

ELECTRORHEOLOGY OF SUSPENSIONS OF VARIOUSLY PROTONATED POLYANILINE PARTICLES UNDER STEADY AND OSCILLATORY SHEAR

MARTIN STENICKA^{1*}, VLADIMIR PAVLINEK¹, PETR SÁHA¹, NATALIE V. BLINOVA²,
JAROSLAV STEJSKAL², OTAKAR QUADRAT²

¹ Polymer Centre, Faculty of Technology, Tomas Bata University in Zlin, TGM 275, 762 72 Zlin, Czech Republic

² Institute of Macromolecular Chemistry, Academy of Sciences of the Czech Republic, Heyrovsky Square 2, 162 06 Prague 6, Czech Republic

* Email: stenicka@ft.utb.cz

Fax: x420.576.031444

Received: 11.11.2009, Final version: 4.1.2010

ABSTRACT:

Electrorheological (ER) and dielectric properties of silicone-oil suspensions of polyaniline (PANI) particles protonated with phosphoric and tetrafluoroboric acids to various doping level have been investigated. The particle conductivity was thus varied between the order of 10^{-9} S/cm and 10^{-4} S/cm. The dynamic yield stresses obtained at controlled shear rate mode viscometry, the storage moduli from the oscillatory shear experiments and the dielectric relaxation times from frequency dependences of dielectric constant and loss factor were used as criteria of rigidity or elasticity of ER structures and particle mobility in the electric field. The conductivity of suspension particles plays a decisive role in their ER behaviour. The ER efficiency increased as conductivity of dispersed particles raised, irrespective of the type of employed acid used for the protonation of PANI.

ZUSAMMENFASSUNG:

Elektorrheologische (ER) und dielektrische Eigenschaften von Suspensionen aus Silikonöl und Poly(anilin)-Partikeln, die mit Phosphor- und Tetrafluoroborsäure in unterschiedlicher Stärke protoniert wurden, wurden untersucht. Die Leitfähigkeit der Partikel wurde in der Größenordnung von 10^{-9} S/cm und 10^{-4} S/cm variiert. Die dynamische Fließspannung, die durch Messungen in dem Modus der kontrollierten Scherrate erhalten wurde, der Speichermodul aus oszillatorischen Scherversuchen und die dielektrischen Relaxationszeiten aus der Frequenzabhängigkeit der dielektrischen Konstanten und des Verlustfaktors wurden als Kriterium der Starrheit oder der Elastizität der ER-Strukturen und der Partikelbeweglichkeit in dem elektrischen Feld verwendet. Die Leitfähigkeit der Suspensionspartikel spielt eine entscheidende Rolle in ihrem ER-Verhalten. Die ER-Wirksamkeit nahm mit der Leitfähigkeit der dispergierten Partikel zu, unabhängig von der zur Protonierung von PANI verwendeten Säure.

RÉSUMÉ:

Nous avons étudié les propriétés électrorhéologiques (ER) et diélectriques de suspensions de particules de polyaniline (PANI) protonées avec des acides phosphoriques et tétrafluoroboriques à divers degrés de dopage. La conductivité de la particule a donc ainsi été variée entre 10^{-9} S/cm et 10^{-4} S/cm. Les contraintes dynamiques de seuil obtenues en mode de vitesse de cisaillement contrôlée, les modules élastiques obtenus à partir des expériences de cisaillement oscillatoire et les temps de relaxation diélectrique obtenus à partir de la dépendance fréquentielle de la constante diélectrique et du facteur de perte, ont été utilisés comme critères de rigidité ou d'élasticité des structures ER et de mobilité de la particule dans le champ électrique. La conductivité des particules en suspension joue un rôle décisif dans leur comportement ER. L'efficacité ER augmente avec la conductivité accrue des particules dispersées, sans relation avec le type d'acide utilisé pour la protonation des particules PANI.

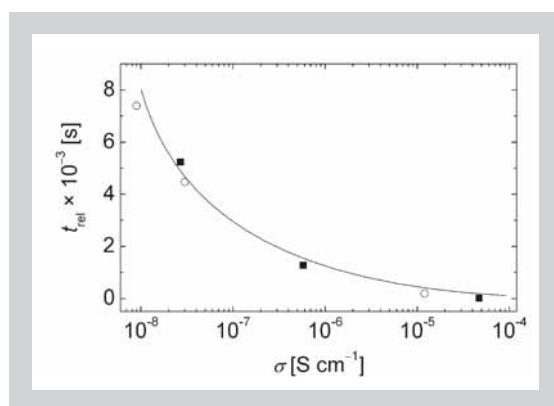
KEY WORDS: polyaniline, electrorheology, steady shear, oscillatory mode, protonation

