pharmacognosy 3rd stage/1stterm

Cyanogenic Glycosides

Dr. Zahraa Shubber Lec .7

Cyanogenic Glycosides

These are glycosides that yield **hydrocyanic acid** as one of their hydrolytic products.

The most common **cyanogenic glycoside** is **amygdalin**, which is found in large quantities in bitter almonds, in kernels of apricots, cherries, peaches and in many other seeds of the Rosaceae.

And **prunasin** which occurs in Prunus serotina.

When amygdalin is hydrolyzed it forms **two molecules** of **glucose** with **benzaldehyde** and **HCN**.

The hydrolysis takes place in three steps:

1. Hydrolysis to give **one molecule of glucose** and **one of mandelonitrile glucoside**.

2. The **second molecule of glucose** is liberated with the formation of **benzaldehyde-cyanohydrin (mandelonitrile)**.

3. The mandelonitrile then breaks down with the formation of **benzaldehyde and hydrocyanic acid**.

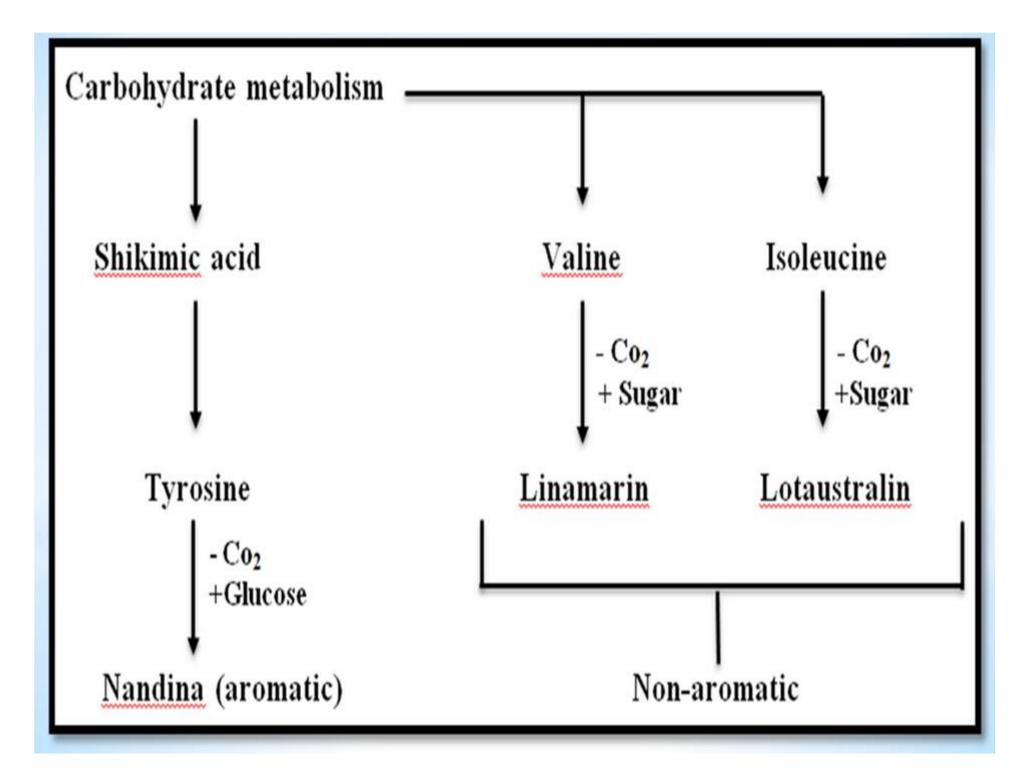
The hydrolysis steps are catalyzed by the presence of an **enzyme emulsin** found in almond kernels.

CH + GH, 05 0- CeH1004-0- CeH1105 HzO amygdalin mandelonitrile glucoside (prunasin) HZO $CH + C_{c}H_{12}O_{c}$ 1) C_H IICN mandelonitrile benzaldehyde

Biosynthesis

Evidences indicates that the biosynthesis of the majority of cyanogenic glycosides proceeds via common amino acids precursors.

The aromatic cyanogens are derived from phenyl alanine and tyrosine; while the non-aromatic ones e.g. linamarin and lotaustralin are derived from valine and isoleucine respectively.



