

2 4. Industrial Processes and Product Use

3 The Industrial Processes and Product Use (IPPU) chapter includes greenhouse gas emissions occurring from
4 industrial processes and from the use of greenhouse gases in products. The industrial processes and product use
5 categories included in this chapter are presented in Figure 4-1 and Figure 4-2. Greenhouse gas emissions from
6 industrial processes can occur in two different ways. First, they may be generated and emitted as the byproducts
7 of various non-energy-related industrial activities. Second, they may be emitted due to their use in manufacturing
8 processes or by end-consumers. Combustion-related energy use emissions from industry are reported in Chapter
9 3, Energy.

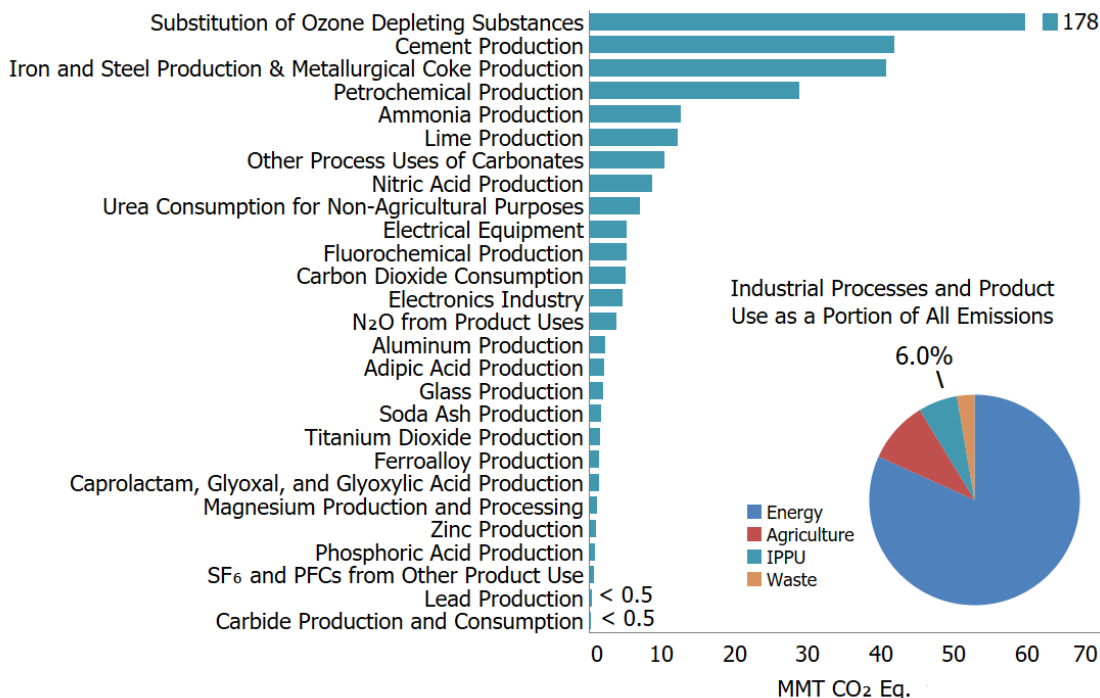
10 In the case of byproduct emissions, the emissions are generated by an industrial process itself and are not directly
11 a result of energy consumed during the process. For example, raw materials can be chemically or physically
12 transformed from one state to another. This transformation can result in the release of greenhouse gases such as
13 carbon dioxide (CO₂), methane (CH₄), nitrous oxide (N₂O), and fluorinated greenhouse gases (e.g., HFC-23). The
14 greenhouse gas byproduct generating processes included in this chapter include iron and steel production and
15 metallurgical coke production, cement production, petrochemical production, ammonia production, lime
16 production, other process uses of carbonates (e.g., flux stone, flue gas desulfurization, ceramics production, non-
17 metallurgical magnesia production, and soda ash consumption not associated with glass manufacturing), nitric acid
18 production, adipic acid production, urea consumption for non-agricultural purposes, aluminum production, HCFC-
19 22 production, other fluorochemical production, glass production, soda ash production, ferroalloy production,
20 titanium dioxide production, caprolactam production, zinc production, phosphoric acid production, lead
21 production, and silicon carbide production and consumption.

22 Greenhouse gases that are used in manufacturing processes or by end-consumers include man-made compounds
23 such as hydrofluorocarbons (HFCs), perfluorocarbons (PFCs), sulfur hexafluoride (SF₆), and nitrogen trifluoride
24 (NF₃). The present contribution of HFCs, PFCs, SF₆, and NF₃ gases to the radiative forcing effect of all anthropogenic
25 greenhouse gases is small; however, because of their extremely long lifetimes, many of them will continue to
26 persist in the atmosphere long after they were first released. In addition, many of these gases have high global
27 warming potentials; SF₆ is the most potent greenhouse gas the Intergovernmental Panel on Climate Change (IPCC)
28 has evaluated. Use of HFCs is growing rapidly since they are the primary substitutes for ozone depleting substances
29 (ODS), which are being phased-out under the Montreal Protocol on Substances that Deplete the Ozone Layer.
30 Hydrofluorocarbons, PFCs, SF₆, and NF₃ are employed and emitted by a number of other industrial sources in the
31 United States, such as the electronics industry, electric power transmission and distribution, aluminum production,
32 PFCs and SF₆ for other product use, and magnesium metal production and processing. Carbon dioxide is also
33 consumed and emitted through various end-use applications. In addition, nitrous oxide is used in and emitted by
34 the electronics industry and anesthetic and aerosol applications.

1 In 2022, IPPU generated emissions of 380.5 million metric tons of CO₂ equivalent (MMT CO₂ Eq.), or 6.0 percent of
 2 total U.S. greenhouse gas emissions.¹ Carbon dioxide emissions from all industrial processes were 168.9 MMT CO₂
 3 Eq. (168,931 kt CO₂) in 2022, or 3.3 percent of total U.S. CO₂ emissions. Methane emissions from industrial
 4 processes resulted in emissions of approximately 0.04 MMT CO₂ Eq. (1 kt CH₄) in 2022, which was 0.01 percent of
 5 U.S. CH₄ emissions. Nitrous oxide emissions from IPPU were 16.1 MMT CO₂ Eq. (61 kt N₂O) in 2022, or 4.2 percent
 6 of total U.S. N₂O emissions. In 2022 combined emissions of HFCs, PFCs, SF₆, and NF₃ totaled 195.5 MMT CO₂ Eq.
 7 Total emissions from IPPU in 2022 were 1.0 percent more than 1990 emissions. Total emissions from IPPU
 8 remained relatively constant between 2021 and 2022, increasing by 0.1 percent due to offsetting trends within the
 9 sector. More information on emissions of greenhouse gas precursors emissions that also result from IPPU are
 10 presented in Section 4.27 of this chapter.

11 The largest source of IPPU-related emissions is the substitution of ozone depleting substances, which accounted
 12 for 46.8 percent of sector emissions in 2022. These emissions have increased by 79.1 percent since 2005, and 3.2
 13 percent between 2021 and 2022. Cement production was the second largest source of IPPU emissions in 2022,
 14 accounting for 11.0 percent of IPPU emissions in 2022. Iron and steel production and metallurgical coke production
 15 was the third largest source of IPPU emissions, accounting for 10.7 percent of the sector total in 2022.

16 **Figure 4-1: Industrial Processes and Product Use Sector Greenhouse Gas Sources**

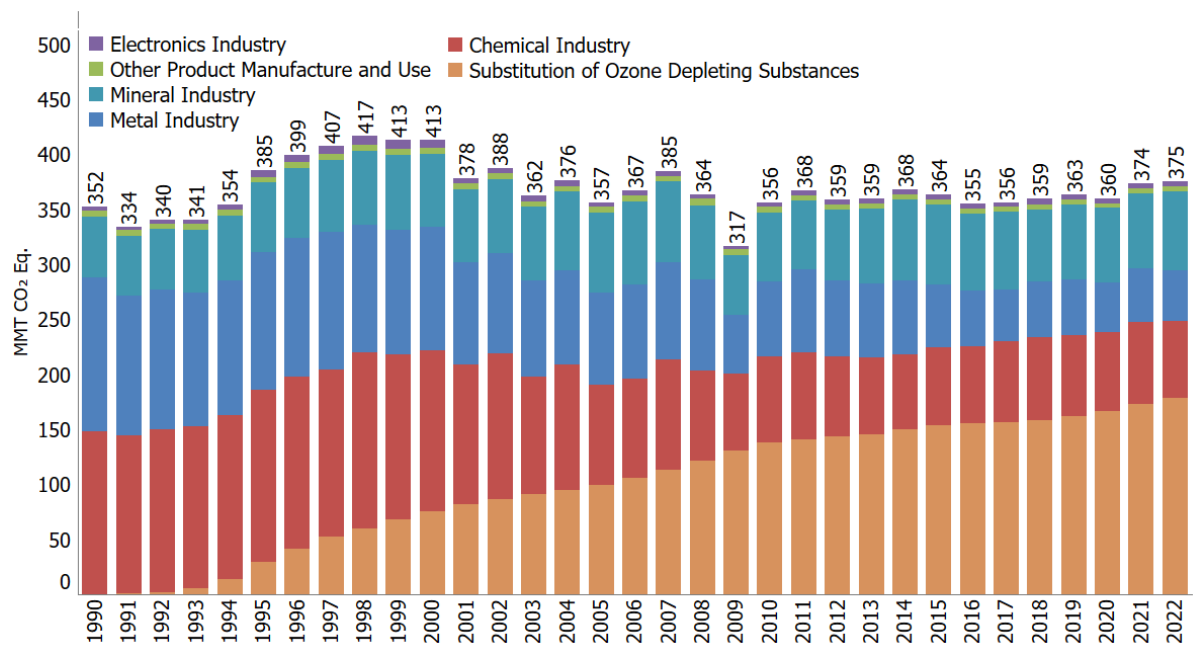


17
 18 The increase in overall IPPU emissions since 1990 reflects a range of emission trends among the emission sources,
 19 as shown in Figure 4-2. Emissions resulting from most types of metal production have declined significantly since
 20 1990, largely due to production shifting to other countries, but also due to transitions to less-emissive methods of
 21 production (in the case of iron and steel) and to improved practices (in the case of PFC emissions from aluminum
 22 production). Carbon dioxide and CH₄ emissions from some chemical production sources (e.g., petrochemical
 23 production, urea consumption for non-agricultural purposes) have increased since 1990, while emissions from

¹ Emissions reported in the IPPU chapter include those from all 50 states, including Hawaii and Alaska, as well as from U.S. Territories.

1 other chemical production sources (e.g., ammonia production, phosphoric acid production) have decreased.
 2 Emissions from mineral sources have either increased (e.g., cement production) or not changed significantly (e.g.,
 3 lime production) since 1990 and largely follow economic cycles. Hydrofluorocarbon emissions from the
 4 substitution of ODS have increased drastically since 1990 and are the largest source of IPPU emissions (46.8
 5 percent in 2022), while the emissions of HFCs, PFCs, SF₆, and NF₃ from other sources have generally declined.
 6 Nitrous oxide emissions from the production of nitric acid have decreased. Some emission sources (e.g., adipic
 7 acid) exhibit varied interannual trends. Trends are explained further within each emission source category
 8 throughout the chapter.

9 **Figure 4-2: Trends in Industrial Processes and Product Use Sector Greenhouse Gas Sources**



10
 11 Table 4-1 summarizes emissions for the IPPU chapter in MMT CO₂ Eq. using IPCC *Fifth Assessment Report* (AR5)
 12 GWP values, following the requirements of the current United Nations Framework Convention on Climate Change
 13 (UNFCCC) reporting guidelines for national inventories (IPCC 2007).² Unweighted gas emissions in kt are also
 14 provided in Table 4-2. The source descriptions that follow in the chapter are presented in the order as reported to
 15 the UNFCCC in the Common Reporting Tables (CRTs), corresponding generally to: mineral industry, chemical
 16 industry, metal industry, and emissions from the uses of HFCs, PFCs, SF₆, and NF₃.

17 Each year, some emission and sink estimates in the IPPU sector of the *Inventory* are recalculated and revised with
 18 improved methods and/or data. In general, recalculations are made to the U.S. greenhouse gas emission estimates
 19 either to incorporate new methodologies or, most commonly, to update recent historical data. These
 20 improvements are implemented consistently across the previous *Inventory's* time series (i.e., 1990 to 2021) to
 21 ensure that the trend is accurate. Key updates to this year's *Inventory* include the addition of new categories
 22 previously not estimated: CO₂ emissions from ceramics production and non-metallurgical magnesia within other
 23 process use of carbonates category; fluorinated gases from fluorochemical production other than HCFC-22 within
 24 the fluorochemical production category; and SF₆ and PFCs from product uses within the other product
 25 manufacture and use category. In addition, there were changes to the petrochemical production methodology to
 26 calculate emissions from methanol production; updates to emission estimates from urea consumption for non-

² See <http://unfccc.int/resource/docs/2013/cop19/eng/10a03.pdf>.

1 agricultural purposes driven by revisions to quantities of urea applied, urea imports, and urea exports; and
 2 revisions to the method for estimating SF₆ emissions from electrical equipment for estimating using CARB data
 3 from electrical equipment in California. Together, these methodological and data updates increased IPPU sector
 4 greenhouse gas emissions an average 8.0 MMT CO₂ Eq. (2.9 percent) across the time series. For more information
 5 on specific methodological updates, please see the Recalculations Discussion section for each category in this
 6 chapter.

7 **Table 4-1: Emissions from Industrial Processes and Product Use (MMT CO₂ Eq.)**

Gas/Source	1990	2005	2018	2019	2020	2021	2022
CO₂	215.2	196.4	164.4	168.2	160.7	168.8	168.9
Cement Production	33.5	46.2	39.0	40.9	40.7	41.3	41.9
Iron and Steel Production & Metallurgical Coke Production	104.7	70.1	42.9	43.1	37.7	41.9	40.7
<i>Iron and Steel Production</i>	99.1	66.2	41.6	40.1	35.4	38.6	37.7
<i>Metallurgical Coke Production</i>	5.6	3.9	1.3	3.0	2.3	3.2	3.0
Petrochemical Production	21.6	27.4	27.2	28.5	27.9	30.7	28.8
Ammonia Production	14.4	10.2	12.7	12.4	13.0	12.2	12.6
Lime Production	11.7	14.6	13.1	12.1	11.3	11.9	12.2
Other Process Uses of Carbonates	7.1	8.5	7.9	9.0	9.0	8.6	10.4
Urea Consumption for Non- Agricultural Purposes	3.8	3.7	6.1	6.2	5.8	6.6	7.1
Carbon Dioxide Consumption	1.5	1.4	4.1	4.9	5.0	5.0	5.0
Glass Production	2.3	2.4	2.0	1.9	1.9	2.0	2.0
Soda Ash Production	1.4	1.7	1.7	1.8	1.5	1.7	1.7
Titanium Dioxide Production	1.2	1.8	1.5	1.3	1.3	1.5	1.5
Aluminum Production	6.8	4.1	1.5	1.9	1.7	1.5	1.4
Ferroalloy Production	2.2	1.4	2.1	1.6	1.4	1.6	1.3
Zinc Production	0.6	1.0	1.0	1.0	1.0	1.0	0.9
Phosphoric Acid Production	1.5	1.3	0.9	0.9	0.9	0.9	0.8
Lead Production	0.5	0.6	0.5	0.5	0.5	0.4	0.4
Carbide Production and Consumption	0.2	0.2	0.2	0.2	0.2	0.2	0.2
Substitution of Ozone Depleting Substances	+	+	+	+	+	+	+
Magnesium Production and Processing	0.1	+	+	+	+	+	+
CH₄	0.1	+	+	+	+	+	+
Carbide Production and Consumption	+	+	+	+	+	+	+
Ferroalloy Production	+	+	+	+	+	+	+
Iron and Steel Production & Metallurgical Coke Production	+	+	+	+	+	+	+
Petrochemical Production	+	+	+	+	+	+	+
N₂O	29.6	22.2	23.1	18.8	20.8	19.7	16.1
Nitric Acid Production	10.8	10.1	8.5	8.9	8.3	7.9	8.6
N ₂ O from Product Uses	3.8	3.8	3.8	3.8	3.8	3.8	3.8
Adipic Acid Production	13.5	6.3	9.3	4.7	7.4	6.6	2.1
Caprolactam, Glyoxal, and Glyoxylic Acid Production	1.5	1.9	1.3	1.2	1.1	1.2	1.3
Electronics Industry	+	0.1	0.2	0.2	0.3	0.3	0.3
HFCs	48.2	121.2	162.9	167.3	169.6	176.3	181.4
Substitution of Ozone Depleting Substances ^a	0.3	99.5	157.9	162.1	166.2	172.6	178.1
Fluorochemical Production	47.8	21.6	4.6	4.8	3.0	3.3	2.9
Electronics Industry	0.2	0.2	0.3	0.3	0.3	0.4	0.3

Gas/Source	1990	2005	2018	2019	2020	2021	2022
Magnesium Production and Processing	0.0	0.0	0.1	0.1	0.1	+	+
PFCs	47.1	7.8	5.8	5.9	5.5	5.4	5.4
Electronics Industry	2.5	3.0	2.9	2.6	2.6	2.7	2.7
Fluorochemical Production	25.2	1.6	1.3	1.7	1.3	1.6	1.8
Aluminum Production	19.3	3.1	1.4	1.4	1.4	0.9	0.8
SF ₆ and PFCs from Other Product Use	0.1	0.1	0.2	0.2	0.2	0.1	0.2
Substitution of Ozone Depleting Substances ^a	NO	+	+	+	+	+	+
Electrical Equipment	+	+	0.0	+	+	+	+
SF₆	35.9	20.0	7.7	8.4	8.1	8.5	7.6
Electrical Equipment	24.7	11.8	5.0	6.1	5.9	6.0	5.1
Magnesium Production and Processing	5.6	3.0	1.1	0.9	0.9	1.2	1.1
Electronics Industry	0.5	0.8	0.8	0.8	0.8	1.0	0.7
SF ₆ and PFCs from Other Product Use	1.4	1.4	0.8	0.6	0.5	0.4	0.6
Fluorochemical Production	3.8	3.0	+	+	+	+	+
NF₃	0.7	0.8	0.7	1.1	1.3	1.2	1.1
Electronics Industry	+	0.4	0.6	0.5	0.6	0.7	0.6
Fluorochemical Production	0.7	0.4	0.1	0.6	0.7	0.5	0.5
Total^b	376.9	368.5	364.6	369.6	366.1	380.0	380.5

+ Does not exceed 0.05 MMT CO₂ Eq.

NO (Not Occurring)

^a Small amounts of PFC emissions from this source are included under HFCs due to confidential business information.

^b Total does not include other fluorinated gases, such as HFEs and PFPEs, which are reported separately in Section 4.23.

Note: Totals may not sum due to independent rounding. Emissions of F-HTFs that are not HFCs, PFCs or SF₆ are not included in *Inventory* totals and are included for informational purposes only in Section 4.23. Emissions presented for informational purposes include HFEs, PFPMEs, perfluoroalkylmorpholines, and perfluorotrialkylamines.

1 **Table 4-2: Emissions from Industrial Processes and Product Use (kt)**

Gas/Source	1990	2005	2018	2019	2020	2021	2022
CO₂	215,218	196,435	164,400	168,171	160,701	168,839	168,931
Cement Production	33,484	46,194	38,971	40,896	40,688	41,312	41,884
Iron and Steel Production & Metallurgical Coke Production	104,740	70,082	42,863	43,095	37,724	41,873	40,672
<i>Iron and Steel Production</i>	<i>99,132</i>	<i>66,161</i>	<i>41,581</i>	<i>40,089</i>	<i>35,398</i>	<i>38,648</i>	<i>37,718</i>
<i>Metallurgical Coke Production</i>	<i>5,608</i>	<i>3,921</i>	<i>1,282</i>	<i>3,006</i>	<i>2,325</i>	<i>3,224</i>	<i>2,954</i>
Petrochemical Production	21,611	27,383	27,200	28,483	27,926	30,656	28,788
Ammonia Production	14,404	10,234	12,669	12,401	13,006	12,192	12,610
Lime Production	11,700	14,552	13,106	12,112	11,299	11,870	12,208
Other Process Uses of Carbonates	7,103	8,472	7,935	8,969	9,012	8,583	10,378
Urea Consumption for Non-Agricultural Purposes	3,784	3,653	6,113	6,150	5,805	6,600	7,053
Carbon Dioxide Consumption	1,472	1,375	4,130	4,870	4,970	4,990	5,000
Glass Production	2,263	2,402	1,989	1,940	1,858	1,969	1,956
Soda Ash Production	1,431	1,655	1,714	1,792	1,461	1,714	1,704
Titanium Dioxide Production	1,195	1,755	1,541	1,340	1,340	1,474	1,474
Aluminum Production	6,831	4,142	1,455	1,880	1,748	1,541	1,446
Ferroalloy Production	2,152	1,392	2,063	1,598	1,377	1,567	1,327
Zinc Production	632	1,030	999	1,026	977	1,007	947
Phosphoric Acid Production	1,529	1,342	937	909	901	874	840
Lead Production	516	553	527	531	450	439	428

Gas/Source	1990	2005	2018	2019	2020	2021	2022
Carbide Production and Consumption	243	213	184	175	154	172	210
Substitution of Ozone Depleting Substances ^a	+	1	3	3	4	4	4
Magnesium Production and Processing	129	4	2	2	3	3	3
CH₄	3	2	2	1	1	1	1
Carbide Production and Consumption	1	+	+	+	+	+	+
Ferroalloy Production	1	+	1	+	+	+	+
Iron and Steel Production & Metallurgical Coke Production	1	1	+	+	+	+	+
Petrochemical Production	+	+	+	+	+	+	+
N₂O	112	84	87	71	79	74	61
Nitric Acid Production	41	38	32	34	31	30	33
N ₂ O from Product Uses	14	14	14	14	14	14	14
Adipic Acid Production	51	24	35	18	28	25	8
Caprolactam, Glyoxal, and Glyoxylic Acid Production	6	7	5	5	4	5	5
Electronics Industry	+	+	1	1	1	1	1
HFCs	M	M	M	M	M	M	M
Substitution of Ozone Depleting Substances ^a	M	M	M	M	M	M	M
Fluorochemical Production	M	M	M	M	M	M	M
Electronics Industry	+	+	+	+	+	+	+
Magnesium Production and Processing	+	+	+	+	+	+	+
PFCs	+	+	+	+	+	+	+
Electronics Industry	+	+	+	+	+	+	+
Fluorochemical Production	M	M	M	M	M	M	M
Aluminum Production	M	M	M	M	M	M	M
SF ₆ and PFCs from Other Product Use	+	+	+	+	+	+	+
Substitution of Ozone Depleting Substances ^a	NO	+	+	+	+	+	+
Electrical Equipment	+	+	0.0	+	+	+	+
SF₆	2	1	+	+	+	+	+
Electrical Equipment	1	1	+	+	+	+	+
Magnesium Production and Processing	+	+	+	+	+	+	+
Electronics Industry	+	+	+	+	+	+	+
SF ₆ and PFCs from Other Product Use	+	+	+	+	+	+	+
Fluorochemical Production	+	+	+	+	+	+	+
NF₃	+	+	+	+	+	+	+
Electronics Industry	+	+	+	+	+	+	+
Fluorochemical Production	+	+	+	+	+	+	+

+ Does not exceed 0.5 kt.

M (Mixture of gases)

NO (Not Occurring)

^a Small amounts of PFC emissions from this source are included under HFCs due to confidential business information.

Note: Totals by gas may not sum due to independent rounding.

- 1 This chapter presents emission estimates calculated in accordance with the *2006 IPCC Guidelines for National Greenhouse Gas Inventories (2006 IPCC Guidelines)* and its refinements. For additional detail on IPPU sources that
- 2 are not included in this *Inventory* report, please review Annex 5, Assessment of the Sources and Sinks of
- 3

1 Greenhouse Gas Emissions Not Included. These sources are not included due to various national circumstances,
2 such as emissions from a source may not currently occur in the United States, data are not currently available for
3 those emission sources (e.g., ceramics, non-metallurgical magnesium production, glyoxal and glyoxylic acid
4 production, CH₄ from direct reduced iron production), emissions are included elsewhere within the *Inventory*
5 report, or data suggest that emissions are not significant (e.g., other various fluorinated gas emissions from other
6 product uses). In terms of geographic scope, emissions reported in the IPPU chapter include those from all 50
7 states, including Hawaii and Alaska, as well as from District of Columbia and U.S. Territories to the extent to which
8 industries are occurring. While most IPPU sources do not occur in U.S. Territories (e.g., electronics manufacturing
9 does not occur in U.S. Territories), they are estimated and accounted for where they are known to occur (e.g.,
10 cement production, lime production, and electrical equipment). EPA will review this on an ongoing basis to ensure
11 emission sources are included across all geographic areas if they occur. Information on planned improvements for
12 specific IPPU source categories can be found in the Planned Improvements section of the individual source
13 category.

14 In addition, as mentioned in the Energy chapter of this report (Box 3-5), fossil fuels consumed for non-energy uses
15 for primary purposes other than combustion for energy (including lubricants, paraffin waxes, bitumen asphalt, and
16 solvents) are reported in the Energy chapter. According to the *2006 IPCC Guidelines*, these non-energy uses of
17 fossil fuels are to be reported under the IPPU, rather than the Energy sector; however, due to national
18 circumstances regarding the allocation of energy statistics and carbon balance data, the United States reports
19 these non-energy uses in the Energy chapter of this *Inventory*. Although emissions from these non-energy uses are
20 reported in the Energy chapter, the methodologies used to determine emissions are compatible with the *2006*
21 *IPCC Guidelines* and are well documented and scientifically based. The methodologies used are described in
22 Section 3.2, Carbon Emitted from Non-Energy Uses of Fossil Fuels and Annex 2.3, Methodology for Estimating
23 Carbon Emitted from Non-Energy Uses of Fossil Fuels. The emissions are reported under the Energy chapter to
24 improve transparency, report a more complete carbon balance, and avoid double counting. For example, only the
25 emissions from the first use of lubricants and waxes are to be reported under the IPPU sector, and emissions from
26 use of lubricants in 2-stroke engines and emissions from secondary use of lubricants and waxes in waste
27 incineration with energy recovery are to be reported under the Energy sector. Reporting non-energy use emissions
28 from only first use of lubricants and waxes under IPPU would involve making artificial adjustments to the non-
29 energy use carbon balance and could potentially result in double counting of emissions. These artificial
30 adjustments would also be required for asphalt and road oil and solvents (which are captured as part of
31 petrochemical feedstock emissions) and could also potentially result in double counting of emissions. For more
32 information, see the Methodology discussion in Section 3.1, CO₂ from Fossil Fuel Combustion, Section 3.2, Carbon
33 Emitted from Non-Energy Uses of Fossil Fuels and Annex 2.3, Methodology for Estimating Carbon Emitted from
34 Non-Energy Uses of Fossil Fuels.

35 Finally, as stated in the Energy chapter, portions of the fuel consumption data for seven fuel categories—coking
36 coal, distillate fuel, industrial other coal, petroleum coke, natural gas, residual fuel oil, and other oil—are
37 reallocated to the IPPU chapter, as they are consumed during non-energy related industrial process activity.
38 Emissions from uses of fossil fuels as feedstocks or reducing agents (e.g., petrochemical production, aluminum
39 production, titanium dioxide, zinc production) are reported in the IPPU chapter, unless otherwise noted due to
40 specific national circumstances. This approach is compatible with the *2006 IPCC Guidelines* and is well documented
41 and scientifically based. The emissions from these feedstocks and reducing agents are reported under the IPPU
42 chapter to improve transparency and to avoid double counting of emissions under both the Energy and IPPU
43 sectors. More information on the methodology to adjust for these emissions within the Energy chapter is
44 described in the Methodology section of CO₂ from Fossil Fuel Combustion (3.1 Fossil Fuel Combustion [CRT Source
45 Category 1A]) and Annex 2.1, Methodology for Estimating Emissions of CO₂ from Fossil Fuel Combustion.
46 Additional information is listed within each IPPU emission source in which this approach applies.

Box 4-1: Methodological Approach for Estimating and Reporting U.S. Emissions and Removals

Consistent with Article 13.7(a) of the Paris Agreement and Article 4.1(a) of the UNFCCC as well as relevant decisions under those agreements, the emissions and removals presented in this report and this chapter are organized by source and sink categories and calculated using internationally accepted methods provided by the Intergovernmental Panel on Climate Change (IPCC) in the *2006 IPCC Guidelines for National Greenhouse Gas Inventories (2006 IPCC Guidelines)* and its supplements and refinements. Additionally, the calculated emissions and removals in a given year for the United States are presented in a common format in line with the reporting guidelines for the reporting of inventories under the Paris Agreement and the UNFCCC. The Parties' use of consistent methods to calculate emissions and removals for their inventories helps to ensure that these reports are comparable. The presentation of emissions and removals provided in the IPPU chapter do not preclude alternative examinations. Rather, this chapter presents emissions and removals in a common format consistent with how Parties are to report inventories under the Paris Agreement and the UNFCCC. The report itself, and this chapter, follows this common format, and provides an explanation of the application of methods used to calculate emissions and removals from industrial processes and from the use of greenhouse gases in products.

