Magnetoelectric coupling in the mixed erythrosiderite $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$

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We present a study of the dielectric, structural, and magnetic properties of the multiferroic or linear magnetoelectric substitution series $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$. Pyroelectric currents, magnetic susceptibilities, and thermodynamic properties were examined on large single crystals of the erythrosiderite compounds and detailed magnetic-field versus temperature phase diagrams are derived for three different substitution levels. With increasing potassium concentration the material is tuned from a multiferroic ($x \leq 0.06$) to a linear magnetoelectric ($x \geq 0.15$) ground state. In contrast to the respective pure parent compounds with x = 0 or 1, however, the ferroelectric or linear magnetoelectric polarization in none of the substituted samples is switchable by external electric fields, because these samples exhibit a significant electric polarization already above the magnetic ordering transition. The polarization arises at a higher-lying structural phase transition that is examined by THz spectroscopy and, on a deuterated pure single crystal, by comprehensive neutron-diffraction experiments. The structural phase transition is attributed to an ordering of NH_4^+ tetrahedra but does not break inversion symmetry in the pure material, while a finite K content causes pyroelectricity.

I. INTRODUCTION

During the last decade spin-driven multiferroics with complex antiferromagnetic order, e.g., spiral or cycloidal spin structures, inspired intensive research activities [1– 6]. These magnetic ordering phenomena can induce a ferroelectric polarization, thus ferroelectric and magnetic order are strongly coupled. These so-called type-II multiferroics are improper ferroelectrics; their polarization is rather weak but can be strongly influenced by external magnetic fields. Studies on multiple spin-driven multiferroic materials, e.g., TbMnO₃ [7], Ni₃V₂O₈ [8], MnWO₄ [9, 10], and NaFe X_2O_6 (X = Si, Ge) [11–14], determined the underlying coupling mechanisms. Most members show two magnetic ordering transitions with a paraelectric intermediate phase with incommensurate and collinear order and a low-temperature ferroelectric phase arising from an inversion-symmetry breaking magnetic structure.

Erythrosiderites are another class of materials in which multiferroicity was found [15, 16]. In these compounds with A_2 [Fe X_5 (H₂O)], where A stands for an alkali-metal or an ammonium ion and X for a halide ion, the members are structurally similar, mostly crystallizing in Pnma structure [17–23], see Fig. 1. Erythrosiderites consist of isolated complex groups of [FeCl₅(H₂O)]²⁻, where the magnetic iron(III) is six-fold octahedrally coordinated, and of alkali-metal ions or NH₄⁺ groups. Hydrogen bonds via O-H-Cl interconnect neighboring [FeCl₅(H₂O)]²⁻ octahedra along the **b** axis and stabilize the structure. Interestingly, only one member of this group, (NH₄)₂[FeCl₅(H₂O)], shows multiferroic order [15]. In a two-step magnetic phase transition the

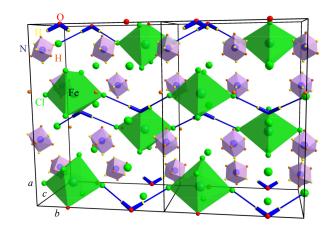


FIG. 1. High-temperature structure of $(NH_4)_2[FeCl_5(H_2O)]$ with symmetry *Pnma*. Fe (green) is surrounded by five Cl ions (green) and one H₂O molecule (O is shown in red, H in yellow). Thick blue lines are OH bonds, thin blue lines mark hydrogen bonds between FeCl₅(H₂O) octahedra forming zigzag chains along **b**, with Fe-Fe distances of 6.47 Å. However, the distances to 8 Fe ions of neighboring chains are only slightly larger (≤ 7.04 Å). The NH₄⁺ groups (magenta tetrahedra) are disordered as is modeled by superposing an NH₄ tetrahedron with its inversion with respect to the N position, with H drawn in orange and yellow, respectively.

ammonium compound first orders antiferromagnetically around 7.2 K [24]. In the analogous deuterated material, a collinear sinusoidal spin structure was reported for this phase [25, 26]. At the second transition at 6.7 K the spins reorient and an electric polarization arises [15]. The low-temperature structure was identified to possess a cycloidal order, which causes electric polarization via the inverse Dzyaloshinskii-Moriya interaction [27, 28]. By applying a magnetic field, the polarization can be turned by 90° , which can be explained by a spin-flop transition at $H_{\rm SF}$ and thus a change of the cycloidal spin structure towards a structure with quasi-collinear antiferromagnetic moments perpendicular to a [29]. This transition shows strong magnetoelastic coupling [15, 30] and in the high-field regime the polarization can be explained with the p-d hybridization mechanism [29]. Magnetic saturation is reached around 30 T [31]. K₂[FeCl₅(H₂O)] exhibits collinear antiferromagnetic order below $T_{\rm N}$ = 14.3 K [16, 32]. In this phase, applied magnetic fields induce a polarization via the linear magnetoelectric effect $P_i = \alpha_{ij} \mu_0 H_j$ [16]. Furthermore, for the multiferroic ammonium compound a high-temperature structural phase transition $(T_{\rm S} = 79 \,\mathrm{K})$ is reported, which was attributed to an ordering of hydrogen atoms of the ammonium groups [15]. An analogous transition thus cannot occur in the alkali-metal erythrosiderites and was consequently not observed there.

In this paper, we study the influence of chemical substitution on the low-temperature ordered phases in the substitution series $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ with three different potassium concentrations of x = 0.06, 0.15,and 0.45. This substitution systematically tunes the system from a cycloidal spin structure with a multiferroic ground state towards a collinear antiferromagnet with linear magnetoelectric coupling. Most interestingly, the high-temperature structural transition in $[(NH_4)_{1-x}K_x]_2$ [FeCl₅(H₂O)] is found to be accompanied by a significant change of electric polarization in all partially substituted compounds, which opens a new possibility to control antiferromagnetic domains. Usually, antiferromagnetic antiphase domains cannot be controlled by an external magnetic field. Only for a linear magnetoelectric material and by combining magnetic and electric fields one may drive the antiferromagnetic domains [33– 35]. In $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ the electric polarization induced by the structural phase transition can take the role of the electric field. Thus, it becomes possible to drive antiferromagnetic domains just by an external magnetic field.