Drugs containing cyanophore glycosides

1. Wild cherry: is the dried stem bark of **Prunus serotine** (Rosaceae).

Constituents: contains a cyanogenic glycoside prunasin (mandelonitrile glucoside).

Uses: wild cherry, in the form of syrup is employed as a flavored vehicle. It has been used as a **sedative expectorant**.





2. Bitter almond:

Is dried ripe kernel of **Prunus amygdalus**. (Fam: Rosaceae). **Uses: anticancer** studies has been made for an amygdalin containing preparations known as laetrile or vitamin B17, and the possibility for **control of sickle cell** anemia with cyanogenic glycosides has been noted.



Isothiocyanate glycosides

 $N \equiv C - O - H$ Cyanic acid $H - N \equiv C \equiv O$ Isocyanic acid $R - N \equiv C \equiv O$ Isocyanates $R - N \equiv C \equiv S$ Isothiocyanate

The seeds of several mustard family plants contain glycosides, the aglycones of which are isothiocyanates.

The most important glycosides are **sinigrin** from black mustard, **sinalbin** from white mustard and **gluconapin** from ripe seed.

When hydrolyzed by the enzyme myrosinase, they yield the mustard oils which are responsible for activity.

Biosynthesis

Aglycones of isothiocyanate glycosides may consist of either aliphatic or aromatic derivatives.

The aliphatics are derived from acetate while the aromatics from shikimic acid.

Mustard

Black mustard is the dried ripe seeds of varieties of Brassica nigra (Fam:Brassicaceae)

Constituenets: although black mustard consists of fixed oil 30-35%, its principal constituent is the glycoside sinigrin, which is accompanied by the enzyme myrosinase. Upon the addition of water to the crushed or powdered seeds, the myrosinase causes hydrolysis of **sinigrin**.

Uses: black mustard is a **local irritant** and an **emetic**.

Externally, the drug is a **rubefacient**.

It's used as a **condiment**.



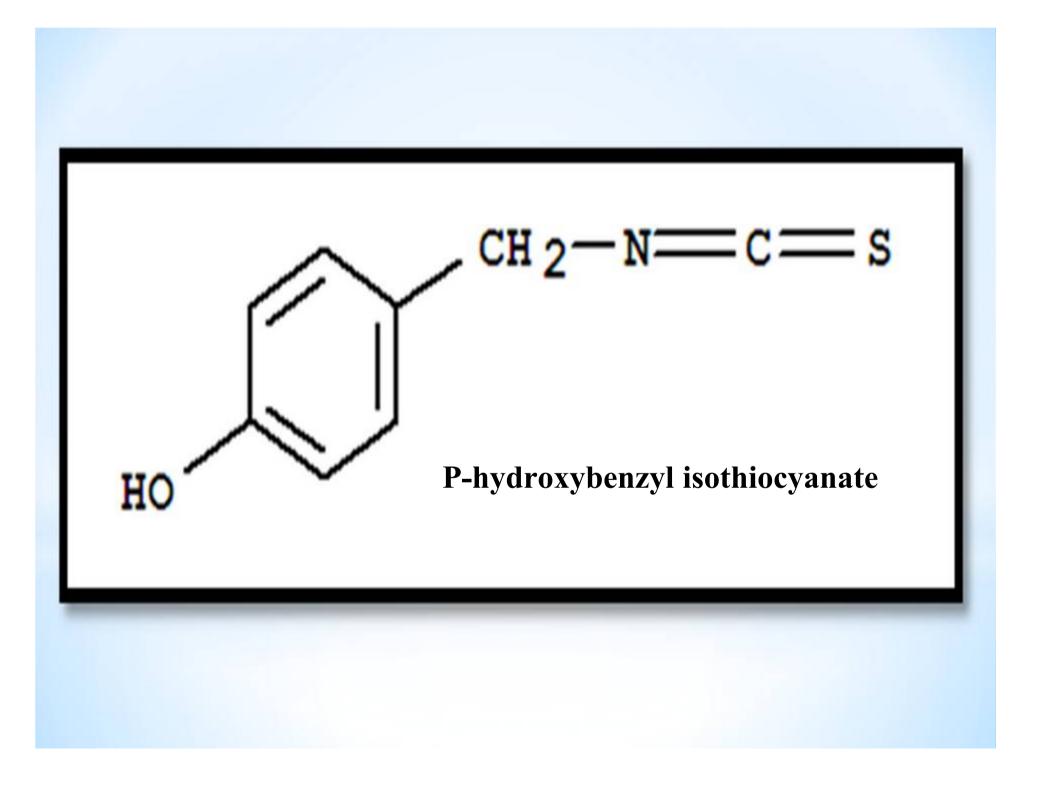
White mustard

➢ Is the dried ripe seeds of Brassica alba (Fam: Brassicaceae).

