

# Pyroxenes

- <http://www.es.mq.edu.au/ndaczko/Web/GEOS254/GEOS254.html>
- Pyroxenes are a major component of the mantle (peridotite is olivine plus pyroxene) and the oceanic crust (basalt, dolerite, gabbro are mainly calcium-rich plagioclase plus pyroxene).
- Pyroxenes are also present in many high grade (high temperature) metamorphic rocks. The calcium magnesium pyroxene diopside forms in meta-limestone even at low temperature.
- Fragments of pyroxene are present in many volcaniclastic sandstones.

# PYROXENES $X Y Z_2 O_6$

(X=Ca,Mg,Fe,Na); Y= Mg,Fe,Al,Ti; Z=Si)

- Most common pyroxenes are solid solutions of
- Diopside  $CaMgSi_2O_6$  Hedenbergite  $CaFeSi_2O_6$
- Enstatite  $MgMgSi_2O_6$  Ferrosilite  $FeFeSi_2O_6$

Two other important pyroxene end-members are  
Jadeite  $NaAlSi_2O_6$       Aegirine  $NaFe^{3+}Si_2O_6$

3 other well recognised pyroxenes are:

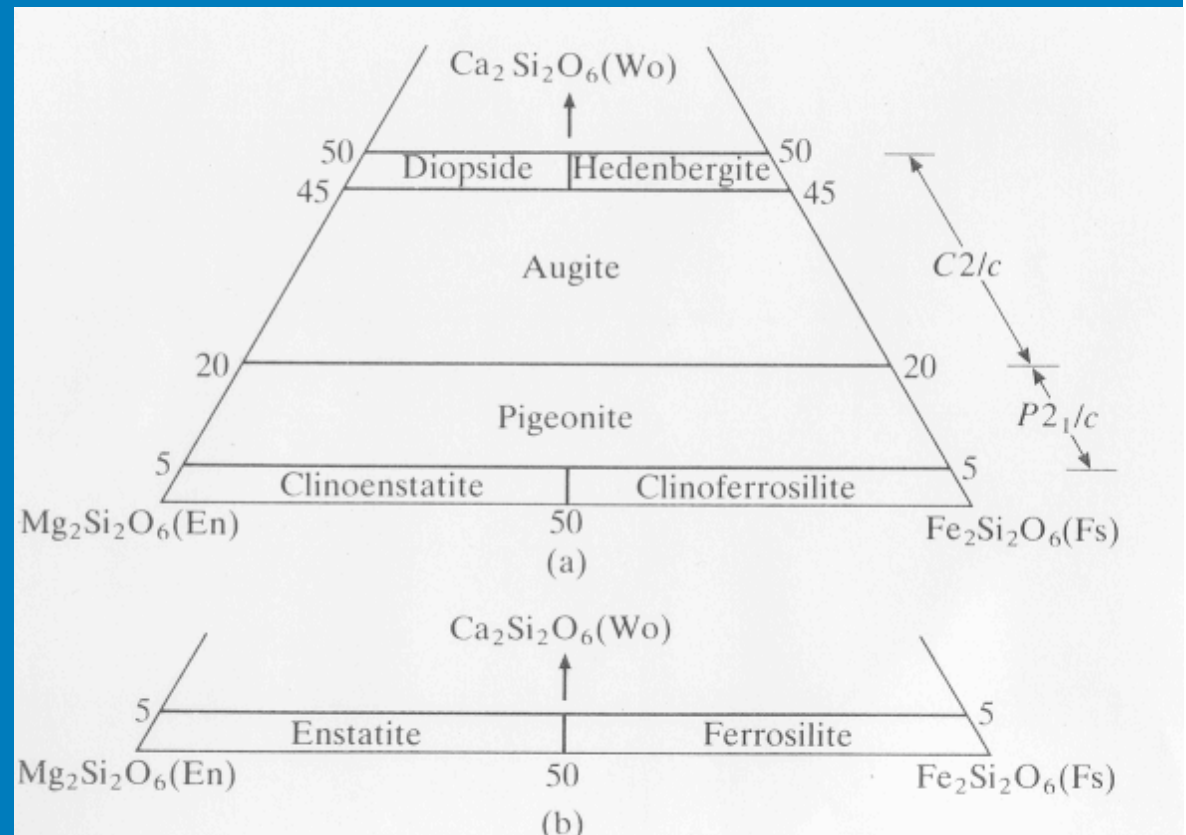
Omphacite (about half augite/half jadeite) in eclogites.