II. EXPERIMENTAL

Single crystals of $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ were grown from aqueous solutions. Both end members of this series, $(NH_4)_2[FeCl_5(H_2O)]$ and $K_2[FeCl_5(H_2O)]$ show incongruent solubility and have to be grown with an excess of FeCl₃ and a surplus of HCl. This was also applied for the growth of the mixed crystals. Here, a molar ratio of the solute constituents, FeCl₃ and $[(1-x) NH_4Cl + x$ KCl], of 2 : 1 was used. Crystal growth was performed by slow temperature reduction between 40 °C and 36 °C over a period of six months using three solutions with $x \approx 0.067$, $x \approx 0.15$, and $x \approx 0.5$. Large single crystals with dimensions of about 1 cm^3 were obtained for all three compositions. The lattice constants of these mixed crystals and of the corresponding pure NH₄- and K-based end members were obtained via refinement of X-ray powder diffraction data. Assuming the validity of Vegards's rule, i.e., a linear variation of the lattice constants with x, actual potassium contents of $x \approx 0.06$, 0.15, and 0.45 are derived for the mixed crystals.

The magnetic susceptibility measurements were performed using a commercial SQUID magnetometer (MPMS, Quantum Design) from 2 K to 300 K in a magnetic field of 0.1 T. The temperature and magnetic-field dependencies of the electric polarization were determined by time integration of the induced currents, which were measured by an electrometer (Keithley 6517). Typically, a static electric poling field of about $200 \,\mathrm{V \, mm^{-1}}$ was applied to ensure single-domain states upon cooling, which, however, has no effect in these samples, as will be seen below. The temperature was changed with various rates between 0.5 and $3 \,\mathrm{K\,min^{-1}}$. For the magnetocurrent measurements, the samples were cooled with an electric poling field to various constant temperatures and complete magnetic-field loops $(7 \text{ T} \rightarrow -7 \text{ T} \rightarrow 7 \text{ T})$ were performed with field-sweep rates of $1 \,\mathrm{T}\,\mathrm{min}^{-1}$ keeping the poling field constant. The specific heat was measured between 2 K and 300 K in a commercial calorimeter (PPMS, Quantum Design) by using the thermal relaxation-time method. Terahertz transmittance data were measured on $(NH_4)_2$ [FeCl₅(H₂O)] using a phase-sensitive continuouswave THz spectrometer [36, 37] equipped with a linear polarizer and a He bath cryostat.

The crystal structure was investigated on a $(ND_4)_2[FeCl_5(D_2O)]$ single crystal of $(40 \pm 2) \text{ mg}$ by neutron diffraction using the diffractometer HEiDi [38] at the MLZ. The crystal was mounted with its c axis parallel to the ϕ axis of the diffractometer and cooled with a closed-cycle refrigerator. Wavelengths of 1.170 Å and 0.793 Å were obtained from a Ge (311) resp. Ge (422) monochromator in combination with an Er filter. Data were taken with a single detector.

III. RESULTS AND DISCUSSION

A. Structural phase transition

Figure 2a) shows the specific-heat data that reveal several transitions. The structural transition at $T_{\rm S}$ = 79 K of pure (NH₄)₂[FeCl₅(H₂O)] systematically shifts to lower temperatures with increasing potassium content ($T_{\rm S}(x = 0.06) = 72 \,\mathrm{K}$, $T_{\rm S}(x = 0.15) = 66 \,\mathrm{K}$). For x = 0.45 the structural transition becomes very broad and occurs around 24 K, see Fig. 2d). In addition, there are two magnetic transitions at $T_{\rm N} = 7.25 \,\mathrm{K}$ for the onset of magnetic ordering and at $T_{\rm C} = 6.8 \,\mathrm{K}$, where (NH₄)₂[FeCl₅(H₂O)] becomes ferroelectric, see inset c). For the smallest substitution x = 0.06 (red) there are still two clearly separated peaks at $T_{\rm N} = 7.5 \,\mathrm{K}$ and $T_{\rm C} = 7 \,\mathrm{K}$

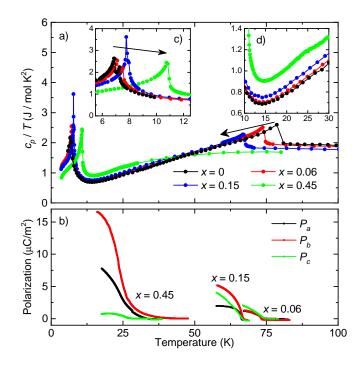


FIG. 2. a) Specific heat of the substitution series $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$. Inset c) resolves the low-temperature transitions and inset d) enlarges the region around the broadened high-temperature transition for x = 0.45 around 24 K. b) Temperature dependence of the polarization components P_a , P_b , and P_c for all different potassium concentrations. The pure $(NH_4)_2[FeCl_5(H_2O)]$ does not show a measurable polarization at T_S .

meaning that both magnetic transitions shift to higher temperature with increasing x. For the higher substitution levels, there is only one anomaly indicating a single second-order phase transition with enhanced transition temperatures $T_{\rm N} = 7.9$ K and 10.8 K for x = 0.15 and 0.45, respectively. The change from two separate transitions to a single transition indicates a change in dielectric and magnetic properties, from spin-spiral multiferroic order for $x \leq 0.06$ to an easy-axis linear magnetoelectric ground state for $x \geq 0.15$, in agreement with the respective ground states of the corresponding pure end members with x = 0 and 1 [15, 16].

The high-temperature structural transition around 79 K was proposed to be an ordering of the $\rm NH_4^+$ groups because it is invisible for X-ray diffraction [15, 27]. This scenario naturally explains the decrease of $T_{\rm S}$ with increasing x, i.e., decreasing $\rm NH_4^+$ concentration, as well as the strong broadening of the corresponding anomaly for x = 0.45, where the high degree of substitution prevents the occurrence of a sharp structural phase transition. Pure (NH₄)₂[FeCl₅(H₂O)] does not show any measurable polarization at $T_{\rm S}$ and a high-temperature polarization neither is observed in K₂[FeCl₅(H₂O)], which does not show the structural phase transition. In contrast, for partially substituted [(NH₄)_{1-x}K_x]₂[FeCl₅(H₂O)] a sizable polarization evolves below $T_{\rm S}$ and its magni-

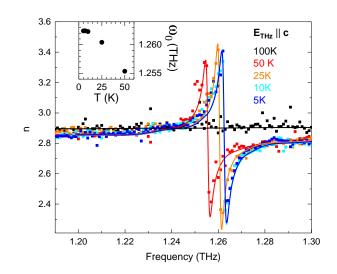


FIG. 3. Refractive index of $(NH_4)_2$ [FeCl₅(H₂O)] for polarization **E** ||*c*. Below T_S , we observe the appearance of a mode at about 1.26 THz, which we attribute to a libration of the NH_4^+ groups. Symbols show the experimental results, solid lines depict fits using a Lorentz oscillator. The inset resolves the temperature dependence of the eigenfrequency of this mode.

tude systematically increases with increasing x for P_a and P_b , whereas P_c shows no clear trend within the experimental resolution, see Fig. 2b). For comparison, the spontaneous polarization in the multiferroic phase of $(NH_4)_2$ [FeCl₅(H₂O)] amounts to approximately $2\,\mu\text{C}\,\text{m}^{-2}$, which is even smaller than the absolute value of the high-temperature polarization for x = 0.06. Surprisingly, the polarization at T_S is not affected by electric fields up to $200\,\text{V}\,\text{mm}^{-1}$. This means that the partially substituted materials are pyroelectric and have a polar structure at least below T_S . We note that these pyrocurrent measurements are only sensitive to polarization changes instead of the absolute polarization.

