefinic carbon signals. Since the original ¹³C assignments for combretastatin D-1 were based on direct one-bond ¹H, ¹³C correlation using the ¹H, ¹³C-COSY⁵ experiment (ambiguous for the quaternary carbons), it became necessary to assign the carbon resonances of the more abundant combretastatin D-1. Therefore, all

Table II. ¹³C NMR Assignments for Combretastatin D-1 (1 with HMBC Correlations) and Combretastatin D-2 (2) in Deuteriochloroform

Deuteriocinororom				
position	1	2	1 (HMBC)	
2	62.56	59.06	C-2 → H-3	
3	52.99	137.74	$C-3 \rightarrow H-2\alpha$, $H-2\beta$, $H-4$	
4	55.84	135.45	C-4 H-2α, H-2β, H-6	
5	132.44	132.01	C-5 → H-4, H-7, H-19	
6	128.83	129.09	C-6 → H-4, H-18	
7	123.95	123.89	C-7 → H-19	
8	156.01	155.6	C-8 → H-7, H-19	
10	149.09ª	149.32°	C-10 → H-12, 11-OH, H-20	
11	142.62ª	142.48ª	C-11 → H-12, 11-OH, H-20	
12	115.38	115.3 9	C-12 → H-13, 11-OH	
13	122.03	121.89	C-13 → H-15α, H-12, H-20	
14	131.90	131.14	C-14 \rightarrow H-12, H-15 α , β , H-16 α , β	
15	26.97	26.89	$C-15 \rightarrow H-16\alpha,\beta, H-20$	
16	31.24	32.42	$C-16 \rightarrow H-15\alpha,\beta$	
17	172.53	173.30	C-17 \rightarrow H-2 α , β , H-15 β , H-16 α , β	
18	126.34	125.68	C-18 → H-4, H-6	
19	123.14	123.89	C-19 → H-7	
20	112.24	112.58	C-20 → H-12, H-13, H-15α	

 a Assignments with identical superscripts in vertical columns may be interchanged.

the carbon resonances of epoxide 1 were assigned by heteronuclear multiple bond connectivity (HMBC) experiments.⁶ The observed connectivities are recorded in Table II. The previous assignments³ remain unchanged except for reversal of the quaternary carbon signals at C-5 and C-14. Both protons at C-2 gave a strong cross correlation with the carbonyl group, clearly confirming presence of the lactone. Similarly, assignment of the chemical shift of C-8 was confirmed by correlation with H-7 and H-19. Definite assignment of C-10 and C-11 remains uncertain because of common correlation cross peaks but this is of little consequence. Combretastatin D-2 must have structure 2 and this was further corroborated by the mass spectral fragmentation pattern (structure 4).

Attempts to convert combretastatin D-1 (1) into D-2 (2) and thereby provide further support for the D-2 structure were unsuccessful. For example, reaction of combretastatin D-1 with Zn/Cu couple⁷ gave hydrocarbon derivative 5a and alcohol 5b. Several other reagents such as $P_2I_4^{\ 8}$

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 (8) Suzuki, H.; Fuchita, T.; Iwasa, A.; Mishina, T. Synthesis 1978, 905.



or sodium iodide with acetonitrile and trifluoroacetic anhydride⁹ were either unreactive under the conditions studied or caused decomposition. When the epoxide group of combretastatin D-1 was hydrogenated to give alcohol **5b**, subsequent treatment with thionyl chloride in pyridine yielded 3,4-deoxy-3-chlorocombretastatin D-1 (**5c**).



Combretastatins D-1 and D-2 both contain a new oxygen heterocyclic ring (17-membered exterior and 15-atom interior). We propose the designation caffrane for this new macrocyclic ring system. Combretastatin D-2 is probably a penultimate biosynthetic precursor of combretastatin D-1 and may originate biosynthethically as noted earlier³ from two units of tyrosine or equivalent via o-phenol coupling, deamination, partial reduction, and lactonization. When additional quantities of combretastatins D-1 and D-2 become available, the biological properties of these biosynthetic products will be further ascertained.

Experimental Section

Synthetic intermediates were used as received from Sigma-Aldrich Co. All chromatographic solvents were redistilled. Sephadex LH-20 (particle size 25-100 μ m) was obtained from Pharmacia Fine Chemicals AB (Uppsala, Sweden) and silica gel 60 (70-230 mesh) was supplied by E. Merck, (Darmstadt, Germany). Analtech, Inc. (Newark, DE) silica gel GHLF U (0.25-mm layer thickness) was employed for thin layer chromatograms. Development was performed with ceric sulfate-sulfuric acid spray reagent (heated at approximately 150 °C for 5-10 min) and/or by use of ultraviolet light. Solvent extracts of aqueous solutions were dried over anhydrous sodium sulfate.

All melting points are uncorrected and were observed with a Kofler-type hot-stage apparatus. Ultraviolet spectra were obtained on a Hewlett-Packard Model 8540A UV/VIS spectrophotometer. Infrared spectra were measured with a Nicolet FT-IR Model MX-1 unit. Nuclear magnetic resonance spectra were obtained with a Bruker AM-400 instrument using deuteriochloroform as solvent and the residual chloroform signal as an internal standard (δ 7.256). The ¹³C NMR multiplicities were determined by using the APT sequence. Mass spectral measurements were performed with a MS-50 instrument at the NSF Regional Facility, University of Nebraska, Lincoln, NE.

Isolation of Combretastatins D-1 (1) and D-2 (2). Fraction A (28.6 g),^{2a-c} obtained after extraction of Combretum caffrum (77 kg) stem wood, was further separated on a column of Sephadex LH-20 (2.5 kg) by partition chromatography using hexane-toluene-methanol (3:1:1) to afford two active fractions (1.97 g, PS ED₅₀ 1.8 × 10⁻² μ g/mL, and 0.54 g, PS ED₅₀ 1.9 μ g/mL). The latter fraction (0.54 g) was chromatographed on a silica gel (0.04-0.063 μ m) flash column (3.0 × 20.0 cm). The column was packed and eluted with hexane-chloroform-acetone (3:2:0.25) to give combretastatin D-1 (1, 180 mg, (2.3 × 10⁻⁴)% yield). For the physical data, consult ref 3.

The fraction weighing 1.97 g was dissolved in hexane-toluene-methanol (3:1:1, 20 mL) and the solution was filtered. The filtrate was chromatographed on a Sephadex LH-20 (200 g) column using the same solvent system. The resulting active fraction (1.35 g, PS ED₅₀ 2.4 × $10^{-2} \mu g/mL$) was dissolved in hexane-ethyl acetate (1:1, 5 mL) and chromatographed on a column (60×2.5 cm) of silica gel (60 g). Gradient elution from $4:1 \rightarrow 1:1$ hexane-ethyl acetate afforded in a 3:1 fraction the PS-active (0.7 g, ED_{50} 1.0 × 10⁻² µg/mL) material. Rechromatography in acetone (2 mL) over a long silica gel column $(100 \times 1.2 \text{ cm}, 45 \text{ g})$ and gradient elution with hexane-ethyl acetate (9:1 \rightarrow 4:1) furnished in the 4:1 fraction pure combretastatin D-2 (2, 5.8 mg, (7.5 \times 10⁻⁶)% yield based on dried plant material), needles from acetone-hexane: mp 148-51 °C; PS ED₅₀ 5.2 μ g/mL; UV λ max (nm) 235 (¢ 7300), 274 (2260), 339 (1050); IR (NaCl) vmax 3436, 3429, 1728, 1519, 1503, 1440, 1215, 1186, 1159, 1110 cm⁻¹; HREIMS m/z 296.1052 (M⁺, 100, calcd for $C_{18}H_{16}O_4$ 296.1049), 237.0916 (20, calcd for C₁₆H₁₃O₂ 237.0916), 180.0426 (5, calcd for C₉H₈O₄ 180.0423), 138.0321 (46, calcd for C7H6O3 138.0317), 135.0450 (50 caled for C₈H₇O₂ 135.0446), 116.0620 (30, caled for C₉H₈ 116.0626), 91.0545 (35, calcd for C_7H_7 91.0548); for ¹H and ¹³C NMR data see Tables I and II, respectively.

Benzyl Bond Hydrogenolysis of Combretastatin D-1 (1 \rightarrow 5b). Method A. To a solution of combretastatin D-1 (1, 10 mg) in a mixture of ethyl acetate-methanol (5:3, 10 mL) was added 5% Pd/C (10 mg). The mixture was hydrogenated under ambient temperature and pressure for 72 h. Catalyst was removed (filtration) and the filtrate was concentrated to give pure alcohol 5b (10 mg, quantitative yield) as needles from ethyl acetate-hexane: mp 191-93 °C; $[\alpha]^{30}_D - 12.6^\circ$ (c 0.95, CHCl₃/CH₃OH, 1:1); IR (KBr) ν_{max} 3200, 1740, 1719, 1521, 1504, 1285, 1220, 1163, 1155, 1142, 1103 cm⁻¹; HREIMS m/z 314.1153 (M⁺, 100, calcd for C₁₈H₁₈O₅ 314.1154), 271.0966 (69, calcd for C₁₆H₁₅O₄ 271.0970), 226.0994); for ¹H NMR data see Table I.

Method B. Combretastatin D-1 (5.0 mg) in ethanol (2 mL) was treated with freshly prepared⁷ zinc/copper couple (100 mg) for 10 days. The solution was filtered and the filtrate concentrated to give a mixture of unreacted starting material and two products. Separation on a preparative silica gel plate using hexane-acetone (7:3) as solvent afforded the less polar hydrocarbon product (5a, 0.8 mg) as a viscous oil (for ¹H NMR data see Table I): HREIMS m/z 298 (M⁺, 6), 135 (10), 115 (60), 107 (65), 91 (100). Unreacted combretastatin D-1 (1.0 mg) was recovered, and the most polar product (20 mg) was identified as alcohol 5b by direct comparison (TLC, ¹H NMR) with the product of method A.

Chlorination of Alcohol 5b. To a cooled (0 °C) solution of alcohol 5b (1.0 mg) in pyridine (0.2 mL) was added thionyl chloride (0.1 mL), and the solution was stirred for 1 h at 0 °C and overnight at room temperature. The solvent was evaporated under a stream of nitrogen and the residue was chromatographed by using a pipet filled with silica gel. Elution of the pipet column with hexane-acetone (3:1) gave 3-chloro-3,4-deoxycombretastatin D-1 (5c, 1.0 mg) as a amorphous powder from acetone-hexane: mp 170–172 °C; IR (NaCl) $\nu_{\rm max}$ 3450, 1740, 1597, 1520, 1506, 1205 cm⁻¹; HREIMS m/z 332.0812 (M⁺, 100, calcd for C₁₈H₁₇O₄³⁶Cl 332.0816), 334.0798 (31, calcd for C₁₈H₁₇O₄³⁷Cl 334.0786), 297.1132 (29, calcd for C₁₈H₁₇O₄ 297.1127); for ¹H NMR data see Table I.

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ANTINEOPLASTIC AGENTS, 178. ISOLATION AND STRUCTURE OF LYCHNOSTATINS 1 AND 2 FROM THE SOUTH AMERICAN LYCHNOPHORA ANTILLANA¹

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ABSTRACT.—Bioassay-guided (P388 lymphocytic leukemia cell line) separation of a $CH_2Cl_2/MeOH$ extract of Lychnophora antillana led to the isolation of two cytostatic (P-388, ED_{50} 2.0 and 0.19 µg/ml, respectively) germacranolides designated lychnostatins 1 [1] and 2 [2]. Structural elucidation was based initially upon high field (400 MHz) nmr and electron impact mass spectral interpretations and unequivocally completed by X-ray crystal structure determinations.

Although many species of the large plant family Compositae are well-known for a variety of reasons, including primitive medical applications, some occur in a few small and relatively unexplored tropical genera. One such genus, the *Lychnophora* of the sub-tribe Lychnophorinae (2,3), contain some twenty-three species indigenous primarily to Brazil. More than twenty years ago, as part of the U.S. National Cancer Institute's (NCI) world-wide exploratory programs directed (by Jonathan L. Hartwell) toward the discovery of new anticancer drugs, specimens of *Lychnophora antillana* Urb. (also known as *Piptocoma antillana*) were collected and evaluated. By 1974, an EtOH extract was found to provide 32–34% life extension against the NCI murine P-388 lymphocytic leukemia (PS system) at 4.9 \mapsto 16 mg/injection. A 1979 Puerto Rican collection gave analogous biological results (including PS cell line ED₅₀ 0.72 µg/ml) and led to the present study.

The plant was extracted with CH₂Cl₂-MeOH (1:1), and the extract was partitioned (4) between MeOH-H₂O (9:1 \mapsto 4:1 \mapsto 3:2) with hexane \mapsto CCl₄ \mapsto CH₂Cl₂ to yield an active CH₂Cl₂-soluble fraction (PS ED₅₀ 0. 15 µg/ml). Separation (PS bioassay-guided) of this fraction on a Si gel column resulted in isolation of lychnostatins 1 [1] and 2 [2] as the major PS-active (ED₅₀ 2.0 and 0. 19 µg/ml) constituents.

Initial structural investigations revealed both cytostatic compounds to be new sesquiterpene lactones of the germacranolide type. While varied biological activity has been reported for a number of such compounds from other genera (5-18), only one example of antineoplastic activity (10) has been reported for germacranolides isolated from the Lychnophora (19-24). Several germacranolides distantly related to lychnostatins 1 and 2 have been isolated from Brazilian Lychnophora (19), Eremanthus (25), and



¹For Part 177 see Pettit and Schaufelbetger (1). ²Deceased March 25, 1981.

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Piptolepis (21, 26) species. One of these, isolated from *Lychnophora blanchetii* (19), was assigned structure **3** a structural isomer of lychnostatin 1.

Ir, ¹H-, ¹³C-nmr, and mass spectral analyses suggested the presence of an α methylene lactone, as well as methacrylate, acetate, and ketone groups. From mass spectral data, it was determined that lychnostatin 1 [1] differed from hychnostatin 2 [2] only by having an additional oxygen atom. In addition, eims exhibited significant peaks corresponding to {M = HOAc}⁺ and {M = HOAc = CH₂ = C(CH₃)CO₂H}⁺ fragment ions, thereby confirming the presence of the ester groups. The spectral data and molecular formula were also consistent with a ten-membered curbon ring bearing the substituents just noted. Extensive ¹H-nmr and ¹³C-nmr decoupling experiments provided sufficient additional information to allow assignment of the α -methacrylate unit adjacent to the lactone. The nmr data also seemed to suggest that a hydroxyl group in lychnostatin 1 was adjacent to the lactone ring. From empirical formula data, the presence of macrocyclic ring unsaturation seemed to be excluded for both compounds. Because neither the complete regio nor stereo relationships of the macrocyclic ring substituents could be definitively ascertained from the above information alone, a number of structural possibilities remained.

In order to establish unambiguously the complete structures of lychnostatins 1 and 2, single crystal X-ray diffraction analyses were undertaken (Table 1). Cell parameters for both lychnostatins were nearly identical, suggesting that each of the compounds had similar cell packing characteristics and conformations. Indeed, this assumption proved to be correct. An X-ray-analysis-derived structure for lychnostatin 1 is shown in Figure 1. The absence of unsaturation in the 10-membered macrocyclic rings for both lychnostatins was thereby established. Although unusual, this result was not without precedent (19,25,27,28). Also established were the orientation of the macrocyclic ring and the relative stereochemistry of the ring substituents for both compounds. The β disposition of the C-4 and C-7 substituents, as well as the C-10 methyl, was readily apparent for the lychnostatins.

For lychnostatin 1, the additional oxygen atom was found to be present as a β oriented C-5 hydroxy group. The two ester substituents attached to the C-8 and C-10 ring atoms of both compounds, as well as the C-6 oxygen atom (which forms part of the *trans*-fused α -methylene lactone ring) were all α -oriented with respect to the 10-membered ring. The more stable *trans*-fusion of the lactone ring to the 10-membered ring is a feature commonly observed for a majority of germacranolide sesquiterpene lactones. The α -methylene– γ -lactone rings of both lychnostatins 1 and 2 exhibited some nonplanarity (endocyclic torsion angle moduli sum of 49° and 55°, respectively). Examples





5 2,3-trans

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Parameters	Compound			
- analice of a	1	2		
Crystal data Molecular formula	С,,,Н,,,О,	C,1H,4O7		
F.W	408.45	392.45		
F(000)	872	840		
Space group	$P2_{1}2_{1}2_{1}$	$P2_{1}2_{1}2_{1}$		
Crystal dimensions (mm)	$0.07 \times 0.10 \times 0.32$	$0.08 \times 0.10 \times 0.45$		
Radiation, A	$CuK\alpha, \lambda = 1.54184$	$CuK\alpha, \lambda = 1.54184$		
Temperature, °C	26 ± 1	26 ± 1		
Cell constants		5 705 (4)		
<i>a</i> , A	5.8/6(2)	5.785(4) 8.003(3)		
b , \mathbf{A}	8.807(1) (0.057(0)	8.902(5)		
$\mathbf{v} \mathbf{A}^{\mathbf{x}}$	2080 3	2093.4		
ν, Λ	2009.1	2003.4 .1		
op g/cm ⁴	1 780	1 245		
ρ_{c} g/cm ⁴	1.202	1.251		
u cm	7.9	7 4		
Collection Parameters				
Instrument	Enraf-Nonius	Enraf-Nonius		
Monochromator	CAD4 diffractometer Graphite crystal, incident	CAD4 Graphite crystal, incident		
Attenuaror	beam Ni foil tactor 11.9	beam Ni foil factor 11.9		
Take-off angle deg	2.8	2.8		
Detector aperature, mm	4.0 to 5.9 horizontal.	4.0 to 5.9 horizontal,		
	4.0 vertical	4.0 vertical		
Crystal-detector dist	21 cm	21 cm		
Scan type	ω-2θ	ω-2θ		
Scan rate, ^{°/} min (in ω)	1 to 5	1 to 5		
Scan width, deg	0.9 + 0.140 tan θ	0.8 ± 0.140 tan θ		
Maximum 20, deg	150.0	150.0		
No. of refl. measured	2676 total, 2537 unique	2602, 2532 unique		
Corrections made	Lorentz-polarization Linear decay, (0.815 to 1.185 on I)	Lorentz-polarization Linear decay, (0.985 to 1.129 on I) Empirical absorption,		
		(0.88 to 0.99 on I)		
Parameters				
Solution method	Direct Methods	Direct Methods		
Hydrogen atoms	Refined, Uiso = 0.06 Å ² , restrained to ride	Refined, Uiso = 0.06 Å ² , restrained to ride		
Refinement	Full matrix least-squares	Full matrix least-squares		
Minimization function	$\Sigma w(Fo - Fc)^2$	Σw(iFol-iFci) ⁺		
Least-squares weights	$1/\sigma^{-}$ (Fo)	1/σ ⁻ (Fo)		
Anomalous dispersion	All non-hydrogen atoms	All non-hydrogen atoms		
Reflections included	21/8 with Fo ⁺ > 3.007(Fo ⁺)	$1297 \text{ with ro}^2 > 5.00(\text{ro})$		
Parameters renned	262	204		
Wainbend P factor	0.045	0.049		
EDS of obs. of upit art	2.30	2 74		
Convergence.	21.00			
Largest shift. A	0.07	0.06		
High peak in final diff.				
map, e/Å	0.20(4)	0.20(5)		
Computer hardware	PDP-11/23, MicroVax II			
Computer software	SDP-PLUS (Enraf-Nonius &	_		
	B. A. Frenz and Assoc., Inc.) CRYSTALS (CCL, Univ. of Oxford)			

TABLE 1. Crystal Data Experimental and Refinement Parameters for Lychnostatin 1 [1] and Lychnostatin 2 [2].

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FIGURE 1. Crystal structure (ORTEP representation) of lychnostatin 1.

covering the range of nearly complete planarity to pronounced non-planarity of the trans-fused lactone ring have been reported (10, 27, 29-32).

No abnormalities were observed for the bond distances and angles of either compound, the values being in good agreement with those reported for similar substances (27, 29, 30, 33-35). Conformational analysis of the data for lychnostatins 1 and 2. with respect to the 10-membered macrocyclic ring, revealed a conformational deviation from that previously proposed and/or observed for cyclodecane/cyclodecanone rings (36-39). Among these are two boat-boat conformations, referred to as the "O-inside" and the "O-outside" conformations, as depicted in Figure 2 In the "O-inside" conformation, the oxygen is approximately perpendicular to the imagined plane of the cyclodecane ring, whereas in the "O-outside" conformation, the oxygen lies approximately in the plane of the ring. The "O-inside" is conformationally favored over the "O-outside," primarily due to the decreased number of destabilizing intra-annular hydrogen atom interactions present in this conformation (36). The conformation assumed by lychnostatins 1 and 2 is depicted in Figure 3 as a twist chair-boat conformation. Although subtly different from the boat-boat "O-inside" conformer, it still maintains one essential distinguishing feature of that conformer, i.e., positioning of the carbonyl oxygen in a perpendicular orientation to the plane of the 10-membered ring.

With lychnostatins 1 and 2, significant intra-annular hydrogen interactions (interatomic bond distance $< 2 \times H$ van der Waals radii or ca. 2.30 Å) occur on both the α and β faces, as signified by arrows in Figure 3. Table 2 summarizes the intra-annular interatomic distances occurring in the lychnostatins. In each case the carbonyl oxygen, O-1, does not seem to participate in any significant intra-annular interactions. All intraannular atomic distances involving O-1 were found to be 2.40 Å or greater. On the other hand, a greater number of hydrogen-hydrogen intra-annular interactions occur in the conformer adopted by lychnostatins 1 and 2, as compared to the "normal" boat-boat

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0-outside

FIGURE 2. Two possible cyclodecanone conformers.

O-inside conformer. Presumably, the *trans*-fusion of the α -methylene– γ -lactone ring to the 6,7 position of the cyclodecanone ring, as well as the α orientation of the ester side chains, provides steric factors contributing to this conformational modification or deviation.

The absolute configuration of lychnostatins 1 and 2 could not be ascertained from the X-ray data, only the relative configuration; nearly identical R values were obtained for both enantiomers. Thus, either the structures depicted by 1 and 2 or their mirror images are equally plausible. A less reliable method (27, 29–31, 40, 41) for affixing absolute stereochemistry about the ring juncture of the α -methylene– γ -lactone and the cyclodecanone ring, based upon cd data, also failed due to interference by the methacrylate moiety with the diagnostic $n \rightarrow \pi^*$ transition curve of the lactone. Finally, utilization of information based solely on the possible biosynthetic pathway previously proposed for the generation of germacranolides must also be excluded, as there are no carbon-carbon double bonds in the 10-membered ring that might indicate its mode of origin. The aforementioned problems concerning absolute configurational assignments and correct classification (12 \rightarrow 6 or 12 \rightarrow 8 lactonization) have been encountered earlier (27,29).

As previously mentioned, Bohlmann *et al.* (19) have investigated *L. hlanchetii* collected in northeast Brazil and assigned structures 3, 4, and 5 to three of the con-



FIGURE 3. Conformation assumed by the lychnostatins

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Atoms	Lychnostatin i		Lychnostatin 2	
	α-face	β-face	α-face	β-face
Н-э-Н-б	2.09 A		2.10 A	•
H-3-H-11	2.36 A		2.43 A	
H-3-H- ⁺	>3.00 Å		>3.00 Á	1
H-+-H-9		>3.00 A		23.00 A
H-4-H-14		2.12 A		2.13 A
H-+		2.54 A		2.58 Å
H-6-H-	2.31 Å		2.26 A	
H-6-H-11	2.30 Å		2.29 A	1
H-7-H-11	2.19 Å		2.20 A	1
H-9-H-14		2.07 Å		2.07 A
H-9-O-1		2.50 A		2.49 A
H-14-O-1		2.40 A		2.40 A

TABLE 2. Intra-annular Atomic Distances in Evchnostatin 1 [1] and Evchnostatin 2 [2] (<3.00 A).

stituents. In addition to these germacranolides, they also identified the two pentacyclic triterpenes, lupeol and lupenone. In the present study, we found betulinic and ursolic acids as representatives of the latter group. More importantly, germacranolide 3 appears to be a structural isomer of lychnostatin 1. From biosynthetic considerations it seems likely that structure 3 may need further refinement.

Lychnostatins 1 [1] and 2 [2] now augment the small number of germacranolides known to exhibit cell growth inhibitory and/or antineoplastic activity (6, 10, 12, 13, 15–18). The lychnostatins are also unusual in that they don't completely satisfy the postulate proposed by Manchand and Blount (17) that antitumor activity requires, in addition to the α -methylene- γ -lactone, an oxygen function or double bond at C-4. Further experiments directed at biological evaluation and unambiguously defining absolute configuration of the lychnostatins are under way.

EXPERIMENTAL

Solvents used for chromatography were redistilled. Ambient cc procedures employed Si gel (70-230 mesh), supplied by E. Merck, Darmstadt. Cc under pressure was carried out using prepacked Lobar Lichro Prep Si gel 60 (40-63 um). Fraction collection was partially automated, using a Gilson microfractionator. The was performed with Si gel GHLF from Analtech. Inc. The tle plates were developed by uv light and/or a ceric sulfate spray reagent.

All melting points are uncorrected and were observed using a Koeffler-type melting point apparatus. Each substance was colorless. It spectra were recorded with a Perkin-Elmer Model 299 spectrophotometer. Optical rotations were determined with a Perkin-Elmer model 241 Automatic Polarimeter. The 100 MHz ¹H-nmr spectra were recorded with a Varian XL-100 instrument and the 400 MHz with a Bruker WH-400 nmr spectrophotometer. The ¹³C spectra were obtained employing a Bruker WH-90 at 22.63 MHz. TMS was used as an internal reference, and δ values are reported in ppm. Mass spectra were obtained using an MAT 312 spectrophotometer.

PLANT MATERIAL.—*L. antillana* (stems and leaves, herbarium specimens NSC B 50°14 maintained by the USDA) was recollected in Puerto Rico in 1979, under auspices of the Economic Botany Laboratory. Agricultural Research Center East, USDA, Beltsville, Maryland, as part of a joint NCI-USDA program directed by Drs. M.I. Suffness and J.A. Duke.