QA/QC and Verification Procedures

The quality of IPPU source categories is assured through application of the U.S. Inventory QA/QC plan outlined in Annex 8. Two types of checks were performed using this plan: (1) general (Tier 1) procedures consistent with Volume 1, Chapter 6 of the *2006 IPCC Guidelines* that focus on annual procedures and checks to be used when gathering, maintaining, handling, documenting, checking, and archiving the data, supporting documents, and files; and (2) source category-specific (Tier 2) procedures that focus on checks and comparisons of the emission factors, activity data, and methodologies used for estimating emissions from the relevant industrial process and product use sources. Examples of these procedures include: checks to ensure that activity data and emission estimates are consistent with historical trends to identify significant changes; that, where possible, consistent and reputable data sources are used and specified across sources; that interpolation or extrapolation techniques are consistent across sources; and that common datasets, units, and conversion factors are used where applicable. The checks also include checking for transcription errors in data inputs required for emission calculations, including activity data and emission factors; and confirmed that estimates were calculated and reported for all applicable source categories for all years. Consistent with the *2006 IPCC Guidelines*, additional category-specific QC procedures were performed for more significant emission categories (such as the comparison of reported consumption with modeled consumption using EPA's Greenhouse Gas Reporting Program (GHGRP) data within Substitution of Ozone Depleting Substances) or sources where significant methodological and data updates have taken place. The QA/QC implementation did not reveal any significant inaccuracies, and all errors identified were documented and corrected. Application of these procedures, specifically category-specific QC procedures and updates/improvements as a result of QA processes (expert, public, and UNFCCC technical expert reviews), are described further within respective source categories, in the Recalculations Discussion and Planned Improvement sections. General QA/QC procedures (Tier 1) and calculation-related QC (category-specific, Tier 2) have been performed for all IPPU sources.

For sources that use data from EPA's Greenhouse Gas Reporting Program (GHGRP), EPA verifies annual facility-level reports through a multi-step process (e.g., including a combination of pre-and post-submittal electronic checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent.³ Based on the results of the verification process, EPA follows up with facilities

³ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general
2 and category-specific QC procedures, including: range checks, statistical checks, algorithm checks, and year-to-year
3 checks of reported data and emissions. See Box 4-2 below for more information on use of GHGRP data in this
4 chapter.

5 For most IPPU categories, activity data are obtained via aggregation of facility-level data from EPA's GHGRP (see
6 Box 4-2 below and Annex 9), national commodity surveys conducted by U.S. Geological Survey (USGS) National
7 Minerals Information Center, U.S. Department of Energy (DOE), U.S. Census Bureau, and industry associations such
8 as Air-Conditioning, Heating, and Refrigeration Institute (AHRI), American Chemistry Council (ACC), and American
9 Iron and Steel Institute (AISI) (specified within each source category). The emission factors used include those
10 derived from the EPA's GHGRP and application of IPCC default factors. Descriptions of uncertainties and
11 assumptions for activity data and emission factors are included within the uncertainty discussion sections for each
12 IPPU source category.

13 **Box 4-2: Industrial Process and Product Use Data from EPA's Greenhouse Gas Reporting** 14 **Program**

EPA collects greenhouse gas emissions data from individual facilities and suppliers of certain fossil fuels and industrial gases through its Greenhouse Gas Reporting Program (GHGRP). The GHGRP applies to direct greenhouse gas emitters, fossil fuel suppliers, industrial gas suppliers, and facilities that inject CO₂ underground for sequestration or other reasons and requires reporting by sources or suppliers in 41 industrial categories. Annual reporting is at the facility level, except for certain suppliers of fossil fuels and industrial greenhouse gases.

In general, the threshold for reporting is 25,000 metric tons or more of CO₂ Eq. per year, but reporting is required for all facilities in some industries. Calendar year 2010 was the first year for which data were collected for facilities subject to 40 CFR Part 98, though some source categories first collected data for calendar year 2011. For more information, see Annex 9, Use of EPA Greenhouse Gas Reporting Program in Inventory.

EPA uses annual GHGRP data in a number of categories to improve the national estimates presented in this *Inventory*, consistent with IPCC guidelines (e.g., minerals, chemicals, product uses). Methodologies used in EPA's GHGRP are consistent with IPCC guidelines, including higher tier methods; however, it should be noted that the coverage and definitions for source categories (e.g., allocation of energy and IPPU emissions) in EPA's GHGRP may differ from those used in this *Inventory* in meeting the Paris Agreement and UNFCCC reporting guidelines (IPCC 2011) and is an important consideration when incorporating GHGRP data in the *Inventory*. In line with the Paris Agreement and UNFCCC reporting guidelines, the *Inventory* is a comprehensive accounting of all emissions from source categories identified in the *2006 IPCC Guidelines*. EPA has paid particular attention to ensuring both completeness and time-series consistency for major recalculations that have occurred from the incorporation of GHGRP data into these categories, consistent with *2006 IPCC Guidelines* and *IPCC Technical Bulletin on Use of Facility-Specific Data in National GHG Inventories*.⁴

For certain source categories in this *Inventory* (e.g., nitric acid production, lime production, cement production, petrochemical production, carbon dioxide consumption, ammonia production, and urea consumption for non-agricultural purposes), EPA has integrated data values that have been calculated by aggregating GHGRP data that are considered confidential business information (CBI) at the facility level. EPA, with industry engagement, has put forth criteria to confirm that a given data aggregation shields underlying CBI from public disclosure. EPA is only publishing data values that meet these aggregation criteria.⁵ Specific uses of aggregated facility-level

⁴ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

⁵ U.S. EPA Greenhouse Gas Reporting Program. Developments on Publication of Aggregated Greenhouse Gas Data, November 25, 2014. See <http://www.epa.gov/ghgreporting/confidential-business-information-ghg-reporting>.

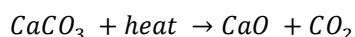
data are described in the respective methodological sections (e.g., including other sources using GHGRP data that is not aggregated CBI, such as aluminum, electronics industry, electrical equipment, HCFC-22 production, and magnesium production and processing). For other source categories in this chapter, as indicated in the respective planned improvements sections,⁶ EPA is continuing to analyze how facility-level GHGRP data may be used to improve the national estimates presented in this *Inventory*, giving particular consideration to ensuring time-series consistency and completeness.

Additionally, EPA's GHGRP has and will continue to enhance QA/QC procedures and assessment of uncertainties within the IPPU categories (see those categories for specific QA/QC details regarding the use of GHGRP data).

4.1 Cement Production (CRT Source Category 2A1)

Cement production is an energy- and raw material-intensive process that results in the generation of carbon dioxide (CO₂) both from the energy consumed in making the clinker precursor to cement and from the chemical process to make the clinker. Emissions from fuels consumed for energy purposes during the production of cement are accounted for in the Energy chapter.

During the clinker production process, the key reaction occurs when calcium carbonate (CaCO₃), in the form of limestone or similar rocks or in the form of cement kiln dust (CKD), is heated in a cement kiln at a temperature range of about 700 to 1,000 degrees Celsius (1,300 to 1,800 degrees Fahrenheit) to form lime (i.e., calcium oxide, or CaO) and CO₂ in a process known as calcination or calcining. The quantity of CO₂ emitted during clinker production is directly proportional to the lime content of the clinker. During calcination, each mole of CaCO₃ heated in the clinker kiln forms one mole of CaO and one mole of CO₂. The CO₂ is vented to the atmosphere as part of the kiln exhaust:



Next, over a temperature range of 1000 to 1450 degrees Celsius, the CaO combines with alumina, iron oxide and silica that are also present in the clinker raw material mix to form hydraulically reactive compounds within white-hot semifused (sintered) nodules of clinker. These "sintering" reactions are highly exothermic and produce few CO₂ process emissions. The clinker is then rapidly cooled to maintain quality and then very finely ground with a small amount of gypsum and potentially other materials (e.g., ground granulated blast furnace slag, etc.) to make portland and similar cements.

Masonry cement consists of plasticizers (e.g., ground limestone, lime, etc.) and portland cement, and the amount of portland cement used accounts for approximately 3 percent of total clinker production (USGS 2023b; 2023c). No additional emissions are associated with the production of masonry cement. Carbon dioxide emissions that result from the production of lime used to produce portland and masonry cement are included in Section 4.2 Lime Production (CRT Source Category 2A2).

Carbon dioxide emitted from the chemical process of cement production is the second largest source of industrial CO₂ emissions in the United States. Cement is produced in 34 states and Puerto Rico. Texas, Missouri, California, and Florida were the leading cement-producing states in 2022 and accounted for approximately 43 percent of total

⁶ Ammonia production, glass production, lead production, and other fluorinated gas production.

1 U.S. production (USGS 2023b). In 2022, shipments of cement were estimated to have increased by about 3% from
2 2021, and net imports increased by about 17 percent compared to 2021 (USGS 2023b).

3 In 2022, U.S. clinker production totaled 80,500 kilotons, which was an increase of 1 percent compared to 2021 and
4 an increase of 25 percent compared to 1990 (EPA 2023). The resulting CO₂ emissions were estimated to be 41.9
5 MMT CO₂ Eq. (41,884 kt) (see Table 4-3 and Table 4-4). The total construction value and cement shipments
6 increased by 11 percent and 4 percent, respectively, during the first nine months of 2022 compared to the same
7 time period in 2021. This increase was attributed to continued economic recovery from the COVID-19 pandemic
8 and the November 2021 passage of the Bipartisan Infrastructure Law. Despite the increases, growth was
9 constrained by increased costs, labor and production shortages, and ongoing supply chain disruptions (USGS
10 2023b).

11 **Table 4-3: CO₂ Emissions from Cement Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Cement Production	33.5	46.2	39.0	40.9	40.7	41.3	41.9

12 **Table 4-4: CO₂ Emissions from Cement Production (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Cement Production	33,484	46,194	38,971	40,896	40,688	41,312	41,884

13 Greenhouse gas emissions from cement production, which are primarily driven by production levels, increased
14 every year from 1991 through 2006 but decreased in the following years until 2009. Emissions from cement
15 production were at their highest levels in 2006 and at their lowest levels in 2009. Emissions in 2009 were
16 approximately 28 percent lower than 2008 emissions and 12 percent lower than 1990 due to the economic
17 recession and the associated decrease in demand for construction materials. Since 2009, emissions have increased
18 by 41 percent due to increasing demand for cement. Cement continues to be a critical component of the
19 construction industry; therefore, the availability of public and private construction funding, as well as overall
20 economic conditions, have considerable impact on the level of cement production.

21 **Methodology and Time-Series Consistency**

22 Carbon dioxide emissions from cement production are estimated using the Tier 2 method from the *2006 IPCC*
23 *Guidelines* as this is a key category, in accordance with the IPCC methodological decision tree and available data.
24 The Tier 2 methodology was used because detailed and complete data (including weights and composition) for
25 carbonate(s) consumed in clinker production are not available,⁷ and thus a rigorous Tier 3 approach is impractical.
26 Tier 2 specifies the use of aggregated plant or national clinker production data and an emission factor, which is the
27 product of the average lime mass fraction for clinker of 65 percent and a constant reflecting the mass of CO₂
28 released per unit of lime. The U.S. Geological Survey (USGS) mineral commodity expert for cement has confirmed
29 that this is a reasonable assumption for the United States (Van Oss 2013a). This calculation yields an emission
30 factor of 0.510 tons of CO₂ per ton of clinker produced, which was determined as follows:

⁷ As discussed further under “Planned Improvements,” most cement-producing facilities that report their emissions to the GHGRP use CEMS to monitor combined process and fuel combustion emissions for kilns, making it difficult to quantify the process emissions on a facility-specific basis. By the end of 2022, the percentage of facilities not using CEMS was 1 percent.

1 **Equation 4-1: 2006 IPCC Guidelines Tier 1 Emission Factor for Clinker (precursor to Equation**
 2 **2.4)**

3
$$EF_{\text{clinker}} = 0.650 \text{ CaO} \times \left[\left(44.01 \frac{\text{g}}{\text{mole}} \text{CO}_2 \right) \div \left(56.08 \frac{\text{g}}{\text{mole}} \text{CaO} \right) \right] = 0.510 \frac{\text{tons CO}_2}{\text{ton clinker}}$$

4 During clinker production, some of the raw materials, partially reacted raw materials, and clinker enters the kiln
 5 line’s exhaust system as non-calcinated, partially calcinated, or fully calcinated cement kiln dust (CKD). To the
 6 degree that the CKD contains carbonate raw materials which are then calcined, there are associated CO₂ emissions.
 7 At some plants, essentially all CKD is directly returned to the kiln, becoming part of the raw material feed, or is
 8 likewise returned to the kiln after first being removed from the exhaust. In either case, the returned CKD becomes
 9 a raw material, thus forming clinker, and the associated CO₂ emissions are a component of those calculated for the
 10 clinker overall. At some plants, however, the CKD cannot be returned to the kiln because it is chemically unsuitable
 11 as a raw material or chemical issues limit the amount of CKD that can be so reused. Any clinker that cannot be
 12 returned to the kiln is either used for other (non-clinker) purposes or is landfilled. The CO₂ emissions attributable
 13 to the non-returned calcinated portion of the CKD are not accounted for by the clinker emission factor and thus a
 14 CKD correction factor should be applied to account for those emissions. The USGS reports the amount of CKD used
 15 to produce clinker, but no information is currently available on the total amount of CKD produced annually.⁸
 16 Because data are not currently available to derive a country-specific CKD correction factor, a default correction
 17 factor of 1.02 (2 percent) was used to account for CKD CO₂ emissions, as recommended by the IPCC (IPCC 2006).⁹
 18 Total cement production emissions were calculated by adding the emissions from clinker production and the
 19 emissions assigned to CKD.

20 Small amounts of impurities (i.e., not calcium carbonate) may exist in the raw limestone used to produce clinker.
 21 The proportion of these impurities is generally minimal, although a small amount (1 to 2 percent) of magnesium
 22 oxide (MgO) may be desirable as a flux. Per the IPCC Tier 2 methodology, a correction for MgO is not used, since
 23 the amount of MgO from carbonate is likely very small and the assumption of a 100 percent carbonate source of
 24 CaO already yields an overestimation of emissions (IPCC 2006).

25 The 1990 through 2012 activity data for clinker production were obtained from USGS (Van Oss 2013a; Van Oss
 26 2013b). Clinker production data for 2013 were also obtained from USGS (USGS 2014). USGS compiled the data (to
 27 the nearest ton) through questionnaires sent to domestic clinker and cement manufacturing plants, including
 28 facilities in Puerto Rico. Clinker production values in the current *Inventory* report utilize GHGRP data for the years
 29 2014 through 2022 (EPA 2023). Clinker production data are summarized in Table 4-5. Details on how this GHGRP
 30 data compares to USGS reported data can be found in the section on QA/QC and Verification.

31 **Table 4-5: Clinker Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Clinker	64,355	88,783	74,900	78,600	78,200	79,400	80,500

Note: Clinker production from 1990 through 2022 includes Puerto Rico (relevant U.S. Territories).

⁸ The USGS *Minerals Yearbook: Cement* notes that CKD values used for clinker production are likely underreported.

⁹ As stated on p. 2.12 of the *2006 IPCC Guidelines*, Vol. 3, Chapter 2: “...As data on the amount of CKD produced may be scarce (except possibly for plant-level reporting), estimating emissions from lost CKD based on a default value can be considered good practice. The amount of CO₂ from lost CKD can vary but range typically from about 1.5 percent (additional CO₂ relative to that calculated for clinker) for a modern plant to about 20 percent for a plant losing a lot of highly calcinated CKD (van Oss 2005). In the absence of data, the default CKD correction factor (CF_{ckd}) is 1.02 (i.e., add 2 percent to the CO₂ calculated for clinker). If no calcinated CKD is believed to be lost to the system, the CKD correction factor will be 1.00 (van Oss 2005)...”

1 Methodological approaches were applied to the entire time series to ensure time-series consistency from 1990
 2 through 2022. The methodology for cement production spliced activity data from two different sources: USGS for
 3 1990 through 2013 and GHGRP starting in 2014. Consistent with the *2006 IPCC Guidelines*, the overlap technique
 4 was applied to compare the two data sets for years where there was overlap, with findings that the data sets were
 5 consistent and adjustments were not needed.

6 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

7 The uncertainties contained in these estimates are primarily due to uncertainties in the lime content of clinker and
 8 in the percentage of CKD recycled inside the cement kiln. Uncertainty is also associated with the assumption that
 9 all calcium-containing raw materials are CaCO₃, when a small percentage likely consists of other carbonate and
 10 non-carbonate raw materials. The lime content of clinker varies from 60 to 67 percent; 65 percent is used as a
 11 representative value (Van Oss 2013a). This contributes to the uncertainty surrounding the emission factor for
 12 clinker which has an uncertainty range of ±3 percent with uniform densities (Van Oss 2013b). The amount of CO₂
 13 from CKD loss can range from 1.5 to 8 percent depending upon plant specifications, and uncertainty was estimated
 14 at ±5 percent with uniform densities (Van Oss 2013b). Additionally, some amount of CO₂ is reabsorbed when the
 15 cement is used for construction. As cement reacts with water, alkaline substances such as calcium hydroxide are
 16 formed. During this curing process, these compounds may react with CO₂ in the atmosphere to create calcium
 17 carbonate. This reaction only occurs in roughly the outer 0.2 inches of the total thickness. Because the amount of
 18 CO₂ reabsorbed is thought to be minimal, it was not estimated. EPA assigned default uncertainty bounds of ±3
 19 percent for clinker production, based on expert judgment (Van Oss 2013b).

20 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-6. Based on the
 21 uncertainties associated with total U.S. clinker production, the CO₂ emission factor for clinker production, and the
 22 emission factor for additional CO₂ emissions from CKD, 2022 CO₂ emissions from cement production were
 23 estimated to be between 39.5 and 43.1 MMT CO₂ Eq. at the 95 percent confidence level. This confidence level
 24 indicates a range of approximately 4 percent below and 4 percent above the emission estimate of 41.3 MMT CO₂
 25 Eq.

26 **Table 4-6: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Cement**
 27 **Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2021 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Cement Production	CO ₂	41.3	39.5	43.1	-4%	+4%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

28 **QA/QC and Verification**

29 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
 30 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the
 31 introduction of the IPPU chapter (see Annex 8 for more details).

32 EPA relied upon the latest guidance from the IPCC on the use of facility-level data in national inventories and
 33 applied a category-specific QC process to compare activity data from EPA’s GHGRP with existing data from USGS
 34 surveys. This was to ensure time-series consistency of the emission estimates presented in the *Inventory*. Total U.S.
 35 clinker production is assumed to have low uncertainty because facilities routinely measure this for economic
 36 reasons and because both USGS and GHGRP take multiple steps to ensure that reported totals are accurate. EPA

1 verifies annual facility-level GHGRP reports through a multi-step process that is tailored to the reporting industry
2 (e.g., combination of electronic checks including range checks, statistical checks, algorithm checks, year-to-year
3 comparison checks, along with manual reviews involving outside data checks) to identify potential errors and
4 ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015). Based on the results of the
5 verification process, EPA follows up with facilities to resolve mistakes that may have occurred.¹⁰ Facilities are also
6 required to monitor and maintain records of monthly clinker production per section 98.84 of the GHGRP regulation
7 (40 CFR 98.84).

8 EPA's GHGRP requires all facilities producing portland cement to report greenhouse gas emissions, including CO₂
9 process emissions from each kiln, CO₂ combustion emissions from each kiln, CH₄ and N₂O combustion emissions
10 from each kiln, and CO₂, CH₄, and N₂O emissions from each stationary combustion unit other than kilns (40 CFR
11 Part 98 Subpart H). Source-specific quality control measures for the cement production category are included in
12 section 98.84, Monitoring and QA/QC Requirements.

13 As mentioned above, EPA compares GHGRP clinker production data (EPA 2023) to the USGS clinker production
14 data (USGS 2023a; USGS 2023c). For the year 2014, 2020, and 2022, USGS and GHGRP clinker production data
15 showed a difference of approximately 1 percent. In 2018, the difference between USGS and GHGRP clinker
16 production data was approximately 3 percent, which resulted in a difference in emissions of about 1.2 MMT CO₂
17 Eq. In 2015, 2016, 2017, 2019, and 2021, that difference was less than 1 percent (0.1 MMT CO₂ Eq.) between the
18 two sets of activity data. The information collected by the USGS National Minerals Information Center surveys
19 continue to be an important data source.

20 Recalculations Discussion

21 No recalculations were performed for the current *Inventory*.

22 Planned Improvements

23 EPA is continuing to evaluate and analyze data reported under EPA's GHGRP that would be useful to improve the
24 emission estimates for the Cement Production source category. Most cement production facilities reporting under
25 EPA's GHGRP use Continuous Emission Monitoring Systems (CEMS) to monitor and report CO₂ emissions, thus
26 reporting combined process and combustion emissions from kilns. In implementing further improvements and
27 integration of data from EPA's GHGRP, the latest guidance from the IPCC on the use of facility-level data in national
28 inventories will be relied upon, in addition to category-specific QC methods recommended by the *2006 IPCC*
29 *Guidelines*.¹¹ EPA's long-term improvement plan includes continued assessment of the feasibility of using
30 additional GHGRP information beyond aggregation of reported facility-level clinker data, in particular
31 disaggregating the combined process and combustion emissions reported using CEMS, to separately present
32 national process and combustion emissions streams consistent with IPCC and UNFCCC guidelines. This long-term
33 planned analysis is still in development and has not been applied for this current *Inventory*.

34 EPA continues to review methods and data used to estimate CO₂ emissions from cement production in order to
35 account for organic material in the raw material and to discuss the carbonation that occurs across the duration of
36 the cement product. Work includes identifying data and studies on the average carbon content for organic
37 materials in kiln feed in the United States and on CO₂ reabsorption rates via carbonation for various cement

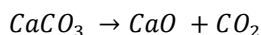
¹⁰ See GHGRP Verification Fact Sheet https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

¹¹ See IPCC Technical Bulletin on Use of Facility-Specific Data in National Greenhouse Gas Inventories http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

1 products. This information is not reported by facilities subject to GHGRP reporting. This is a long-term
2 improvement.

3 4.2 Lime Production (CRT Source Category 4 2A2)

5 Lime is an important manufactured product with many industrial, chemical, and environmental applications. Lime
6 production involves three main processes: stone preparation, calcination, and hydration. Carbon dioxide (CO₂) is
7 generated during the calcination stage, when limestone—consisting of calcium carbonate (CaCO₃) and/or
8 magnesium carbonate (MgCO₃)—is roasted at high temperatures in a kiln to produce calcium oxide (CaO) and CO₂.
9 The CO₂ is given off as a gas and is normally emitted to the atmosphere.



11 Some facilities, however, recover CO₂ generated during the production process for use in sugar refining and
12 precipitated calcium carbonate (PCC) production.¹² PCC is used as a filler or coating in the paper, food, and plastic
13 industries and is derived from reacting hydrated high-calcium quicklime with CO₂, a production process that does
14 not result in net emissions of CO₂ to the atmosphere. Emissions from fuels consumed for energy purposes during
15 the production of lime are included in the Energy chapter.

16 For U.S. operations, the term “lime” actually refers to a variety of chemical compounds. These include CaO, or
17 high-calcium quicklime; calcium hydroxide (Ca(OH)₂), or hydrated lime; dolomitic quicklime ([CaO•MgO]); and
18 dolomitic hydrate ([Ca(OH)₂•MgO] or [Ca(OH)₂•Mg(OH)₂]).

19 The current lime market is approximately distributed across six end-use categories, as follows: metallurgical uses,
20 35 percent; environmental uses, 29 percent; chemical and industrial uses, 21 percent; construction uses, 10
21 percent; miscellaneous uses, 3 percent; and refractory dolomite, 1 percent (USGS 2021). The major uses are in
22 steel making, chemical and industrial applications (such as the manufacture of fertilizer, glass, paper and pulp, and
23 precipitated calcium carbonate, and in sugar refining), flue gas desulfurization (FGD) systems at coal-fired electric
24 power plants, construction, and water treatment, as well as uses in mining, pulp and paper and precipitated
25 calcium carbonate manufacturing (USGS 2023a). Lime is also used as a CO₂ scrubber, and there has been
26 experimentation on the use of lime to capture CO₂ from electric power plants. Both lime (CaO) and limestone
27 (CaCO₃) can be used as a sorbent for FGD systems. Emissions from limestone consumption for FGD systems are
28 reported under Section 4.4 Other Process Uses of Carbonate Production (CRT Source Category 2A4).

29 Emissions from lime production have fluctuated over the time series depending on lime end-use markets –
30 primarily the steel making industry and FGD systems for utility and industrial plants – and also energy costs. One
31 significant change to lime end-use since 1990 has been the increase in demand for lime for FGD at coal-fired
32 electric power plants, which can be attributed to compliance with sulfur dioxide (SO₂) emission regulations of the
33 Clean Air Act Amendments of 1990. Phase I went into effect on January 1, 1995, followed by Phase II on January 1,
34 2000. To supply lime for the FGD market, the lime industry installed more than 1.8 million tons per year of new
35 capacity by the end of 1995 (USGS 2023a). The need for air pollution controls continued to drive the FGD lime
36 market, which had doubled between 1990 and 2019 (2021, 2023a, 2023b).

37 The U.S. lime industry temporarily shut down some individual gas-fired kilns and, in some case, entire lime plants
38 during 2000 and 2001, due to significant increases in the price of natural gas. Lime production continued to

¹² The amount of CO₂ captured for sugar refining and PCC production is reported within the CRTs under CRT Source Category 2H3 “Other”, but within this report, they are included in this chapter.

1 decrease in 2001 and 2002, a result of lower demand from the steel making industry, lime’s largest end-use
 2 market, when domestic steel producers were affected by low priced imports and slowing demand (USGS 2023a).

3 Emissions from lime production peaked in 2006 at approximately 30.3 percent above 1990 levels, due to strong
 4 demand from the steel and construction markets (road and highway construction projects), before dropping to its
 5 second lowest level in 2009 at approximately 2.5 percent below 1990 emissions, driven by the economic recession
 6 and downturn in major markets including construction, mining, and steel (USGS 2023a). In 2010, the lime industry
 7 began to recover as the steel, FGD, and construction markets also recovered (USGS 2023a). Fluctuation in lime
 8 production since 2015 has been driven largely by demand from the steel making industry (USGS 2021). In 2020, a
 9 significant decline in lime production occurred due to plants temporarily closing as a result of the global COVID-19
 10 pandemic (USGS 2023a). This resulted in the lowest level of emissions in 2020 at approximately 3.4 percent below
 11 1990 emissions. Emissions increased annually since then, with 2022 levels similar to emissions in 2019.

12 Lime production in the United States—including Puerto Rico—was reported to be 16,994 kilotons in 2022, an
 13 increase of about 1.3 percent compared to 2021 levels (USGS 2023a). Compared to 1990, lime production
 14 increased by about 7.3 percent. At year-end 2022, 73 primary lime plants were operating in the United States,
 15 including Puerto Rico according to the USGS MCS (USGS 2023a).¹³ Principal lime producing states were, in
 16 alphabetical order, Kentucky, Missouri, Ohio, and Texas (USGS 2023a).

17 U.S. lime production resulted in estimated net CO₂ emissions of 12.2 MMT CO₂ Eq. (12,208 kt) (see Table 4-7 and
 18 Table 4-8). Carbon dioxide emissions from lime production increased by about 2.8 percent compared to 2021
 19 levels. Compared to 1990, CO₂ emissions have increased by about 4.3 percent. The trends in CO₂ emissions from
 20 lime production are directly proportional to trends in production, which are described above.

21 **Table 4-7: CO₂ Emissions from Lime Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Lime Production	11.7	14.6	13.1	12.1	11.3	11.9	12.2

22 **Table 4-8: Gross, Recovered, and Net CO₂ Emissions from Lime Production (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Gross	11,959	15,074	13,609	12,676	11,875	12,586	12,750
Recovered ^a	259	522	503	564	576	716	542
Net Emissions	11,700	14,552	13,106	12,112	11,299	11,870	12,208

Note: Totals may not sum due to independent rounding.

^a For sugar refining and PCC production.

23 Methodology and Time-Series Consistency

24 To calculate emissions, the amounts of high-calcium and dolomitic lime produced were multiplied by their
 25 respective emission factors, consistent with Tier 2 methodology from the *2006 IPCC Guidelines* and in accordance
 26 with the IPCC methodological decision tree and available data. The emission factor is the product of the
 27 stoichiometric ratio between CO₂ and CaO, and the average CaO and MgO content for lime. The CaO and MgO
 28 content for lime is assumed to be 95 percent for both high-calcium and dolomitic lime (IPCC 2006). The emission
 29 factors were calculated as follows:

¹³ In 2022, 68 operating primary lime facilities in the United States reported to the EPA Greenhouse Gas Reporting Program.

1 **Equation 4-2: 2006 IPCC Guidelines Tier 2 Emission Factor for Lime Production, High-Calcium**
2 **Lime (Equation 2.9)**

3
$$EF_{\text{High-Calcium Lime}} = \left[\left(44.01 \frac{\text{g}}{\text{mole}} \text{CO}_2 \right) \div \left(56.08 \frac{\text{g}}{\text{mole}} \text{CaO} \right) \right] \times \left(0.9500 \frac{\text{CaO}}{\text{lime}} \right) = 0.7455 \frac{\text{g CO}_2}{\text{g lime}}$$

4 **Equation 4-3: 2006 IPCC Guidelines Tier 2 Emission Factor for Lime Production, Dolomitic Lime**
5 **(Equation 2.9)**

6
$$EF_{\text{Dolomitic Lime}} = \left[\left(88.02 \frac{\text{g}}{\text{mole}} \text{CO}_2 \right) \div \left(96.39 \frac{\text{g}}{\text{mole}} \text{CaO} \cdot \text{MgO} \right) \right] \times \left(0.9500 \frac{\text{CaO} \cdot \text{MgO}}{\text{lime}} \right) = 0.8675 \frac{\text{g CO}_2}{\text{g lime}}$$

7 Production was adjusted to remove the mass of chemically combined water found in hydrated lime, determined
8 according to the molecular weight ratios of H₂O to (Ca(OH)₂ and [Ca(OH)₂•Mg(OH)₂]) (IPCC 2006). These factors set
9 the chemically combined water content to 27 percent for high-calcium hydrated lime, and 30 percent for dolomitic
10 hydrated lime.