Clear fingerprints of the structural phase transition are also observed in the low-energy excitations. We studied $(NH_4)_2$ [FeCl₅(H₂O)] by phase-sensitive THz spectroscopy in transmittance geometry in the frequency range from 0.1 to 1.4 THz. Data were collected on freshly polished surfaces. The electric field was either parallel to the **b** or the **c** axis. Above $T_{\rm S} = 79 \,\rm K$, the data reveal insulating behavior and a nearly frequency-independent refractive index $n(\omega) \approx 3$. The small observed dispersion in $n(\omega)$ can be described by assuming a phonon mode with an eigenfrequency above 2 THz. However, below $T_{\rm S}$ a feature with an eigenfrequency of about 1.26 THz $(\approx 5.2 \,\mathrm{meV})$ appears in the studied frequency range, see Fig. 3, which signals the structural phase transition. This new mode may either be attributed to a change of selection rules at the phase transition or to a strong shift of its eigenfrequency. The eigenfrequency shifts by about 0.6%between 5 K and 50 K, see inset of Fig. 3. In many ammonium compounds an order-disorder transition of the NH_{4}^{+} group is known, e.g., in the large family of alums [39]. In

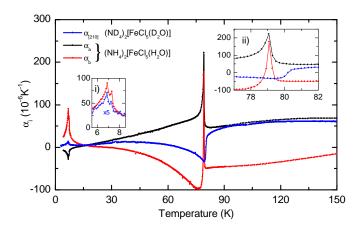


FIG. 4. Thermal expansion $\alpha_{[210]}$ measured along the [2 1 0] direction of $(ND_4)_2$ [FeCl₅(D₂O)] in comparison to the *a*- and the *b*-axis expansions of $(NH_4)_2$ [FeCl₅(H₂O)] from Ref. [15]. An enlarged view of the low-temperature transitions is shown in inset i) with $\alpha_{[210]}$ multiplied by a factor of 5 to allow for a better comparison. Inset ii) enlarges the structural transition-temperature regime.

an ordered structure with preferred NH₄ orientation, the ammonium ions may show libration, i.e., torsional oscillation around the equilibrium orientation. Librational frequencies vary strongly for different compounds [40]. Excitation energies of, e.g., 5.9 meV and 4 meV were observed in neutron scattering on $(NH_4)_2PdCl_6$ [41] and $K_{1-x}(NH_4)_xSCN$ [42], respectively, suggesting a common origin with the 5.2 meV mode in $(NH_4)_2$ [FeCl₅(H₂O)]. A disordered phase, in contrast, may show a much lower potential barrier between different orientations and thus a very different excitation energy.

Since the K substituted samples exhibit a finite polarization below the structural transition temperature, it appeared interesting to reanalyze the character of this transition. Therefore, the crystal structure of deuterated (ND₄)₂[FeCl₅(D₂O)] was studied in the paramagnetic state at 100 K (above $T_{\rm S}$) and at 10 K (below $T_{\rm S}$) by single-crystal neutron diffraction. Thermal expansion measurements presented in Fig. 4 show that the isotope exchange hardly affects the transition temperatures. The magnetic and multiferroic transition temperatures remain unchanged, whereas the structural transition temperature slightly increases from 79 K to 80.2 K.

The results of temperature-dependent neutron diffraction studies are presented in Fig. 5. The (0 4 0) Bragg reflection splits below the structural phase transition indicating a monoclinic distortion, as it can be expected for a ferroelastic transition from orthorhombic to monoclinic symmetry. The occurrence of monoclinic domains hides the breaking of the mirrorplane perpendicular to the *b* axis in the orthorhombic phase. Therefore, the two monoclinic domain orientations correspond to a clockor anticlockwise arrangement of the γ angle in the *a*, *b* plane and the (040) from the first domain orientation is rotated only by $2 \cdot (\gamma - 90)$ degrees from the (0-40)

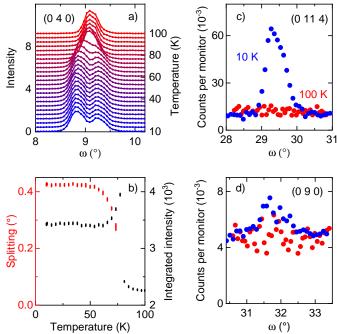


FIG. 5. Neutron diffraction results on the structural phase transition in $(ND_4)_2$ [FeCl₅(D₂O)]. Panel a) shows the temperature dependence of the (0 4 0) ω -scan profile, that exhibits a splitting in the low-temperature phase due to the appearance of monoclinic domains (neutron wavelength of 1.170 Å). Note, that the indices of Bragg reflections do not change across the phase transition, as we use the non standard setting $P2_1/a$ for the low-temperature phase, in which the monoclinic axis is c. Panel b) shows the splitting of the (0 4 0) profiles and the integrated intensity. Panel c) and d) show the ω -scans obtained above and below the phase transition for (0 11 4) and (0 9 0), respectively. Note that both reflections are forbidden in the high-temperature symmetry.

reflection of the second domain. Therefore, the splitting angle of the (040) reflection allows us to determine the monoclinic angle. Note, however, that there is no longitudinal splitting expected as both contributions posses exactly the same d value. The temperature dependence of this monoclinic distortion well agrees with an orderdisorder character of the transition. There is a clear peak in the integrated intensity of this reflection just at $T_{\rm S}$, which points to the impact of the ferroelastic domains on the extinction. Close to the phase transition the domain structure severely reduces the extinction conditions and thereby enhances the scattering signal, and also the fixed domain structure at lower temperature corresponds to a lower crystal quality, less extinction and thus higher intensities. The glide plane n in space group Pnma implies a special selection rule that k + l must be even for (0 k l) reflections. The observation of reflections at low temperature that violate these special selection rules, see Fig. 5 c) and d), unambiguously shows that this glide plane is broken below $T_{\rm S}$. There are, however, no superstructure reflections that can be associated with a loss of translation symmetry. The structural distortion must be

Fe
N
Cl
O

100 K



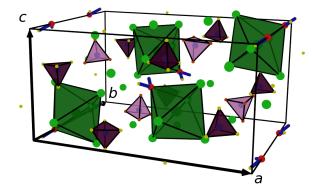


FIG. 6. Crystal structure of $(ND_4)_2$ [FeCl₅(D₂O)] at 100 K (top) and 10 K (bottom). ND₄⁺ tetrahedra of the same shading are symmetry equivalent.

described by a Γ mode [43].