EXTRACTION AND SOLVENT PARTITIONING.—The stems and leaves of L. antillana (54 kg) were extracted with CH₂Cl₂-MeOH (1:1) (320 liters) at ambient temperature for 4 days. Decantation of solvent and subsequent dilution with H₂O (25% by volume) allowed the chlorocarbon phase to separate. The CH₃Cl₂ was removed to yield a viscous, brown gum (1145 g) which was further purified by partitioning (4) employing the sequence MeOH-H₂O (9: 1) \leftrightarrow 4: 1 \leftrightarrow 3:2) against, respectively, hexane \rightarrow CCl₄ \rightarrow CH₂Cl₂. An aliquot (40 g) of the CH₂Cl₂ fraction (total weight, 315 g) was chromatographed on a column of Si gel (800 g). Elution with hexane-Me₂CO (9: 1) (6.0 liters to 10.0 liters) yielded fraction A (0.18 g) as a colorless solid. Further elution (13.0 to 18.0 liters) resulted in isolation of lychnostatin 1 [1] as a colorless solid (0.31 g, 5.7×10^{-5} % yield).

Rechromatography of fraction A (see above) on a column of Si gel (18 g, dry packed column) and elution with CH₂Cl₂ (200 ml) atforded a single product, lychnostatin 2 [2] (20 mg, 3.7×10^{-67} yield). Further elution with CH₂Cl₂-MeOH (44:1) gave a mixture (0, 10 g) which was rechromatographed on Si gel (Lobar B column). Gradient elution with CH₂Cl₂-MeOH (99:1 \mapsto 44:1) (400 ml total) led to a pure compound (34 mg) that recrystallized from MeOH/CH₂Cl₂ to give berulinic acid (20 mg), mp 304–307°. Further elution with the same solvents yielded another minor product (22 mg), which proved to be ursolic acid, mp 260–265°. Both triterpene carboxylic acids were identified by comparison (tlc, ir, ¹H nmr) with authentic specimens.

LYCHNOSTATIN 1 [1].—Recrystallization from Me₂CO/hexane afforded crystals melting at 228–230°. tlc R, 0.85 in CHCl₄-MeOH (9:1); $[\alpha]^{24}$ tb +89° (z = 1.0, CHCl₄); ir (KBr) ν max 3580, 3430, 1780, 1732, 1710, 1702 (sh), 1660, 1633, 1460, 1380, 1308, 1276, 1267, 1253, 1154, 1120, 1095, 1073, 1060, 1020, 993, 960, 816, 805, 763, 700, 600 cm⁻¹; ¹H nmr (400 MHz, CDCl₄) δ 1.08 (d, 3H, J = 8 Hz, Me-15), 1.62 (1H, br s, -OH), 1.74 (3H, s, Me-14), 1.94 (3H, s, Me-3'), 1.70–2.10 (3H, m, -CH₂-), 2.03 (3H, s, Me-2"), 2.32 (1H, dd, J = 16, 2 Hz, -CH₂), 2.66–2.74 (3H, m, -CH-), 3.0° (1H, d, J = 10 Hz, H-7), 3.41 (1H, dd, J = 8, 7 Hz, H-5), 4.35 (1H, d, J = 8 Hz, H-6), 4.84 (1H, m, H-8), 5.65 (2H, br s, H-13a or H-13b), H-4'a or H-4'b), 6.16 (1H, br s, H-4'a or H-4'b), 6.24 (1H, br s, H-15a or H-13b); ¹¹°C nmr (22.63 MHz, CDCl₄), 207.61 (s, C-1), 169.56 (s, C-12), 168.45 (s, C-1"), 165.82 (s, C-1'), 135.93 (s, C-11 or C-2'), 134.92 (s, C-11 or C-2'), 126.31 [t (2C), C-13 and C-4'], 84.91 (s, C-10), 82.44 (d, C-6), 77.50 (d, C-5), 70.7° (d, C-8), 44.29 (d, C-7), 41.07 (t, C-2 or C-9), 35.32 (t, C-2 or -9), 32.07 (d, C-4), 24.05 (t, C-3), 22.42 (q, C-14), 21.38 (q, C-2"), 20.24 (q, C-3'), 18.23 (q, C-15); eims m/z [M]* 408, [M = HOAc]* 348. [M = H₂O = C₄H₅O₂]* 305, [M = HOAc = C₄H₅O₂]* 245. Anal. calcd for C₂₁H₂₈O₈, C 61.75, H 6.91; found C 61.66, H 6.67%.

LYCHNOSTATIN 2 [2].—Recrystallization of lychnostatin 2 [2] from Me₂CO/hexane provided fine needles: mp 190–193°; [α]³⁰D +20.9° (α = 0.67, CHCl₄); ir (KBr) ν max 2950, 1780, 1740, 1712, 1645, 1460, 1385, 1312, 1300, 1275, 1178, 1156, 1126, 1105, 1065, 1022, 955, 878, 810, 741, 614 cm⁻¹. ¹H nmr (100 MHz, CDCl₄) **b** 1.04 (3H, d, J = 6 Hz, Me-15), 1.80 (3H, s, Me-14), 1.96 (3H, s, Me-3'), 1.4–2.2 (4H, m, -CH₂-), 2.06 (3H, s, Me-2''), 2.23 (1H, dd, J = 15, 2 Hz, -CH₂-), 2.70 (3H, m, -CH-), 3.05 (2H, m), 4.37 (1H, m, H-6), 4.96 (1H, dd, J = 8, 2 Hz, H-5), 5.73 (2H, d, J = 2 Hz, H-13a or H-13b, H-4'a or H-4'b), 6.18 (1H, br s, H-4'a or H-4'b), 6.35 (1H, br s, H-13a or H-13b); ¹¹C-nmr (22.63 MHz, CDCl₄) **b** 208.29 (s, C-1), 169.62 (s, C-12), 168.97 (s, C-1''), 165.92 (s, C-1'), 135.86 (s, C-11 or C-2'), 134.69 (s, C-11 or C-2'), 126.53 (t, C-13 or C-4'), 124.88 (t, C-13 or C-4'), 84.26 (s, C-10), 77.86 (d, C-6), 68.24 (d, C-8), 46.99 (d, C-7), 43.44 (t, C-5), 30.09 (t, C-2 or C-9), 35.91 (t, C-2 or C-9), 29.70 (d, C-4), 27.13 (t, C-3), 23.98 (q, C-14), 21.51 (q, C-2''), 21.25 (q, C-3'), 18.20 (q, C-15); spsims *m/z* [M + Na]⁺ 415; eims *m/z* [M]⁺ 392, [M = CH₂CO]⁺ 350, [M = HOAc]⁺ 332, [M = C₅H₅O₁]⁺ 263, [M = HOAc = C₄H₆O₃]⁺ 246, hrfabms *m/z* [M + ⁻Li]⁺ 399.1996 (calcd C₂₁H₂₅O₇ + ⁻Li, 399.195° 2).

X-RAY CRYSTALST. ICTURE DETERMINATIONS OF LYCHNOSTATIN 1 [1] AND LYCHNOSTATIN 2 [2].—Preliminary examinations and data collections for lychnostatins 1 and 2 were performed at room temperature by the moving-crystal, moving-counter technique with background measurements made on both sides of the peak using an Enraf-Nonius CAD-4 automatic diffractometer. Crystal data, collection, and refinement parameters for the two compounds are summarized in Table 1. In each case, data was corrected for Lorentz and polarization effects. For lychnostatin 2, an additional semi-empirical absorption correction was also applied [the absorption correction being based on a scries of psi scans (42)]. Space group assignments for each compound were derived on the basis of Laue symmetry and observed systematic extinctions. Cell dimensions were determined from least-squares refinement, using the setting angles of 25 carefully measured reflections. The structures were solved by direct methods (43). Scattering factors were taken from Cromer and Waber (44). Initial stages of refinement were performed using the SDP-PLUS (45) software package; final refinements were done with CRYSTALS (46). Anomalous dispersion corrections were made in Fc (47) for both compounds; the values of $\Delta F'$ and $\Delta F''$ were those of Cromer (48); extinction coefficients were refined on both compounds.⁴ A perspective view (49) displaying all essential conformational and configurational features for lychnostatins 1 and 2 appears in Figure 1.

Colorless crystals of lychnostatin 1, arising from MeOH-H₂O solution, were used in mass spectral,

Atomic coordinates for these structures have been deposited with the Cambridge Crystallographic Data Centre and can be obtained on request from Dr. Olga Kennard, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1 EW, UK.

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density and X-ray data collections. The observed density and mass spectral data indicated a single molecule of lychnostatin 1 per asymmetric unit in space group $P_{2_1} P_{1_2} P_{1_2}$. All unique reflections (one octant) were collected. Structural solution proceeded without incident, the nonhydrogen atoms being located readily. The remaining hydrogen atom coordinates were calculated at ideal positions and assigned fixed coordinates and isothermal parameters during subsequent structure-factor full-matrix least-squares refinements. Here the function minimized for least-squares was $\Sigma w (|F| - |Fc|)^2$ with the weight w defined as $1/\sigma^2(FO)$. Refinement was continued until convergence to a residual of R = 0.045 and $R_w = 0.044$.

The crystal structure of lychnostatin 2 was performed on a fine needle-shaped crystal obtained from Me₂CO/heptane solution. Observed density measurements again indicated one molecule per asymmetric unit corresponding to the $P2_12_12_1$ space group. Solution by direct methods proceeded with some difficulty. After a number of unsuccessful preliminary attempts, a starting set of seven reflections was used (from 400 reflections) with the largest E's (minimum E of 1.29) in order to generate 12,659 relationships. In addition, the lower limit of probability of acceptance of phases determined by the sigma 1 formula being included in the starting set was extremely low (i.e., 0.650). In this manner, a total of 200 possible phase sets were generated; the phase set with the highest overall figure of merit (2.99) provided an E map which revealed all 28 nonhydrogen atoms. Hydrogen atom coordinates again were calculated, fixed, and assigned isotropic thermal parameters in subsequent least-squares methods. Refinement converged to a residual of R = 0.049 and $R_w = 0.038$.

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ANTINEOPLASTIC AGENTS, 107. ISOLATION OF ACTEOSIDE AND ISOACTEOSIDE FROM CASTILLEJA LINARIAEFOLIA¹

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ABSTRACT.—The southwestern Indian paintbrush, *Castilleja linariaefolia*, yielded extracts that displayed in vivo activity against murine P-388 (PS) lymphocytic leukemia. Separation guided by PS cell line inhibition led to isolation of cytotoxic compounds that were identified as the known glycosides acteoside [1] (ED₅₀ 2.6 μ g/ml) and isoacteoside [2] (ED₅₀ 10 μ g/ml). The identifications were established by spectral measurements and degradation studies. Mannitol was also found in this plant.

In mountainous areas of northern Arizona (2) and southern Utah, Castilleja linariaefolia Benth. is known as Indian paintbrush. While some 50 species of the large (3000 species and 220 genera) Scrophulariaceae family have been used in primitive cancer treatment, only one has been represented by the Castilleja genus (3), namely, the Mexican (Yucatan) Castilleja communis Benth. (4).

Extracts from C. linariaefolia gave confirmed activity against the Walker carcinosarcoma 256 (intramuscular WM system) in the rat. Each of the major plant parts appeared to contain the anticancer constituent(s), with the flowers showing the highest activity (90% inhibition of tumor growth at 266 mg/ kg). When the murine P-388 lymphocytic leukemia (PS system) became available and a key fraction showed T/C 125-165% (10-40 mg/kg), we discontinued use of the WM system. [The PS in vitro studies were conducted in our laboratory according to procedures developed by the National Cancer Institute, and PS in vivo bioassays were performed under the auspices of the NCI (5)] Inconsistent results were obtained with both in vivo systems making fractionation difficult. Monitoring fractionation of C. linariaefolia with in vitro PS

eventually yielded two of the PS cytostatic constituents. These were found to be the known glycosides acteoside (6-9) and isoacteoside (8). These caffeoyl glycosides have been isolated from a Labiatae species (8), and acteoside also occurs in a Gesneriaceae (7) and two Oleaceae (6,9) species; neither glycoside had hitherto been found in a plant of the Scrophulariaceae. The research was completed using a series of recently developed (10) experimental procedures augmented by dccc. The dccc technique has previously been used in the separation of iridoid glycosides from Castilleja miniata (11). Interestingly, dccc was the only effective method found for separation of myricoside, a bioactive substance closely resembling accesside in structure (12). As part of the current study, Dmannitol also was isolated.

Acteoside [1] was identified on the basis of detailed spectral analyses (uv, ir, etc.) and by identification of alkaline and acid hydrolysis products and confirmed by comparison with an authentic sample provided by Professor I. Nishioka, Faculty of Pharmaceutical Sciences, Kyushu University. The general spectral features of isoacteoside [2] closely resembled those of accesside, except that the Lrhamnose unit methyl group signal in the ¹H-nmr spectrum was shifted from δ 1.12 to 1.27 ppm. Also, the C-6 and C-3 D-glucose unit carbon signals in the ¹³C-nmr spectrum were shifted from δ 62,43 to 64.70 and from 81.64 to 84.15

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ppm, respectively. These data suggested an isomeric relationship, which was confirmed by a series of methylation, acetylation, and hydrolysis experiments (11, 13, 14).

Acteoside and isoacteoside exhibited moderate to weak cytotoxic activity in the PS in vitro system (ED₅₀ 2.6 and 10 $\mu g/ml$, respectively). Because of the antibacterial (12) and cAMP phosphodiesterase inhibitory (15) activity shown by several closely related natural products, these glycosides are worthy of further biological study.

EXPER!MENTAL

GENERAL EXPERIMENTAL PROCEDURES.-Paper partition chromatography (ppc) was performed by the ascending method using Toyo Roshi No. 50 paper and n-BuOH-HOAc-HO (4.1.2). Mp s were determined on a Yanagimoto micro melting point apparatus and are uncorrected. Ir were recorded with an Hitachi EPI-G2 spectrometer. The ¹H-nmr spectra were measured with an Hitachi R 40 spectrometer at 90 MHz and the ''C-nmr spectra with a JEOL JNM FX-200 spectrometer at 50.3 MHz. Chemical shifts are given in ppm (δ) downfield from TMS as an internal standard. Solution phase sims (16) mass spectra were obtained using a Varian MAT 312 spectrometer equipped with a modified capillaritron source and 0.14 M Nal in sulfolane as liquid phase. The eims were recorded using an Hitachi M-70 spectrometer.

PLANT MATERIAL.—The recollection (original in August 1966, Kaibab, N.F., Arizona) of *C. linariaefolia* (ca. 500 kg dry wt), used here was made in August 1978, in the Dixie National Forest, Garfield Co., Utah, at an elevation of "000-8000 fr. Taxonomic identification was made by one of us (CPP) in the USDA Laboratory and by E. Lehto in the Department of Botany at Arizona State University. Herbarium specimens are maintained in the Department of Botany and in the Cancer Research Institute at Arizona State University.

PLANT EXTRACTION. - Dried plant material (leaves, stems, and roots, 36 kg) from the 1978 collection was extracted with a mixture (160 liters) of CH₃Cl₃ and MeOH (1:1) (10) at ambient temperature for 3 weeks. The extract was separated into CH₃Cl₃ and H₃O phases on addition of 25% by volume of H₃O. The aqueous phase was adjusted by the addition of MeOH and CH 3CI 3 to achieve a 4:1:2 ratio for H3O-MeOH-CH3Cl3, the plant was extracted with this mixture for weeks. Addition of 15% by volume of H₃O resulted in separation of the CH2Cl2 phase which was combined with that obtained from the first partition. Concentration gave a CH-Cl- extract (712 g; PS in vitro ED₅₀ 19 µg/ml; PS in vivo inactive at 12.5-100 mg/kg). The H₂O phase was concentrated to give an extract (3.5 kg) that was marginally active (in vivo T/C 120% at 25 mg/ kg; PS in vitro $ED_{50} \ge 100 \ \mu g/ml$).

SOLVENT PARTITION SEQUENCE. —A portion of the rf₂O extract (180 g) was successively partitioned between MeOH-H₂O (9:1) (1800 ml)→(4:1)→(1:1) with hexane (5×1800 ml), CCl₁ (3×1800 ml), and CH₂Cl₂ (3×1800 ml), respectively. Concentration of the partitioned fractions gave hexane (1.5 g; PS in vitro ED₅₀ $24 \mu g$ /ml), CCl₂ (4.6 g; PS in vitro ED₅₀ $251 \mu g$ /ml), and H₂O (142.8 g; PS in vitro ED₅₀ $36 \mu g$ /ml) fractions. None of these fractions exhibited activity in the PS in vivo system when tested at dose levels of 3.12-25 mg/kg. However, in a number of earlier experiments, fractions had been obtained at this stage showing T/C 165 at 40 mg/kg.

ISOLATION OF ACTEOSIDE [1] AND ISOAC-TEOSIDE [2].—An aliquot (10.22 g) of the H₂O fraction was chromatographed on Sephadex LH-20 (805 g; 77 × 8 cm) using MeOH-H₂O (4:1) as eluent. Fractions were monitored by ppc. After elution of 5.1 g of material, a fraction (2 g) was obtained from which D-mannitol (0.252 g) was isolated as colorless needles, mp 171-172°, and found to be identical (ir, tlc) with an authentic sample. Further elution gave 1.8 g of material (inactive against PS in vitro), followed by a fraction (1.3 g) that was active in vitro (ED₅₀-4, 1 µg/ ml, PS in vivo inactive at 3.12-25 mg/kg). A 1.94-g portion of the in-vitro-active fraction (obtained after repeating the Sephadex LH-20 chromatographic step) was treated with MEOH. The soluble fraction (1.38 g) was dissolved in a minimum volume of the upper layer prepared from CHCl₃-MeOH-H₃O (5:5.7:3) and placed in the transfer tube of a dccc apparatus filled with the upper layer of the same solvent system as stationary phase. The flow rate of the moving lower phase was 0.66 ml/min. Three fractions were collected on the basis of monitoring by ppc.

Acteoside [1] (0.388 g) was isolated as a pale yellow amorphous powder from the third fraction. The first fraction (0, 174 g) was again subjected to dccc to give isoacteoside [2] (0, 107 g) as a light brown amorphous powder. Acteoside [1] exhibited the following physical properties: mp 145-149° [lit. (6) mp 147-150°]; solution phase sims m/z [M + H] ^{*} 625 (21%), 480 (10%), 472 (24%), $\{M + H = caffeovl-3, 4-dihydroxyphen$ ethyl]⁺ 325 (100%). The identity was confirmed by direct comparison of the 'H- and 'C-nmr spectra with those of an authentic sample. Treatment of acteoside with Ac2O/pyridine followed by Si gel cc gave the peracetate as a colorless amorphous powder: uv (MeOH) λ max 283 (log \in 4.19) nm; the cd spectrum was identical with that already published (12). Isoacteoside [2] exhibited mp 136–139° and solution phase sims m/z625 [M + H]* (22%), 480 (29%), 472 (30%), 325 $[M + H = caffeovl-2, 4-dihvdroxyphenethyl]^*$ (100%). Acetylation and chromatography of the product gave the peracetate as a colorless amorphous powder: uv (MeOH) λ max 280 (log ϵ +.53) nm. Acteoside [1] and isoacteoside [2] both showed activity against the PS cell line (ED₅₀ 2.6 and 10 µg/ml, respectively),

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ANTINEOPLASTIC AGENTS, 195. ISOLATION AND STRUCTURE OF ACERATIOSIDE FROM ACERATIUM MEGALOSPERMUM¹

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ABSTRACT.—A new tetralin glucoside [1], named aceratioside, has been found to be a weakly cytostatic (PS ED_{50} 9 µg/ml) constituent in leaves produced by the Australian rain forest tree Aceratium megalospermum. Structural determination of aceratioside was primarily accomplished with results from a series of acetylation, methylation, and high field nmr (including heteronuclear multiple bond correlation) experiments.

The largest (200 of 350 species) genus, *Elaeocarpus*, of the subtropical to tropical Elaeocarpaceae (2) contains the *Elaeocarpus* alkaloids (3), a series of indolizidines that have received principal phytochemical attention. Chemical constituents of the genus *Aceratium* (4) in this family have remained essentially unexplored. *Aceratium* species occur primarily in the rain forests of Papuasia (5), and their relative inaccessibility probably accounts for the lack of prior chemical investigation.

Early (1962) in the U.S. National Cancer Institute's (NCI) world-wide evaluation of plants with potential anticancer constituents, extracts of the Queensland tree (to 15 m high) Aceratium megalospermum (F. v. M.) von Balgooy (6) began to be evaluated, and in the period 1964–66 they were confirmed as active against the NCI KB cell line (ED₅₀ 0.17 µg/ml) and Walker carcinosarcoma (72% tumor reduction at 22 mg/kg). This lead was pursued with a 1979 re-collection of Ac. megalospermum leaves and the NCI P-388 lymphocytic leukemia cell line (PS system).

A CH₂Cl₂-MeOH (1:1) extract of the leaves (55 kg) was successively partitioned between MeOH-H₂O (9:1 \mapsto 3:2) and hexane \mapsto CH₂Cl₂. The PS activity was found concentrated in the CH₂Cl₂ (ED₅₀ 0.07 µg/ml) and residual MeOH/H₂O (ED₅₀ 0.16 µg/ml) fractions. The more pronounced cell growth inhibitory activity of the former fraction was due to the presence of cucurbitacins; cf. Bittner *et al.* (7). The latter fraction was separated by Si gel chromatography to afford the cytostatic constituent, accratioside [1], in 5.0 × 10⁻³% yield, as a colorless powder (PS ED₅₀ 9 µg/ml) from Me₂CO/CHCl₃.

The high resolution fab mass spectrum of aceratioside [1] showed a molecular ion at m/z 371 [M + H]⁺ corresponding to the molecular formula $C_{17}H_{22}O_9$ and suggested seven double bond equivalents. Uv and it spectra indicated the presence of an aromatic ring system bearing a phenolic hydroxyl group (bathochromic shift by addition of NaOMe). The it spectrum also exhibited absorption bands for hydroxyl groups (3446–3380 cm⁻¹), a carboxyl type carbonyl (1682 cm⁻¹) group, and an aromatic ring. The 400 MHz ¹H-nmr spectrum (Table 1) showed a complex pattern, and the proton spin systems required resolution employing 2D ¹H, ¹H COSY (8). The latter 2D data revealed structural segments consisting of meta-coupled aromatic protons, ArCH₂CH₂CH(X)CH₂-Ar (subunit A), and -OCH(O)-CH(

The ¹³C-nmr spectrum (Table 2) of 1 exhibited signals for seventeen carbons, deduced to be three simple methylenes, an oxymethylene, a shielded methine, five oxymethines, two olefinic methines (one relatively shielded at 102.7 ppm), two olefinic quaternary carbons, two oxygenated quaternary carbons, and a shielded carbonyl car-

¹For part 194, see Drexler et al. (1).



bon. The proton-bearing carbons were correlated to the protons attached to them by interpreting a ${}^{1}\text{H}{}^{-13}\text{C}$ COSY (9,10) spectrum. For example, the meta-coupled proton appeared at δ 6.95 and was correlated to the carbon at δ 102.7, reminiscent of an aromatic carbon ortho to two oxygen-bearing carbons.

Acetylation of aceratioside yielded a pentaacetate 2, and methylation (MeI and K_2CO_3 in refluxing Me₂CO) gave the methyl ether 3. The aromatic methyl ether and methyl ester signals appeared at δ 3.85 and 3.77 and in the ir at 1726 cm⁻¹. These results clearly suggested that subunit A contained a carboxyl group and subunit B a trihydroxypyranose with a hydroxymethyl group. Because the aromatic ring contained only one free hydroxy group, the other aromatic ring bonded oxygen atom must form an ether linkage. These facts suggested a 1,2,3,4-tetrahydronaphthalene-2-carboxylic acid glycoside. Placement of both oxygen atoms in a meta orientation in the aromatic ring was achieved using nOe methods. Irradiation of the isolated proton triplet at δ 3.42 (H-1) in the ¹H-nmr spectrum of aceratioside [1] significantly enhanced the signal at δ 6.65 (H-8) and placed oxygens at C-5 and C-7 (meta).

Exact position of the glycoside linkage was resolved on the basis of nOe studies (see 4) and application of 2D nmr heteronuclear multiple bond connectivity (HMBC) (11– 13) methods. With the methylation product 3, nOe irradiation of methoxy singlets at δ 3.85 and 3.77 ppm enhanced signals for the aromatic protons at δ 6.35 and 6.32 ppm, respectively. The observed nOe between the methyl group of the methyl ester and H-8 was explained by a partial boat conformation 4 for the cyclohexene ring, thereby forcing the methyl ester closer to the aromatic ring. In this conformation H-2 and H-4a appear in a quasi 1,3 diaxial relationship or even closer if the molecule resides in a semi chair conformation [5]. One likely explanation of the nOe results depends on the two semi boat/chair conformations 4 and 5 existing in equilibrium.

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Proton	Compound			
	$1(in C_5D_5N + D_2O)$	2 (in CDCl ₃)	3 (in CDCl ₃)	
H-la	2.44, dt, 5.4, 12.6 3.42, dt, 5.4, 12.6 1.80, m 1.75, m 1.88, m 2.29, dd, 2.3, 14.2 2.54, dd, 5.3, 14.6 6.957d, 2.8 6.65, d, 2.8 4.81, d, 8 4.20, dd, 8, 9 4.14, dd, 9, 9 3.84, dd, 9, 9 4.10, ddd, 3, 9, 11 4.77, dd, 11, 11 5.00, dd, 3, 11	2.30, m 2.88, dt, 3.5, 10.2 1.69, m 1.78, m 1.79, m 2.44, m 2.52, m 6.78, d, 2.7 6.73, d, 2.7 5.05, d, 6.8 5.24, dd, 6.8, 8.6 5.18, t, 8.6 5.29, t, 10 3.73, dt, 3.5, 10 3.98, dd, 3.5, 12 4.40, dd, 3.5, 12	2.39, dt, 4.4, 13 3.02, dt, 4.6, 13 1.78, m 1.87, m 2.27, m 2.53, td, 4.2, 16 6.35, d, 2.8 6.32, d, 2.8 4.30, d, 7.7 3.63, t, 8.5 3.53, t, 8.6 3.51, t, 8.8 3.65, m 4.49, dd, 12, 4 4.50, dd, 12, 12	
Λc		2.02, 2.04, 2.07, 2.24, 2.29 —		
ОН	—		2.40, 2.67, 4.84	

 TABLE 1.
 ¹H-nmr Assignments for Aceratioside [1], Aceratioside Pentaacetate [2] and the Methyl Ether 3.