11 The *2006 IPCC Guidelines* (Tier 2 method) also recommends accounting for emissions from lime kiln dust (LKD)
12 through application of a correction factor. LKD is a byproduct of the lime manufacturing process typically not
13 recycled back to kilns. LKD is a very fine-grained material and is especially useful for applications requiring very
14 small particle size. Most common LKD applications include soil reclamation and agriculture. Emissions from the
15 application of lime for agricultural purposes are reported in the Agriculture chapter under 5.5 Liming (CRT Source
16 Category 3G). Currently, data on annual LKD production is not readily available to develop a country-specific
17 correction factor. Lime emission estimates were multiplied by a factor of 1.02 to account for emissions from LKD
18 (IPCC 2006). See the Planned Improvements section associated with efforts to improve uncertainty analysis and
19 emission estimates associated with LKD.

20 Lime emission estimates were further adjusted to account for the amount of CO₂ captured for use in on-site
21 processes. All the domestic lime facilities are required to report these data to EPA under its GHGRP. The total
22 national-level annual amount of CO₂ captured for on-site process use was obtained from EPA's GHGRP (EPA 2023)
23 based on reported facility-level data for years 2010 through 2022. The amount of CO₂ captured/recovered for non-
24 marketed on-site process use is deducted from the total gross emissions (i.e., from lime production and LKD). The
25 net lime emissions are presented in Table 4-7 and Table 4-8. GHGRP data on CO₂ removals (i.e., CO₂
26 captured/recovered) was available only for 2010 through 2022. Since GHGRP data are not available for 1990
27 through 2009, IPCC "splicing" techniques were used as per the *2006 IPCC Guidelines* on time-series consistency
28 (IPCC 2006, Volume 1, Chapter 5).

29 Lime production data (i.e., lime sold and non-marketed lime used by the producer) by type (i.e., high-calcium and
30 dolomitic quicklime, high-calcium and dolomitic hydrated lime, and dead-burned dolomite) for 1990 through 2022
31 (see Table 4-9) were obtained from U.S. Geological Survey (USGS) Minerals Yearbook (USGS 2023a) and are
32 compiled by USGS to the nearest ton. Dead-burned dolomite data are additionally rounded by USGS to no more
33 than one significant digit to avoid disclosing company proprietary data. Production data for the individual
34 quicklime (i.e., high-calcium and dolomitic) and hydrated lime (i.e., high-calcium and dolomitic) types were not
35 provided prior to 1997. These were calculated based on total quicklime and hydrated lime production data from
36 1990 through 1996 and the three-year average ratio of the individual lime types from 1997 to 1999. Natural
37 hydraulic lime, which is produced from CaO and hydraulic calcium silicates, is not manufactured in the United
38 States (USGS 2023a). Total lime production was adjusted to account for the water content of hydrated lime by
39 converting hydrate to oxide equivalent based on recommendations from the IPCC and using the water content
40 values for high-calcium hydrated lime and dolomitic hydrated lime mentioned above, and is presented in Table
41 4-10 (IPCC 2006). The CaO and CaO•MgO contents of lime, both 95 percent, were obtained from the IPCC (IPCC
42 2006).

1 **Table 4-9: High-Calcium- and Dolomitic-Quicklime, High-Calcium- and Dolomitic-Hydrated,**
 2 **and Dead-Burned-Dolomite Lime Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
High-Calcium Quicklime	11,166	14,100	12,400	11,300	10,700	11,200	11,500
Dolomitic Quicklime	2,234	2,990	2,810	2,700	2,390	2,700	2,640
High-Calcium Hydrated	1,781	2,220	2,430	2,430	2,320	2,430	2,410
Dolomitic Hydrated	319	474	265	267	252	244	244
Dead-Burned Dolomite	342	200	200	200	200	200	200

3 **Table 4-10: Adjusted Lime Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
High-Calcium	12,466	15,721	14,174	13,074	12,394	12,974	13,259
Dolomitic	2,800	3,522	3,196	3,087	2,766	3,071	3,011

Note: Minus water content of hydrated lime.

4 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
 5 through 2022.

6 **Uncertainty– TO BE UPDATED FOR FINAL REPORT**

7 The uncertainties contained in these estimates can be attributed to slight differences in the chemical composition
 8 of lime products and CO₂ recovery rates for on-site process use over the time series. Although the methodology
 9 accounts for various formulations of lime, it does not account for the trace impurities found in lime, such as iron
 10 oxide, alumina, and silica. Due to differences in the limestone used as a raw material, a rigid specification of lime
 11 material is impossible. As a result, few plants produce lime with exactly the same properties.

12 In addition, a portion of the CO₂ emitted during lime production will actually be reabsorbed when the lime is
 13 consumed, especially at captive lime production facilities. As noted above, lime has many different chemical,
 14 industrial, environmental, and construction applications. In many processes, CO₂ reacts with the lime to create
 15 calcium carbonate (e.g., water softening). Carbon dioxide reabsorption rates vary, however, depending on the
 16 application. For example, 100 percent of the lime used to produce precipitated calcium carbonate reacts with CO₂,
 17 whereas most of the lime used in steel making reacts with impurities such as silica, sulfur, and aluminum
 18 compounds. Quantifying the amount of CO₂ that is reabsorbed would require a detailed accounting of lime use in
 19 the United States and additional information about the associated processes where both the lime and byproduct
 20 CO₂ are “reused.” Research conducted thus far has not yielded the necessary information to quantify CO₂
 21 reabsorption rates.¹⁴ Some additional information on the amount of CO₂ consumed on site at lime facilities,
 22 however, has been obtained from EPA’s GHGRP.

23 In some cases, lime is generated from calcium carbonate byproducts at pulp mills and water treatment plants.¹⁵
 24 The lime generated by these processes is included in the USGS data for commercial lime consumption. In the
 25 pulping industry, mostly using the Kraft (sulfate) pulping process, lime is consumed in order to causticize a process
 26 liquor (green liquor) composed of sodium carbonate and sodium sulfide. The green liquor results from the dilution
 27 of the smelt created by combustion of the black liquor where biogenic carbon (C) is present from the wood. Kraft

¹⁴ Representatives of the National Lime Association estimate that CO₂ reabsorption that occurs from the use of lime may offset as much as a quarter of the CO₂ emissions from calcination (Males 2003).

¹⁵ Some carbide producers may also regenerate lime from their calcium hydroxide byproducts, which does not result in emissions of CO₂. In making calcium carbide, quicklime is mixed with coke and heated in electric furnaces. The regeneration of lime in this process is done using a waste calcium hydroxide (hydrated lime) [CaC₂ + 2H₂O → C₂H₂ + Ca(OH)₂], not calcium carbonate [CaCO₃]. Thus, the calcium hydroxide is heated in the kiln to simply expel the water [Ca(OH)₂ + heat → CaO + H₂O], and no CO₂ is released.

mills recover the calcium carbonate “mud” after the causticizing operation and calcine it back into lime—thereby generating CO₂—for reuse in the pulping process. Although this re-generation of lime could be considered a lime manufacturing process, the CO₂ emitted during this process is mostly biogenic in origin and therefore is not included in the industrial processes totals (Miner and Upton 2002). In accordance with IPCC methodological guidelines, any such emissions are calculated by accounting for net carbon fluxes from changes in biogenic carbon reservoirs in wooded or crop lands (see the Land Use, Land-Use Change, and Forestry chapter).

In the case of water treatment plants, lime is used in the softening process. Some large water treatment plants may recover their waste calcium carbonate and calcine it into quicklime for reuse in the softening process. Further research is necessary to determine the degree to which lime recycling is practiced by water treatment plants in the United States.

Another uncertainty is the assumption that calcination emissions for LKD are around 2 percent. EPA assigned uncertainty ranges of ±2 percent for the LKD correction factor based on expert judgment (RTI 2023). The National Lime Association (NLA) has commented that the estimates of emissions from LKD in the United States could be closer to 6 percent. They also note that additional emissions (approximately 2 percent) may also be generated through production of other byproducts/wastes (off-spec lime that is not recycled, scrubber sludge) at lime plants (Seeger 2013). Publicly available data on LKD generation rates, total quantities not used in cement production, and types of other byproducts/wastes produced at lime facilities are limited. NLA compiled and shared historical emissions information and quantities for some waste products reported by member facilities associated with generation of total calcined byproducts and LKD, as well as methodology and calculation worksheets that member facilities complete when reporting. There is uncertainty regarding the availability of data across the time series needed to generate a representative country-specific LKD factor. Uncertainty of the activity data is also a function of the reliability and completeness of voluntarily reported plant-level production data. EPA assigned uncertainty ranges of ±1 percent for lime production, based on expert judgment (USGS 2012). Further research, including discussion with NLA and data is needed to improve understanding of additional calcination emissions to consider revising the current assumptions that are based on the *2006 IPCC Guidelines*. More information can be found in the Planned Improvements section below.

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-11. Lime CO₂ emissions for 2022 were estimated to be between 11.6 and 12.1 MMT CO₂ Eq. at the 95 percent confidence level. This confidence level indicates a range of approximately 2 percent below and 2 percent above the emission estimate of 11.9 MMT CO₂ Eq.

Table 4-11: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Lime Production (MMT CO₂ Eq. and Percent)

Source	Gas	2021 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
Lime Production	CO ₂	11.9	11.6	12.1	-2%	+2%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as noted in the introduction of the IPPU chapter (see Annex 8 for more details).

1 More details on the greenhouse gas calculation, monitoring and QA/QC methods associated with reporting on CO₂
2 captured for onsite use applicable to lime manufacturing facilities can be found under Subpart S (lime
3 manufacturing) of the GHGRP regulation (40 CFR Part 98).¹⁶ EPA verifies annual facility-level GHGRP reports
4 through a multi-step process (e.g., combination of electronic checks and manual reviews) to identify potential
5 errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2023).¹⁷ Based on the
6 results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The
7 post-submittals checks are consistent with a number of general and category-specific QC procedures, including:
8 range checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.

9 Recalculations Discussion

10 No recalculations were performed for the 1990 through 2021 portion of the time series.

11 Planned Improvements

12 EPA plans to review GHGRP emissions and activity data reported to EPA under Subpart S of the GHGRP regulation
13 (40 CFR Part 98), and aggregated activity data on lime production by type in particular. In addition, initial review of
14 data has identified that several facilities use CEMS to report emissions. Under Subpart S, if a facility is using a
15 CEMS, they are required to report combined combustion emissions and process emissions. EPA continues to
16 review how best to incorporate GHGRP and notes that particular attention will be made to also ensuring time-
17 series consistency of the emissions estimates presented in future *Inventory* reports, consistent with IPCC and
18 UNFCCC guidelines. This is required because the facility-level reporting data from EPA's GHGRP, with the program's
19 initial requirements for reporting of emissions in calendar year 2010, are not available for all inventory years (i.e.,
20 1990 through 2009) as required for this *Inventory*. In implementing improvements and integration of data from
21 EPA's GHGRP, the latest guidance from the IPCC on the use of facility-level data in national inventories will be
22 relied upon.¹⁸

23 Future improvements involve improving and/or confirming the representativeness of current assumptions
24 associated with emissions from production of LKD and other byproducts/wastes as discussed in the Uncertainty
25 section, per comments from the NLA provided during a prior Public Review comment period for a previous (1990
26 through 2018) *Inventory*. EPA met with NLA in summer of 2020 for clarification on data needs and available data
27 and to discuss planned research into GHGRP data. Previously, EPA met with NLA in spring of 2015 to outline
28 specific information required to apply IPCC methods to develop a country-specific correction factor to more
29 accurately estimate emissions from production of LKD. In 2016, NLA compiled and shared historical emissions
30 information reported by member facilities on an annual basis under voluntary reporting initiatives from 2002
31 through 2011 associated with generation of total calcined byproducts and LKD. Reporting of LKD was only
32 differentiated for the years 2010 and 2011. This emissions information was reported on a voluntary basis
33 consistent with NLA's facility-level reporting protocol, which was also provided to EPA. To reflect information
34 provided by NLA, EPA updated the qualitative description of uncertainty. At the time of this *Inventory*, this planned
35 improvement is in process and has not been incorporated into this current *Inventory* report.

¹⁶ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

¹⁷ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

¹⁸ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

4.3 Glass Production (CRT Source Category 2A3)

Glass production is an energy and raw-material intensive process that results in the generation of carbon dioxide (CO₂) from both the energy consumed in making glass and the glass production process itself. Emissions from fuels consumed for energy purposes during the production of glass are included in the Energy sector.

Glass production employs a variety of raw materials in a glass-batch. These include formers, fluxes, stabilizers, and sometimes colorants. The major raw materials (i.e., fluxes and stabilizers) that emit process-related CO₂ emissions during the glass melting process are limestone, dolomite, and soda ash. The main former in all types of glass is silica (SiO₂). Other major formers in glass include feldspar and boric acid (i.e., borax). Fluxes are added to lower the temperature at which the batch melts. Most commonly used flux materials are soda ash (sodium carbonate, Na₂CO₃) and potash (potassium carbonate, K₂O). Stabilizers make glass more chemically stable and keep the finished glass from dissolving and/or falling apart. Commonly used stabilizing agents in glass production are limestone (CaCO₃), dolomite (CaCO₃MgCO₃), alumina (Al₂O₃), magnesia (MgO), barium carbonate (BaCO₃), strontium carbonate (SrCO₃), lithium carbonate (Li₂CO₃), and zirconia (ZrO₂) (DOE 2002). Glass makers also use a certain amount of recycled scrap glass (cullet), which comes from in-house return of glassware broken in the production process or other glass spillage or retention, such as recycling or from cullet broker services.

The raw materials (primarily soda ash, limestone, and dolomite) release CO₂ emissions in a complex high-temperature chemical reaction during the glass melting process. This process is not directly comparable to the calcination process used in lime manufacturing, cement manufacturing, and process uses of carbonates (i.e., limestone/dolomite use) but has the same net effect in terms of generating process CO₂ emissions (IPCC 2006).

The U.S. glass industry can be divided into four main categories: containers, flat (window) glass, fiber glass, and specialty glass. The majority of commercial glass produced is container and flat glass (EPA 2009). The United States is one of the major global exporters of glass. Domestically, demand comes mainly from the construction, auto, bottling, and container industries. There are more than 1,700 facilities that manufacture glass in the United States, with the largest companies being Corning, Guardian Industries, Owens-Illinois, and PPG Industries.¹⁹

The glass container sector is one of the leading soda ash consuming sectors in the United States. In 2022, glass production accounted for 48 percent of total domestic soda ash consumption (USGS 2023). Emissions from soda ash production are reported in Section 4.12 soda ash production (CRT Source Category 2B7).

In 2022, 2,250 kilotons of soda ash, 1,370 kilotons of limestone, 925 kilotons of dolomite, and 1.9 kilotons of other carbonates were consumed for glass production (USGS 2023; EPA 2023). Use of soda ash, limestone, dolomite, and other carbonates in glass production resulted in aggregate CO₂ emissions of 2.0 MMT CO₂ Eq. (1,956 kt), which are summarized in Table 4-12 and Table 4-13. Overall, emissions have decreased by 14 percent compared to 1990. Emissions decreased by 1 percent compared to 2021 levels.

Emissions from glass production have remained relatively consistent over the time series with some fluctuations since 1990. In general, these fluctuations were related to the behavior of the export market and the U.S. economy. Specifically, the extended downturn in residential and commercial construction and automotive industries between 2008 and 2010 resulted in reduced consumption of glass products, causing a drop in global demand for limestone/dolomite and soda ash and resulting in lower emissions. Some commercial food and beverage package manufacturers are shifting from glass containers towards lighter and more cost-effective polyethylene terephthalate (PET) based containers, putting downward pressure on domestic consumption of soda ash (USGS

¹⁹ Excerpt from Glass & Glass Product Manufacturing Industry Profile, First Research. Available online at: <http://www.firstresearch.com/Industry-Research/Glass-and-Glass-Product-Manufacturing.html>.

1 1995 through 2015b). Glass production in 2022 was steady, changing by no more than 2 percent over the course of
2 the year (Federal Reserve 2023).

3 **Table 4-12: CO₂ Emissions from Glass Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Glass Production	2.3	2.4	2.0	1.9	1.9	2.0	2.0

4 **Table 4-13: CO₂ Emissions from Glass Production (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Glass Production	2,263	2,402	1,989	1,940	1,858	1,969	1,956

5 Methodology and Time-Series Consistency

6 Carbon dioxide emissions were calculated based on the 2006 IPCC Guidelines Tier 3 method, in accordance with
7 the IPCC methodological decision tree and available data, by multiplying the quantity of input carbonates (i.e.,
8 limestone, dolomite, soda ash, and other carbonates) by the carbonate-based emission factor (in metric tons
9 CO₂/metric ton carbonate) and the average carbonate-based mineral mass fraction.

10 2010 through 2022

11 The methodology for estimating CO₂ emissions from glass production for years 2010 through 2022 used the
12 quantities of limestone, dolomite, and a group of other carbonates (i.e., barium carbonate, potassium carbonate,
13 lithium carbonate, and strontium carbonate) used for glass production, obtained from GHGRP (EPA 2023). USGS
14 data on the quantity of soda ash used for glass production was used because it was obtained directly from the
15 soda ash producers and includes use by smaller artisanal glass operations, which are excluded in the GHGRP data.

16 GHGRP collects data from glass production facilities with greenhouse gas emissions greater than 25,000 metric
17 tons CO₂ Eq. The reporting threshold is used to exclude artisanal glass operations that are expected to have much
18 lower greenhouse gas emissions than the threshold. These smaller facilities have not been accounted for yet for
19 this portion of the time series for limestone, dolomite, or other carbonates due to limited data. Facilities report the
20 total quantity of each type of carbonate used in glass production each year to GHGRP, with data collection starting
21 in 2010 (EPA 2023).

22 Using the total quantities of each carbonate, EPA calculated the metric tons of emissions resulting from glass
23 production by multiplying the quantity of input carbonates (i.e., limestone, dolomite, soda ash, and other
24 carbonates) by carbonate-based emission factors in metric tons CO₂/metric ton carbonate (limestone, 0.43971;
25 dolomite, 0.47732; soda ash, 0.41492; and other carbonates, 0.262), and by the average carbonate-based mineral
26 mass fraction for each year. IPCC default emission factors were used for limestone, dolomite, and soda ash, and
27 the emission factor for other carbonates is based on expert judgment (RTI 2022).

28 1990 through 2009

29 Data from GHGRP on the quantity of limestone, dolomite, and other carbonates used in glass production are not
30 available for 1990 through 2009. Additionally, USGS does not collect data on the quantity of other carbonates used
31 for glass production.

32 To address time-series consistency, total emissions from 1990 to 2009 were calculated using the Federal Reserve
33 Industrial Production Index for glass production in the United States as a surrogate for the total quantity of
34 carbonates used in glass production. The production index measures real output expressed as a percentage of real
35 output in a base year, which is currently 2017 (Federal Reserve 2023). Since January 1971, the Federal Reserve has
36 released the monthly glass production index for NAICS code 3272 (Glass and Glass Product Manufacturing) as part
37 of release G.17, "Industrial Production and Capacity Utilization" (Federal Reserve 2023). The monthly index values
38 for each year were averaged to calculate an average annual glass production index value. Total annual process
39 emissions were calculated by taking a ratio of the average annual glass production index for each year to the

1 average annual glass production index for base year 2017, and multiplying by the calculated 2017 emissions
 2 (process-related) based on GHGRP data.

3 Emissions from limestone, dolomite, and other carbonate consumption were disaggregated from total annual
 4 emissions, using the average percent contribution of each to annual emissions from these three carbonates for
 5 2010 through 2014 based on GHGRP data: 64.5 percent limestone, 35.5 percent dolomite, and 0.1 percent other
 6 carbonates.

7 The methodology for estimating CO₂ emissions from the use of soda ash for glass production and data sources for
 8 the amount of soda ash used in glass production are consistent with the methodology used for 2010 through 2022.
 9 The average mineral mass fractions for soda ash are only available starting in 2010. The average carbonate-based
 10 mineral mass fractions from the GHGRP, averaged across 2010 through 2014, indicate that soda ash contained
 11 98.7 percent sodium carbonate (Na₂CO₃). This averaged value is used to estimate emissions for 1990 through 2009.
 12 The years 2010 to 2014 were used to determine the average carbonate-based mineral mass fractions because
 13 those years were deemed to better represent historic glass production from 1990 to 2009.

14 Data on soda ash used for glass production for 1990 through 2022 were obtained from the U.S. Bureau of Mines
 15 (1991 and 1993a), the USGS *Minerals Yearbook: Soda Ash* (USGS 1995 through 2015b), and USGS *Mineral Industry*
 16 *Surveys for Soda Ash* (USGS 2017 through 2023). Data on limestone, dolomite, and other carbonates used for glass
 17 production and on average carbonate-based mineral mass fraction for 2010 through 2022 were obtained from
 18 GHGRP (EPA 2023). The quantities of limestone, dolomite, and other carbonates were calculated for 1990 through
 19 2009 using the Federal Reserve Industrial Production Index (Federal Reserve 2023).

20 The amount of limestone, dolomite, soda ash, and other carbonates used in glass production each year and the
 21 annual average Federal Reserve production indices for glass production are shown in Table 4-14.

22 **Table 4-14: Limestone, Dolomite, Soda Ash, and Other Carbonates Used in Glass Production**
 23 **(kt) and Average Annual Production Index for Glass and Glass Product Manufacturing**

Activity	1990	2005	2018	2019	2020	2021	2022
Limestone	1,409	1,690	1,442	1,370	1,334	1,397	1,370
Dolomite	714	857	871	883	824	893	925
Soda Ash	3,177	3,050	2,280	2,220	2,130	2,280	2,250
Other Carbonates	2	3	2	2	2	2	1.9
Total	5,302	5,599	4,596	4,475	4,289	4,572	4,547
Production Index ^a	94.3	113.1	102.5	99.8	92.4	88.3	86.8

^a Average Annual Production Index uses 2017 as the base year.

Note: Totals may not sum due to independent rounding.

24 As discussed above, methodological approaches were applied to the entire time series to ensure consistency in
 25 emissions from 1990 through 2022. Consistent with the *2006 IPCC Guidelines*, the overlap technique was applied
 26 to compare USGS and GHGRP data sets for 2010 through 2022. To address the inconsistencies, adjustments were
 27 made as described above.

28 **Uncertainty– TO BE UPDATED FOR FINAL REPORT**

29 The methodology in this *Inventory* report uses GHGRP data for the average mass fraction of each mineral used in
 30 glass production. These minerals are limestone, dolomite, soda ash, and other carbonates (barium carbonate
 31 (BaCO₃), potassium carbonate (K₂CO₃), lithium carbonate (Li₂CO₃), and strontium carbonate (SrCO₃)). The mass
 32 fractions are reported directly by the glass manufacturers, for each year from 2010 to 2022.

33 The methodology uses the quantities of limestone, dolomite, and other carbonates used in glass manufacturing
 34 which is reported directly by the glass manufacturers for years 2010 through 2022 and the amount of soda ash
 35 used in glass manufacturing which is reported by soda ash producers for the full time series. EPA assigned an
 36 uncertainty range of ±5 percent for all carbonate quantities and the Federal Reserve Industrial Production Index

for glass production, and using this suggested uncertainty provided in Section 2.4.2.2 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023). EPA assigned an uncertainty range of ± 2 percent for the carbonate emission factors, ± 2 percent for the mineral mass fractions, and ± 1 percent for the calcination fraction, and using this suggested uncertainty provided in Section 2.4.2.1 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023).

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-15. In 2022, glass production CO₂ emissions were estimated to be between 1.9 and 2.0 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 3 percent below and 3 percent above the emission estimate of 2.0 MMT CO₂ Eq.

Table 4-15: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Glass Production (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
Glass Production	CO ₂	2.0	1.9	2.0	-3%	+3%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details). For the GHGRP data, EPA verifies annual facility-level reports through a multi-step process (e.g., including a combination of pre-and post-submittal electronic checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).²⁰ Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including: range checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.

Recalculations Discussion

During annual QC, a transcription error for the 1990 value of CO₂ emissions from glass production was identified and corrected in Table 4-12 and Table 4-13. No recalculations were performed due to this transcription error and no other errors were identified for the 1990 through 2021 portion of the time series.

Planned Improvements

EPA plans to evaluate updates to uncertainty levels for the activity data and mineral mass fraction values from EPA's GHGRP. This is a near-term planned improvement.

Some glass producing facilities in the United States do not report to EPA's GHGRP because they fall below the reporting threshold for this industry. EPA will continue ongoing research on the availability of data to better assess the completeness of emission estimates from glass production and how to refine the methodology to ensure

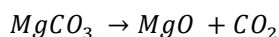
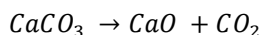
²⁰ GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 complete national coverage of this category. When reporting began in 2010, EPA received data from more facilities
2 that were above the reporting threshold than expected, and total emissions for these reporting facilities were
3 higher than expected for all glass production facilities in the United States (EPA 2009). Research will include
4 reassessing previous assessments of GHGRP industry coverage using the reporting threshold of 25,000 metric tons
5 CO₂ Eq. This is a medium-term planned improvement.

6 4.4 Other Process Uses of Carbonates (CRT 7 Source Category 2A4)

8 Limestone (CaCO₃), dolomite (CaCO₃MgCO₃),²¹ and other carbonates such as soda ash, magnesite, and siderite are
9 basic materials used by a wide variety of industries, including construction, agriculture, chemical, metallurgy, glass
10 production, ceramics production, and environmental pollution control. This section addresses mineral industry use
11 of these carbonates: limestone, dolomite, soda ash, and magnesite. Emissions from the use of these carbonates are
12 organized into four subcategories: other process uses of carbonates (i.e., limestone and dolomite consumption),
13 other uses of soda ash, ceramics production, and non-metallurgical magnesia production.

14 For industrial applications, carbonates are heated sufficiently enough to calcine the material and generate CO₂ as a
15 byproduct.



18 Examples of such applications include limestone used as a flux or purifier in metallurgical furnaces, as a sorbent in
19 flue gas desulfurization (FGD) systems for utility and industrial plants, and as a raw material for the production of
20 glass, lime, and cement.

21 Emissions from limestone and dolomite used in the production of cement, lime, glass, and iron and steel are
22 excluded from the other process uses of carbonates category and reported under their respective source categories
23 (e.g., Section 4.3, glass production). Emissions from soda ash production are reported under Section 4.12, soda ash
24 production (CRT Source Category 2B7). Emissions from soda ash consumption associated with glass manufacturing
25 are reported under Section 4.3, glass production (CRT Source Category 2A3). Emissions from the use of limestone
26 and dolomite in liming of agricultural soils are included in the Agriculture chapter under Section 5.5, liming (CRT
27 Source Category 3G). Emissions from fuels consumed for energy purposes during these processes are accounted for
28 in the Energy chapter under Section 3.1, fossil fuel combustion (CRT Source Category 1A). Both lime (CaO) and
29 limestone (CaCO₃) can be used as a sorbent for FGD systems. Emissions from lime consumption for FGD systems
30 and from sugar refining are reported under Section 4.3, lime production (CRT Source Category 2A2). Emissions
31 from the use of dolomite in primary magnesium metal production are reported under Section 4.20, magnesium
32 production and processing (CRT Source Category 2C4).

33 Limestone and dolomite are widely distributed throughout the world in deposits of varying sizes and degrees of
34 purity. Large deposits of limestone occur in nearly every state in the United States, and significant quantities are
35 extracted for industrial applications. In 2018, the leading limestone producing states were Texas, Florida, Ohio,
36 Missouri, and Pennsylvania, which contributed 46 percent of the total U.S. output (USGS 2022a). Dolomite deposits
37 are found in the United States, Canada, Mexico, Europe, Africa, and Brazil. In the United States, the leading dolomite

²¹ Limestone and dolomite are collectively referred to as limestone by the industry, and intermediate varieties are seldom distinguished.

1 producing states are Pennsylvania, New York, and Utah which currently contribute more than a third of the total
2 U.S. output (USGS 2022a).

3 Internationally, two types of soda ash are produced: natural and synthetic. In 2019, 93 percent of the global soda
4 ash production came from China, the United States, Russia, Germany, India, Turkey, Poland, and France. The United
5 States only produces natural soda ash and only in two states: Wyoming and California (USGS 2021c).

6 Ceramics include the production of bricks and roof tiles, vitrified clay pipes, refractory products, expanded clay
7 products, wall and floor tiles, table and ornamental ware (i.e., household ceramics), sanitary ware, technical
8 ceramics (e.g., aerospace, automotive, electronic, or biomedical applications), and inorganic bonded abrasives. Most
9 ceramic products are made from one or more different types of clay (e.g., shales, fire clay, and ball clay) with
10 varying carbonate contents. The process of manufacturing ceramic products, regardless of the product type or
11 scale, is essentially the same. This process consists of raw material processing (grinding, calcining, and drying),
12 forming (wet or dry process), firing (single or multiple stage firing process), and final processing. Process CO₂
13 emissions are produced during the calcination process in the kiln or dryer, where carbonates are heated to high
14 temperatures which results in metal oxides and CO₂. In 2018, the leading clay producing states were Georgia,
15 Wyoming, Texas, Alabama, and North Carolina, which contributed 60 percent of the total U.S. output (USGS 2022e).

16 Non-metallurgical magnesia production comprises of three categories of magnesia products: calcined magnesia,
17 deadburned magnesia, and fused magnesia. Magnesia is produced by calcining magnesite (MgCO₃) which results in
18 the release of CO₂. Non-metallurgical magnesia is used in agricultural, industrial, refractory, and electrical insulating
19 applications. Specific applications include fertilizers, construction materials, plastics, and flue gas desulphurization.
20 China, Russia, and Turkey account for 83 percent of global production capacity of magnesia from magnesite (USGS
21 2022f). In the United States, only one facility located in Nevada produces non-metallurgical magnesia using
22 magnesite as the raw material.