Full data sets of Bragg reflection intensities were collected at $100 \,\mathrm{K}$ (in total 3.335 reflections) and at $10 \,\mathrm{K}$ $(5\,204 \text{ reflections})$. The data collected at 100 K (10 K)yield an internal R value of 2.02% (1.53%) when the data are reduced according to space group Pnma $(P2_1/a)$. The refinement with the 100 K data in space group Pnma yields two different orientations of ND₄⁺ tetrahedra with almost equal partial occupation of 56.13(16) % and 43.87(16)%, respectively. All refined structure models contain isotropic atomic displacement parameters (ADP) for the Fe atoms and anisotropic ADPs for all other atoms. Nearly opposing D atoms of the two possible orientations are constrained to have the same ADPs and a total occupancy of 1. Further details of the refinements are given in the Appendix VA. This disordered structure is illustrated in Figs. 1 and 6, and the corresponding structural parameters together with the R values are given in Table I.

The low-temperature neutron diffraction data unambiguously show that the structural phase transition in $(ND_4)_2[FeCl_5(D_2O)]$ at T_S results from orientational order of the ND_4^+ tetrahedra, see the Appendix, in accordance with Ref. [27]. The ordering results in broken n

100 K, $Pnma$ $a = 13.579, b = 9.922, c = 6.947 \text{ Å}$									
	R(obs) =	= 4.86 %	wR(obs) =	= 5.95%					
	R(all) =	= 7.97 %	wR(all) =	6.09%					
	x	y	z	$U_{\rm iso}$					
Fe1	0.11807(8)	1/4	0.1900(2)	7.1(3)					
Cl1	0.10580(6)	0.00997(8)	0.1763(1)	10.7(3)					
Cl2	0.00679(9)	1/4	0.4576(2)	11.4(4)					
Cl3	0.25069(9)	1/4	0.4004(2)	9.5(4)					
Cl4	0.22553(9)	1/4	-0.0756(2)	11.2(4)					
O1	-0.0028(2)	1/4	0.0000(4)	14.9(6)					
D1	-0.0342(1)	0.1712(1)	-0.0502(3)	25.2(5)					
N1	0.13865(6)	0.00028(9)	0.6653(2)	13.5(3)					
D2	0.1356(2)	0.0079(4)	0.8133(5)	46(1)					
D3	0.1067(3)	-0.0817(3)	0.6225(6)	44(1)					
D4	0.2120(2)	0.0040(4)	0.6352(5)	40(1)					
D5	0.1059(2)	0.0835(3)	0.6115(6)	40(1)					
10	$K, P2_1/a$		b = 9.960, c =	= 6.925 Å					
		$\gamma = 90.2^{\circ}$							
	· · · ·	= 3.38 %	wR(obs) =						
	R(all) =	= 5.39 %	wR(all) =						
	x	y	z	$U_{\rm iso}$					
Fe1	0.11923(6)	0.2499(1)	0.1875(1)	2.8(2)					
$Cl1_1$	0.10315(7)	0.0103(1)	0.1825(1)	4.7(3)					
$Cl1_2$	0.10981(7)	0.4897(1)	0.1667(1)	4.5(2)					
Cl2	0.00651(6)	0.2551(1)	0.4567(1)	4.6(2)					
Cl3	0.25158(6)	0.2489(1)	0.4001(1)	4.6(2)					
Cl4	0.22700(6)	0.2451(1)	-0.0768(1)	4.9(2)					
O1	-0.0017(1)	0.2492(2)	-0.0033(2)	7.1(3)					
$D1_1$	-0.0346(1)	0.1705(2)	-0.0530(3)	20.1(6)					
$D1_2$	-0.0353(1)	0.3280(2)	-0.0500(3)	22.0(6)					
$N1_1$	0.13640(7)	-0.0028(1)	0.6643(2)	7.1(3)					
$N1_2$	0.13987(7)	0.4949(1)	0.6700(2)	7.4(3)					
$D2_2$	0.1315(1)	0.4918(2)	0.8162(3)	22.5(5)					
$D3_2$	0.1110(2)	0.5825(2)	0.6176(3)	27.2(6)					
$D4_2$	0.2132(1)	0.4888(2)	0.6400(3)	26.1(6)					
$D5_2$	0.1031(1)	0.4161(2)	0.6086(3)	22.5(6)					
$D6_1$	0.1407(1)	-0.0077(3)	0.5172(3)	26.9(6)					
$D7_1$	0.1702(2)	0.0829(2)	0.7138(3)	25.6(6)					
$D8_1$	0.0637(1)	-0.0019(2)	0.7017(3)	21.1(5)					
D9_1	0.1702(1)	-0.0852(2)	0.7227(3)	24.1(6)					

TABLE I. $(ND_4)_2$ [FeCl₅(D₂O)]: Results of the structural refinements with the neutron-diffraction data sets obtained at 100 K and 10 K in space groups *Pnma* and *P*2₁/*a*, respectively. Isotropic atomic displacements parameters are given in 10⁻³Å². Lattice parameters were taken from reference 15 (besides γ deduced from the splitting of the (0 4 0) reflection) and anisotropic atomic displacement parameters are given in Table II in the Appendix.

and m symmetries resulting in space group $P2_1/a$ (with the non standard setting). We did not find any evidence for an additional breaking of the inversion symmetry. These ordered tetrahedra result in a harder libration mode that becomes visible in THz spectroscopy.

Although a satisfying description of the Bragg reflection intensities is obtained within the centrosymmetric but monoclinic structure, this solution does not rule out the appearance of a small but finite electric polarization in the structurally distorted phase for the partially K-substituted samples. Since the lattice becomes monoclinic in the distorted structure, there are only two equivalent space diagonals, along which the finite polarization could appear, but the measured polarization, see Fig. 2b) does not follow this explanation, pointing to a more exotic mechanism. The experimentally determined polarization vector is only allowed in a triclinic structure, which is not supported by diffraction data on the substituted compounds. An in-grown potassium concentration gradient could cause a built-in electric field and a corresponding polar structure, as it is proposed for ferroelectric $KTiOPO_4$ [44, 45]. However, pure $(NH_4)_2$ [FeCl₅(H₂O)] is paraelectric, because the dipole moments of the NH_4^+ groups cancel each other so that the total structure is not polar. However, this balance can be perturbed by the partial substitution with nonpolar K^+ ions, in particular, if they stay mobile around the structural phase transition. The inversion symmetry in Pnma can be broken by partial ordering of the K substitution, but this loss of inversion symmetry will remain hidden through the disorder of the NH_4^+ groups in the high-temperature phase. However, the ordering of these NH_4^+ groups group can imply strong pyroelectric effects at the structural phase transition due to a finite dipole moment in the low-symmetry environment of the NH_{4}^{+} tetrahedron, that is missing at a K site.

