

Kalipyrochlore**(H₂O)(Nb, Ti)₂(O, OH)₆•H₂O**

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Crystal Data: Cubic. *Point Group:* 4/m $\overline{3}$ 2/m. As octahedra, to 1 cm, which may be corroded.

Physical Properties: *Fracture:* [Uneven] (by analogy to the pyrochlore group). *Tenacity:* [Brittle.] *Hardness* = 4–4.5 *D(meas.)* = 3.40–3.48 *D(calc.)* = 3.40–3.44

Optical Properties: Transparent. *Color:* Greenish. *Luster:* [Vitreous to resinous.] *Optical Class:* Isotropic. *n* = 1.93–1.99

Cell Data: *Space Group:* Fd3m. *a* = 10.604(1) *Z* = 8

X-ray Powder Pattern: n.d.

Chemistry:	(1)	(2)	(3)	(1)	(2)	(3)
U ₃ O ₈	0.10	0.08		PbO	0.01	0.02
Nb ₂ O ₅	78.60	80.05	75.69	MgO	0.11	0.11
Ta ₂ O ₅	0.06	0.11		MnO	0.08	0.06
V ₂ O ₅	0.02	0.03		CaO	0.41	0.13
TiO ₂	4.10	4.12	5.01	BaO	0.28	0.38
ZrO ₂	0.66	0.37		SrO	2.60	1.73
SnO ₂		0.06		Na ₂ O	0.35	0.58
ThO ₂	0.12	0.17		K ₂ O	2.73	2.76
Al ₂ O ₃	0.18			F	0.38	0.11
RE ₂ O ₃	0.43	0.50		H ₂ O ⁺	8.54	8.37
Fe ₂ O ₃	0.20	0.13		–O = F ₂	[0.16]	[0.05]
FeO	0.06	0.07		Total	[99.86]	[99.89]
						[100.74]

(1) Lueshe, Congo; original total given as 100.192%, corrected for goyazite 7.8%, ilmenite 2.0%, (rutile, brookite, anatase) 1.3%, calcite 0.9%, kaolinite 0.55%, goethite 0.3%, H₂O^{+130°C}. (2) Do.; original total given as 100.066%, corrected for goyazite 7.5%, ilmenite 1.9%, (rutile, anatase) 1.1%, calcite 1.6%, kaolinite 0.4%, goethite 0.2%, H₂O^{+130°C}. (3) Do.; by electron microprobe, H₂O from crystal-structure analysis; corresponds to [(H₂O)_{0.99} Sr_{0.05} Ca_{0.01}]_{Σ=1.05} (Nb_{1.80} Ti_{0.20})_{Σ=2.00} [O_{4.06}(OH)_{1.94}]_{Σ=6.00} [(H₂O)_{0.86} K_{0.14}]_{Σ=1.00}.

Mineral Group: Pyrochlore group and subgroup; K_A > 20%; (Nb + Ta)_B > 2Ti_B; Nb_B > Ta_B.

Occurrence: In alluvial deposits and residual soils from a carbonatite deposit, formed from pyrochlore by the leaching of Na, Ca, and F in waters rich in K ions.

Association: Na–Ca pyrochlores, lueshite, columbite, fersmite, ilmenite, rutile, barian goyazite.

Distribution: In the Lueshe carbonatite, 150 km north of Goma, Kivu Province, Congo (Zaire).

Name: Originally for a member of the *pyrochlore* group with significant potassium, KALIum, although recent work shows only a minor amount of that element in natural material.

Type Material: National Museum of Natural History, Washington, D.C., USA, 136440.

References: (1) Hogarth, D.D. (1977) Classification and nomenclature of the pyrochlore group. Amer. Mineral., 62, 403–410, esp. 405–406 [kalipyrochlore]. (2) Van Wambeke, L. (1978) Kalipyrochlore, a new mineral of the pyrochlore group. Amer. Mineral., 63, 528–530. (3) Ercit, T.S., F.C. Hawthorne, and P. Černý (1994) The structural chemistry of kalipyrochlore, a “hydropyrochlore”. Can. Mineral., 32, 415–420.

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