

Lecture 28 (12/1/21)

TODAY

- Reading: Ch11; 370-381
- Problems: Ch11; 5,6,8,9,12,13,15

NEXT

- Reading: Ch7; 220-235
- Problems: Ch7; 1,2,3,5,9

Lipids & Membranes

A. Lipids

1. Roles
2. Classes
 - a. Fatty Acids
 - b. Fats
 - c. Waxes
 - d. Membrane lipids
 - e. Terpenes

B. Membranes

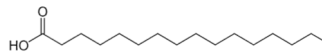
1. Introduction
2. The 4 S's
 - a. Size
 - b. Solubility
 - c. Shape
 - d. Stability
3. Models for Membrane structure
 - a. Old Model
 - b. Data
 - c. Fluid Mosaic Model
 - d. Testing the model
4. The Red-Blood Cell Membrane
5. Membrane Asymmetry
 - a. transverse
 - b. lateral
 - c. anchoring
6. Membrane Fluidity

Lipids: Classes

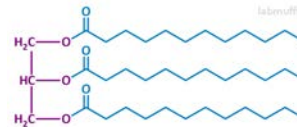
Biological molecules that are characterized by low solubility in water, that is, are relatively hydrophobic.

Classes of Lipids They have a high hydrocarbon content

1. Fatty acids

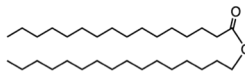


2. Fats (triglycerides)

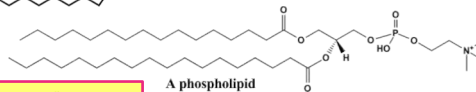


Structure of a fat

3. Waxes

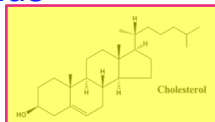


4. Membrane Lipids



A phospholipid

5. Isoprenes



Cholesterol

Lipids: Fatty Acids

Common Biological Fatty Acids

Number of Carbons	Common Name	Systematic Name	Symbol	Structure
Saturated fatty acids				
12	Lauric acid	Dodecanoic acid	12:0	$\text{CH}_3(\text{CH}_2)_{10}\text{COOH}$
14	Myristic acid	Tetradecanoic acid	14:0	$\text{CH}_3(\text{CH}_2)_{12}\text{COOH}$
16	Palmitic acid	Hexadecanoic acid	16:0	$\text{CH}_3(\text{CH}_2)_{14}\text{COOH}$
18	Stearic acid	Octadecanoic acid	18:0	$\text{CH}_3(\text{CH}_2)_{16}\text{COOH}$
20	Arachidic acid	Eicosanoic acid	20:0	$\text{CH}_3(\text{CH}_2)_{18}\text{COOH}$
22	Behenic acid	Docosanoic acid	22:0	$\text{CH}_3(\text{CH}_2)_{20}\text{COOH}$
24	Lignoceric acid	Tetracosanoic acid	24:0	$\text{CH}_3(\text{CH}_2)_{22}\text{COOH}$
Unsaturated fatty acids (all double bonds are cis)				
16	Palmitoleic acid	9-Hexadecenoic acid	16:1 (Δ^9)	$\text{CH}_3(\text{CH}_2)_5\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$
18	Oleic acid	9-Octadecenoic acid	18:1 (Δ^9)	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$
18	Linoleic acid	9,12-Octadecadienoic acid	18:2 ($\Delta^9,12$)	$\text{CH}_3(\text{CH}_2)_4(\text{CH}=\text{CHCH}_2)_2(\text{CH}_2)_6\text{COOH}$
18	α -Linolenic acid	9,12,15-Octadecatrienoic acid	18:3 ($\Delta^9,12,15$)	$\text{CH}_3\text{CH}_2(\text{CH}=\text{CHCH}_2)_5(\text{CH}_2)_6\text{COOH}$
18	γ -Linolenic acid	6,9,12-Octadecatrienoic acid	18:3 ($\Delta^{6,9,12}$)	$\text{CH}_3(\text{CH}_2)_4(\text{CH}=\text{CHCH}_2)_3(\text{CH}_2)_3\text{COOH}$
20	Arachidonic acid	5,8,11,14-Eicosatetraenoic acid	20:4 ($\Delta^{5,8,11,14}$)	$\text{CH}_3(\text{CH}_2)_4(\text{CH}=\text{CHCH}_2)_4(\text{CH}_2)_5\text{COOH}$
24	Nervonic acid	15-Tetracosenoic acid	24:1 (Δ^{15})	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_{13}\text{COOH}$

