THE CRYSTAL STRUCTURE OF ARAPOVITE, U⁴⁺ (Ca,Na)₂ (K_{1-x} \square_x) [Si₈ O₂₀], $x \approx 0.5$, A NEW MINERAL SPECIES OF THE STEACYITE GROUP FROM THE DARA-I-PIOZ MORAINE, TIEN-SHAN MOUNTAINS, TAJIKISTAN

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ABSTRACT

The crystal structure of arapovite, ideally U^{4+} (Ca,Na)₂ (K_{1-x} \square_x) [Si₈ O₂₀], $x \approx 0.5$, tetragonal, a 7.5505(4), c 14.7104(9) Å, V 838.6(1) Å³, space group P4/mcc, Z = 2, D calc. 3.365 g/cm³, was refined to an R_1 index of 2.9% based on 528 observed [$F_0 > 4\sigma F$] unique reflections measured with Mo $K\alpha$ radiation on a Bruker P4 diffractometer with a 1K CCD area detector. In the crystal structure of arapovite, there is one tetrahedrally coordinated site solely occupied by Si, with <Si-O> = 1.617 Å. There are two [8]-coordinated sites, A and B, occupied by ($U^{4+}_{0.59}$ Th_{0.26} Ca_{0.10} REE_{0.04}) and (Ca_{1.23} Na_{0.68} REE_{0.08} Ba_{0.01}), with <A-O> = 2.403 and <B-O> = 2.490 Å. There is one [12]-coordinated site, C, partly occupied by K ($K_{0.52}$ $\square_{0.48}$), with <C-O> = 3.103 Å. In the crystal structure of arapovite, (SiO₄) tetrahedra form a double four-membered ring: [Si₈O₂₀]. Eight-coordinated A and B polyhedra share common edges to form a (001) sheet. The sheets are connected through [Si₈O₂₀] groups to form a framework; C atoms are situated in large cages within the framework. The topology of the arapovite structure is identical to that of turkestanite, Th (Ca,Na)₂ (K_{1-x} \square_x) Si₈ O₂₀ (H₂O)_n, and steacyite, Th (Na,Ca)₂ (K_{1-x} \square_x) Si₈ O₂₀, $x \approx 0.5$. The topology of silicate minerals with [Si₈O₂₀]⁸-units, including ekanite, Th Ca₂ Si₈O₂₀, and litidionite, Na₂ K₂ Cu₂ Si₈O₂₀, is discussed.

Keywords: arapovite, U⁴⁺ silicate, crystal structure, [Si₈O₂₀] double ring, Dara-i-Pioz, Tajikistan.

Sommaire

Nous avons affiné la structure cristalline de l'arapovite, dont la formule idéale est U^{4+} (Ca,Na) $_2$ ($\mathrm{K}_{1-x}\square_x$) [Si₈ O₂₀], $x\approx0.5$, tétragonale, a 7.5505(4), c 14.7104(9) Å, V 838.6(1) Å 3 , groupe spatial P4/mcc, Z=2, D (calc.) 3.365 g/cm^3 , jusqu'à un résidu R_1 de 2.9% en utilisant 528 réflexions distinctes observées [$F_0 > 4\sigma F$] et mesurées avec rayonnement Mo $K\alpha$ et un diffractomètre Bruker P4 muni d'un détecteur à aire 1K de type CCD. Dans la structure de l'arapovite, il y a un site à coordinence tétraédrique uniquement occupé par le Si, avec <Si-O> = 1.617 Å. Il y a deux sites à coordinence [8], A et B, Q0 cocupent ($Q^{4+}_{0.59}$ Th_{0.26} Ca_{0.10} TR_{0.04}) et (Ca_{1.23} Na_{0.68} TR_{0.08} Ba_{0.01}), avec <A-O> = 2.403 et <B-O> = 2.490 Å. Il y a un site à coordinence [12], C, partiellement rempli par le K ($X_{0.52}$ $Z_{0.48}$), avec <C-O> = 3.103 Å. Dans cette structure, les tétraèdres (SiO₄) forment un anneau double à quatre membres: [Si₈O₂₀]. Les polyèdres à huit coins renfermant A et B partagent des arêtes communes pour former un feuillet (001). Ces feuillets sont connectés grâce aux groupes [Si₈O₂₀] pour former une trame; les atomes de C occupent des cages volumineuses dans cette trame. La topologie de la structure de l'arapovite est identique à celle de la turkestanite, Th (Ca,Na) $_2$ (X_{1-x}) Si₈ O₂₀ (X_{1-x}) Si₈ O₂₀, X_{1-x} et elle de la steacyite, Th (Na,Ca) $_2$ (X_{1-x}) Si₈ O₂₀, et la litidionite, Na $_2$ K $_2$ Cu $_2$ Si₈O₂₀.