Aegirine-augite (about equal amounts of both) in trachytes (green)

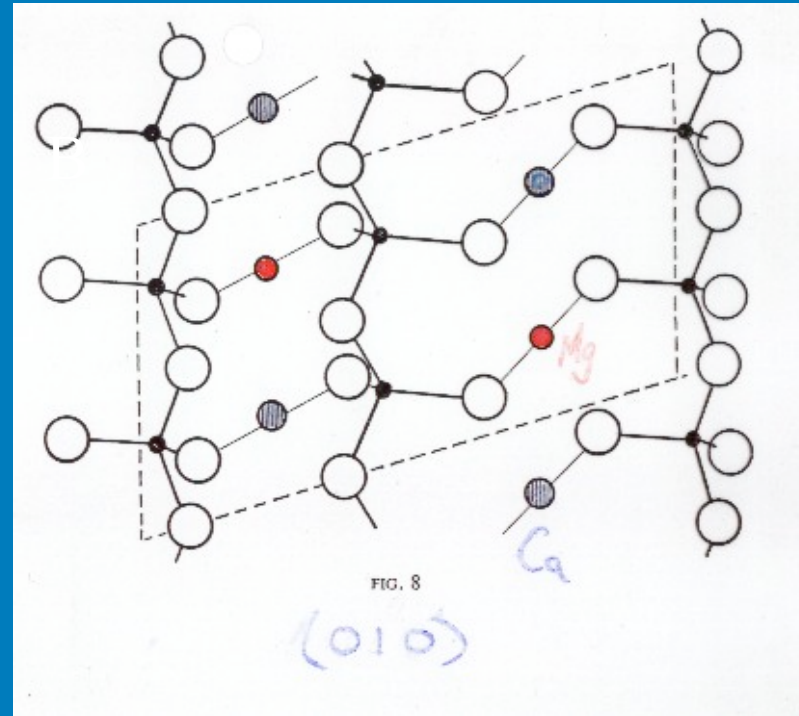
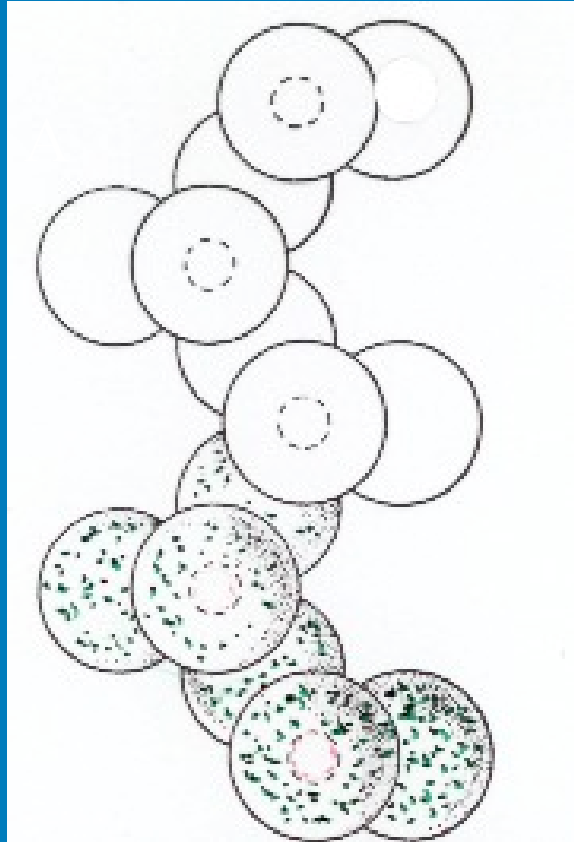
Titan-augite (Ti replaces Fe) in intra-plate basalts (pink/mauve)

# The pyroxene quadrilateral

- Most monoclinic or “clino”-pyroxenes are close to the Di – He join.
- Most orthorhombic or “ortho”-pyroxenes are close to the En – Fs join.
- Pigeonite is a calcium-poor clinopyroxene that occurs in some dolerites.
- Augite is a common igneous clinopyroxene.



