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5 **Discovery of terrestrial allabogdanite (Fe,Ni)<sub>2</sub>P, and the effect of Ni and Mo substitution on the**  
6 **barringerite-allabogdanite high-pressure transition**

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29 **Abstract**

30 Minerals formed at high pressures are sensitive indicators of extreme pressure-temperature conditions  
31 that occur in nature. The discovery of the high-pressure polymorph of  $(\text{Fe,Ni})_2\text{P}$ , allabogdanite in the  
32 surficial pyrometamorphic rocks of the Hatrurim Formation (the Mottled Zone) surrounding the Dead  
33 Sea basin in Israel is the first terrestrial occurrence of a mineral previously only found in iron  
34 meteorites. Stepwise annealing experiments demonstrate that allabogdanite is metastable at ambient  
35 pressure and that it irreversibly transforms into its low-pressure polymorph, barringerite, upon heating  
36 to  $850\pm 50^\circ\text{C}$ . High-pressure high-temperature diamond-anvil cell (DAC) experiments confirm the  
37 results of annealing experiments. The DAC data indicate that Hatrurim allabogdanite is metastable  
38 below 7.4 GPa, and the low- to high-pressure phase transition (barringerite $\rightarrow$ allabogdanite) occurs at  
39  $25\pm 3$  GPa and  $1400\pm 100^\circ\text{C}$ . The observed transition pressure of Hatrurim allabogdanite is significantly  
40 higher than that of pure synthetic  $\text{Fe}_2\text{P}$  (8 GPa), due to partial substitution of Fe for Ni (4 wt.%) and Mo  
41 (2.5 wt.%). Because the influence of substituting impurities on the conditions of phase transitions can  
42 be unexpectedly strong, our findings confirm that caution should be exercised when extrapolating data  
43 from experiments on synthetic compounds to natural systems. Based on the discovery of terrestrial  
44 allabogdanite  $(\text{Fe,Ni})_2\text{P}$  coupled with experiments probing the phase transitions in this natural  
45 composition, we contend that terrestrial allabogdanite formed via transformation from barringerite and  
46 posit potential scenarios of its formation.

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51 **Keywords:** allabogdanite, barringerite, phosphide,  $\text{Fe}_2\text{P}$ , high pressure, crystal structure, phase  
52 transitions, Fe–Ni–P system

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## Introduction

55 The area of the Eastern Mediterranean in the historical region of the Southern Levant bears the  
56 remnants of an enigmatic rock complex known as the Mottled Zone or the Hatrurim Formation (Figure  
57 1) (Gross 1977; Vapnik et al. 2007). This formation represents the world's largest field of sedimentary  
58 rocks subjected to combustion metamorphism (pyrometamorphism) – the processes of surficial high-  
59 temperature calcination and fusion of sediments that produces mineral assemblages resembling cement  
60 clinkers or basic igneous rocks (Sokol et al. 2019). Numerous outcrops of the Mottled Zone span an  
61 area of  $150 \times 200 \text{ km}^2$  on both sides of the Dead Sea, encompassing the Judean and Negev deserts in  
62 Israel, Palestinian Authority and Transjordan Plateau in West-Central Jordan, with several patches  
63 mapped as far north as the Sea of Galilee (Kinneret) (the Maqarin site) (Figure 1). These exposed  
64 outcrops are likely remnants of wider fields deflated by later erosion (Matmon 2017). The largest  
65 preserved Mottled Zone massifs are the Daba-Siwaqa complex in West-Central Jordan ( $>300 \text{ km}^2$ ) and  
66 the Hatrurim Basin in Israel ( $\sim 50 \text{ km}^2$ ) (Figure 1). Geological dating of the Mottled Zone has produced  
67 a range of ages, with the majority falling between 16 and 2.5 Ma (million years) (see a review in Sokol  
68 et al. 2019).

69 The mineralogy of the Mottled Zone rock suite is diverse and contains both highly reduced (e.g.,  
70 Britvin et al. 2015; Galuskina et al. 2017) and highly oxidized mineral assemblages (Sokol et al. 2011;  
71 Juroszek et al. 2020), including those more characteristic of meteorites than terrestrial rocks (Weber  
72 and Bischoff 1994; Galuskina et al. 2016). Phosphides are typical accessory minerals in the Mottled  
73 Zone and have been identified in several localities at both sides of the Dead Sea, in Israel and West-  
74 Central Jordan (Britvin et al. 2015, 2019a,c, 2020a,b,c). Phosphides typically form under highly  
75 reducing conditions and contain phosphorus in an oxidation state lower than zero. These minerals are  
76 commonly encountered in iron meteorites, whereas phosphorus normally occurs in its oxidized form  
77 in minerals from the Earth's crust. Since phosphides are considered the most likely source of low-

78 valent phosphorus in the prebiotic history of our planet (Bryant and Kee 2006; Gull et al. 2015; Pasek  
79 2017), their discovery in the Mottled Zone has prompted the hypothesis of a terrestrial source of  
80 prebiotic phosphorus on early Earth (Britvin et al. 2015). The superficial combustion of sediments that  
81 give rise to the Mottled Zone mineral assemblage occurred at high-temperatures (from 800 to 1400°C)  
82 but low-pressures (near-atmospheric); hence all hypotheses aimed at explaining the origin of this rock  
83 suite follow a “low-pressure” paradigm (Gross 1977; Vapnik et al. 2007; Sokol et al. 2019). The recent  
84 discovery of allabogdanite,  $(\text{Fe,Ni})_2\text{P}$ , a high pressure polymorph of barringerite in the Mottled Zone,  
85 which is described in this paper, calls into question this prevailing low-pressure paradigm. However,  
86 because allabogdanite may contain substantial amounts of Ni and Mo substituting for Fe, a direct  
87 comparison of its formation conditions with those determined for synthetic high-pressure  $\text{Fe}_2\text{P}$  is  
88 problematic. Therefore, we performed high-pressure experiments aimed at determining the stability  
89 field of the Dead Sea allabogdanite to inform the discussion of what geological processes may have  
90 produced the allabogdanite-bearing assemblages of the Mottled Zone.

91 Allabogdanite (orthorhombic, structure type  $\text{Co}_2\text{Si}$ ) and barringerite (hexagonal, structure type  
92  $\text{Fe}_2\text{P}$ ) are polymorphs of  $\text{Fe}_2\text{P}$ . The relationship between their unit cells is sketched in Figure 2. Both  
93 minerals commonly contain Ni substituting for Fe (Britvin et al. 2019b, 2020c). Barringerite was  
94 discovered in the Ollague pallasite (Buseck 1969). This mineral, with substantial Ni, was subsequently  
95 reported from several meteoritic and terrestrial occurrences (a brief review is given in Britvin et al.  
96 2020c). In all cases, the mineral with composition  $(\text{Fe,Ni})_2\text{P}$  was described as barringerite, as no other  
97 polymorphs of  $(\text{Fe,Ni})_2\text{P}$  were known. Recently, a complete solid solution between barringerite and  
98 transjordanite (natural hexagonal  $\text{Ni}_2\text{P}$ ) was shown (Britvin et al. 2020c). Allabogdanite was discovered  
99 in the Onello meteorite (iron anomalous, Ni-rich ataxite) (Britvin et al. 2002). The high-pressure origin  
100 of the mineral was not suggested at that time, because no experimental data on phase transitions of  
101  $(\text{Fe,Ni})_2\text{P}$  existed. Subsequently, Dera et al. (2008) showed that orthorhombic  $\text{Fe}_2\text{P}$  (i.e., having

102 allabogdanite structure) is a high-pressure phase metastable below 8 GPa. Recently, allabogdanite was  
103 found in two more Ni-rich ataxites (Santa Catharina and Barbianello) (Britvin et al. 2019b).  
104 Allabogdanite from meteorites is enriched in Ni (up to 0.8 Ni atoms per formula unit). However, the  
105 stability field of Fe<sub>2</sub>P–Ni<sub>2</sub>P solid solutions with this structure is not known. Pure Ni<sub>2</sub>P does not  
106 transform to allabogdanite-type polymorph even at 50 GPa (Dera et al. 2009). Allabogdanite was not  
107 encountered on Earth prior to the present work.