The HMBC technique was used to assign (Table 2) all the carbon resonances. Strong and multiple correlation peaks were observed for all the protons two or three bonds away from carbon atoms. The H-1' (δ 4.81) proton was correlated to aromatic ring C-4a (δ 138.8), C-3' (74.6), and C-5' (δ 78.0). Relationship of the carboxyl group (δ 173.3) to H-3 (δ 1.75, 1.88 ppm), H-2 (δ 1.80), and H-4 (δ 2.29, 2.54 ppm) was also established. Proton H-6 (δ 6.95) was related to C-4a, -5, -7, and -8, and H-8 with C-1, -4a, -6, and -7. For other connectivities refer to Table 2. Interestingly, whenever a proton was arranged in a W-type spatial relationship with a carbon, a four-bond-apart HMBC correlation was observed. The HMBC experiments allowed placement of the glycoside at C-5 and confirmed the substitution pattern of the aromatic ring and carboxylic acid at position C-2. The glycoside linkage was assigned β on the basis of the anomeric proton coupling constants for the doublet H-1' in aceratioside [1] (J = 8 Hz), pentaacetate 2 (J = 6.8 Hz), and methyl ester 3 (J = 7.7 Hz). Acid hydrolysis of aceratioside led to the isolation of glucose and assignment of structure 1 to aceratioside.

The cd spectrum of methyl ester 3 in MeOH exhibited two relatively weak negative Cotton effect curves at λ max 230 and 270 nm due to the phenethyl chromophore (14). On the basis of direct comparison of the cd spectrum with that of the 1,2,3,4-tetrahydronaphthalene-(2S and 2R)-carboxylic acids (15), the absolute configuration of aceratioside [1] at C-2 was assigned S. [Cd spectra of 1,2,3,4-tetrahydronaphthalene-(2S and 2R)-carboxylic acids in MeOH are: 2S [α]D = 55°; \in (nm) 0 (280), = 1.2 (270), 0 (250), 0 (225), =33.3 (210); 2R: [α]D + 28.5°; \in (nm) 0 (280), +0.7 (270), 0 (250), +34.4 (210).]

Certain tetralins have shown a variety of biological activities. For example, 6-substituted 1,2,3,4-tetrahydro-1-naphthoic acids have exhibited moderate anti-inflammatory activity (16) in mice. Other 1,2,3,4-tetrahydro-1-naphthoic acids have shown plant-growth-regulating activities (17), and 5,7-dihydroxy-2-aminotetralins have

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Carbon	Compound			
	1 (in C, D, N)	2 (in CDCl ₃)	HMBC of $1 (in C_5 D_5 N)$	
C-1 C-2 C-3 C-4 C-4 C-4a C-5 C-6 C-7 C-8 C-8 C-8a C-9 C-1' C-2' C-3' C-4' C-5' C-6 C-7 C-8 C-9 C-1' C-2' C-3' C-4' C-5 C-6 C-7 C-8 C-6 C-7 C-8 C-6 C-7 C-7 C-8 C-6 C-7 C-6 C-7 C-8 C-7 C-6 C-7 C-7 C-8 C-6 C-7 C-6 C-7 C-7 C-7 C-8 C-2' C-3' C-4' C-5' C-6' C-6' C-7 C-6 C-7 C-6 C-7 C-6 C-7 C-7 C-7 C-6 C-7 C-6 C-7 C-6 C-7 C-7 C-6 C-7 C-6 C-7 C-6 C-7 C-2' C-6'	31.2 31.5 25.8 34.8 138.8 152.4 102.7 156.7 107.8 138.0 173.3 108.4 75.4 74.6 73.7 78.0 64.7	29.1 28.7 25.6 34.1 138.6 146.8 114.7 143.1 120.5 142.9 173.5 100.5 72.2 71.0 68.7 73.1 60.6 20.6(×3), 20.8 21.1, 168.2	H-1 \mapsto C-2, -3, -4a, -8, -8a H-2 \mapsto C-3 H-3 \mapsto C-1, -4, -9 H-4 \mapsto C-2, -3, -9 H-4 \mapsto C-2, -3, -9 H-6 \mapsto C-4a, -5, -7, -8 H-8 \mapsto C-1, -4a, -6, -7 H-1' \mapsto C-4a, -3', -5' H-2' \mapsto C-1', -2', -4' H-4' \mapsto C-1', -2', -4' H-4' \mapsto C-3', -5', -6' H-5' \mapsto C-6', -2', -4' H-6' \mapsto C-5'	
		169.4, 170.3		

 TABLE 2.
 ¹³C-nmr Assignments for Aceratioside [1] and Aceratioside Pentaacetate [2] and HMBC Correlations for Glucoside 1.

exhibited dopaminergic and adrenergic action in dogs (18). Further biological evaluation of aceratioside is under way.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Analtech Si gel GF (0.25 mm) plates were used for tlc and developed with either 3% ceric sulfate in 3 N H₂SO₄ spray and/or iodine vapor. Stationary phases used for gravity or flash cc were E. Merck (Darmstadt) Si gel (70–230 mesh for gravity and 40–63 mesh for flash), Whatman Si gel LPS-1 (13–24 mesh), or Sephadex LH-20 (Pharmacia Fine Chemicals AB, Uppsala, Sweden).

Melting points were observed with a Kofler-type hot stage apparatus. Optical rotation values were recorded using a Perkin-Elmer 241 polarimeter. The uv spectra were obtained in MeOH solution with a Hewlett-Packard 8450A UV/vis spectrophotometer. A Nicolett MX-1 FT spectrophotometer was employed for ir measurements. TMS, residual CHCl₃ (7.256 ppm), or CH₂Cl₂ (5.32 ppm) was used as an internal reference in all nmr experiments, which were determined with Bruker AM 400 (¹H, ¹³C), or WH 90 (¹H) instruments. Chemical shifts are recorded in ppm. CDCl₃ was used as nmr solvent unless otherwise mentioned. The hreims and sp-sims (fabms) were recorded with a Kratos MS 50 instrument in the NSF regional mass spectrometry facility at the University of Nebraska.

PLANT MATERIAL.—The leaves of Ac. megalospermum, earlier (1875) classified as Aristotelia megalosperma, were collected in the rain forests of Queensland, Australia in 1979 as part of the NCI-USDA programs directed by Drs. John L. Hartwell, Matthew Suffness, and J. Duke (USDA). The plant was assigned NCI B631963, and a herbarium specimen is maintained by the USDA, Beltsville, Maryland.

ISOLATION OF ACERATIOSIDE [1] (7-HYDROXY-1,2,3,4-TETRAHYDRONAPHTHALENE-2-CAR-BOXYLIC ACID-5- β -D-GLUCOSIDE]. —Dried leaves (50 kg) of *Ac. megalospermum* were extracted at ambient temperature with 400 liters of CH₂Cl₂-MeOH (1:1), and the extract was converted to a CH₂Cl₂ fraction (2.07 kg, PS ED₅₀ 0.3 µg/ml) by addition of H₂O (100 liters). The CH₂Cl₂ fraction (1.55 kg) was dissolved in 8 liters of MeOH-H₂O (9:1) and extracted with hexane (4 × 4 liters). The MeOH-H₂O phase was adjusted to 3:2 by addition of H₂O and extracted with CH₂Cl₂ (3 × 4 liters) to give CH₂Cl₂ (358 g, PS

ED₅₀ 0.07 µg/ml) and residual aqueous fractions (93 g, PS ED₅₀ 0. 16 µg/ml, PS T/C 121 at 25 mg/kg). An 11.8-g aliquot of the latter fraction was chromatographed on a column of Si gel employing gradient elution with CH₂Cl₂-MeOH (49: 1 \mapsto 17:3) to give an active fraction that crystallized from Me₂CO/CHCl₄ to furnish aceratioside [1] (0.32 g, 0.005% yield) as a buff colored powder: mp 273–275°, R₁0.35 [Si gel, CH₂Cl-MeOH (9:1)]; [α]D + 10° (c=1.2, MeOH); hr fabms m/z [M + H]⁺ 371.1366 (100%) for C₁₇H₂₃O₉ (calcd 371.1341); eims m/z [M]⁺ 370 (8%), {M - H₂O]⁺ 352 (5%), [M - 2H₂O]⁺ 334 (10), [M - glucose + H]⁺ 208 (73), [M - glucose - CO]⁺ 180 (100), [M - aglycone]⁺ 163 (24), [M - glucose - 2 × CO]⁺ 152 (30); uv (MeOH) λ max 226 (ϵ 5124), 281 (2032); uv (MeOH + NaOCH₃) λ max 230 (ϵ 5442), 235 (5255), 290 (3097); ir (KBr) ν max 3446, 3380, 1682, 1607, 1492, 1343, 1165, 1147, 1084, 1071, 1054, 1037 cm⁻¹; ¹H nmr see Table 1; ¹³C nmr see Table 2. Anal. calcd for C₁₇H₂₂O₉·1.5 H₂O: C 51.38, H 6.09; found C 50.88, H 5.83.

ACERATIOSIDE PENTAACETATE [2].—A solution of glycoside 1 (50 mg) in pyridine/Ac₂O (1 ml each) was acetylated at room temperature (overnight). After addition of EtOH, the solvent was evaporated at reduced pressure. The pentaacetate (70 mg) crystallized from EtOH as an amorphous powder: mp 189–191°; R_f 0.2 or 0.72 [Si gel, hexane-EtOAc (9:5 or 1:1)]; [α]D = 10° (c = 1.2, MeOH); hreims m/z [M]⁺ 580.1886 (2%) (calcd for C₂₇H₃₂O₁₄, 580.1792), {M = 2 × Ac = OHAc}⁺ 436.1356 (6%) (calcd for C₂₁H₂₄O₁₀, 436.1370), [C₂₁H₂₂O₉]⁺ 418.1281 (14), {M = 2 × OHAc = 2 × Ac}⁺ 376.1144 (16), [C₁-H₁₈O₇]⁺ 334.1046 (7), [C₁₅H₁₆O₆]⁺ 292.0948 (20), [C₁₃H₁₄O₅]⁺ 250.0833 (80), [C₁₁H₁₂O₄]⁺ 208.0739 (100), [C₁₀H₁₂O₃]⁺ 180.0789 (61), [C₈H₈O₃]⁺ 152.0474 (34); ir (NaCl) 1757, 1476, 1370, 1214, 1199, 1069, 1035 cm⁻¹; ¹H nmr see Table 1; ¹³C nmr see Table 2.

METHYLATION OF ACERATIOSIDE. —To a solution of aceratioside [1] (11.3 mg) in anhydrous $Me_2CO(3 ml)$ was added anhydrous K_2CO_3 (50 mg) and MeI (1 ml). The mixture was heated at reflux for 3 h. The K_2CO_3 was removed by filtering the solution, and the filtrate was purified by chromatographic separation on a small Si gel column. Elution with EtOAc-hexane-MeOH (9:1:1) provided methyl ester **3** (8 mg) as an amorphous pewder: mp 208–210°, $R_fO.32$ [Si gel, hexane-CH₂Cl₂-MeOH-H₂O(10:80:10:1)], hreims m/z [M]⁺ 398.1575 (3%) (calcd for $C_{19}H_{26}O_9$, 398.1577), [$C_{13}H_{16}O_4$]⁺ 236.1047 (21); cd (MeOH) ϵ (nm) 0 (283), -1.0 (270), 0 (256), 0 (240), -7.6 (230), 0 (220); ir (NaCl) 3500, 1726, 1598, 1492, 1450, 1350, 1210, 1095, 1075, 1055, 1040 cm⁻¹; ¹H nmr see Table 1; ¹³C nmr see Table 2.

HYDROLYSIS OF ACERATIOSIDE.—The glucoside 1 (14.3 mg) was heated in a refluxing solution of 1 N HCl-MeOH (1:1) (5 ml) for 24 h. EtOAc (15 ml) was added, and the EtOAc phase was washed with $H_2O(2 \times 5 \text{ ml})$ and dried (anhydrous Na₂SO₄). Solvent was removed from both phases. The EtOAc extract was found to be a complex mixture, whereas the aqueous phase yielded glucose [identified by direct comparison with an authentic specimen using tlc on a cellulose plate with pyridine-EtOAc-HOAc-H₂O (5:5:1:3) as mobile phase and by ¹H-nmr comparison]. The pentaacetate derivative was also found to be identical with an authentic sample of glucose pentaacetate by the same procedures.

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Antineoplastic agents. 168. Isolation and structure of axinohydantoin¹

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Western (Palau) and Eastern (State of Truk) Caroline Islands and Papua New Guinea sponges of the genera Axinella and Hymeniacidon were found to contain the cytostatic (PS $ED_{50} 2.5$ and $2.0 \mu g/mL$) and antineoplastic (PS T/C 143 at 3.6 mg/kg and T/C 138 at 3.6 mg/kg) pyrrologuanidines 1a and 1b. The related hydantoin 2, designated axinohydantoin, was also isolated from an Axinella sp. and its structure was assigned by X-ray crystallographic techniques. Present experience with sponges in the Axinella and Hymeniacidon genera suggests that the previously known hymenialdisine (1b) and analogous imidazole derivatives may be widely distributed among these and related orange colored Porifera.

Key words: axinohydantoin, hymenialdisine, Axinella, Hymeniacidon, cystostatic.

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On a trouvé que les éponges du genera Axinella et du Hymeniacidon des iles Caroline occidentale (Palau) et orientale (État de Truk) ainsi que de la Nouvelle Guinée contiennent des pyrrologuanidines 1a et 1b qui sont des cytostatiques (PS ED₅₀ 2,5 et 2,0 µg/mL) et des antinéoplasiques (PS T/C 143 à 3,6 mg/kg et T/C 138 à 3,6 mg/kg). À partir d'un Axinella sp., on a aussi isolé l'hydantoïne apparentée 2, appelée axinohydantoïne, et on a déterminé sa structure à l'aide de la diffraction des rayons-X. L'expérience acquise avec les éponges de l'Axinella et de l'Hymeniacidon genera suggère que l'hyménialdisine (1b), qui était connue antérieurement, ainsi que les dérivés imidazoles analogues sont peut-être très répandus dans ces éponges et dans les Porifera apparentés de couleur orange.

Mots clés : axinohydantoïne, hyménialdisine, Axinella, Hymeniacidon, cytostatique.

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Introduction

Early (2) in our evaluation of marine animals as new sources of potentially useful anticancer drugs, good leads were uncovered among the Porifera, and this initial (1966–1968) promise is now being amply realized (3, 4). In 1979 in Palau we collected a *Hymeniacidon* species (at -40 m) and an *Axinella* sp. that provided extracts with confirmed levels of activity against the U.S. National Cancer Institute's (NCI) murine P388 lymphocytic leukemia (PS system). Other PS active sponge collections were completed in 1981 (Papua New Guinea) and 1985 (Truk, Federated States of Micronesia) that included *Axinella carteri* (Dendy) and a *Hymeniacidon* species.

Initial extracts of each sponge were found to provide a confirmed level of activity against the PS system. By means of PS (*in vitro*) bioassay guided separation procedures, these sponge species led to two cytostatic and antineoplastic alkaloids (1*a*,*b*) accompanied in the case of Axinella sp. by a closely related, but marginally (PS ED₅₀ 18 μ g/mL) inactive, component (2). The PS active marine alkaloids proved to be identical⁶ with the known (5-7) hymenialdisine (1*b*, ref. 6, PS

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 $ED_{50} 2.0 \,\mu g/mL$ and T/C 138 at 3.6 mg/kg)⁷ and its debromo derivative 1*a* (refs. 5-7, PS $ED_{50} 2.5 \,\mu g/mL$ and T/C 143 at 3.6 mg/kg).



The unequivocal X-ray crystal structure of hymenialdisine was nicely established by Cimino *et al.* (5) and reconfirmed in the following year by the Kitagawa group (6). In turn, these advances simplified characterization of the companion substance from *Axinella* sp., herein named axinohydantoin (2), as a closely related compound. But establishing the exact geometrical configuration for its hydantoin–lactam sp^2 bond required the following crystal structure determination.

Axinohydantoin (2) crystallized from methanol as yellow prisms, which corresponded to $C_{11}H_9BrN_4O_3$ (by hreims) and with one mole of methanol. The structure was solved using

¹For contribution 167 refer to ref. 1.

⁶By comparison with authentic specimens provided by Dr. I. Kitagawa (see ref. 6).

⁷Interestingly, hymenialdisine was previously active in the KB cell line, but *in*active employing the P388 leukemia: cf. ref. 5. Perhaps the initial negative results were due to the sparingly soluble properties of this pyrrologuanidine.



FIG. 1. ORTEP view of a single molecule of 2, with 50% thermal ellipsoids.

MULTAN (8) and refined to R = 0.054 using anisotropic temperature factors for Br and oxygens other than O-3, and isotropic temperature factors for the other non-hydrogens and for hydrogens (unrefined) in calculated positions. HN-1, HN-11, and HN-9 were calculated to be 0.95 Å along a line to the respective oxygen. These positions differed very little from those calculated assuming bonding to trigonal atoms. HN-4 has been shown at the calculated position assuming a trigonal N-4 that is too far (2.4 Å) from the closest O-10 for significant hydrogen bonding; it may actually bend somewhat toward this O-10.

The structure deduced for axinohydantoin (2, Fig. 1) was found to be closely related to that of hymenial disine (1b), with reversal of configuration at the C7-C8 double bond being the most interesting difference. In turn, this suggested that axinohydantoin was not simply a hydrolysis product of guanidine 1b. The most prominent bond length difference between the two structures occurs at C10-O10, with 1.23 Å in hydantoin 2 compared to 1.33 Å for C11-N11 in 1b. No significant differences in bond angles were observed. An angle of 36° was observed between the least-squares planes of the two nearly planar five-membered rings in hydantoin 2, compared to 43.8° in guanidine 1b. In both cases, the seven-membered ring has adopted a boat conformation with C-5 at the prow, and similar torsion angles except for C2-C3-N4-C5 expanding from -10.5° in 1b to -15° in 2, and C2-C13-C7-C6 contracting from 41.1° in 1b to 31° in 2. The twist angle C13-C7-C8-C12 about the carbon-carbon double bond increases from 0.5° in 1b to 10° in 2, presumably to relieve the steric interaction between O-12 and HC-14.

The arrangement of intermolecular hydrogen bonds govern-

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ing the packing in hydantoin 2 (see supplementary material)⁸ was found to be completely different than that in guanidine 1b (6). The hydantoin ring in each molecule was found linked to the hydantoin rings of two other molecules via base-pairing interactions across centers of symmetry. The pyrrole NH proved to be hydrogen bonded to the methanol solvate oxygen and in turn to O-3. Only a few substances (9–12) with a hydantoin system have been isolated from sponges, and one of these, midpacamide, found by Scheuer and co-workers (9) in an unidentified Marshall Island sponge, may be biogenetically related to axinohydantoin. From evidence now in hand, pyrroles 1 and 2 and related substances may prove to be ubiquitous Porifera biosynthetic products.

Experimental

General methods

Marine sponge taxonomic identification was performed in the Smithsonian Institution where voucher specimens are deposited in the collections of the Department of Invertebrate Zoology, National Museum of Natural History. All solvents employed were redistilled. Size exclusion chromatography was accomplished with Sephadex LH-20 (particle size: $25-100 \mu$ m) suppled by Pharmacia Fine Chemicals, Uppsala, Sweden. Thin-layer chromatography was carried out with silica gel GHLF Uniplates (Analtech Inc.) and with RP-8 precoated plates (layer thickness: 0.25 mm) from E. Merck, Darmstadt, Germany. High-speed countercurrent chromatography was accomplished with an Ito multilayer coil extractor-separator (P.C. Inc., Potomac, MD) using 2.6 mm i.d. tubing, and an FMI Lab Pump.

Melting points are uncorrected and were determined on a Kofler-type hot-stage apparatus. Ultraviolet spectra were recorded employing a Hewlett-Packard model 8450 uv/vis spectrophotometer and ir spectra with a Nicolet ft-ir model MX-1 instrument. The nmr spectra were measured in DMSO- d_6 using a Bruker AM-400 instrument and are recorded in ppm downfield to TMS (assignments bearing the same superscript may be reversed). The ¹³C nmr multiplicities were determined with APT experiments based on an average coupling constant of 135 Hz. The eims spectra were recorded with a Kratos AEI 5076 spectrometer at the NSF Regional Facility. University of Nebraska, Lincoln, Nebraska.

Palau Porifera (Axinella sp. and Hymeniacidon sp.) collection and extraction.

The initial collection of Axinella sp. (Demospongiae class, Axinellida order, Axinellidae family) in Palau, Western Caroline Islands, was conducted in May, 1979. The sponge displayed a brownish-yellow exterior of irregular mass. A 2-propanol–CHCl₃ extract gave confirmatory in vivo activity with PS T/C 201 at 100 mg/kg. PS $ED_{50} =$ 2.5 µg/mL. A scale-up re-collection (220 kg wet wt.) of this sponge was completed in March, 1985, and preserved in 2-propanol. The preserving solution was separated from the sponge, concentrated to an aqueous slurry, and extracted with CH₂Cl₂ (13). The remaining sponge material was re-extracted with 2-propanol–CH₂Cl₂ (1:1): the extract was separated, solvent removed, and the residue partitioned between CH₂Cl₂ and H₂O At this early stage a solid precipitate appeared at the CH₂Cl₂-H₂O interface. The precipitate was separated and amounted to

⁸Tables of observed and calculated structure factor amplitudes, calculated hydrogen coordinates, isotropic temperature factors, bond lengths and angles, torsion angles, and a packing diagram may be purchased from the Depository of Unpublished Data, CISTI, National Research Council of Canada, Ottawa, Ont., Canada K1A 0S2.

Tables of positional parameters and bond distances have also been deposited with the Cambridge Crystallographic Data Centre, and can be obtained on request from The Director, Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, U.K.
1.8 kg (PS T/C 227 at 294 mg/kg, ED_{50} 2.8 μ g/mL). Analogous solid fractions were also obtained at this initial CH_2Cl_2 step during separations of the sponge extracts summarized below.

The orange sponge Hymeniacidon sp. (Demospongiae class, Halichondrida order, Hymeniacidonidae family) was collected (1979), re-collected (in 1985, 218 kg wet wt.), and extracted (2-propanol extract showed PS T/C 130 at 5.5 mg/kg and ED₅₀ 27 μ g/mL) as summarized above. Removal of solvent from the initial 2-propanol extract led to an aqueous concentrate that contained 1.2 kg of a solid fraction with PS ED₅₀ = 3.7 μ g/mL.

Isolation of hymenialdisine (1b) and axinohydantoin (2)

A 10 g aliquot from the 1.8 kg of solid precipitate noted above was dissolved in CH₃OH (400 mL) and separated by size exclusion chromatography on a column of Sephadex LH-20 (100 × 10 cm) to yield two major marine alkaloid fractions. When fraction 1 (elution volume: 12.0–12.6 L) was allowed to stand for 24 h at room temperature, axinohydantoin (2) slowly crystallized as yellow needles (30 mg): mp >350°C; tlc on silica gel $R_f = 0.83$, 1-BuOH–AcOH 50% (95:5); tlc on RP-8 $R_f = 0.51$, CH₃OH–AcOH 55% (1:1); uv (CH₃OH) λ_{max} : 264sh (log $\varepsilon = 3.88$), 345 (log $\varepsilon = 4.16$) nm; ir (KBr) ν_{max} : 1740, 1702, 1638, 1480, 1425, 1407 cm⁻¹; ¹H nmr (DMSO- d_6) δ : 2.67 (2H, m, H-6), 3.22 (2H, m, H-5), 6.66 (1H, s, H-14), 7.89 (1H, t, HN-4), 9.83, 10.91 (2 × 1H, s, HN-9, HN-11), 12.35 (1H, s, HN-1); ¹³C nmr (DMSO- d_6) δ : 36.2 (t, C-6), 38.5 (t, C-5), 101.6 (s, C-15), 113.9 (d, C-14), 120.0 (s, C-13), 121.2 (s, C-7), 125.5 (s, C-2), 126.5 (s, C-8), 153.8 (s, C-10), 162.7 (s, C-3), 163.3 (s, C-12); hreims m/z: 325.9842 and 323.9834 (C₁₁H₉N₄O₃Br requires 325.9836).

Fraction 2 (elution volume: 12.9-13.5 L) yielded a crystalline precipitate (100 mg), which was identified as hymenialdisine (1b) by comparison (uv, ir, ¹H nmr, eims) with an authentic sample (6).

Truk Porifera (Axinella carteri) collection and extraction

In May 1985, approximately 1 kg of an orange-yellow sponge subsequently identified as Axinella carteri (Dendy), was collected in the Truk Lagoon, Federated States of Micronesia, at -13 to -24 m. The preserving solution (2-propanol) was removed and this extract proved toxic down to 50 mg/kg against the PS leukemia. The 2-propanol extract was partitioned between CH₂Cl₂ and H₂O and the resulting CH₂Cl₂ extract was successively partitioned (13) between 9:1-4:1-1:1 MeOH:H₂O with hexane-CCl₄-CH₂Cl₂. The final CH₂Cl₂ extract showed PS T/C 135 at 100 mg/kg and PS cell line ED₅₀ = 1.2 µg/mL.