B. Low-temperature transitions

In Fig. 7, we present the magnetic susceptibility of the substitution series $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)].$ As discussed in Ref. 15, the susceptibility of $(NH_4)_2$ [FeCl₅(H₂O)] shows a high-temperature Curie-Weiss behavior and an *ac* easy-plane anisotropy below $T_{\rm N}$ indicating a spin-spiral order. For x = 0.06, we find an analogous behavior of $\chi(T)$, in particular the evolution of an easy-plane anisotropy indicating that this compound hosts a cycloidal spin texture as well. For potassium concentrations of 0.15 (panel c) and 0.45 (panel d), both χ_b and χ_c remain constant below $T_{\rm N}$ of 7.9 K and 10.8 K, respectively. In contrast, χ_a drops towards zero, which indicates collinear antiferromagnetic order with moments along a as it is also found in $K_2[FeCl_5(H_2O)]$ [16]. As already known from the specific-heat data (Fig. 2), the Néel temperature increases with increasing x and the transition only splits for x = 0 and 0.06. For all substitution levels, the hightemperature susceptibility follows a Curie-Weiss law with nearly the same effective magnetic moment of approximately 5.95 $\mu_{\rm B}$ (see insets), which well agrees with the expected value of 5.92 $\mu_{\rm B}$ for S = 5/2. Additionally, the absolute value of the Curie-Weiss temperature increases slightly with increasing potassium content, from $\theta = -34.9 \,\mathrm{K}$ to $-37.6 \,\mathrm{K}$. The fact that $|\theta| > T_{\mathrm{N}}$ indicates magnetic frustration in all systems. Because $|\theta|$ increases, the net antiferromagnetic exchange becomes stronger with increasing x, but as T_N increases even

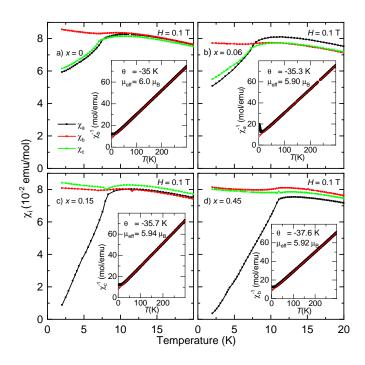


FIG. 7. Susceptibility of the substitution series $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ with a) x = 0, b) x = 0.06, c) x = 0.15, d) x = 0.45. The insets show the Curie-Weiss fits.

faster, the frustration parameter $f = |\theta|/T_{\rm N}$ is reduced continuously from $f \approx 4.8$ in $(\rm NH_4)_2[\rm FeCl_5(H_2O)]$ to 2.8 in K₂[FeCl₅(H₂O)]. This suggests that the change of the magnetic order from cycloidal to collinear is directly linked to the reduced magnetic frustration.

Figure 8 displays the temperature-dependent polarization of the substitution series. Again, we include the data for $(NH_4)_2$ [FeCl₅(H₂O)] from Ref. 15, which show a typical type-II multiferroic behavior with a rather small polarization. In low magnetic fields, the polarization is almost constant and mainly points along \boldsymbol{a} , but switches towards the c direction due to a spin-flop transition occurring around 5 T [15]. Panel b) shows the analogous data for the smallest potassium content, which also exhibits a spontaneous polarization at zero magnetic field that hardly varies in small fields. As discussed above, $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ already shows a finite polarization below the structural transition. Therefore, the polarization change occurring at low-temperature is poled by this internal polarization which causes an internal electric field $E_{\rm int} = \frac{2}{3\epsilon_0}P \approx 750 \,\mathrm{V \, mm^{-1}}$. Consequently, external poling fields of $200 \,\mathrm{V \, mm^{-1}}$ cannot switch the direction of the additional polarization. Thus, we call this phase pseudo-ferroelectric and pseudomultiferroic. In larger magnetic fields, the additional polarization along a again almost fully vanishes, because it switches towards the c direction, in analogy to the pure $(NH_4)_2$ [FeCl₅(H₂O)] (see below).

For higher substitution levels, x = 0.15 and 0.45, the magnetoelectric behavior qualitatively changes. As

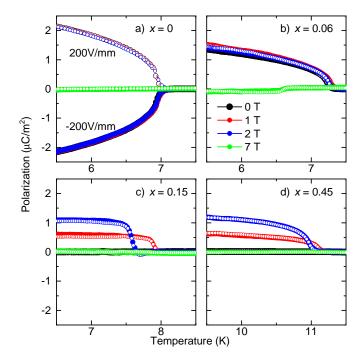


FIG. 8. Temperature dependence of the polarization P_a for the substitution series $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ with a) x = 0, b) x = 0.06, c) x = 0.15, d) x = 0.45. Upon cooling, electric poling fields of $\pm 200 \,\mathrm{V \,mm^{-1}}$ (open and closed symbols, respectively) were applied and removed at base temperature. The polarization was then obtained by numerical integration of the pyrocurrent measured upon heating.

shown in panels c) and d), there no longer is an additional spontaneous polarization in zero magnetic field, but small applied magnetic fields induce an extra electric polarization, whose magnitude first linearly increases with magnetic field and then abruptly vanishes because again the polarization switches towards the *c* direction (see below). This kind of linear magnetoelectric behavior is very similar to that of the pure $K_2[FeCl_5(H_2O)]$ and other alkaline-metal-based erythrosiderites [16]. One main difference is, however, that again the polarization cannot be switched by electric poling fields of 200 V mm⁻¹, because of the polarization that appears already at T_S (see Fig. 2b).