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## 109 **Materials and Methods**

110 **Scanning electron microscopy (SEM) and electron microprobe analysis (EMPA).** Chemical  
111 homogeneity of phosphides was inspected in the polished rock sections with an Hitachi S-3400N SEM  
112 equipped with an Oxford Instruments AzTec Energy X-Max 20 EDX detector. EMPA data were  
113 acquired with an INCA WAVE 500 WDX spectrometer using the following standards: GaP (P),  
114 metallic Cr, V, Fe, Co and Ni (all *K*α lines), Mo (MoLα), at 20 kV and 15 nA. **Electron backscatter**  
115 **diffraction (EBSD).** Polished sections intended for EBSD study were subjected to reactive ion etching  
116 (RIE) with Ar<sup>+</sup> ions using an Oxford Instruments IonFab-300 instrument (500 V, 2.4 mA cm<sup>-2</sup> flow  
117 current), etching time was 20 min. EBSD analyses were carried out by means of an Oxford Instruments  
118 Nordlys-HKL EBSD detector operated at 30 kV and 1.5 nA in focused beam mode with a 70° tilted  
119 stage. The samples were coated with a ~2 nm carbon film and grounded with carbon tape.  
120 Identification of allabogdanite and barringerite was performed by matching their EBSD patterns with  
121 the reference structures (ICSD 94577 and 70115). The details of EBSD procedures are given in Table  
122 S1. **Single-crystal X-ray diffraction (SCXRD) and crystal structure.** Allabogdanite grain #1 (Figure  
123 3) and barringerite grain #5, as well as two-phase grain #3 were extracted for crystal structure  
124 measurements. SCXRD datasets were collected by means of a Bruker Kappa APEX DUO CCD  
125 diffractometer (MoKα, 50 kV, 0.6 mA) and processed using standard SCXRD routines incorporated

126 into the Bruker APEX2 software. Crystal structures were solved and refined using the *SHELX*-2018 set  
127 of programs (Sheldrick 2015) incorporated into the Olex2 graphical user shell (Dolomanov et al. 2009).  
128 The details of SCXRD study can be retrieved from CIF files in Supplementary Data. **Annealing**  
129 **experiments, powder X-ray diffraction (PXRD) and post-annealing X-ray single-crystal diffraction**  
130 **(SCXRD)**. Allabogdanite grains #7 and #13 (Figure 3) were subjected to stepwise heating in a Netzsch  
131 STA 449 F1 differential scanning calorimeter at a ramp rate of  $10^{\circ} \text{ min}^{-1}$ , using 4-mm Pt pan at  $50 \text{ mL}$   
132  $\text{ min}^{-1}$  Ar flow (99.9994 % Ar). In order to avoid possible reaction between allabogdanite and Pt upon  
133 heating, mineral grains were placed onto 1-mm thick diamond plate used as an interface between  
134 allabogdanite and Pt pan. The structural state of stepwise-heated grains was inspected using non-  
135 destructive PXRD including the following set-up: Rigaku RAXIS Rapid II diffractometer (semi-  
136 cylindrical image plate,  $r=127.4 \text{ mm}$ , Co  $K\alpha$ , rotating anode, 40 kV, 15 mA, microfocus X-ray optics).  
137 PXRD image acquisitions were conducted in Debye-Scherrer geometry using a Gandolfi method, with  
138 an exposure time 60 min. The obtained images were transformed into the profile data using osc2xrd  
139 program (Britvin et al. 2017a) and fitted with a Pawley method implemented into a Bruker Topas 4.2  
140 software. The first heating run of grain #7 was interrupted at  $600^{\circ}\text{C}$ , followed by cooling the grain to  
141 room temperature, picking up the annealed grain and inspecting its structural state. Subsequent  
142 heating/PXRD runs were performed stepwise with  $50^{\circ}$  step. The allabogdanite→barringerite transition  
143 of grain #7 was detected between  $800$  and  $850^{\circ}\text{C}$  (Figure 7). The same transition of the grain #13 was  
144 detected between  $750$  and  $800^{\circ}\text{C}$ . Therefore, average transition temperature was found to be  $800\pm 50^{\circ}\text{C}$ .  
145 The grain #13 (originally composed solely of allabogdanite) was studied by means of SCXRD *after*  
146 thermally induced transition to barringerite. The SCXRD procedures were the same as described above.  
147 The rotation matrices of barringerite formed after thermally induced allabogdanite→barringerite  
148 transition are given in Table 4. **In situ X-ray single-crystal high-pressure-high-temperature**  
149 **experiments**. The experiments in laser-heated diamond anvil cells (LHDAC) were carried out at the

150 P02.2 Extreme Conditions Beamline (ECB) of PETRA III storage ring of Deutsches Elektronen  
151 Synchrotron (DESY) light source (Hamburg, Germany). The layout of a beamline involves a 1-circle  
152 ( $\omega$ -rotation) goniometer equipped with a stage holder designed for X-ray studies in diamond-anvil cells  
153 (DAC). Symmetric Mao-type DACs with culet diameters of 400 and 150  $\mu\text{m}$  were used for pressure  
154 generation in the experiments #1 and #2, respectively. The sample chambers with approximate  
155 diameters of 200 and 80  $\mu\text{m}$  were obtained by drilling the preintended rhenium gaskets.  $\sim 5$   $\mu\text{m}$  ruby  
156 balls were placed in each sample chamber for initial pressure estimation (using luminescence  
157 spectrum), whereas 10-15  $\mu\text{m}$  chips of Au foil were used for pressure determination during the  
158 experiments. 20-30  $\mu\text{m}$  single-crystal fragments of Mo- and Ni-enriched allabogdanite have been  
159 placed in each DAC along with the pressure standards. The DACs were loaded with neon as a pressure-  
160 transmitting medium using an in-house gas loading system at PETRA III. The monochromatic X-ray  
161 beam had a cross-section of  $2 \times 2$   $\mu\text{m}^2$  and a wavelength of 0.2892  $\text{\AA}$ . The Perkin Elmer XRD 1621  
162 detector (CsI bonded amorphous silicon,  $2048 \times 2048$  pixels, 200  $\mu\text{m}$  square pixel size) was placed  $\sim 404$   
163 mm away from the DAC. The detector was calibrated with a  $\text{CeO}_2$  powder standard using the  
164 DIOPTAS program (Prescher and Prakapenka 2015). Prior to the high-pressure-high-temperature  
165 experiments, the unit-cell parameters of allabogdanite were determined at ambient conditions using the  
166 single-crystal X-ray datasets collected from  $-32$  to  $32^\circ$  ( $\omega$ ) at the frame sweep of  $0.5^\circ$ . Data processing  
167 and integration routines were carried out by means of a Rigaku Oxford Diffraction CrysAlis Pro  
168 package, using orthoenstatite as an external calibration standard. The 200W on-axis Yb-fiber laser  
169 (1064 nm) allowed double-sided heating of a  $\sim 15 \times 15$   $\mu\text{m}^2$  area of the allabogdanite single crystals  
170 between 1300-1600°C. The temperature was determined with a precision  $\pm 100^\circ\text{C}$  using a black-body  
171 excitation spectrum. Each series of single-crystal X-ray experiments included stepwise compression of  
172 the crystal in DAC, followed by a double-sided laser heating for 5-8 minutes. During the heating, the  
173 laser beam was gradually moved across the crystal, in order to achieve as uniform heating as possible.

174 After cooling the crystal for ~2 min, the single-crystal X-ray datasets were collected at each pressure  
175 step. In the experiment #1, the initial pressure was set at 2.0 GPa. The crystal was then compressed to  
176 4.8 GPa followed by the first laser heating to 1350°C. The complete allabogdanite→barringerite  
177 transition was observed after 5 min of heating time. The two additional compression-heating runs were  
178 conducted up to 19.6 GPa at 1350°C, without reaching the expected low- to high-pressure phase  
179 transition point. The experiment #2 was started at 7.4 GPa. The allabogdanite→barringerite phase  
180 transition was completed within 8 min of laser heating to 1350°C. After several compression/heating  
181 steps, the barringerite→allabogdanite conversion was accomplished at 28.4 GPa and 1400°C for 8 min.  
182 Further compression/heating runs to 39 GPa and 1400°C did not reveal any phase transitions.

## 183 **Results**

### 184 **Occurrence of terrestrial allabogdanite**

185 Allabogdanite was identified in several rock samples collected in the midstream of the Halamish Wadi  
186 (Nahal Halamish) in the Hatrurim Basin, Israel, near the southern sub-basin of the Dead Sea (Figure  
187 1b). Phosphide assemblages of the Halamish Wadi were the first discovered in the Mottled Zone  
188 (Britvin et al. 2015, 2017b, 2020c and the references cited therein). Allabogdanite is associated with  
189 barringerite; both minerals typically occur as rounded nodules up to 0.5 mm in diameter, disseminated  
190 in a matrix dominated by a micro-brecciated diopside (up to 70%) (Figure 3). Chemical composition of  
191 the latter corresponds to pure  $\text{CaMgSi}_2\text{O}_6$  as the contents of other elements lie below detection limit of  
192 our WDX detector (~0.05 wt.%) and single-crystal X-ray diffraction confirmed the diopside structure.  
193 No other anhydrous silicates were detected. However, the microbreccia is cemented by a range of  
194 poorly crystallized, late hydrothermal (or possibly secondary) hydrous silicates of Ca, Mg, Fe and  
195 sometimes Ni, as well as calcite and aragonite. Small grains of hematite, magnetite and pyrrhotite are  
196 sporadically encountered. A notable chemical feature of these microbreccias is the almost complete  
197 lack of Al-containing minerals.