In October 1985, approximately 148 kg (wet wt.) of the sponge was recollected and preserved in MeOH. The MeOH solution was decanted, and the sponge was ground and extracted with MeOH:CH₂Cl₂ (1:1). The original MeOH solution was concentrated to an aqueous phase and extracted with CH₂Cl₂ (3×) followed by 1-BuOH. Study of this 1-BuOH fraction was discontinued when PS results showed minimal activity.

When the ambient temperature extraction of the sponge with MeOH:CH₂Cl₂ was completed, the aqueous MeOH phase was separated and concentrated to an aqueous phase, which was extracted with 1-BuOH (15 L). The 1-BuOH phase was concentrated, redissolved in MeOH (1.5 L), and dried to give a 232 g fraction (PS ED₅₀ 1.4 μ g/mL). A 97 g aliquot of the MeOH soluble fraction was treated with 1-BuOH (800 mL, 50°C, 12 h) and the relatively insoluble part (50 g, PS ED₅₀ 1.5 μ g/mL) was collected. The MeOH (600 mL) sparingly soluble portion weighed 4.26 g (PS ED₅₀ 0.11 μ g/mL).

Papua New Guinea Porifera (Hymeniacidon sp.) collection and extraction

The collection (May 1981, near Motapure Island, Papua New Guinea) and recollection (October 1983, 44 kg wet wt.) of an orange *Hymeniacidon* sp. as well as the large scale extraction (crude extract PS T/C 136 at 100 mg/kg and ED₅₀ 24 μ g/mL) and solvent partitioning were performed as described above for *A. carteri*. In this case, when the 934 g initial CH₂Cl₂ fraction was subjected to further separation by MeOH:H₂O with the hexane \rightarrow CCl₄ \rightarrow CH₂Cl₂ sequence, a total

TABLE 1. Positional and thermal parameters

		and the second		
Atom	x	у	Z	B (Å ²)
Br	0.56111(6)	0.0097(2)	0.18172(7)	4.40(3)
O3	0.7915(4)	-0.210(1)	0.0865(4)	3.4(2)*
012	0.7371(3)	-0.194(1)	0.4049(4)	3.0(2)
O10	0.9511(3)	-0.093(1)	0.5586(3)	3.3(2)
ОМ	0.6407(4)	-0.152(2)	0.0121(5)	7.3(3)
N1	0.6929(4)	-0.069(1)	0.1562(4)	2.3(2)*
N4	0.8760(4)	-0.055(1)	0.1624(5)	2.9(2)*
N11	0.8375(4)	-0.161(1)	0.4951(4)	2.5(2)*
N9	0.9106(4)	-0.056(1)	0.4357(4)	2.0(2)*
C15	0.6581(5)	-0.024(2)	0.2056(5)	2.5(2)*
C14	0.7046(4)	-0.000(2)	0.2706(5)	2.2(2)*
C13	0.7725(4)	-0.036(1)	0.2599(5)	1.8(2)*
C2	0.7630(5)	-0.073(1)	0.1875(5)	2.0(2)*
C3	0.8108(5)	-0.120(2)	0.1407(6)	2.8(2)*
C5	0.8961(5)	0.083(2)	0.2145(5)	2.6(2)*
C6	0.9053(4)	0.024(2)	0.2915(5)	2.4(2)*
C7	0.8393(4)	-0.031(1)	0.3144(5)	1.8(2)*
C8	0.8468(5)	-0.069(1)	0.3839(5)	1.9(2)*
C12	0.7982(5)	-0.147(1)	0.4248(5)	2.0(2)*
C10	0.9045(5)	-0.104(2)	0.5026(5)	1.9(2)*
СМ	0.5809(8)	-0.093(2)	-0.0359(8)	6.3(4)*

*Starred atoms were refined isotropically.

Anisotropically refined atoms are given in the form of the isotropic equivalent thermal parameter, defined as $8\pi^2(U_{11} + U_{22} + U_{33})/3$.

of 135 g (PS ED₅₀ 14 μ g/mL) of a solid interfacial fraction was collected and used to isolate hymenial disines 1*a* and 1*b*.

Isolation of hymenial disines Ia and Ib-Procedure A

An aliquot (250 mL) of the preceding Axinella carteri MeOH (600 mL) solution was applied to a column of Sephadex LH-20 (1.9 kg in MeOH). A total of 460 fractions G 20 mL each were collected and a fraction weighing 0.73 g (PS $ED_{50} 2.2 \ \mu g/mL$) was further separated using high speed countercurrent distribution with an Ito coil. A 50 mg aliquot was applied (6 mL) in 1-BuOH:HOAc:H₂O (4:1:5) to the coil with the 1-BuOH phase as stationary (upper) and the aqueous part as mobile (lower) phase. Fractions (120) of 6.5 mL each were collected; fractionation was monitored with ultraviolet detection (254 nm). The fractions were neutralized (pH 7) with aqueous NaOH and refrigerated. Debromohymenialdisine 1a. 9 mg, PS $ED_{50} = 3.0 \ \mu g/mL$, crystallized from fractions 28–33 and was identical (tlc, ms, nmr) with an authentic sample (6).

The MeOH less soluble fraction (4.26 g) described above was extracted with MeOH $(5 \times 25 \text{ mL})$ at 40°C and the solution filtered to give 3.73 g of residue. A 0.90 g portion was triturated with DMSO (10 mL). The soluble portion (0.25 g) was chromatographed on a column of Sephadex LH-20 in MeOH to provide 0.13 g of hymenialdisine (1b) as yellow crystals (PS ED₅₀ 0.62 µg/mL), identical (tlc and ms comparisons) with an authentic sample (6).

Procedure B

The 1.2 kg fraction (see above) from the Palau Hymeniacidon sp. was further separated by successive Soxhlet extraction (20 g aliquot) with CH_2Cl_2 (6 × 5 L), EtOH (6 × 5 L), and 1-BuOH (6 × 5 L) to give respectively 35 g (PS ED₅₀ 26 μ g/mL), 500 g (PS ED₅₀ 8.6 μ g/mL), and 106 g (PS ED₅₀ 2.6 μ g/mL) fractions. A 10 g sample of the 1-BuOH fraction in MeOH was subjected to chromatography on a column of Sephadex LH-20 (500 g) to give 26 individual (by the comparisons) fractions using 4:1 CH₂Cl₂:MeOH. Of these, 56 mg proved to be largely debromohymenialdisine 1*a* (PS ED₅₀ 1.4 μ g/mL) and hymenialdisine (5.5 mg, 1*b*, PS ED₅₀ 7.5 μ g/mL) by comparison nmr and the.

Procedure C

The 135 g fraction from the Papua New Guinea Hymeniacidon sp.

was extracted (Soxhlet procedure with two stainless steel 1-gallon extractors) with EtOH to yield a 22 g alcohol soluble fraction. Treatment of this fraction with CH₂Cl₂:CH₃OH (1:1) yielded a precipitate (4.3 g, PS ED₅₀ 8.5 μ g/mL), which was extracted with hot 1-BuOH. The 1.5 g 1-BuOH soluble fraction was preabsorbed onto silica gel and separated by chromatography on a column (3 × 62 cm) of silica gel (180 g). Gradient elution with 95:5 CH₂Cl₂:CH₃OH with increments of MeOH provided fractions that yielded (0.17 g and 0.06 g respectively) debromohymenialdisine (1*a*, PS T/C 143 at 3.6 mg/kg and ED₅₀ 2.5 μ g/mL). Both 1*a* and 1*b* were identified by direct comparison with authentic samples (6) employing tlc, ¹³C and ¹H nmr, ms, uv, and ir spectral data.

X-ray structure of axinohydantoin (2)

$C_{11}H_8BrN_4O_3 \cdot CH_3OH$ fw = 357.17

Monoclinic, C2/c, a = 19.558(2), b = 7.505(1), c = 19.092(3) Å, $\beta = 103.78(1)^\circ$, V = 2754.3 Å³, Z = 8, $\rho_c = 1.72$; Nicolet P2₁ diffractometer, crystal $0.17 \times 0.13 \times 0.08$ mm, 23°C, MoK_{a1}, $\lambda = 0.71073$ Å, 20/ θ scans, 2 θ_{max} 50°; 836 of 2446 reflections with $F_o^2 > 3\sigma$ (F_o^2) used; solved by direct methods; R = 0.053, $R_w = 0.055$ excluding unobserved reflections. Coordinates and isotropic temperature factors of non-hydrogens are given in Table 1.⁸

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ISOLATION AND STRUCTURE OF BRYOSTATINS 14 AND 151

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SUMMARY: Further investigation of constituents from the marine bryozoan Bugula neritina employing new 1,000 Kg recollections from the Gulf of Mexico and Eastern Pacific Ocean (California) has led to isolation and structural determination of two previously undetected members of the bryostatin (1-13) series, bryostatins 14 (14) and 15 (15). Structural analyses were conducted primarily with high field (400 MHz) NMR and high resolution mass spectral techniques. Both new bryostatins significantly inhibited growth of the P388 lymphocytic leukemia.

Discovery² of bryostatins 1-13 (<u>cf</u>. 1-13) has made available a new class of important biochemical probes³ with considerable clinical potential.⁴ For example, with fresh samples of human myeloid leukemia, bryostatin 1 generally caused differentiation responses leading to macrophage-like morphology.^{4a} Again, with peripheral blood cells from β -chronic lymphocytic leukemia patients, this substance triggered activation and differentiation^{4b} and





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is undergoing clinical evaluation. In order to meet potential clinical supply requirements for the bryostatins, it became necessary to increase (to 1,000 kg, damp wt.) the size of *Bugula neritina* recollections from the Gulf of Mexico^{2f} (Florida) and further explore such challenging quantities of biomass from the Eastern Pacific Ocean (California).²¹ We now report the isolation and structural elucidation of bryostatin 14 (14) from the Gulf of Mexico specimens in 1.02 x 10^{-5} % yield and bryostatin 15 (15) in 8.6 x 10^{-7} % yield from the Pacific Ocean collection of this remarkable bryozoan.

Bryostatins 4-8 (4-8) and 10 (10) were again²¹ isolated from the Gulf of Mexico recollection (1986) and this allowed application of contemporary NMR techniques (HMBC and NOE) to make some refinements in position assignments (Table 1). In turn, these twodimensional NMR correlations aided the characterization of bryostatin 14 (14), separated from the previously known bryostatins by chromatography.

Bryostatin 14 (14, P388 ED₅₀ 0.33 μ g/mL) exhibited a FAB mass spectral base peak at m/z831 $[M+Li]^+$ corresponding to molecular formula $C_{42}H_{64}O_{16}$. The EIMS of bryostatin 14 typically^{2a, 2h} did not show a molecular ion. Fragments at m/z 806, 788 and 770 suggested loss of three hydroxyl groups. The ¹H-NMR spectra of bryostatin 14 indicated the presence of the bryopyran ring.^{2b} Both ¹H-¹H COSY and ¹H-¹³C chemical shift correlation spectra allowed assignment of most 1 H and 13 C signals (Tables 2 and 3). Exceptions involved several overlapped ¹H-NMR signals and ¹³C-NMR signals for carbons without proton bonds. A doublet at & 75.13 clearly indicated a free hydroxyl group at the C-20 position. A pivalate was evident from the strong signals in the ¹H (δ 1.16, 9 protons) and ¹³C (δ 27.14)-NMR spectra. The shift of the H-7 signal downfield to δ 5.10 suggested attachment of the pivalate group at C-7. Other ¹H and ¹³C NMR signals were consistent with the assigned structure. However, a methyl singlet at δ 1.24 assigned to H-33 was at lower field than expected (Table 2). Unequivocal support for the bryostatin 14 structural assignment (14) was obtained by acetylation (acetic anhydride/pyridine) to afford diacetate 14a. As one result, the H-33 signal shifted upfield (from δ 1.24 in bryostatin 14 to δ 1.04 or 0.93 in acetate derivative 14a). Acetylation of bryostatin 5 (5) yielded a single product identical with diacetate 14a. Thus, the paramagnetic shift of the H-33 signal of bryostatin 14 was due to the C-20 hydroxyl group.

Additional evidence for the structure previously assigned bryostatin 5 and thence bryostatin 14 was obtained by HMBC (¹H-detected multiple-bond heteronuclear multiple-quantum coherence)⁵ NMR experiments that established attachment of the pivalate group at C-7, and unambiguous assignments for each of the carbonyl carbon atoms as well as C-8, C-9, C-13, C-18, C-19 and C-21 (Table 4). Finally, NOE difference spectroscopy was applied to assign the geminal dimethyl groups (C-28, 29 and C-32, 33). Irradiation of the H-28 signal at δ 0.91 enhanced the H-7 signal (at δ 5.10) and a broadened doublet signal at δ 1.64 (H-10 α). Reciprocal irradiation of the H-7 signal enhanced the signal at δ 0.91. In contrast, irradiation of the signal at δ 0.98 enhanced only signals at δ 2.03 (doublet of doublets, H-10 β) and 1.40 (H-6 β). Therefore, the signal at δ 0.91 was assigned to the C-28 methyl and the signal at 0.98 to the C-29 methyl hydrogens. Irradiation of H-20 enhanced dramatically

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Bryo	statin 4	Bryostatin 5	Bryostatin b	Bryostatin 8	Bryostatin 10
C-1	172.29	172.37	172.48	172.24	172.66
2	42.14	42.19	42.17	42.30	42.14
3	68.46	68.53	68.48	68.47	68.15
4	39.91	39.96	39.85	39.87	39.78
5	65.48	65.56	65.62	65.72	65.63
6	33,19	33.24	33.38	33.38	33.20
7	72 64	72 66	72.71	72.54	72.61
8	41 20	41.26	41.04	41.02	41.27
ġ	101 73	101 79	101.85	101.82	101.79
10	41 86	41 92	41.91	41.96	42.05
11	71 45	71 51	71 54	71 51	71 35
12	44 13	44 17	44 16	44 14	44 19
13	157 23	157 18	157.06	156 71	157 07
14	36 48	36.46	36 44	36 37	36 57
15	78 91	78 98	79 04	79 08	78 95
16	129 66	129 73	129 68	129 56	130 45
17	138 97	138 96	138 96	139 13	137 90
18	44 75	44 80	44 80	44 82	44 75
19	98 84	98 85	98 83	98 89	100 94
20	76.04	74.45	74 42	74 24	39 78
21	151 83	151 73	151 68	151 81	157 03
22	31 20	31 23	31 20	31 22	36 11
22	64 70	64 74	64 72	64 71	64 62
24	25 81	35 85	35 85	35 89	35 73
25	73 50	73 67	73 76	73 69	73 84
26	70.03	70.09	70.13	70 18	70 21
27	10.03	10.05	19 66	19 78	19 66
28	21 00	21.06	21 08	21 06	21 06
29	16 92	16 98	16 92	16 88	17 06
10	113 99	114 10	116 16	114 31	114 19
31	166 73	166 79	166 81	166 72	166 82
32	10 79	10 86	10 71	19 78	20.39
22	24 57	24 63	26 60	24 57	20.35
34	110 50	110 73	110 72	110 63	115 73
25	166 00	167 00	166 00	166 00	167 00
22	100.70 51 00	53 16	£1 12	51 07	51 07
20	51.08	51.10	51.15	51.07	50.84
ינ יו פ	170 22	179 32	173 60	173 40	178 14
R2 I 21	1/0.32	1/0.32	26 56	36 5/	20.05
2	39.08	39.07	10.50	19 54	37.05
	27.09	27.15	17 66	13 65	27.10
4.	27.09	27.13	T3'00	13.03	47.10 27 16
	27.09	27.10	160.36	173 00	27.10
к I" Он	1/2.00	107.3/	01 /0 107.00	26 54	
2"	30.41 10 17	21.49	21.40	20.34	
3"	10.1/			10.22	
4.7	13.38			13.01	

Table 1. The ¹³C-NMR chemical shift assignments for bryostatins 4, 5, 6, 8, and 10 recorded at 100.6 MHz, δ ppm in CDCl₃ solution.

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		14		14a	15	15
н-	2	2.52	t(11)	2.39 dd(12.5,14)	2.45 brs	R
	2	2.42	dd(2.2,11)	2.37 d(14)	2.45 brs	2" 5.93 d(15.4)
	3	4.12	brd(11)	4.08*	4.16 m	3" 7.24 m
	4	1.97	brdd(11,15)	1.95*	1.75 m	4" 6.39 dd(11,15.5)
	4	1.54	brd(15)	1.53 dt(15,2)	1.60 m	5" 6.07 dd(7.5,15.5)
	5	4.20	brt(11)	4.18 brt(11)	4.20 brt(10)	6" 4.38 q(7.5)
	6a	1.67	brdd(5,12)	1.68 brdd(5,12)	1.77 m	7" 1.59 m
	6β	1.40	dt(12,12)	1.38 dt(12,12)	1.48 m	1.72 m
	7 a	5.10	dd(5,12)	5.05 dd(5,12)	5.14 dd(5,12)	8" 0.94 t(7.5)
	10a	1.64	brd(15)	1.62 brd(15)	1.65 m	
	10 <i>β</i>	2.03	dd(8,15)	2.03 dd(8,15)	2.07 m	
	11	3.91	ddd(4,8,13)	3.78 brdd(8,13)	3.81 m	
	12	2.15	brt(13)	2.15 brt(13)	2.21 brt(13)	
	12	2.05	dd(8,14)	2.0*	2.07	
	14	3.65	5rd(14)	3.58 brd(14)	3.68*	
	14	1.90	brdd(12,14)	1.85 brdd(12,14)	1.96 brt(12)	
	15	4.02	brdd(8.5,12)	4.05 brdd(8.5,12)	4.07 m	
	16	5.30	dd(8.5,16)	5.24 dd(8.5,16)	5.29 dd(8.5,16)	
	17	5.75	d(16)	5.71 d(16)	5.78 d(16)	
	20	3.89	d(7.5)	5.04 s	5.21 s	
	22	3.68		3.62*	3.68*	
	22	2.15	brdd(11,13)	1.98 brdd(11,13)	2.10 m	
	23	3.98	brt(11)	3.93 brt(11)	4.01 m	
	24	1.86		1.82*	1.81 m	
	24	1.80	brdd(11, 14)	1.72 ddd(3,11,13)	1.81 m	
	25	5.12	brm	5.25 brm	5.20 m	
	26	3.74	dq(7,6.5)	4.95 dq(7,6.5)	3.81 m	
	27	1.18	d(6.5)	1.16 d(6.5)	1.23 d(6.5)	
	28	0.91	S	0.95 ^b s	0.94 ^b s	
	29	0.98	S	0.88 ^b s	0.99 ^b s	
	30	5.65	brs	5.62 brs	5.68 brs	
	32	1.12	S	0.93 ^b s	0.99 ^b s	
	33	1.24	8	1.04 ^b s	1.14 ^b s	
	34	5.77	brs	5.91 d(1.8)	6.00 brs	
	36	3.68	8	3.64 s	3.69 s	
	37	3.66	S	3.61 s	3.66 s	
R ₂	3'	1.16	5	1.13 s		
•	4'	1.16	S	1.13 s		
	5'	1.16	8	1.13 s		
20	-OH	4.22	d(7.5)	$2.01 s(R_1)$	2.04 $s(R_2)$	
			,	2.09 s(R)		

Table 2. The 1 H NMR data for compounds 14, 14a and 15 recorded at 400 MHz in CDCl₃ (some J values were measured with J-resolved 2D NMR).

a,

Couplings obscured due to overlapping. Assignments for these signals may be interchanged. Ъ,

	14	14a	15	15
C-1	172.57p	170.82	172.24	R
2	42.25p	42.21	41.95	1" 165.10
3	68.53n	68.60	68.48	2" 121.75
4	40.00p	39.98	39.87	3" 144.58
5	65.58n	65.83 -	65.73	4" 130.82
6	33.24p	33.20	33.34	5" 141.39
7	72.62n	71.47	72.88	6" 86.86
8	41.25p	41.18	40.98	7" 25.44
9	101.76p	101.76	101.83	8" 9.57
10	41.92p	41.96	42.28	
11	71.48n	72.14	71.49	
12	44.11p	44.05	44.14	
13	156.70p	156.06	156.67	
14	36.46p	36.33	36.36	
15	79.01n	79.03	79.08	
16	129.56n	129.54	129.58	
17	139.15n	139.21	139.09	
18	44.93p	44.77	44.88	
19	99.38p	98.88	98.95	
20	75.13n	71.47	74.33	
21	156.94p	151.63	151.73	
22	30.56p	31.13	31.30	
23	64.40n	64.60	64.72	
24	35.78p	35.90	35.84	
25	73.74n	74.32	73.63	
26	70.04n	70.35	67.00	
27	19.54n	19.88	19.80	
28	21.02n	21.01	21.14	
29	16.93n	16.85	16.82	
30	114.23n	114.63	114.32	
31	166./2p	167.00	166./2	
32	19.88n	16.49	24.65	
33	24.66n	24.50	29.69	
34	116.69n	119.68	119.74	
32	167.09p	166.64	166.98	
30	51.04n	51.08	51.05	
_3/	51.04n	51.05	51.05	
K ₂ 1'	178.23p	1//.93		
2,	39.03p	39.01		
37-54	27.14n	77.15		
K1		1/0.30;21.47	1/0.99;21.06	
ĸ		169.28;21.18		

Table 3. The ¹³C-NMR assignments for bryostatin 14 (14), derived diacetate (14a) and bryostatin 15 (15) recorded at 100.6 MHz, δ ppm in CDCl₃ solution; The n (negative, 3 or 1 protons) and p (positive, 2 or no protons) are APT results.

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Proton position		Proton position		
н- 2	C-1, C-3, <u>C-4</u>	22	C-21, C-23, C-24, C-34	
4	C-2	23	C-22	
4	C-2	24	C-23	
5	C-3 –	26	C-24, C-25, C27	
6	C-5, <u>C-8</u>	27	C-25, C-26	
6	C-4, C-5, <u>C-8</u>	28	C-7, C-8, C-9, C-29	
7	C-1', C-6, C-8, C-28, C-29	29	C-7, C-8, C-9, C-28	
10	C-8, C-9, C-11, C-12	30	C-12, C-13, C-14, C-31	
10	C-9, C-11	32	C-17, C-18, C-19, C-33	
12	C-11, C-13, C-14, C-30	33	C-17, C-18, C-19, C-32	
12	C-11, C-13, C-14, C-30	34	C-20, C-21, C-22, C-35	
14	C-13	36	C-31	
14	C-13, C-15, C-16, C-30	37	C-35	
15	C-17	R ₂		
16	C-14, C-18	31-51	C-1'. C-2'	
17	C-15, C-18, C-19, C-32, C-33	C-9 OH	C-9	
20	C-21. C-34	C-19 OH	C-18, C-19, C-20	
22	C-21, C-20, C-34	C-20 OH	C-19, C-20	

Table 4. Bryostatin 14 (14) 1 H- and 13 C multiple bond correlations (HMBC) recorded at 500 MHz in CDCl₃ solution.⁴

a, Underlined positions correspond to weak signals; H-3' correlated also with C-4', 5'; H-4' with C-3', 5' and H-5' with C-3', 4'.

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the H-34 signal at δ 5.77 and to some extent the methyl signal at δ 1.12, but not at δ 1.24. In keeping with this result, irradiation of the signal at δ 1.12 (C-32 hydrogen) enhanced the signals for H-20 (δ 3.89) and H-16 (δ 5.30) whereas irradiation of the H-33 methyl signal (at δ 1.24) increased the H-17 signal (δ 5.75). Thus, the methyl hydrogen signals at δ 1.12 and 1.24 were assigned respectively to C-32 and C-33.

For reisolation of bryostatins 1 (1) and 2 (2) from a more recent (1987) recollection (~1,000 kg, damp wt) of California Bugula neritina we initiated separation of the crude extract as previously described²¹ and then devised a very useful high speed countercurrent distribution (HSCCD)⁶ technique followed by further separation using HPLC and recrystallization to yield 8.6 mg (8.6 $\times 10^{57}$ % yield) of bryostatin 15 (15, P388 ED₅₀ 1.4 μ g/mL).

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The FAB mass spectrum of bryostatin 15 gave $[M+Li]^*$ at m/z 927 corresponding to molecular formula $C_{47}H_{68}O_{18}$ (16 mass units more than bryostatin 1 at mol. wt. 904). The ¹H NMR (400 MHz) spectrum revealed a macrocyclic lactone possessing an octadienoate side chain similar to bryostatin 1 (1). But chemical shifts of hydrogen signals in the olefinic region suggested a substitution change in the octadienoate side chain. The C-4" hydrogen signal of this ester appeared at δ 6.39 (dd, J 15.5, 11 Hz) and the C-5" at δ 6.07 (dd, J 15.5, 7.5 Hz) downfield compared to their counterparts in bryostatin 1 at δ 6.16 (dd, J 8.5, 2.4 and 4.8, 1.5 Hz respectively). At other positions in the C-20 ester the 2" hydrogen signal showed a doublet at δ 5.93 (J 15.4 Hz). The 3" hydrogen appeared slightly downfield at δ 7.28 (multiplet) compared to that of bryostatin 1 at δ 7.25 (multiplet). These observations were further supported and confirmed by 2D COSY and ¹³C NMR experiments which led to a firm structure assignment for bryostatin 15 (15).

The biosynthetic processes orchestrated by *Bugula neritina* have produced a very useful series of bryostatins for detailed structure/activity studies. Whether by endogenous and/or exogenous biosyntheses, we now have a number of subtle structural modifications in hand that would be very difficult to realize by total⁷ or semisyntheses. In turn, further biological evaluation of these substances should provide important insights for future anticancer drug design.