We also determined the magnetic-field dependence of the polarization directly by integrating the measured magnetic-field dependent (magneto-)current. In Fig. 9, we present both polarization components P_a and P_c as a function of $\boldsymbol{H} \parallel \boldsymbol{a}$. For x = 0, panel a) shows an approximately constant polarization along \boldsymbol{a} up to $\mu_0 H \simeq 5 \text{ T}$, where it switches to \boldsymbol{c} and then linearly increases with further increasing magnetic field, i.e., $P_c = \alpha_{ca}\mu_0 H_a$ [15]. Figure 9b) displays the polarization components P_a and P_c for x = 0.06, which behave similarly to the pure compound. However, there are also clear differences. First of all, the switching field is reduced to 3.5 T for x = 0.06and also the linear magnetoelectric coefficient of P_c is reduced to $\alpha_{ca} = \partial P_c / \partial (\mu_0 H_a) = 0.53 \text{ ps} \text{ m}^{-1}$. As discussed above, for x = 0.06 the sign of the polarization is

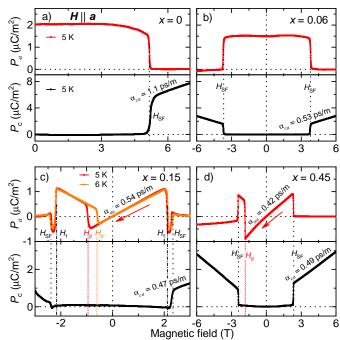


FIG. 9. Magnetic-field dependence of the polarization components P_a (red) and P_c (black) for $\boldsymbol{H} \| \boldsymbol{a}$ for the substitution series [(NH₄)_{1-x}K_x]₂[FeCl₅(H₂O)] with a) x = 0, b) x = 0.06, c) x = 0.15, d) x = 0.45. All polarization curves were obtained by numerical integration of the magnetocurrent measured as a function of increasing magnetic field. For x = 0, electric poling fields of 200 V mm⁻¹ were applied. $H_{\rm SF}$ denotes the spin-flop field, which is accompanied by a polarization switch from the \boldsymbol{a} to the \boldsymbol{c} direction. For x = 0.15, an intermediate phase with inverted polarization P_a occurs in the field range $H_1 < H < H_{\rm SF}$. Switching of magnetoelectric domains at H_d is observed for x = 0.15 and x = 0.45. The arrows indicate the field-sweep direction because this domain switching is hysteretic, see Fig. 12 in the Appendix V B. The linear magnetoelectric coefficients $\alpha_{ij} = \partial P_i / \partial (\mu_0 H_j)$ are also given.

not switchable by applying an electric field, in contrast to the pure x = 0 compound. Above $H_{\rm SF}$, the polarization P_c is of the same sign for both magnetic-field directions and cannot be inverted by electric fields up to $200 \,\mathrm{V \, mm^{-1}}$, again in contrast to the x = 0 compound [15]. The reason is that the preexisting internal electric field for x = 0.06 fixes the direction of P_c , which implies that opposite linear magnetoelectric domains are realized for positive and negative magnetic field.

For larger x, see panels c) and d), the dielectric properties change drastically, because these materials are no longer multiferroic or pseudo-multiferroic. We start the discussion with the x = 0.45 sample. Above $H_{\rm SF}$, this material is again linear magnetoelectric with the sign of P_c remaining unchanged for both magnetic-field directions, in analogy to the x = 0.06 sample. Below $H_{\rm SF}$, the x = 0.45 sample also shows linear magnetoelectric coupling $P_a = \alpha_{aa}\mu_0 H_a$. Remarkably, there is a polarization sign reversal at $H_d = -1.6$ T on decreasing magnetic field for 5 K, and below this flop the slope $\partial P_a/\partial(\mu_0 H)$ is inverted as well. In the linear magnetoelectric phase, there are in general two possible domains, which are usually switchable by applying an external electric field (in addition to H). In the present case, however, the hightemperature polarization, see Fig. 2b), provides an intrinsic electric field that poles the magnetoelectric domains. Due to this combination, it is possible to switch between linear magnetoelectric domains with a magnetic field even without any external electric field. This domain switching shows a symmetric hysteresis behavior with respect to the field-sweep direction, see Fig. 12 in Appendix V B.

For x = 0.15, see panel c), the material is still linear magnetoelectric as we find a linear polarization P_a for small magnetic fields. The domain switching field H_d decreases with increasing temperature, indicating activated behavior. Between $|H_1| = 2.2 \text{ T}$ and $|H_{\text{SF}}| =$ 2.4 T, an intermediate phase occurs with a negative and essentially constant polarization. Above H_{SF} , the polarization switches to the *c* direction, as for x = 0.45, and grows linearly in magnetic fields. The intermediate regime might be a precursor of the ferroelectric intermediate phase known for $(\text{NH}_4)_2[\text{FeCl}_5(\text{H}_2\text{O})]$.

To summarize these findings, the spin-flop phases above $H_{\rm SF}$ are linear magnetoelectric for all samples. Interestingly, all mixed samples even quantitatively show very similar magnetoelectric coefficients below and above the spin-flop transition with $\alpha_{ca} \approx \alpha_{aa} \approx 0.5 \,\mathrm{ps}\,\mathrm{m}^{-1}$, whereas the x = 0 parent compound has a larger $\alpha_{ca} =$ $1.1 \,\mathrm{ps}\,\mathrm{m}^{-1}$. Concerning the switching of magnetoelectric domains, it is not possible to switch the polarization above $H_{\rm SF}$ by an external electric field of up to $200 \,\mathrm{V \, mm^{-1}}$. This means that opposite antiferromagnetic domains are realized for larger positive and negative magnetic fields, i.e., for $\pm H > |H_{\rm SF}|$, in order to align the induced magnetoelectric polarization with P_c that is present already above $T_{\rm N}$. In particular, these are single-domain states even in zero electric field, whereas usual linear magnetoelectrics with spin-flop transitions, e.g., $(NH_4)_2$ [FeCl₅(H₂O)] [15] require the application of both a magnetic and an electric field to reach a singledomain state above $H_{\rm SF}$.

Finally, we present the resulting temperature vs. magnetic-field phase diagrams with $H \parallel a$ for all investigated substitution levels in Fig. 10. Pure $(NH_4)_2$ [FeCl₅(H₂O)] is a type-II multiferroic, which has been characterized in detail [15, 27, 29, 31]. For x =0.06 we still find a two-step magnetic ordering transition $(T_{\rm N} = 7.25 \,\mathrm{K}, T_{\rm C} = 7 \,\mathrm{K})$, which is typical for multiferroics with a cycloidal spin texture. Similar to the pure ammonium-based compound, there is an intermediate-field phase around 3 T that vanishes above 4K. The low-field regime shows a pseudo-multiferroic non-zero spontaneous polarization P_a . The term pseudomultiferroic is used because the polarization cannot be switched by applying electrical poling fields (of the order of $200 \,\mathrm{V \, mm^{-1}}$) because the structure is already polar for $T > T_{\rm C}$.