198 Allabogdanite is steel-gray colour with metallic lustre; the crushed surfaces are metallic-white.  
199 The mineral is brittle and lacks recognizable cleavage; the Moh's hardness is between 5 and 6. In the  
200 polished sections under reflected light, allabogdanite is yellowish-white with no observable  
201 birefringence and very weak anisotropy. These properties are similar to those of meteoritic  
202 allabogdanite (Britvin et al. 2002) and make allabogdanite optically indistinguishable from barringerite  
203 (Figure 3, 4). The peripheral parts of allabogdanite (and barringerite) grains are commonly corroded  
204 and replaced by the rims of murashkoite, FeP (Britvin et al. 2019c) (Figure 4).

205

### 206 **Chemical composition and structural study**

207 A phosphide-rich section of one of the diopside microbreccias was chosen for further examination.  
208 Each phosphide grain was subjected to electron microprobe analysis (Table 1) and crystallographic  
209 investigation (Table 2). With the exception of grain #9, allabogdanite and barringerite are close to  
210 endmember compositions with 1.6  $\pm$  0.2 and 1.8  $\pm$  0.2 wt% Ni, respectively, while Co and Mo are  
211 below 0.5 wt%. We conducted structure analysis with single-crystal diffraction for allabogdanite and  
212 barringerite crystals of nearly endmember compositions (Table 2). The detailed results of structure  
213 refinements can be retrieved from crystallographic information file (CIF) in the Supplementary Data.

214 Electron backscatter diffraction (EBSD) reveals no orientational relationship between  
215 allabogdanite and barringerite (Figs. 3, 4). Barringerite occurs as single-crystal grains, many of which  
216 are in part, or completely, replaced by allabogdanite (Figure 4). The latter forms fine multiple  
217 interpenetration twins (trillings) formed by the rotation of twin domains about the *b*-axis, as revealed  
218 by EBSD orientation mapping (Figure 3, 4). The compositions of allabogdanite and barringerite sectors  
219 of those grains are identical (Table 1) and they can be distinguished only by their crystal structure. In  
220 order to unambiguously confirm the presence of allabogdanite trillings, a two-phase grain #3 (Figure 4)  
221 was extracted from the polished section for single-crystal structural examination. The latter confirmed

222 (Table 3) the existence of a three-component interpenetrating twin (trilling) of allabogdanite (Figure  
223 4b). It is important with regard to further implications that allabogdanite trillings do not have any  
224 regular crystallographic relations to barringerite – the polymorphs are intergrown in a random way  
225 (Figure 4c). An extreme example of multiple intergrowth and twinning of allabogdanite is shown in  
226 Figure 5. It can be seen that the twin boundaries of allabogdanite domains do not coincide with, but  
227 instead penetrate or abut the habit and interphase boundaries of phosphide grains. The occurrence of  
228 twinning distinguishes terrestrial allabogdanite from the mineral of meteoritic origin (Britvin et al.  
229 2019b) which always appear as single crystals (Figure 6).

230

### 231 **Metastable state of terrestrial allabogdanite**

232 The metastable state of allabogdanite was confirmed from stepwise annealing experiments aimed at  
233 determining the allabogdanite→barringerite transition temperature. The experiment conditions are  
234 reported in the Supplementary Data. It was found that: (1) allabogdanite→barringerite transition occurs  
235 at  $800\pm 50^{\circ}\text{C}$  (Figure 7), (2) this transition is irreversible at atmospheric pressure, (3) barringerite  
236 formed by thermally induced transition forms multi-crystal grains rather than single crystals and, (4)  
237 orientation relationships between twin domains of primary allabogdanite are not inherited by the newly  
238 formed barringerite (Table 4). Therefore, one can suggest that allabogdanite→barringerite conversion  
239 proceeds via reconstructive, rather than displacive phase transformation, similar to the process  
240 previously reported for synthetic  $\text{Fe}_2\text{P}$  (Dera et al. 2008). The physical integrity of mixed  
241 allabogdanite-barringerite grains is maintained due to the small difference (0.8 - 2.5%) in molar volume  
242 (and thus density) between allabogdanite and barringerite (Table 2 and 5), mitigating residual strains in  
243 the crystals. The 1-bar transition temperature presented here is consistent with the earlier data reported  
244 by Senateur et al. (1976) for synthetic  $\text{Fe}_2\text{P}$ , but  $200^{\circ}\text{C}$  lower than that reported by Dera et al. (2008).

245

## 246 ***In situ* high-pressure high-temperature experiments**

247 In order to determine the phase transition conditions for the terrestrial allabogdanite, a fragment of the  
248 most Ni- and Mo-rich grain was chosen (denoted by “X” in Figure 5a and Table 1), as it was expected  
249 that the incorporation of Ni and Mo may substantially affect the pressure-temperature phase boundary.  
250 This grain was crushed and studied by *in situ* high-pressure high-temperature single-crystal X-ray  
251 diffraction in diamond anvil cells (DAC). The details of the experiments are provided in the  
252 Supplementary Data. The results are summarized in Table 5 and graphically expressed in Figure 8.  
253 Based on initial heating runs performed at 4.8 and 7.4 GPa and  $1350\pm 100^\circ\text{C}$ , it was found that  
254 allabogdanite irreversibly transforms into barringerite within  $\sim 5\text{-}8$  min at both pressures. This means  
255 that in order to be preserved (though metastable) at ambient or low-pressure conditions, allabogdanite  
256 has to be quenched below transition temperature at pressures exceeding 7.4 GPa.

257         Additionally, a series of high-temperature compression experiments were performed on the  
258 fragments of the same Ni- and Mo-rich grain (“X” in Figure 5a), to determine the onset of  
259 barringerite→allabogdanite transition. It is noteworthy that, in contrast to the conditions of single-  
260 crystal experiment reported by Dera et al. (2008), each compression step in our experiments was  
261 accompanied by laser heating. Based on our experiments at  $1400\pm 100^\circ\text{C}$ , the barringerite →  
262 allabogdanite transition of the Ni- and Mo-bearing grain occurs at the unexpectedly high pressure of  
263  $25\pm 3$  GPa (Table 5, Figure 8). The measured transition pressure of this grain is significantly greater  
264 than the 8 GPa obtained for pure synthetic  $\text{Fe}_2\text{P}$  (Senateur et al. 1976; Dera et al. 2008). Another  
265 important result of these experiments is that in contrast to the observations of synthetic  $\text{Fe}_2\text{P}$  (Dera et  
266 al. 2008), Hatrurim Formation allabogdanite has a higher density at the transition point than  
267 barringerite. The observed change in density ( $D$ ) is small (0.8%) and thus fits the  $\Delta D/D$  trend typical of  
268 polymorph pairs (Tschauner 2019).

269

270

## Implications

271 Though allabogdanite was discovered 18 years ago (Britvin et al. 2002), it wasn't until 2008 that it was  
272 shown to be a high-pressure phase (Dera et al. 2008), as the high-pressure relationships in the Fe–Ni–P  
273 system have attracted less attention compared to the silicate and oxide systems (Tomioka and Miyahara  
274 2017; Tschauner 2019). The Fe–Ni–P ternary is one of the most important reduced systems in the Solar  
275 System, as phosphides were condensed during the early stages of the nebula formation (Pasek 2019)  
276 and are argued to be substantial constituents of planetary interiors (Scott et al. 2007, 2008; Gu et al.  
277 2011, 2014; He et al. 2019). This work is the first to report terrestrial allabogdanite and the first high-  
278 pressure study of natural allabogdanite. Previous work on synthetic Fe<sub>2</sub>P found that 8 GPa is the low-  
279 pressure limit of phase transformation (Senateur et al. 1973; Dera et al. 2008; Minin et al. 2019). The  
280 unexpectedly high transition pressure of 25±3 GPa for natural allabogdanite can be explained by the  
281 doping effects of Ni (4 wt.%) and Mo (2.5 wt.%), consistent with previous studies that have shown that  
282 synthetic Ni<sub>2</sub>P is stable in the barringerite-type structure up to at least 50 GPa (Dera et al. 2009). The  
283 influence of Mo substitution on the phase transition is less well understood, because both synthetic and  
284 natural MoNiP (monipite) adopt the barringerite structure at ambient conditions (Guérin and Sergent  
285 1977a; Ma et al. 2014), whereas synthetic FeMoP crystallizes in the orthorhombic (allabogdanite-type)  
286 system (Guérin and Sergent 1977b). The difference in the phase boundaries in synthetic versus natural  
287 terrestrial allabogdanite shown in this study underline the importance of exercising caution whilst using  
288 data on endmembers to extrapolate to more compositionally complex natural systems as the influence  
289 of substituting impurities on the conditions of phase transitions can be significant.