EXPERIMENTAL

GENERAL PROCEDURES. Solvents used for column chromatography were freshly distilled. Sephadex LH-20, particle size 25-100 μ m, used in gel permeation and partition column chromatographic separations was obtained from Pharmacia Fine Chemicals AB, Uppsala, Sweden. The P.C. Inc. Ito-Multilayer Coil Separator-Extractor Model 1 was employed for high speed countercurrent distribution (HSCCD). An FMI lab pump Model RP SYX (Fluid Metering Inc., Oyster Bay, N.Y.) delivered the mobile phase and fractions were collected using Gilson FC-220 race track and FC-80 microfractionators. Thin layer chromatography silica gel plates were obtained from Analtech, Inc. The TLC plates were viewed under shortwave UV-light and then developed by 20% sulfuric acid and/or anisaldehyde-acetic acid spray reagent followed by heating at approximately 150°C. Uncorrected melting points were observed with a Kofler-type mp apparatus. Optical rotations were determined employing a Perkin-Elmer Model 241

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polarimeter. IR spectra were recorded with a Nicolet MX-1 FT-IR Spectrometer and UV spectra were obtained by using a Hewlett-Packard 8450 UV-VIS spectrometer. In high pressure liquid chromatography separations Phenomenex Prepex (particle size 5-20 μ , ϕ 10.0 mm x 25 cm) C-8 was used in reversed phase mode and Phenomenex Prepex (particle size 5-20 μ , ϕ 10.0 mm x 25 cm) silica gel was used in normal phase mode using Altex (Model 110A) solvent metering pumps and Gilson HM UV detection at 254 nm. The ¹H NMR, ¹³C NMR, 2D COSY, ¹H-¹³C correlation, and NOE were recorded with a Bruker AM-400 instrument equipped with cryomagnet and ASPECT-3000 computer. The HMBC data were recorded using a Varian 500 NMR spectrometer. Mais spectra (70 eV and FAB) were obtained employing a Kratos MS-50 spectrometer.

Bryostatin 14 (14). Approximately 1,000 kg (damp wt.) of Bugula neritina was recollected in May, 1986 in the Gulf of Mexico near Florida, USA. The animal was preserved in 2-propanol and subjected to separation as previously discussed for a 50 kg, 1984 collection.²¹ The methylene chloride fraction from the solvent partition sequence was diluted with a mixture of ethyl acetate-2-propanol-water (12:1:6, 55 L). Separation of the organic phase (33.6 L) yielded 906.5 g of P388 lymphocytic leukemia cell line active fraction which was subjected to a series of steric exclusion and partition column chromatographic steps using Sephadex LH-20 and silica gel, similar to those previously described.²² Further purification was performed with high speed countercurrent distribution employing hexane-ethyl acetate-methanol-water (3:7:5:5), with the upper layer as mobile phase and lower layer as stationary phase (detailed below for bryostatin 15) followed by HPLC using hexane-isopropanol (9:1) in normal phase and methanol-water (4:1) in reversed phase. By these techniques the known bryostatin 4 (306 mg, 3.06 x10⁻⁵% yield),²⁴ bryostatin 5 (187 mg, 1.87 x 10⁻⁵¹ yield),²⁴ bryostatin 6 (32.5 mg, 3.25 x 10⁻⁶¹ yield),²¹ bryostatin 7 (3.1 mg, 3.1 x 10⁻⁷% yield),²¹ bryostatin 8 (23.5 mg, 2.35 x 10⁻⁶% yield),^{2f} and bryostatin 10 (39.0 mg, 3.9 x 10⁻⁶% yield)^{2h} were obtained accompanied by 102 mg $(1.02 \times 10^{-5}$ yield) of the new bryostatin 14 (14) as an amorphous powder: mp 174-176°C, $[\alpha]^{22}_{D} = +41.3^{\circ}$, (c 0.92, CH₂Cl₂); HRFABMS (3-NBA/LII as matrix), found, 831.4346 (calc. for C₄₂H₆₄O₁₆Li 831.4354); EIMS 70 eV, m/z, 806[M-H₂O]* (19%), 788[806-H₂O⁺] (100%), 774[806-CH₃OH]⁺ (80%), 770[788-H₂O)]⁺ (33%); FABMS, m/z, 831[M+Li]⁺ for mol. wt. 824 corresponding to $C_{42}H_{64}O_{16}$; IR (thin film) ν_{max} cm⁻¹: 3460 (OH), 1730 (COO), 1650 (C=C), 1170 (COOR); NMR (¹H and ¹³C) appear in Tables 2-4; and P388 $ED_{50} = 0.33 \ \mu g/ml$.

Acetylation of bryostatin 5 (5). To bryostatin 5 (5, 0.8 mg) in acetic anhydride (100 μ l) was added 50 μ l of pyridine. The reaction course was monitored by TLC. After 10 hr, reaction was complete and afforded 0.8 mg of bryostatin 5 26-acetate (14a); R_f=0.61 (in 5:4 hexane:ethyl acetate), 0.90 (in 3:1:1 toluene:ethyl acetate:methanol), 0.40 (in 9:1 methanol-water, RP C-8 plate).

<u>Acetylation of bryostatin 14 (14)</u> A sample (0.8 mg) of bryostatin 14 (14) was acetylated as summarized above for bryostatin 5 (5), except for a 20 hr reaction period. HPLC yielded 0.5 mg of pure bryostatin 14 20,26-diacetate identical (by TLC and 400 MHz ¹H-NMR) with bryostatin 5 26-diacetate and exhibiting $[\alpha]^{22}_{D} + 73.2^{\circ}$ (c 0.59, CH₂Cl₂).

Bryostatin 15 (15). Bugula neritina (1,000 kg, damp wt) was recollected from the U.S. Southern California Coast in 1987. The animal was preserved in 2-propanol and this solution

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was partitioned with methylene chloride to produce a fraction (6.13 kg) that was partitioned between hexane and 9:1 methanol-water. The hexane fraction was evaporated under reduced pressure to produce 4.3 kg of hexane extract. The methanol water portion was adjusted to a concentration of 3:2 and extracted with methylene chloride. Removal of solvent from the methylene chloride fraction afforded 1.25 kg of a P388 cell line active fraction, which was subjected to a series of steric exclusion and partition column chromatographic steps using Sephadex LH-20.21 Typically, 40 to 45 g aliquots of active methylene chloride fraction were applied to Sephadex LH-20 in 1:1 methylene chloride-methanol. Fractions containing bryostatins 1 and 2 were located by TLC (95:5 methylene chloride-methanol) giving a combined weight of 348.7 g. Partition chromatography of 26 g aliquots on Sephadex LH-20 using hexane:toluene:methanol (3:1:1) provided separate fractions enriched in bryostatin 1 (37.51 g) and bryostatin 2 (14.11 g). High speed countercurrent distribution allowed further purification of bryostatins 1 and 2. A biphasic solvent system was prepared from hexaneethyl acetate-methanol-water (4.5:1.5:1:0.3). HSCCD was accomplished with the Ito horizontal flow-through coil planet centrifuge using the planet gear drive at \sim 800 rpm. The column consisted of 2.6 mm PTFE tubing (approx. volume 375 ml). The column was filled with the lower (stationary) phase of the two phase solvent system and counter-balanced. An 0.80 g aliquot (from 37.51 g of bryostatin 1 enriched fraction) dissolved in 3-4 ml of stationary phase was pumped into the column. The upper phase of the solvent system (the mobile phase) was pumped into the column from a tail to head direction with planetary motion of the column. An FMI lab pump maintained pressure at 25-30 psi. Fractions were collected and monitored by TLC. From 140 fractions (18 ml each) collected (2.5 L mobile phase) over a 5 hr period, fractions 50-120 contained bryostatin 1. The resulting bryostatin 1 fraction weighed 0.18 g. The HSCCD was repeated (45x) to give 4.4 g total of rearly pure bryostatin 1. Similar experimental procedures were followed for purification of the bryostatin 2 containing fraction. The solvent system was prepared using hexane-e-+ 1 acetate-methanol-water (4:2.5:1.2:0.5). The lower phase was stationary. After HSCCD (14x) a combined fraction (3.36 g) enriched in bryostatin 2 was obtained. Subsequent flash column chromatography (silica gel) with 1:1 hexane-ethyl acetate for bryostatin 1 separation and 9:1 ethyl acetatehexane for bryostatin 2 separation, and crystallization from ethyl acetate-hexane produced bryostatin 1 as an amorphous solid, mp 226-30°C (1.5 g, 1.5 x 10^{-4} % yield) and bryostatin 2. mp 186-187°C, (2.0 g, 2.0 x 10^{-4} % yield). The mother liquor from the bryostatin 1 crystallization was separated by HPLC with a Prepex 5-20 silica column (10 mm x 25 cm). Elution with hexane-methylene chloride-methanol (14:8:1) at a flow rate of 0.8 ml/min gave bryostatin 15 as an amorphous solid, mp 140-141°C (8.6 mg, 8.6 x 10^{-7} % yield) and bryostatin 8 (5.6 mg, 5.6 x 10^{-7} % yield). The known bryostatins 1, 2 and 8 were identified by direct comparison (principally by 400 MHz ¹H NMR and TLC) with authentic samples.

Bryostatin 15 exhibited FABMS m/z 927 [M+Li]* for mol. wt. 920 corresponding to $C_{42}H_{68}O_{16}$; [α]²⁵_D + 26° (c 0.27, CH₃OH); UV (CH₃OH) λ_{max} 227 nm (ϵ 25,995); IR (thin film) ν_{max} 3464. 2923, 2845, 1735, 1470, 1375, 1360, 1240, 1135, 1090 cm⁻¹. For the ¹H and ¹³C NMR see Tables 2 and 3.

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ANTINEOPLASTIC AGENTS, 177.¹ ISOLATION AND STRUCTURE OF PHYLLANTHOSTATIN 6

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ABSTRACT.—The isolation and structural elucidation of a new *Phyllanthus* glycoside, phyllanthostatin 6 [7], was summarized. Phyllanthostatin 6 [7] was isolated from the roots of *Phyllanthus acuminatus* (Euphorbiaceae) and was found to inhibit ($ED_{50} = 0.35 \ \mu g/ml$) growth of the murine P-388 lymphocytic leukemia cell line. Two other new constituents were shown to be didesacetylphyllanthostatin 3 [9] and descinnamoylphyllanthocindiol [10]. Structure determinations were achieved employing hrfabms and 2D-nmr spectroscopy. Application of an hplc separation technique to the *Phyllanthus* glycosides and development of a new isolation procedure for the major antineoplastic constituent, phyllanthoside [1], are also described.

The Central American tree Phyllanthus acuminatus Vahl (Euphorbiaceae) has been found to produce a new series of potentially useful antineoplastic glycosides. From 1978 to 1986, Costa Rican collections of the roots and stems of this tree were investigated; these investigations led to the isolation and structural elucidation of phyllanthoside [1], phyllanthostatin 1 [2], the related phyllanthostatins 2 [6], 3 [8], 4 [3], and 5 [4] (2,3), and two cytostatic lignans (4). Recently, three new lignans were isolated from the Indian medicinal plant Phyllanthus niruri (5). Because of strong activity against human neoplastic cell lines representing breast, CNS (TE671), colon (Colo 205), lung, ovary, and melanoma (Lox) cancers combined with curative levels of activity against the U.S. National Cancer Institute's (NCI) murine B16 melanoma, the phyllanthoside-phyllanthostatin 1 ortho acid equilibrium product has been undergoing preclinical development by the NCI Division of Cancer Treatment and is now in phase 1 clinical trial. Subsequently, Smith and colleagues completed the first total synthesis of phyllanthoside (6), phyllanthostatin 1 (7) and phyllanthostatin 2 (8). A variety of syntheses are now available for the aglycone, phyllanthocin (9).



¹For Part 176, see Kamano et al. (1).

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The present report summarizes a procedure for improving the yield of phyllanthoside [1]. In addition, a sixth member [7] of the cytostatic phyllanthostatin series and two inactive (NCI murine P-388 lymphocytic leukemia cell line, PS system) transformation products have been discovered. Earlier (2) phyllanthoside was isolated in yields ranging from $7.4 \times 10^{-4}\%$ to $1.4 \times 10^{-2}\%$. In the present study we found that extraction of the dry root with CH₂Cl₂ is an efficient and selective way to obtain a phyllanthoside-rich (corresponding to ~1% of the root) crude extract. The extract was efficiently separated by size exclusion chromatography on Sephadex LH-20 [elution with *n*-hexane-CH₂Cl₂ (1:3) and *n*-hexane-CH₂Cl₂-Me₂CO (1:3:1)] followed by high-speed countercurrent distribution (hsccd) with *n*-hexane-CH₂Cl₂-MeOH-H₂O (2:4:5:2) as solvent system. By this means phyllanthoside [1] was quickly isolated (only 4 steps) in yields as high as 0.2%, and rearrangement and degradation of glycoside 1 were minimized.

The hplc technique we previously developed for detection of phyllanthostatin A (4) was very helpful in developing the new isolation procedure for phyllanthoside. Inspection of the hplc analyses corresponding to crude CH_2Cl_2 (Figure 1a) and MeOH (Figure 1b) extracts of *P. acuminatus* illustrates this point. Both extracts displayed a large peak assigned to phyllanthoside [1], the most dominant phyllanthostatin constituent in crude extracts of *P. acuminatus*.



FIGURE 1. Hplc separation of a *Phyllanthus acuminatus* CH₂Cl₂ extract (a) and an MeOH extract (b) using RP-18 Si gel with a linear gradient of MeCN-H₂O (3:7→7:3) (photodiode array detector). Identified peaks are noted with corresponding structure numbers.

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Figure 2 illustrates the separation of a mixture of phyllanthoside [1] and its isomers 2-4. Phyllanthus glycosides 7/9 and 6/8, which coeluted (Figure 1) on an RP-18 hplc column, were easily separated on RP-8 (aqueous MeOH). However, the latter system was less powerful for the separation of complex samples such as total extracts. Finally, a freshly prepared CH_2Cl_2 extract of the original *P. acuminatus* roots (collected in 1978) was analyzed as just described. Again, large amounts of phyllanthoside and only traces of phyllanthostatin 1 were detected. Compared to a 1986 sample, the chrematogram (not shown) obtained from the 1978 sample displayed a much larger peak at Rt 6.7 min, assigned to phyllanthostatin 3 [8], as well as additional peaks between Rt 4 and 6 min (nonidentified degradation products).



FIGURE 2. Hplc separation of phyllanthoside [1] isomers on a column of RP-8 Si gel using a MeOH-H₂O (2:3→9:1) gradient.

Phyllanthostatin 6 [7] was isolated from a fresh 1986 MeOH extract of *P. acuminatus*. The MeOH extract (22 g) was separated by size exclusion chromatography (Sephadex LH-20; MeOH), affording a phyllanthostatin-6-rich fraction (4.95 g), which was further purified by hsccd (4, 10–12) using a CH_2Cl_2 -MeOH- H_2O (5:5:3) solvent system. Semi-preparative reversed-phase hplc finally afforded 12 mg of phyllanthostatin 6 [7] ($3.7 \times 10^{-3}\%$ yield, PS ED₅₀ 0.35 µg/ml). Didesacetylphyllanthostatin 3 [9] and descinnamoylphyllanthocindiol [10] were both isolated from a fraction prepared during an earlier (1983) large-scale isolation of the phyllanthostatins (2). Separation of these relatively polar *Phyllanthus* constituents was accomplished by a size exclusion (Sephadex LH-20), hsccd, and reversed-phase liquid chromatographic sequence. The pure compounds 9 (105 mg) and 10 (0.75 g) were found to be inactive against the PS cell line. Structural determinations were conducted as follows.

Phyllanthostatin 6 [7] showed spectroscopic (uv, ir, nmr) properties similar to those of the phyllanthostatins. Acid hydrolysis of the glycoside 7 afforded two hexoses with the same tlc mobility as glucose and 6-deoxyglucose. By hrfabms the molecular formula was established as $C_{36}H_{48}O_{16}$. The ion observed at m/z 613 [(M + Na) – 146]⁺ indicated that deoxyglucose was the terminal hexose of the disaccharide moiety of phyllanthostatin 6. Both the ¹H- and ¹³C-nmr chemical shifts were assigned based on 2D nmr experiments (¹H, ¹H-COSY and ¹H, ¹³C-COSY). Chemical shifts of the cinnamoyl-sequiterpene moiety were identical to those previously assigned to the phyllanthostatin aglycone (2). For example, the epoxide was confirmed by carbon resonances at δ 49.80 (C-14), 71.02 (C-7), and 102.08 (C-8) and by the C-14 protons at δ 2.93 ppm.

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From nmr and ms spectra it became apparent that the phyllanthostatin 6 disaccharide was not acetylated. The disaccharide proton resonances were fully assigned by ¹H, ¹H-COSY and double-quantum filtered phase-sensitive COSY experiments (13, 14). Figure 3 shows the sugar resonances between δ 2.9 and 4.2 ppm with the corresponding correlation peaks (double-quantum filtered phase-sensitive COSY spectrum). Interpretation of the latter spectrum compared to a normal ¹H, ¹H-COSY was simplified by less overlapping of the correlation peaks on and close to the diagonal. The C-14 protons, for example, appeared as a very weak signal, whereas the normal ¹H, ¹H-COSY spectrum showed a prominent signal at δ 2.93 ppm. Complete correlation between sugar protons was observed from S-1 through S-6 and from S-1' through S-6', respectively. The ¹H, ¹³C-COSY spectrum showed that the S-2 proton was correlated to the carbon resonance at δ 82.15 ppm typical of glycosylation at this position and confirming glucose as the inner sugar. Chemical shifts assigned to the terminal sugar were typical of 6-deoxy-D-glucose (2). The coupling constants (J = 8 Hz) of both anomeric



COSY spectrum of phyllanthostatin 6 [7] carbohydrate moiety (400 MHz, CDCl₃). 1409

protons confirmed the β linkage of the 2-0-(6-deoxy-D-glucopyranosyl)-D-glucopyranosyl unit. The anomeric proton and carbon of the inner glucose unit displayed chemical shifts (δ 5.48/92.40 ppm) identical to those observed with phyllanthostatin 3 [8]. Overall assignment of the phyllanthostatin 6 [7] chemical shifts were in agreement with those reported (2) for the other phyllanthostatins. Thus, structure 7 was assigned to this new member of the series.

The molecular formula of didesacetylphyllanthostatin 3 [9] was determined by hrfabms to be $C_{36}H_{50}O_{16}$. Except for ¹³C-nmr chemical shifts recorded for C-7 (85.33 ppm), C-8 (106.31 ppm), and C-14 (66.61 ppm), indicating a diol unit at C-7–C-14 (2), glycoside 9 displayed spectroscopic properties similar to those of phyllanthostatin 6 [7]. The ¹H- and ¹³C-nmr resonances assigned to positions 1–15 and 1'–9' were in accord with those of phyllanthostatin 3, but chemical shifts of the sugar moiety were more typical of a β -linked phyllanthose unit and indeed agreed with the corresponding data for didesacetylphyllanthoside [5] (2). Hence, this component was assigned to didesacetylphyllanthostatin 3 [9].

Descinnamoylphyllanthocindiol [10] gave the same tlc color reaction (browngray-pink after 24 h) upon development with anisaldehyde, as observed with the phyllanthostatins. Lack of uv absorption suggested absence of the cinnamoyl ester, and the molecular formula (by hrfabms), $C_{15}H_{24}O_7$, suggested lack of a disaccharide unit (confirmed by nmr analyses). Assignments for the ¹H and ¹³C chemical shifts were achieved by 2D nmr techniques and indicated a phyllanthocindiol (2) analogue with a hydroxyl at C-10 and a carboxyl group at C-3. Compared to glycoside 9, carbon resonances C-9 to C-11 of glycoside 10 appeared at higher (C-11 at δ 34.22, C-9 at δ 36.76 ppm) and lower (C-10 at δ 68.22 ppm) fields in agreement with a hydroxyl group at C-10. Structure 10 was thereby identified as descinnamoylphyllanthocindiol.

Both the improved procedure herein summarized for isolation of phyllanthoside and its useful total synthesis (6,7) have diminished the problem of future supplies of this substance and the isomeric phyllanthostatin 1. Isolation of phyllanthostatin 6 appears to complete the series of principal antineoplastic and/or cytostatic glycosides produced by *P. acuminatus*.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES .- All hplc-grade solvents (Omnisolv) were obtained from EM Science, and all other solvents were redistilled. Adsorption cc was performed with Si gel 60 (70-230 mesh, E. Merck, Darmstadt, Germany). Reversed-phase Si gel chromatography was accomplished with RP-8 Lober columns (size B, 40-63 µm, E. Merck) and size exclusion chromatography with Sephadex LH-20 (particle size: 25-100 µm) supplied by Pharmacia Fine Chemicals, Uppsala, Sweden. Tlc was carried out with Si gel GHLF Uniplates (Analtech Inc.). The tlc plates were examined under uv light and developed with an anisaldehyde spray reagent. High-speed countercurrent distribution (hsccd) was performed with an Ito Multilayer Coil Extractor-Separator (P.C. Inc., Potomac, Maryland) using 2.6 mm i.d. tubing, FMI Lab Pump, Linear recorder, and Gilson Model Holochrome uv/vis detector (2.5 mm/3.2 µl cell) with a Micro Fractionator. The hplc-uv/vis separations were accomplished with Ultremex 3 µm RP-8 and RP-18 columns (100×4.6 mm i.d.; Phenomenex, Rancho Palos Verde, California). The mobile phase was delivered by two Gilson Model 302 pumps using an Apple II programmer and a Rheodyne 7161 injector with a 0.5 µm in-line precolumn filter (Rainin). Linear gradient elution was carried out with MeCN-H₂O (3:7->7:3) (RP-18) and MeOH-H₂O (2:3->9:1) (RP-8) within 15 min at a flow rate of 1 ml/ min. All pure compounds for hplc analyses were dissolved in MeOH (~0.1 mg/ml). Roots of P. acuminatus (1 g powder) were extracted at room temperature with MeOH or CH_2Cl_2 (3 × 30 ml solvent). The MeOH extract (78 mg) and the CH₂Cl₂ extracts (16 mg of 1978 samples and 8 mg of 1986 specimens) were dissolved or suspended in 2 and 1 ml of MeOH, respectively. These solutions were each passed through a Sep-Pak C-18 cartridge (Waters). The cartridges were washed with MeOH until 4 ml (MeOH extract) and 2 ml (CH2Cl2) elustes were collected. These solutions (10 µl) were injected for hplc analyses. Signals were identified by comparison of retention times with those of authentic samples by co-injection. More than 20 nonidentified minor peaks with uv spectra typical of the phyllanthostatins were observed. The same results were obtained when crude extracts were separated on RP-8 Si gel with MeOH-H2O (2:3-9:1) as eluent.

Melting points are uncorrected and were determined using a Kofler-type hot-stage apparatus. Optical rotations were measured with a Perkin-Elmer Model 241 Automatic Polarimeter. The uv spectra were recorded employing a Hewlett-Packard Model 8450A uv/vis spectrophotometer and ir spectra with a Nicolet Ft-ir Model MX-1 instrument. Nmr spectra were measured using a Bruker AM-400 instrument and are recorded in ppm downfield to TMS. Assignments bearing the same superscript may be reversed. The ¹³C-nmr multiplicities were determined with APT experiments based on an average coupling constant of 135 Hz. Normal 2D homonuclear and heteronuclear shift correlated spectra were recorded using standard pulse sequences (15–17). The double-quantum filtered phase-sensitive COSY experiment was pursued following the procedure of Wuethrich and co-workers (13, 14). Eims spectra were obtained using a Varian MAT 312 spectrometer. Fabms spectra were recorded with a MS-50 instrument at the NSF Regional Facility, University of Nebraska, Lincoln.

PLANT MATERIAL AND EXTRACTION.—The 1986 collection of *P. acuminatus* roots was obtained in Costa Rica, and a voucher specimen is preserved in our Institute. The dry powdered roots (345 g) were extracted at room temperature successively with *n*-hexane, CH_2Cl_2 , and MeOH $(3 \times 6$ liters each solvent), yielding 0.6, 3.6, and 25.0 g extracts, respectively. Extraction, solvent partitioning, and chromatographic separation of *P. acuminatus* (1978 collection) were performed as described by Pettit *et al.* (2).

ISOLATION OF PHYLLANTHOSIDE [1].—The 1986 CH_2Cl_2 extract (3.6 g) was separated on a column of Sephadex LH-20 (60 × 4 cm i.d.) with *n*-hexane– CH_2Cl_2 (1:3) as initial solvent system. After eluting with 2.5 liters, the solvent was changed to *n*-hexane– CH_2Cl_2 -Me₂CO (1:3:1), and a phyllanthosiderich (0.833 g) fraction was eluted. An aliquot of this fraction (106 mg) was purified by hsccd with the solvent system *n*-hexane– CH_2Cl_2 -Me₂CO (1:3:1), and a phyllanthosiderich (0.833 g) fraction was eluted. An aliquot of this fraction (106 mg) was purified by hsccd with the solvent system *n*-hexane– CH_2Cl_2 -MeOH– H_2O (2:4:5:2). The sample was dissolved in a mixture (4 ml) of stationary and mobile phase (6:1) and introduced in the coil through the head inlet. The coil was rotated at 800 rpm, and the mobile phase (lower layer) was pumped at a flow rate of 200 ml/h. Detection (uv) was set at 280 nm and fractions collected every 1.5 min. Retention of the stationary phase was 45%. Pure (by hplc) phyllanthoside (93 mg) was obtained (elution volume 265–335 ml) and its identity confirmed by comparison with an authentic sample (ir, ¹H nmr, and ¹³C nmr).