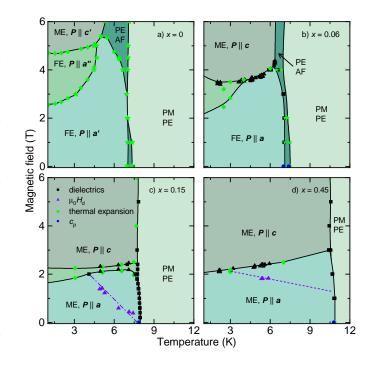


FIG. 10. Phase diagrams of the substitution series $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ with a) x = 0, b) x = 0.06, c) x = 0.15, d) x = 0.45 for $H \parallel a$. Phase boundaries are based on anomalies in thermal expansion and magnetostriction (not discussed here), pyrocurrent and magnetocurrent (dielectrics), and specific heat measurements. The points of $\mu_0 H_d$ stem from the magnetic-field dependent anti-phase domain switching discussed for Fig. 9. The phases labeled ME show the linear magnetoelectric effect, FE phases are (pseudo-)ferroelectric, PM = paramagnetic, AF = antiferromagnetic, PE = paraelectric.

The exact direction of the polarization in the intermediate phase around 3T was not part of this study. but one may suspect that there might be a slight reorientation of the polarization as it is the case in the pure $(NH_4)_2$ [FeCl₅(H₂O)] due to a reorientation of the magnetic moments, see Refs. 15 and 29. The potassium content of x = 0.15 suppresses the spontaneous polarization and changes the cycloidal spin structure towards a collinear order with a single antiferromagnetic transition at 7.9 K. The material is linear magnetoelectric with the polarization pointing along a with $H \parallel a$. When performing magnetic-field hysteresis loops, see Appendix VB, it is possible to switch between the two different linear-magnetoelectric domains. The switching is thermally activated and the switching field strongly increases with decreasing temperature. Below 4K, switching is no longer possible because there is a transition to an intermediate phase existing in the field range from 2.2 T to 2.4 T. This intermediate phase has an inverted polarization that seems to be field-independent, although the limited field range prevents an unambiguous conclusion. If the polarization does not follow the linear magnetoelectric effect, this phase could be a distorted

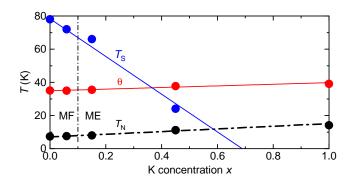


FIG. 11. Potassium concentration dependence of the transition temperatures $T_{\rm N}$, $T_{\rm S}$, and the magnetic Curie-Weiss temperature θ for [(NH₄)_{1-x}K_x]₂[FeCl₅(H₂O)].

cycloid as in $(NH_4)_2$ [FeCl₅(H₂O)] and thus be a pseudomultiferroic phase. Interestingly, this phase is stable up to the ordering temperature, which differs from the intermediate phase in the materials with less potassium content. The stability of this intermediate phase indicates that the potassium concentration of 0.15 is close to the border between a multiferroic and a magnetoelectric ground state. For x = 0.45, see panel d), T_N increases to 10.8 K. As a function of increasing field there is only one spin-flop transition at H_{SF} . For both, x = 0.15 and 0.45, it is possible to switch the magnetoelectric domains for $|H| < |H_{SF}|$ with a magnetic field alone. Moreover, for all x > 0, opposite single-domain states are realized in the respective spin-flop phases above positive and negative magnetic fields H_{SF} .

Figure 11 compares the substitution dependence of the Néel temperature (T_N) , the Curie-Weiss temperature (θ) , and the structural transition temperature (T_S) of $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$. By changing the potassium concentration x we can tune the magnetic ordering temperature T_N and the Curie-Weiss temperature, which both increase linearly with x, while the structural phase transition temperature T_S decreases linearly.

IV. CONCLUSION

Pyrocurrent measurements across the structural phase transition occurring in $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ well above the magnetic order reveal the appearance of electric polarization for finite K content while no such signal is observed in the pure material. Using a deuterated crystal, neutron diffraction experiments characterize this phase transition as an order-disorder transition of the ND₄⁺ groups. In the high-temperature phase the ND₄⁺ tetrahedra are disordered, which we model by the simultaneous occupation of a ND₄⁺ tetrahedron orientation and its inversion with respect to the central N position. Below the structural phase transition only one of these orientations is occupied at each N site. This ordering breaks a glide plane resulting in the occurrence of special extinction rule violating reflections and the lattice becomes monoclinically distorted while the translation symmetry does not change. Furthermore, a libration mode of the ordered NH_4^+ is observed by THz spectroscopy. The low-temperature structure is well described in a centrosymmetric space group, and further loss of this inversion symmetry or the modeling in a noncentrosymmetric isotropy subgroup does not improve the description of the diffraction data. Considering the absence of electric polarization appearing at this transition in pure $(NH_4)_2$ [FeCl₅(H₂O)] we can conclude that the low-temperature symmetry of the pure material remains non-polar. The continuous emergence of polarization with increasing K content can be understood if the K insertion is not fully random but breaks inversion symmetry. In this case the ordering of the NH_4^+ groups carrying a dipole moment can cause additional electric polarization; the materials with finite K content are thus pyroelectric.

The magnetic and dielectric properties of the substitution series $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ systematically vary with x. The potassium concentration affects the magnetic exchange reducing the frustration (from f = 4.8 for x = 0 to 2.8 for x = 1). Furthermore, the magnetic order transforms from a cycloidal ground state ($x \leq 0.06$) towards a collinear antiferromagnet ($x \geq 0.15$) and, simultaneously, multiferroicity and spontaneous polarization are suppressed. The materials at higher K content exhibit linear magnetoelectric behavior. Due to the preexisting pyroelectric polarization it is possible to drive the linear-magnetoelectric antiferromagnetic domains without applying an electric field. Here, an external magnetic field is sufficient to switch between otherwise energetically degenerate states.

ACKNOWLEDGMENTS

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V. APPENDIX

A. Structure refinement of $(ND_4)_2[FeCl_5(D_2O)]$

In the refinements of the crystal structure with the data taken at 100 K, the D atoms within one tetrahedron are constrained to have the same occupancy. The refinement of the model with two independent tetrahedra describes the data well with satisfactory reliability parameters, given in Table I. In order to further simplify the model and to reduce the number of free parameters, we constrain the second D_4 tetrahedron to arise from the first one just by inversion with respect to the central N atom, which results in an only slightly worsened description, (wR(all) = 6.09%).