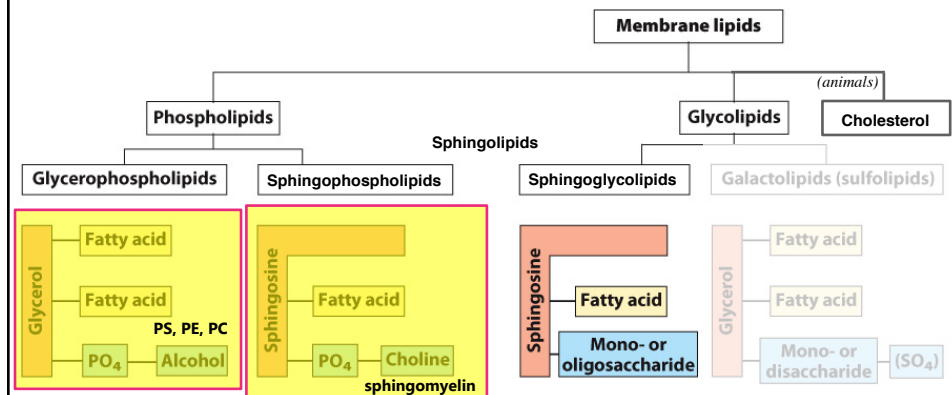
Need to Know: Common names, structure, symbol (e.g., 18:3 $\Delta^{9,12,15}$)

Lipids: Membrane Lipids

Classification of Membrane Lipids

Two major categories based on the structure and function:

1. Lipids that contain phosphate
2. Lipids that do not contain phosphate
 - each can be further separated into:
 - Glycerol-based and sphingosine-based



Lipids: Membranes

Introduction

The 4 S's

Size
Solubility
Shape
Stability

Models for Membrane structure

Old Model
Data
Fluid Mosaic Model
Testing the model

The Red-Blood Cell Membrane

Membrane Asymmetry

Lipids
transverse
lateral
Protein
anchoring
glycoproteins

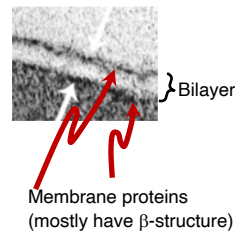
Membrane Fluidity

Lipids: Membranes

Models for Membrane Structure

OLD MODEL (ca. 1940-1970)
Sandwich model proposed by Danielli-Davson.

Based on the structures in the EM



Scientifically, this is a good MODEL because it is clearly TESTABLE!

This model makes several testable predictions:

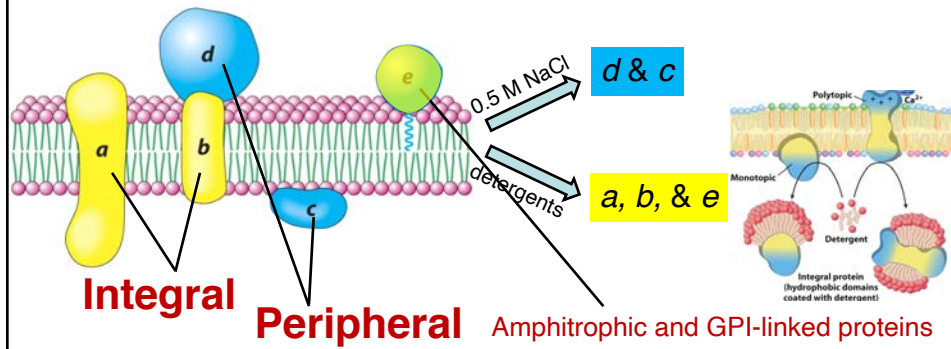
- 1) Protein-lipid interactions should be mostly electrostatic; proteins should have lots of charged groups.
- 2) Should be able to "wash" nearly all membrane proteins off the membranes with high salt.
- 3) Isolated membrane proteins should show lots of β -structure
- 4) Importantly, NO PROTEINS ON THE INSIDE

Lipids: Membranes

Models for Membrane Structure

TESTING OLD MODEL: DATA

- 1) & 2) Wash isolated membranes with high-salt solutions or changes in pH.
- Removes some but not all proteins
 - This leads to an operational definition of **peripheral** (those that wash off with 0.5 M salt), an **integral** (those that remain after washing) **membrane proteins**

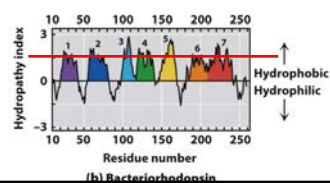
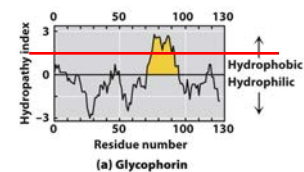
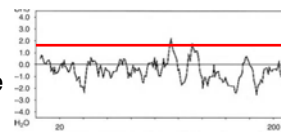
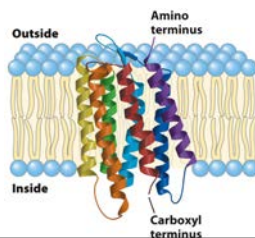


