#### Hypermagnesemia in gastrointestinal tract

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#### Overview

Serum magnesium concentration
 [Mg<sup>2+</sup>] > 2.6 mg/dL in adults

- Renal failure OR Excessive intake of magnesium

• Typically asymptomatic unless [Mg<sup>2+</sup>] > 4.8 mg/dL

Jahnen-Dechent W, Ketteler M. Magnesium basics. Clin Kidney J. 2012 Chang WT, Radin B, McCurdy MT. Calcium, magnesium, and phosphate abnormalities in the emergency department. Emerg Med Clin North Am. 2014 May

#### Overview

- Mechanisms of hypermagnesemia include:
  - 1. Excess intake via oral, rectal, or IV medications
  - 2. Decreased renal excretion renal failure
  - 3. Redistribution intra- to extracellular space with acidosis
  - 4. Release from cells (for example, due to tumor lysis syndrome)

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#### **Clinical Sequelae of Magnesium Disturbances**

	mg/dL	Manifestation
Severe hypomagnesemia	<1.2	Tetany, seizures, arrhythmias
Mild to moderate hypomagnesemia	1.2-1.8	Neuromuscular irritability, hypocalcemia, hypokalemia
Normal	1.8-2.5	Normal magnesium level
Hypermagnesaemia	2.5-4.8	Typically asymptomatic
Mild hypermagnesaemia	4.8-7.0	Lethargy, drowsiness, flushing, nausea and vomiting, diminished deep tendon reflexes
Moderate hypermagnesaemia	7.0-12	Somnolence, loss of deep tendon reflexes, hypotension, electrocardiographic changes (bradycardia)
Severe hypermagnesaemia	>12	Complete heart block, cardiac arrest, apnea, paralysis, coma

Jesse B. Hall, Gregory A. Schmidt, John P. Kress; Principles of Critical Care, 4<sup>th</sup> edition

Magnesium and cardiovascular complications of chronic kidney disease, Nat. Rev. Nephrol. 2015

### Incidence/Prevalence

- Moderate-to-severe hypermagnesemia is rare in persons with normal renal function
- Hypermagnesemia reported to be common in hospitalized patients

- 2.8% - 13.5% hospitalization patient

• Upper limit of [Mg<sup>2+</sup>]: 2 - 2.6 mg/dL

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#### Cause

- 1. Excess intake
  - Ingestion of magnesium
    - Magnesium-contain medication
      - Cathartics, antacids, laxatives, and dietary supplements
  - Magnesium-containing enemas
  - IV infusion for seizure prophylaxis in eclampsia

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#### Cause

- 2. Decreased renal excretion
  - Renal insufficiency
  - Familial hypocalciuric hypercalcemia (CaSR inactivation)
- 3. Other causes (redistribution and release)
  - Redistribution from intra- to extracellular space with acidosis
  - Release from cells, such as in tumor lysis syndrome
  - Primary adrenal insufficiency
  - Hypothyroidism

- 2<sup>nd</sup> intracellular cation
- Shares chemical properties with calcium but has distinct biologic properties and regulation
- [Mg<sup>2+</sup>] homeostasis depends:
  - 1. Intestine: uptake
  - 2. Bone: storage system
  - 3. Kidneys: [Mg<sup>2+</sup>] excretion

- [Mg<sup>2+</sup>] required 3.0-4.5 mg/kg/day
  - Female: 310-320 mg/day
  - Male: 400-420 mg/day
    - Green vegetables, nuts, seeds, and unprocessed cereals
    - About 10% of daily intake comes from drinking water



Magnesium in man: implications for health and disease, Physiol Rev. 2015

 Reported 85% of whole-body magnesium does not exchange or does so very slowly with an estimated half-life of 1,000 hours.