(Traduit par la Rédaction)

Mots-clés: arapovite, silicate de U⁴⁺, structure cristalline, anneau double [Si₈O₂₀], Dara-i-Pioz, Tajikistan.

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Introduction

Arapovite, U⁴⁺ (Ca,Na)₂ (K_{1-x} \square_x) [Si₈ O₂₀], $x \approx 0.5$, is a new mineral species from the moraine of the Darai-Pioz glacier, the Alai mountain ridge, Tien-Shan Mountains, northern Tajikistan (Agakhanov et al. 2004). Arapovite is the seventh silicate with a double-ring module [Si_nO_{2.5n}] described from the Dara-i-Pioz moraine. There are three types of double rings of tetrahedra known for silicates. A three-membered double ring $[Si_6O_{15}]$ has been recently described in moskvinite-(Y), Na₂ K Y Si₆ O₁₅, from the same locality (Sokolova et al. 2003). Four-membered [Si₈O₂₀] double-rings occur in three minerals with the general formula ${}^{[8]}A {}^{[8]}B_2 {}^{[12]}C$ $[Si_8O_{20}]$ (H₂O)_n, where A = Th, REE; B = Ca, Na; C = K, \square (vacancy): steacyite, Th (Na,Ca)₂ (K_{1-x} \square _x) Si₈ O₂₀ (H₂O)_n (Richard & Perrault 1972, Perrault & Szymański 1982), iraqite-(La), REE (Na,Ca)₂ ($K_{1-x}\square_x$) $Si_8 O_{20} (H_2O)_n$ (Livingstone et al. 1976), and turkestanite, Th (Ca,Na)₂ (K_{1-x} \square_x) Si₈ O₂₀ (H₂O)_n, x =0.5 (Kabalov et al. 1998); the crystal structure of iraqite-(La) has not yet been refined. Six-membered doublerings, ideally [Si₁₂O₃₀], occur in the 18 minerals of the milarite group with the general formula ${}^{[6]}A_2 {}^{[9]}B_2 {}^{[12]}C$ $^{[18]}D^{[4]}T(2)_3^{[4]}T(1)_{12} O_{30}$, where T(1) = Si, Al; $T(2) = T(1)_{12}^{[4]}T(1)_{12}^{[4]}$ Li, Be, B, Mg, Al, Si, Mn^{2+} , Zn; A = Al, Fe^{3+} , Sn^{4+} , Mg, Zr, Fe²⁺, Ca, Na, Y, REE; B = Na, H₂O, K, \square ; C = K, Na, Ba, \square ; $D = \square$ (Forbes *et al.* 1972, Hawthorne *et al.* 1991). Here, we report the crystal structure and crystal chemistry of arapovite.

TABLE 1. CHEMICAL COMPOSITION (wt.%) AND UNIT FORMULA* (apfu) FOR ARAPOVITE

SiO ₂	55.08	Si	8.00
UO ₂	18.17	ΣT	8.00
ThO ₂	7.75		
CaO	8.43	U ⁴⁺	0.59
Dy_2O_3	0.35	Th	0.26
Sm_2O_3	0.23	Ca	0.10
Pr_2O_3	0.19	Dy	0.02
Na₂O	2.39	Sm	0.01
Nd_2O_3	0.92	Pr	0.01
Ce_2O_3	0.56	ΣΑ	0.99
BaO	0.18		
K₂O	2.78	Ca	1.23
Σ	97.03	Na	0.68
		Nd	0.05
		Ce	0.03
		Ba	0.01
		ΣΒ	2.00
		K	0.52
		ΣC	0.52
		_	
		0	20

^{*} The structural formula calculated on the basis of 8 Si

ELECTRON-MICROPROBE ANALYSIS

The chemical composition of arapovite (Table 1) was established on annealed material with a Cameca SX 100 electron microprobe operating in wavelength-dispersion mode with an accelerating voltage of 15 kV, a specimen current of 15 nA, a beam size of 20 µm and counting times on peak and background of 20 and 10 s, respectively. The following standards and crystals were used for *K*, *L* or *M* X-ray lines: Na: albite; Si, Ca: diopside; Al, K: orthoclase; Th: ThO₂; Sr: strontianite; Ba: barite; U: UO₂; Pb: PbTe. Data were reduced using the X–PHI procedure (Merlet 1992).