Lipids: Membranes

Models for Membrane Structure

TESTING OLD MODEL: DATA

- 3) Isolated membrane proteins should show lots of β -structure.
- Peripheral membrane proteins looked like cytosolic proteins
 - CD showed there was actually more α -helix than β -structure
 - Integral membrane proteins had patches of hydrophobic residues in their sequence



Lipids: Membranes

Models for Membrane Structure

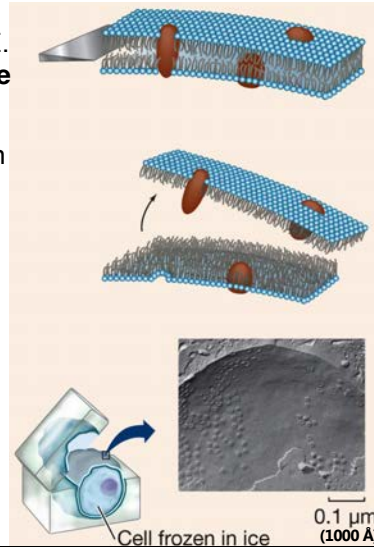
TESTING OLD MODEL: DATA

4) Importantly, NO PROTEINS ON THE INSIDE.

- So, lets look: performed **Freeze-fracture EM** on cell membranes
- This immediately became an explanation for Integral membrane proteins.

OMG!!!
NOT smooth inside!

Oops, maybe
proteins DO span the
membrane.



Lipids: Membranes

Models for Membrane Structure

NEW MODEL (1972)

Fluid Mosaic Model proposed by
SJ Singer & GL Nicholson

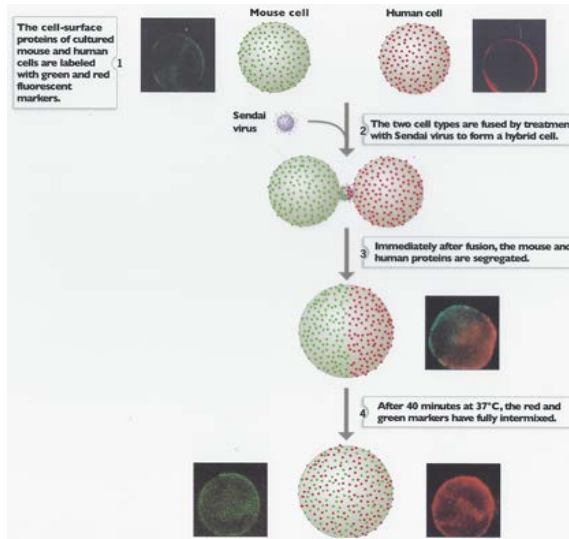
- Lipids form a viscous, two-dimensional solvent into which proteins are inserted and integrated more or less deeply.
- Proteins can either be embedded in or associated with the membrane:
 - Integral membrane proteins are firmly associated with the membrane, often spanning the bilayer.
 - Peripheral membrane proteins are weakly associated and can be removed easily.
 - Some are non-covalently attached.
 - Some are linked to membrane lipids (amphitrophic)(more later).



This model was also testable!

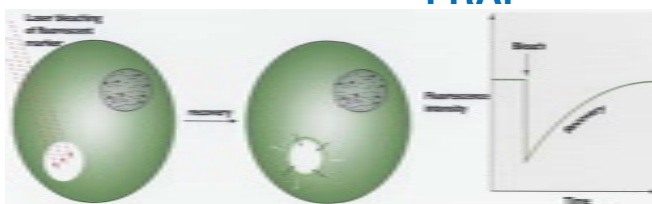
Lipids: Membranes

Testing Fluid Mosaic Model of Membrane Structure: Cellular Fusion



Lipids: Membranes

Testing Fluid Mosaic Model of Membrane Structure: FRAP

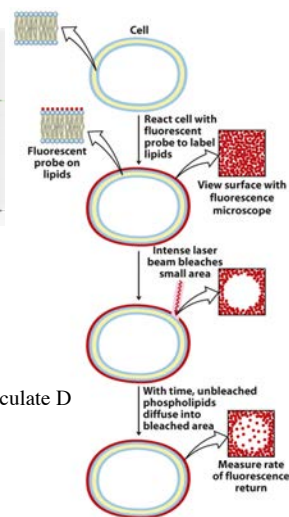


- **F**luorescence **R**ecovery **A**fter **P**hotobleaching (FRAP) allows us to monitor lateral diffusion by monitoring the rate of fluorescence return.
- From the rate of return of fluorescently labeled lipids, the rate of diffusion of a lipid in the leaflet can be determined.

