Phytochemistry, Vol. 26, No. 12, p. 3372, 1987. Printed in Great Britain.

GALIPEIN, A COUMARIN FROM GALIPEA TRIFOLIATA

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(Revised received 9 April 1987)

Key Word Index—Galipea trifoliata; Rutaceae; coumarins; ramosin; phebalosin; 7-isopentenyloxy-8-(trans-1',2'-epoxy-3'-methyl-3'-butenyl)coumarin; galipein.

Abstract—In addition to phebalosin and ramosin, the air-dried stem and root barks of *Galipea trifoliata* contain a third previously unreported coumarin identified as 7-isopentenyloxy-8-(*trans*-1',2'-epoxy-3'-methyl-3'-butenyl)coumarin.

INTRODUCTION

In our continuing phytochemical studies of Rutaceae, we report here coumarins from stem and root barks of *Galipea trifoliata*.

RESULTS AND DISCUSSTION

The petrol extracts of the root bark and of the stem bark from *Galipea trifoliata*, on chromatographic separation, each afforded, three coumarins. Two were identified as ramosin (7-isopenteryl oxy-8-isopenterylcoumarin) [1] and phebalosin 7-methoxy-8-(*trans*-1',2'-epoxy-3'methyl-3'-butenylcoumarin) [2]. The third (1), a new natural compound, was isolated as colourless prisms from acetone.

The UV spectrum of 1 in methanol exhibited maxima at 223, 246, 257, 312 (sh) and 324 nm. There was no bathochromic shift on addition with sodium hydroxide. The IR spectrum afforded strong peaks at 1740, 1710, 1605 and 1220 cm⁻¹. This suggested that 1 was a non-phenolic and 7-O-substituted coumarin [3].

In the ₁H NMR spectrum, a pair of doublets at $\delta 6.18$ and 7.62 (each 1H, d, J = 10 Hz) was characteristic of H-3 and H-4 in the coumarin nucleus. Two ortho-coupled aromatic protons at $\delta 7.29$ and 6.85 (each 1H, d, J = 9 Hz) were attributed to H-5 and H-6 in the aromatic nucleus. A 7-isopentenyloxy group was revealed by two methyl singlets at $\delta 1.76$ and 1.78, a doublet at $\delta 4.62$ for methylene protons and a multiplet at $\delta 5.44$ for an olefinic proton. An 8-isoprenoid side chain was characterized by a vinylic methyl group at $\delta 1.86$ and two multiplets at $\delta 5.02$ and 5.26 corresponding to the presence of C=CH₂ as in phebalosin. An AB quartet centred at $\delta 3.93$ (1H, d, J= Hz), 3.97 (1H, d, J = 2 Hz) revealed the presence of the epoxide oxymethine protons, the vicinal coupling constant of 2 Hz indicating a *trans* configuration [4].

The UV, IR and ¹H NMR spectra suggested that 1 was 7-isopentenyloxy-8-(*trans*-1',2'-epoxy-3'-methyl-3'-butenyl) coumarin for which we propose the name of galipein.

EXPERIMENTAL

Materials. Stem bark and root bark of *Galipea trifoliata* Aublet (voucher samples. No. 47, are deposited at the Herbarium of O.R.S.T.O.M. Center of Cayenne in French Guyana) collected near Säul (French Guyana) were sliced, then air-dried and powdered.

Centrifugal thin layer chromatographic (CTLC) separations were carried out using a Chromatotron from Harrison Research. All mps are uncorr. ¹H NMR were measured at 90 MHz in CDCl₃ using TMS as int. standard. EIMS were recorded at 70 eV.

Extraction, isolation and purification. The air-dried powdered material were separately stirred at room temp. with petrol (68–80°). Filtration, followed by removal of petrol, gave a residue which was chromatographed on a silica gel column using hexane containing increasing amounts of EtOAC. The coumarins collected were isolated and purified by CTLC on silicagel with hexane–EtOAC as eluent (9/1 to 7/3).