It contains the enzyme myrosinase and a glucoside, sinalbin, which upon hydrolysis yields p-hydroxybenzyl isothiocyanate, a pungent-tasting but almost odorless oil.
It also contains 20-25% of fixed oil.





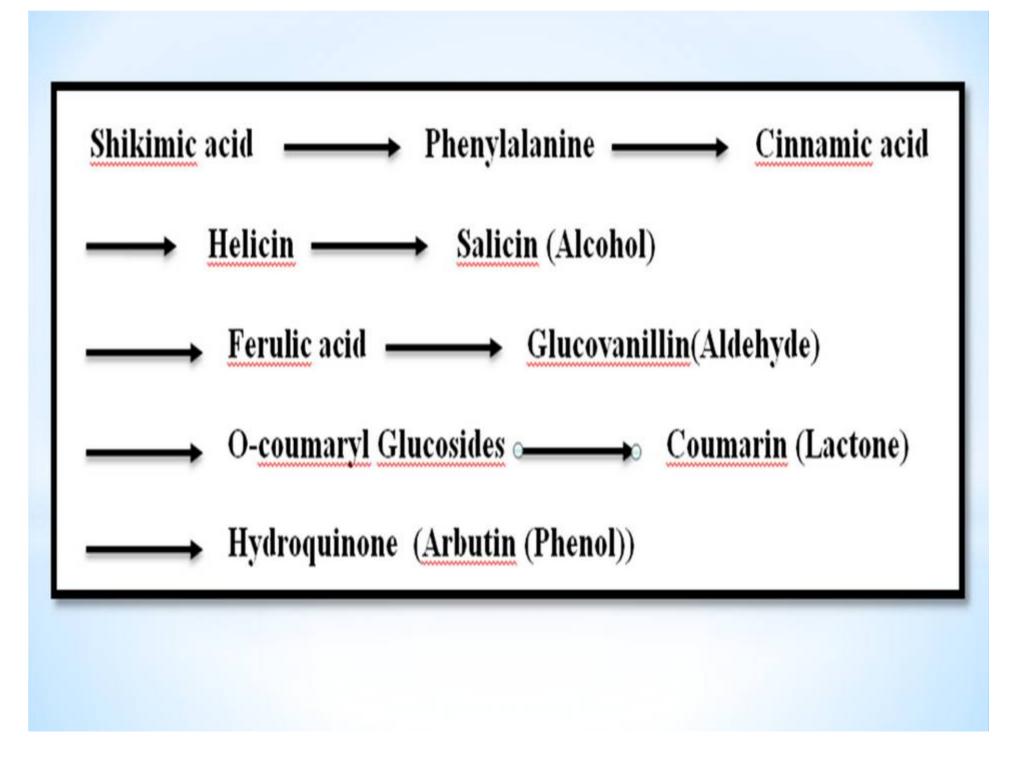


Alcohol Glycosides

Biosynthesis

Because of the close relationship between this group and others like aldehyde, lactone and phenol, the biosynthesis of all these groups will be discussed together.

Available evidence indicates that the aromatic nuclei of these compounds derive from precursors formed from shikimic acid pathway:



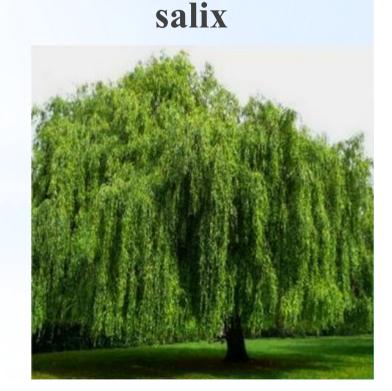
Salicin

is an alcohol glycoside obtained from several species of salix and populous (Fam: salicaceae).