# PYROXENE STRUCTURE

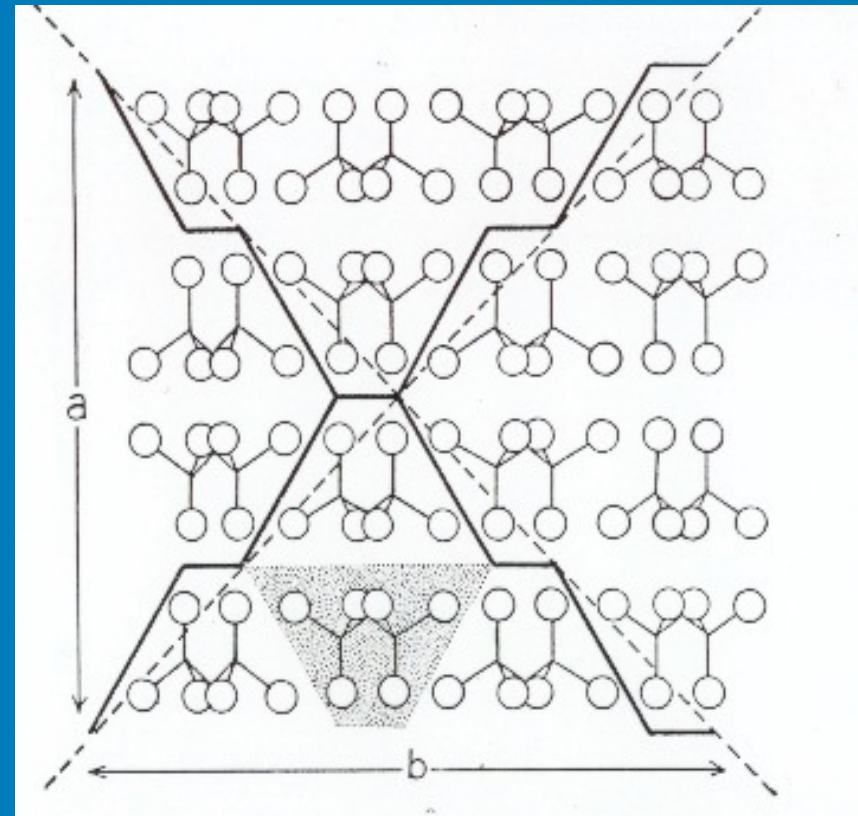


- A: Single chain parallel to c-axis.
- Looking along the b-axis of diopside (clinopyroxene). Red atoms are Mg, grey are Ca.