23 In 2022, 18,659 kilotons (kt) of limestone, 2,050 kt of dolomite, 2,391 kt of soda ash, and 388 kt of magnesite were
24 consumed for these emissive applications, which excludes consumption for the production of cement, lime, glass,
25 and iron and steel (Willett 2023, USGS 2022b). Usage of limestone, dolomite, soda ash, and magnesite resulted in
26 aggregate CO₂ emissions of 10.4 MMT CO₂ Eq. (10,378 kt) (see Table 4-16 and Table 4-17). The 2022 emissions
27 increased 21 percent compared to 2021, primarily as a result of increased limestone consumption attributed to
28 sulfur oxide removal. Growth in the public and private construction markets contributed to an increase in
29 consumption of crushed stone in 2022. Overall emissions have increased 46 percent from 1990 through 2022.

30 **Table 4-16: CO₂ Emissions from Other Process Uses of Carbonates (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Other Uses of Carbonates	4.8	6.2	6.3	7.4	7.4	7.0	8.8
Other Uses of Soda Ash ^a	1.4	1.3	1.1	1.0	1.0	1.0	1.0
Ceramics Production	0.8	0.8	0.4	0.4	0.4	0.4	0.4
Non-Metallurgical Magnesia Production	0.1	0.2	0.2	0.2	0.2	0.2	0.2
Total	7.1	8.5	7.9	9.0	9.0	8.6	10.4

^a Soda ash consumption not associated with glass manufacturing.

Note: Totals may not sum due to independent rounding.

31 **Table 4-17: CO₂ Emissions from Other Process Uses of Carbonates (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Other Uses of Carbonates	4,843	6,155	6,283	7,386	7,441	6,972	8,781
Other Uses of Soda Ash ^a	1,390	1,305	1,069	1,036	958	979	992
Ceramics Production	757	822	414	395	397	401	401
Non-Metallurgical Magnesia Production	113	191	169	152	216	231	204
Total	7,103	8,472	7,935	8,969	9,012	8,583	10,378

^a Soda ash consumption not associated with glass manufacturing.

Note: Totals may not sum due to independent rounding.

1 Methodology and Time-Series Consistency

2 Other Uses of Carbonates (Limestone and Dolomite Consumption)

3 Carbon dioxide emissions from other uses of carbonates, specifically limestone and dolomite consumption, were
4 calculated using a Tier 2 method from the *2006 IPCC Guidelines*, in accordance with the IPCC methodological
5 decision tree and available data, by multiplying the quantity of limestone or dolomite consumed by the emission
6 factor for limestone or dolomite calcination, respectively: 0.43971 metric ton CO₂/metric ton carbonate for
7 limestone and 0.47732 metric ton CO₂/metric ton carbonate for dolomite.²² This methodology was used for flux
8 stone, flue gas desulfurization systems, chemical stone, mine dusting or acid water treatment, and acid
9 neutralization. Flux stone used during the production of iron and steel was deducted from the other process uses
10 of carbonates source category estimate and attributed to the iron and steel production source category estimate.
11 Similarly, limestone and dolomite consumption for glass manufacturing, cement, and lime manufacturing are
12 excluded from this category and attributed to their respective categories.

13 Consumption data for 1990 through 2022 of limestone and dolomite used for flux stone, flue gas desulfurization
14 systems, chemical stone, mine dusting or acid water treatment, and acid neutralization (see Table 4-18) were
15 obtained from the U.S. Geological Survey (USGS) *Minerals Yearbook: Crushed Stone Annual Report* (1995a through
16 2023a), preliminary data for 2022 from USGS Crushed Stone Commodity Expert (Willett 2023), American Iron and
17 Steel Institute limestone and dolomite consumption data (AISI 2018 through 2021), and the U.S. Bureau of Mines
18 (1991 and 1993a), which are reported to the nearest ton. In addition, the estimated values for limestone and
19 dolomite consumption for flux stone used during the production of iron and steel were adjusted using emissions
20 data from the EPA's Greenhouse Gas Reporting Program (GHGRP) subpart Q for the iron and steel sector to
21 account for the impacts of the COVID-19 pandemic in 2020 and 2021. Iron and steel GHGRP process emissions data
22 decreased by approximately 8 percent from 2021 to 2022 (EPA 2023). This adjustment method is consistent with
23 the method used in Section 4.17 iron and steel production (CRT Source Category 2C1) and metallurgical coke
24 production.

25 During 1990 and 1992, the USGS did not conduct a detailed survey of limestone and dolomite consumption by
26 end-use; therefore, data on consumption by end use for 1990 was estimated by applying the 1991 ratios of total
27 limestone and dolomite consumption by end use to total 1990 limestone and dolomite consumption values.
28 Similarly, the 1992 consumption figures were approximated by applying an average of the 1991 and 1993 ratios of
29 total limestone and dolomite use by end uses to the 1992 total values.

30 In 1991, the U.S. Bureau of Mines, now known as the USGS, began compiling production and end use information
31 through surveys of crushed stone manufacturers. Manufacturers provided different levels of detail in survey
32 responses, so information was divided into three categories: (1) production by end-use, as reported by
33 manufacturers (i.e., "specified" production); (2) production reported by manufacturers without end-uses specified
34 (i.e., "unspecified-reported" production); and (3) estimated additional production by manufacturers who did not
35 respond to the survey (i.e., "unspecified-estimated" production). Additionally, each year the USGS withholds data
36 on certain limestone and dolomite end-uses due to confidentiality agreements regarding company proprietary
37 data. For the purposes of this analysis, emissive end-uses that contained withheld data were estimated using one
38 of the following techniques: (1) the value for all the withheld data points for limestone or dolomite use was
39 distributed evenly to all withheld end-uses; (2) the average percent of total limestone or dolomite for the withheld
40 end-use in the preceding and succeeding years; or (3) the average fraction of total limestone or dolomite for the
41 end-use over the entire time period.

²² *2006 IPCC Guidelines*, Volume 3: Chapter 2, Table 2.1.

1 A large quantity of crushed stone was reported to the USGS under the category “unspecified uses.” A portion of
 2 this consumption is believed to be limestone or dolomite used for emissive end uses. The quantity listed for
 3 “unspecified uses” was, therefore, allocated to all other reported end-uses according to each end-use’s fraction of
 4 total consumption in that year.²³

5 **Table 4-18: Limestone and Dolomite Consumption from Other Uses of Carbonates (kt)**

Activity	1990	2005	2018	2019	2020	2021	2022
Limestone	10,016	10,465	12,816	15,146	13,707	12,788	17,891
Dolomite	919	3,254	1,356	1,520	2,962	2,826	1,915
Total	10,935	13,719	14,172	16,667	16,669	15,614	19,806

Note: Totals may not sum due to independent rounding.

6 **Other Uses of Soda Ash**

7 Carbon dioxide emissions from soda ash consumption were calculated using a Tier 1 method from the *2006 IPCC*
 8 *Guidelines*, in accordance with the IPCC methodological decision tree and available data. Excluding glass
 9 manufacturing which is reported under Section 4.3 Glass Production (CRT Source Category 2A3), most soda ash is
 10 consumed in chemical production, with minor amounts used in soap production, pulp and paper, flue gas
 11 desulfurization, and water treatment. As soda ash is consumed for these purposes, CO₂ is usually emitted. In these
 12 applications, it is assumed that one mole of carbon is released for every mole of soda ash used. Thus,
 13 approximately 0.113 metric tons of carbon (or 0.415 metric tons of CO₂) are released for every metric ton of soda
 14 ash consumed. The activity data for soda ash consumption for 1990 to 2022 (see Table 4-19) were obtained from
 15 the *U.S. Geological Survey (USGS) Minerals Yearbook for Soda Ash* (1994 through 2015b) and *USGS Mineral*
 16 *Industry Surveys for Soda Ash* (USGS 2017a, 2018, 2019, 2020b, 2021d, 2022b, 2023b). Soda ash consumption data
 17 were collected by the USGS from voluntary surveys of the U.S. soda ash industry.

18 **Table 4-19: Other Uses of Soda Ash Consumption Not Associated with Glass Manufacturing**
 19 **(kt)**

Activity	1990	2005	2018	2019	2020	2021	2022
Soda Ash ^a	3,351	3,144	2,576	2,497	2,310	2,360	2,391

^a Soda ash consumption is sales reported by producers which exclude imports. Historically, imported soda ash is less than 1 percent of the total U.S. consumption (Kostick 2012).

20 **Ceramics Production**

21 Carbon dioxide emissions from ceramics production were calculated using a Tier 1 method from the *2006 IPCC*
 22 *Guidelines*, in accordance with the IPCC methodological decision tree and available data, by multiplying the
 23 quantity of clay consumed for emissive purposes by a national default carbonate content value of clay of 10
 24 percent, default limestone fraction of 85 percent and dolomite fraction of 15 percent, and by the emission factor
 25 for limestone or dolomite calcination, respectively: 0.43971 metric ton CO₂/metric ton of limestone and 0.47732
 26 metric ton CO₂/metric ton of dolomite.²⁴ To estimate annual process CO₂ emissions, EPA evaluated the end-uses of
 27 each type of clay published by USGS to identify the emissive end-uses that fall into the ceramics production
 28 subcategory. The emissive end-uses were organized into three groups: ceramics, glass, and floor & tile;
 29 refractories; and heavy clay products. The total limestone and dolomite consumption from the three emissive
 30 groupings for ceramics production for 1990 through 2021 (See Table 4-20) were obtained from USGS (Simmons
 31 2023). Data for 2021 was used as a proxy for 2022 because 2022 ceramics production data was not available at the
 32 time of publication.

²³ This approach was recommended by USGS, the data collection agency.

²⁴ *2006 IPCC Guidelines*, Volume 3: Chapter 2, Table 2.1.

1 **Table 4-20: Limestone and Dolomite Consumption from Ceramics Production (kt)**

Activity	1990	2005	2018	2019	2020	2021	2022
Limestone	1,444	1,569	791	755	758	765	765
Dolomite	255	277	140	133	134	135	135
Total	1,699	1,846	930	888	892	900	900

Note: Totals may not sum due to independent rounding.

2 **Non-Metallurgical Magnesia Production**

3 Carbon dioxide emissions from non-metallurgical magnesia production were calculated using a Tier 1 method from
 4 the *2006 IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available data, by
 5 multiplying the quantity of magnesium ore extracted from the mine and processed at the facility by the carbonate
 6 content for magnesite or limestone, respectively, and by the emission factor for magnesite or limestone
 7 calcination, respectively: 0.52197 metric ton CO₂/metric ton carbonate for magnesite and 0.43971 metric ton
 8 CO₂/metric ton carbonate for limestone.²⁵ A USGS report on magnesite and brucite deposits at Gabbs, Nevada lists
 9 the carbonate content of magnesite as 98 percent magnesite and 1 percent limestone with traces of other
 10 minerals (USGS 1948). In the absence of other data, all magnesium ore extracted from the mine is assumed to be
 11 used for non-metallurgical magnesium production. Magnesium ore extracted from the mine and processed at the
 12 facility for non-metallurgical magnesia production for 2002 through 2022 (See Table 4-21) was obtained from the
 13 Nevada Division of Environmental Protection (McNeece 2023). This data was not available for 1990 through 2001.
 14 To address this gap in data availability and time-series consistency, carbonate consumption for 1990 through 2001
 15 were estimated by multiplying the average ratio of magnesium ore consumption to production capacity for 2002
 16 to 2004 by the production capacity of the facility in Nevada.

17 **Table 4-21: Magnesite and Limestone Consumption from Non-Metallurgical Magnesia**
 18 **Production (kt)**

Activity	1990	2005	2018	2019	2020	2021	2022
Magnesite	214	363	321	289	410	439	388
Limestone	2	4	3	3	4	4	4
Total	216	367	325	292	414	443	392

Note: Totals may not sum due to independent rounding.

19 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
 20 through 2022. Consistent with the *2006 IPCC Guidelines*, the overlap technique was applied for non-metallurgical
 21 magnesia production to compare the magnesium ore consumption data to USGS production capacity data for 2002
 22 through 2022. To address inconsistencies, adjustments were made, as described above.

23 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

24 The uncertainty levels presented in this section account for uncertainty associated with activity data. Data on
 25 limestone and dolomite consumption are collected by USGS through voluntary national surveys. USGS contacts the
 26 mines (i.e., producers of various types of crushed stone) for annual sales data. Data on other carbonate
 27 consumption are not readily available. The producers report the annual quantity sold to various end-users and
 28 industry types. USGS estimates the historical response rate for the crushed stone survey to be approximately 70
 29 percent, and the rest is estimated by USGS. Large fluctuations in reported consumption exist, reflecting year-to-
 30 year changes in the number of survey responders. The uncertainty resulting from a shifting survey population is
 31 exacerbated by the gaps in the time series of reports. The accuracy of distribution by end use is also uncertain
 32 because this value is reported by the producer/mines and not the end user. Additionally, there is significant
 33 inherent uncertainty associated with estimating withheld data points for specific end uses of limestone and

²⁵ *2006 IPCC Guidelines*, Volume 3: Chapter 2, Table 2.1.

dolomite. Lastly, much of the limestone consumed in the United States is reported as “other unspecified uses;” therefore, it is difficult to accurately allocate this unspecified quantity to the correct end-uses. EPA contacted the USGS National Minerals Information Center Crushed Stone commodity expert to assess the current uncertainty ranges associated with the limestone and dolomite consumption data compiled and published by USGS. During this discussion, the expert confirmed that EPA’s range of uncertainty was still reasonable (Willett 2017). EPA assigned an uncertainty range of ±10 percent for limestone and dolomite consumption, based on expert judgement (Willett 2017). EPA assigned an uncertainty range of ±5 percent for soda ash consumption, and using this suggested uncertainty provided in Volume 3, Chapter 2, Section 2.4.2.2 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023).

Uncertainty in the estimates also arises in part due to variations in the chemical composition of limestone. In addition to calcium carbonate, limestone may contain smaller amounts of magnesia, silica, and sulfur, among other minerals. The exact specifications for limestone or dolomite used as flux stone vary with the pyrometallurgical process and the kind of ore processed. EPA assigned an uncertainty range of ±3 percent for the CO₂ emission factors for limestone and dolomite consumption, and using this suggested uncertainty provided in Volume 3, Chapter 2, Section 2.5.2.1 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023). EPA assigned an uncertainty range of ±2 percent for the CO₂ emission factor for soda ash consumption, and using this suggested uncertainty provided in Volume 3, Chapter 2, Section 2.4.2.1 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023).

For emissions from soda ash consumption, the primary source of uncertainty results from the fact that these emissions are dependent upon the type of processing employed by each end-use. Specific emission factors for each end-use are not available, so a Tier 1 default emission factor is used for all end-uses. Therefore, there is uncertainty surrounding the emission factors from the consumption of soda ash. Additional uncertainty comes from the reported consumption and allocation of consumption within sectors that is collected on a quarterly basis by the USGS. Efforts have been made to categorize company sales within the correct end-use sector. EPA assigned an uncertainty range of ±2 percent for the CO₂ emission factor for soda ash consumption. The uncertainty range is derived from the default ranges for soda ash consumption for glass production in Volume 3, Chapter 2, Section 2.4.2.1 of the *2006 IPCC Guidelines* which is representative of soda ash consumption not associated with glass production, based on expert judgment (RTI 2023).

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-20. Carbon dioxide emissions from other process uses of carbonates in 2022 were estimated to be between 7.1 and 9.2 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 11 percent below and 14 percent above the emission estimate of MMT CO₂ Eq.

Table 4-20: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Other Process Uses of Carbonates (MMT CO₂ Eq. and Percent)

Source	Gas	2021 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
Other Process Uses of Carbonates	CO ₂		7.1	9.2	-11%	+14%

^a Range of emission estimates predicted by Monte Carlo stochastic *simulation* for a 95 percent confidence interval.

1 QA/QC and Verification

2 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
3 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the
4 introduction of the IPPU chapter (see Annex 8 for more details).

5 Recalculations Discussion

6 For the current *Inventory*, emissions estimates for new subcategories ceramics production and non-metallurgical
7 magnesia production were incorporated across the entire time series. No other recalculations were performed for
8 the 1990 through 2021 portion of the time series.

9 Planned Improvements

10 EPA plans to review the uncertainty ranges assigned to activity data. This planned improvement is currently
11 planned as a medium-term improvement.

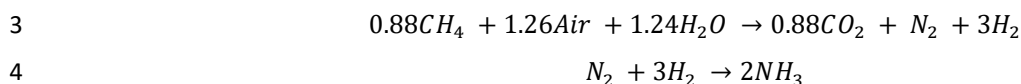
12 4.5 Ammonia Production (CRT Source 13 Category 2B1)

14 Emissions of carbon dioxide (CO₂) occur during the production of synthetic ammonia (NH₃), primarily through the
15 use of natural gas, petroleum coke, or naphtha as a feedstock. The natural gas-, naphtha-, and petroleum coke-
16 based processes produce CO₂ and hydrogen (H₂), the latter of which is used in the production of ammonia. The
17 brine electrolysis process for production of ammonia does not lead to process-based CO₂ emissions. Due to
18 national circumstances, emissions from fuels consumed for energy purposes during the production of ammonia
19 are accounted for in the Energy chapter. More information on this approach can be found in the Methodology
20 section below.

21 Ammonia production requires a source of nitrogen (N) and hydrogen (H). Nitrogen is obtained from air through
22 liquid air distillation or an oxidative process where air is burnt and the residual nitrogen is recovered. In the United
23 States, the majority of ammonia is produced using a natural gas feedstock as the hydrogen source. One synthetic
24 ammonia production plant located in Kansas is producing ammonia from petroleum coke feedstock. In some U.S.
25 plants, some of the CO₂ produced by the process is captured and used to produce urea rather than being emitted
26 to the atmosphere. In 2022, 16 companies operated 35 ammonia producing facilities in 16 states. Approximately
27 60 percent of domestic ammonia production capacity is concentrated in Louisiana, Oklahoma, and Texas (USGS
28 2023).

29 Synthetic ammonia production from natural gas feedstock consists of five principal process steps. The primary
30 reforming step converts methane (CH₄) to CO₂, carbon monoxide (CO), and hydrogen (H₂) in the presence of a
31 catalyst. Only 30 to 40 percent of the CH₄ feedstock to the primary reformer is converted to CO and CO₂ in this
32 step of the process. The secondary reforming step converts the remaining CH₄ feedstock to CO and CO₂. In the shift
33 conversion step, the CO in the process gas from the secondary reforming step (representing approximately 15
34 percent of the process gas) is converted to CO₂ in the presence of a catalyst, water, and air. Carbon dioxide is
35 removed from the process gas by the shift conversion process, and the H₂ is combined with the nitrogen (N₂) gas in
36 the process gas during the ammonia synthesis step to produce ammonia. The CO₂ is included in a waste gas stream
37 with other process impurities and is absorbed by a scrubber solution. In regenerating the scrubber solution, CO₂ is
38 released from the solution.

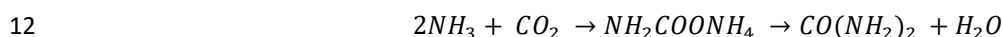
1 The conversion process for conventional steam reforming of CH₄, including the primary and secondary reforming
 2 and the shift conversion processes, is approximately as follows:



5 To produce synthetic ammonia from petroleum coke, the petroleum coke is gasified and converted to CO₂ and H₂.
 6 These gases are separated, and the H₂ is used as a feedstock to the ammonia production process, where it is
 7 reacted with N₂ to form ammonia.

8 Not all of the CO₂ produced during the production of ammonia is emitted directly to the atmosphere. Some of the
 9 ammonia and some of the CO₂ produced by the synthetic ammonia process are used as raw materials in the
 10 production of urea [CO(NH₂)₂], which has a variety of agricultural and industrial applications.

11 The chemical reaction that produces urea is:



13 Only the CO₂ emitted directly to the atmosphere from the synthetic ammonia production process is accounted for
 14 in determining emissions from ammonia production. The CO₂ that is captured during the ammonia production
 15 process and used to produce urea does not contribute to the CO₂ emission estimates for ammonia production
 16 presented in this section. Instead, CO₂ emissions resulting from the consumption of urea are attributed to the urea
 17 consumption or urea application source category (under the assumption that the carbon stored in the urea during
 18 its manufacture is released into the environment during its consumption or application). Emissions of CO₂ resulting
 19 from agricultural applications of urea are accounted for in Section 5.6 urea fertilization (CRT Source Category 3H).
 20 Emissions of CO₂ resulting from non-agricultural applications of urea (e.g., use as a feedstock in chemical
 21 production processes) are accounted for in Section 4.6 Urea Consumption for Non-Agricultural Purposes.

22 Emissions from fuel used for energy at ammonia plants are accounted for in the Energy chapter. The consumption
 23 of natural gas and petroleum coke as fossil fuel feedstocks for NH₃ production are adjusted for within the Energy
 24 chapter as these fuels were consumed during non-energy related activities. More information on this methodology
 25 is described in Annex 2.1, Methodology for Estimating Emissions of CO₂ from Fossil Fuel Combustion.

26 Total emissions of CO₂ from ammonia production in 2022 were 12.6 MMT CO₂ Eq. (12,610 kt) and are summarized
 27 in Table 4-21 and Table 4-22. Ammonia production relies on natural gas as both a feedstock and a fuel, and as
 28 such, market fluctuations and volatility in natural gas prices affect the production of ammonia. Since 1990,
 29 emissions from ammonia production have decreased by 12 percent. Emissions in 2022 increased by about 3
 30 percent from the 2021 levels. One facility in Kansas produces ammonia from petroleum coke and began operations
 31 in 2000. All other facilities use natural gas as feedstock.

32 Emissions from ammonia production increased steadily from 2015 to 2018, due to the addition of new ammonia
 33 production facilities and new production units at existing facilities in 2016, 2017, and 2018. Agriculture continues
 34 to drive demand for nitrogen fertilizers, accounting for approximately 88 percent of domestic ammonia
 35 consumption (USGS 2023).

36 **Table 4-21: CO₂ Emissions from Ammonia Production (MMT CO₂ Eq.)**

Source	1990	2005	2018	2019	2020	2021	2022
Ammonia Production	14.4	10.2	12.7	12.4	13.0	12.2	12.6

37 **Table 4-22: CO₂ Emissions from Ammonia Production (kt CO₂)**

Source	1990	2005	2018	2019	2020	2021	2022
Ammonia Production	14,404	10,234	12,669	12,401	13,006	12,192	12,610

1 Methodology and Time-Series Consistency

2 Estimates of CO₂ emissions from the production of synthetic ammonia are estimated using a country-specific
3 approach consistent with Tier 3 method from the *2006 IPCC Guidelines*, in accordance with the IPCC
4 methodological decision tree and available data (IPCC 2006). Emissions from fuel used for energy at ammonia
5 plants are accounted for in the Energy chapter. This approach differs slightly from the *2006 IPCC Guidelines* which
6 indicates that “in the case of ammonia production no distinction is made between fuel and feedstock emissions
7 with all emissions accounted for in the IPPU Sector.” Disaggregated data on fuel used for ammonia feedstock and
8 fuel used for energy for ammonia production are not available in the United States. The Energy Information
9 Administration (EIA), where energy use data are obtained for the *Inventory* (see the Energy chapter), does not
10 provide data broken out by industrial category. EIA data are only available at the broad industry sector level.
11 Furthermore, the GHGRP data used to estimate emissions are based on feedstock use and not fuel use.

12 Petroleum Coke Feedstock

13 Since 2000, one facility in the United States has produced ammonia using petroleum coke as a feedstock. For 2010
14 to 2022, CO₂ emissions from the production of synthetic ammonia from petroleum coke feedstock were estimated
15 using CO₂ emissions reported by the facility to GHGRP (EPA 2018; EPA 2023).

16 For 2006 to 2009, CO₂ emissions from the production of synthetic ammonia from petroleum coke feedstock were
17 estimated by multiplying the following: quantity of petroleum coke feedstock reported by the facility (CVR 2008
18 through 2022); the *Inventory* heating content value for petroleum coke (consistent with values used in the Energy
19 chapter); the petroleum coke carbon content; and a stoichiometric CO₂/C factor of 44/12.

20 For 2000 to 2005, the quantity of petroleum coke feedstock was not available and was estimated by multiplying
21 the average ratio of petroleum coke feedstock quantity to ammonia production quantity produced from
22 petroleum coke from 2006 through 2010 by total ammonia production for 2000 to 2005 (ACC 2023). The years
23 2006 to 2010 were used to determine the average ratio of petroleum coke feedstock quantity to the ammonia
24 quantity produced from petroleum coke because that period was deemed to better represent historic ammonia
25 production from petroleum coke for the period from 2000 to 2005.

26 For 2000 to 2005, CO₂ emissions from the production of synthetic ammonia from petroleum coke feedstock were
27 estimated by multiplying the following: the average ratio of petroleum coke feedstock quantity to ammonia
28 production quantity; total ammonia production quantity (ACC 2023); the *Inventory* heating content value for
29 petroleum coke (consistent with values used in the Energy chapter); the petroleum coke carbon content; and a
30 stoichiometric CO₂/C factor of 44/12.

31 Natural Gas Feedstock

32 For 2017 through 2022, facilities directly reported to GHGRP the quantity of natural gas feedstock used for
33 ammonia production along with the carbon content of the natural gas feedstock (EPA 2018, EPA 2023).

34 For 2010 through 2016, the quantity of natural gas feedstock was calculated using GHGRP process CO₂ emissions
35 for 2010 through 2016, average molecular weight of the feedstock from 2017 through 2021, and average carbon
36 content from 2017 through 2021. Data from years 2017 to 2021 were used to determine the average molecular
37 weight and the average carbon content because that period better represents historic ammonia production from
38 2010 to 2016. Using all available data from 2017 to 2021 allowed for the maximum number of data points
39 available at the time of adopting this methodology to ensure that the average was representative. The averages
40 were not updated using later data to exclude any new facilities that might not be representative of facilities that
41 were operating during the earlier years of the GHGRP.

42 For 2010 to 2022, CO₂ emissions from the production of synthetic ammonia from natural gas feedstock were
43 estimated using the CO₂ emissions reported to the GHGRP (EPA 2018, EPA 2023) and subtracting the CO₂ emissions

1 from the production of synthetic ammonia from petroleum coke feedstock as determined in the Petroleum Coke
2 Feedstock section above.

3 For 1990 to 2009, the quantity of natural gas feedstock was not available and was estimated by multiplying the
4 average ratio of natural gas feedstock quantity to ammonia production quantity from 2010 through 2014 by total
5 ammonia production for each year for 1990 to 2009 (ACC 2023). The years 2010 to 2014 were used to determine
6 the average ratio of natural gas feedstock quantity to ammonia production because that period better represents
7 historic ammonia production from 1990 to 2009.²⁶ For 1990 to 2009, CO₂ emissions from the production of
8 synthetic ammonia from natural gas feedstock were estimated using the natural gas feedstock quantity as
9 determined above and the *Inventory* CO₂ emissions factor and heating content value for natural gas (consistent
10 with values used in the Energy chapter).

11 Urea Production Adjustments

12 Emissions of CO₂ from ammonia production from both feedstocks and for all years from 1990 to 2022 were
13 adjusted to account for the use of some CO₂ emissions resulting from ammonia production as a raw material in the
14 production of urea. The CO₂ emissions reported for ammonia production are reduced by a factor of 0.733, which
15 corresponds to a stoichiometric CO₂/urea factor of 44/60, assuming complete conversion of ammonia (NH₃) and
16 CO₂ to urea (IPCC 2006; EFMA 2000), and multiplied by total annual domestic urea production.

17 All synthetic ammonia production and subsequent urea production are assumed to be from the same process—
18 conventional catalytic reforming of natural gas feedstock, with the exception of ammonia production from
19 petroleum coke feedstock at the one facility located in Kansas.

20 **Table 4-23: Total Ammonia Production, Total Urea Production, and Recovered CO₂ Consumed**
21 **for Urea Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Total Ammonia Production	15,425	10,143	16,010	16,410	17,020	15,420	16,800
Total Urea Production	7,450	5,270	10,700	11,400	11,500	10,521	11,272
Recovered CO ₂ Consumed for Urea Production	5,463	3,865	7,847	8,360	8,433	7,715	8,266

22 Total ammonia production, total urea production, and recovered CO₂ consumed for urea production are shown in
23 Table 4-23. Total ammonia production data for 2011 through 2022 were obtained from American Chemistry
24 Council (ACC 2023). For years 1990 through 2011, ammonia production data were obtained from the Census
25 Bureau of the U.S. Department of Commerce (U.S. Census Bureau 1991 through 1994, 1998 through 2011) as
26 reported in *Current Industrial Reports Fertilizer Materials and Related Products* annual and quarterly reports. Data
27 on facility-level process emissions for 2010 through 2022 and data on natural gas feedstock used and carbon
28 content of the natural gas feedstock starting in 2017 were obtained from GHGRP (EPA 2018; EPA 2023). Natural
29 gas and petroleum coke heating values come from national-level data (EIA 2023), and natural gas and petroleum
30 coke carbon contents are the same as used in the Energy chapter calculations.

31 Data on urea production for 2010 through 2022 were obtained from GHGRP (EPA 2018, EPA 2023). Urea
32 production data for 2009 through 2010 were obtained from the U.S. Census Bureau (U.S. Census Bureau 2010 and
33 2011). Urea production data for 1990 through 2008 were obtained from the USGS *Minerals Yearbook: Nitrogen*
34 (USGS 1994-2009). The U.S. Census Bureau ceased collection of urea production statistics in 2011.

²⁶ The number of facilities reporting to GHGRP has increased since 2010: 22 facilities reported from 2010 to 2012; 23 from 2013 to 2015; 26 in 2016; 28 in 2017 and 29 from 2018 to 2022. Using data from 2010 to 2014 excludes the newer facilities that might not be representative of facilities in earlier years.

1 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
 2 through 2022. The methodology for ammonia production spliced activity data from different sources: U. S. Census
 3 Bureau data for 1990 through 2010, ACC data beginning in 2011, and GHGRP data beginning in 2010 and
 4 2017. Consistent with the *2006 IPCC Guidelines*, the overlap technique was applied to compare the two data sets
 5 for years where there was overlap, with findings that the data sets were consistent and adjustments were not
 6 needed.