290 The experimentally reconstructed formation conditions of allabogdanite from the Mottled Zone  
291 imply that the terrestrial allabogdanite formed from the low-pressure barringerite precursor must have  
292 been subjected to pressures of at least 25±3 GPa. However, there is a major obstacle in adopting this  
293 high-pressure formation scenario, which is the association of terrestrial allabogdanite with pure

294 diopside. Previous studies indicate that diopside is stable under compression up to 17 GPa at 1000-  
295 1400°C but decomposes into a mixture of silicate phases at higher pressures (Kim et al. 1994; Oguri et  
296 al. 1997; Akaogi et al. 2004). Taking into account the accuracy of our pressure measurements ( $\pm 3$   
297 GPa), there is a gap of at least 5 GPa between the upper stability limit of diopside and the recorded  
298 barringerite→allabogdanite transition point. This discrepancy makes it unlikely that allabogdanite  
299 formed as a result of an *in-situ* high-pressure shock event, e.g., a hypothetical Earth-meteoroid  
300 hypervelocity collision (French and Koeberl 2010) or any natural process mimicking high-pressure gas  
301 explosions (e.g., Wilk and Kenkmann 2016). Furthermore, there is no geological evidence for a large  
302 impact structure in the well-studied Dead Sea area (e.g., Garfunkel and Ben-Avraham 1996), including  
303 an absence of other high pressure minerals. The mineral composition and texture of the allabogdanite-  
304 bearing diopside microbreccia (Figure 2) resembles mantle diopsidites from ophiolite complexes  
305 (Python et al. 2007; Ishimaru and Arai 2011; Akizawa and Arai 2014). However, there are no  
306 documented mantle lithologies across a well-explored area surrounding the Dead Sea Transform (e.g.,  
307 Ten Brink et al. 1999). Therefore, the origin of terrestrial allabogdanite in the rocks of the Mottled  
308 Zone remains unresolved and adds to the number of mineralogical enigmas of this unusual  
309 metamorphic complex.

310  
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471 **Captions to the Figures**

472

473 **Figure 1.** The Mottled Zone (Hatrum Formation). **(a)** Distribution of the Mottled Zone outcrops  
474 (brown patches) in the Southern Levant. The largest areas are the Hatrum Basin (Israel) and Daba-  
475 Siwaqa complex (Jordan). **(b)** The Hatrum Basin in the Negev Desert, Israel, and Nahal Halamish –  
476 the locality of allabogdanite-bearing assemblages. **(c)** Eastward view on the Hatrum Basin, Israel  
477 (hills beyond the road 258). Width of panorama is ca. 10 km. The base for Figure 1a,b was adapted  
478 from Britvin et al. (2015) (CC 4.0); the panorama was created by S.N.B.

479

480 **Figure 2.** The idealized relationship between the unit cells of barringerite and allabogdanite. The  
481 dashed lines define the allabogdanite cell. Projection onto  $\{001\}$ (barringerite) =  $\{010\}$ (allabogdanite).  
482 The *c*-axis of barringerite coincides with (and approximately equal to) the *b*-axis of allabogdanite and  
483 is perpendicular to the image plane.  $c(\text{allabogdanite}) \approx \frac{2}{3}a(\text{barringerite})\sqrt{3}$ .

484

485 **Figure 3.** General view of an allabogdanite-barringerite assemblage from the Hatrum Basin. **(a)**  
486 Polished cross-section of diopside microbreccia containing allabogdanite and barringerite grains (white  
487 pebbles), photo in reflected light. **(b)** The same fragment; electron backscatter diffraction (EBSD) maps  
488 superimposed onto the SEM image of back-scattered electrons (BSE). Yellow grains denote  
489 barringerite (colored irrespective of crystallographic orientation), mottled ones are allabogdanite. Grain  
490 numbers correspond to those indicated in Table 1. The area corresponding to Figure 4 is marked by red  
491 square. Di – diopside.

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493 **Figure 4.** Detailed view of barringerite-allabogdanite grain #3 (a red square on Figure 3). **(a)** Photo in  
494 reflected light. Mur - murashkoite (FeP) rims replacing allabogdanite and barringerite. **(b)** EBSD  
495 inverse pole figure (IPF) map of grains #3 and #5. Note that grain #3 is composed of a barringerite

496 single crystal and allabogdanite trilling. Allabogdanite domains are shown by labels colored according  
497 to the scheme of trilling shown in (c). (c) Orientation of barringerite single crystal and three domains of  
498 cyclic allabogdanite twin (trilling) in grain #3 (see Table 3). Abg, allabogdanite; Brr, barringerite; Mur,  
499 murashkoite. Extended information on EBSD mapping of grains #3 and #5 are given in Supplementary  
500 Data (Figure S1).

501  
502 **Figure 5.** Multiply twinned allabogdanite grain intergrown with pyrrhotite-4C. (a) SEM BSE image.  
503 The “X” letter marks the grain which was used for high-pressure high-temperature DAC experiments.  
504 (b) Photo in reflected light. (c) EBSD IPF orientation map. Black areas indicate unindexed points and  
505 unmapped regions. Abg, allabogdanite; Po, pyrrhotite.

506  
507 **Figure 6.** Single crystals of allabogdanite in the Onello iron meteorite. Polished section after nital  
508 etching. SEM BSE image. Legend: Abg, allabogdanite;  $\alpha$ -(Fe,Ni), kamacite;  $\gamma$ -(Fe,Ni), taenite.

509  
510 **Figure 7.** X-ray diffraction patterns illustrating phase transition of allabogdanite to barringerite upon  
511 heating at atmospheric pressure. (a) Allabogdanite (grain #7 in Figure 3) heated under argon flow to  
512 800°C. (b) The same grain heated to 850°C.

513  
514 **Figure 8.** Phase transitions of allabogdanite from the Dead Sea upon compression and heating in the  
515 laser-heated diamond anvil cells (LHDAC). Fragments of a grain depicted on Figure 5a; chemical  
516 composition is given in Table 1. Data on the unit-cell parameters are presented in Table 5. Note the  
517 change of calculated density ( $D_{\text{calc}}$ , left scale) and unit-cell parameters (right scale) during phase  
518 transitions at 7.4 and 28.4 GPa, respectively. The errors (esd's) are smaller than the size of the symbols.

519 The orientation relationships between unit-cell axes of allabogdanite and barringerite are shown in  
520 Figure 2. Abg, allabogdanite; Brr, barringerite; RT, room temperature.

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522

523 **Table 1.** Elemental composition of allabogdanite and barringerite from the Dead Sea basin.

Allabogdanite, wt.%						Allabogdanite, <i>apfu</i> ( $\Sigma = 3$ ) <sup>a</sup>					Grain # <sup>b</sup>
Fe	Ni	Co	Mo	P	Total	Fe	Ni	Co	Mo	P	
76.23	1.74	0.33	0.47	21.36	100.13	1.95	0.04	0.008	0.007	0.99	#01
75.72	1.62	0.14	0.31	21.94	99.73	1.94	0.04	0.003	0.005	1.01	#03 <sup>c</sup>
76.24	1.95	0.18	0.41	21.35	100.13	1.95	0.05	0.004	0.006	0.99	#07
76.33	1.45	0.14	0.19	21.65	100.20 <sup>d</sup>	1.95	0.04	0.003	0.003	1.00	#12
76.66	1.44	0.16	0.29	21.59	100.14	1.96	0.04	0.004	0.004	1.00	#13
76.24	1.64	0.19	0.33	21.58	100.07	1.95	0.04	0.005	0.005	1.00	Avg. of 5
0.34	0.21	0.08	0.11	0.24							Std. dev.
71.88	4.13	0.15	2.47	21.40	100.03	1.86	0.10	0.004	0.037	1.00	Figure 5a <sup>e</sup>

Barringerite, wt.%						Barringerite, <i>apfu</i> ( $\Sigma = 3$ ) <sup>a</sup>					Grain # <sup>b</sup>
Fe	Ni	Co	Mo	P	Total	Fe	Ni	Co	Mo	P	
76.04	1.79	0.33	0.34	21.14	99.64	1.96	0.04	0.008	0.005	0.98	#02
75.84	1.65	0.14	0.20	21.93	99.76	1.94	0.04	0.003	0.003	1.01	#03 <sup>c</sup>
76.18	1.95	0.31	0.32	21.67	100.43	1.94	0.05	0.007	0.005	1.00	#04
75.73	1.81	0.15	0.23	21.64	99.56	1.95	0.04	0.004	0.003	1.00	#05
75.52	2.08	0.12	0.38	21.59	99.69	1.94	0.05	0.003	0.006	1.00	#06
76.07	1.79	0.16	0.68	21.47	100.17	1.95	0.04	0.004	0.010	0.99	#08
77.08	2.07	0.13	0.28	21.20	100.76	1.97	0.05	0.003	0.004	0.98	#10
76.45	1.76	0.20	0.31	21.29	100.01	1.96	0.04	0.005	0.005	0.99	#11
76.41	1.52	0.14	0.43	21.67	100.59 <sup>f</sup>	1.95	0.04	0.003	0.006	1.00	#14
75.86	1.68	0.17	0.18	21.45	99.34	1.96	0.04	0.004	0.003	1.00	#15
76.12	1.81	0.19	0.34	21.51	99.97	1.95	0.04	0.005	0.005	1.00	Avg. of 10
0.45	0.18	0.07	0.14	0.24							Std. dev.
56.19	22.36	0.18	0.25	21.16	100.14	1.45	0.55	0.004	0.004	0.99	#09, max Ni
73.36	5.34	0.18	0.29	21.32	100.49	1.88	0.13	0.004	0.004	0.98	#09, min Ni