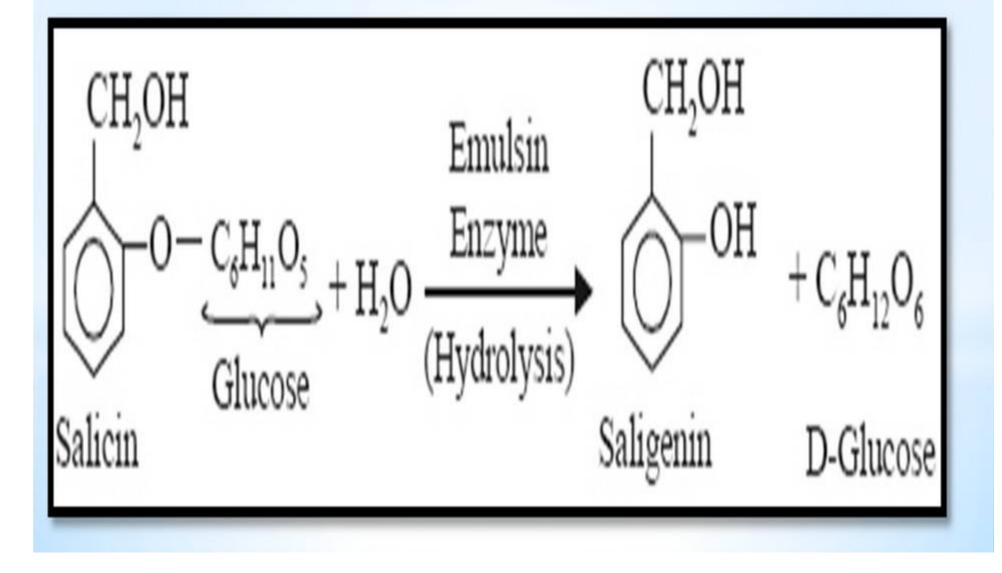
The principal sources are **salix purpurea** and **S.fragilis**. Salicin is hydrolyzed by emulsin into D-glucose and saligenin (salicyl alcohol).

Populous





Salicin is used as antirheumatic, probably through the formation of salicylic acid.



Aldehyde Glycosides

Vanillin is the aglycone developed during the curing of vanilla beans.

>Vanillin is methylprotocatechuic aldehyde.

Vanilla or vanilla bean is the full grown unripe fruit of Vanilla planifolia often known in commerce as Mexican or of Vanilla tahitensis, known in commerce as Tahiti vanilla (Fam: Orchidaceae).





Constituents: vanilla contains two glycosides:

1. Glucovanillin: upon hydrolysis with enzyme gives vanillin and glucose.

2. Glucovanillic alcohol: upon hydrolysis gives vanillic alcohol and glucose. Vanillic alcohol may oxidize to vanillic aldehyde (vanillin).

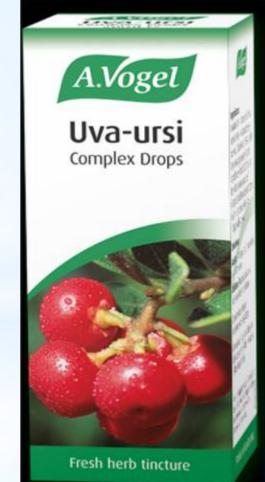
Uses: vanilla in the form of vanilla tincture, is used as **flavoring agent**, also as a source of vanilla.

Phenol Glycosides

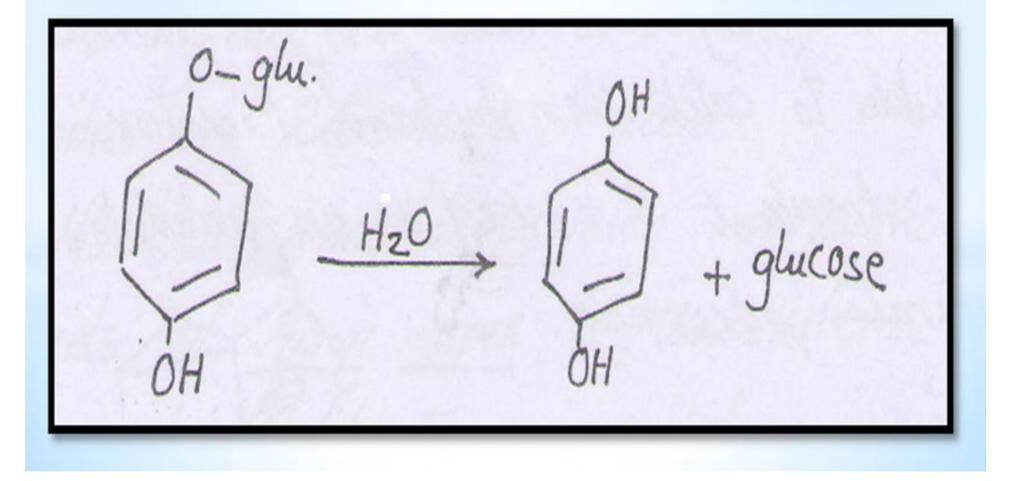
 Upon hydrolysis gives phenolic compounds and sugar.
Uva ursi: is the dried leaf of Arctostaphylos uva-ursi (Fam: Ericaceae).