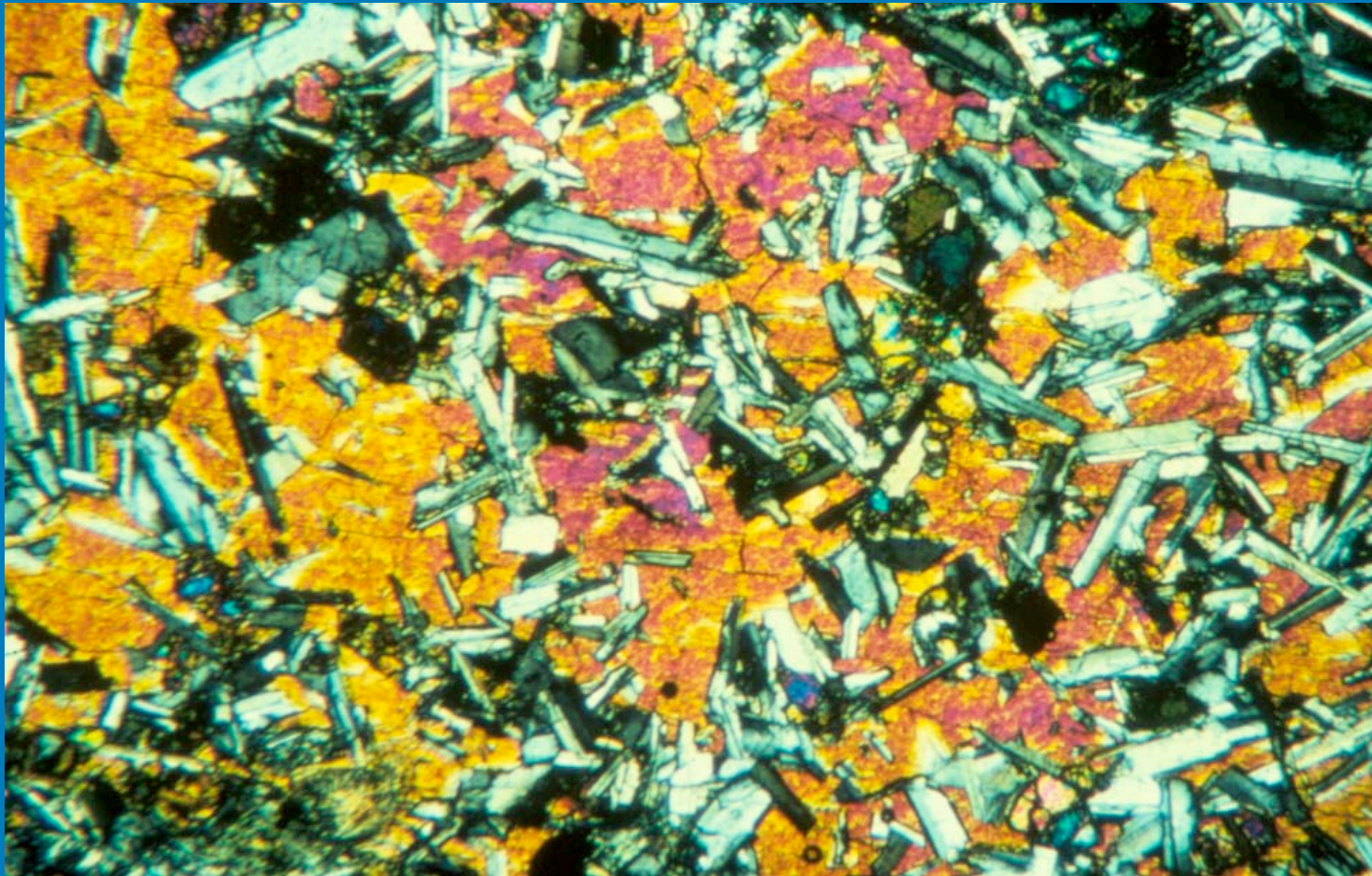


# Pyroxene cleavage

- Pyroxene has two cleavages at  $87^\circ$ . True angle seen in sections cut normal to the c-axis.
- Sections cut parallel to the C-axis will show one direction of cleavage.