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525 <sup>a</sup> *apfu* = atoms per formula unit. <sup>b</sup> Grain numbers correspond to those indicated in Figure 3. <sup>c</sup> Intergrowth of  
526 allabogdanite and barringerite (Figure 4). <sup>d</sup> Contains vanadium, 0.44 wt.% (0.01 atoms per formula unit). <sup>e</sup> The  
527 grain used in high-pressure high-temperature experiments. <sup>f</sup> Contains vanadium, 0.42 wt.% (0.01 atoms per  
528 formula unit).  
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**Table 2.** Crystallographic data for allabogdanite and barringerite grains #1 and #5

	Allabogdanite #1	Barringerite #5
Crystal system	Orthorhombic	Hexagonal
Space group	<i>Pnma</i>	$\bar{P}62m$
<i>a</i> (Å)	5.832 (2)	5.8618 (13)
<i>b</i> (Å)	3.5971 (13)	
<i>c</i> (Å)	6.713 (2)	3.4598 (11)
<i>V</i> (Å <sup>3</sup> )	140.83 (9)	102.95 (6)
<i>Z</i>	4	3
<i>V/Z</i> (Å <sup>3</sup> )	35.20	34.32
<i>D</i> <sub>calc</sub> (g cm <sup>-3</sup> )	6.729	6.903

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**Table 3.** Euler orientation matrices of allabogdanite domains in grain #03 (Figure 4)<sup>a</sup>

Domain #	Mass fraction (%)	Isolated / overlapped reflections	Orientation matrix
1	41	4560 / 1596	[1 0 0] [0 1 0] [0 0 1]
2	32	4560 / 1592	$[-\frac{1}{2} 0 -\frac{3}{4}]$ [0 1 0] [1 0 $-\frac{1}{2}$ ]
3	27	4553 / 1599	$[\frac{1}{2} 0 -\frac{3}{4}]$ [0 1 0] [1 0 $\frac{1}{2}$ ]

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<sup>a</sup> Domain #2 is rotated by  $-120^\circ$ ; domain #3 by  $60^\circ$  around [010]. See datablock Allabogdanite\_Trilling\_hklf5 in the attached CIF file. Software used: CrysAlisPro (2018).

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**Table 4.** Rotation matrices of allabogdanite and barringerite domains before and after thermally induced allabogdanite→barringerite transition (grain #13)

Domain #	Mass fraction (%)	Rotation angle (°)	Rotation axis		
			<i>h</i>	<i>k</i>	<i>l</i>
Allabogdanite before transition					
1	39	0	0	1	0
2	23	120.02	0	1	0
3	24	59.70	0	1	0
Barringerite after transition					
1	48	0	1	0	0
2	40	-146.49	0.61	-0.06	-0.79
3	2	90.11	0	-0.13	-0.99

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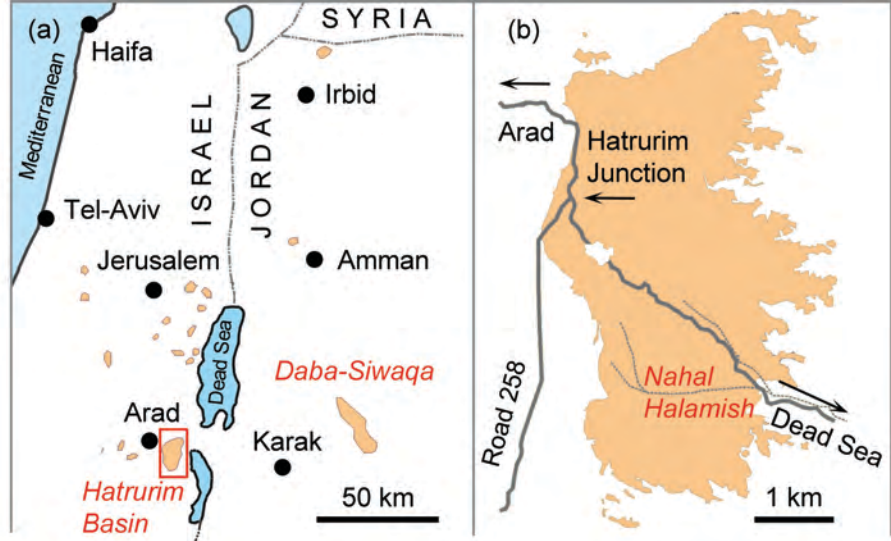
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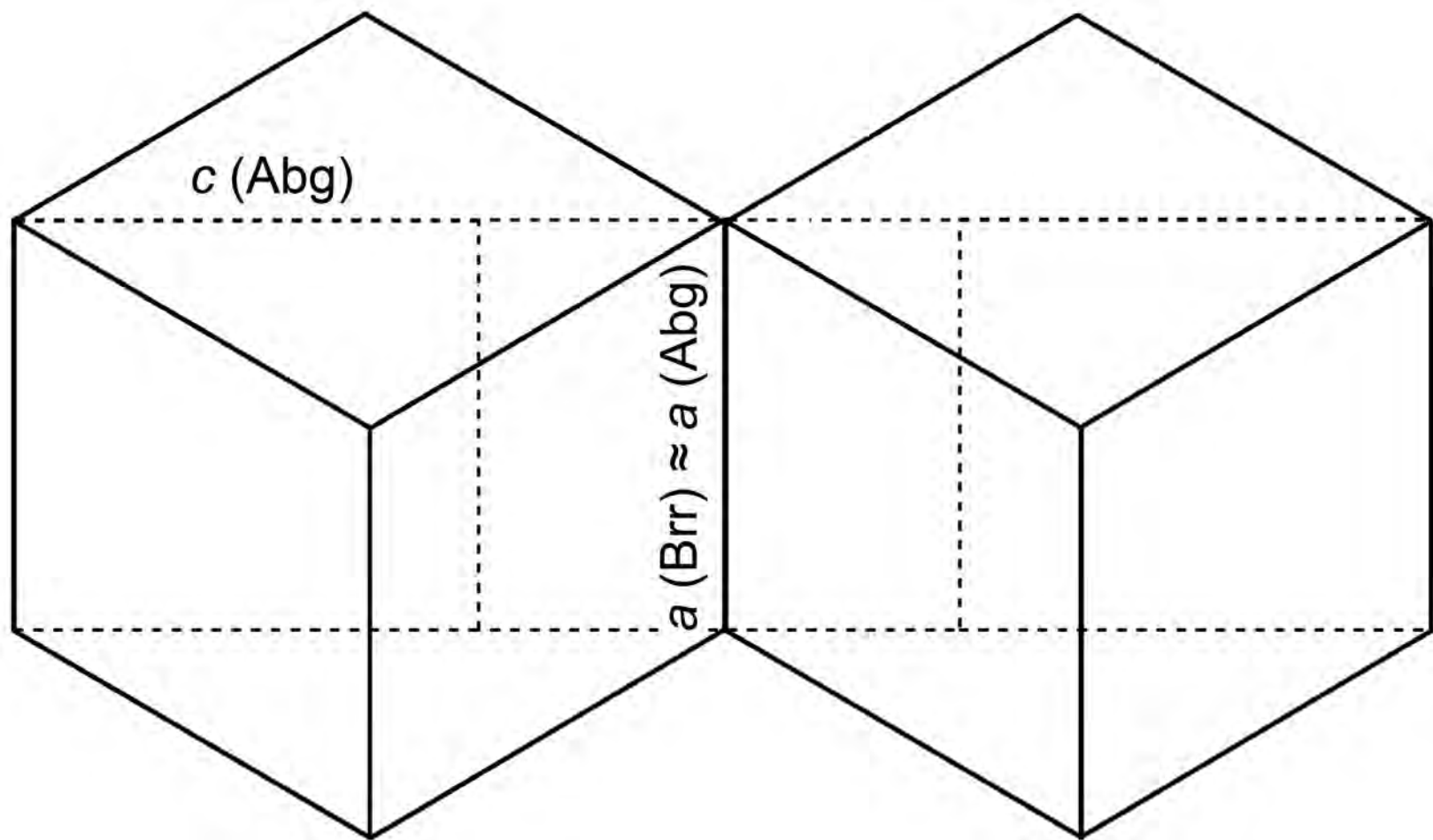
561  
562**Table 5.** *In situ* phase transitions of allabogdanite upon compression/heating in diamond anvil cells <sup>a</sup>