Starting from the high-temperature Pnma phase, we first consider possible symmetry reductions that result in a centrosymmetric structure and that do not break translation symmetry. The modes Γ_2^+ , Γ_3^+ , and Γ_4^+ lead to space groups $P2_1/a$, $P2_1/n$, and $P2_1/m$, respectively [43, 46]. They can be distinguished by their selection rules. We can rule out space group $P2_1/n$ because we observed reflections of the form (0 k 0) with odd k and (0 k l) with odd k + l, see Fig. 5c) and d). Space group $P2_1/a$ (we use a non-conventional setting in order to allow for direct comparison with the high-temperature *Pnma* structure) is in accordance with all observed, respectively not observed reflections. Space group $P2_1/m$ allows reflections of the form (h k 0) with odd h, which are not observed, but this does not rule out this space group, because the reflections might be below the detection level. Structural refinements were thus carried out in space groups Pnma, $P2_1/a$, and $P2_1/m$ using the software JANA2006 [47]. The R values [48] unambiguously show that the neutron diffraction data at 10 K are best described in space group $P2_1/a$. Among the symmetry reductions resulting in a non-centrosymmetric low-temperature structure [43] only space group $Pmc2_1$ is in accordance with the observed superstructure reflections, but a satisfactory description of the data cannot be achieved in this symmetry. A non-centrosymmetric structure can also be obtained by further reducing the symmetry from $P2_1/a$ to Pa, but the refinements of this model give no evidence for such a distortion. We thus conclude that the low-temperature symmetry corresponds to $P2_1/a$ in agreement with the absence of a spontaneous electric polarization at the structural transition in the pure material. This monoclinic symmetry $P2_1/a$ is in agreement with the occurrence of ferroelastic domains of two different orientation states at $T_{\rm S}$ and with the orientation of the optical indicatrix within these domains, as it was found in Ref. 49.

The refinement of the structure model in space group $P2_1/a$ with the 10 K data yields a structure, in which each of the 8 tetrahedra within the unit cell strongly prefers one of the two possible orientations (95.3(3)%)resp. 95.5(2)% indicating an almost perfect orientational ordering. Due to the low occupation of the minority tetrahedra their structural parameters cannot be reliably refined. We therefore simplified the structure model by assuming perfect tetrahedron ordering which again only slightly reduces the reliability values (wR(all) =4.41%). As already discussed with the (040) peak profile, the crystal shows twinning of monoclinic domains, which was considered during the refinements finding a domain ratio of 69 to 31 percent. This observation well agrees with the scan profiles shown in Fig. 5). The resulting structure is shown in Fig. 6 and structural parameters together with the R values are given in Ta-

$100\mathrm{K}$, Pnma					
	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Cl1	12.7(4)	5.0(4)	14.5(6)	-0.6(3)	-2.6(4)	-0.5(3)
Cl2	10.9(6)	9.7(6)	13.6(8)	0	3.3(5)	0
Cl3	9.3(5)	10.6(6)	8.6(8)	0	-0.9(5)	0
Cl4	12.1(6)	12.4(6)	9.2(9)	0	1.3(5)	0
O1	14.1(9)	9.8(9)	21(1)	0	-9.5(9)	0
D1	27.4(7)	17.9(7)	30(1)	-4.6(6)	-8.4(7)	-3.7(7)
N1	14.0(5)	13.2(5)	13.2(7)	0.8(4)	-0.5(4)	2.1(5)
D2	55(2)	67(2)	16(2)	-4(2)	1(1)	-2(2)
D3	50(2)	22(1)	60(3)	-13(1)	-12(2)	-6(1)
D4	23(1)	55(2)	43(2)	0(1)	4(1)	-4(2)
D5	36(1)	23(1)	61(3)	7(1)	-7(1)	11(1)
_10 K,	$P2_1/a$					
	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Cl1_1	5.1(4)	1.9(4)	7.0(5)	-0.8(4)	-0.7(3)	0.2(4)
$Cl1_2$	6.2(4)	2.5(4)	4.7(5)	0.0(4)	-0.2(3)	
Cl2	4.4(3)	4.3(4)	5.2(4)	-0.7(4)	()	-1.0(4)
Cl3	4.9(3)	4.5(4)	4.5(4)	-0.7(4)		0.4(5)
Cl4	5.7(4)	3.9(4)	5.0(5)	0.3(4)	0.8(3)	-0.2(4)
O1	6.6(4)	5.7(5)	9.2(6)	-1.2(8)	-3.6(5)	2(1)
$D1_1$	24(1)	11.9(9)	24(1)	-5.4(7)	-6.7(8)	-4.1(8)
$D1_2$	22(1)	14(1)	30(1)	3.2(7)	-8.5(9)	2.3(8)
$N1_1$	8.2(4)	5.9(5)	7.1(6)	-1.4(4)	-0.3(4)	-1.1(5)
$N1_2$	8.1(4)	6.2(5)	8.1(6)	-0.7(4)	-0.4(4)	-0.6(5)
$D2_2$	27.8(9)	28(1)	12.0(9)	-0.7(9)		-0.5(8)
$D3_2$	36(1)	16.4(9)	29(1)	7.3(8)	-6.6(9)	3.6(8)
$D4_2$	13.0(7)	34(1)	32(1)	1.9(8)	4.4(7)	-2(1)
$D5_2$	25(1)	16.4(8)	26(1)	-6.0(8)		-5.5(8)
$D6_1$	34.5(9)	33(1)	13(1)	-1(1)	1.2(7)	-0.5(9)
$D7_1$	26(1)	14.9(9)	36(1)	-7.3(8)	-3.9(8)	-6.1(9)
$D8_1$	12.5(7)	26.2(9)	25(1)	-0.9(8)	3.3(6)	-2.7(9)
D9_1	25(1)	17.0(9)	31(1)	6.8(8)	-4.0(8)	6.2(8)

TABLE II. Anisotropic displacement parameters of the two structural refinements with the data taken at 100 K and 10 K on $(ND_4)_2$ [FeCl₅(D₂O)]; all values are given in 10^{-3} Å².

ble I. There are two independent distorted ND_4^+ groups in space group $P2_1/a$ that are related through the *n* glide plane in *Pnma*; the centrosymmetry, however, persists so that for each of these independent ND_4^+ groups there exists an inverted one.

B. Switching of magnetoelectric domains

Figure 12 shows the magnetic-field dependence of the polarization component P_a obtained during a full hysteresis loop on increasing and decreasing magnetic field for the potassium concentrations of x = 0.15 and x = 0.45. In both materials, the magnetoelectric polarization is poled to be positive when entering the magnetoelectric phase either at H_1 or at $H_{\rm SF}$ for x = 0.15 or x = 0.45, respectively, due to the internal field that stems from the polarization arising below the structural phase transition. As expected for a linear magnetoelectric, the magnetoelectric polarization changes sign when the magnetic field inverts at H = 0. However, this negative polarization is

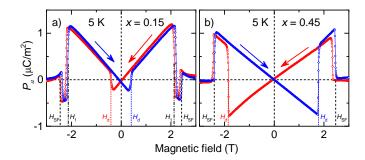


FIG. 12. Hysteresis loops of the magnetic-field dependent polarization component P_a for H || a of $[(NH_4)_{1-x}K_x]_2[FeCl_5(H_2O)]$ with a) x = 0.15 and b) x = 0.45. All polarization curves were obtained by numerical integration of the magnetocurrent measured as a function of increasing (blue) and decreasing (red) magnetic field.

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