**Ophitic microstructure (dolerite)  
plagioclase is included in large  
pyroxene crystals.**

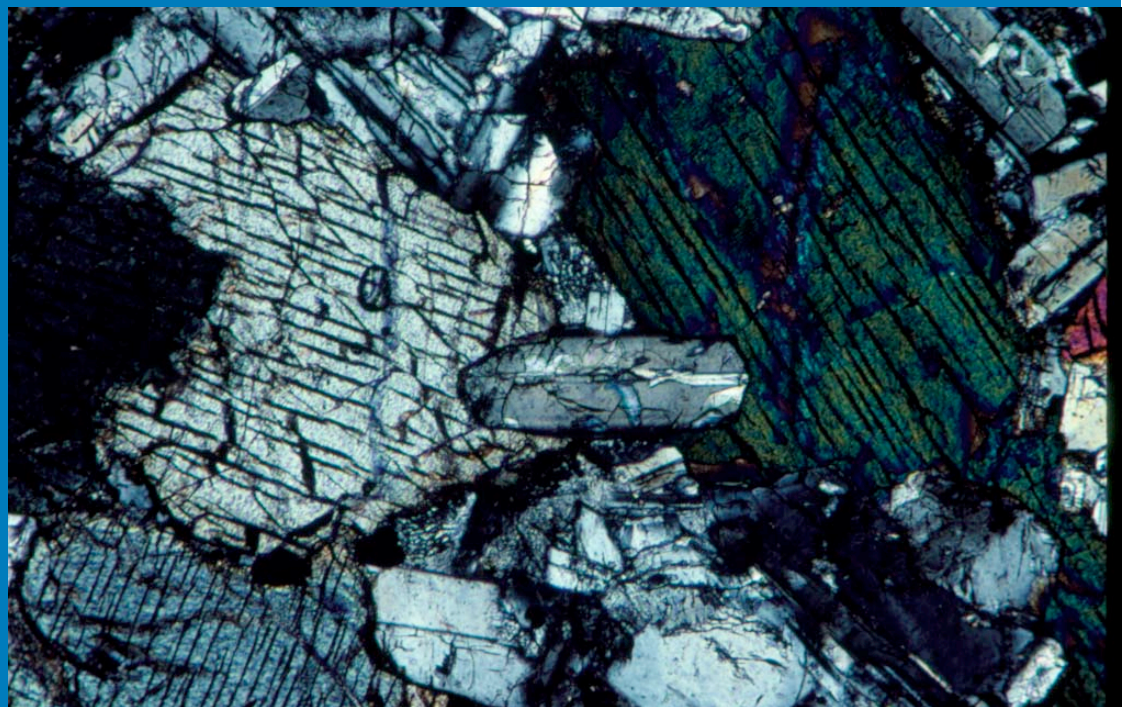
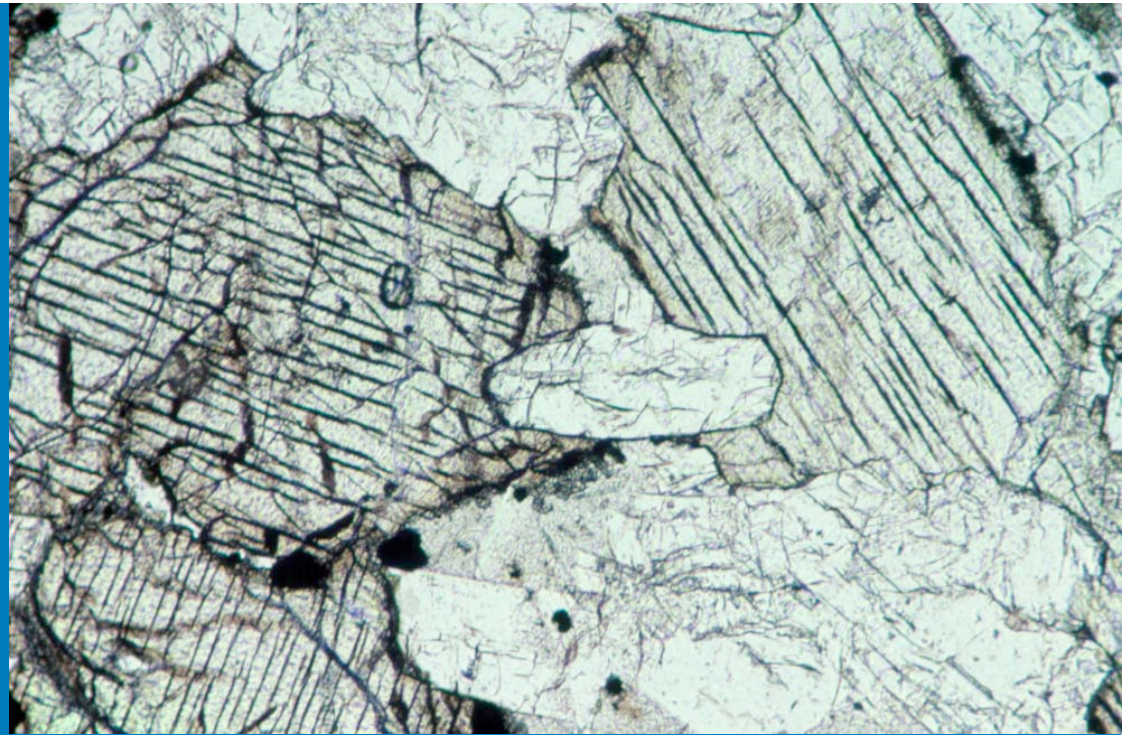