Concentrations of REE elements were determined titrimetrically after chromatographic partition on unannealed material. On the basis of 8 Si apfu (atoms per formula unit), the following chemical formula for arapovite was obtained: $(U^{4+}_{0.59} \, Th_{0.26} \, Ca_{0.10} \, Dy_{0.02} \, Sm_{0.01} \, Pr_{0.01})_{\Sigma 0.99} \, (Ca_{1.23} \, Na_{0.68} \, Nd_{0.05} \, Ce_{0.03} \, Ba_{0.01})_{\Sigma 2.00} \, (K_{0.52} \, \Box_{0.48})_{\Sigma 1.00} \, Si_8 \, O_{20}$, with an ideal formula of U^{4+} (Ca,Na)₂ $(K_{1-x} \, \Box_x)$ [Si₈ O_{20}], $x \approx 0.5$.

COLLECTION OF X-RAY DATA AND STRUCTURE REFINEMENT

The highly metamict natural crystals of arapovite gave an indistinct X-ray-diffraction pattern, and it was impossible to refine the structure. Arapovite was annealed for 3.5 hours up to 1100° C in air. Single-crystal X-ray-diffraction data for an annealed single crystal of arapovite were collected with a Bruker P4 diffractometer equipped with Mo $K\alpha$ X-radiation and a 1K CCD detector. Integrated intensities of 8780 reflections with $\overline{10} < h < 10$, $\overline{10} < k < 10$, $\overline{20} < l < 20$ were collected up to $2\theta = 60.08^{\circ}$ using 60 s per frame. The refined cell-parameters (Table 2) were obtained from 4657 reflections (I > 10 oI). An empirical absorption correction (SADABS, Sheldrick 1998) was applied.

The crystal structure of arapovite was refined to the final R_1 index of 2.9% using atom coordinates of turkestanite (Kabalov et al. 1998) as the initial model and the SHELXTL 5.1 system of programs (Sheldrick 1997). Scattering factors for neutral atoms were taken from the International Tables for X-ray Crystallography (1992). Site occupancies were refined for the A, B and C sites with the scattering curves of U, Ca and K. Details of the X-ray data collection and structure refinement are given in Table 2. Residual weak maxima [two maxima of 3.4 e and 1.1 e located 0.75 Å and 1.74 Å from the U site, respectively, and one maximum of 1.1 e located 0.87 Å from the K site] were present in difference-Fourier maps calculated at the final stages of refinement. Final atom-parameters for anapovite are given in Table 3, selected interatomic distances are presented in Table 4, refined site-scattering values are given in Table 5, and a bond-valence analysis is shown in Table 6. Structure factors may be obtained from The Depository of Unpublished Data, CISTI, National Research Council, Ottawa, Ontario K1A 0S2, Canada.

DESCRIPTION OF THE CRYSTAL STRUCTURE

Cation coordination

There is one Si site, tetrahedrally coordinated by O atoms and fully occupied by Si with an <Si-O> distance of 1.617 Å. The shortest Si-O distance, Si-O(3) = 1.577(3) Å, involves a bond to a non-bridging O atom (Table 6), whereas all other O atoms are bonded to two Si atoms. There are two sites, A and B, coordinated by eight O atoms with <A-O> and <B-O> distances of 2.403 and 2.490 Å, respectively. The total refined scattering at the A and B sites is 79.4(4) and 37.1(3) epfu (electrons per formula unit) and the analogous scattering calculated from the unit formula (Tables 1, 5) is 82.2 and 37.4 epfu, respectively. Thus the A and B site-populations were assigned as 0.59 U⁴⁺ + 0.26 Th + 0.10 Ca + 0.02 Dy + 0.01 Sm + 0.01 Pr and 1.23 Ca + 0.68 Na +