Constituents: contains the glycoside **arbutin**





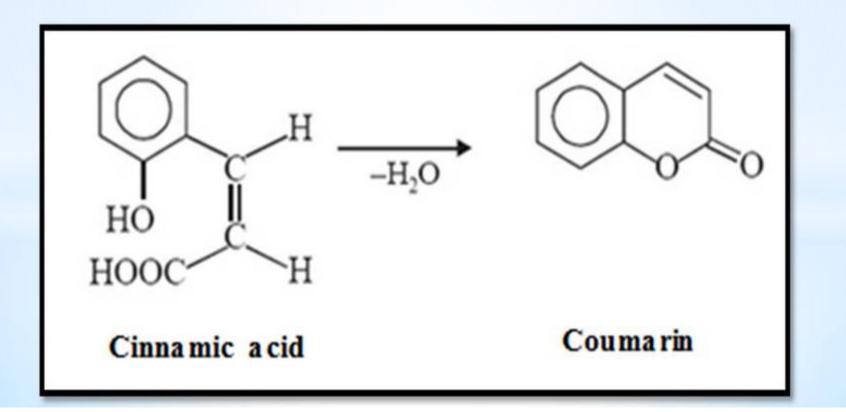
Uses: urva ursi has been used for many years as a **diuretic** and as an **astringent**.



Lacton Glycosides

Coumarins

They are lactons derived from O-hydroxy cinnamic acid by ring closure between ortho hydroxyl group and carboxyl group of side chain formed coumarin.



Coumarin occurs as colorless crystals and has a characteristic fragrant odor and bitter, aromatic, burning taste. Its soluble in alcohol.

It is widely distributed in nature, it occurs in:

- **1. Tonka beans:** in the dried seeds of **Dipteryx odorata**, and **Dipteryx oppositifolia**. Family: Lamiaceae.
- **2. Sweet vernal grass: Anthoxanthum odoratum.** Family: poaceae.





Coumarin and extracts of tonka beans were formerly used as **flavoring agents**.

- ➢ However, because of coumarin hepatotoxicity and coumarin-drug interactions that occur in a number of therapeutic substances, the FDA has banned the use of coumarin and coumarin containing materials for flavoring purposes.
- Some coumarin derivatives still find application for their **anticoagulant properties**.