**Close up of plagioclase partly included in pyroxene in sub-ophitic dolerite.**

**Top PPL, bottom is XP  
This pyroxene is colourless with high relief (cleavage shows clearly).**

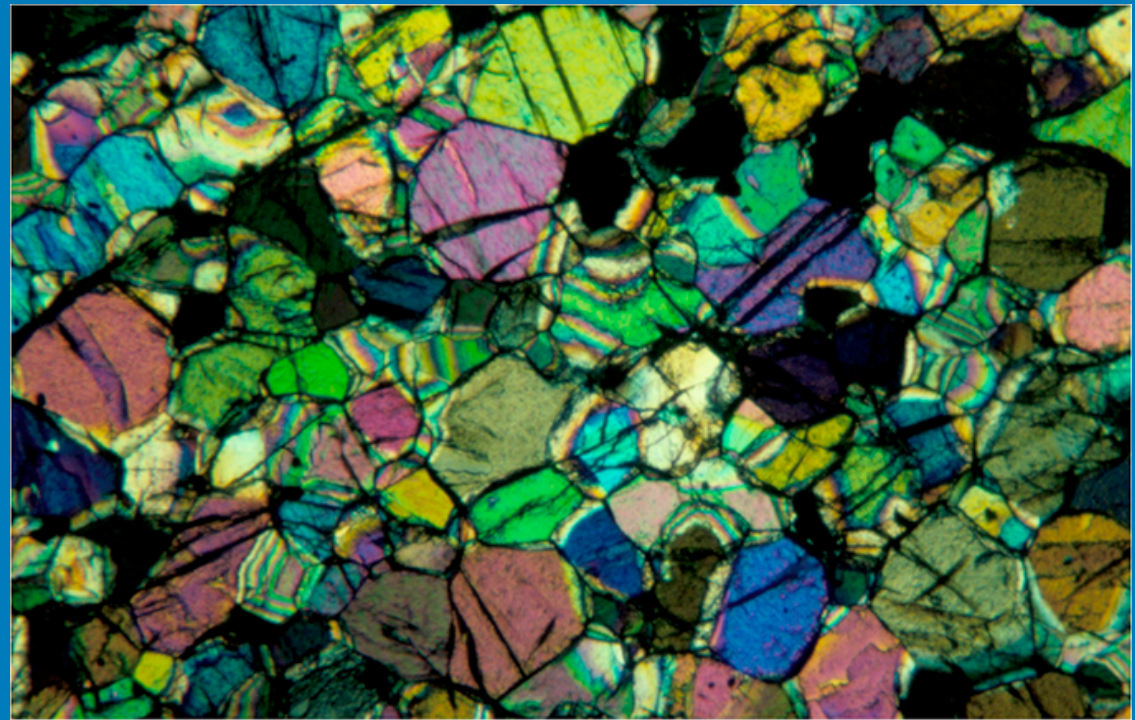
**The two grains on the left have interference colours towards the lower end of the range.**