Run <sup>b</sup>	<i>P</i> (GPa)	<i>T</i> (°C) <sup>c</sup>	Phase <sup>d</sup>	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	<i>V/Z</i> (Å <sup>3</sup> )	<i>D</i> <sub>calc</sub> (g cm <sup>-1</sup> )
0	0.0	–	Abg	5.7800(1)	3.5637(1)	6.6478(3)	34.23	7.008
1	2.0	–	Abg	5.7429(13)	3.5581(15)	6.6245(11)	33.84	7.089
1	4.8	–	Abg	5.6953(5)	3.5484(6)	6.5973(4)	33.33	7.197
2	7.4	–	<i>Abg</i>	<i>5.6487(9)</i>	<i>3.5424(14)</i>	<i>6.571(18)</i>	<i>32.87</i>	<i>7.298</i>
2	7.4	1350	<i>Brr</i>	<i>5.7747(4)</i>		<i>3.4439(2)</i>	<i>33.15</i>	<i>7.236</i>
2	13.5	1350	Brr	5.7360(8)		3.3787(4)	32.09	7.476
1	19.6	1350	Brr	5.677(15)		3.3694(8)	31.34	7.654
2	22.0	1350	Brr	5.6356(4)		3.3682(2)	30.88	7.769
2	28.4	–	<i>Brr</i>	<i>5.5954(5)</i>		<i>3.3368(3)</i>	<i>30.16</i>	<i>7.955</i>
2	28.4	1400	<i>Abg</i>	<i>5.423(7)</i>	<i>3.4558(8)</i>	<i>6.3853(9)</i>	<i>29.92</i>	<i>8.019</i>
2	32.2	1400	Abg	5.346(2)	3.4474(3)	6.3958(3)	29.47	8.142
2	39.0	1400	Abg	5.307(3)	3.4356(4)	6.3701(5)	29.04	8.262