Distance = $\sqrt{4Dt}$ Set distance by radius of laser target area, measure t , calculate D
 D is rate of diffusion in m^2/sec

- Rates of lateral diffusion are high (up to $1 \mu m^2/sec$).
 - A lipid can circumnavigate an *E.coli* cell in one second.
 - A protein has a variable rate: 0.4 to $0.0001 \mu m^2/sec$.

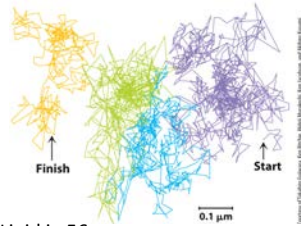
..... but not so fast!



Lipids: Membranes

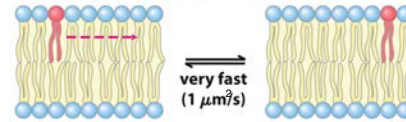
Testing Fluid Mosaic Model of Membrane Structure: FRAP

Lateral Movement is fast

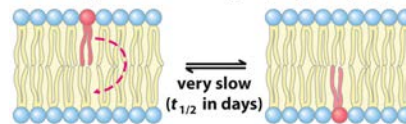


Transverse Movement is SLOW

Uncatalyzed lateral diffusion



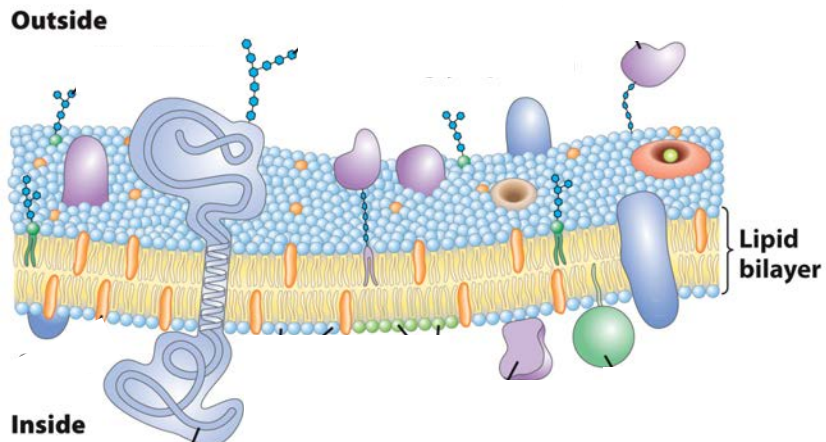
Uncatalyzed transbilayer ("flip-flop") diffusion



Spontaneous flips from one leaflet to another are rare because the charged head group must transverse the hydrophobic tail region of the membrane.

Lipids: Membranes

The Fluid Mosaic Model: Details



Lipids: Membranes

Introduction

The 4 S's

- Size
- Solubility
- Shape
- Stability

Models for Membrane structure

- Old Model
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- Testing the model

The Red-Blood Cell Membrane

Membrane Asymmetry

Lipids

- transverse
- lateral

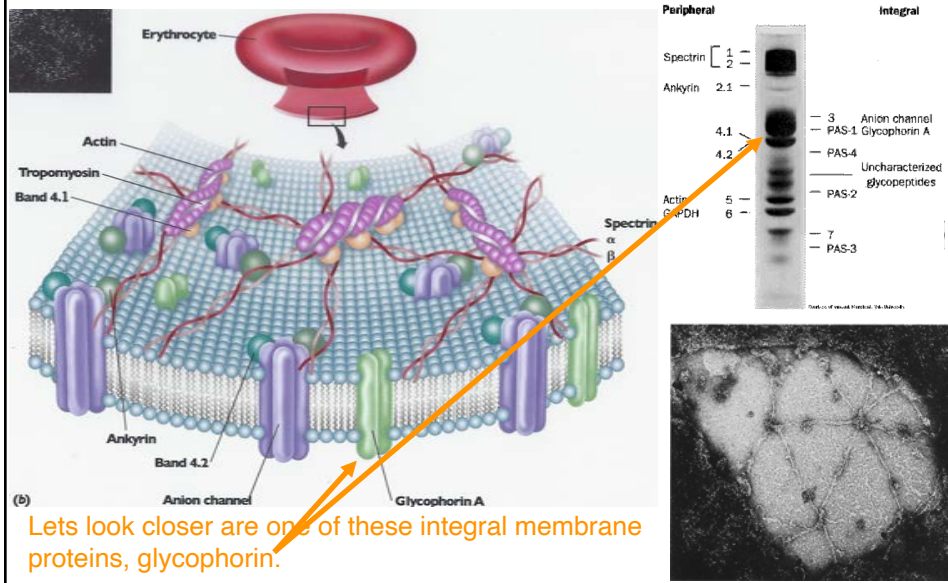
Protein

- anchoring
- glycoproteins

Membrane Fluidity

Lipids: Membranes

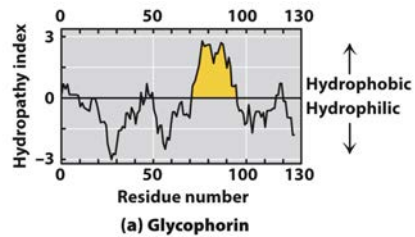
The Structure of the Red-blood Cell Membrane