## Pathophysiology - Kidney

- Mechanisms of Mg absorption in kidney
  - Maintaining magnesium homeostasis
  - Renal excretion: 0.5% 70% of filtered load
    depending on magnesium balance
  - Filter 2,400 mg/day
    - 95% reabsorbed, and 3%-5% is excreted in urine
    - Proximal tube (10%)
    - Ascending loop of Henle (60%-70%)



Magnesium in man: implications for health and disease, Physiol Rev. 2015

## Pathophysiology – GI tract

- Mechanisms of Mg absorption in intestine
  - Predominantly in small intestine (distal jejunum and ileum)
    via paracellular pathway (80-90%)
  - Smaller amounts are absorbed in the colon, mainly via a transcellular pathway (10-20%)
    - 24%-76% of dietary magnesium is absorbed
    - Absorption is determined by magnesium status



Regulation of magnesium balance: lessons learned from human genetic disease, Clin Kidney J. 2012 TRPM6 and TRPM7—Gatekeepers of human magnesium metabolism, Biochimica et Biophysica Acta, 2007 Intestinal Absorption and Factors Influencing Bioavailability of Magnesium, Current nurtition Food science, 2017

Mg Salts	Carbonate	Chloride	Citrate	Fumarate	Gluconate
Solubility	Nearly insoluble	High	Very good	Good	Moderate
Bioavailability	Extremely low	Good	Good	Good	Good
Delivery system	Tablets	Enteric coated tablets	Liquid tablets	Tablets	Tablets, liquid
Side effects	GI distress, diarrhea	GI distress, diarrhea	Laxative		GI distress, diarrhea

**Bioavailability of US commercial magnesium preparations.** Magnes. Res. 2001 Bioavailability and Pharmacokinetics of Magnesium Salts to Humans, Am J of therapeutic, 2001

Mg Salts	Glycinate	L-lactate	Oxide	K Mg citrate	DL-aspartate
Solubility	Good	Excellent	Extremely low	High solubility	Good
Bioavailability	Good	Excellent	Extremely low	Good	
Delivery system	Ingestion	Sustained release caplets	Tablets, capsule	Tablets	Tablets
Side effects		Minor GI disturbances	Emesis, diarrhea	No GI side effect	

**Bioavailability of US commercial magnesium preparations.** Magnes. Res. 2001 Bioavailability and Pharmacokinetics of Magnesium Salts to Humans, Am J of therapeutic, 2001

Mg Salts	L-aspartate	Hydroxide	Salicylate	Sulfate	Aminoate
Solubility	Good		Freely soluble	Moderately soluble	
Bioavailability			80-100%		
Delivery system	Tablets	Tablets	Tablets	lv solution	Tablets
Side effects		Occasional regurgitatio n and mild diarrhea			

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## **Bioavailability of Mg preparation**

- The complexity of [Mg<sup>2+</sup>] absorption
  - Direct method: [Mg<sup>2+</sup>] after oral administration
    - Insufficient to investigate: Rate and amount of [Mg<sup>2+</sup>]
      Rapid homeostasis: renal excretion + storage in compartment
  - Indirect method: faecal or urinary [Mg<sup>2+</sup>]
    excretion after oral Mg2+ administration
    - Limitation: diet, long-term
  - Isotopic method: <sup>25</sup>Mg<sup>2+</sup> and <sup>26</sup>Mg<sup>2+</sup>
    - highly abundant in nature; relative low sensitivity

## **Bioavailability of Mg preparation**

 The bioavailability and intestinal absorption efficacy of orally ingested [Mg<sup>2+</sup>] are influenced by **endogenous** and **exogenous** factors.