Coumarin derivatives

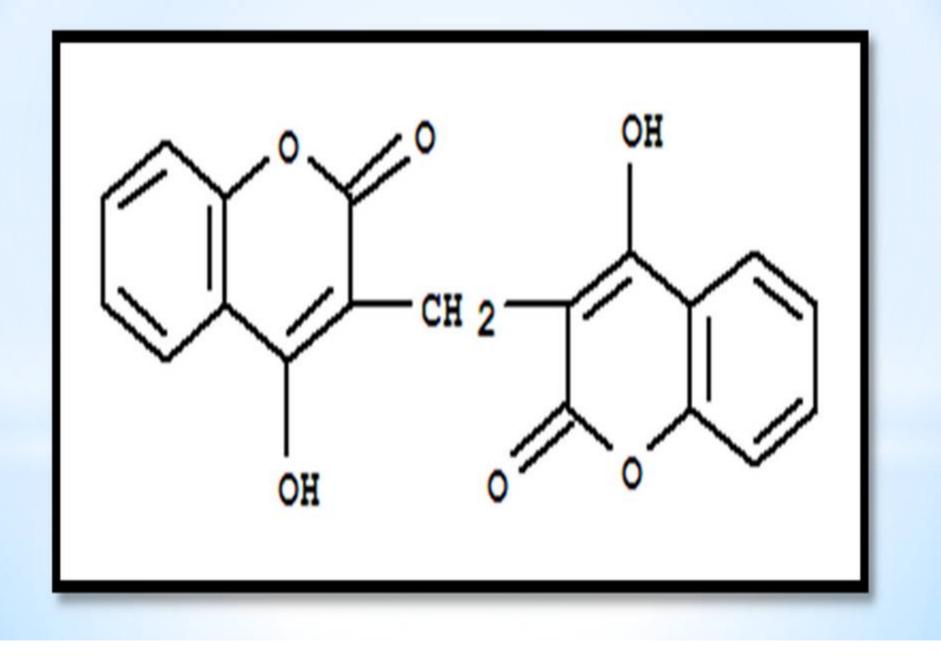
- 1. Hydroxycoumarin
- 2. Furocoumarin

1. Hydroxycoumarin: examples

A. Scopoletin : obtained from the barks of Viburnum pronifolium. Family: Caprifoliaceae. Has antispasmodic activity.



B. Bishydroxycoumarin or dicumarol:



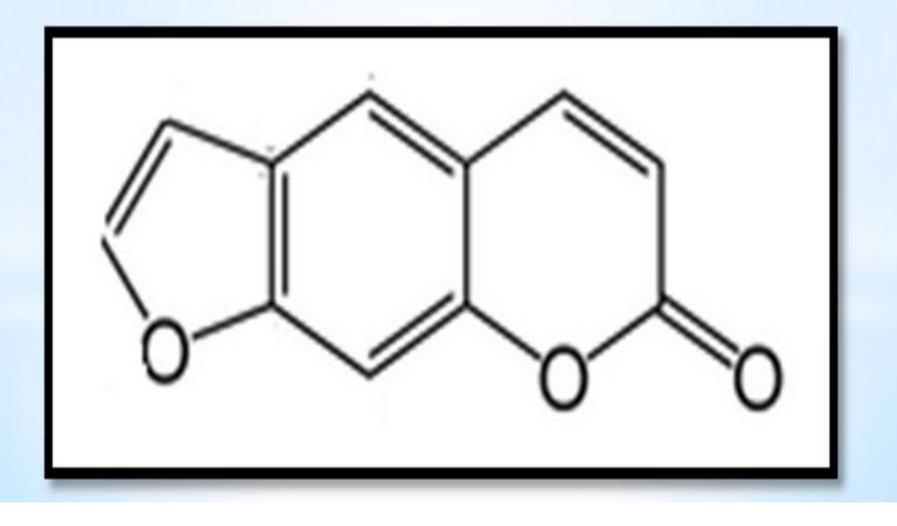
Related to coumarin. It was obtained originally from improperly cured leaves and flowering tops of Melilotus officinalis. Family:Lamiaceae.

A number of synthetic analogs of bishydroxycoumarin like warfarin also are used in anticoagulant therapy and used for the prophylaxis and treatment of venous thrombosis and pulmonary embolism.



2. Furocoumarins: example: psoralens

are photosensitizing furocoumarins that occur in a number of plant families, including the Apiaceae and Rutaceae, where they are a common cause of phototoxicity.



Methoxsalen,

a constituent of Ammi majus, is used to facilitate repigmentation in idiopathic vitilligo and for symptomatic control of sever psoriasis.

> Methoxsalen may be applied topically or taken orally.





In general, coumarins have antimicrobial and antifungal activity, photosensitizing, blood thinning, antispasmodic, antiasthmatic, hypotensive properties. Warfarin also used as rat poison.



Chromones

Represent a small group of plant phenolics that are isomeric with coumarins. The best known chromones is the furochromone khellin, which is a **potent coronary vasodilator.**

Khellin:

- is a furochromone found in the fruit of Ammi visnaga, family: Apiaceae.
- ► It has been used for **urethral spasm and renal colic**.
- It is a potent coronary vasoldilator and bronchodilator in the treatment of coronary insufficiency, angina pectoris, and bronchial asthma.
- ➢ It is an antiasthmatic, antiallergic agent that acts locally in the lung to which it is applied by inhalation.



Ammi visnaga



Thank You