1 INTRODUCTION

Electrorheological (ER) effect [1], known as an outstanding change in rheological behaviour of ER suspension after the application of external electric field has been attracting strong attention since its discovery 60 years ago. The particles randomly dispersed in the non-conducting carrier

medium are polarized in the presence of the electric field, resulting in particle fibrillation with string-like or columnar structures oriented along the field direction. In the electric field of intensity of several kilovolts per millimetre, this abrupt change sets in order of milliseconds. Thus, a dramatic increase in viscosity of several orders of

Figure 10:
The dependence of relaxation time, t_{rel} on the conductivity, σ of the PANI particles. PA (solid) and TA (open symbols).



of the higher conducting particles, the conductivity is an important factor controlling their mobility in the electric field. Measured dielectric values were approached by virtue of Havriliak-Negami empirical Equation [36].

$$\epsilon^* = \epsilon'_{\infty} + \frac{(\epsilon'_0 - \epsilon')}{(1 + (i\omega t_{rel})^a)^b} \quad (2)$$

Here, ϵ'_0 and ϵ'_{∞} are the limit values of relative permittivity at the frequencies below and above the relaxation frequencies, t_{rel} is a relaxation time, a is the scattering degree of t_{rel} and b is related to the asymmetry of the t_{rel} spectrum. Large values of a mean a great scattering of t_{rel} . When a differs much from zero and b significantly from unity, the t_{rel} spectrum becomes more asymmetrical. In our case, however, the spectra are practically symmetric and the parameter b tends to one. Thus the Havriliak-Negami Equation changes to the Cole-Cole Equation

$$\epsilon^* = \epsilon'_{\infty} + \frac{(\epsilon'_0 - \epsilon'_{\infty})}{1 + (i\omega t_{rel})^a} \quad (3)$$

often successfully used for evaluation of dielectric properties of ER suspensions [37, 38].

4 CONCLUSIONS

Our findings revealed, that protonation of PANI particles with various acids at a same relatively high protonation degree may provide material with different ER property. On the other hand, conductivity of particles proved to be a universal factor controlling both viscoelastic properties of the suspension material as well as particle mobility in the electric field.

ACKNOWLEDGEMENTS

The acknowledgement for the financial support to the Ministry of Education, Youth and Sports of the Czech Republic (MSM 7088352101) and the Czech Grant Agency (202/09/1626).

REFERENCES

- [1] Winslow WM: US Patent 2 417 850 (1947).
- [2] Block H, Kelly JP: Electro-rheology, *J. Phys. D-Appl. Phys.* 21 (1988) 1661–1677.
- [3] Jordan TC, Shaw MT: Electrorheology, *IEEE Trans. Electron. Insul.* 24 (1989) 849–878.
- [4] Block H, Kelly JP, Qin A, Watson T: Materials and mechanisms in electrorheology, *Langmuir* 6 (1990) 6–14.
- [5] Conrad H, Sprecher AF: Characteristics and mechanisms of electrorheological fluids, *J. Stat. Phys.* 64 (1991) 1073–1091.
- [6] Blackwood KM, Block H: Semi-conducting polymers in electrorheology: a modern approach to smart fluids, *Trends Polym. Sci.* 14 (1993) 98–104.
- [7] Martin JE, Adolf D, Halsey TC: Electrorheology of a model colloidal fluid, *J. Colloid Interface Sci.* 167 (1994) 437–452.
- [8] Parthasarathy M, Klingenberg DJ: Electrorheology: mechanisms and models, *Mater. Sci. Eng. R* 17 (1996) 57–103.
- [9] See H: Mechanisms of magneto- and electro-rheology: recent progress and unresolved issues, *Appl. Rheol.* 11 (2001) 70–82.
- [10] Hao T: Electrorheological fluids, *Adv. Mater.* 13 (2001) 1847–1856.
- [11] Schneider S, Eibl S: Review of the electrorheological (ER) effect of polyurethane-based ER fluids, *Appl. Rheol.* 18 (2008) 23956–23963.
- [12] Pavlínek V, Sába P, Kitano T, Tanegashima T: Influence of the electric field on the electrorheological behaviour of crystalline cellulose suspensions in silicone oil, *Appl. Rheol.* 9 (1999) 64–68.
- [13] Alanis E, Romero G, Martinez C, Alvarez L, Mechetti C: Characteristic times of microstructure formation in electrorheological fluids, determined by viscosity and speckle activity measurements, *Appl. Rheol.* 15 (2005) 38–45.
- [14] Sung JH, Cho MS, Choi HJ, Jhon MS: Electrorheology of semiconducting polymers, *J. Int. Eng. Chem.* 10 (2004) 1217–1229.
- [15] Park SM, Lee HJ: Recent advances in electrochemical studies of π -conjugated polymers, *Bull. Korean Chem. Soc.* 26 (2005) 697–706.
- [16] Cheng Q, Pavlínek V, Belza T, Lengálová A, He Y, Li C: The effect of polypyrrole loading on the electrorheological properties of polypyrrole/SBA-15 suspensions, *Int. J. Mod. Phys. B* 21 (2007) 5026–5032.
- [17] Quadrat O, Stejskal J: Polyaniline in electrorheology, *J. Ind. Eng. Chem.* 12 (2006) 352–361.
- [18] Pavlínek V, Sába P, Peréz-González J, de Vargas L, Stejskal J, Quadrat O: Analysis of the yielding behaviour of electrorheological suspensions by controlled shear stress experiments, *Appl. Rheol.* 16 (2006) 14–18.