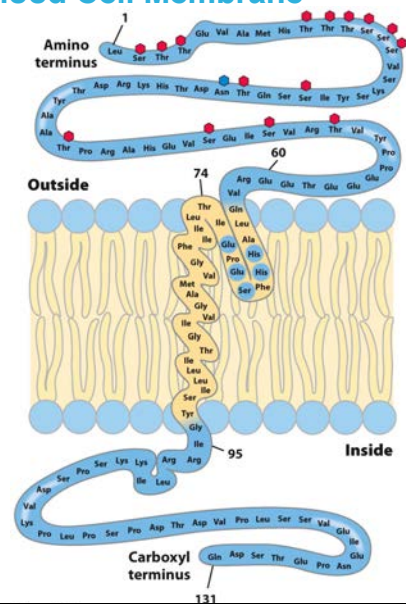
Lipids: Membranes

The Structure of the Red-blood Cell Membrane

Glycophorin



Nonpolar Amino Acids of Integral Membrane Proteins are within the Membrane



Lipids: Membranes

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Membrane Fluidity

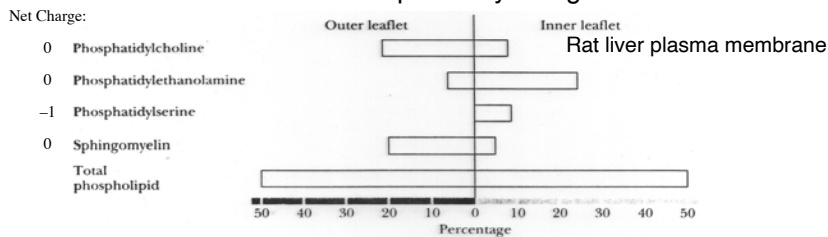
Lipids: Membranes

Asymmetry

- Membranes are very asymmetric.
- All kinds of asymmetry: Components—lipids, proteins
Types—transverse, lateral

– Lipids (transverse):

- Two leaflets have different lipid compositions.
- The outer leaflet is often more positively charged.



- If Phosphatidylserine is found outside, it has a special meaning:
 - platelets: activates blood clotting
 - other cells: marks the cell for destruction

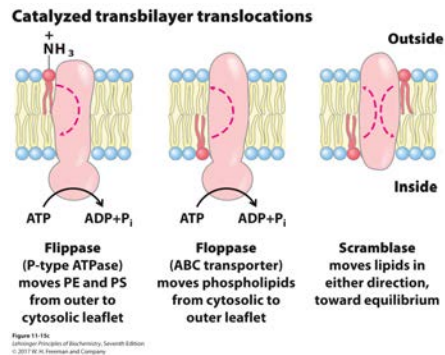
How is this asymmetry maintained?

Lipids: Membranes

Asymmetry

– Lipids (transverse): Flippases

- Special enzymes catalyze transverse diffusion.
 - Though often referred to by category name “flippase,” there are unique unidirectional and bidirectional enzymes to catalyze lipid movement.
- Some flippases use energy of ATP to move lipids against the concentration gradient.



Lipids: Membranes

Asymmetry

Lipids (lateral):

1) On the inner leaflet, can induce phosphoserine to coalesce with calcium.

2) On the outer leaflet, can induce "raft" formation; the coalescence of particular membrane lipids (cholesterol, sphingoglycolipids, sphingomyelin, etc.)

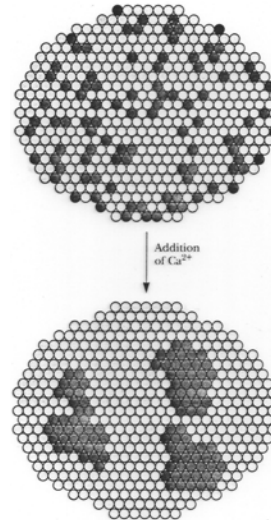


Figure 9.28 An illustration of the concept of lateral phase separations in a membrane. Phase separations of phosphatidylserine (green circles) can be induced by divalent cations such as Ca^{2+} .