Isopentenyloxy-8-(trans-1',2'-epoxy-3'-methyl-3'-butenyl)coumarin, mp 88–90°; IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1740, 1710, 1605, 1280, 1220, 1090, 780; MS m/z (rel. int. %): 312 [M]⁺ (2) (C₁₉H₂₀O₄), 244 [M-C₅H₈] (32), 230 [M-C₅H₆O] (50), 215 [244-CHO] (33), 175 [244-C₄H₅O] (100); ¹H NMR: δ 7.29 and 6.85 (each, 1H, d, J = 9 Hz, H-5 and H-6), 7.62 and 6.18 (each, 1H, d, J = 10 Hz, H-4 and H-3), 5.44 (1H, m, 7-0-CH₂-CH =C(Me)₂) 5.26 and 5.02 (each, 1H, m-MeC=CH₂), 4.62 (2H, d, J= 7 Hz, 7-0-CH₂-), 3.97 and 3.89 (each, 1H, d, J = 2 Hz, trans -HC₋CH-). 1.86 (3H, t, MeC=CH₂), 1.78 and 1.76 (each, 3H, s, O

Me).

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b/i/ 0

Carbon no.

1

2

3

4

7

8

9

10

11

12

13

14

15

16

19

22

17, 21

18, 20

5,6

of 6 except for a (-) signal at δ 5.60 due to a methoxy group.

Table 2. ¹³C NMR assignment (attached proton test) of compound 4

EXPERIMENTAL

Mps: uncorr. ¹H NMR was run at (CD₂Cl₂) 100 MHZ and 400 MHz and ¹³C NMR spectra (CD₂Cl₂) at 74.2 MHz. HRMS, ¹³C NMR and micro-analysis were performed at the Department of Chemistry, University of British Columbia, Canada.

Isolation and identification of phenolics. The dried powdered seeds (500 g) of M. dactyloides Gaertn., from Hanguranketa, Sri Lanka, were extracted with Me₂CO. The concd Me₂CO extract was defatted with petrol giving a viscous brown solid (165 g), which after CC on silica gel using petrol: EtOAc mixtures of increasing proportions yielded compounds 1-6, which were purified by prep. TLC.

1-(2,6-Dihydroxyphenyl)tetradecan-1-one (1). Needles mp, 91-91.5° (petrol, lit [3] 91-92); Found: C, 74.89%; H, 9.96%; C20H32O3 requires: C, 74.96%; H, 10.06%, UV 2 max nm, 223, 268, 339, λ_{max}^{EiOĤ/OH-} nm 239, 283, 390; IR ν_{max}^{KBr} 3360 (br), 1640 (s) $^1\mathrm{H}\,\mathrm{NMR}$ (400 MHz, $\mathrm{CD}_2\mathrm{Cl}_2)$ $\delta 9.37$ (2H, br s, OH-16, 20 $\mathrm{D}_2\mathrm{O}$ exchangeable), 7.22 (1H, t, J=8 Hz, H-18), 6.38 (2H, d, J=8 Hz, H-17, 19), 3.12 (2H, t, J = 7 Hz, H-2), 1.69 (2H, m, H-3), 1.27 (20H, m, H-4-13), 0.88 (3H, t, J = 7 Hz, H-14); HRMS, m/z (rel. int.%) 320.2356 ($C_{20}H_{32}O_3$, 10), 302.2246 ($C_{20}H_{30}O_0$, 17), 189.0920 (C₁₂H₁₃O₂, 17), 165.0553 (C₉H₉O₃, 26), 152.0473 (C₈H₈O₃, 31), 137.0247 (C₇H₅O₃, 100) ¹³C NMR assignments are given in Table 1.

Malabaricone A (2). Crystals (petrol) mp, 80-82° (lit. [4] 81-82°).

Malabaricone D (3). Pale yellow crystals (toluene) mp, 89-91° (lit [4] 90-91°).