# Eclogite (ultra high grade metabasalt) with pyroxene and garnet (black)

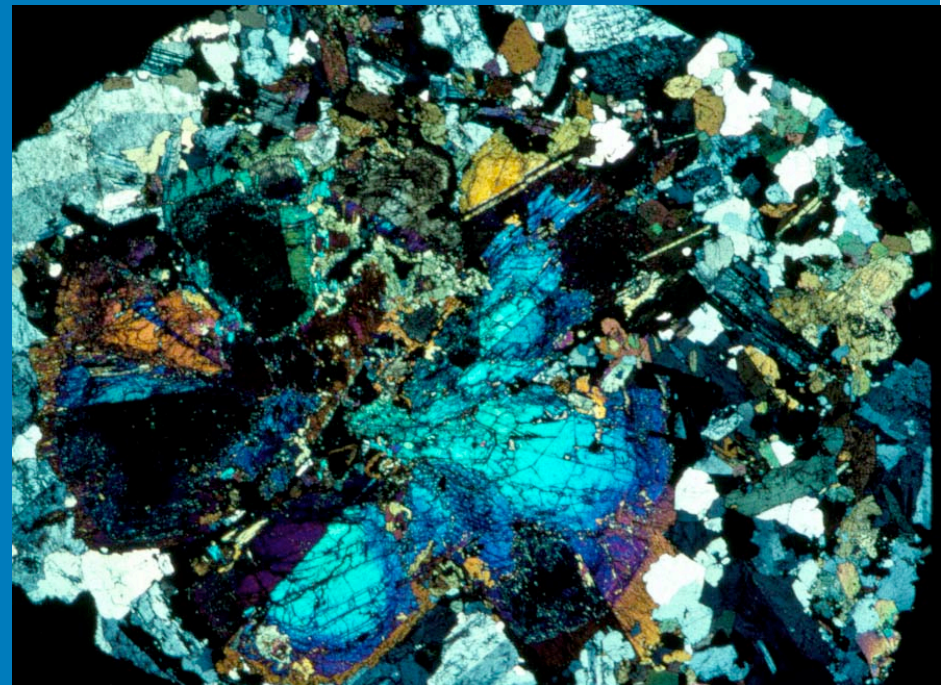
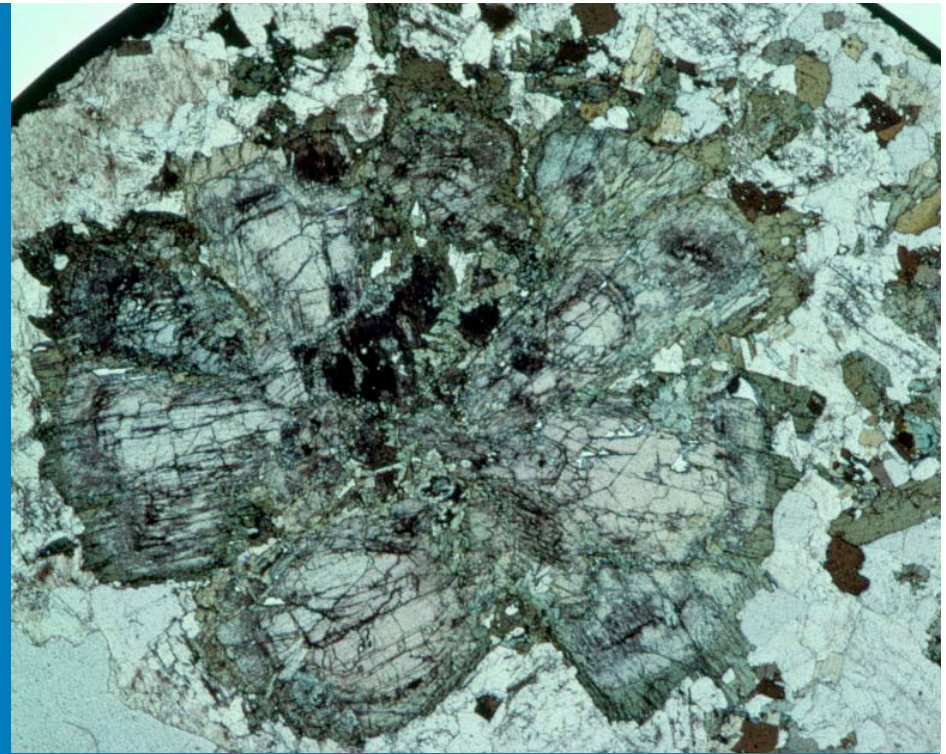
The pyroxene in eclogite is  
A solid solution between  
Augite and  
jadeite