Lipids: Membranes

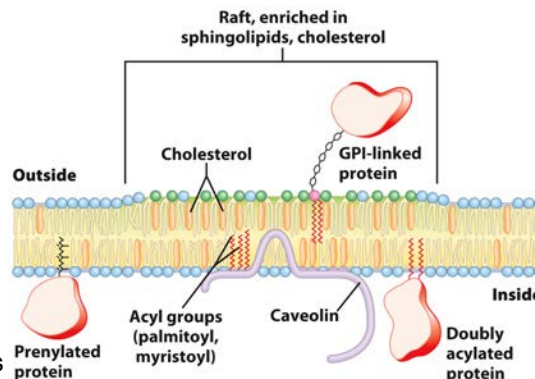
Asymmetry

Lipids (lateral): Membrane Rafts

Lipid distribution in a single leaflet is not random or uniform.

Lipid rafts:

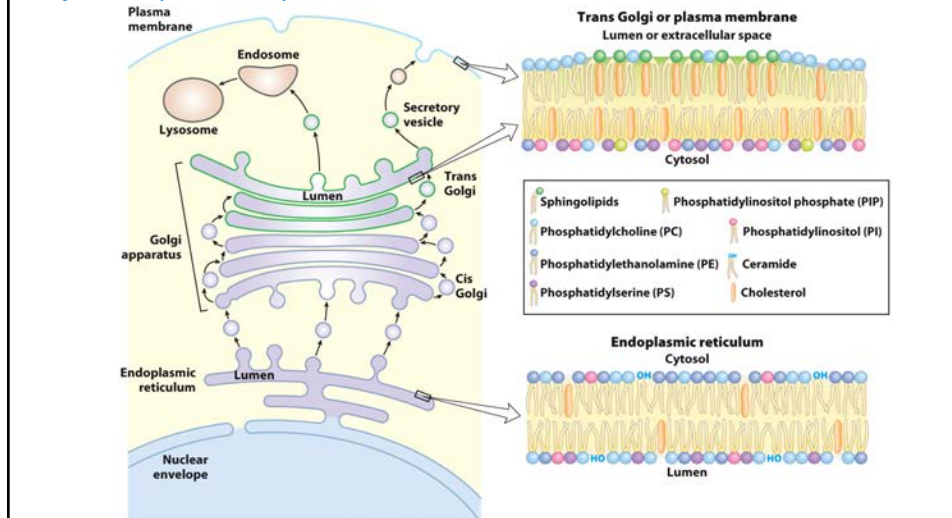
- contain clusters of sphingoglycolipids with longer-than-usual tails and cholesterol
- are more ordered (not as fluid)
- contain specific doubly or triply acylated proteins
- allow segregation of proteins in the membrane



Lipids: Membranes

Asymmetry

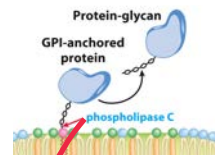
Lipids (lateral): Membrane Rafts



Lipids: Membranes

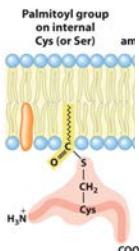
Proteins: Asymmetry

- 1) anchored
- 2) glycoproteins



- Some membrane proteins are lipoproteins containing a covalently linked lipid molecule.
- The lipid part inserts into the membrane.
- The protein is then anchored to the membrane.
- These are the aforementioned "Amphitropic" membrane proteins

- This allows targeting of proteins to the membrane, either internally (FA or isoprenes), or externally (GPI-linked).
- Some, such as GPI anchors are found only on the outer face of plasma membrane.
- This is a reversible process.