ISOLATION OF PHYLLANTHOSTATIN 6 [7]. - The MeOH extract (22 g) of P. acuminatus (1986 collection) was separated by size exclusion chromatography on Sephadex LH-20 (100 × 10 cm i.d., MeOH), and 7 fractions were collected. Fraction 5 (4.95 g, elution volume 6150-6700 ml) was further separated by haced with the solvent system CH2Cl2-MeOH-H2O (5:5:3). Samples (2 g) were dissolved in a mixture of ise haced was conducted with the lower, organic phase and a flow upper (13 ml) and lower (2 ml) phase rate of 200 ml/h. A uv detector was so, at 280 nm and fractions collected every 1.5 min. Fractions eluted between 120 and 150 min after sample introduction were combined and afforded 38 mg of almost pure phyllanthostatin 6 [7]. Combined fractions were purified by semi-preparative hplc (RP-8, Prepex 5-20 μ m, 250 × 10 mm, Phenomenex) with aqueous MeOH-H₂O (3:7) at a flow rate of 2 ml/min, yielding 12 mg (3.7 × 10⁻³% yield) of phyllanthostatin 6 [7]: amorphous solid; mp 136–139°; tlc on Si gel $R_f 0.12$ $[CH_2Cl_2-MeOH (9:1)], R_{c} 0.25 [CH_2Cl_2-MeOH-H_2O (5:5:3) (lower phase)]; [\alpha]^{25}D + 12.0° (c = 0.25, 0.25) [CH_2Cl_2-MeOH (9:1)], Ch_2Cl_2-MeOH (9:1)], Ch_2Ch_2Ch_2-MeOH (9:1)], Ch_2Ch_2-MeOH (9:1)], Ch_2-MeOH (9:1)]$ CH_2Cl_2 ; hrfabms $m/z [M + Na]^+$ 759.2839 (calcd for $C_{36}H_{48}O_{16}Na$, 759.2840) with $\Delta = 0.1$ ppm, [(M + Na) - 146]⁺ 613; uv & max (MeOH) 277 nm; ir (KBr) v max 3422, 2940, 1745, 1707, 1635, **1450**, 1315, 1281, 1169, 1123, 1075, 1021 cm⁻¹; ¹H nmr (CDCl₃) δ 0.83 (3H, d, J = 6 Hz, H-15), 1.20 (3H, d, J = 6 Hz, S-6'), 1.27 (2H, m, H-2), 1.57 (H-1), 1.63 (H-9), 1.76 (H-4), 1.91 (H-9), 1.94 (H-11), 1.98 (H-1, H-6), 2.32 (H-4), 2.50 (H-3), 2.93 (2H, br s, H-14), 2.97 (S-2), 3.04 (S-4'), 3.25 (\$-5'), 3.30 (\$-2'), 3.37 (\$-5), 3.43 (H-12), 3.44 (\$-3'), 3.46 (\$-4), 3.66 (\$-3), 3.75 (\$-6), 3.98 (1H, dd, J = 11.5 Hz, H-12), 4.13 (1H, d, J = 8 Hz, S-1'), 4.42 (H-5), 5.14 (H-10), 5.48 (1H, d, J = 7.7 Hz, S-1), 6.56(1H, d, J = 16.1 Hz, H-2'), 7.39(3H, brs, H-5', H-7', H-9'), 7.56(2H, brs, H-6', H-8'), 7.78(1H, d, J = 16.1 Hz, H-3'); ¹³C nmr (CDCl₃) δ 12.73 (q, C-15), 17.76 (q, S-6'), 21.84 (t, C-1), 25.62 (t, C-2), 29.47 (t, C-4), 33.14 (d, C-11), 34.31 (t, C-9), 37.04 (d, C-3), 38.21 (d, C-6), 49.80 (t, C-14), 61.64 (t, S-6), 62.79 (t, C-12), 69.15 (d, S-4), 69.70 (d, C-10), 71.02 (s, C-7), 72.16 (d, S-5')*, 72.61 (d, C-5), 74.87 (d, S-2'), 75.15 (d, S-4'), 76.03 (d, S-3')^b, 76.10 (d, S-3), 76.30 (d, S-5)^b, 82.15 (d, S-2), 92.40 (d, S-1), 102.08 (s, C-8), 104.50 (d, S-1'), 118.57 (d, C-2'), 128.29 (2 × d, C-6', C-8'), 129.24 (2×d, C-5', C-9'), 130.61 (d, C-7'), 134.34 (s, C-4'), 145.01 (d, C-3'), 166.99 (s, C-1'), 174.38 (s, C-13).

HYDROLYSIS.—A solution of phyllanthostatin 6 [7] (2 mg) in MeOH (2 ml) and 2 N HCl (10 ml) was heated at reflux for 30 min, diluted with H₂O, and extracted with CHCl₃. The aqueous phase was neutralized (NaHCO₃), the solvent was evaporated, and the sugars were extracted with pyridine. Glucose and 6-deoxyglucose were detected in the extract by tlc on Si gel using the solvent system EtOAc-MeOH-H₂O-HOAc (65:15:15:30) followed by spraying with anisaldehyde reagent and heating to reveal spots at R_f 0.58 and R_f 0.70 characteristic of D-glucose and 6-deoxy-D-glucose, respectively. ISOLATION OF DIDESACETYLPHYLLANTHOSTATIN 3 [9] AND DESCINNAMOYLPHYLLANTHOCIN-DIOL [10]. —A fraction obtained from an earlier large-scale isolation of phyllanthoside (2) was separated by size exclusion chromatography on Sephadex LH-20 in MeOH (100 × 10 cm i.d.; 100 g and 92 g samples) yielding 11 fractions. Part (6 g) of the major fraction (111 g; elution volume 4550–5925 ml) was further separated by hsccd with the solvent system CH₂Cl₂-MeOH-H₂O (5:5:3). The organic layer was used as mobile phase and was passed at a flow rate of 400 ml/h. Retention of the stationary phase was about 50%. Samples (3 × 2 g) were dissolved in a 20 ml mixture of both phases, and fractions were collected every minute. Fractions eluted between volumes 150 and 250 ml were combined (0.36 g) and further purified by reversed-phase liquid chromatography with MeOH-H₂O (3:2 \rightarrow 7:3) (Lobar RP-8, size B) to afford 105 mg of didesacetylphyllanthostatin 3 (7 × 10⁻⁶% yield). Another aliquot (6.3 g) of the main fraction was separated by reversed-phase liquid chromatography with MeOH-H₂O (1:3) (Lobar RP-8, size B, 3 × 2.1 g samples). Fractions containing diol 10 were combined in MeOH solution and further purified on a column of Sephadex LH-20, yielding 0.75 g of descinnamoylphyllanthocindiol [10] (5 × 10⁻⁵% yield).

Didesacetylphyllanthostatin 3 [9] was isolated as an amorphous solid: mp 135-139°; tlc on Si gel R 0.08 [CH₂Cl₂-MeOH (9:1)]; [α]²³D +9.1° (c = 0.11, CH₂Cl₂); hrfabms m/z [M + Li]⁺ 745.3236 (calcd for $C_{36}H_{50}O_{16}Li$, 745.3260), $\Delta = 3.3$ ppm; uv λ max (MeOH) 277 nm; ir (KBr) ν max 3433, 2940, 1745, 1707, 1635, 1445, 1309, 1281, 1233, 1169, 1117, 1074 cm⁻¹; ¹H nmr (CDCl₃)δ0.85 (3H. d. J = 5.6 Hz), 1.21 (3H, d, J = 5.5 Hz, S-6'), 1.25 (3H, d, J = 5.5 Hz, S-6)^a, 1.32 (H-2), 1.38 (H-1), 1.59 (H-1), 1.73 (H-4), 1.82 (H-6), 1.94 (H-9, H-11), 2.02 (H-2), 2.14 (H-9), 2.17 (H-4), 2.51 (H-3), 3.00 (S-2), 3.04 (S-4), 3.05 (S-4'), 3.23 (S-5')^b, 3.25 (S-2'), 3.38 (S-5)^b, 3.42 (S-3'), 3.49 (H-12, H-14), 3.57 (S-3), 3.93 (H-14), 4.01 (H-12), 4.16 (1H, d, J = 7.7 Hz in C₅D₅N, S-1'), 4.18 (H-5), 5.13 (H-10), 5.45 (1H, d, J = 8.1 Hz in C₅D₅N, S-1), 6.50 (1H, d, J = 15.8 Hz, H-2'), 7.46 (H-5', H-7'), 7.55 (H-6', H-8'), 7.75 (1H, d, J = 15.8 Hz, H-3'); ¹³C nmr (CDCl₃) δ 12.67 (q, C-15), 17.61 (q, S-6')^e, 17.82 (q, S-6)^a, 20.47 (t, C-1), 26.11 (t, C-2), 29.47 (t, C-4), 33.21 (d, C-11), 35.32 (t, C-9), 36.87 (d, C-3), 43.22 (d, C-6), 62.77 (t, C-12), 66.21 (t, C-14), 70.06 (d, C-10), 72.10 (d, S-2'), 72.70 (d, S-5)^b, 72.83 (d, C-5), 74.58 (d, S-5')^b, 75.07 (2d, S-4, S-4'), 75.88 (d, S-3'), 76.23 (d, S-3), 81.70 (d, S-2), 85.33 (s, C-7), 92.28 (d, S-1), 104.15 (d, S-1'), 106.31 (s, C-8), 118.56 (d, C-2'), 128.32 (2 × d, C-6', C-8'), 129.22 (2×d, C-5', C-9'), 130.52 (d, C-7'), 134.35 (d, C-4'), 145.15 (d, C-3'), 167.22 (s, C-1'), 174.72 (s, C-13).

Descinnamoylphyllanthocindiol [10] was obtained as an amorphous solid: mp 60–65°; tlc on Si gel R_f 0.18 [CH₂Cl₂-Me₂CO-H₂O (20:80:5)]; [α]²³D +92° (c=0.25, MeOH); hrfabms m/z 323.1688 [M + Li]⁺ (calcd for C₁₅H₂₄O₇Li, 323.1683), Δ = 1.5 ppm; ir (KBr) ν max 3456, 2954, 1707, 1455, 1417, 1390, 1121, 1082, 1040, 1022, 985 cm⁻¹; ¹H nmr (CDCl₃) δ 0.90 (3H, d, J = 6.8 Hz, H-15), 1.30–1.39 (H-2), 1.51–1.60 (H-1), 1.72–1.82 (H-1, H-4, H-6, H-9, H-11), 2.05–2.13 (H-2, H-9), 2.18–2.22 (H-4), 2.59–2.65 (H-3), 3.44 (1H, d, J = 11.5 Hz, H-14), 3.51 (H-12), 3.79 (H-12), 3.86 (H-10), 4.00 (1H, d, J = 11.5 Hz, H-14), 4.20 (H-5); ¹³C nmr [CDCl₄-MeOD (9:1)] δ 12.56 (q, C-15), 20.21 (r, C-1), 25.74 (r, C-2), 29.33 (r, C-4), 34.22 (d, C-11), 36.76 (r, C-9), 36.84 (d, C-3), 42.71 (d, C-6), 62.03 (r, C-12), 65.19 (r, C-14), 68.22 (d, C-10), 74.01 (d, C-5), 84.11 (s, C-7), 107.52 (s, C-8), 180.00 (s, C-13, in MeOD).

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APPENDIX A

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Table I.	Prescreen Submissions for Grant Period
Table II.	Prescreen Actives (2/6/89 - 10/31/90)
Table III.	Prescreen Actives (11/1/90 - 7/5/91)
Table IV.	Full Screen Submissions for Grant Period
Table V.	Special Sample Submissions for Grant Period

Table I

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Prescreen Submissions for Grant Period

All plants and marine animals listed represent crude extracts unless otherwise noted.

Date	<u>Total</u>	<u>Plants</u>	<u>Marines</u>	<u>Mycelliums</u>	<u>Synthetics</u>
In BRFF Repository	1770	81	1401	288	
11 II	161		19* 142		
5/22/90	144		144		
6/26/90	183	21	162		
8/20/90	152	32	120		
8/23/90	111		111		
9/21/90	200	3	197		
10/11/90	200	27	173		
11/13/90	200	196	4		
11/29/90	189	69	120		
12/13/90	161		161		
1/15/91	15				15
1/23/91	195		195		
2/4/91	30	30*			
2/19/91	193	54	139		
2/28/91	191	191			
3/13/91	298	255	43		
3/27/91	200		200		
4/10/91	198	3	195		

<u>Date</u>	Total	<u>Plants</u>	<u>Marines</u>	<u>Mycelliums</u>	<u>Synthetics</u>
4/24/91	197	37	160		
5/8/91	211	103	9* 99		
5/22/91	200	200			
6/6/91	200	140	60		
6/21/91	200	200			
	<u> </u>		-		
	5,799	1,642	3,854	288	15

* = fractions

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Summary:	1,612	Plant Crudes
-	30	Plant Fractions
	3,826	Marine Crudes
	28	Marine Fractions
	288	Mycelliums
	15	Synthetics

1420 of the total submissions showed activity in one or more of the viruses and have been submitted for full screen testing. This figure equals 24% of the total number of samples submitted for prescreen testing during the grant period.

Table II

Prescreen Actives (2/6/89 - 10/31/90)

Drug_Num	AVS_Num	Virus
B23-11C	4855	PT.YF
B720914	7332	YF
B720916	7099	PT
B720933	8513	VEE
B720937	7100	PT
B720940	7101	РТ
B720941	7102	PT
B720951	7103	PT
B720952	7104	PT
B720958	7105	PT
B720963	7106	PT
B720979	7107	PT
B720998	7349	PT
8721000	7350	PT
B721011	7298	PT
B721021	7299	PT
8721031	/368	PT
8721045	/3/3	PT VF
8721043	7375	PT YF
B721053	7377	YF
B721054	7378	YF
B721055	7379	PT
B721060	7300	PT
B721061	7301	PT
B721062	7302	PT
8721063	7383	PT
B721064	7303	PT
8721092	1312	PT
B721095	1315	PT DO MO
B/21166	1908	PI,IP
N/211/3	7108	ነ <i>г</i> թኖ
B/211// B721178	7386	PT
B721257	7390	YF
B721260	7391	рт
B721295	8309	PT
B721348	8311	PT
B721374	7304	PT
B721377	7305	PT
B721378	7306	PT
H/21392	/30/	רו זיל עתי
8/21011	1001	r + , i r D7
8/21032	1010	
8/61300 8791609	103U 7109	PT, TP
8721372 8721404	7300	г 1 рт
8721596	7310	PT
B721604	8321	PT
R721611	7311	PT
B721616	7399	PT
B721628	7403	PT

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Drug_Num	n AVS_Num	Virus
B721743		PT
. B721749	6584	PT
B721787	6585	PT
B721818	6586	PT
B721823	6248	PT,YF
B721826	6587	PT
B721838	6588	PT
B721880	6589	PT
B721892	6590	PT
B721899	6591	PT,YF
B721905	6592	PT
B721908	6593	PT
B721910	6594	РТ
. B721917	6595	PT
B721925	6596	PT
B721953	6597	PT,YF
B721958	6598	PT
B721979	6599	PT
B722006	6600	PT
B722048	6601	РТ
B722052	6602	PT
B722054	6603	PT
B722060	9180	PT
B722076	6604	PT
B722077	6605	PT
B722078	6606	PT
B722080	6607	PT
B722081	6608	PT
B722087	6609	PT,YF
B722089	6610	PT
B722091	6611	PT
B722094	6612	PT
B722109	6613	YF
B722111	6614	YF
8722116	9183	PT , YF
B722117	9184	PT
B722141	6615	PT
B722162	6616	PT
B722165	6617	PT,YF
B722168	6618	PT
B722181	6619	
B/22182	6620	P1,IF
B/22183	6621	PT, IF
B/22222 B700004	6622	PT
B722224 B700009	0023 6624	P1 9 T
B722220	6625	PT
87777710	6626	•• рт Vг
R700061	6627	PT VP
R700046	6628	PT YF
B722247	5629	PT.YF
R700070	8238	PT
R7999RA	8230	рт
5,22200	0237	

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Drug_Num	AVS_Num	Virus
B722518	7109	YF
B722525	7110	рт
B722559	8240	PT
B722591	8241	PT
B722607	8242	PT
B722628	8244	PT
B722632	8245	PT
B722634	8246	PT
B722689	8228	PT
B722743	8250	PT 200
B722745	8251	PI
B722752	7405	P1 PT
B722805	7111	P1 PT
B722808	7112	<u> </u>
B722811	/113	ይፐ
B722823	8271	VF
B722824	/114	PT
B722849	7113	PT
B/22854	7400	PT
B/2286/	7312	Tq
B/220/1	7314	PT
B/220/4	7315	PT
B122003	7316	PT
B/22000	7317	PT
B/22003	7318	PT
B722904	7116	PT
B723044	7117	PT
B723061	7443	VEE
B723062	7445	PT
B723096	7439	VEE
B723103	7442	VEE
B723106	7408	PT
8723110	7410	PT
B723123	7438	YF
B723136	7446	VEE
B723141	7440	YF,VEE
B723148	7441	VEE
B723203	7444	VEE
B723247	8201	VEE
B723250	8212	V МБ РТ
B723268	8216	VEE
8723275	8217	PT
B723278	8218	PT
B723280	0000	VEE
B723286	8220	UUF
B723289	0221	PT_VEE
B723290	8222	DT
B723297	8223	r i Ver
B723315	8226	vee VFF
8723318	822/	VEE
8777779	8210	7 LJ M

Drug_Num	AVS_Num	Virus
B723409	8204	VEE
B723412	8206	YF
B723414	8208	VEE
B724373	6630	PT
B724379	6631	PT
B724382	6632	PT
B724384	6633	PT
B724385	6634	PT
B724387	6635	PT
B724394	6636	PT
B724396	6637	PT
B724405	6638	PT
B724406	6639	PT
B724411	6640	PT
B724413	6641	PT
B724415	6642	PT
B724416	6643	PT
B724417	6644	PT
B724418	6645	PT
B724420	6646	PT
B724423	6647	PT
B724433	6648	PT
B724434	6649	PT
B724436	6650	PT
B724439	6651	PT
B724442	6652	PT ,YF
B724447	6653	PT
B724453	6654	PT
B724455	6655	PT
B724456	6656	PT
B724457	6657	PT
B724458	6658	PT
B724466	6659	PT,YF
B/24468	6660	YF
B/24508	6661	PT
8724509	6662	P.1.
B/24512	6663	PT
B/2451/	6664	P1
B/24519	6665	PT
8/24521	6666	PT DT
B/24323	6667	
B/24220	6666	Г.L рт
8724520	6609	DT VE
B724JJU B704535	6670	11,11 DT
0/24333 979/6//	00/1 4679	Г.L DT
D/24244 D77660	00/2 4472	1.7
0/24347 879/863	6676	EL DT
8/24333 879/660	00/4 4675	ר ג דעם
8/24338 879/622	00/3	rT DT
0/24300 D79/602	00/0	r 1 om
D/24380	00//	F1 DT
B/2439U	00/0	11 11 11
8/24592	00/9	rī,if

Drug_Num	AVS_Num	Virus	
B724596	6680	PT	
B724607	6681	PT	
B724610	6682	YF	
B724618	6683	YF	
B724627	6684	PT	
B724633	6685	PT	
B724642	6686	РТ	
B724644	6687	PT	
8724652	6688	PT	
B724654	6689	PT	
8724657	6690	PT	
B724661	6691	PT	
8724664	6692	PT	
B724667	6693	PT	
B724670	6250	PT	
B724697	6251	PT	
B724698	6252	PT	
B724701	6253	YF	
B724712	6254	PT	
B724714	6255	PT	
B724716	6256	PT	
B724719	6257	PT	
B724720	6258	PT	
B724722	6259	PT	
B724724	6694	PT	
B724728	6695	PT	
B724729	6696	PT	
B124732	6697	PT	
B724740	6261	PT	
B724762	6262	PT	
B724764	6263	PT	
B724769	6264	PT	
B724772	6698	YF	
B724781	6265	PT	
B724783	6699	PT	
B724785	6700	PT	
B724797	6266	PT	
B724812	6701	YF	
B724820	6268	PT	
8724825	6269	PI PT	
B724832	6702	VF	
B724844	6270	VE	
8724852	6272	1F	
8/24855	6703	PT	
D/2486U	6274	PT	
B724863	6275	PT,YF	
8/24866	6704	PT PT	
B724885	6/05	rT pm	
B724886	62//	ri Ve	
8724898	6706	I.F DT	
DG48989 8848989	8238 8350	Г L DT	
D048990	8239	ri DT	
0047033	0234	E T	

AVS_Num	Virus
8236	PT
8237	PT
4276	YF
4279	PT,YF
4280	PT,YF
7320	PT
7321	YF
7414	PT
7415	PT
7417	PT
7418	PT
7424	PT
	AVS_Num 8236 8237 4276 4279 4280 7320 7321 7414 7415 7417 7418 7424

Table III

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Prescreen Actives (11/1/90 - 7/5/91)

Virus	<u>Ctrl. B No.</u>	AVS No.
рт	604736-F046	11029
PT YF	604736-F047	11030
рт РТ	604736-F056	
PT YF	604736-F057	11031
ቦ ግ	604736-F058	
PT	631963-F010	
PT	631963-F015	11032
PT.VEE	631963-F016	11033
PT	634131-F028	
YF.VEE	634131-F031	11034
YF, VEE	634131-F032	11035
PT	634131-F064	
PT,YF,VEE	634131-F065	11036
PT, YF, VEE	634131-F068	11037
PT,YF	642761-F016	11038
PT	642761-F017	11039
PT	642761-F018	11040
P.L	644263-F106	11041
PT,VEE	644263-F107	11042
PT	678018-F243	11043
PT,YF	678018-F244	11044
PT , YF	678018-F245	11045
VEE	678018-F246	11046
PT, YF	706269	9213
PT	706308	
РŤ	709724	
PT.VEE	709752	9206/9399
VEE	709761	9400
PT	709763	
PT.YF.VEE	710030	9207/9214/9407
PT, YF, VEE	710041	9237/9401
PT	710043	
PT	710046	
PT	710057	
VEE	710064	9402
VEE	710068	9403
VEE	710072	9404
PT	710100	9211
рт	710110	
PT	710124	
VEE	710126	9405

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PT,VEE PT	710128 710130	9406
PT	710131	
PT	/1013/	
PT	710150	
PT	710152	
PI DT	710154	
ri Ver	710157	9408
DT DT	710161	
PT.VEE	710162	9409
PT	710174	
PT	710176	
PT	710179	
PT	710183	
PT	710210	
PT VEE	711707	9410
VEE	711710	9411
PT, VEE	711711	9412
PT, VEE	711712	9413
YF	711714	
PT,VEE	711715	
YF,VEE	711716	
PT,VEE	/11/1/	
YF	/11/18	
YF,VEE	711720	
YF	711720	
Yr	/11/22	
YF	712291	
ΥF	712292	
YF	/12295	
YF	712290	
YF	712300	
Yr	/12500	
PT	714994	9121
рŤ	714997	9122
ΥF	715001	9123
VEE	715010	9124
VEE	715011	9125
VEE	715012	9126
PT	715022	912/
PT	715023	9120
PT	715026	9129
PT,VEE	12000	2130

	71 50 60	
YF	/15062	
YF	/15068	
PT,YF,VEE	715070	
YF	715074	
YF	715075	
YF	715078	
YF	715082	
YF	715089	
PT	715091	
PT VEE	715093	
VEE	715095	
DT	715101	
VEE	715111	
VEE DT VEE	715112	
FI,VEE	715113	
	715100	
	715161	11047
PT, YF, VEC	715141	
VEE	/15100	
YF	715180	
PT	/15186	
РТ	/15234	
VEE	715284	
VEE	715627	
	710057	
PT	/1825/	
VEE	/18546	
PT	718548	
PT	718550	
PT,YF,VEE	718551	
PT	718553	
PT.VEE	718574	11048
PT.YF	718577	11049
PT	718580	11050
ער דיר	718582	11051
ר ז חיד	718583	11052
	718586	11053
PI	710507	11054
PT	710507	1105.
VEE	710505	11055
PT	/18595	11055
PT	718598	11057
PT,VEE	718599	11057
РT	718603	11050
YF,VEE	718636	11059
VEE	718640	
VEE	718642	
VEE	718645	
PT,YF	718648	11060
YF	718649	11061
PT VF	718656	11062
YF VEE	718789	11063
VE VEE	718799	11064
12,700		

PT.VEE	718807	11065
VEE	718813	11066
PT.VEE	718819	11067
PT	718821	11068
VEE	718825	
PT.VEE	718838	11069
VEE	718840	11070
PT.VEE	718841	11071
PT	718842	11072
YF	718843	11073
PT	718849	11074
YF	718860	11075
PT	718862	11076
YF	718876	11077
VFF	718882	11078
VEE	718889	11079
VEE	718917	11080
VEE	71892/	11081
DT VE VEE	71903/	11082
ri, ir, vee pr	718035	11083
I L VEC	7190/0	11084
VEE	710742	11004
VEE	718954	11005
VEE	718909	11000
VEL	/189/8	11007
PT,VEE	/189/9	11088
PT,VEE	718982	11089
PT	/18983	11090
PT	718990	11091
PT	718996	11092
1100	71000/	11003
VEE DT VEE	719004	1100/
PI,VEE	719006	11094
PI, YF, VEE	719007	11000
YF	719008	11090
YF	719215	9131
YF	/19216	9132
YF	/1921/	9133
YF,VEE	719218	9134
YF	719219	9135
YF	719220	9136
YF,VEE	719221	9137
PT,YF	719222	9138
YF	719224	9139
YF	719225	9140
YF	719226	9141
YF	719228	9142
VEE	719229	9143
VEE	719230	9144
YF	719231	
РТ	719232	9145
PT	719238	9146
YF.VEE	719240	9147
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PT,YF	719241	9148
PT, YF, VEE	719242	9149
YF, VEE	719244	9150
PT,YF	719245	9151
YF	719246	9152
PT,YF	719247	9153
PT	719255	9154
PT	719256	9155
YF	719257	9156
YF	719258	9157
YF	719259	9158
PT VF	719260	9159
VEE	719265	9160
VEE	719266	9161
PT VEE	719267	9162
VFF	719272	9163
VE	719274	9164
IF	710274	9165
VEE	/192/0	9109
	720202	11097
PI,VEE	720392	11098
YF	720394	11090
VEE	720396	11100
PT,VEE	/20398	11100
VEE	720399	11101
ΥF	720400	11102
VEE	720405	11103
PT,YF	720409	11104
YF	720410	11105
YF	720412	11106
YF	720415	11107
РТ	720418	11108
YF	720419	11109
РТ	720424	11110
YF VEE	720425	11111/11112
YE VEE	720427	11113/11114
PT VE	720430	11115/11116
VE	720431	11117
VE	720433	11118
IT VEE	720435	11119
VEE	720434	11120
VEE	720435	11121
1F VD	720430	11122
	720445	11123
P1,1F	720445	0166
VEE	720820	0167
11	720821	710/
PT,VEE	720825	2100
Ϋ́F	/20828	A10A
РТ	720832	91/0
PT,VEE	720834	9171
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YF,VEE	721643	9172