- [19] Stejskal J, Prokes J, Trchová M: Reprotonation of polyaniline: a route to various conducting polymer materials, *React. Funct. Polym.* 68 (2008) 1355–1361.
- [20] Choi HJ, Jhon MS: Electrorheology of polymers and nanocomposites, *Soft Matter* 5 (2009) 1562–1567.
- [21] MacDiarmid AG: Synthetic metals: a novel role for organic polymers, *Synth. Met.* 125 (2002) 11–22.
- [22] Sapurina I, Stejskal J: The mechanism of the oxidative polymerization of aniline and the formation of supramolecular polyaniline structures, *Polym. Int.* 57 (2008) 1295–1325.
- [23] Blinova NV, Stejskal J, Trchová M, Prokes J: Control of polyaniline conductivity and contact angles by partial protonation, *Polym. Int.* 57 (2008) 66–69.
- [24] Jang WH, Kim JW, Choi HJ, Jhon MS: Synthesis and electrorheology of camphorsulfonic acid doped polyaniline suspensions, *Colloid Polym. Sci.* 279 (2001) 823–827.
- [25] Zhang Z, Wei Z, Wan M: Nanostructures of polyaniline doped with inorganic acids, *Macromolecules* 35 (2002) 5937–5942.
- [26] Hong CH, Choi HJ: Shear stress and dielectric analysis of H_3PO_4 doped polyaniline based electrorheological fluid, *J. Macromol. Sci. Part B-Phys.* 46 (2007) 683–692.
- [27] Choi HJ, Cho MS, To K: Electrorheological and dielectrical characteristics of semiconductive polyaniline-silicone oil suspensions, *Physica A* 254 (1998) 272–279.
- [28] Lee KH, Park BJ, Song DH, Chin IJ, Choi HJ: The role of acidic *m*-cresol in polyaniline doped by camphorsulfonic acid, *Polymer* 50 (2009) 4372–4377.
- [29] Hwang JY, Cho MS, Choi HJ, Jhon MS: Synthesis of polyaniline using stabilizer and its electrorheological properties, *Synth. Met.* 135-136 (2003) 21–22.
- [30] Stenicka M, Pavlínek V, Sába P, Blinova NV, Stejskal J, Quadrat O: Conductivity of flowing polyaniline suspensions in electric field, *Colloid Polym. Sci.* 286 (2008) 1403–1409.
- [31] Stejskal J, Gilbert RG: Polyaniline. Preparation of a conducting polymer (IUPAC technical report), *Pure Appl. Chem.* 74 (2002) 857–867.
- [32] Herschel WH, Bulkley R: Konsistenzmessungen von Gummi-Benzol-Lösungen, *Kolloid Z.* 39 (1926) 291–300.
- [33] Wu CW, Conrad H: A modified conduction model for the electrorheological effect, *J. Phys. D-Appl. Phys.* 29 (1996) 3147–3153.
- [34] David LC: Polarization forces and conductivity effects in electrorheological fluids, *J. Appl. Phys.* 72 (1992) 1334–1340.
- [35] Lan Y, Xu X, Men S, Lu K: The conductivity dependence of the shear stress in electrorheological fluids, *Appl. Phys. Lett.* 73 (1998) 2908–2910.
- [36] Havriliak SJr, Havriliak SJ: Dielectric and mechanical relaxation in materials, Hanser, Munich (1997).
- [37] Cho MS, Cho YH, Choi HJ, Jhon MS: Synthesis and electrorheological characteristics of polyaniline-coated poly(methyl methacrylate) microspheres: size effect, *Langmuir* 19 (2003) 5875–5881.
- [38] Kim SG, Lim JY, Sung JH, Choi HJ, Seo Y: Emulsion polymerized polyaniline synthesized with dodecylbenzenesulfonic acid and its electrorheological characteristics: temperature effect, *Polymer* 48 (2007) 6622–6631.