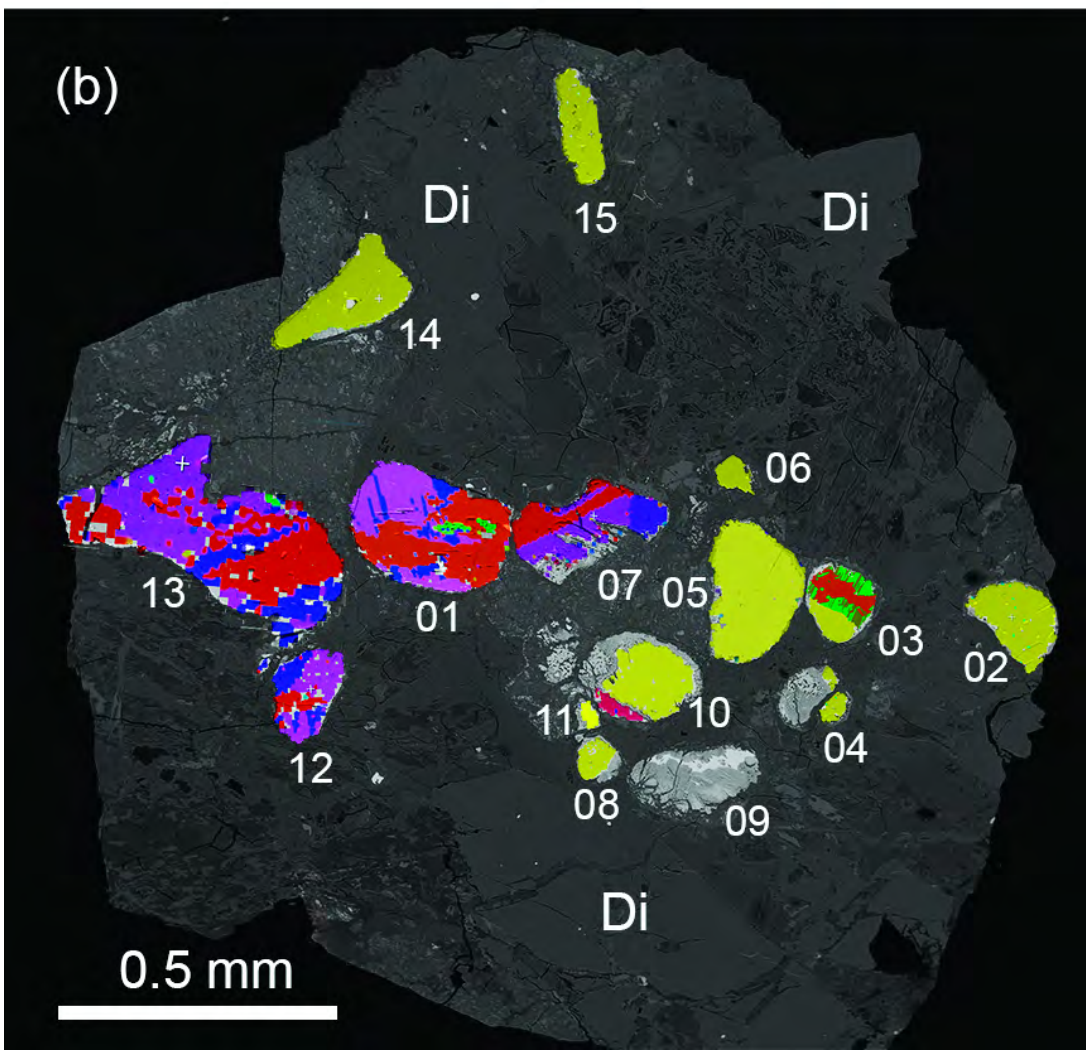
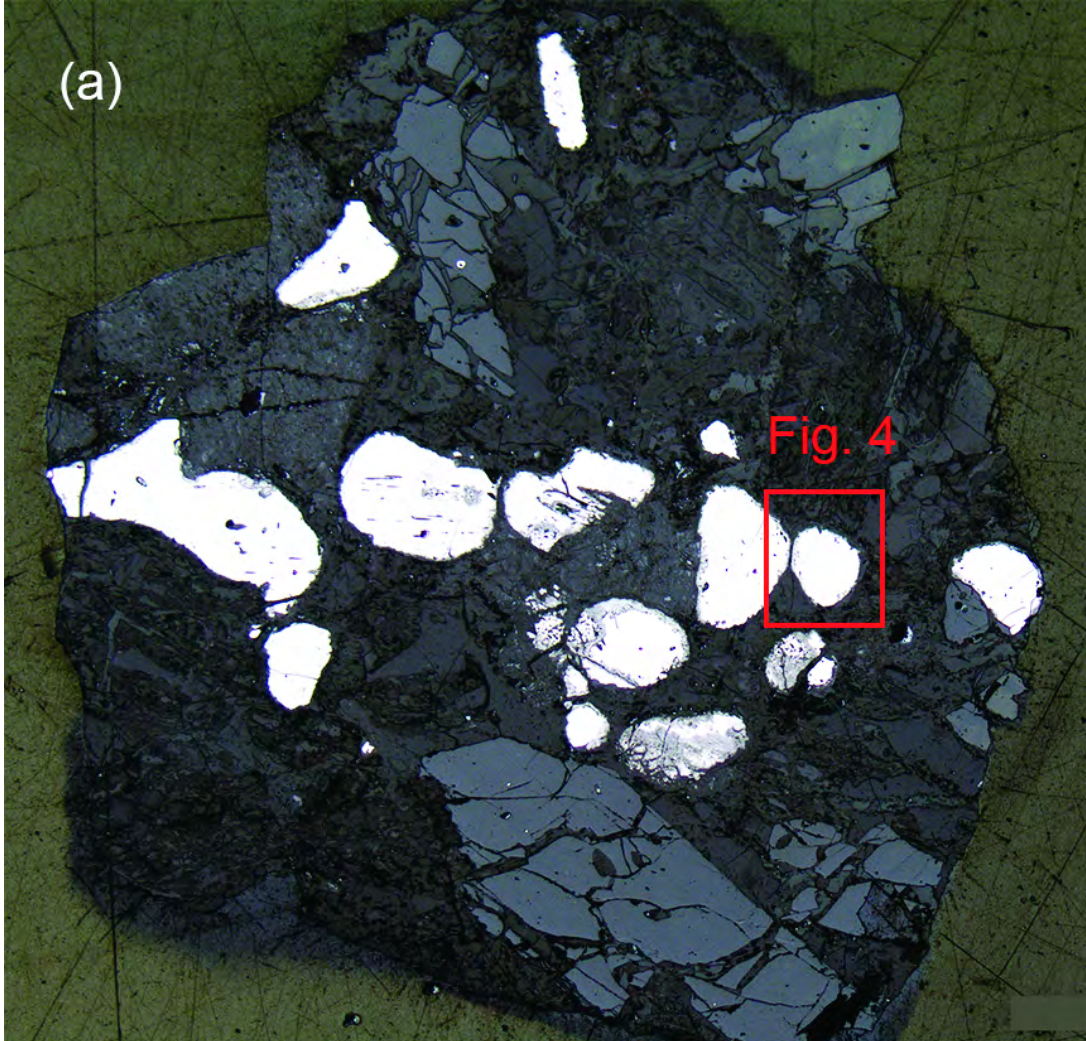
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564 <sup>a</sup> Single-crystal fragments of a grain marked as “X” in Figure 5a (last analysis record in Table 1). Records  
565 related to phase transition points are typed in italic. <sup>b</sup> Run #0 – measurement under ambient conditions; runs #2  
566 and #3 – compression/heating experiments. <sup>c</sup> Accuracy of temperature measurements: ±100 °C. <sup>d</sup> Abg,  
567 allabogdanite; Brr, barringerite.  
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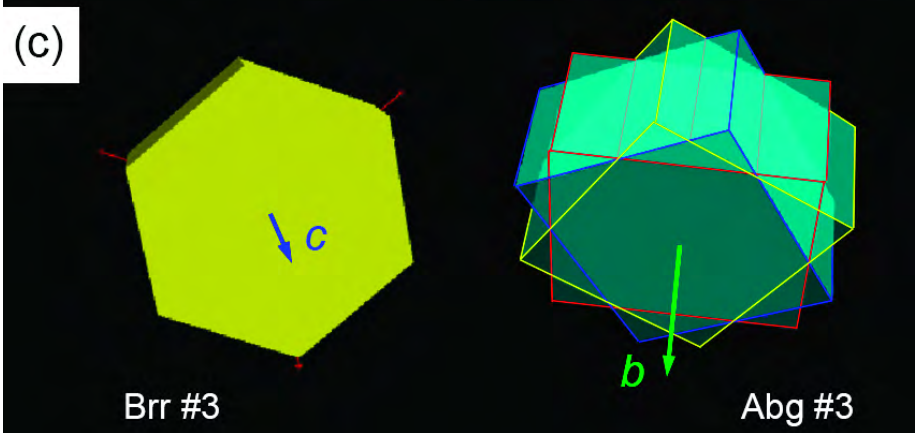
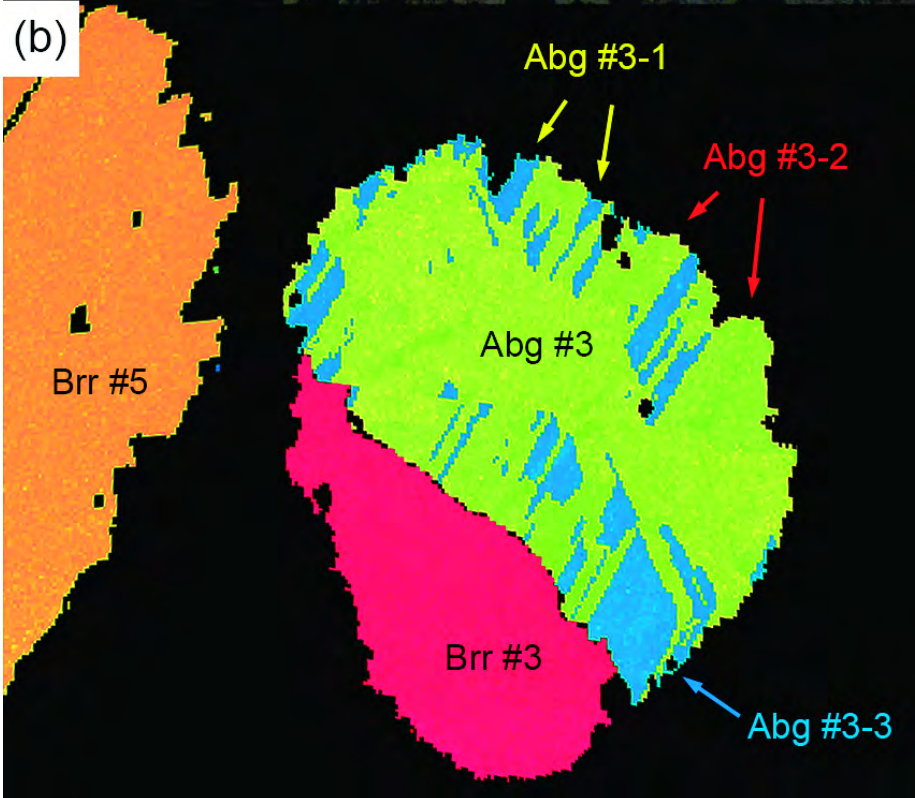
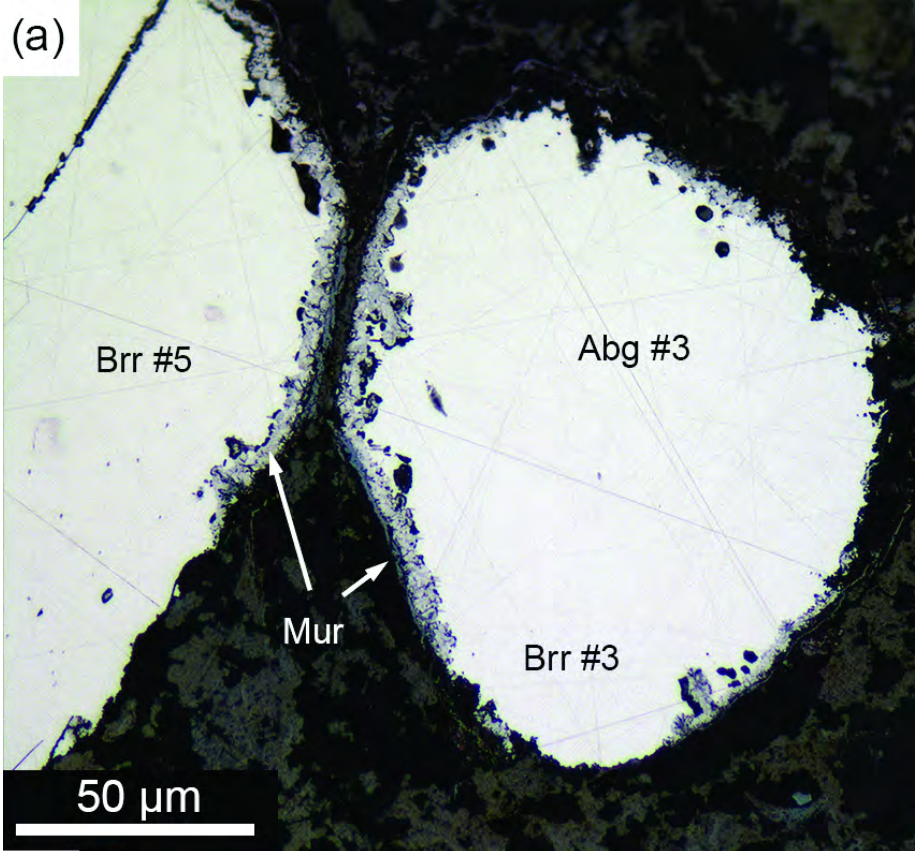