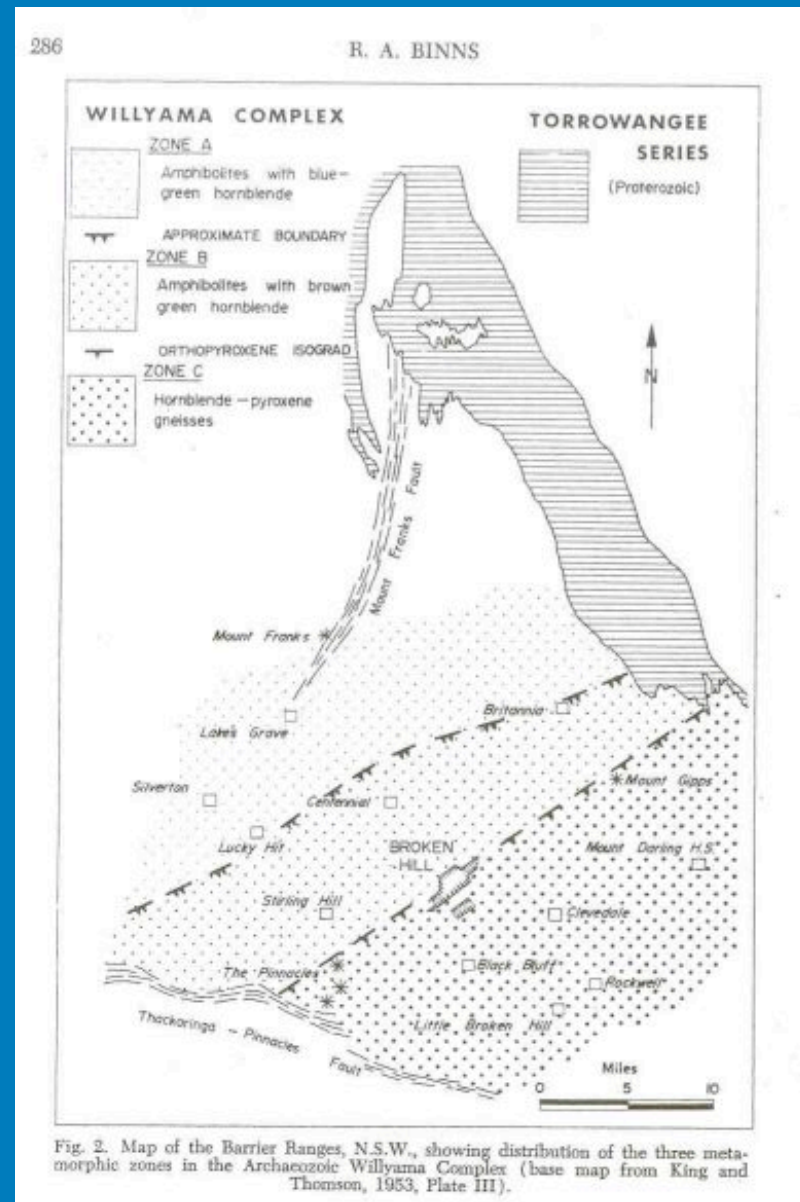
**Radiating aggregate of pyroxene in quartz monzonite with hornblende forming a rim around the pyroxene. Typical bright interference colours of clinopyroxene.**

**Clinopyroxene exhibits inclined extinction for most grains with one cleavage. Most orthopyroxene grains show straight or parallel extinction (black under X-polars) when cleavage (one) is N-S or E-W**



# Pyroxene in Meta- basalts.

- At Broken Hill some metabasalts in zone 2 have clinopyroxene but it is only at the highest grade (Zone 3) that hornblende starts to react out to produce orthopyroxene and clinopyroxene.





# Olivine

## $(\text{Mg,Fe})_2\text{SiO}$

- $\text{Mg}_2\text{SiO}_4$  is forsterite (Fo);  $\text{Fe}_2\text{SiO}_4$  is fayalite (Fa). Complete solid solution.
- Mantle olivine is about Fo90-93.
- Basalts, gabbros have olivine Fo80-40.
- Rare dolerites, ferrogabbros and syenites have fayalite (pale yellow in thin section).
- Meta-dolomitic limestone has Fo100 (dolomite + quartz  $\rightarrow$  forsterite + calcite +  $\text{CO}_2$ )
- Meta banded iron formations Fa100,
- Fo-rich olivine does not co-exist with quartz. Fa-rich does.
- Olivine is not stable in a low temp. wet environment and is replaced by serpentine. This replacement is commonly pseudomorphous (cracks).