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YF	721746	9173
VEE	721777	9174
YF	721778	
PT,YF,VEE	721781	
VEE	721786	9175
YF	722117	9184
PT,YF	722161	9185
YF, VEE	722174	9187
YF	722179	9188
VEE	722184	
YF	722186	9189
YF	722190	9190
YF	722194	9191
۲۲	722210	
YF	722215	9194
YF	722229	9195
VEE	722230	6625
РТ	722231	
YF	722233	9196
YF	722236	9198
PT,YF	722240	
YF	722242	9199
YF	722245	9201
PT	722246	6628
YF	722248	9202
YF	722278	
VEE	722504	9205
YF.VEE	722506	
PT.YF.VEE	722508	
YFVEE	722510	
PT.VEE	722872	9208
PT.YF	722873	9209
YF	722876	9210
PT	722883	7315/9211
YF	722884	9212
PT	722886	7316/9213
PT VEE	722889	7317/9214
VEE	722899	9216
YF VEE	722890	9215
PT	722902	8374/9217
VEE	722905	9218
VEE	722908	9219
YF	722909	
VEE	722911	9220
VEE	722914	9221
PT.VEE	722915	9222
VEE	722917	9223
PT VEE	722920	
PT.YF	722921	
VEE	722922	9224

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VEE	722926	9225
PT,VEE	722929	
VEE	722931	9226
VEE	722932	9227
YF	/22933	
VEE	722935	9228
PT,VEE	722936	9229
PT,YF	722937	
VEE	722938	9230
YF,VEE	722942	
VEE	722944	9231
VEE	722950	9232
PT,YF,VEE	722965	9233
VEE	722976	9234
VEE	722987	9235
VEE	722992	9236
VE	722003	
	723003	
ri VE	723004	
IF DT VEE	723000	
PI,VEE	723033	
PT, YF, VEE	723047	0238
PI, IF	723109	9230
PT	/231/2	9239
VEE	723420	9240
VEE	/23424	9241
VEE	723425	9242
VEE	/23420	9243
YF, VEE	/2342/	
PT,YF	723428	0244
VEE	/23429	9244
VEE	723430	9245
PT,VEE	723431	00/6
VEE	723432	9246
YF	723435	9247
YF	723436	9248
YF	723438	9249
YF	723441	9250
YF	723444	9251
VEE	723448	
VEE	723449	9252
YF	723452	9253
VEE	723455	9254
VEE	723457	0055
PT,YF	723459	9255
VEE	723460	
VEE	723463	
VEE	723465	
PT	723466	9256
PT,VEE	723467	9257
VEE	723468	
	701/71	9258
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P1	723471	7230
PT, VEE	723472	9259
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VEE	723473	9260
	723477	,200
	723482	9261
	723403	9262
16	723404	9263
YF	723403	9205
VEE	723486	9204
VEE	723487	9205
PT	723493	9200
YF	723801	9207
VEE	723802	9200
PT,VEE	/2380/	9209
PT,YF	723816	
VEE	723821	
YF	723822	
YF,VEE	/23826	
VEE	723827	
PT,YF,VEE	723828	
PT,VEE	723829	
PT	723830	
VEE	723831	
PT,YF,VEE	723833	
PT,YF	723835	
рт	723837	
VEE	723838	
VEE	723839	
VEE	723840	
VEE	723842	
PT	723846	
VEE	723849	
PT.YF	723864	
YF	723868	
YF	723871	
DT .	723872	
VE	723875	
1 r DT	723876	
F I DT	723070	
ri DT VE	723879	
ri,ir VF	723881	
PT VEE	723882	
PT YF VEE	723883	
VEE	723885	
VFF	723886	
VF	723887	
1 r DT	723808	
	723000	
ri, ir	723700	
YF	123903	
VEE	/23904	

PT,VEE	723908
PT.VEE	723909
PT, VEE	723910
YF, VEE	723911
YF,VEE	723912
VEE	723915
YF,VEE	723917
YF	723918
VEE	723922
YF	723932
PT	723933
PT	723934
PT.VEE	723937
VEE	723939
YF	723944
PT.VEE	723946
VEE	723948
PT YF	723949
YF	723950
VEE	723951
PT VF	723954
DT VE VEF	723957
DT VE	723958
ri,ir Vee	723050
VEE	723737
VEE	723900
VEE	723962
VEE	/23963
VEE	723964
VEE	/23965
VEE	723970
YF	723972
PT	723973
PT,VEE	723975
PT	723976
PT	723977
PT	723979
VEE	723980
VEE	723983
VEE	723985
PT	723989
VEE	72 39 92
YF	723993
VEE	723994
VEE	723995
VEE	723996
VEE	723997
PT	723998
VEE	723999
PT VFF	724001
PT	724004

VEE	724005	
VEE	724009	
VEE	724010	
VEE	724011	
VEE	724012	
VEE	724015	
VEE	724019	
VEE	724021	
PT,YF,VEE	724025	
VEE	724029	
VEE	724030	
VEE	724033	
VEE	724035	
VEE	724040	
YF	724042	
PT	724045	
VEE	724046	
?	724055	11589
PT	724058	11590
?	724060	11591
YF,VEE	724062	11592
YF	724063	11593
YF	724067	11594
?	724069	11595
VEE	724070	11596
PT,YF	724073	11597
PT	724075	11598
PT	724076	11599
VEE	724078	11600
РТ	724081	11601
YF	724086	11602
VEE	724087	11603
YF	724091	11604
VEE	724093	11605
YF	724100	11606
PT,YF	724108	11607
YF	724109	11608
PT	724110	11609
YF,VEE	724111	11610
YF	724112	11611
PT,VEE	724113	11612
VEE	724114	11613
YF	724124	11614
YF	724127	11615
YF	724130	11616
РТ	724131	11617
VEE	724135	11618
PT	724136	11619
YF	724137	11620
PT	724139	11621
PT,YF	724140	11622
PT	724141	11623

PT	724143	11624
YF	724145	11625
PT, VEE	724148	11626
PT	724151	11627
PT	724155	11628
VEE	724163	11629
РТ	724164	11630
PT,VEE	724165	11631
PT.VEE	724170	11632
PT	724172	11633
VEE	724173	11634
PT.VEE	724181	11635
PT	724182	11636
PT.VEE	724184	11637
PTYF	724186	11638
PTYF	724187	11639
PT	724192	11640
VEE	724194	11641
VEE	724198	11642
PT	724199	11643
PT	724201	11644
PT.VEE	724202	11645
VEE	724203	11646
VEE	724205	11647
VEE	724207	11648
PT	724209	11649
PT.VEE	724210	11650
VEE	724212	11651
VE VEE	724214	11652
VF VEE	724215	11653
PT	724216	11654
VEE	724217	11655
PT VEE	724219	11656
YF VEE	724220	11657
VEF	724221	11658
VEE	724223	11659
VEE	724225	11660
VEE	724225	11661
PT VF	724225	11662
DT	724220	11663
VFF	724227	11664
PT VF VFE	724220	11665
VER	724230	11666
VEE	724232	11667
VEE	724233	11668
YF	724238	
PT VEE	724240	
VEE	724242	
YF VEF	724243	
PT	724244	
PT	724247	
YF	724248	

VEE	724249
YF	724251
PT,VEE	724252
PT, VEE	724253
VEE	724254
YF	724256
VEE	724257
VEE	724258
VEE	724260
YF	724261
PT,YF	724266
PT,YF	724268
YF	724270
YF	724271
PT,YF	724275
YF	724279
PT	724280
YF	724281
PT	724283
YF	724287
VEE	724288
PT, VEE	724289
YF	724290
PT	724291
PT	724293
YF	724294
РТ	724296
YF	724297
YF	724298
YF	724299
YF	724300
PT	724301
РТ	724303
PT,YF	724308
PT	724310
YF	724312
YF	724313
YF	724316
VEE	724320
PT YF	724323
YF. VEE	724324
YF.VEE	724327
VEE	724335
PT VEE	724337
YF	724338
VEE	724339
PT.VEE	724340
PT	724348
PT.YF	724354
YF	724355
PT	724356
YF	724358

VEE	724361	
VEE	724363	
VEE	724365	
VEE	724366	
VEE	724367	
VEE	724368	
YF	725005	
YF,VEE	725006	11323
YF,VEE	725007	11324
YF	725008	11325
DT VEE	848631	9270
DT	848633	9271
PT	848634	9272
PT	848635	9273
PT VEE	848649	9274
PT	848650	9275
PT.YF.VEE	848653	9276
PT VEE	848654	9277
PT.VEE	848656	9278
PT	848657	9279
PT	848659	9280
PT	848660	9281
PT	848662	9282
PT	848663	9283
PT.YF	848669	9284
PT.VEE	848670	9285
PT.VEE	848671	9286
PT	848673	9287
PT	848676	9288
PT	848678	9289
PT	848679	9290
PT	848681	9291
PT.VEE	848691	9292
PT VEE	848699	9293
PT	848700	9294
PT	848703	9295
PT	848706	9296
VEE	848709	9297
РТ	848711	9298
PT	848716	9299
PT	848717	9300
PT	848720	9301
VEE	848722	9302
PT	848724	9303
PT, VEE	848725	9304
VEE	848727	9305
VEE	848728	9306
PT	848729	9307
VEE	848730	9308

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VEE	040/31	9310
PT	040/32	9311
VEE	848734	9312
PT, VEE	948734	9313
VEE	848736	9314
VEE	848737	2315
VEL DT VEF	848738	9316
FI,VEC	848739	9317
FI DT VEE	848740	9318
FI,VEE	848741	9319
FI VE VEE	848742	9320
IF, VEE	848745	9321
f i DT	848747	9322
r i pr	848748	9323
r i pr	848749	9324
	848750	9325
	848751	9326
ri DT VF VFF	848752	9327
PI, IF, VEE	848753	9328
FI,VEE	848755	9329
	848757	9330
	848767	9331
	848771	9332
IF VF	848774	9333
1 F D'F	848782	9334
r i DT	848788	9335
r L DT	848790	9336
r i DT	848792	9337
ri or vre	848793	9338
PI,VEC	848794	9339
PI,IF	848795	9340
PI, IF, VEE	848795	9341
YF, VEE	9/9707	9342
PT,VEE	040777	9343
VEE	040790	9344
YF	848800 848801	9345
VEE	040001	9346
PT, VEE	848804	9347
VEE	8//8808	9348
VEE Ve vee	848809	9349
IF,VED	848810	9350
VEC DT VEC	848811	9351
PI,VEE DT VEE	848812	
FL,VEE DT VE VEE	848814	9352
PI, IF, VEC	8/8816	9353
1 ⁻¹	848838	9355
ri Dr ve ver	878830	9356
PT, YE, VEE	040037	9357
PT, VEL	040041	9358
PT,VEL	040043	9359
PT,VEE	04004)	9360
PT,VEE	040040	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,

PT	848861	9361
PT,VEE	848864	9362
PT	848867	9363
VEE	848869	9364
PT,VEE	848870	9365
PT	848873	9366
PT	848874	9367
PT	848876	9368
PT	848879	9369
PT	848880	9370
PT	848881	9371
PT	848882	9372
PT	848883	9373
VEE	848892	9374
PT,VEE	848893	9375
PT,VEE	848895	9376
PT	848896	9377
PT	848897	9378
РТ	848899	9379
PT	848900	9380
PT	848901	9381
PT	848903	9382
PT	848904	9383
PT,YF	848905	9384
РТ	848906	9385
PT,VEE	848907	9386
РТ	848909	9387
PT,VEE	848911	9388
PT	848913	9389
PT,YF	848914	9390
YF	848916	9391
YF	848917	9392
PT	848920	9393
PT	848921	9394
YF	848924	9395
РŢ	848926	9396

YF,VEE	849188
PT	849190
PT,YF,VEE	849192
VEE	849195
YF,VEE	849196
VEE	849197
PT	849199
VEE	849200
VEE	849201
VEE	849202
VEE	849204
VEE	849206
VEE	849216

VEE	849219	
VEE	849220	
PT	849239	9397
VEE	849240	9398
PT.YF	849259	
PT	849268	11387
VEE	849269	11388
PT.VEE	849271	11389
PT.VEE	849272	11390
, РТ	849273	
PT	849275	11391
 Pፒ	849277	11392
?	849278	11393
?	849280	11394
PT	849281	11395
PT	849283	11396
PT	849284	11397
PT	849285	11398
PŤ	849289	11399
YF	849291	
PT	849292	11400
VEE	849295	11401
PT	849296	
PT VEE	849297	11402
PT. YF. VEE	849298	11403
PT	849299	11404
YF VEE	849300	11405
VEE	849302	11406
PT	849305	
VFF	849307	11407
VEE	849308	14408
V EL	0,,,,,,	
<u>Virus</u>	<u>Ctrl GRP No.</u>	AVS No
PT	18058	11124
PT	18060	11125
РТ	18061	11126
YF	18062	1112/
РТ	18063	11128
YF	18066	11129
YF	18068	11130
DT VEF	19457	11131
FI,VEE DT VE UEE	19458	11132
ri, ir, vee	19459	11133
(1,1) DT UEE	10/62	11134
ri,vee	10463	11135
ri,ir VD VCD	10/6/	11136
IF,VEE	10/65	11137
VEE	19467	11138
V N	1740/	

РТ	19469	11139
PT, VEE	19470	11140
PTVEE	19471	11141
PT	19472	11142
PT,YF,VEE	19473	11143
PT	19474	11144
YF	19479	11145
РТ	19480	11146
PT, VEE	19481	11147/11148
PT, VEE	19486	11149
PT, VEE	19490	11150
PT, VEE	19491	11151
PT	19492	11152
РТ	19493	11153
PT,VEE	19501	11154
PT,YF,VEE	19502	11155
PT	19509	11156
PT	19510	11157
PT	19511	11158
PT	19512	11159
YF	19514	11160
FT	19516	11326
PT	19521	11327
?	19522	11328
?	19523	11329
VEE	19525	11330
PT	19528	
PT	19529	11331
РТ	19533	11332
Ъ.Ц	19538	11333
РТ	19539	11334
PT	19541	11335
PT	19544	11336
PT	19545	11337
?	19550	11338
PT	19573	11339
РТ	19574	11340
РТ	19576	11341
PT	19578	11342
PT	19579	11343
PT	19583	11344
VEE	19584	11345
VEE	19594	11346
VEE	19598	11347
?	19599	11348
PT	19601	11349
PT,VEE	19602	11350
PT	19603	11351

РТ	19605	11352
рт	19608	11353
A A DT	19610	11354
r 1 pm	10611	11355
FI DT	19011	11256
PT	19612	11250
?	19619	11357
PT,YF	19621	11358
PT	19622	
VEE	19623	11359
PT	19624	
PT	19626	
VF VFF	19631	11360
VEE	19634	11361
VEE DT VEE	10435	11362
FI,VEE	19635	11363
VEE	19636	11266
PT, VEE	19640	11304
PT,VEE	19641	
PT	19642	
VEE	19643	11365
VEE	19647	11366
PT	19649	11367
?	19653	11368
VEE	19655	11369
PT	19657	11370
DT VE	19664	11371
	19004	11271
PI,VEE	19670	11372
VEE	19671	113/3
VEE	19672	11374
PT,YF	19674	11375
PT,YF	19675	11376
PT.YF	19676	11377
PT	19679	11378
2	19680	11379
፡ pт	19681	11380
	10605	11201
	19085	11201
PT,VEE	19686	11302
VEE	19689	
PT,VEE	19691	11383
PT	19692	
PT	19695	
YF	19696	11384
PT	19698	
PT VEE	19699	
,		
PT, VEE	19700	11385
PT	19701	11386
PT,YF,VEE	21188	11161
PT,YF	21189	11162
YF	21190	11163

PT,YF	21191	11164
PT	21194	11165
PT	21195	11166
PT,VEE	21197	11167
VEE	21199	11168
YF	21201	11169
VEE	21205	11170
PT,YF,VEE	21207	11171
VEE	21214	11172
PT,VEE	21220	11268
РТ	21221	11269
PT	21222	11270
PT	21223	11271
PT,VEE	21224	11272
PT,YF	21225	11273
PT,YF,VEE	21226	11274
VEE	21228	11275
PT	21229	11276
PT,YF,VEE	21232	11277
PT,VEE	21233	11278
PT,YF,VEE	21234	11279
YF	21235	
PT,YF,VEE	21236	11280
PT,YF	21238	11281
VEE	21242	11282
PT,YF	21247	11283
YF,VEE	21248	11284
YF	21251	11285
PT,VEE	21253	11286
YF, VEE	21255	11287
YF	21261	11288
PT	21263	11289
PT,YF,VEE	21264	11290
PT	21266	11291
PT, VEE	21267	11292
PT	21268	11293
PT, YF, VEE	21272	11294
YF	21275	11295
PT,YF	21278	11296
YF	21281	11431
YF	21282	11297
PT	21285	11298
PT	21286	11299
PT	21287	11300
YF	21288	11301
PT	21289	11302
PT,YF	21291	11303
PT,YF	21292	11304
?	21293	11305
VEE	21294	11306

PT	21297	11307
PT YF	21301	11308
VE VEE	21303	11309
PT VE	21305	11310
DT	21307	11311
DT VE	21307	11312
rl,ir DT VF	21314	11313
r1,1r	21313	1131/
P1	21317	11214
P1	21320	11315
PT,YF	21321	11217
YF	21323	11317
PT,YF,VEE	21325	11310
VEE	21326	11319
PT	21328	11320
PT	21329	11321
PT	21330	11322
YF	21668	11409
РТ	21669	
YF	21671	11410
PT YF VEE	21680	11411
РТ	21681	
PT .	21682	
PT	21683	
VE	21687	11412
VE	21688	
DT VEE	21689	11413
DT	21690	11414
VEE	21691	11415
DT	21695	
	21075	
	21090	11416
VEE	21095	11410
PT	21700	
YF	21703	11417
VEE	21704	11418
PT,YF,VEE	21705	11419
PT,VEE	21706	11420
PT,VEE	21708	11421
PT, VEE	23174	11422
PT VEE	23175	11423
PT	23176	
VEE	23184	11424
PT,VEE	23185	11425
PT,VEE	23186	11426
PT	23187	

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РT	23188	11427
PT	23189	11428
PT VE VEE	23190	11429
PT	23191	11430
YF	23192	11432
VF	23193	11433
VF	23194	11434
VE VEE	23195	11435
YF	23198	11436
YF	23199	11437
••	_	
PT.YF	23200	11438
YF	23202	11439
PT, YF, VEE	23203	11440
PTYF	23204	11441
PT, VEE	23206	11442
?	23207	11443
PT, YF, VEE	23210	11444
VEE	23211	11445
PT,YF	23213	11446
YF	23214	11447
VEE	23216	11448
VEE	23218	11449
VEE	23221	11450
PT.YF	23222	11451
PT.VEE	23225	11452
РТ	23226	11453
PT	23227	11454
VEE	23229	11455
VEE	23230	11456
PT.VEE	23233	11457
VEE	23234	11458
YF.VEE	23235	11459
VEE	23236	11450
VEE	23237	11461
PT.VEE	23238	11462
PT.VEE	23240	11463
PT	23242	11464
PT.YF.VEE	23243	11465
VEE	23244	11466
PT, YF, VEE	23245	11467
PT.VEE	23247	11468
YF VEE	23248	11469
PT, VEE	23249	11470
VEE	23250	11471
PT,VEE	23252	11472
YF, VEE	23253	11473
PT, VEE	23254	11474
PT, YF, VEE	23255	11475
YF	23256	11476
?	23257	11477

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YF, VEE	23258	11478
YF	23260	11479
YF,VEE	23261	11480
YF, VEE	23262	11481
YF	23263	11482
YF	23264	11483
PT,VEE	23265	11484
VEE	23267	11485
VEE	23269	11486
PT,YF,VEE	23270	11487
VEE	23271	11488
VEE	23272	11489
VEE	23273	
РТ	23274	11490
VEE	23275	11491
YF	23276	11492
VEE	23278	11493
PT.VEE	23279	11494
YF	23280	11495
PT.VEE	23282	11496
?	23283	11497
PT.YF.VEE	23284	11498
VEE	23285	11499
PT	23286	11500
PT YF	23287	11501
VFF	23288	11502
VEL	23290	11502
11 V65	23290	11504
VEE	23292	11505
PT VF	23292	11505
	23275	11500
VEE	23306	11507
PT	23307	
PT VEE	23308	11508
PT	23311	11509
VEF	23316	11510
VEE	23317	11511
VEF	23318	11512
755 77	23310	11512
11	23313	11514
	23322	11514
VEE	23325	11515
VEE	23325	11517
11	23320	11517
f D	22221	11610
1	23320	11519
VEL	23329	11520
	23330	11521
PT,VEE	23331	11522
PT	23333	11523
VEE	23334	11524
PT,VEE	23335	11525

PT	23337	11526
PT.VEE	23338	11527
PT.YF	23339	11528
VEE	23342	11529
PT VEF	23343	11530
DT VEE	23345	11531
ri,vec	23344	11522
P1,YF	23340	11552
?	23347	11533
YF	23348	11534
YF	23349	11535
YF	23350	11536
?	23353	11537
PT VEE	23354	11538
PT	23355	11539
VEE	23355	11540
VEE	23350	11540
VEE	23357	11541
YF	23359	11542
VEE	23361	11543
PT	23363	11544
YF	23364	11545
VEE	23365	11546
VEE	23366	11547
VEE	23369	11548
PT VEE	23370	11540
II,VEE	23370	11550
VEE	23371	11550
PT,YF	233/2	11551
VEE	23374	11552
VEE	23375	11553
VEE	23376	11554
PT.VEE	23377	11555
PT	23378	11556
VFF	23379	11557
VEE	23380	11558
VEE	23300	11550
VEE	23302	11559
VEE	23383	11560
VEE	23386	11561
VEË	23387	11562
VEE	23388	11563
VEE	23389	11564
VEE	23391	11565
VEE	23392	11566
VEE	23394	11567
VEE	23396	11568
VEE	23397	11569
VEE	23308	11570
VCC DT	23370	11570
r I	23373	
	224.00	11231
ri,vee	23400	115/1
7	23401	11572
VEE	23402	11573
VEE	23404	11574

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PT	23405	11575
VEE	23407	11576
PT,YF,VEE	23408	11577
VEE	23410	11578
PT,YF,VEE	23412	11579
PT	23417	
VEE	23418	11580
?	23426	11581
PT	23427	
PT,VEE	23428	11582
PT,VEE	23429	11583
VEE	23430	11584
?	23433	11585
VEE	23434	11586
PT	23437	11587
PT	23439	11588

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Table IV

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Full Screen Submissions for Grant Period

ASU No.	<u>Number of</u> Fractions	Virus	AVS No.	<u>Priority</u>
B611679	5 8 3	SFS	6975-6979 9439-9446 11181-11183	High
B619208	4		6980-6983	
B619315	5 6	SFS,VEE,VV	6984 - 6988 9447 - 9452	High
B619467	7 1		6989-6995 6994	
B624784	1		6000	
B630654	5		6996-7000	
B636725	7		7001-7007	
B662371	6 1		7008-7013 7009	
B677577	5 1		7014-7018 7017	
B680433	1		3966	
B705008	2 6	HIV HIV	8472-8473 8480-8485	
B705028	5	HIV	8433-8437	High
B706399	6	HIV	702, 8467-8471	-
B708007	2	HIV	8460-8461	
B708116	6	HIV	8454-8459	
B708122	12	HIV	8419-8430	
B708143	12	HIV	5426-5437	
B710052	16	HIV	8438-8453	
B711932	2	HIV	8465-8466	
B712294	2	HIV	8431-8432	

ASU No.	<u>Number of</u> Fractions	Virus	AVS No.	<u>Priority</u>
B712550	1 8	HIV Vee, yf	703 8343-8350	
B715449	13	HIV	8486-8498	
B716543	3	HIV	8462-8464	
B721116	5 3	Dengue,JE,YF HIV	6579-6583	
B721160	5	VEE	6733-6737	
B721557	4	HIV	8474-8477	
B721562	2	HIV	8478-8479	
B723123	5	YF	9453-9457	High
B723344	14	JE	6738-6751	
B724441	2	SFS	8351-8352	
B724957	3	RNA		
B805951	5		7019-7023	
B818539	5		7024 - 7028	
B827298	5 5 1		7029-7033 11184-11188 7031	High
B832245	5		7034-7038	
B832248	5		7039-7043	
B833909	9 1		7044-7052 7048	
B835741	6		7053-7058	
B836710	6		7059-7064	
B836749	7 3 1	YF	7065-7071 9458-9460	High
B841474	4		7072-7075	
B842649	7		7076 - 7082	

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<u>ASU No.</u>	<u>Number of</u> Fractions	Virus	AVS No.	Priority
B848528	5 2 5 2		7083-7087 7085, 7087 11260-11265 7083, 7087	High
B848990	2		11190-11191	High
B853791	11 1		7088-7098 11265	High
GRP-23148	3 (a plant	:)		

GRP-23465 1 (a plant)

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Table V

Special Sample Submissions for Grant Period

Sample	Weight	AVS No.
Lycorine	2 gm.	2563
Pancratistatin	114.9 mg. 41.8 mg. 200 mg.	361
Crude extract with Pancratistatin	1.56 gm.	
TRK-BS-1-10 TRK-BS-5 TRK-BS-8 TRK-BS-13		5787-5796 5791 5794 5797
Trichosanthin (GLQ 223)		5999
Narciclasine	4.09 gm.	2812
Isonarciclasine	200 mg. 1.58 gm.	4223
<u>cis</u> -Dihydronarciclasine	200 mg. 2.0 gm.	4590
<u>trans</u> -Dihydronarciclasine	50 mg. 20 mg. 200 mg.	4591
Balanitin 4-7 (B816351)		6001-6004
<u>cis</u> -Dihydro-7-deoxynarciclasine	40.1 mg. 200 mg.	4592
Streptimidone	2.068 gm.	4796
<u>trans</u> -Dihydro-7-deoxynarciclasine	18.6 mg. 10.1 mg. 100 mg.	4609
<u>iso</u> -7-Deoxynarciclasine	43.7 mg.	4527
Justicidin B	2.0 gm. 2.0 gm.	346
GRP-B3-5B		4742
GRP-B5-5F		4747

<u>Sample</u>		Weight	<u>AVS No.</u>
GRP-B45-1C			6789
GRP-B45-3A			6790
GRP-18072	DiMeVal-Val-MeVal-Pro (a synth	etic)	
GRP-18073	Dolapyrrolidone (a synthetic)		
GRP-18074	(a synthetic)		
GRP-18056	(a synthetic)		6793
GRP-18075	(a synthetic)		
GRP-18076	(a synthetic)		
GRP-18077	(a synthetic)		
GRP-18078	(a synthetic)		
GRP-18079	(a synthetic)		
GRP-18080	(a synthetic)		
GRP-22906	(a microorganism)		6791
GRP-23148	(a plant)		6792
SB-D-45E			4102
B631963 - K	D84 (NSC 374923)		2793
Dolastatin	15 (1 segment and 3 units)		
Dolastatin	15		
Bryostatin	1 10 x 100 μg		2712
Dolastatin	10 10 x 100 μg		2715
Stylotellin	1 (B722095)		

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A P P E N D I X B

Experimental Summaries

SENI-SYNTHETIC APPROACHES TO PANCEATISTATIN



Narciclasine-3,4-acetonide

To a solution of narciclasine (1.0 g, 3.25 mmol) in dimethylformamide (5 mL) and dimethoxypropane (5 mL) was added p-toluene sulfonic acid (100 mg). The solution was stirred at room temperature overnight. Acetonide precipitated out of solution. Pyridine (1 mL) and water (50 mL) was added and the mixture was stirred at room temperature for 30 minutes. The precipitate was collected by filtration, washed with water and dried at 64°C over P_2O_5 under high vacuum to give as an amorphous powder, narciclasine-3,4-acetonide (1.05 g, 92.9%), mp. 275-7°, IR (NaCl) ν_{max} 3500, 3150, 1637, 1625, 1596, 1464, 1437, 1337, 1201, 1079, 1038, 1019 cm⁻¹, ¹HNMR & (CDCl₃) 1.39 (s, 3H, CH₃), 1.53 (s, 3H, CH₃), 2.47 (d, J = 4.1 Hz, 1H, OH), 4.10-4.13 (m, 3H, H-3,4,4a), 4.39 (dd, J = 6.7, 4.1 Hz, 1H, H-2), 6.05 (ABq, J = 1.2 Hz, 2H, -OCH₂O-), 6.21 (brs, 1H, NH), 6.32 (dd, J = 3, 1.2 Hz, 1H, H-1), 6.70 (s, 1H, H-10), 9.2 (s, 1H, OH).