TABLE 2. MISCELLANEOUS REFINEMENT DATA FOR

ARAPOVITE				
a (Å)	7.5505(4)			
c	14.7104(9)			
V (Å ³)	838.6(1)			
Space group	P4/mcc			
Z	2			
Absorption coefficient (mm ⁻¹)	9.64			
F(000)	799.0			
$D_{\text{calc.}}(g/\text{cm}^3)$	3.365			
Crystal size (mm)	0.120 x 0.100 x 0.080			
Radiation/filter	MoKα/graphite			
2θ-range for data collection (°)	60.08			
R(int) (%)	7.3			
Reflections collected	8780			
Independent reflections $F_o > 4\sigma F_o $	649 528			
Refinement method	Full-matrix least squares on F^2 , fixed weights \propto to 1/ σ $ F_{\sigma}^2 $			
Goodness of fit on F2	1.064			
Final $R_{\text{(obs)}}(\%)$ $[F_o > 4\sigma F_o]$	$R_1 = 2.9$			
R indices (all data) (%)	$R_1 = 4.3$ $wR_2 = 7.6$ GoF = 1.064			

0.05 Nd + 0.03 Ce + 0.01 Ba, following the unit formula (Table 1). There is one C site coordinated by twelve O atoms, with a <C-O> distance of 3.103 Å. The [12]-coordinated C site is partly occupied by K: 0.52 K + 0.48 \square .

Structure topology

Eight (SiO₄) tetrahedra share corners to form a four-membered double-ring of the form [Si₈O₂₀]. The (AO_8) and (BO_8) polyhedra share edges to form a sheet parallel to (001) (Fig. 1a). In the sheet, each A polyhedron shares common edges with four B polyhedra, and each B polyhedron shares common edges with two A polyhedra, the ratio A:B being 1:2. These sheets are linked in the c direction by [Si₈O₂₀] double-rings, forming a mixed-polyhedron framework (Fig. 1b). Large cages of the framework occlude [12]-coordinated C sites (Fig. 1c). The C cations are situated at the same level along the c direction as the [Si₈O₂₀] double-rings. The topology of the arapovite structure is identical to that of steacyite (Richard & Perrault 1972) and turkestanite (Kabalov et al. 1998).

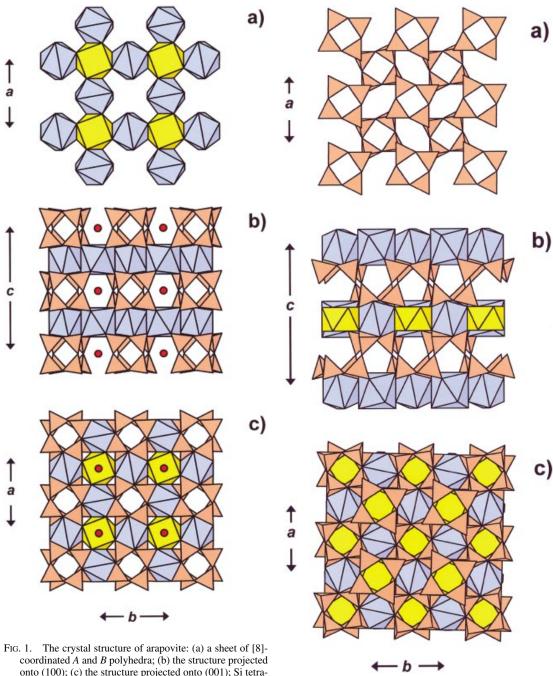
RELATED MINERALS

Other minerals with [Si₈O₂₀] units

Two minerals contain a silicate unit of different bond-topology but the same stoichiometry as that in arapovite: ekanite, Th Ca₂ [Si₈O₂₀], a 7.483(3), c 14.893(6) Å, V 833.9 Å³, Z = 2, space group I422 (Szymański et al. 1982), and litidionite, Na₂ K₂ Cu₂ $[Si_8O_{20}]$, a 9.800(10), b 8.010(10), c 6.970(10) Å, α 114.1(1), β 99.5(1), γ 105.6(1)°, V 456.8 Å³, Z = 2, space group $P\overline{1}$ (Pozas *et al.* 1975). The [Si₈O₂₀] units in these minerals are a sheet and a chain, respectively. The sheet in the crystal structure of ekanite consists of four- and eight-membered rings of tetrahedra (Fig. 2a). Polyhedra containing [8]-coordinated Th and Ca form a sheet identical to that in the structure of arapovite (Fig. 1a). In the crystal structure of ekanite, $[Si_8O_{20}]$ sheets and sheets of Th and Ca polyhedra alternate along the c direction (Fig. 2b). Figure 2c shows a general view of the mixed framework in the crystal structure of ekanite,

