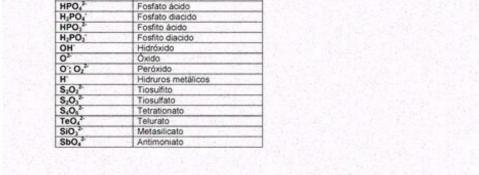


## Tabla de aniones y cationes completa

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ANIONES	TABLA DE ANIONES	CATIONES	NOMBRE
F	Fluoruro	Cr <sup>8+</sup>	Cromo VI (Crómico)
CI	Cloruro	Cr3.	Cromo III (Cromoso)
Br	Bromuro	Fe2	Fierro II (Ferroso)
r	Yoduro	Fe <sup>3+</sup>	Fierro III (Férrico)
S <sup>2.</sup>	Sulfuro	Co2+	Cobalto II (Cobaltoso)
Se <sup>2</sup>	Selenuro	Co3+	Cobalto III (Cobáltico)
N <sup>3</sup>	Nitruro	Ni <sup>2+</sup>	Niguel II (Nigueloso)
p3-	Fosfuro	Ni <sup>3+</sup>	Niquel III (Niquelico)
C4 ; C,2	Carburo	Cu	Cobre I (Cuproso)
SIt	Siliciuro	Cu2*	Cobre II (Cúprico)
NO;	Nitrato	Hg224	Mercurio I (Mercuroso
NO <sub>1</sub>	Nitrito	Hg <sup>2</sup>	Mercurio II (Mercúrico
PO,3	Fosfito	Sn2*	Estaño II (Estanoso)
PO,3	Fosfato	Sn4*	Estaño IV (Estánico)
BO3 <sup>2</sup>	Borato	Pb2*	Plomo II (Plumboso)
CO32.	Carbonato	Pb4*	Plomo IV (Plúmbico)
AsO3	Arsenito	Au	Oro I (Auroso)
AsO4	Arseniato	Au <sup>2+</sup>	Oro III (Aurico)
SO32	Sulfito	Zn2*	Cinc II (cincico)
SO42	Sulfato	NH4	Amonio
CIO.	Perciorato	Ag	Plata I (Argentico)
CIO	Clorato	H,	Ácido
CIO2	Clorito	100 - C - C - C - C - C - C - C - C - C -	
CIO.	Hipoclorito	1	Constant Constant of the
MnO42	Manganato	11.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.2.	
MnO4	Permanganato	100 UP82	a start and the start of the
CrO42	Cromato		
Cr <sub>2</sub> O <sub>7</sub> <sup>2</sup>	Dicromato	the first of the second	
CN'	Cianuro	122 122	Personal and a second second
SCN'	Tiocianato o sulfocianuro	-	
C2O42	Oxalato	14-12-12-12-12-12-12-12-12-12-12-12-12-12-	
CH3COO.	Acetato		
[Fe(CN),] <sup>4</sup>	Ferrocianuro		
[Fe(CN)]3-	Ferricianuro		
HCO3	Carbonato acido o bicarbonato		
HSO;	Sulfito acido o bisulfito		
HSO4	Sulfato acido o bisulfato	10-10-10-10-10-10-10-10-10-10-10-10-10-1	
HS	Sulfuro acido o bisulfuro		
HPO.2	Ecolato ócido		



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With elements, two electrons gain anions with two charges. 1 2 3 4 5 Ammonium NH4 Barium BA ALLUMINUM AL Carbon with Surmin SB Hydrogen H Strontium Sr.B. Thinking system We can know which elements form cationic anionsb'The "USER" is required to grant the corresponding credits to the "UNAM" as the generating body and, when possible, to mention the body or university agency that contributed to the generation of the information, as well as any third parties who holds the intellectual property rights on any content hosted on the site, specifying the date of recent update.\xc3\xa1s according to the following model: UNAM MODEL: UNAM, "Name of the resource", [Name/Acronym of the university body or agency who contributed to the generation\xc3\xb3of the information]; INTERNET LINK OF THE DATA CONSULTED, and the update date in numerical format [YYYY-MM-DD], made available so that they are easily accessible to users. THIRD PARTY COPYRIGHT TEMPLATE (NAME OF INDIVIDUAL OR LEGAL PERSON, COPYRIGHT OR COPYRIGHT HOLDER), "Asset Name"; INTERNET LINK OF THE DATA CONSULTED, and the update date in numerical format [YYYY-MM-DD], made available so that they are easily accessible to users. Periodic table of anions and cations.

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Aluminum, a member of the IIIA family, loses three electrons to form cation 3. With these two elements, when combined, lithium transfers electrons to the final level, loses electrons., and chlorine gains negatively charged electrons to complete the final level Be 2 Beryllium Mg2 Magnesium Hg2 Mercury II FE 2 Iron II CA 2 May 27, 2019 - List of cations and anions on the topic of chemistry. National Autonomous University of Mexico.