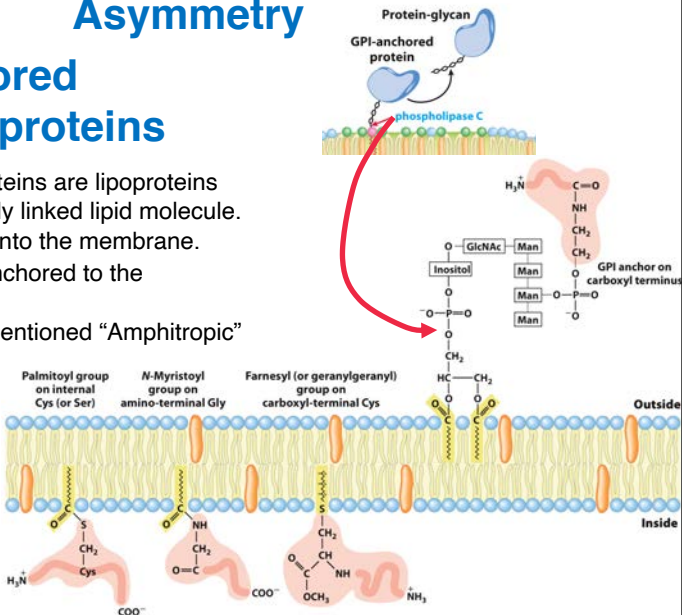
Lipids: Membranes

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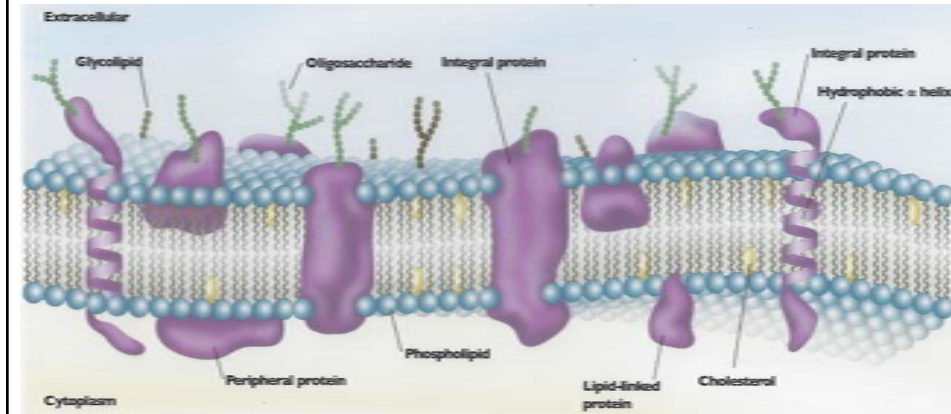


Lipids: Membranes

Proteins: **Asymmetry**

- 1) anchored
- 2) glycoproteins

Found **ONLY** ever on the **OUTSIDE** of Cells



Lipids: Membranes

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Membrane Asymmetry

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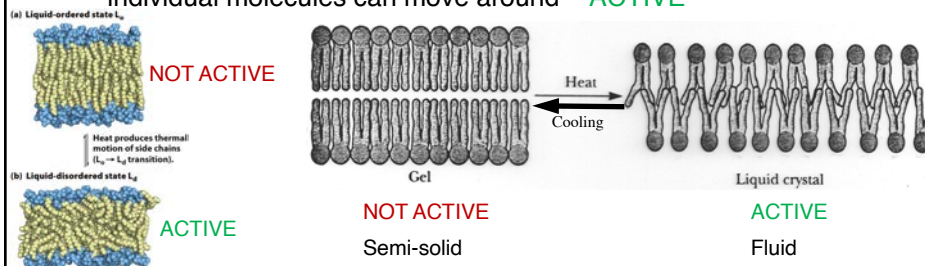
Membrane Fluidity

Lipids: Membranes

Fluidity

Membrane Phase Transition

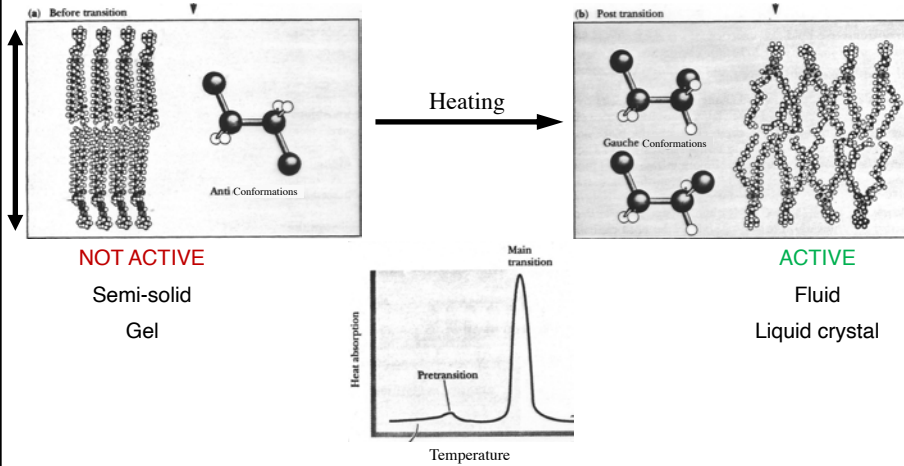
- Cells must maintain fluidity for membranes to function properly.
- Depending on their composition and the temperature, the lipid bilayer must maintain a fluid phase; if cooled, it undergoes a phase transition and goes to a gel state.
 - liquid-ordered state (i.e., “gel phase”): individual molecules do not move around – **NOT ACTIVE**
 - liquid-disordered (or liquid crystal) state (i.e., “fluid phase”): individual molecules can move around – **ACTIVE**