7 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

8 The uncertainties presented in this section are primarily due to how accurately the emission factor used represents
 9 an average across all ammonia plants using natural gas feedstock. Uncertainty in the back calculation of natural gas
 10 feedstock used for 1990 through 2009 also exists. Using the average ratio of natural gas feedstock quantity to
 11 ammonia production, determined using GHGRP data from 2010 to 2014, does not account for efficiency gains in
 12 ammonia production since 1990 (e.g., potential decreases in gas usage per ton of ammonia, manufacturing shift
 13 from steam-driven turbines to electrical-drive turbines). Uncertainties are also associated with ammonia
 14 production estimates and the assumption that all ammonia production and subsequent urea production was from
 15 the same process—conventional catalytic reforming of natural gas feedstock, with the exception of one ammonia
 16 production plant located in Kansas that is manufacturing ammonia from petroleum coke feedstock. Uncertainty is
 17 also associated with the representativeness of the emission factor used for the petroleum coke-based ammonia
 18 process. It is also assumed that ammonia and urea are produced at co-located plants from the same natural gas
 19 raw material. The uncertainty of the total urea production activity data, based on USGS *Minerals Yearbook:
 20 Nitrogen* data, is a function of the reliability of reported production data and is influenced by the completeness of
 21 the survey responses. EPA assigned an uncertainty range of ± 5 percent for ammonia production and a range of ± 2
 22 percent for urea production, natural gas feedstock quantity, petroleum coke feedstock quantity, and carbon
 23 content of natural gas feedstock, and using these suggested uncertainty provided in Section 3.2.3.2 of the *2006
 24 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023).

25 Recovery of CO₂ from ammonia production plants for purposes other than urea production (e.g., commercial sale,
 26 etc.) has not been considered in estimating the CO₂ emissions from ammonia production, as data concerning the
 27 disposition of recovered CO₂ are not available. Such recovery may or may not affect the overall estimate of CO₂
 28 emissions depending upon the end use to which the recovered CO₂ is applied. Further research is required to
 29 determine whether byproduct CO₂ is being recovered from other ammonia production plants for application to
 30 end uses that are not accounted for elsewhere; however, for reporting purposes, CO₂ consumption for urea
 31 production is provided in this chapter.

32 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-24. Carbon dioxide
 33 emissions from ammonia production in 2022 were estimated to be between 11.8 and 12.6 MMT CO₂ Eq. at the 95
 34 percent confidence level. This indicates a range of approximately 4 percent below and 4 percent above the
 35 emission estimate of 12.2 MMT CO₂ Eq.

36 **Table 4-24: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Ammonia**
 37 **Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.)			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Ammonia Production	CO ₂	12.2	11.8	12.6	-4%	+4%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

1 QA/QC and Verification

2 General quality assurance/quality control (QA/QC) procedures were applied to ammonia production emission
3 estimates consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of 2006
4 *IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details). More details
5 on the greenhouse gas calculation, monitoring and QA/QC methods applicable to ammonia facilities can be found
6 under Subpart G (Ammonia Production) of the regulation (40 CFR Part 98).²⁷ EPA verifies annual facility-level
7 GHGRP reports through a multi-step process (e.g., combination of electronic checks and manual reviews) to
8 identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent.²⁸ Based on
9 the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred.
10 The post-submittals checks are consistent with a number of general and category-specific QC procedures, including
11 range checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.

12 More details on the greenhouse gas calculation, monitoring, and QA/QC methods applicable to reporting of urea
13 produced at ammonia production facilities can be found under Section 4.6 Urea Consumption for Non-Agricultural
14 Purposes.

15 Recalculations

16 For 2021, the urea consumption value was changed from a rounded value to a more precise unrounded value. As a
17 result, recalculations were performed for emissions from ammonia for years 1990 through 2021. Compared to the
18 previous *Inventory*, total CO₂ emissions from the production of ammonia production (from natural gas and
19 petroleum coke feedstocks) decreased by less than 1 percent (15 kt) in 2021.

20 Planned Improvements

21 Currently the *Inventory* does not separately track fuel energy use for ammonia production. To be more consistent
22 with 2006 *IPCC Guidelines*, EPA is considering whether to include natural gas fuel use as part of ammonia
23 production emissions as a future improvement. The data are still being evaluated as part of EPA's efforts to
24 disaggregate other industrial sector categories' energy use in the Energy chapter of the *Inventory*. If possible, this
25 will be incorporated in future *Inventory* reports. If incorporated, the fuel energy use and emissions will be removed
26 from current reporting under Energy to avoid double counting.

27 4.6 Urea Consumption for Non-Agricultural 28 Purposes

29 Urea is produced using ammonia (NH₃) and carbon dioxide (CO₂) as raw materials. All urea produced in the United
30 States is assumed to be produced at ammonia production facilities where both ammonia and CO₂ are generated.
31 There were 35 plants producing ammonia in the United States in 2022, with two additional plants sitting idle for
32 the entire year (USGS 2023b).

33 The chemical reaction that produces urea is:



²⁷ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

²⁸ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 This section accounts for CO₂ emissions associated with urea consumed exclusively for non-agricultural purposes.
 2 Emissions of CO₂ resulting from agricultural applications of urea are accounted for in Section 5.6 Urea Fertilization
 3 (CRT Source Category 3H) of the Agriculture chapter.

4 The industrial applications of urea include its use in adhesives, binders, sealants, resins, fillers, analytical reagents,
 5 catalysts, intermediates, solvents, dyestuffs, fragrances, deodorizers, flavoring agents, humectants and
 6 dehydrating agents, formulation components, monomers, paint and coating additives, photosensitive agents, and
 7 surface treatments agents. In addition, urea is used for abating nitrogen oxide (NO_x) emissions from coal-fired
 8 power plants and diesel transportation motors.

9 Emissions of CO₂ from urea consumed for non-agricultural purposes in 2022 were estimated to be 7.1 MMT CO₂
 10 Eq. (7,053 kt) and are summarized in Table 4-25 and Table 4-26. Net CO₂ emissions from urea consumption for
 11 non-agricultural purposes have increased by approximately 86 percent from 1990 to 2022 and increased by
 12 approximately 7 percent from 2021 to 2022.

13 **Table 4-25: CO₂ Emissions from Urea Consumption for Non-Agricultural Purposes (MMT CO₂**
 14 **Eq.)**

Source	1990	2005	2018	2019	2020	2021	2022
Urea Consumption	3.8	3.7	6.1	6.2	5.8	6.6	7.1

15 **Table 4-26: CO₂ Emissions from Urea Consumption for Non-Agricultural Purposes (kt CO₂)**

Source	1990	2005	2018	2019	2020	2021	2022
Urea Consumption	3,784	3,653	6,113	6,150	5,805	6,600	7,053

16 Methodology and Time-Series Consistency

17 Emissions of CO₂ resulting from urea consumption for non-agricultural purposes are estimated using a country-
 18 specific method consistent with the Tier 1 method used to estimate emissions from ammonia production in the
 19 2006 IPCC Guidelines which states that the “CO₂ recovered [from ammonia production] for downstream use can be
 20 estimated from the quantity of urea produced where CO₂ is estimated by multiplying urea production by 44/60,
 21 the stoichiometric ratio of CO₂ to urea” (IPCC 2006). The amount of urea consumed in the United States for non-
 22 agricultural purposes is multiplied by a factor representing the amount of CO₂ used as a raw material to produce
 23 the urea. This method is based on the assumption that all of the carbon in urea is released into the environment as
 24 CO₂ during use.

25 The amount of urea consumed for non-agricultural purposes in the United States is estimated by deducting the
 26 quantity of urea fertilizer applied to agricultural lands, which is obtained directly from the Agriculture chapter (see
 27 Table 5-25), from the total domestic supply of urea as reported in Table 4-27. The domestic supply of urea is
 28 estimated based on the amount of urea produced plus urea imports and minus urea exports. A factor of 0.733 tons
 29 of CO₂ per ton of urea consumed is then applied to the resulting supply of urea for non-agricultural purposes to
 30 estimate CO₂ emissions from the amount of urea consumed for non-agricultural purposes. The 0.733 tons of CO₂
 31 per ton of urea emission factor is based on the stoichiometry of carbon in urea. This corresponds to a
 32 stoichiometric ratio of CO₂ to urea of 44/60, assuming complete conversion of carbon in urea to CO₂ (IPCC 2006;
 33 EFMA 2000).

34 Urea production data for 1990 through 2008 were obtained from the U.S. Geological Survey (USGS) *Minerals*
 35 *Yearbook: Nitrogen* (USGS 1994 through 2009a). Urea production data for 2009 through 2010 were obtained from
 36 the U.S. Census Bureau (2011). The U.S. Census Bureau ceased collection of urea production statistics in 2011.
 37 Urea production data for 2011 through 2022 were obtained from GHGRP (EPA 2018; EPA 2023a; EPA 2023b).

38 Urea import data for 2022 were not available at the time of publication and were estimated using 2021 values.
 39 Urea import data for 2013 to 2021 were obtained from the USGS *Minerals Yearbook: Nitrogen* (USGS 2023a). Urea

1 import data for 2011 and 2012 were taken from U.S. Fertilizer Import/Exports from the United States Department
 2 of Agriculture (USDA) Economic Research Service Data Sets (U.S. Department of Agriculture 2012). USDA
 3 suspended updates to this data after 2012. Urea import data for the previous years were obtained from the U.S.
 4 Census Bureau *Current Industrial Reports Fertilizer Materials and Related Products* annual and quarterly reports for
 5 1997 through 2010 (U.S. Census Bureau 2001 through 2011), The Fertilizer Institute (TFI 2002) for 1993 through
 6 1996, and the United States International Trade Commission Interactive Tariff and Trade DataWeb (U.S. ITC 2002)
 7 for 1990 through 1992 (see Table 4-27).

8 Urea export data for 2022 were not available at the time of publication and were estimated using 2021 values.
 9 Urea export data for 2013 to 2021 were obtained from the USGS *Minerals Yearbook: Nitrogen* (USGS 2023a). Urea
 10 export data for 1990 through 2012 were taken from U.S. Fertilizer Import/Exports from USDA Economic Research
 11 Service Data Sets (U.S. Department of Agriculture 2012). USDA suspended updates to this data after 2012.

12 **Table 4-27: Urea Production, Urea Applied as Fertilizer, Urea Imports, and Urea Exports (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Urea Production	7,450	5,270	10,700	11,400	11,500	10,521	11,272
Urea Applied as Fertilizer	3,296	4,779	6,732	6,865	6,998	7,131	7,265
Urea Imports	1,860	5,026	5,110	4,410	4,190	5,880	5,880
Urea Exports	854	536	743	559	777	270	270
Urea Consumed for Non-Agricultural Purposes	5,160	4,981	8,335	8,386	7,915	9,000	9,617

13 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
 14 through 2022. The methodology for urea consumption for non-agricultural purposes spliced activity data from
 15 different sources: USGS data for 1990 through 2008, U. S. Census Bureau data for 2009 and 2010, and GHGRP data
 16 beginning in 2011. Consistent with the *2006 IPCC Guidelines*, the overlap technique was applied to compare the
 17 data sets for years where there was overlap, with findings that the data sets were consistent and adjustments
 18 were not needed.

19 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

20 There is limited publicly available data on the quantities of urea produced and consumed for non-agricultural
 21 purposes. Therefore, the amount of urea used for non-agricultural purposes is estimated based on a balance that
 22 relies on estimates of urea production, urea imports, urea exports, and the amount of urea used as fertilizer. EPA
 23 uses an uncertainty range of ± 5 percent for urea imports and urea exports, consistent with the ranges for activity
 24 data that are not obtained directly from plants, and using this suggested uncertainty provided in section 3.2.3.2 of
 25 the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023). The primary uncertainties associated
 26 with this source category are associated with the accuracy of these estimates as well as the fact that each estimate
 27 is obtained from a different data source. Because urea production estimates are no longer available from the
 28 USGS, there is additional uncertainty associated with urea produced beginning in 2011. There is also uncertainty
 29 associated with the assumption that all of the carbon in urea is released into the environment as CO₂ during use.

30 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-28. Carbon dioxide
 31 emissions associated with urea consumption for non-agricultural purposes during 2022 were estimated to be
 32 between 4.8 and 5.2 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 4
 33 percent below and 4 percent above the emission estimate of 5.0 MMT CO₂ Eq.

34 **Table 4-28: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Urea
 35 Consumption for Non-Agricultural Purposes (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower	Upper	Lower	Upper

			Bound	Bound	Bound	Bound
Urea Consumption for Non-Agricultural Purposes	CO ₂	5.0	4.8	5.2	-4%	+4%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details).

More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to reporting of urea production occurring at ammonia facilities can be found under Subpart G (Ammonia Manufacturing) of the regulation (40 CFR Part 98).²⁹ EPA verifies annual facility-level GHGRP reports through a multi-step process (e.g., combination of electronic checks and manual reviews) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent.³⁰ Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including range checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions. EPA also conducts QA checks of GHGRP reported urea production data against external datasets including the USGS Minerals Yearbook data. The comparison shows consistent trends in urea production over time.

Recalculations Discussion

Based on updated quantities of urea applied for agricultural uses for 2017 through 2021, updated urea imports from USGS for 2021, and updated urea exports from USGS for 2021, recalculations were performed for 2017 through 2021. Compared to the previous *Inventory*, CO₂ emissions from urea consumption for non-agricultural purposes increased by less than 1 percent for 2017 (46 kt CO₂) and 2018 (2 kt CO₂), decreased by less than 1 percent for 2019 (4 kt CO₂) and 2020 (10 kt CO₂) and increased by 32 percent for 2021 (1,611 kt CO₂).

Planned Improvements

At this time, there are no specific planned improvements for estimating CO₂ emissions from urea consumption for non-agricultural purposes.

4.7 Nitric Acid Production (CRT Source Category 2B2)

Nitrous oxide (N₂O) is emitted during the production of nitric acid (HNO₃), an inorganic compound used primarily to make synthetic commercial fertilizers. Nitric acid is also a major component in the production of adipic acid—a feedstock for nylon—and explosives. Virtually all of the nitric acid produced in the United States is manufactured by the high-temperature catalytic oxidation of ammonia (EPA 1998). There are two different nitric acid production

²⁹ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

³⁰ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 methods: weak nitric acid and high-strength nitric acid. The first method utilizes oxidation, condensation, and
 2 absorption to produce nitric acid at concentrations between 30 and 70 percent nitric acid. High-strength acid (90
 3 percent or greater nitric acid) can be produced from dehydrating, bleaching, condensing, and absorption of the
 4 weak nitric acid. Most U.S. plants were built between 1960 and 2000. As of 2022, there were 31 active nitric acid
 5 production plants, including one high-strength nitric acid production plant in the United States (EPA 2010; EPA
 6 2023).

7 The basic process technology for producing nitric acid has not changed significantly over time. During this process,
 8 N₂O is formed as a byproduct and released from reactor vents into the atmosphere. Emissions from fuels
 9 consumed for energy purposes during the production of nitric acid are included in the Energy chapter.

10 Nitric acid is made from the reaction of ammonia (NH₃) with oxygen (O₂) in two stages. The overall reaction is:



12 Currently, the nitric acid industry in the United States controls emissions of NO and NO₂ (i.e., NO_x), using a
 13 combination of non-selective catalytic reduction (NSCR) and selective catalytic reduction (SCR) technologies. In the
 14 process of destroying NO_x, NSCR systems are also very effective at destroying N₂O. Five nitric acid plants had NSCR
 15 systems installed between 1964 and 1977, over half due to the finalization of the Nitric Acid Plant New Source
 16 Performance Standards (NSPS) which went into effect in 1971. Four additional nitric acid plants had NSCR systems
 17 installed between 2016 and 2018, as a result of EPA Consent Decrees to control NO_x emissions more effectively.
 18 NSCR systems are used in approximately one-third of the weak acid production plants. For N₂O abatement, U.S.
 19 facilities are using both tertiary (i.e., NSCR and SCR) and secondary controls (i.e., catalysts added to the ammonia
 20 reactor to lessen potential N₂O production).

21 Emissions from the production of nitric acid are generally directly proportional to the annual amount of nitric acid
 22 produced because emissions are calculated as the product of the total annual production and plant-specific
 23 emission factors. There are a few instances, however, where that relationship has not been directly proportional.
 24 For example, in 2015 and 2019, nitric acid production decreased and emissions increased compared to the
 25 respective preceding years; in 2016, nitric acid production increased and emissions decreased compared to
 26 2015. N₂O emissions for those years are calculated based on data from the GHGRP as discussed in the
 27 Methodology section below. According to data from plants reporting to GHGRP, plant-specific operations can
 28 affect the emission factor used, including: (1) site-specific fluctuations in ambient temperature and humidity, (2)
 29 catalyst age and condition, (3) process changes, such as fluctuations in process pressure or temperature and
 30 replacing the ammonia catalyst, (4) the addition, removal, maintenance, and utilization of abatement technologies,
 31 and (5) the number of nitric acid trains, which are reaction vessels where ammonia is oxidized to form nitric acid.
 32 Changes in those operating conditions for the years in question (2015, 2016, and 2019) caused changes in emission
 33 factors, which resulted in emissions changing disproportionately to production in those years.

34 Nitrous oxide emissions from this source were estimated to be 8.6 MMT CO₂ Eq. (33 kt of N₂O) in 2022 and are
 35 summarized in Table 4-29 and Table 4-30. Emissions from nitric acid production have decreased by 20 percent
 36 since 1990, while production has increased by 9 percent over the same time period (see Table 4-29 and Table
 37 4-30). Emissions have decreased by 33 percent since 1997, the highest year of production in the time series. From
 38 2021 to 2022, nitric acid production increased by 1 percent, leading to an overall increase in emissions from nitric
 39 acid production of 9.4 percent from 2021 to 2022.

40 **Table 4-29: N₂O Emissions from Nitric Acid Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Nitric Acid Production	10.8	10.1	8.5	8.9	8.3	7.9	8.6

41

42 **Table 4-30: N₂O Emissions from Nitric Acid Production (kt N₂O)**

Year	1990	2005	2018	2019	2020	2021	2022
Nitric Acid Production	41	38	32	34	31	30	33

1 Methodology and Time-Series Consistency

2 Emissions of N₂O from nitric acid production are estimated using methods provided by the *2006 IPCC Guidelines*, in
3 accordance with the IPCC methodological decision tree and available data. For 2010 through 2022, a Tier 3 method
4 was used to estimate emissions based on GHGRP data. For 1990 through 2009, a Tier 2 method was used to
5 estimate emissions from nitric acid production based on U.S. Census Bureau data.

6 2010 through 2022

7 Process N₂O emissions and nitric acid production data were obtained directly from EPA's GHGRP for 2010 through
8 2022 by aggregating reported facility-level data (EPA 2018, EPA 2023).

9 Since 2010, in the United States, all nitric acid facilities that produce weak nitric acid (30 to 70 percent) have been
10 required to report annual greenhouse gas emissions data to EPA as per the requirements of the GHGRP (Subpart
11 V). Beginning with 2018, the rule was changed to include facilities that produce nitric acid of any strength. The only
12 facility that produces high-strength nitric acid also produces weak nitric acid. All N₂O emissions from nitric acid
13 production originate from the production of weak nitric acid.

14 Process emissions and nitric acid production reported to the GHGRP provide complete estimates of greenhouse
15 gas emissions for the United States because there are no reporting thresholds. While facilities are allowed to stop
16 reporting to the GHGRP if the total reported emissions from nitric acid production are less than 25,000 metric tons
17 CO₂ Eq. per year for five consecutive years or less than 15,000 metric tons CO₂ Eq. per year for three consecutive
18 years, no facilities have stopped reporting as a result of these provisions.³¹ All nitric acid facilities are required to
19 either calculate process N₂O emissions using a site-specific emission factor that is the average of the emission
20 factor determined through annual performance tests for each nitric acid train under typical operating conditions or
21 directly measure process N₂O emissions using monitoring equipment.³²

22 Emissions from facilities vary from year to year, depending on the amount of nitric acid produced with and without
23 abatement technologies and other conditions affecting the site-specific emission factor. To maintain consistency
24 across the time series and with the rounding approaches taken by other data sets, GHGRP nitric acid data are
25 rounded and are shown in Table 4-31.

26 1990 through 2009

27 Using GHGRP data for 2010,³³ country-specific N₂O emission factors were calculated for nitric acid production with
28 abatement and without abatement (i.e., controlled and uncontrolled emission factors). The following 2010
29 emission factors were derived for production with abatement and without abatement: 3.3 kg N₂O/metric ton
30 HNO₃ produced at plants using abatement technologies (e.g., tertiary systems such as NSCR systems) and 5.99 kg
31 N₂O/metric ton HNO₃ produced at plants not equipped with abatement technology. Country-specific weighted
32 emission factors were derived by weighting these emission factors by percent production with abatement and
33 without abatement over time periods 1990 through 2008 and 2009. These weighted emission factors were used to
34 estimate N₂O emissions from nitric acid production for years prior to the availability of GHGRP data (i.e., 1990

³¹ See 40 CFR 98.2(i)(1) and 40 CFR 98.2(i)(2) for more information about these provisions.

³² Facilities must use standard methods - either EPA Method 320 or ASTM D6348-03 for annual performance tests - and must follow associated QA/QC procedures consistent with category-specific QC of direct emission measurements during these performance tests.

³³ National N₂O process emissions, national production, and national share of nitric acid production with abatement and without abatement technology was aggregated from the GHGRP facility-level data for 2010 to 2017 (i.e., percent production with and without abatement).

1 through 2008 and 2009). A separate weighted emission factor is included for 2009 due to data availability for that
 2 year.

3 EPA verified the installation dates of N₂O abatement technologies for all facilities based on GHGRP facility-level
 4 information and confirmed that all abatement technologies were accounted for in the derived emission factors
 5 (Icenhour 2020). No changes to N₂O abatement levels from 1990 through 2008 or for 2009 were made due to the
 6 review of GHGRP-reported N₂O abatement installation dates. Due to the lack of information on abatement
 7 equipment utilization, it is assumed that once abatement technology was installed in facilities, the equipment was
 8 consistently operational for the duration of the time series considered in this report (especially NSCRs).

9 The country-specific weighted N₂O emission factors were used in conjunction with annual production to estimate
 10 N₂O emissions for 1990 through 2009, using the following equations:

11 **Equation 4-4: 2006 IPCC Guidelines Tier 3: N₂O Emissions From Nitric Acid Production**
 12 **(Equation 3.6)**

$$E_i = P_i \times EF_{weighted,i}$$

$$EF_{weighted,i} = [(\%P_{c,i} \times EF_c) + (\%P_{unc,i} \times EF_{unc})]$$

15 where,

- 16 E_i = Annual N₂O Emissions for year i (kg/yr)
- 17 P_i = Annual nitric acid production for year i (metric tons HNO₃)
- 18 EF_{weighted,i} = Weighted N₂O emission factor for year i (kg N₂O/metric ton HNO₃)
- 19 %P_{c,i} = Percent national production of HNO₃ with N₂O abatement technology (%)
- 20 EF_c = N₂O emission factor, with abatement technology (kg N₂O/metric ton HNO₃)
- 21 %P_{unc,i} = Percent national production of HNO₃ without N₂O abatement technology (%)
- 22 EF_{unc} = N₂O emission factor, without abatement technology (kg N₂O/metric ton HNO₃)
- 23 i = year from 1990 through 2009

- 24
- 25 • For 2009: Weighted N₂O emission factor = 5.46 kg N₂O/metric ton HNO₃.
- 26 • For 1990 through 2008: Weighted N₂O emission factor = 5.66 kg N₂O/metric ton HNO₃.
- 27

28 Nitric acid production data for the United States for 1990 through 2009 were obtained from the U.S. Census
 29 Bureau (U.S. Census Bureau 2008, 2009, 2010a, 2010b) (see Table 4-31). EPA used GHGRP facility-level information
 30 to verify that all reported N₂O abatement equipment were incorporated into the estimation of N₂O emissions from
 31 nitric acid production over the full time series (EPA 2021).

32 **Table 4-31: Nitric Acid Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Production (kt)	7,200	6,710	8,210	8,080	7,970	7,800	7,860

33 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
 34 through 2022. The methodology for nitric acid production spliced activity data from two different sources: U. S.
 35 Census Bureau production data for 1990 through 2009 and GHGRP production data starting in 2010. Consistent
 36 with the 2006 IPCC Guidelines, the overlap technique was applied to compare the two data sets for years where
 37 there was overlap, with findings that the data sets were consistent and adjustments were not needed.

Uncertainty – TO BE UPDATED FOR FINAL REPORT

Uncertainty associated with the parameters used to estimate N₂O emissions includes the share of U.S. nitric acid production attributable to each emission abatement technology (i.e., utilization) over the time series (especially prior to 2010), and the associated emission factors applied to each abatement technology type. While some information has been obtained through outreach with industry associations, limited information is available over the time series (especially prior to 2010) for a variety of facility level variables, including plant-specific production levels, plant production technology (e.g., low or high pressure, etc.), and abatement technology destruction and removal efficiency rates. Production data prior to 2010 were obtained from National Census Bureau, which does not provide uncertainty estimates with their data. Facilities reporting to EPA’s GHGRP must measure production using equipment and practices used for accounting purposes. While emissions are often directly proportional to production, the emission factor for individual facilities can vary significantly from year to year due to site-specific fluctuations in ambient temperature and humidity, catalyst age and condition, nitric acid production process changes, the addition or removal of abatement technologies, and the number of nitric acid trains at the facility. At this time, EPA does not estimate uncertainty of the aggregated facility-level information. As noted in the QA/QC and verification section below, EPA verifies annual facility-level reports through a multi-step process (e.g., combination of electronic checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent. The annual production reported by each nitric acid facility under EPA’s GHGRP and then aggregated to estimate national N₂O emissions is assumed to have low uncertainty. EPA assigned an uncertainty range of ±5 percent for facility-reported N₂O emissions, and using this suggested uncertainty provided in section 3.4.3.2 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023). EPA assigned an uncertainty range of ±2 percent for nitric acid production, and using this suggested uncertainty provided in section 3.3.3.2 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023).

The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-32. Nitrous oxide emissions from nitric acid production were estimated to be between 7.5 and 8.3 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 5 percent below to 5 percent above the 2021 emissions estimate of 7.9 MMT CO₂ Eq.

Table 4-32: Approach 2 Quantitative Uncertainty Estimates for N₂O Emissions from Nitric Acid Production (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Nitric Acid Production	N ₂ O	7.9	7.5	8.3	-5%	+5%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of the *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details). More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to nitric acid facilities can be found under Subpart V: Nitric Acid Production of the GHGRP regulation (40 CFR Part 98).³⁴

³⁴ See Subpart V monitoring and reporting regulation http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

1 The main QA/QC activities are related to annual performance testing, which must follow either EPA Method 320 or
2 ASTM D6348-03. EPA verifies annual facility-level GHGRP reports through a multi-step process that is tailored to
3 the Subpart (e.g., combination of electronic checks including range checks, statistical checks, algorithm checks,
4 year-to-year comparison checks, along with manual reviews) to identify potential errors and ensure that data
5 submitted to EPA are accurate, complete, and consistent. Based on the results of the verification process, EPA
6 follows up with facilities to resolve mistakes that may have occurred (EPA 2015).³⁵ EPA's review of observed
7 trends noted that while emissions have generally mirrored production, in 2015 and 2019 nitric acid production
8 decreased compared to the previous year and emissions increased. While review is ongoing, based on feedback
9 from the verification process to date, these changes are due to facility-specific changes (e.g., in the nitric
10 production process and management of abatement equipment).

11 Recalculations Discussion

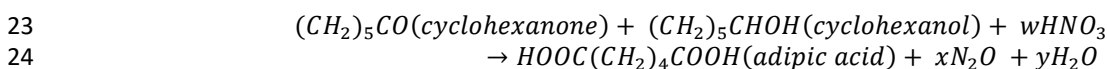
12 No recalculations were performed for the 1990 through 2021 portion of the time series.

13 Planned Improvements

14 Pending resources, EPA is considering a near-term improvement to both review and refine quantitative uncertainty
15 estimates and the associated qualitative discussion.

16 4.8 Adipic Acid Production (CRT Source 17 Category 2B3)

18 Adipic acid is produced through a two-stage process during which nitrous oxide (N₂O) is generated in the second
19 stage. The first stage of manufacturing usually involves the oxidation of cyclohexane to form a
20 cyclohexanone/cyclohexanol mixture. The second stage involves oxidizing this mixture with nitric acid to produce
21 adipic acid. Nitrous oxide is generated as a byproduct of the nitric acid oxidation stage and is emitted in the waste
22 gas stream (Thiemens and Trogler 1991). The second stage is represented by the following chemical reaction:



25 Process emissions from the production of adipic acid vary with the types of technologies and level of emission
26 controls employed by a facility. In 1990, two major adipic acid-producing plants had N₂O abatement technologies
27 in place and, as of 1998, three major adipic acid production facilities had control systems in place (Reimer et al.
28 1999). In 2022, thermal reduction was applied as an N₂O abatement measure at one adipic acid facility (EPA 2023).
29 Emissions from fuels consumed for energy purposes during the production of adipic acid are accounted for in the
30 Energy chapter.

31 Worldwide, only a few adipic acid plants exist. The United States, Europe, and China are the major producers, with
32 the United States accounting for the largest share of global adipic acid production capacity in recent years. In 2022,
33 the United States had two companies with a total of two adipic acid production facilities (one in Texas and one in
34 Florida), following the ceased operations of a third major production facility at the end of 2015 (EPA 2023).