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TRPM6 and TRPM7—Gatekeepers of human magnesium metabolism, Biochimica et Biophysica Acta, 2007

- In principle, [Mg<sup>2+</sup>] uptake is higher
  - Multiple low doses > single, large dose
  - Mg salt: less relevant than is often thought.
    - Bioavailability: organic > inorganic Mg salts

- Effects on other electrolytes
  - Hypocalcemia
    - Inhibition of parathyroid hormone (PTH) by magnesium
    - [Ca<sup>2+</sup>] and [PTH] fell after magnesium infusion in 7 pregnant women being treated to suppress labor
  - Hyperkalemia
    - [Mg<sup>2+</sup>] blocking potassium channels in cortical collecting duct, resulting in reduced potassium clearance

• Correlations: clinical presentation and [Mg<sup>2+</sup>]

– [Mg<sup>2+</sup>] < 4.8 mg/dL are usually asymptomatic</p>

Deep tendon reflexes may be diminished with
 [Mg<sup>2+</sup>] > 6.1 mg/dL

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• Interpreting serum magnesium

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- The serum [Mg<sup>2+</sup>]: not reflect total body status
- Avoid hemolysis : RBC [Mg<sup>2+</sup>] > serum [Mg<sup>2+</sup>]
- No clinically available "ionized" magnesium test to diagnose deficiency based on free magnesium

- Additional test: causes of alterations in calcium levels
  - parathyroid hormone
  - vitamin D
  - phosphate
  - muscle enzymes (to assess for rhabdomyolysis)
- urinary 24-hour magnesium collection

Moe SM. Disorders involving calcium, phosphorus, and magnesium. Prim Care. 2008 Jun Jahnen-Dechent W, Ketteler M. Magnesium basics. Clin Kidney J. 2012 Chang WT, Radin B, McCurdy MT. Calcium, magnesium, and phosphate abnormalities in the emergency department. Emerg Med Clin North Am. 2014 May

#### Electrocardiography

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- In patient with unexplained bradycardia or tachycardia (not diagnostic)
  - prolonged PR, QRS, and QT intervals
  - complete heart block
  - atrial fibrillation and asystole

#### **Additional test**

- Few case report showed
  - Abdominal computed tomography
    - Magnesium-containing bezoars
      - Magnesium oxide, aluminum and magnesium hydroxide

#### Management

- Most cases of sufficiently treated by
  - Discontinuing magnesium-containing supplements
  - Volume replacement with IV saline
- Loop diuretics: enhance magnesium excretion
  - Prevent volume overload from volume repletion
  - Monitor [Ca<sup>2+</sup>], because increased calcium excretion

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#### Management

- Cardiac dysfunction or hypotension and respiratory depression, additionally
  - Calcium to counteract effects of magnesium
    - [Mg<sup>2+</sup>] interfere with synaptic transmission by blocking calcium channels; formulations used include
      - Elemental calcium 100-200 mg IV over 5-10 minutes
      - Calcium chloride 10% solution 5-10 mL (500-1,000 mg) IV
      - Calcium gluconate 1 g IV over 3 minutes



#### Management

 For patients with severe renal dysfunction and/or severe hypermagnesemia, consider dialysis

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# Mg product develop hypermagnesemia in normal renal function patient.



#### Management – case experience

- Regularly measured serum magnesium in
  - Geriatric patients
  - Administered magnesium oxide for prolonged periods



### **Prognosis and prevention**

- With adequate renal function typically recover and normalize magnesium levels once supply of magnesium has been stopped.
- Reduced prescription magnesium-containing antacids or cathartics to patients with renal insufficiency.

	Fleet Phospho-Soda	Bow-klean Powder	GI Klean Powder	
	Sodium phosphate	Magnesium oxide, Sodium picosulfate, citric acid anhydrous	Sodium sulfate, sodium bicarbonate	
清腸乾淨度	76.6-78.8%	72.5%	65.7-77.6%	
自費價格	336元	578元	352元	
Side effect	13.6-32.2% Renal toxicity, hyperphosphatemia, hypo-K/Ca	4.3-26.7% Hypermagnesemia Hypo-Na/K	10.4-33.1%	
Benefit	Cheapest + cleanest		No renal adverse event	
Disadvantage	Renal toxicity	Electrolyte imbalance		
Contraindication	Renal insufficiency	Renal insufficiency		
	Intestinal obstruction, intestinal perforation			

#### 感謝您的指導與聆聽