TABLE 3. FINAL ATOM POSITIONS AND DISPLACEMENT PARAMETERS (Ų) FOR ARAPOVITE

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	х	у	Z	U_{eq}	U ₁₁	U ₂₂	U ₃₃	U ₂₃	U ₁₃	U ₁₂
Α	0	0	1/4	0.0097(2)	0.0071(2)	0.0071(2)	0.0149(3)	0	0	0
В	0	1/2	1/4	0.0141(6)	0.0134(9)	0.0124(9)	0.0165(9)	0	0	0
C	0	0	0	0.0339(13)	0.0288(18)	0.0288(18)	0.0442(32)	0	0	0
Si	0.2594(2)	0.3354(2)	0.10846(7)	0.0107(3)	0.0093(6)	0.0097(6)	0.0132(6)	0.0001(4)	0.0008(4)	0.0001(4)
O(1)) -0.2362(6)	0.6845(8)	0	0.0229(11)	0.0263(27)	0.0295(28)	0.0129(20)	0	0	0.0000(22)
0(2)	0.4538(4)	0.2567(5)	0.1361(2)	0.0172(8)	0.0105(16)	0.0191(17)	0.0221(16)	0.0036(13)	0.0004(12)	0.0005(12)
O(3)) -0.1026(4)	0.2508(4)	0.3357(2)	0.0157(7)	0.0130(16)	0.0138(15)	0.0202(14)	-0.0026(13)	0.0037(13)	-0.0001(12)



onto (100); (c) the structure projected onto (001); Si tetrahedra are orange, A and B polyhedra are yellow and blue, Catoms are shown as red circles.

Fig. 2. The crystal structure of ekanite: (a) the Si-O sheet; (b) the structure projected onto (100); (c) the structure projected onto (001); Si tetrahedra are orange, [8]-coordinated Ca and Th polyhedra are blue and yellow, respectively.

formed through linkage of (SiO₄) tetrahedra and [8]-coordinated polyhedra.

The chain $[\mathrm{Si}_8\mathrm{O}_{20}]^{8-}$ in the crystal structure of litidionite is a combination of four- and eight-membered rings (Fig. 3a). Polyhedra containing [5]-coordinated Ca and polyhedra containing [7]-coordinated Na share common edges to form chains along [001] (Figs. 3b, c). The (SiO₄) tetrahedra and (CaO₅) and (NaO₇) polyhedra share common vertices to form a mixed-polyhedron framework (Fig. 3c).

Distribution of minerals of $(Si_nO_{2.5n})$ stoichiometry

There are twenty-three minerals that contain double rings of tetrahedra of the form $[Si_nO_{2.5n}]$, where n = 3, 4and 6. There are also eleven minerals that have similar stoichiometries but bond topologies not involving silicate rings. As described above, six minerals have nonring [Si₈O₂₀] units, and four minerals have [Si₈O₂₀] double-rings. The 18 minerals of the milarite-group have six-membered [Si₁₂O₃₀] double-rings in their crystal structure. There is only one mineral with the same stoichiometry as milarite, but with a different topology: miserite, $K_2 Ca_{10} (Y,REE)_2 [Si_2O_7]_2 [Si_12O_{30}] (OH)_2 F_2$, contains a chain of the form [Si₁₂O₃₀] (Scott 1976). A unit of the form [Si₆O₁₅] occurs in nine minerals. A three-membered double-ring occurs in the crystal structure of moskvinite-(Y), Na₂ K (Y,REE) [Si₆O₁₅]. A chain of the form [Si₆O₁₅] occurs in the crystal structures of five minerals. An extended (or elpidite) chain was described in crystal structures of elpidite, Na₂ Zr [Si₆O₁₅] (H₂O)₃ (Cannillo et al. 1973) and epididymite,

TABLE 4. SELECTED INTERATOMIC DISTANCES (Å) AND ANGLES (°) IN ARAPOVITE

Si-O(1)	1.612(1)	A-O(3) x8	2.403(3)		
Si-O(2)	1.635(4)				
Si-O(2)a	1.643(3)	B-O(2) x4	2.586(3)		
Si-O(3)	1.577(3)	B-O(3) x4	2.393(3)		
<si0></si0>	1.617	<bo></bo>	2.490		
Sib-O(1)-Sic Si-O(2)-Sid	163.6(4) 143.5(2)	C-O(1) x4 C-O(3) x8 <c-o></c-o>	2.976(6) 3.167(3) 3.103		