³⁵ See GHGRP Verification Factsheet https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 Adipic acid is a white crystalline solid used in the manufacture of synthetic fibers, plastics, coatings, urethane
 2 foams, elastomers, and synthetic lubricants. Commercially, it is the most important of the aliphatic dicarboxylic
 3 acids, which are used to manufacture polyesters. Eighty-four percent of all adipic acid produced in the United
 4 States is used in the production of nylon 6,6; 9 percent is used in the production of polyester polyols; 4 percent is
 5 used in the production of plasticizers; and the remaining 4 percent is accounted for by other uses, including
 6 unsaturated polyester resins and food applications (ICIS 2007). Food grade adipic acid is used to provide some
 7 foods with a “tangy” flavor (Thiemens and Trogler 1991).

8 Compared to 1990, national adipic acid production in 2022 has increased by 3 percent to approximately 780,000
 9 metric tons (ACC 2023). Nitrous oxide emissions from adipic acid production were estimated to be 2.1 MMT CO₂
 10 Eq. (8 kt N₂O) in 2022 and are summarized in Table 4-33 and Table 4-34. Over the period 1990 through 2022,
 11 facilities have reduced emissions by 84.5 percent due to the widespread installation of pollution control measures
 12 in the late 1990s. The main reason for the 68 percent decrease in N₂O emissions from adipic acid production
 13 between 2021 and 2022 is increased utilization of N₂O abatement equipment at one adipic acid production facility.

14 Significant changes in the amount of time that the N₂O abatement device at one facility was in operation has been
 15 the main cause of fluctuating emissions in recent years. These fluctuations are most evident for years where trends
 16 in emissions and adipic acid production were not directly proportional: (1) between 2016 and 2017, (2) between
 17 2017 and 2018, (3) between 2019 and 2020, (4) between 2020 and 2021, and (5) between 2021 and 2022. As
 18 noted above, changes in control measures and abatement technologies at adipic acid production facilities,
 19 including maintenance of equipment, can result in annual emission fluctuations. Little additional information is
 20 available on drivers of trends, and the amount of adipic acid produced is not reported under EPA’s GHGRP.

21 **Table 4-33: N₂O Emissions from Adipic Acid Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Adipic Acid Production	13.5	6.3	9.3	4.7	7.4	6.6	2.1

22 **Table 4-34: N₂O Emissions from Adipic Acid Production (kt N₂O)**

Year	1990	2005	2018	2019	2020	2021	2022
Adipic Acid Production	51	24	35	18	28	25	8

23 Methodology and Time-Series Consistency

24 Emissions of N₂O from adipic acid production are estimated using methods provided by the *2006 IPCC Guidelines*,
 25 in accordance with the IPCC methodological decision tree and available data. For 2010 through 2022, a Tier 3
 26 method was used to estimate emissions. For 1990 through 2009, emissions are estimated using both Tier 2 and
 27 Tier 3 methods. Due to confidential business information (CBI), plant names are not provided in this section;
 28 therefore, the four adipic acid-producing facilities that have operated over the time series will be referred to as
 29 Plants 1 through 4. As noted above, one currently operating facility uses thermal reduction as an N₂O abatement
 30 technology.

31 2010 through 2022

32 All emission estimates for 2010 through 2022 were obtained through analysis of GHGRP data (EPA 2010 through
 33 2023). Facility-level greenhouse gas emissions data were obtained from EPA’s GHGRP for the years 2010 through
 34 2022 (EPA 2010 through 2023) and aggregated to national N₂O emissions. Consistent with IPCC Tier 3 methods, all
 35 adipic acid production facilities are required to either calculate N₂O emissions using a facility-specific emission

1 factor developed through annual performance testing under typical operating conditions or directly measure N₂O
2 emissions using monitoring equipment.³⁶

3 **1990 through 2009**

4 For years 1990 through 2009, which were prior to EPA’s GHGRP reporting, for both Plants 1 and 2, emission
5 estimates were obtained directly from the plant engineers and account for reductions due to control systems in
6 place at these plants during the time series. These prior estimates are considered CBI and hence are not published
7 (Desai 2010, 2011). These estimates were based on continuous process monitoring equipment installed at the two
8 facilities.

9 For Plant 4, 1990 through 2009 N₂O emissions were estimated using the following Tier 2 equation from the 2006
10 *IPCC Guidelines*:

11 **Equation 4-5: 2006 IPCC Guidelines Tier 2: N₂O Emissions From Adipic Acid Production** 12 **(Equation 3.8)**

$$13 \quad E_{aa} = Q_{aa} \times EF_{aa} \times (1 - [DF \times UF])$$

14 where,

15 E_{aa} = N₂O emissions from adipic acid production, metric tons

16 Q_{aa} = Quantity of adipic acid produced, metric tons

17 EF_{aa} = Emission factor, metric ton N₂O/metric ton adipic acid produced

18 DF = N₂O destruction factor

19 UF = Abatement system utility factor

20 The adipic acid production is multiplied by an emission factor (i.e., N₂O emitted per unit of adipic acid produced),
21 which has been estimated to be approximately 0.3 metric tons of N₂O per metric ton of product (IPCC 2006). The
22 “N₂O destruction factor” in the equation represents the percentage of N₂O emissions that are destroyed by the
23 installed abatement technology. The “abatement system utility factor” represents the percentage of time that the
24 abatement equipment operates during the annual production period. Plant-specific production data for Plant 4
25 were obtained across the time series through personal communications (Desai 2010, 2011). The plant-specific
26 production data were then used for calculating emissions as described above.

27 For Plant 3, 2005 through 2009 emissions were obtained directly from the plant (Desai 2010, 2011). For 1990
28 through 2004, emissions were estimated using plant-specific production data and the IPCC factors as described
29 above for Plant 4. Plant-level adipic acid production for 1990 through 2003 was estimated by allocating national
30 adipic acid production data to the plant level using the ratio of known plant capacity to total national capacity for
31 all U.S. plants (ACC 2023; CMR 2001, 1998; CW 1999; C&EN 1992 through 1995). For 2004, actual plant production
32 data were obtained and used for emission calculations (CW 2005).

33 Plant capacities for 1990 through 1994 were obtained from *Chemical & Engineering News*, “Facts and Figures” and
34 “Production of Top 50 Chemicals” (C&EN 1992 through 1995). Plant capacities for 1995 and 1996 were kept the
35 same as 1994 data. The 1997 plant capacities were taken from *Chemical Market Reporter*, “Chemical Profile: Adipic
36 Acid” (CMR 1998). The 1998 plant capacities for all four plants and 1999 plant capacities for three of the plants
37 were obtained from *Chemical Week*, Product Focus: Adipic Acid/Adiponitrile (CW 1999). Plant capacities for the
38 year 2000 for three of the plants were updated using *Chemical Market Reporter*, “Chemical Profile: Adipic Acid”

³⁶ Facilities must use standard methods, either EPA Method 320 or ASTM D6348-03 for annual performance testing, and must follow associated QA/QC procedures during these performance tests consistent with category-specific QC of direct emission measurements.

1 (CMR 2001). For 2001 through 2003, the plant capacities for three plants were held constant at year 2000
 2 capacities. Plant capacity for 1999 to 2003 for the one remaining plant was kept the same as 1998.
 3 National adipic acid production data (see Table 4-35) from 1990 through 2022 were obtained from the American
 4 Chemistry Council (ACC 2023).

5 **Table 4-35: Adipic Acid Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Production (kt)	755	865	825	810	710	760	780

6 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
 7 through 2022. The methodology for adipic acid production spliced activity data from multiple sources: plant-
 8 specific emissions data and publicly available plant capacity data for 1990 through 2009 and GHGRP emission data
 9 starting in 2010. Consistent with the *2006 IPCC Guidelines*, the overlap technique was applied to compare the two
 10 data sets for years where there was overlap, with findings that the data sets were consistent and adjustments
 11 were not needed.

12 Uncertainty – TO BE UPDATED FOR FINAL REPORT

13 Uncertainty associated with N₂O emission estimates includes the methods used by companies to monitor and
 14 estimate emissions. While some information has been obtained through outreach with facilities, limited
 15 information is available over the time series on these methods, abatement technology destruction and removal
 16 efficiency rates, and plant-specific production levels. EPA assigned an uncertainty range of ±5 percent for facility-
 17 reported N₂O emissions, and using this suggested uncertainty provided in section 3.4.3.2 of the *2006 IPCC*
 18 *Guidelines* is appropriate based on expert judgment (RTI 2023).

19 The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-36. Nitrous oxide
 20 emissions from adipic acid production for 2022 were estimated to be between 6.3 and 6.9 MMT CO₂ Eq. at the 95
 21 percent confidence level. These values indicate a range of approximately 5 percent below to 5 percent above the
 22 2021 emission estimate of 6.6 MMT CO₂ Eq.

23 **Table 4-36: Approach 2 Quantitative Uncertainty Estimates for N₂O Emissions from Adipic**
 24 **Acid Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.)			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Adipic Acid Production	N ₂ O	6.6	6.3	6.9	-5%	+5%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

25 QA/QC and Verification

26 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
 27 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of the *2006 IPCC Guidelines* as described in the
 28 introduction of the IPPU chapter (see Annex 8 for more details).

1 More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to adipic acid facilities
2 can be found under Subpart E (Adipic Acid Production) of the GHGRP regulation (40 CFR Part 98).³⁷ The main
3 QA/QC activities are related to annual performance testing, which must follow either EPA Method 320 or ASTM
4 D6348-03. EPA verifies annual facility-level GHGRP reports through a multi-step process (e.g., combination of
5 electronic checks and manual reviews) to identify potential errors and ensure that data submitted to EPA are
6 accurate, complete, and consistent (EPA 2015).³⁸ Based on the results of the verification process, EPA follows up
7 with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a
8 number of general and category-specific QC procedures, including range checks, statistical checks, algorithm
9 checks, and year-to-year comparisons of reported data.

10 Recalculations Discussion

11 No recalculations were performed for the 1990 through 2021 portion of the time series.

12 Planned Improvements

13 EPA reviewed GHGRP facility reported information on the date of abatement technology installation in order to
14 better reflect trends and changes in emissions abatement within the industry across the time series. The facility
15 using the facility-specific emission factor developed through annual performance testing has reported no
16 installation and no utilization of N₂O abatement technology. The facility using direct measurement of N₂O
17 emissions has reported the use of thermal reduction as an N₂O abatement technology but is not required to report
18 the date of installation.

19 4.9 Caprolactam, Glyoxal and Glyoxylic 20 Acid Production (CRT Source Category 2B4)

21 Caprolactam

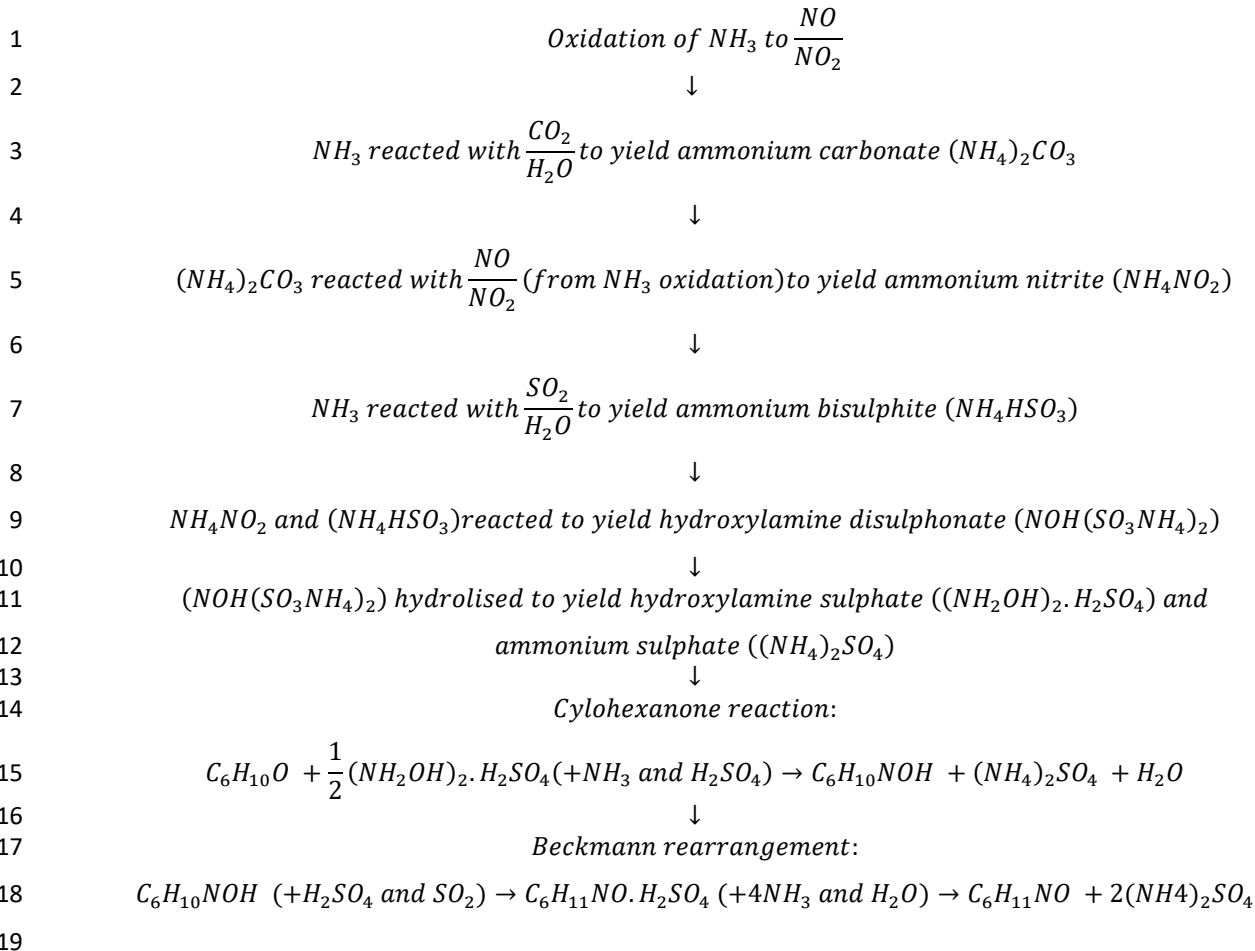
22 Caprolactam (C₆H₁₁NO) is a colorless monomer produced for nylon-6 fibers and plastics. A substantial proportion
23 of the fiber is used in carpet manufacturing. Most commercial processes used for the manufacture of caprolactam
24 begin with benzene, but toluene can also be used. The production of caprolactam can give rise to emissions of
25 nitrous oxide (N₂O).

26 During the production of caprolactam, emissions of N₂O can occur from the ammonia oxidation step, emissions of
27 carbon dioxide (CO₂) from the ammonium carbonate step, emissions of sulfur dioxide (SO₂) from the ammonium
28 bisulfite step, and emissions of non-methane volatile organic compounds (NMVOCs). Emissions of CO₂, SO₂ and
29 NMVOCs from the conventional process are unlikely to be significant in well-managed plants. Modified
30 caprolactam production processes are primarily concerned with elimination of the high volumes of ammonium
31 sulfate that are produced as a byproduct of the conventional process (IPCC 2006).

32 In the most commonly used process where caprolactam is produced from benzene, benzene is hydrogenated to
33 cyclohexane which is then oxidized to produce cyclohexanone (C₆H₁₀O). The classical route (Raschig process) and
34 basic reaction equations for production of caprolactam from cyclohexanone are (IPCC 2006):

³⁷ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

³⁸ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.



20 In 2004, three facilities produced caprolactam in the United States (ICIS 2004). Another facility, Evergreen
 21 Recycling, was in operation from 2000 to 2001 (ICIS 2004; Textile World 2000) and from 2007 through 2015 (Shaw
 22 2015). Caprolactam production at Fibrant LLC (formerly DSM Chemicals) in Georgia ceased in 2018 (Cline 2019). As
 23 of 2022, two companies in the United States produced caprolactam at two facilities: AdvanSix (formerly
 24 Honeywell) in Virginia (AdvanSix 2023) and BASF in Texas (BASF 2023).

25 Nitrous oxide emissions from caprolactam production in the United States were estimated to be 1.3 MMT CO₂ Eq.
 26 (5 kt N₂O) in 2022 and are summarized in Table 4-37 and Table 4-38. National emissions from caprolactam
 27 production decreased by approximately 10.5 percent over the period of 1990 through 2022. Emissions in 2022
 28 increased by approximately 9.8 percent from the 2021 levels. This annual increase returned caprolactam
 29 production to levels consistent with 2017 before the COVID-19 pandemic.

30 **Table 4-37: N₂O Emissions from Caprolactam Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Caprolactam Production	1.5	1.9	1.3	1.2	1.1	1.2	1.3

31 **Table 4-38: N₂O Emissions from Caprolactam Production (kt N₂O)**

Year	1990	2005	2018	2019	2020	2021	2022
Caprolactam Production	6	7	5	5	4	5	5

1 Glyoxal

2 Glyoxal is mainly used as a crosslinking agent for vinyl acetate/acrylic resins, disinfectant, gelatin hardening agent,
3 textile finishing agent (permanent-press cotton, rayon fabrics), and wet-resistance additive (paper coatings) (IPCC
4 2006). It is also used for enhanced oil-recovery. It is produced from oxidation of acetaldehyde with concentrated
5 nitric acid, or from the catalytic oxidation of ethylene glycol, and N₂O is emitted in the process of oxidation of
6 acetaldehyde.

7 Glyoxal (ethanedial) (C₂H₂O₂) is produced from oxidation of acetaldehyde (ethanal) (C₂H₄O) with concentrated
8 nitric acid (HNO₃). Glyoxal can also be produced from catalytic oxidation of ethylene glycol (ethanediol)
9 (CH₂OHCH₂OH).

10 Glyoxylic Acid

11 Glyoxylic acid is produced by nitric acid oxidation of glyoxal. Glyoxylic acid is used for the production of synthetic
12 aromas, agrochemicals, and pharmaceutical intermediates (IPCC 2006).

13 Preliminary data suggests that glyoxal and glyoxylic acid may be produced in small quantities domestically but are
14 largely imported to the United States. EPA does not currently estimate the emissions associated with the
15 production of glyoxal and glyoxylic acid because activity data are not available. See Annex 5 for more information.

16 Methodology and Time-Series Consistency

17 Emissions of N₂O from the production of caprolactam are calculated using the Tier 1 methodology from *the 2006*
18 *IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available data. The Tier 1 equation
19 is as follows:

20 Equation 4-6: 2006 IPCC Guidelines Tier 1: N₂O Emissions From Caprolactam Production 21 (Equation 3.9)

$$22 E_{N_2O} = EF \times CP$$

23 where,

24 E_{N_2O} = Annual N₂O Emissions (kg)

25 EF = N₂O emission factor (default) (kg N₂O/metric ton caprolactam produced)

26 CP = Caprolactam production (metric tons)

27 During the caprolactam production process, N₂O is generated as a byproduct of the high temperature catalytic
28 oxidation of ammonia (NH₃), which is the first reaction in the series of reactions to produce caprolactam. The
29 amount of N₂O emissions can be estimated based on the chemical reaction shown above. Based on this formula,
30 which is consistent with an IPCC Tier 1 approach, approximately 111.1 metric tons of caprolactam are required to
31 generate one metric ton of N₂O, resulting in an emission factor of 9.0 kg N₂O per metric ton of caprolactam (IPCC
32 2006). When applying the Tier 1 method, the *2006 IPCC Guidelines* state that it is good practice to assume that
33 there is no abatement of N₂O emissions and to use the highest default emission factor available in the guidelines.
34 In addition, EPA did not find support for the use of secondary catalysts to reduce N₂O emissions, such as those
35 employed at nitric acid plants.

36 The activity data for caprolactam production (see Table 4-39) from 1990 to 2022 were obtained from the American
37 Chemistry Council's *Guide to the Business of Chemistry* (ACC 2023). EPA will continue to analyze and assess
38 alternative sources of production data as a quality control measure.

1 **Table 4-39: Caprolactam Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Production (kt)	626	795	530	515	480	510	560

2 Carbon dioxide and methane (CH₄) emissions may also occur from the production of caprolactam, but currently the
3 IPCC does not have methodologies for calculating these emissions associated with caprolactam production.

4 Methodological approaches, consistent with the *2006 IPCC Guidelines*, have been applied to the entire time series
5 to ensure consistency in emissions from 1990 through 2022.

6 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

7 Estimation of emissions of N₂O from caprolactam production can be treated as analogous to estimation of
8 emissions of N₂O from nitric acid production. Both production processes involve an initial step of NH₃ oxidation,
9 which is the source of N₂O formation and emissions (IPCC 2006). Therefore, uncertainties for the default emission
10 factor values in the *2006 IPCC Guidelines* are an estimate based on default values for nitric acid plants. In general,
11 default emission factors for gaseous substances have higher uncertainties because mass values for gaseous
12 substances are influenced by temperature and pressure variations and gases are more easily lost through process
13 leaks. The default values for caprolactam production have a relatively high level of uncertainty due to the limited
14 information available (IPCC 2006). EPA assigned uncertainty bounds of ±5 percent for caprolactam production,
15 based on expert judgment. EPA assigned an uncertainty range of ±40 percent for the N₂O emission factor, and
16 using this suggested uncertainty provided in Section 3.5.2.1 of the *2006 IPCC Guidelines* is appropriate based on
17 expert judgment (RTI 2023).

18 The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-40. Nitrous oxide
19 emissions from Caprolactam, Glyoxal and Glyoxylic Acid Production for 2022 were estimated to be between 0.8
20 and 1.6 MMT CO₂ Eq. at the 95 percent confidence level. These values indicate a range of approximately 32
21 percent below to 32 percent above the 2022 emission estimate of 1.2 MMT CO₂ Eq.

22 **Table 4-40: Approach 2 Quantitative Uncertainty Estimates for N₂O Emissions from**
23 **Caprolactam, Glyoxal and Glyoxylic Acid Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Caprolactam Production	N ₂ O	1.2	0.8	1.6	-32%	+32%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

24 **QA/QC and Verification**

25 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
26 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of the *2006 IPCC Guidelines* as described in the
27 introduction of the IPPU chapter (see Annex 8 for more details).

28 **Recalculations Discussion**

29 Recalculations were performed for 2020 and 2021 to reflect updated caprolactam production data from the
30 American Chemistry Council's *Guide to the Business of Chemistry* (ACC 2023). Compared to the previous *Inventory*,
31 annual N₂O emissions decreased by 2 percent in 2020 and 2021, with a decrease of 0.02 MMT CO₂ Eq. in 2020 and
32 2021.

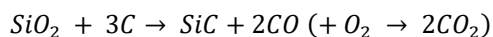
1 Planned Improvements

2 Pending resources, EPA will research other available datasets for caprolactam production and industry trends,
3 including facility-level data. EPA continues to research available activity data and emissions associated with the
4 production of glyoxal and glyoxylic acid. Preliminary data suggests that glyoxal and glyoxylic acid may be produced
5 in small quantities domestically but are largely imported to the United States. See Annex 5 for more information.
6 This planned improvement is subject to data availability and will be implemented in the medium- to long-term.

7 4.10 Carbide Production and Consumption 8 (CRT Source Category 2B5)

9 Carbon dioxide (CO₂) and methane (CH₄) are emitted from the production of silicon carbide (SiC), a material used
10 for industrial abrasive applications as well as metallurgical and other non-abrasive applications in the United
11 States. Emissions from fuels consumed for energy purposes during the production of silicon carbide are accounted
12 for in the Energy chapter. Additionally, some metallurgical and non-abrasive applications of SiC are emissive at
13 high temperatures due to the SiC oxidation temperature (Biscay 2021). While emissions should be accounted for
14 where they occur based on *2006 IPCC Guidelines*, emissions from SiC consumption are accounted for here until
15 additional data on SiC consumption by end-use are available.

16 To produce SiC, silica sand or quartz (SiO₂) is reacted with carbon (C) in the form of petroleum coke. A portion
17 (about 35 percent) of the carbon contained in the petroleum coke is retained in the SiC. The remaining carbon is
18 emitted as CO₂, CH₄, or carbon monoxide (CO). The overall reaction is shown below, but in practice, it does not
19 proceed according to stoichiometry:



21 Carbon dioxide and CH₄ are also emitted during the production of calcium carbide, a chemical used to produce
22 acetylene. Carbon dioxide is implicitly accounted for in the storage factor calculation for the non-energy use of
23 petroleum coke in the Energy chapter.

24 Markets for manufactured abrasives, including SiC, are heavily influenced by activity in the U.S. manufacturing
25 sector, especially in the aerospace, automotive, furniture, housing, and steel manufacturing sectors. Specific
26 applications of abrasive-grade SiC in 2018 included antislip abrasives, blasting abrasives, bonded abrasives, coated
27 abrasives, polishing and buffing compounds, tumbling media, and wire-sawing abrasives (USGS 2021).
28 Approximately 50 percent of SiC is used in metallurgical applications, which include primarily iron and steel
29 production, and other non-abrasive applications, which include use in advanced or technical ceramics and
30 refractories (USGS 2023a; Washington Mills 2023).

31 As a result of the economic downturn in 2008 and 2009, demand for SiC decreased in those years. Low-cost
32 imports, particularly from China, combined with high relative operating costs for domestic producers, continue to
33 put downward pressure on the production of SiC in the United States. Consumption of SiC in the United States has
34 recovered somewhat from its low in 2009 to 2020; 2021 consumption data was withheld to avoid disclosing
35 company proprietary data (USGS 1991b through 2021), and 2022 USGS data has not yet been released.

36 Silicon carbide was manufactured by two facilities in the United States, one of which produced primarily non-
37 abrasive SiC (USGS 2021). USGS production values for the United States consists of SiC used for abrasives and for
38 metallurgical and other non-abrasive applications (USGS 2021). During the COVID-19 pandemic in 2020, the U.S.
39 Department of Homeland Security considered abrasives manufacturing part of the critical manufacturing sector,
40 and as a result, pandemic “stay-at-home” orders issued in March 2020 did not affect the abrasives manufacturing
41 industry. These plants remained at full operation (USGS 2021a). In 2022, imports and exports continued to recover
42 from the negative effects of the COVID-19 pandemic (USGS 2023b). Consumption of SiC increased by

1 approximately 27 percent from 2021 to 2022, rising above pre-pandemic levels (U.S. Census Bureau 2005 through
2 2022).

3 Carbon dioxide emissions from SiC production and consumption in 2022 were 0.2 MMT CO₂ Eq. (210 kt CO₂), which
4 are about 14 percent lower than emissions in 1990 (see Table 4-41 and Table 4-42). Approximately 50 percent of
5 these emissions resulted from SiC production, while the remainder resulted from SiC consumption. Methane
6 emissions from SiC production in 2022 were 0.01 MMT CO₂ Eq. (0.5 kt CH₄) (see Table 4-41 and Table 4-42). These
7 tables indicate minor changes in emissions in recent years.

8 **Table 4-41: CO₂ and CH₄ Emissions from Silicon Carbide Production and Consumption (MMT**
9 **CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Production							
CO ₂	0.2	0.1	0.1	0.1	0.1	0.1	0.1
CH ₄	+	+	+	+	+	+	+
Consumption							
CO ₂	0.1	0.1	0.1	0.1	0.1	0.1	0.1
Total	0.2	0.2	0.2	0.2	0.2	0.2	0.2

+ Does not exceed 0.05 MMT CO₂ Eq.

Note: Totals may not sum due to independent rounding.

10 **Table 4-42: CO₂ and CH₄ Emissions from Silicon Carbide Production and Consumption (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Production							
CO ₂	170	92	92	92	92	92	105
CH ₄	1	+	+	+	+	+	+
Consumption							
CO ₂	73	121	93	84	62	80	105

+ Does not exceed 0.5 kt.

Note: Totals by gas may not sum due to independent rounding.

11 Methodology and Time-Series Consistency

12 Emissions of CO₂ and CH₄ from the production of SiC are calculated using the Tier 1 method from the *2006 IPCC*
13 *Guidelines*, in accordance with the IPCC methodological decision tree and available data. Emissions of CO₂ and CH₄
14 from the consumption of SiC are a country-specific source calculated using a country-specific methodology based
15 on available data. The *2006 IPCC Guidelines* do not provide guidance for estimating emissions from use of SiC or SiC
16 consumption, but the country-specific methodology used is compatible with the *2006 IPCC Guidelines* and
17 consistent with a Tier 1 approach. Annual estimates of SiC production were multiplied by the default emission
18 factors, as shown below:

19 **Equation 4-78: 2006 IPCC Guidelines Tier 1: Emissions from Carbide Production (Equation**
20 **3.11)**

$$E_{sc,CO_2} = EF_{sc,CO_2} \times Q_{sc}$$

$$E_{sc,CH_4} = EF_{sc,CH_4} \times Q_{sc} \times \left(\frac{1 \text{ metric ton}}{1000 \text{ kg}} \right)$$

23 where,

24 E_{sc,CO_2} = CO₂ emissions from production of SiC, metric tons

25 EF_{sc,CO_2} = Emission factor for production of SiC, metric ton CO₂/metric ton SiC

26 Q_{sc} = Quantity of SiC produced, metric tons

1 E_{sc,CH_4} = CH₄ emissions from production of SiC, metric tons

2 EF_{sc,CH_4} = Emission factor for production of SiC, kilogram CH₄/metric ton SiC

3 Emission factors were taken from the 2006 IPCC Guidelines:

- 4 • 2.62 metric tons CO₂/metric ton SiC
- 5 • 11.6 kg CH₄/metric ton SiC

6 Production data includes silicon carbide manufactured for abrasive applications as well as for metallurgical and
7 other non-abrasive applications (USGS 2021).

8 Silicon carbide industrial abrasives production data for 1990 through 2022 were obtained from the U.S. Geological
9 Survey (USGS) *Minerals Yearbook: Manufactured Abrasives* (USGS 1991a through 2021; USGS 2023a). Silicon
10 carbide production data published by USGS have been rounded to the nearest 5,000 metric tons to avoid
11 disclosing company proprietary data. For the period 1990 through 2001, reported USGS production data include
12 production from two facilities located in Canada that ceased operations in 1995 and 2001. Using SiC production
13 data from Canada (ECCC 2022), U.S. SiC production for 1990 through 2001 was adjusted to reflect only U.S.
14 production.