Lipids: Membranes

Fluidity

Membrane Phase Transition

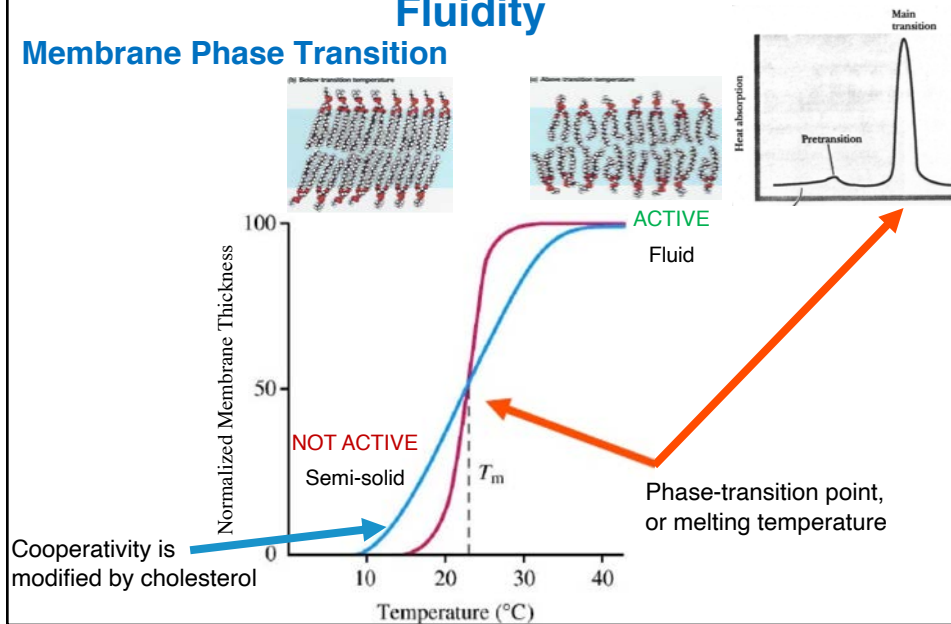


Plot membrane thickness vs. temperature

Lipids: Membranes

Fluidity

Membrane Phase Transition

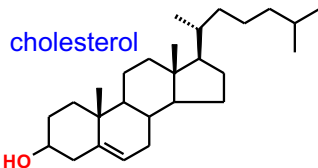


Lipids: Membranes

Fluidity

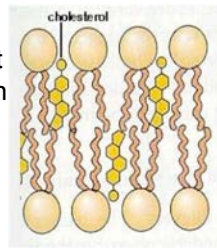
Membrane Phase Transition

Cholesterol Increases Membrane Rigidity and Permeability

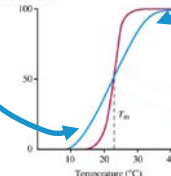


- Due to its small polar head, it lies deeper in the bilayer than the phospholipids
- Cell membranes of many eukaryotes contain **sterols**.
 - **cholesterol** in animals
 - *phytosterols* in plants
 - *ergosterol* in fungi
- Cell membranes of aerobic prokaryotes contain **hopanols**.

Cholesterol: A “Fluidity Buffer”



- Below T_m - cholesterol disrupts close packing of acyl chains \Rightarrow *increases* fluidity
- Above T_m - cholesterol constrains motion of acyl chains \Rightarrow *decreases* fluidity
- Broadens/abolishes phase transitions



Lipids: Membranes

Fluidity

Organisms Can Adjust the Temperature of the Phase Transition by Changing the Membrane Composition

- Membrane fluidity is determined mainly by the fatty acid composition and melting point.
- The temperature of the phase transition (T_m):
 - Melting temperature **higher** with more **saturated** fatty acids.
 - Melting temperature **higher** with **longer** fatty acids.
 - Melting temperature **lower** with more **unsaturated** fatty acids.
 - Melting temperature **lower** with **shorter** fatty acids.
- Therefore, at higher temperatures, cells need more **long**, **saturated** fatty acids.
- And at lower temperatures, cells need more **shorter**, **unsaturated** fatty acids.

Lipids: Membranes

Fluidity

TABLE 11-2	Fatty Acid Composition of <i>E. coli</i> Cells Cultured at Different Temperatures			
	Percentage of total fatty acids ^a			
	10 °C	20 °C	30 °C	40 °C
Myristic acid	4	4	4	8
Palmitic acid	18	25	29	48
Palmitoleic acid	26	24	23	9
Oleic acid	38	34	30	12
Hydroxymyristic acid	13	10	10	8
Ratio of unsaturated to saturated ^b	2.9	2.0	1.6	0.38

SOURCE: Data from A. G. Marr and J. L. Ingraham, *J. Bacteriol.* 84:1260, 1962.

^aThe exact fatty acid composition depends not only on growth temperature but on growth stage and growth medium composition.

^bRatios calculated as the total percentage of 16:1 plus 18:1 divided by the total percentage of 14:0 plus 16:0. Hydroxymyristic acid was omitted from this calculation.