1-(2,6-Dihydroxyphenyl)-9-(4-hydroxy-3-methoxyphenyl) nonan-1-one (4). White needles (toluene) mp 109-111°; found: C, 71.03%; H, 7.48%, $C_{22}H_{28}O_5$ requires: C, 70.95%; H, 7.58%; UV λ_{max}^{EiOH} nm 340, 272, 269, 220 $\lambda_{max}^{EiOH/OH-}$ nm 386, 283, 238; IR v_{max}^{KBr} cm⁻¹ 3450 (s), 3340 (br), 1635 (s), ¹H NMR (CD₂Cl₂ run at 400 MHz) 89.50 (2H, s, OH-17, 21, D₂O exchangeable), 7.20 (1H, $t, J\,{=}\,8$ Hz, H-19), 6.98 (1H, $d, J\,{=}\,8$ Hz, H-15), 6.73 (1H, d, J=2 Hz, H-11), 6.65 (1H, dd, J=2 and 8 Hz, H-14), 6.37 (2H, d, J =8 Hz, H-20, 18), 5.60 (1H, s, D₂O exchangeable, OH-13), 3.80 (3H, s, H-22), 3.00 (2H, t, J = 7 Hz, H-2), 2.44 (2H, t, J = 7 Hz, H-1)9), 1.64 (4H, m, H-3, 8), 1.32 (8H, m, H-4-7); HRMS m/z (rel. int.%)

Table 1. ¹³C NMR assignment (attached proton test) of compound 1

Carbon no.	$\delta(\text{ppm})$	Intensity
1	207.79	6
2	44.82	32
3	24.41	48
	{ 29.66	90
4-11	29.55	75
	29.40	52
	29.37	· 48 ·
12	31.93	17
13	22.70	23
14	14.13	-14
15	110.03	v. weak
16,20	161.03	15
17,19	108.52 135.57	-100
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PHY

	δ (ppm)	Intensity
	207.95	6
	44.71	41
	24.37	46
	29.12	47
	29.02	69
	28.75	45
	31.25	45
	35.08	61
	136.35	12
	110.58	-48
	145.05	8
۰.	144.52	5

-41

- 39

6

15

-101

-50

-19

372.1947 (C₂₂H₂₈O₅, 52), 262.1568 (C₁₆H₂₂O₃, 21), 234.1618 (C15H22O2, 15), 165.0548 (C9H9O3, 22), 137.0605 (C8H9O2, 100), 137.0244 (C₇H₅O₃, 79); ¹³C NMR assignments are given in Table 2.

114.57

119.93

110.05

161.18

108.42

135.62

56.01

Oxidation of 4. Compound 4 (75 mg) was heated in HOAc (0.3 ml) with one drop of concd H_2SO_4 at 60° for 15 min. Acetylated 4 (30 mg) in Me₂CO was stirred with a soln of KMnO₄ (0.2 g) in H₂O (0.5 ml) and Me₂CO (2 ml), decolourized with NaHSO₃ in dil. H₂SO₄ and extracted with Et₂O. Prep. TLC of the extract on silica gel with the upper layer of toluene-HOAc-H₂O (2:3:1) [5] gave acetylvanillic acid which was characterized by mmp and co-TLC with an authentic sample.

Malabaricone B (5). Pale yellow crystals (toluene) mp 100-102° (lit. [4] 102°).

Malabaricone C (6). Yellow crystals (toluene) mp 122-124° (lit. [4] 123-124°).

Acknowledgements-We thank International Seminar, Uppsala, Sweden for financial assistance to obtain the spectral data; Professor David Dolphin, Department of Chemistry, University of British Columbia, Canada, for ¹³CNMR, HRMS, ¹HNMR and micro-analysis; Dr K K Purushothaman, Captain Sirinivasamurthi Drug Research Institute for Avyurveda, for providing the authentic samples. Dr M H Jayasuriya, Royal Botanical Garden Peradeniya, Sri Lanka, for identification of plant materials and Mrs S D Tennekoon for secretarial assistance.

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