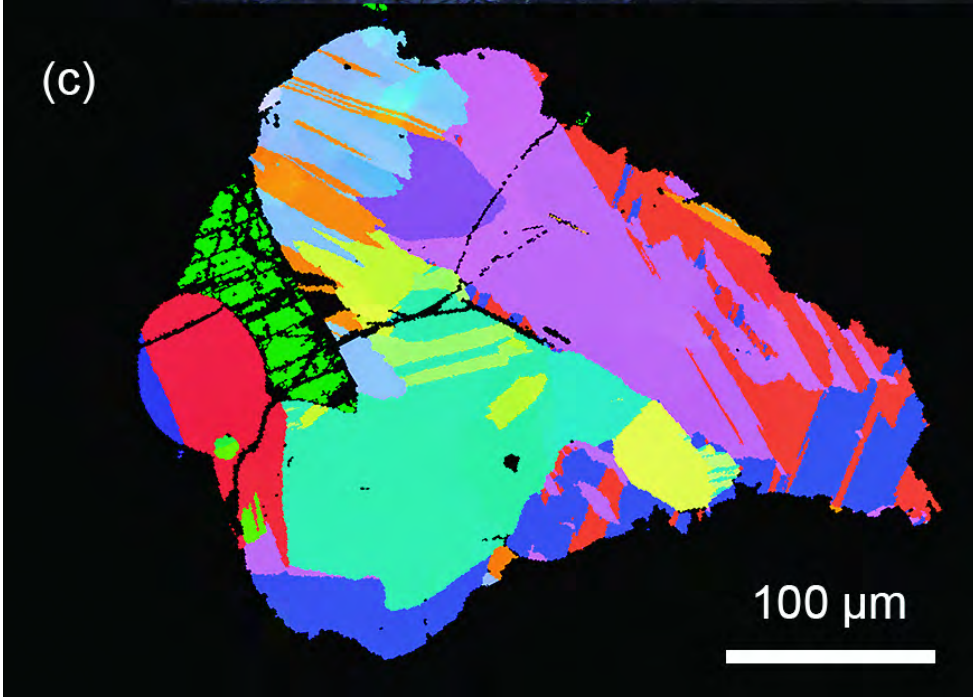
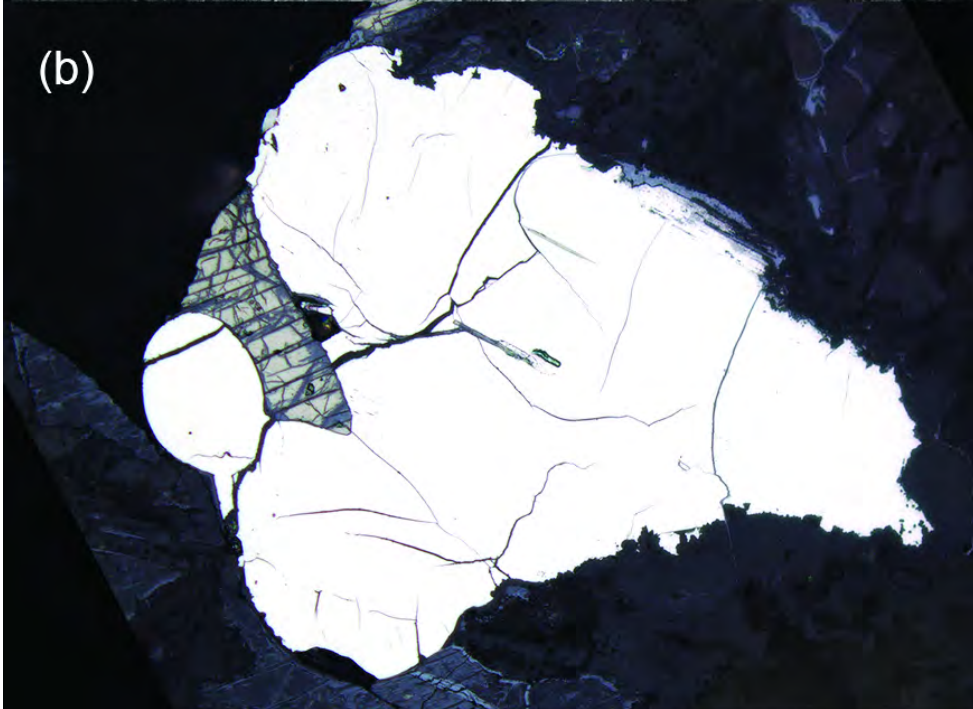
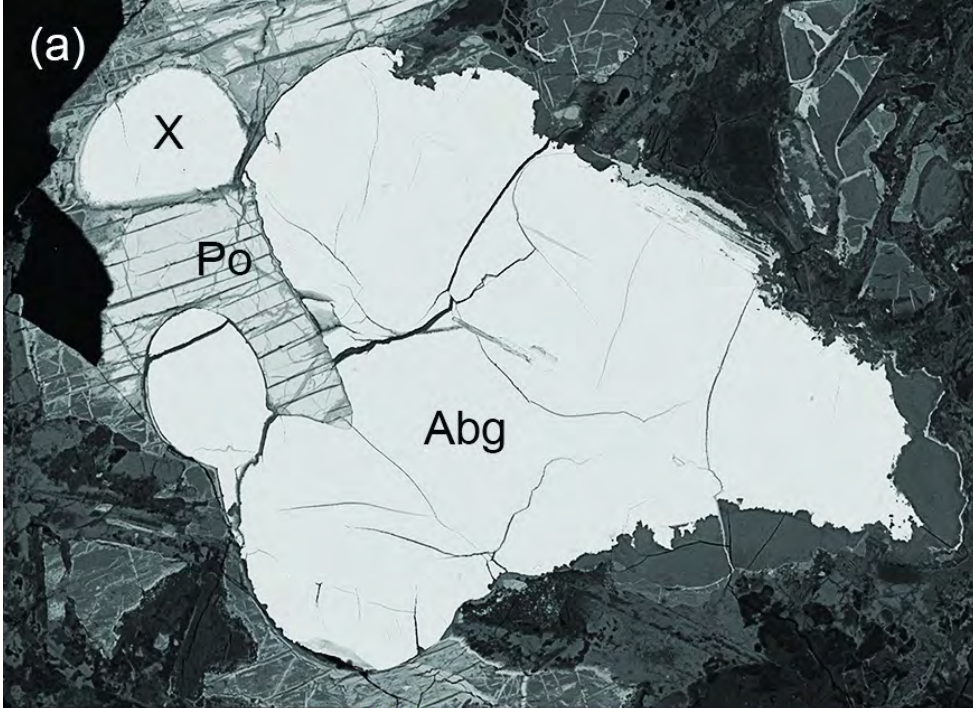




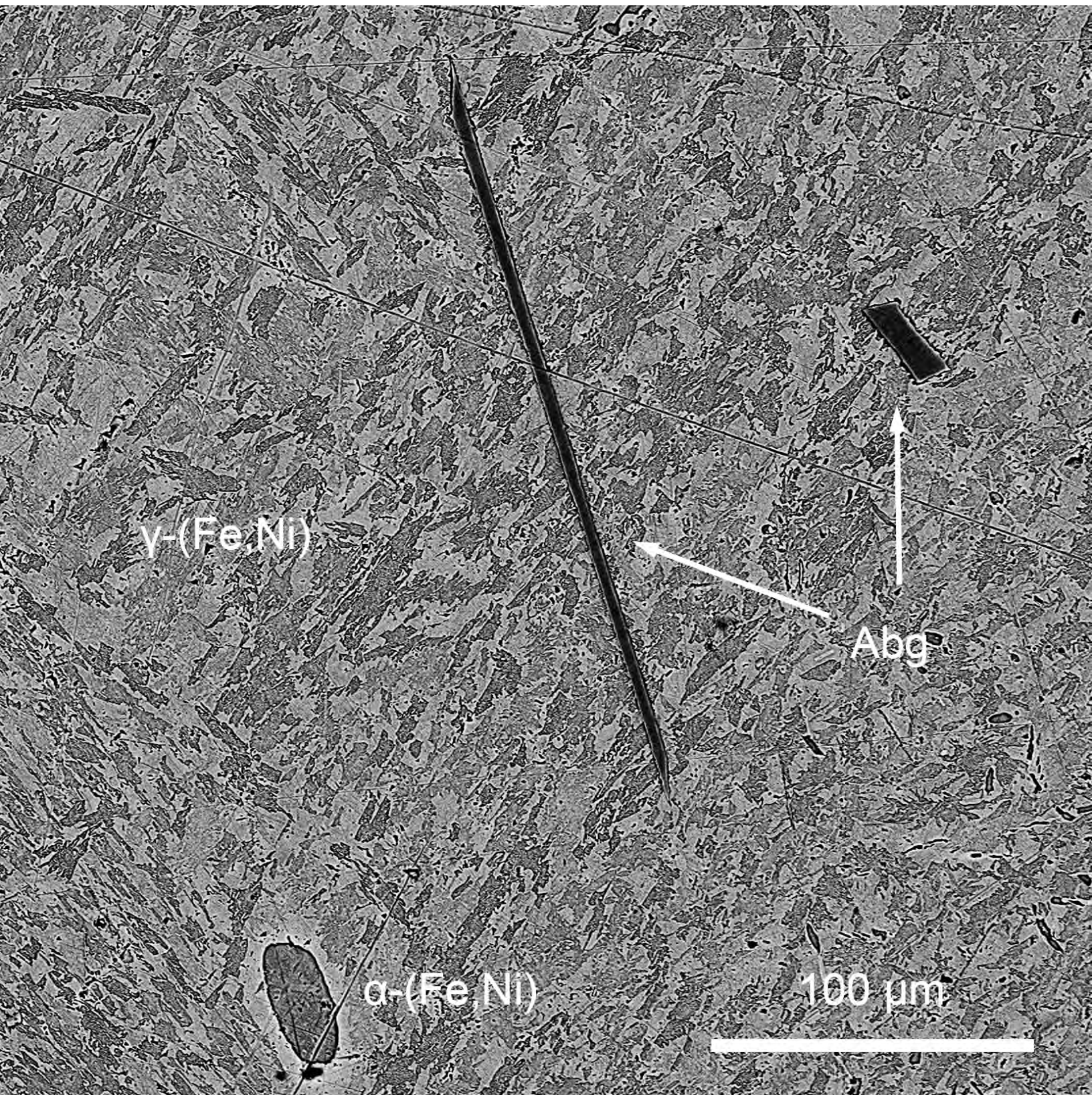












$\gamma$ -(Fe,Ni)

$\alpha$ -(Fe,Ni)

Abg

100  $\mu\text{m}$