15 Emissions from SiC consumption are estimated for the entire time series using USGS consumption data (USGS
16 1991b through 2021) and data from the U.S. International Trade Commission (USITC) database on net imports and
17 exports of SiC (U.S. Census Bureau 2005 through 2022) (Table 4-43). Total annual SiC consumption (utilization) was
18 estimated by subtracting annual exports of SiC from the annual total of national SiC production and annual
19 imports.

20 The method for estimating emissions of CO₂ from SiC consumption from metallurgical uses and non-abrasive uses
21 are similar:

22 Emissions from metallurgical uses were calculated by multiplying the annual utilization of SiC for
23 metallurgical uses (reported annually in the USGS Minerals Yearbook: Silicon) by the carbon content of SiC
24 (30.0 percent), which was determined according to the molecular weight ratio of SiC. Because USGS
25 withheld consumption data for metallurgical uses from publication for 2017, 2018, and 2021 due to
26 concerns of disclosing company-specific sensitive information, SiC consumption for 2017 and 2018 were
27 estimated using 2016 values, and SiC consumption for 2021 was estimated using the 2020 value.
28 Additionally, as the USGS has not yet released the 2022 data, SiC consumption for 2022 was estimated
29 using the 2020 value.

- 30 • Emissions of CO₂ from SiC consumption for other non-abrasive uses were calculated by multiplying the
31 annual SiC consumption for non-abrasive uses by the carbon content of SiC (30 percent). The annual SiC
32 consumption for non-abrasive uses was calculated by multiplying the annual SiC consumption (production
33 plus net imports) by the percentage used in metallurgical and other non-abrasive uses (50 percent) (USGS
34 1991a through 2021; USGS 2023a) and then subtracting the SiC consumption for metallurgical use.

35 The petroleum coke portion of the total CO₂ process emissions from silicon carbide production is adjusted for
36 within the Energy chapter, as these fuels were consumed during non-energy related activities. Additional
37 information on the adjustments made within the Energy sector for non-energy use of fuels is described in both the
38 Methodology section of CO₂ from Fossil Fuel Combustion (Section 3.1) and Annex 2.1, Methodology for Estimating
39 Emissions of CO₂ from Fossil Fuel Combustion.

40 **Table 4-43: Production and Consumption of Silicon Carbide (Metric Tons)**

Year	1990	2005	2018	2019	2020	2021	2022
Production	65,000	35,000	35,000	35,000	35,000	35,000	40,000
Consumption	132,465	220,149	168,526	152,412	113,756	146,312	191,133

1 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
 2 through 2022.

3 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

4 Silicon carbide production data published by the USGS is rounded to the nearest 5,000 tons and has been
 5 consistently reported at 35,000 tons since 2003 to avoid disclosure of company proprietary data. This translates to
 6 an uncertainty range of ± 7 percent for SiC production (USGS 2021). There is uncertainty associated with the
 7 emission factors used because they are based on stoichiometry as opposed to monitoring of actual SiC production
 8 plants. An alternative is to calculate emissions based on the quantity of petroleum coke used during the
 9 production process rather than on the amount of silicon carbide produced; however, these data were not
 10 available. For CH₄, there is also uncertainty associated with the hydrogen-containing volatile compounds in the
 11 petroleum coke (IPCC 2006). EPA assigned an uncertainty of ± 10 percent for the Tier 1 CO₂ and CH₄ emission
 12 factors for the SiC production processes, and using this suggested uncertainty provided in Section 3.6.3.1 of the
 13 *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023). There is also uncertainty associated with
 14 the use or destruction of CH₄ generated from the process, in addition to uncertainty associated with levels of
 15 production, net imports, consumption levels, and the percent of total consumption that is attributed to
 16 metallurgical and other non-abrasive uses. EPA assigned an uncertainty range of ± 5 percent for the primary data
 17 inputs for consumption (i.e., crude imports, ground and refined imports, crude exports, ground and refined
 18 exports, utilization [metallurgical applications]) to calculate overall uncertainty from SiC production, and using this
 19 suggested uncertainty provided in Section 3.6.3.2 of the *2006 IPCC Guidelines* is appropriate based on expert
 20 judgment (RTI 2023).

21 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-44. Silicon carbide
 22 production and consumption CO₂ emissions from 2021 were estimated to be between 10 percent below and 10
 23 percent above the emission estimate of 0.17 MMT CO₂ Eq. at the 95 percent confidence level. Silicon carbide
 24 production CH₄ emissions were estimated to be between 10 percent below and 11 percent above the emission
 25 estimate of 0.01 MMT CO₂ Eq. at the 95 percent confidence level.

26 **Table 4-44: Approach 2 Quantitative Uncertainty Estimates for CH₄ and CO₂ Emissions from**
 27 **Silicon Carbide Production and Consumption (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.)			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
Silicon Carbide Production and Consumption	CO ₂	0.17	0.16	0.19	-10%	+10%
Silicon Carbide Production	CH ₄	+	+	+	-10%	+11%

+ Does not exceed 0.05 MMT CO₂ Eq.

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

28 **QA/QC and Verification**

29 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
 30 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the
 31 introduction of the IPPU chapter (see Annex 8 for more details).

32 **Recalculations Discussion**

33 No recalculations were performed for the time series compared to the previous *Inventory*.

1 Planned Improvements

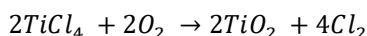
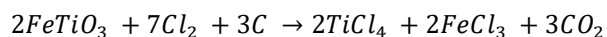
2 EPA is initiating research for data on SiC consumption by end-use for consideration in updating emissions
3 estimates from SiC consumption and to account for emissions where they occur. This planned improvement is
4 subject to data availability and will be implemented in the medium- to long-term given significance of emissions.

5 EPA has not integrated aggregated facility-level GHGRP information to inform estimates of CO₂ and CH₄ from SiC
6 production and consumption. The aggregated information (e.g., activity data and emissions) associated with silicon
7 carbide did not meet criteria to shield underlying confidential business information (CBI) from public disclosure.

8 EPA plans to examine the use of GHGRP silicon carbide emissions data for possible use in emission estimates
9 consistent with both Volume 1, Chapter 6 of the 2006 IPCC Guidelines and the latest IPCC guidance on the use of
10 facility-level data in national inventories. This planned improvement is ongoing and has not been incorporated into
11 this *Inventory* report. This is a long-term planned improvement.

12 4.11 Titanium Dioxide Production (CRT 13 Source Category 2B6)

14 Titanium dioxide (TiO₂) is manufactured using one of two processes: the chloride process and the sulfate process.
15 The chloride process uses petroleum coke and chlorine as raw materials and emits process-related carbon dioxide
16 (CO₂). Emissions from fuels consumed for energy purposes during the production of titanium dioxide are
17 accounted for in the Energy chapter. The sulfate process does not use petroleum coke or other forms of carbon as
18 a raw material and does not emit CO₂. The chloride process is based on the following chemical reactions and does
19 emit CO₂:



22 The carbon in the first chemical reaction is provided by petroleum coke, which is oxidized in the presence of the
23 chlorine and FeTiO₃ (rutile ore) to form CO₂. Since 2004, all TiO₂ produced in the United States has been produced
24 using the chloride process, and a special grade of “calcined” petroleum coke is manufactured specifically for this
25 purpose.

26 The principal use of TiO₂ is as a white pigment in paint, lacquers, and varnishes. It is also used as a pigment in the
27 manufacture of plastics, paper, and other products. In 2022, U.S. TiO₂ production totaled 1,100,000 metric tons
28 (USGS 2023b). Five plants produced TiO₂ in the United States in 2022.

29 Emissions of CO₂ from titanium dioxide production in 2022 were estimated to be 1.5 MMT CO₂ Eq. (1,474 kt CO₂),
30 which represents an increase of 23 percent since 1990 (see Table 4-45 and Table 4-46). Compared to 2021,
31 emissions from titanium dioxide production remained the same because production was consistent from 2021 to
32 2022. Annual production dipped in 2019 and 2020 and increased in 2021 and 2022.

33 **Table 4-45: CO₂ Emissions from Titanium Dioxide (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Titanium Dioxide	1.2	1.8	1.5	1.3	1.3	1.5	1.5

34 **Table 4-46: CO₂ Emissions from Titanium Dioxide (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Titanium Dioxide	1,195	1,755	1,541	1,340	1,340	1,474	1,474

1 Methodology and Time-Series Consistency

2 Emissions of CO₂ from TiO₂ production are calculated using a Tier 1 method from the *2006 IPCC Guidelines*, in
3 accordance with the IPCC methodological decision tree and available data. Annual national TiO₂ production is
4 multiplied by chloride process-specific emission factors provided by IPCC (IPCC 2006). The Tier 1 equation is as
5 follows:

6 Equation 4-9: 2006 IPCC Guidelines Tier 1: CO₂ Emissions from Titanium Production (Equation 7 3.12)

$$8 \quad E_{td} = EF_{td} \times Q_{td}$$

9 where,

10 E_{td} = CO₂ emissions from TiO₂ production, metric tons

11 EF_{td} = Emission factor (chloride process), metric ton CO₂/metric ton TiO₂

12 Q_{td} = Quantity of TiO₂ produced, metric tons

13 The petroleum coke portion of the total CO₂ process emissions from TiO₂ production is adjusted for within the
14 Energy chapter as these fuels were consumed during non-energy related activities. Additional information on the
15 adjustments made within the Energy sector for Non-Energy Use of Fuels is described in both the Methodology
16 section of CO₂ from Fossil Fuel Combustion (Section 3.1 Fossil Fuel Combustion) and Annex 2.1, Methodology for
17 Estimating Emissions of CO₂ from Fossil Fuel Combustion.

18 Data were obtained for the total amount of TiO₂ produced each year. For years prior to 2004, it was assumed that
19 TiO₂ was produced using the chloride process and the sulfate process in the same ratio as the ratio of the total U.S.
20 production capacity for each process. As of 2004, the last remaining sulfate process plant in the United States
21 closed; therefore, 100 percent of production since 2004 used the chloride process (USGS 2005). An emission factor
22 of 1.34 metric tons CO₂/metric ton TiO₂ was applied to the estimated chloride-process production (IPCC 2006). It
23 was assumed that all TiO₂ produced using the chloride process was produced using petroleum coke, although
24 some TiO₂ may have been produced with graphite or other carbon inputs.

25 The emission factor for the TiO₂ chloride process was taken from the *2006 IPCC Guidelines*. Titanium dioxide
26 production data and the percentage of total TiO₂ production capacity that used the chloride process for 1990
27 through 2018 (see Table 4-47) were obtained through the *U.S. Geological Survey (USGS) Minerals Yearbook:
28 Titanium* (USGS 1991 through 2022). Production data for 2019 and 2020 were obtained from the USGS Minerals
29 Yearbook: Titanium, advanced data release of the 2020 tables (USGS 2023a). Production data for 2020 and 2021
30 were obtained from the *Minerals Commodity Summaries: Titanium and Titanium Dioxide* (USGS 2023b).³⁹ Data on
31 the percentage of total TiO₂ production capacity that used the chloride process were not available for 1990
32 through 1993, so data from the 1994 USGS Minerals Yearbook were used for these years. Because a sulfate
33 process plant closed in September 2001, the chloride process percentage for 2001 was estimated (Gambogi 2002).
34 By 2002, only one sulfate process plant remained online in the United States, and this plant closed in 2004 (USGS
35 2005).

36 **Table 4-47: Titanium Dioxide Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Production	979	1,310	1,150	1,000	1,000	1,100	1,100

³⁹ EPA has not integrated aggregated facility-level GHGRP information for titanium dioxide production facilities (40 CFR Part 98 Subpart EE). The relevant aggregated information (activity data, emission factor) from these facilities did not meet criteria to shield underlying CBI from public disclosure.

1 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
 2 through 2022.

3 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

4 Each year, the USGS collects titanium industry data for titanium mineral and pigment production operations. If
 5 TiO₂ pigment plants do not respond, production from the operations is estimated based on prior year production
 6 levels and industry trends. Variability in response rates fluctuates from 67 to 100 percent of TiO₂ pigment plants
 7 over the time series. EPA currently uses an uncertainty range of ±5 percent for the primary data inputs (i.e., TiO₂
 8 production and chloride process capacity values) to calculate overall uncertainty from TiO₂ production, and using
 9 this suggested uncertainty provided in Section 3.7.3.2 of the *2006 IPCC Guidelines* is appropriate based on expert
 10 judgment (RTI 2023). Additionally, the EPA uses an uncertainty range of ±15 percent for the CO₂ chloride process
 11 carbon consumption rate, and using this uncertainty provided in Section 3.7.2.2 of the *2006 IPCC Guidelines* is
 12 representative of operations in the United States. based on expert judgment (RTI 2023).

13 Although some TiO₂ may be produced using graphite or other carbon inputs, information and data regarding these
 14 practices were not available. Titanium dioxide produced using graphite inputs, for example, may generate differing
 15 amounts of CO₂ per unit of TiO₂ produced as compared to that generated using petroleum coke in production.
 16 While the most accurate method to estimate emissions would be to base calculations on the amount of reducing
 17 agent used in each process rather than on the amount of TiO₂ produced, sufficient data were not available to do
 18 so.

19 As of 2004, the last remaining sulfate-process plant in the United States closed. Since annual TiO₂ production was
 20 not reported by USGS by the type of production process used (chloride or sulfate) prior to 2004 and only the
 21 percentage of total production capacity by process was reported, the percent of total TiO₂ production capacity that
 22 was attributed to the chloride process was multiplied by total TiO₂ production to estimate the amount of TiO₂
 23 produced using the chloride process. Finally, the emission factor was applied uniformly to all chloride-process
 24 production, and no data were available to account for differences in production efficiency among chloride-process
 25 plants. In calculating the amount of petroleum coke consumed in chloride-process TiO₂ production, literature data
 26 were used for petroleum coke composition. Certain grades of petroleum coke are manufactured specifically for
 27 use in the TiO₂ chloride process; however, this composition information was not available. EPA assigned an
 28 uncertainty range of ±15 percent for the Tier 1 CO₂ emission factor for the titanium dioxide (chloride route)
 29 production process, and using this uncertainty provided in Table 3.9 of the *2006 IPCC Guidelines* is representative
 30 of operations in the United States based on expert judgment (RTI 2023).

31 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-47. Titanium dioxide
 32 consumption CO₂ emissions from 2022 were estimated to be between 1.3 and 1.7 MMT CO₂ Eq. at the 95 percent
 33 confidence level. This indicates a range of approximately 13 percent below and 13 percent above the emission
 34 estimate of 1.5 MMT CO₂ Eq.

35 **Table 4-48: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Titanium**
 36 **Dioxide Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Titanium Dioxide Production	CO ₂	1.5	1.3	1.7	-13%	+13%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

1 QA/QC and Verification

2 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
3 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of the *2006 IPCC Guidelines* as described in the
4 introduction of the IPPU chapter (see Annex 8 for more details).

5 Recalculations Discussion

6 Updated USGS data on TiO₂ production was available for 2019 and 2020, resulting in updated emissions estimates
7 for those years. Compared to the previous *Inventory*, emissions for 2019 decreased by 9 percent (134 kt CO₂), and
8 emissions for 2020 increased by 12 percent (147 kt CO₂).

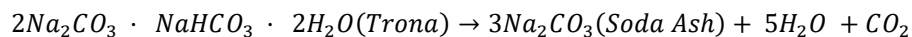
9 Planned Improvements

10 EPA plans to examine the use of GHGRP titanium dioxide emissions and other data for possible use in emission
11 estimates consistent with both Volume 1, Chapter 6 of the *2006 IPCC Guidelines* and the latest IPCC guidance on
12 the use of facility-level data in national inventories.⁴⁰ This planned improvement is ongoing and has not been
13 incorporated into this *Inventory* report. This is a long-term planned improvement given significance of these
14 emissions.

15 4.12 Soda Ash Production (CRT Source 16 Category 2B7)

17 Carbon dioxide (CO₂) is generated as a byproduct of calcining trona ore to produce soda ash and is eventually
18 emitted into the atmosphere. In addition, CO₂ may also be released when soda ash is consumed. Emissions from
19 soda ash consumption not associated with glass production are reported under Section 4.4 Other Process Uses of
20 Carbonates (CRT Category 2A4), and emissions from fuels consumed for energy purposes during the production
21 and consumption of soda ash are accounted for in the Energy chapter.

22 Calcining involves placing crushed trona ore into a kiln to convert sodium bicarbonate into crude sodium carbonate
23 that will later be filtered into pure soda ash. The emission of CO₂ during trona-based production is based on the
24 following reaction:



26 Soda ash (sodium carbonate, Na₂CO₃) is a white crystalline solid that is readily soluble in water and strongly
27 alkaline. Commercial soda ash is used as a raw material in a variety of industrial processes and in many familiar
28 consumer products such as glass, soap and detergents, paper, textiles, and food. The largest use of soda ash is for
29 glass manufacturing. Emissions from soda ash used in glass production are reported under Section 4.3, Glass
30 Production (CRT Source Category 2A3). In addition, soda ash is used primarily to manufacture many sodium-based
31 inorganic chemicals, including sodium bicarbonate, sodium chromates, sodium phosphates, and sodium silicates
32 (USGS 2018b). Internationally, two types of soda ash are produced: natural and synthetic. The United States
33 produces only natural soda ash and is second only to China in total soda ash production. Trona is the principal ore
34 from which natural soda ash is made.

⁴⁰ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

1 The United States represents about one-fifth of total world soda ash output (USGS 2023a). Only two states
 2 produce natural soda ash: Wyoming and California. Of these two states, net emissions of CO₂ from soda ash
 3 production were only calculated for Wyoming where trona ore is used.⁴¹ Soda ash end uses in 2022 (excluding
 4 glass production) consisted of chemical production, 54 percent; other uses, 17 percent; wholesale distributors
 5 (e.g., for use in agriculture, water treatment, and grocery wholesale), 10 percent; soap and detergent
 6 manufacturing, 9 percent; flue gas desulfurization, 7 percent; water treatment, 2 percent; and pulp and paper
 7 production, 1 percent (USGS 2023b).⁴²

8 U.S. natural soda ash is competitive in world markets because it is generally considered a better-quality raw
 9 material than synthetically produced soda ash, and most of the world’s soda ash is synthetic. Although the United
 10 States continues to be a major supplier of soda ash, China surpassed the United States in soda ash production in
 11 2003, becoming the world’s leading producer.

12 In 2022, CO₂ emissions from the production of soda ash from trona ore were 1.7 MMT CO₂ Eq. (1,704 kt CO₂) (see
 13 Table 4-49 and Table 4-50). Total emissions from soda ash production in 2022 decreased by approximately 1
 14 percent compared to emissions in 2021, as soda ash production returned to 2018 levels observed before the
 15 COVID-19 pandemic. Emissions have increased by approximately 19 percent from 1990 levels.

16 Trends in emissions have remained relatively constant over the time series with some fluctuations since 1990. In
 17 general, these fluctuations were related to the behavior of the export market and the U.S. economy. The U.S. soda
 18 ash industry saw a decline in domestic and export sales caused by adverse global economic conditions in 2009,
 19 followed by a steady increase in production through 2019 before a significant decrease in 2020 due to the COVID-
 20 19 pandemic.

21 **Table 4-49: CO₂ Emissions from Soda Ash Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Soda Ash Production	1.4	1.7	1.7	1.8	1.5	1.7	1.7

22 **Table 4-50: CO₂ Emissions from Soda Ash Production (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Soda Ash Production	1,431	1,655	1,714	1,792	1,461	1,714	1,704

23 Methodology and Time-Series Consistency

24 Carbon dioxide emissions from soda ash production are calculated using a Tier 1 method from the *2006 IPCC*
 25 *Guidelines*, in accordance with the IPCC methodological decision tree and available data. During the soda ash
 26 production process, trona ore is calcined in a rotary kiln and chemically transformed into a crude soda ash that
 27 requires further processing. Carbon dioxide and water are generated as byproducts of the calcination process.
 28 Carbon dioxide emissions from the calcination of trona ore can be estimated based on the chemical reaction
 29 shown above. Based on this formula and the IPCC default emission factor of 0.0974 metric tons CO₂ per metric ton

⁴¹ In California, soda ash is manufactured using sodium carbonate-bearing brines instead of trona ore. To extract the sodium carbonate, the complex brines are first treated with CO₂ in carbonation towers to convert the sodium carbonate into sodium bicarbonate, which then precipitates from the brine solution. The precipitated sodium bicarbonate is then calcined back into sodium carbonate. Although CO₂ is generated as a byproduct, the CO₂ is recovered and recycled for use in the carbonation stage and is not emitted. A facility in a third state, Colorado, produced soda ash until the plant was idled in 2004. The lone producer of sodium bicarbonate no longer mines trona ore in the state. For a brief time, sodium bicarbonate was produced using soda ash feedstocks mined in Wyoming and shipped to Colorado. Prior to 2004, because the trona ore was mined in Wyoming, the production numbers given by the USGS included the feedstocks mined in Wyoming and shipped to Colorado. In this way, the sodium bicarbonate production that took place in Colorado was accounted for in the Wyoming numbers.

⁴² Percentages may not add up to 100 percent due to independent rounding.

1 of trona ore, both of which are consistent with an IPCC Tier 1 approach, one metric ton of CO₂ is emitted when
 2 approximately 10.27 metric tons of trona ore are processed (IPCC 2006).

3 Data is not currently available for the quantity of trona used in soda ash production. Because trona ore is used
 4 primarily for soda ash production, EPA assumes that all trona ore production was used in soda ash production. The
 5 activity data for trona ore production (see Table 4-51) for 1990 through 2022 were obtained from the U.S.
 6 Geological Survey (USGS) *Minerals Yearbook for Soda Ash* (1994 through 2015b) and USGS *Mineral Industry*
 7 *Surveys for Soda Ash* (USGS 2016 through 2017, 2018a, 2019, 2020, 2021, 2022b, 2023b). Soda ash production⁴³
 8 data were collected by the USGS from voluntary surveys of the U.S. soda ash industry. EPA will continue to analyze
 9 and assess opportunities to use facility-level data from EPA’s GHGRP to improve the emission estimates for the
 10 soda ash production source category consistent with IPCC⁴⁴ and UNFCCC guidelines.

11 **Table 4-51: Trona Ore Used in Soda Ash Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Trona Ore Use ^a	14,700	17,000	17,600	18,400	15,000	17,600	17,500

^a Trona ore use is assumed to be equal to trona ore production.

12 Methodological approaches were applied to the entire time series to ensure consistency in emissions estimates
 13 from 1990 through 2022.

14 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

15 Emission estimates from soda ash production have relatively low associated uncertainty levels because reliable
 16 and accurate data sources are available for the emission factor and activity data for trona-based soda ash
 17 production. One source of uncertainty is the purity of the trona ore used for manufacturing soda ash. The emission
 18 factor used for this estimate assumes the ore is 100 percent pure and likely overestimates the emissions from soda
 19 ash manufacture. The average water-soluble sodium carbonate-bicarbonate content for ore mined in Wyoming
 20 ranges from 85.5 to 93.8 percent (USGS 1995c).

21 EPA is aware of one facility producing soda ash from a liquid alkaline feedstock process, based on EPA’s GHGRP.
 22 Soda ash production data was collected by the USGS from voluntary surveys. A survey request was sent to each of
 23 the five soda ash producers, all of which responded, representing 100 percent of the total production data (USGS
 24 2023b). EPA assigned an uncertainty range of ±5 percent for trona production, and using the this suggested
 25 uncertainty provided in Section 3.8.2.2 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI
 26 2023). EPA assigned an uncertainty range of -15 percent to 0 percent range for the trona emission factor, based on
 27 expert judgment on the purity of mined trona (USGS 1995c).

28 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-52. Soda ash production
 29 CO₂ emissions for 2021 were estimated to be between 1.5 and 1.8 MMT CO₂ Eq. at the 95 percent confidence
 30 level. This indicates a range of approximately 9 percent below and 8 percent above the emission estimate of 1.7
 31 MMT CO₂ Eq.

32 **Table 4-52: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Soda Ash**
 33 **Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.)	(%)
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⁴³ EPA has assessed the feasibility of using emissions information (including activity data) from EPA’s GHGRP program. At this time, the aggregated information associated with production of soda ash did not meet criteria to shield underlying confidential business information (CBI) from public disclosure.

⁴⁴ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Soda Ash Production	CO ₂	1.7	1.5	1.8	-9%	+8%
^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.						

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details).

Recalculations Discussion

No recalculations were performed for the 1990 through 2021 portion of the time series.

Planned Improvements

EPA is assessing planned improvements for future reports, but at this time has no specific planned improvements for estimating CO₂ emissions from soda ash production.

4.13 Petrochemical Production (CRT Source Category 2B8)

The production of some petrochemicals results in carbon dioxide (CO₂) and methane (CH₄) emissions. Petrochemicals are chemicals isolated or derived from petroleum or natural gas. Carbon dioxide emissions from the production of acrylonitrile, carbon black, ethylene, ethylene dichloride, ethylene oxide, and methanol, and CH₄ emissions from the production of acrylonitrile are presented here and reported under IPCC Source Category 2B8. The petrochemical industry uses primary fossil fuels (i.e., natural gas, coal, petroleum, etc.) for non-fuel purposes in the production of carbon black and other petrochemicals. Emissions from fuels and feedstocks transferred out of the system for use in energy purposes (e.g., indirect or direct process heat or steam production) are currently accounted for in the Energy sector. The allocation and reporting of emissions from feedstocks transferred out of the system for use in energy purposes to the Energy chapter is consistent with the *2006 IPCC Guidelines*.

Worldwide, more than 90 percent of acrylonitrile (vinyl cyanide, C₃H₃N) is made by way of direct ammoxidation of propylene with ammonia (NH₃) and oxygen over a catalyst. This process is referred to as the SOHIO process, named after the Standard Oil Company of Ohio (SOHIO) (IPCC 2006). The primary use of acrylonitrile is as the raw material for the manufacture of acrylic and modacrylic fibers. Other major uses include the production of plastics (acrylonitrile-butadiene-styrene [ABS] and styrene-acrylonitrile [SAN]), nitrile rubbers, nitrile barrier resins, adiponitrile, and acrylamide. All U.S. acrylonitrile facilities use the SOHIO process (AN 2014). The SOHIO process involves a fluidized bed reaction of chemical-grade propylene, ammonia, and oxygen over a catalyst. The process produces acrylonitrile as its primary product, and the process yield depends on the type of catalyst used and the process configuration. The ammoxidation process produces byproduct CO₂, carbon monoxide (CO), and water from the direct oxidation of the propylene feedstock and produces other hydrocarbons from side reactions.

Carbon black is a black powder generated by the incomplete combustion of an aromatic petroleum- or coal-based feedstock at a high temperature. Most carbon black produced in the United States is added to rubber to impart strength and abrasion resistance, and the tire industry is by far the largest consumer. The other major use of carbon black is as a pigment. The predominant process used in the United States to produce carbon black is the

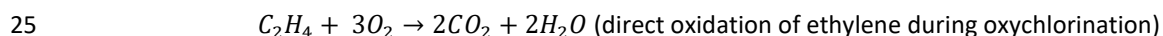
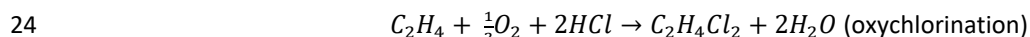
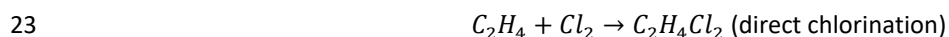
1 furnace black (or oil furnace) process. In the furnace black process, carbon black oil (a heavy aromatic liquid) is
2 continuously injected into the combustion zone of a natural gas-fired furnace. Furnace heat is provided by the
3 natural gas and a portion of the carbon black feedstock; the remaining portion of the carbon black feedstock is
4 pyrolyzed to carbon black. The resultant CO₂ and uncombusted CH₄ are released from thermal incinerators used as
5 control devices, process dryers, and equipment leaks. Three facilities in the United States use other types of
6 carbon black processes. Specifically, one facility produces carbon black by the thermal cracking of acetylene-
7 containing feedstocks (i.e., acetylene black process), a second facility produces carbon black by the thermal
8 cracking of other hydrocarbons (i.e., thermal black process), and a third facility produces carbon black by the open
9 burning of carbon black feedstock (i.e., lamp black process) (EPA 2000).

10 Ethylene (C₂H₄) is consumed in the production processes of the plastics industry including polymers such as high,
11 low, and linear low density polyethylene (HDPE, LDPE, LLDPE); polyvinyl chloride (PVC); ethylene dichloride;
12 ethylene oxide; and ethylbenzene. Virtually all ethylene is produced from steam cracking of ethane, propane,
13 butane, naphtha, gas oil, and other feedstocks. The representative chemical equation for steam cracking of ethane
14 to ethylene is shown below:



16 Small amounts of CH₄ are also generated from the steam cracking process. In addition, CO₂ and CH₄ emissions
17 result from combustion units.

18 Ethylene dichloride (C₂H₄Cl₂) is used to produce vinyl chloride monomer, which is the precursor to polyvinyl
19 chloride (PVC). Ethylene dichloride was also used as a fuel additive until 1996 when leaded gasoline was phased
20 out. Ethylene dichloride is produced from ethylene by either direct chlorination, oxychlorination, or a combination
21 of the two processes (i.e., the “balanced process”); most U.S. facilities use the balanced process. The direct
22 chlorination and oxychlorination reactions are shown below:



26 In addition to the byproduct CO₂ produced from the direct oxidation of the ethylene feedstock, CO₂ and CH₄
27 emissions are also generated from combustion units.