a: y, -x+1, z; b: -x, -y+1, -z; c: -x, -y+1, z; d: -y+1, x, z

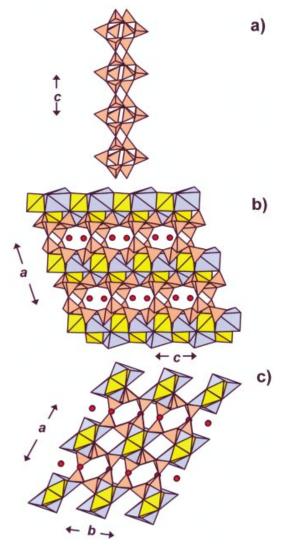


Fig. 3. The crystal structure of litidionite: (a) the Si–O chain; (b) the structure viewed down the *b* axis; (c) the structure viewed down the *c* axis. The [5]-coordinated Ca polyhedra are yellow, the [7]-coordinated Na polyhedra are blue, the Si tetrahedra are orange, and the K atoms are shown as red circles.

TABLE 5. REFINED SITE-SCATTERING VALUES (epfu) AND ASSIGNED SITE-POPULATIONS (apfu)
FOR ARAPOVITE

	Refined site- scattering	Site population	Calculated site-scattering	<x-o>_{calc}*</x-o>	<x-o>_{obs} (Å)</x-o>
Α	79.4(4)	0.59 U ⁴⁺ + 0.26 Th + 0.10 Ca + 0.02 Dy + 0.01 Sm + 0.01 P	r 82.2	2.28	2.40
В	37.1(4)	1.23 Ca + 0.68 Na + 0.05 Nd + 0.03 Ce + 0.01 Ba	37.4	2.49	2.49
С	9.1(1)	0.52 K + 0.48 □	9.9	3.02	3.10

^{*}Calculated by summing constituent ionic radii; values from Shannon (1976)

TABLE 6. BOND-VALENCES* (vu) FOR ARAPOVITE						
Α	В	С	Si	Σ		
		0.041 ^{x4} 1	1.027 ^{x2} →	2.095		
	0.186 ^{x4} 1		0.966 0.946	2.098		
0.450 ^{x8} ↓	0.277*41	0.023 ^{x8} 1	1.128	1.878		
3.603	1.853	0.348	4.067			
	0.450 ^{x8} 1	A B 0.186 ^{x4} 1 0.450 ^{x8} 1 0.277 ^{x4} 1	A B C 0.041*4 i 0.186*4 i 0.450*8 i 0.277*4 i 0.023*8 i	A B C Si 0.041**1 1.027**2- 0.186**1 0.966 0.946 0.946 0.450**1 0.277**1 0.023**8 1.128		

^{*} Bond-valence curves are from Brown (1981)

Na₂ Be₂ [Si₆O₁₅] (H₂O) (Pobedimskaya & Belov 1956). A kinked (or tuhualite) chain was described in the crystal structures of tuhualite, (Na,K) Fe²⁺ Fe³⁺ [Si₆O₁₅] (Merlino 1969), zektzerite, Na Li Zr [Si₆O₁₅] (Ghose & Wan 1978), and emeleusite, Na₂ Li Fe³⁺ [Si₆O₁₅] (Johnsen *et al.* 1978). A sheet of the form [Si₆O₁₅], with of 4-, 6- and 8-membered rings of tetrahedra, occurs in crystal structures of armstrongite, Ca Zr [Si₆O₁₅] (H₂O)_{2.5} (Kashaev & Sapozhnikov 1978), dalyite, K₂ Zr [Si₆O₁₅] (Fleet 1965), and davanite, K₂ Ti⁴⁺ [Si₆O₁₅] (Gebert 1983).

Figure 4 shows the distribution of minerals of $(Si_nO_{2.5n})$ stoichiometry. Where n = 3, double-rings are rare and the structure topology is dominated by other connectivities. Where n = 4, there is a similar number of double-rings as other connectivities. Where n = 6, double-rings dominate the topology of the structure, and other connectivities are rare.

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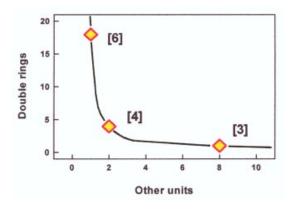


Fig. 4. The distribution of minerals of $(Si_nO_{2.5n})$ stoichiometry; n is shown in square brackets.

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