2,7-Di-[(tert-butyldimethyl)-silyloxy]-nerciclasine-3,4-acetonide

Dissepropylethyl amine (1.8 ml, 10.35 mmol) was added (under argon) to a heated (60°C) solution of narciclasine 3,4-acetonide (800 mg, 2.3 mmol) in dimethylformamide (8 ml) followed by tert-butyldimethylsilyl chloride (1.04 g, 6.9 mmol). The resulting reddish solution was stirred at room temperature overnight and monitored by TLC (hexane: acetons, 4:1). After completion, water (50 ml) was added and the viscous mixture was poured into ether (450 mL). The ethereal solution was washed with 10% aqueous citric acid (50 mL), water (2 x 100 mL), dried and evaporated under reduced pressure to give a gum which was crystallized from ethanol to afford colorless flakes of disilyloxy derivative (1.2 g, 90.5%), mp. 207-9°C, $[\alpha]_{0}^{30}$ +61.2°, (c, 2.5, CHCl₃), IR (NaCl) ν_{max} 3250, 2952, 2930, 2857, 1676, 1480, 1381, 1362, 112, 1057, 837 cm⁻¹, ¹HNMR & (CDCl₃) 0.148, 0.152 (s, 6H, 2xCH₃), 0.219, 0.225 (s, 6H, 2xCH₃), 0.945 (s, 9H, C(CH₃)₃), 1.335 (s, 3H, CH₃), 1.467 (s, 3H, CH₃), 3.969-4.019 (m, 2H, 2xCH), 4.065 (dd, J = 7.1, 5.2 Hz, 1H, CH), 4.305 (quint, J = 2.5 Hz, 1H, CH), 5.902 (brs, 1H, NH), 5.967 (d, J = 1.2 Hz, 1H, 1/20CH₂O), 5.984 (d, J = 1.2 Hz, 1H, 1/20CH₂O), 6.153 (brt, J = 2.3 Hz, 1H, H-1), 6.799 (s, 1H, H-10).



1,10b-(α)-and-(β)-Epoxy-2,7-di-[(tert-butyldimethyl)silyloxy]-narciclasine-3,4-acetonide

To a stirred solution of 2,7-di-[(tert-butyldimethyl)-silyloxy}narciclasine-3,4-acetonide (240 mg, 0.42 mmol) in CH₂Cl₂ (10 mL) was added 0.2M phosphate (pH 8.0) buffer (10 ml, prepared from Na₂HPO, and NaH₂PO,). The biphasic mixture was cooled to 0°C and m-chloroperbenzoic acid (215 mg, 1.2 mmol) was added and the mixture was stirred 20 min. at 0°C and then 4 hrs. at room temperature. The reaction was carefully monitored (TLC, hexane:acetone, 17:3) and upon completion CH_2Cl_2 (100 mL) was added. The organic phase was separated and washed with 5% aqueous sodium thiosulfate (2 x 25 mL), water (25 mL), 5% aqueous sodium carbonate (3 x 25 mL), water (25 mL), dried (Na₂SO₄), and evaporated under reduced pressure to produce a colorless powder. Chromatography (VLC) on neutral SiO₂ and elution with hexane-acetone (95:5) gave an inseparable mixture ($\alpha:\beta$, ¹HNMR of hydrogenolyzed product) of J = 8.0Hz, 1H), 4.158-4.266 (m, 3H), 5.763 (brs, 1H, NH), 5.972 (d, 1H, J = 1.6 Hz, 1H, 1/20CH₂O), 6.014 (d, J = 1.6 Hz, 1H, 1/20CH₂O), 7.294 (s, 1H, H-10), EIMS (m/z) 591 (2%), 576 (10), 534 (100), 518 (4), 476 (10), 430 (10), 344 (4), 316 (6).



(major product)

la and 1β -Hydroxy-2,7-di-[(tert-butyldimethyl)silyloxy]-10b, 4a-<u>cis</u> and <u>trans</u>-dihydro-narciclasine-3,4-acetonide

To a solution of epoxide mixture (see above) (100 mg, 0.17 mmol) in methanol:ethylacetate (1:3, 20 mL) was added 10% palladium supported on carbon (100 mg). The reaction mixture was evacuated and flushed with hydrogen (5x) and then hydrogenated at ambient temperature and pressure for 1 hr using a hydrogen filled balloon. The catalyst was removed by filtration and the filtrate concentrated to dryness to give a powder (97 mg), purified on PLC (SiO₂, hexane:ethylacetate, 4:1) to give a mixture (1:17, ¹HNMR analysis) of 1 β , 10ba and 1 α , 10b β alcohols (80 mg, 80%, found to lose the phenolic sily1 group in solution), as an amorphous powder from acetone-hexane, mp 116-9°, [α]₀³⁰ +7.5° (c, 0.55, CHCl₃), IR (NaCl) ν_{max} 3250, 2952, 2929, 1675, 1473, 1382, 1360, 1250, 1220, 1111, 1069, 1042, 839 cm⁻¹, ¹HNMR of major product (1 α -hydroxy isomer), δ (CDCl₃) 0.096, 0.158 (s, 3H each, Si(CH₃)₂), 0.231 (s, 6H, C(CH₃)₃), 1.403 (s, 3H, CH₃), 1.531 (s, 3H, CH₃), 2.310 (brs, 1H, OH), 2.926 (brs, 1H, H-10b), 3.745 (dd, J = 7.5, 3.3 Hz, 1H, H-2), 3.828 (brs, 1H, H-4a), 4.109 (dd, J = 5.4, 1.5 Hz, 1H, H-4), 4.196 (d, J = 4.0 Hz, 1H, H-1), 4.217 (dd, J = 7.3, 5.3 Hz, 1H, H-3), 5.115 (brs, 1H, NH), 5.930 (d, J = 1.3 Hz, 1H, 1/20CH₂O), 5.993 (d, J = 1.3 Hz, 1H, 1/20CH₂O), 6.415 (s, 1H, H-10).



(major product)

$la, 2a, 3\beta, 4\beta, 7$ -pentscetyl-10b, 4a-<u>cis</u>-dihydro-narciclasine (la, 10b β -isopancratistatin pentsecetate) and Pancratistatin pentsecetate

To a cooled (0°C) solution of the mixture of silylather (see above, 15 mg, 0.025 mmol) in methanol (2 mL) and water (0.5 ml) was added acatic acid (0.5 ml) and trifluroacatic acid (0.5 ml). The solution was stirred at 0°C for

2 hrs and then stored in a refrigerator overnight. Solvents were removed under reduced pressure and the resulting product was dried under high vacuum over phosphorous pentoxide for 4 hrs. The product was then acetylated using pyridine (0.5 mL) and acetic anhydride (0.5 mL) at room temperature overnight followed by heating at 60° C for 1 hr. The reaction mixture was quenched with methanol and the volatile materials were evaporated through azeotropic distillation with methanol and cyclohexans. Product was found to be a (1:9)mixture of pancratistatin pentaacetate (detected only by NMR spectrum of the mixture) and $lo, 10b\beta$ -isopancratistatin pentaacetate. The products were separated on a column of silica gel by elution with CH₂Cl₂:CH₃OH, 99:1 to give 9.0 mg of an amorphous powder from CH_2Cl_2 -hexans of $l\alpha_1 2\alpha_1 3\beta_1 4\beta_1 7$ -pentaacetyl-10b,4a-<u>cis</u>-dihydro-narciclasine, mp.165-9°, [a]₀³⁰ +135 (c, 0.2, CHCl₃), IR (NaCl) ν_{max} 3341, 1778, 1751, 1677, 1481, 1371, 1248, 1226, 1192, 1084, 1035 cm⁻¹, ³HNMR & (CDCl₃) assignment based on ³H, ¹H-COSY spectra, 1.893, 1.979, 2.027, 2.166, 2.351 (each s, 3H each, 5 x OCOCH₃), 3.305 (t, J = 3.8 Hz, 1H, H-10b), 3.933 (t, J = 2.5 Hz, 1H, H-4a), 5.405 (dd, J = 10.7, 3.2 Hz, 1H, H-2), 5.460 (brs, 1H, NH), 5.470 (dd, J - 10.7, 2.3 Hz, 1H, H-3), 5.488 (brs, 1H, H-4), 5.532 (t, J = 3.4 Hz, 1H, H-1), 6.068 (d, J = 1.2 Hz, 1H, $1/20CH_2O$), 6.085 (d, J = 1.2 Hz, 1H, 1/20CH₂O), 6.626 (s, 1H, H-1O), the chemical shift for NH shifted downfield at δ 5.590 in dilute solutions (ca. 1.5 mg/0.5 mL). Cis relationship of the protons at H-4a, H-10b and H-1 established by NOE measurement. Thus strong NOE's were observed between H-10b, H-1, H-2, H-10, H-4a, and H-4a also gave NOE enhancement to NH). The NOE's also establishes proof for the chair conformation of ring C.



(major product)

la-Isopancratistatin

Palladium/carbon (10%, 80 mg) was added to the epoxide mixture described above (80 mg, 0.14 mmol) in anhydrous THF (45 mL) and the hydrogenolysis was performed as described in the previous experiment for 8 hrs. The filtrate obtained after removal of the catalyst was concentrated to dryness to give a 2:1 mixture of trans: cis dihydro product, by ¹HNNR analysis. The products were separated on PLC (hexana-acetone (17:3) to give trans dihydro product (42 mg, 52.3%) and cis dihydro product (20 mg, 24.8%), identical (¹HNNR and TLC) with the cis product obtained in the previous hydrogenolysis reaction. Slightly impure trans dihydro product was obtained as an amorpho's powder from acetone-hexane; ¹HNNR 6 (CDCl₃) of major product: 0.104, 0.138, 0.190, 0.199 (each s, 3H each, 2 x Si(CH₃)₂), 0.902, 0.971 (each s, 9H each, 2 x C(CH₃)₃), ¹ 1.312, 1.433 (each s, 3H each, C(CH₃)₂), 2.814 (dd, J = 14.3, 7.7 Hz, 1H, H-10b), 3.085 (d, J = 3.7 Hz, 1H, OH), 3.457 (dd, J = 14.1, 8.0 Hz, 1H, H-4a), 3.866 (dd, J = 7.2, 5.0 Hz, 1H, H-2), 4.055 (m, 1H, H-1), 4.207 (t, J = 8.5 Hz, 2H, H-3,4), 5.696 (s, 1H, NH), 5.910 (d, J = 1.2 Hz, 1H, 1/20CH₂O), 5.950 (d, J = 1.2 Hz, 1H, 1/20CH₂O), 6.998 (s, 1H, H-1O).



To a cooled (0°C) solution of <u>trans</u> product (25 mg 0.042 mmol) in THF:CH₃OH:H₂O (1.5:2:1, 4.5 mL) was added acetic acid (0.5 mL) and trifluoroacetic acid (1.0 mL) and stirred at the same temperature for 1 hr. After storing overnight in the refrigerator, the reaction was not complete and required heating to 40°C for 8 hrs. Solvents were removed under reduced pressure and the product was purified by flash chromatography on silica gel. The product, la-isopancratistatin (11.1 mg, 81%), eluted with a 9:1 mixture of CH₂Cl₂:CH₃OH and was obtained as an amorphous powder, mp. 325-7, IR (KBr) ν_{max} 3500-3300, 1679, 1470, 1337, 1285, 1210, 1141, 1089, 1064, 802, 725 cm⁻¹, ¹HNMR & (DMSO-d₅+D₂O) 2.80 (dd, 10.9, 10.9 Hz, 1H, H-10b), 3.31 (dd, J = 13.2, 10.5 Hz, 1H, H-4a), 3.76 (m, 2H), 3.81 (t, J = 3.6 Hz, 1H), 3.84 (brs, 1H), 5.97, 6.00 (only two AB lines visible, 2H, OCH₂O), 7.27 (s, 1H, H-10).



 $1\alpha, 2\alpha, 3\beta, 4\beta, 7$ -Pentascetyl isopancratistatin

la-Isopancratisatin (2.7 mg) was treated with acetic anhydride (0.2 mL) in pyridine (0.2 ml) at 50°C for 2 hrs. The mixture, quenched with methanol, was reduced to dryness under a nitrogen stream. Product was chromatographed on a column of silica gel and eluted with $CH_2Cl_2:CH_3OH$ (49:1) to give an amorphous powder of Lo-isopancratistatin pentaacetate (3.4 mg, 76.5%), mp. 146-8, IR (NaCl) ν_{max} 3350, 2930, 2850, 1755, 1676, 1482, 1370, 1248, 1226, 1176, 1084, 1059, 1033 cm⁻¹, ¹HNMR & (CDCl₃) 2.06 (s, 3H, COCH₃), 2.09 (s, 3H, COCH₃), 2.12 (s, 3H, COCH₃), 2.18 (s, 3H, COCH₃), 2.36 (s, 3H, COCH₃), 3.44 (t, J = 11.9 Hz, 1H, H-10b), 3.84 (t, J = 11.0 Hz, 1H, H-4a), 5.25 (dd, J = 10.8, 3.0 Hz, 1H, H-4), 5.42 (dd, J = 11.4, 3.3 Hz, 1H, H-1), 5.44 (t, J = 3.3 Hz, 1H, H-1), 5.53 (t, J = 3.8 Hz, 1H, H-2), 6.06 (d, J = 1.2 Hz, 1H, 1/20CH₂O), 6.07 (d, J = 1.2 Hz, 1H, 1/20CH₂O), 6.54 (s, 1H, H-1O), spectrum was assigned on the basis of 2D-COSY, and the stereochemistry by NOEDS data.



la-Hydroxy-2-[(tert-butyldimethyl)silyloxy]-10b, 4a-<u>cis</u> and <u>trans-</u> isopancratistatin-3,4-acetonide and la-hydroxy-2-[(tertbutyldimethyl)silyloxy]-A(10b,4a)-isopancratistatin-3,4-acetonide

The silyloxy epoxide mixture (150 mg) was dissolved in THF (10 mL) and hydrogenolyzed using hydrogen-10% Pd/C (50 mg) as described. Chromatography on a silica gel column and elution with hexane-acetone (7:3) gave 7desilylated products (interestingly desilation was occurring during the hydrogenolysis reaction as crystallized epoxide was free of benzoic acid by NMR) trans:cis:A(10b,4s) in the ratio of (5:3:5). The 10b,4a trans product (50 mg, 38%), crystallized from methanol as shining flakes, mp. 274-5; IR (NaCl) ν_{max} 3530, 3360, 2952, 2939, 1678, 1466, 1373, 1361, 1345, 1260, 1230, 1085, 1071 cm⁻¹; ¹H NMR & (CDCl₃) 0.15 (s, 3H, SiCH₃), 0.18 (s, 3H, SiCH₃), 0.94 (s, 9H, SiC(CH₃), 1.37 (s, 3H, CH₃), 1.48 (s, 3H, CH₃), 2.90 (dd, J = 14.3, 6.9 Hz, 1H, H-10b), 3.13 (d, J = 2.7 Hz, 1H, OH), 3.57 (dd, J = 14.5, 7.9 Hz, 1H, H-4a), 3.92 (dd, J = 6.8, 5.2 Hz, 1H, H-2), 4.13 (ddd, J = 7.4, 4.6, 2.2 Hz, 1H, H-1), 4.25 (t, J = 6.7 Hz, 1H, H-4), 4.29 (t, J = 6.7 Hz, 1H, H-3), 6.03 (ABq, J = 3.0 Hz, 2H, OCH₂O), 6.04 (brs, 1H, NH), 6.93 (s, 1H, H-1O), 12.48 (s, 1H, ArOH), (assignment was made on the basis of a 2D-COSY analysis and stereochemical assignment was accomplished by NOEDS measurement).



Continued elution of the column with the same solvent gave a mixture of <u>cis</u> and 10b,4a (A) product. The <u>cis</u> product could not be separated

but crystallization of the mixture from methanol yielded pure 10b, 4a (Δ) olefinic product (55 mg, 41.44), recrystallized from methanol as flakes, mp. 266-8; IR (NaCl) ν_{max} 3535 (brs), 2989, 2959, 2931, 2897, 2857, 1677, 1625, 1485, 1422, 1373, 1253, 1215, 1117, 1086, 1036 cm⁻¹; ¹H NMR & (CDCl₃) 0.17 (s, 3H, SiCH₃), 0.21 (s, 3H, SiCH₃), 0.97 (s, 9H, SiC(CH₃), 1.49 (s, 3H, CH₃), 1.50 (s, 3H, CH₃), 2.90 (s, 1H, OH), 3.86 (dd, J = 7.8, 3.5 Hz, 1H, H-2), 4.50 (t, J = 7.5 Hz, 1H, H-3), 4.76 (d, J = 3.4 Hz, 1H, H-1), 5.11 (d, J = 7.0 Hz, 1H, H-4), 6.10 (d, J = 1.6 Hz, 1H, 1/20CH₂0), 6.11 (d, J = 1.6 Hz, 1H, 1/20CH₂0), 6.83 (s, 1H, H-10), 9.85 (s, 1H, NH), 12.65 (s, 1H, OH), Assignment is based on 2D-COSY spectrum and stereochemistry was determined by NOEDS measurement.

Hydrogenolysis of the silyloxy epoxide mixture on a scale better than the one reported here in different solvents (ethyl acetate, mixture of ethyl acetate and methanol) produced similar products. Hydrogenolysis in methanol mostly produced the Δ (10b, 4a) product.



2,7-Discetoxy-marciclasine-3,4-acetonide

Acetonide was prepared from narciclasine (1 g) as described before and all the solvents were evaporated under reduced pressure to give a crude product which was acetylated with acetic anhydride (3 mL) - pyridine (3 mL) at 60°C for 6 hrs. Solvents were evaporated under reduced pressure after addition of methanol and then chromatographed on a silica gel column and eluted with hexane-ethyl acetate-methylene chloride (3:1:2) to give pure diacetate (1.2 g, 85.4%) as an amorphous powder from acetone-hexane, mp. 130-33 °C; IR (NaCl) ν_{max} 3350, 1775, 1745, 1671, 1482, 1373, 1233, 1210, 1177, 1081, 1031 cm⁻¹; ¹H NMR 6 (CDCl₃) 1.39 (s, 3H, CH₃), 1.52 (s, 3H, CH₃), 2.21 (s, 3H, COCH₃), 2.39 (s, 3H, COCH₃), 4.12 (dd, J = 7.8, 7.8 Hz, 1H, H-3), 4.16 (brs, 1H, H-4a), 4.31 (dd, J = 7.5, 5.8 Hz, 1H, H-4), 5.39 (dd, J = 4.9, 2.3 Hz, 1H, H-2), 6.02 (brs, 1H, NH), 6.09 (brs, 2H, OCH₂O), 6.12 (t, J = 3.2 Hz, 1H, H-1), 6.98 (s, 1H, H-10).



1,10b-(a)-Epoxy-2,7-diacetoxy-marciclasine-3,4-acetomide

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To a solution of narciclasine acstonide diacetate (1.0 g, 2.32 mmol) in CH_2Cl_2 (60 mL) was added 0.2 M soadium phosphate buffer (pH 8, 60 mL) followed by m-chloroperbenzoic acid (1.4 g, 3.5 molar equivalent). The reaction mixture was stirred at room temperature overnight and then CH_2Cl_2 (500 mL) was added and the organic layer was separated, washed with 5% solution of sodium thiosulfate (3 x 200 mL), 5% solution of sodium carbonate (3 x 200 mL), water (2 x 100 mL), dried (Na₂SO₄) and evaporated to give almost pure epoxide as a powder, crystallized from acetone-hexame (700 mg, 67.5%) as amorphous granules, mp. 231-232 °C; IR (NaCl) ν_{max} 3370, 1792, 1749, 1682, 1500, 1365, 1345, 1209, 1175, 1083, 1032 cm⁻¹; ¹H NMR & (CDCl₃) 1.31 (s, 3H, CH₃), 1.43 (s, 3H, CH₃), 2.20 (s, 3H, COCH₃), 2.36 (s, 3H, COCH₃), 4.00 (d, J = 6 Hz, 1H, H-4a), 4.03 (s, 1H, H-1), 4.27 (apparent t, J = 8.1 Hz, H-1, H-3), 4.38 (dd, J = 7.9, 6.3 Hz, 1H, H-4), 5.33 (d, J = 6.1 Hz, 1H, H-2), 5.82 (brs, 1H, NH), 6.07 (ABq, J = 1.2 Hz, OCH₂O), 6.43 (s, 1H, H-1O).

SYNTHETIC TRANSFORMATION OF 7-DEOXYNARCICLASINE



7-deoxynarciclasine

iso-7-deoxynarciclasine



cis-dihydro-7-deoxynarciclasine

trans-dihydro-7-

-deoxynarciclasine

<u>Hydrogenation of 7-deoxynarciclasine</u>. A solution of 7-deoxynarciclasine (1.02 g, 3.4 mmol) in methanol-ethanol (400 ml, 1:1) was degassed with nitrogen, platinum oxide (57 mg) was carefully added and the resulting mixture was hydrogenated at ambient temperature and pressure for 24 hrs. The reaction mixture was filtered through Celite and concentrated <u>in vacuo</u> to afford crude <u>iso-7-deoxynarciclasine</u> (150 mg, dark brown solid) which crystallized from pyridine-hexane as a powder (100 mg, 9.8% yield). Identity with earlier sample of the compound was confirmed by mp and nmr. The mother liquor contained <u>cis</u> and <u>trans</u>-dihydro-7-deoxynarciclasine.



cis and trans-dihydronarciclasine triacetate. The crystallization residue from <u>iso</u>-7-deoxynarciclasine (see above) was concentrated to dryness and treated with acetic anhydride (7 ml) and pyridine (10 ml) at 60° for 6 hrs. Methanol was added and the resulting solution was concentrated to dryness. The mixture was flash chromatographed over silica gel using CH_2Cl_2 :MeOH (99.4-0.6) twice to furnish <u>trans</u>-dihydro-7-deoxynarciclasine (112 mg, 7.6% yield) and <u>cis</u>-dihydro-7-deoxynarciclasine (752 mg, 51.2% yield). Identity with earlier sample of the compound was confirmed by mp and nmr.



<u>Trans-dihydro-7-deoxynarciclasine</u>. To a solution of the triacetate (112 mg) in methanol (20 ml) was added a saturated solution of barium hydroxide (6 ml). After heating for 15 min. at 100°C, the mixture was cooled, saturated with solid CO_2 , stirred at room temperature overnight and filtered. The filtrate was evaporated to dryness and the product was crystallized from methanol to give <u>trans</u>-dihydro-7-deoxynarciclasine (51 mg, 65% yield). Identity with earlier sample of the compound was confirmed by mp and nmr.



<u>cis-dihydro-7-deoxynarciclasine</u>. To a solution of the triacetate (750 mg) in methanol (80 ml) was added potassium carbonate (300 mg). The mixture was stirred at room temperature for 2 hrs, then filtered through a column of Sephadex LH-20. Elution with CH_2Cl_2 :MeOH (3:2) removed the product from the column. Crystallization from acetone-MeOH afforded <u>cis</u>-dihydro-7-deoxynarciclasine as crystals (419 mg, 80t yield). Identity with earlier sample of the compound was confirmed by mp and nmr.