28 Ethylene oxide (C₂H₄O) is used in the manufacture of glycols, glycol ethers, alcohols, and amines. Approximately 70
29 percent of ethylene oxide produced worldwide is used in the manufacture of glycols, including monoethylene
30 glycol. Ethylene oxide is produced by reacting ethylene with oxygen over a catalyst. The oxygen may be supplied to
31 the process through either an air (air process) or a pure oxygen stream (oxygen process). The byproduct CO₂ from
32 the direct oxidation of the ethylene feedstock is removed from the process vent stream using a recycled carbonate
33 solution, and the recovered CO₂ may be vented to the atmosphere or recovered for further utilization in other
34 sectors, such as food production (IPCC 2006). The combined ethylene oxide reaction and byproduct CO₂ reaction is
35 exothermic and generates heat, which is recovered to produce steam for the process. The ethylene oxide process
36 also produces other liquid and off-gas byproducts (e.g., ethane that may be burned for energy recovery within the
37 process. Almost all facilities, except one in Texas, use the oxygen process to manufacture ethylene oxide (EPA
38 2008).

39 Methanol (CH₃OH) is a chemical feedstock most often converted into formaldehyde, acetic acid and olefins. It is
40 also an alternative transportation fuel, as well as an additive used by municipal wastewater treatment facilities in
41 the denitrification of wastewater. Methanol is most commonly synthesized from a synthesis gas (i.e., “syngas” – a
42 mixture containing H₂, CO, and CO₂) using a heterogeneous catalyst. There are a number of process techniques
43 that can be used to produce syngas. Worldwide, steam reforming of natural gas is the most common method;
44 most methanol producers in the United States also use steam reforming of natural gas to produce syngas. Other
45 syngas production processes in the United States include partial oxidation of natural gas and coal gasification.

1 Emissions of CO₂ and CH₄ from petrochemical production in 2022 were 28.8 MMT CO₂ Eq. (28,788 kt CO₂) and
 2 0.005 MMT CO₂ Eq. (0.17 kt CH₄), respectively (see Table 4-53 and Table 4-54). Carbon dioxide emissions from
 3 petrochemical production are driven primarily from ethylene production, while CH₄ emissions are only from
 4 acrylonitrile production. Since 1990, total CO₂ emissions from petrochemical production increased by 33 percent,
 5 and CH₄ emissions declined by 22 percent. Emissions of CO₂ were 6 percent lower in 2022 than in 2021, and
 6 emissions of CH₄ were 12 percent higher in 2022 than in 2021. The increase in CO₂ emissions since 1990 is due
 7 primarily to increased ethylene and methanol production, which have been driven by the increased natural gas
 8 production in the United States. The reduction in CO₂ emissions since 2021 is due to a reduction in emissions from
 9 ethylene production, despite an increase in ethylene production. Since CH₄ emissions from acrylonitrile are
 10 calculated using a Tier 1 approach based on production as the activity data, the increase in CH₄ emissions since
 11 1990 and the decrease since 2021 correspond with changes in the production levels for acrylonitrile.

12 **Table 4-53: CO₂ and CH₄ Emissions from Petrochemical Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
CO₂	21.6	27.4	27.2	28.5	27.9	30.7	28.8
Carbon Black	3.4	4.3	3.4	3.3	2.6	3.0	3.1
Ethylene	13.1	19.0	19.4	20.7	20.7	22.8	20.7
Ethylene Dichloride	0.3	0.5	0.4	0.5	0.5	0.4	0.4
Ethylene Oxide	1.1	1.5	1.3	1.4	1.7	1.9	1.7
Methanol	2.5	0.8	1.4	1.6	1.6	1.7	2.0
Acrylonitrile	1.2	1.3	1.3	1.0	0.9	0.9	1.0
CH₄	+	+	+	+	+	+	+
Acrylonitrile	+	+	+	+	+	+	+
Total	21.6	27.4	27.2	28.5	27.9	30.7	28.8

+ Does not exceed 0.05 MMT CO₂ Eq.

Note: Totals may not sum due to independent rounding.

13 **Table 4-54: CO₂ and CH₄ Emissions from Petrochemical Production (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
CO₂	21,611	27,383	27,200	28,483	27,926	30,656	28,788
Carbon Black	3,381	4,269	3,440	3,300	2,610	3,000	3,060
Ethylene	13,126	19,024	19,400	20,700	20,700	22,800	20,700
Ethylene Dichloride	254	455	440	503	456	376	428
Ethylene Oxide	1,123	1,489	1,300	1,370	1,680	1,930	1,650
Methanol	2,513	821	1,370	1,620	1,630	1,700	2,000
Acrylonitrile	1,214	1,325	1,250	990	850	850	950
CH₄	+	+	+	+	+	+	+
Acrylonitrile	+	+	+	+	+	+	+

+ Does not exceed 0.5 kt CH₄.

Note: Totals by gas may not sum due to independent rounding.

14 Methodology and Time-Series Consistency

15 Emissions of CO₂ and CH₄ were calculated using the estimation methods provided by the *2006 IPCC Guidelines*, in
 16 accordance with the IPCC methodological decision tree and available data, and country-specific methods from
 17 EPA's GHGRP. The *2006 IPCC Guidelines* Tier 1 method was used to estimate CO₂ and CH₄ emissions from
 18 production of acrylonitrile,⁴⁵ and a country-specific approach similar to the IPCC Tier 2 method was used to
 19 estimate CO₂ emissions from production of carbon black, ethylene oxide, ethylene, ethylene dichloride, and

⁴⁵ EPA has not integrated aggregated facility-level GHGRP information for acrylonitrile production. The aggregated information associated with production of these petrochemicals did not meet criteria to shield underlying CBI from public disclosure.

1 methanol, as CO₂ emissions from petrochemical production is a key category. The Tier 2 method for
2 petrochemicals is a total feedstock carbon mass balance method used to estimate total CO₂ emissions, but it is not
3 applicable for estimating CH₄ emissions.

4 As noted in the *2006 IPCC Guidelines*, the Tier 2 total feedstock carbon mass balance method is based on the
5 assumption that all of the carbon input to the process is converted either into primary and secondary products or
6 into CO₂. Further, the guideline states that while the total carbon mass balance method estimates total carbon
7 emissions from the process, it does not directly provide an estimate of the amount of the total carbon emissions
8 emitted as CO₂, CH₄, or non-CH₄ volatile organic compounds (NMVOCs). This method accounts for all the carbon as
9 CO₂, including CH₄.

10 A methodology refinement for emissions from methanol production was implemented this year to transition from
11 a Tier 1 method to a country-specific approach similar to a Tier 2 method, using the process CO₂ emissions reported
12 to subpart X of the GHGRP. As part of this refinement, CH₄ emissions from methanol production for every year in
13 the time series are now included in the CO₂ emissions estimates to avoid double counting because the GHGRP
14 reporting method is a mass balance method under which all carbon input to the process is assumed to be
15 converted either into primary and secondary products or into CO₂.

16 Note, a small subset of facilities reporting under EPA's GHGRP use Continuous Emission Monitoring Systems
17 (CEMS) to monitor CO₂ emissions from process vents and/or stacks from stationary combustion units. These
18 facilities are required to also report CO₂, CH₄ and N₂O emissions from combustion of process off-gas in flares. The
19 CO₂ emissions from flares are included in aggregated CO₂ results. Preliminary analysis of aggregated annual reports
20 shows that flared CH₄ and N₂O emissions are less than 500 kt CO₂ Eq./year. EPA's GHGRP team is still reviewing
21 these data across reported years, and EPA plans to address this more completely in future reports.

22 **Carbon Black, Ethylene, Ethylene Dichloride, and Ethylene Oxide**

23 **2010 through 2022**

24 Carbon dioxide emissions and national production for carbon black, ethylene, ethylene dichloride, and ethylene
25 oxide were aggregated directly from EPA's GHGRP dataset for 2010 through 2022 (EPA 2023).

26 These emissions reflect application of a country-specific approach similar to the IPCC Tier 2 method and were used
27 to estimate CO₂ emissions from the production of carbon black, ethylene, ethylene dichloride, ethylene oxide. In
28 2022, data reported to the GHGRP included 3,060,000 metric tons of CO₂ emissions from carbon black production;
29 20,700,000 metric tons of CO₂ from ethylene production; 428,000 metric tons of CO₂ from ethylene dichloride
30 production; and 1,650,000 metric tons of CO₂ from ethylene oxide production.

31 Since 2010, EPA's GHGRP requires all domestic producers of petrochemicals to report annual emissions and
32 supplemental emissions information (e.g., production data, etc.) under Subpart X to facilitate verification of
33 reported emissions. Most petrochemical production facilities are required to use either a mass balance approach
34 or CEMS to measure and report emissions for each petrochemical process unit to estimate facility-level process
35 CO₂ emissions; ethylene production facilities also have a third option. The mass balance method is used by most
36 facilities⁴⁶ and assumes that all the carbon input is converted into primary and secondary products or is emitted to
37 the atmosphere as CO₂. To apply the mass balance, facilities must measure the volume or mass of each gaseous
38 and liquid feedstock and product, mass rate of each solid feedstock and product, and carbon content of each
39 feedstock and product for each process unit and sum for their facility. To apply the optional combustion
40 methodology, ethylene production facilities must measure the quantity, carbon content, and molecular weight of
41 the fuel to a stationary combustion unit when that fuel includes any ethylene process off-gas. These data are used
42 to calculate the total CO₂ emissions from the combustion unit. The facility must also estimate the fraction of the

⁴⁶ A few facilities producing ethylene dichloride, ethylene, and methanol used CO₂ CEMS; those CO₂ emissions have been included in the aggregated GHGRP emissions presented here.

1 emissions that is attributable to burning the ethylene process off-gas portion of the fuel. This fraction is multiplied
2 by the total emissions to estimate the emissions from ethylene production. The QA/QC and Verification section
3 below has a discussion of non-CO₂ emissions from ethylene production facilities.

4 All non-energy uses of residual fuel and some non-energy uses of “other oil” are assumed to be used in the
5 production of carbon black; therefore, consumption of these fuels is adjusted for within the Energy chapter to
6 avoid double-counting of emissions from fuel used in the carbon black production presented here within IPPU
7 sector. Additional information on the adjustments made within the Energy sector for non-energy use of fuels is
8 described in both the Methodology section of CO₂ from Section 3.1 Fossil Fuel Combustion (CRT Source Category
9 1A) and Annex 2.1, Methodology for Estimating Emissions of CO₂ from Fossil Fuel Combustion.

10 **1990 through 2009**

11 Prior to 2010, for carbon black, ethylene, ethylene dichloride, and ethylene oxide processes, an average national
12 CO₂ emission factor was calculated based on the GHGRP data and applied to production for earlier years in the
13 time series (i.e., 1990 through 2009) to estimate CO₂ emissions. For these 4 types of petrochemical processes, CO₂
14 emission factors were derived from EPA’s GHGRP data by dividing annual CO₂ emissions for petrochemical type “i”
15 with annual production for petrochemical type “i” and then averaging the derived emission factors obtained for
16 each calendar year 2010 through 2013 (EPA 2023). The years 2010 through 2013 were used in the development of
17 carbon dioxide emission factors as these years are more representative of operations in 1990 through 2009 for
18 these facilities. The average emission factors for each petrochemical type were applied across all prior years
19 because petrochemical production processes in the United States have not changed significantly since 1990,
20 though some operational efficiencies have been implemented at facilities over the time series.

21 The average country-specific CO₂ emission factors that were calculated from the GHGRP data are as follows:

- 22 • 2.59 metric tons CO₂/metric ton carbon black produced
- 23 • 0.79 metric tons CO₂/metric ton ethylene produced
- 24 • 0.040 metric tons CO₂/metric ton ethylene dichloride produced
- 25 • 0.46 metric tons CO₂/metric ton ethylene oxide produced

26 Annual production data for carbon black for 1990 through 2009 were obtained from the International Carbon
27 Black Association (Johnson 2003 and 2005 through 2010). Annual production data for ethylene, ethylene
28 dichloride, and ethylene oxide for 1990 through 2009 were obtained from the American Chemistry Council’s (ACC)
29 *Business of Chemistry* (ACC 2023).

30 **Methanol**

31 **2015 through 2022**

32 Carbon dioxide emissions and national production for methanol were aggregated directly from EPA’s GHGRP data
33 for 2015 through 2022 (EPA 2023). These emissions reflect application of a country-specific approach similar to the
34 IPCC Tier 2 method and were used to estimate CO₂ emissions from the production of methanol. In 2022, data
35 reported to the GHGRP included 2,000,000 metric tons of CO₂ emissions from methanol production.

36 As noted above, since 2010, EPA’s GHGRP requires all domestic producers of petrochemicals to report annual
37 emissions and supplemental emissions information (e.g., production data, etc.) under Subpart X to facilitate
38 verification of reported emissions. Methanol production facilities are required to use either a mass balance
39 approach or CEMS to measure and report emissions for each methanol process unit to estimate facility-level
40 process CO₂ emissions. Most methanol production facilities use the mass balance method. As noted above, when
41 using the mass balance method, facilities must measure the volume or mass of each gaseous and liquid feedstock
42 and product, mass rate of each solid feedstock and product, and carbon content of each feedstock and product for
43 each process unit and sum for their facility.

1 **1990 through 2009**

2 In this *Inventor*, the methanol production quantities and CO₂ emissions for 2015 through 2021 were revised to
3 use GHGRP as the data source. The average country-specific CO₂ emission factor from the GHGRP data for these
4 years was determined to be 0.26 metric tons CO₂/metric ton methanol produced, which is significantly lower than
5 the 0.67 value that was used in previous versions of the *Inventor*. The value is lower because many of the
6 methanol production process units constructed or overhauled since 2015 are more efficient than older process
7 units. As a result, it was determined that the emission factor based on GHGRP data would not be appropriate for
8 prior years in the time series. Instead, it was assumed that most of the older units in operation prior to 2015 were
9 based on conventional steam reforming with both a primary reformer unit and a secondary reformer unit, which is
10 the default process unit identified in the *2006 IPCC Guidelines*. It was also assumed that the default CO₂ emission
11 factor of 0.67 metric tons CO₂/metric ton of methanol produced in the *2006 IPCC Guidelines* for this type of unit
12 also accurately represents the total CO₂ emissions from such process units as per a mass balance approach. Thus,
13 the average country-specific CO₂ emission factor for 1990 through 2009 was estimated to be 0.67 metric tons
14 CO₂/metric ton of methanol produced. Annual methanol production data for 1990 through 2009 were obtained
15 from the ACC's *Business of Chemistry* (ACC 2023).

16 **2010 through 2014**

17 For 2010 through 2014, annual country-specific CO₂ emission factors were calculated for each year based on linear
18 interpolation between the assumed emission factor of 0.67 for 2009 to the emission factor of 0.355 for 2015 that
19 was calculated from the GHGRP data. 2009 was chosen as a starting point for the linear decline since that
20 corresponded to roughly the low point in methanol production after which new plants started to come online.
21 GHGRP data for methanol in 2010 through 2014 cannot be used because they did not meet criteria to shield
22 underlying CBI from public disclosure. Annual methanol production data for 2010 through 2014 were obtained
23 from the ACC's *Business of Chemistry* (ACC 2023).

24 **Acrylonitrile**

25 Carbon dioxide and methane emissions from acrylonitrile production were estimated using the Tier 1 method in
26 the *2006 IPCC Guidelines*. Acrylonitrile emissions represent about 3 percent of total petrochemical emissions in
27 2022 so a Tier 1 approach is deemed acceptable, and higher Tier methods could not be used due to data
28 sensitivities which are described below. Annual acrylonitrile production data were used with IPCC default Tier 1
29 CO₂ and CH₄ emission factors to estimate emissions for 1990 through 2022. Emission factors used to estimate
30 acrylonitrile production emissions are as follows:

- 31 • 0.18 kg CH₄/metric ton acrylonitrile produced
- 32 • 1.00 metric tons CO₂/metric ton acrylonitrile produced

33 Annual acrylonitrile production data for 1990 through 2022 were obtained from ACC's *Business of Chemistry* (ACC
34 2023). EPA is unable to apply the aggregated facility-level GHGRP information for acrylonitrile production needed
35 for a Tier 2 approach due to sensitive nature of reported data. The aggregated information associated with
36 production of these petrochemicals did not meet criteria to shield underlying CBI from public disclosure.

37 Production of each type of petrochemical are shown in Table 4-55.

38 **Table 4-55: Production of Selected Petrochemicals (kt)**

Chemical	1990	2005	2018	2019	2020	2021	2022
Carbon Black	1,307	1,651	1,280	1,210	990	1,140	1,170
Ethylene	16,542	23,975	30,500	32,400	33,500	34,700	35,400
Ethylene Dichloride	6,283	11,260	12,500	12,600	11,900	11,500	12,100
Ethylene Oxide	2,429	3,220	3,310	3,800	4,680	4,860	5,310
Methanol	3,750	1,225	5,830	6,460	6,580	7,110	8,030
Acrylonitrile	1,214	1,325	1,250	990	850	850	950

1 As noted earlier in the introduction section of the Petrochemical Production section, the allocation and reporting
2 of emissions from both fuels and feedstocks transferred out of the system for use in energy purposes to the Energy
3 chapter differs slightly from the *2006 IPCC Guidelines*. According to the *2006 IPCC Guidelines*, emissions from fuel
4 combustion from petrochemical production should be allocated to this source category within the IPPU chapter.
5 Due to national circumstances, EIA data on primary fuel for feedstock use within the energy balance are presented
6 by commodity only, with no resolution on data by industry sector (i.e., petrochemical production). In addition,
7 under EPA's GHGRP, reporting facilities began reporting in 2014 on annual feedstock quantities for mass balance
8 and CEMS methodologies (79 FR 63794), as well as the annual average carbon content of each feedstock (and
9 molecular weight for gaseous feedstocks) for the mass balance methodology beginning in reporting year 2017 (81
10 FR 89260).⁴⁷ The United States is currently unable to report non-energy fuel use from petrochemical production
11 under the IPPU chapter due to CBI issues. Therefore, consistent with *2006 IPCC Guidelines*, fuel consumption data
12 reported by EIA are modified to account for these overlaps to avoid double-counting. More information on the
13 non-energy use of fossil fuel feedstocks for petrochemical production can be found in Annex 2.3.

14 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
15 through 2022. The methodology for ethylene production, ethylene dichloride production, and ethylene oxide
16 production spliced activity data from two different sources: ACC for 1990 through 2009 and GHGRP for 2010
17 through 2022. The methodology for methanol production spliced activity data from two different sources: ACC for
18 1990 through 2014 and GHGRP for 2015 through 2022. Consistent with the *2006 IPCC Guidelines*, the overlap
19 technique was applied to compare the two data sets for years where there was overlap. For ethylene production,
20 the data sets were determined to be consistent, and adjustments were not needed. For ethylene dichloride
21 production, ethylene oxide production, and methanol production, the data sets were determined to be
22 inconsistent. The GHGRP data includes production of ethylene dichloride and ethylene oxide as intermediates,
23 while it is unclear if the ACC data does. Methanol production data from GHGRP are significantly higher than the
24 ACC data for every year since 2015; the reason for the difference is not clear. Therefore, no adjustments were
25 made to the ethylene dichloride, ethylene oxide, and methanol activity data for 1990 through 2009 because the
26 *2006 IPCC Guidelines* indicate that it is not good practice to use the overlap technique when the data sets are
27 inconsistent. The methodology for carbon black production also spliced activity data from two different sources:
28 ICBA for 1990 through 2009 and GHGRP for 2010 through 2022. The overlap technique was applied to these data
29 for 2010 and 2011. The data sets were determined to be consistent, and adjustments were not needed.

30 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

31 The CO₂ and CH₄ emission factors used for methanol and acrylonitrile production are based on a limited number of
32 studies. Using plant-specific factors instead of default or average factors could increase the accuracy of the
33 emission estimates; however, such data were not available for the current *Inventory* report. For methanol, EPA
34 assigned an uncertainty range of ±30 percent for the CO₂ emission factor and -80 percent to +30 percent for the
35 CH₄ emission factor, consistent with the ranges in Table 3.27 of the *2006 IPCC Guidelines*. For acrylonitrile, EPA
36 assigned an uncertainty range of ±60 percent for the CO₂ emission factor and ±10 percent for the CH₄ emission
37 factor, consistent with the ranges in Table 3.27 of the *2006 IPCC Guidelines*. The results of the quantitative
38 uncertainty analysis for the CO₂ emissions from carbon black production, ethylene, ethylene dichloride, and
39 ethylene oxide are based on reported GHGRP data. Refer to the Methodology section for more details on how
40 these emissions were calculated and reported to EPA's GHGRP. EPA assigned CO₂ emissions from carbon black,
41 ethylene, ethylene dichloride, and ethylene oxide production an uncertainty range of ±5 percent, consistent with
42 the ranges in Table 3.27 of the *2006 IPCC Guidelines*. In the absence of other data, these values have been
43 assessed as reasonable. There is some uncertainty in the applicability of the average emission factors for each
44 petrochemical type across all prior years. While petrochemical production processes in the United States have not

⁴⁷ See <https://www.epa.gov/ghgreporting/historical-rulemakings>.

1 changed significantly since 1990, some operational efficiencies have been implemented at facilities over the time
2 series.

3 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-56. Petrochemical
4 production CO₂ emissions from 2022 were estimated to be between 28.4 and 31.7 MMT CO₂ Eq. at the 95 percent
5 confidence level. This indicates a range of approximately 5 percent below to 6 percent above the emission
6 estimate of 30.0 MMT CO₂ Eq. Petrochemical production CH₄ emissions from 2022 were estimated to be between
7 0.11 and 0.39 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 57 percent
8 below to 47 percent above the emission estimate of 0.3 MMT CO₂ Eq.

9 **Table 4-56: Approach 2 Quantitative Uncertainty Estimates for CH₄ Emissions from**
10 **Petrochemical Production and CO₂ Emissions from Petrochemical Production (MMT CO₂ Eq.**
11 **and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.)			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Petrochemical Production	CO ₂	30.0	28.4	31.7	-5%	+6%
Petrochemical Production	CH ₄	0.3	0.11	0.39	-57%	+47%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

12 QA/QC and Verification

13 For petrochemical production, QA/QC activities were conducted consistent with the U.S. Inventory QA/QC plan, as
14 described in the QA/QC and Verification Procedures section of the IPPU chapter and Annex 8. Source-specific
15 quality control measures for this category included the QA/QC requirements and verification procedures of EPA's
16 GHGRP. More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to
17 petrochemical facilities can be found under Subpart X (Petrochemical Production) of the regulation (40 CFR Part
18 98).⁴⁸ EPA verifies annual facility-level GHGRP reports through a multi-step process (e.g., combination of electronic
19 checks and manual reviews) to identify potential errors and ensure that data submitted to EPA are accurate,
20 complete, and consistent (EPA 2015).⁴⁹ Based on the results of the verification process, EPA follows up with
21 facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of
22 general and category-specific QC procedures, including range checks, statistical checks, algorithm checks, and year-
23 to-year checks of reported data and emissions. EPA also conducts QA checks of GHGRP reported production data
24 by petrochemical type against external datasets.

25 For ethylene, ethylene dichloride, ethylene oxide, and methanol, it is possible to compare CO₂ emissions calculated
26 using the GHGRP data to the CO₂ emissions that would have been calculated using the Tier 1 approach if GHGRP
27 data were not available. For ethylene, the GHGRP emissions were within ±8 percent of the emissions calculated
28 using the Tier 1 approach prior to 2018; for 2018 through 2022, the GHGRP emissions were between 76 percent
29 and 87 percent of what would be calculated using the Tier 1 approach. For ethylene dichloride, the GHGRP
30 emissions are typically higher than the Tier 1 emissions by up to 25 percent, but in 2010 and 2021, GHGRP
31 emissions were slightly lower than the Tier 1 emissions. For ethylene oxide, GHGRP emissions typically vary from

⁴⁸ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

⁴⁹ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 the Tier 1 emissions by up to ±20 percent, but in 2021 and 2022, the GHGRP emissions were significantly higher
2 than the Tier 1 emissions. This was likely due to GHGRP data capturing the production of ethylene oxide as an
3 intermediate in the onsite production of ethylene glycol.

4 For methanol, GHGRP production data was consistently higher than ACC production data in all years between 2015
5 and 2022. Even though the GHGRP production was higher than the ACC production, the GHGRP CO₂ emissions
6 estimated using the methodology refinement in this *Inventory* are significantly lower than the emissions calculated
7 using the Tier 1 approach in all years between 2015 and 2022. Additionally, there is a trend towards increasing
8 differences over these years starting with an 873 kt CO₂ difference in 2015 and increasing to a 3,000 kt CO₂
9 difference in 2022. GHGRP emissions were between 43 percent and 61 percent of the Tier 1 emissions in 2015 and
10 2018, respectively.

11 EPA's GHGRP mandates that all petrochemical production facilities report their annual emissions of CO₂, CH₄, and
12 N₂O from each of their petrochemical production processes. Source-specific quality control measures for the
13 Petrochemical Production category included the QA/QC requirements and verification procedures of EPA's GHGRP.
14 The QA/QC requirements differ depending on the calculation methodology used.

15 As part of a planned improvement effort, EPA has assessed the potential of using GHGRP data to estimate CH₄
16 emissions from ethylene production. As discussed in the Methodology section above, CO₂ emissions from ethylene
17 production in this chapter are based on data reported under the GHGRP, and these emissions are calculated using
18 a Tier 2 approach that assumes all of the carbon in the fuel (i.e., ethylene process off-gas) is converted to CO₂.
19 Ethylene production facilities also calculate and report CH₄ emissions under the GHGRP when they use the optional
20 combustion methodology. The facilities calculate CH₄ emissions from each combustion unit that burns off-gas from
21 an ethylene production process unit using a Tier 1 approach based on the total quantity of fuel burned, a default
22 higher heating value, and a default emission factor. Because multiple other types of fuel in addition to the ethylene
23 process unit off-gas may be burned in these combustion units, the facilities also report an estimate of the fraction
24 of emissions that is due to burning the ethylene process off-gas component of the total fuel. Multiplying the total
25 emissions by the estimated fraction provides an estimate of the CH₄ emissions from the ethylene production
26 process unit. These ethylene production facilities also calculate CH₄ emissions from flares that burn process vent
27 emissions from ethylene processes. The emissions are calculated using either a Tier 2 approach based on
28 measured gas volumes and measured carbon content or higher heating value, or a Tier 1 approach based on the
29 measured gas flow and a default emission factor. Nearly all ethylene production facilities use the optional
30 combustion methodology under the GHGRP. The CH₄ emissions from ethylene production under the GHGRP have
31 not been included in this chapter because this approach double counts carbon (i.e., all of the carbon in the CH₄
32 emissions is also included in the CO₂ emissions from the ethylene process units). EPA continues to assess the
33 GHGRP data for ways to better disaggregate the data and incorporate it into the *Inventory*.

34 These facilities are also required to report emissions of N₂O from combustion of ethylene process off-gas in both
35 stationary combustion units and flares. Facilities using CEMS (consistent with a Tier 3 approach) are also required
36 to report emissions of CH₄ and N₂O from combustion of petrochemical process-off gases in flares. Preliminary
37 analysis of the aggregated reported CH₄ and N₂O emissions from facilities using CEMS and N₂O emissions from
38 facilities using the optional combustion methodology suggests that these annual emissions are less than 0.4
39 percent of total petrochemical emissions, which is not significant enough to prioritize for inclusion in the report at
40 this time. Pending resources and significance, EPA may include these N₂O emissions in future reports to enhance
41 completeness.

42 Future QC efforts to validate the use of Tier 1 default emission factors and report on the comparison of Tier 1
43 emission estimates and GHGRP data are described below in the Planned Improvements section.

44 Recalculations Discussion

45 A methodology refinement for calculating emissions from methanol production was implemented in this *Inventory*.
46 For 2015 through 2021, these changes resulted in a reduction in the reported CO₂ emissions between 43 percent
47 (873 kt) in 2015 to 61 percent (2,110 kt) in 2018. For 2010 to 2014, the refinement resulted in a reduction between

1 7.8 percent (38.3 kt) in 2010 to 39.1 percent (552 kt) in 2014. There was no change in CO₂ emissions for 1990
2 through 2009.

3 Additionally, reported CH₄ emissions under the GHGRP method were reduced to zero in all years of the time series
4 because, as noted above in the Methodology and Time Series Consistency section, the GHGRP method is
5 consistent with a Tier 2 mass balance method in the *2006 IPCC Guidelines* and as such, it is based on the
6 assumption that all carbon input to the process is converted either to primary and secondary products or to CO₂.
7 Furthermore, for prior time series the emission factor used was also determined to be equivalent to a mass-based
8 approach and therefore accounted for all carbon as CO₂.

9 **Planned Improvements**

10 Improvements include completing category-specific QC of activity data and emission factors, along with further
11 assessment of CH₄ and N₂O emissions to enhance completeness in reporting of emissions from U.S. petrochemical
12 production, pending resources, significance and time-series consistency considerations. For example, EPA is
13 planning additional assessment of ways to use CH₄ data from the GHGRP in the *Inventory*. One possible approach
14 EPA is assessing would be to adjust the CO₂ emissions from the GHGRP downward by subtracting the carbon that is
15 also included in the reported CH₄ emissions, per the discussion in the Petrochemical Production QA/QC and
16 Verification section, above. As of this current report, timing and resources have not allowed EPA to complete this
17 analysis of activity data, emissions, and emission factors and remains a priority improvement within the IPPU
18 chapter.

19 Pending resources, a secondary potential improvement for this source category would focus on continuing to
20 analyze the fuel and feedstock data from EPA's GHGRP to better disaggregate energy-related emissions and
21 allocate them more accurately between the Energy and IPPU sectors of the *Inventory*. It is important to ensure no
22 double counting of emissions between fuel combustion, non-energy use of fuels, and industrial process emissions.
23 For petrochemical feedstock production, EPA review of the categories suggests this is not a significant issue since
24 the non-energy use industrial release data includes different categories of sources and sectors than those included
25 in the IPPU emissions category for petrochemicals. As noted previously in the methodology section, data
26 integration is not available at this time because feedstock data from the EIA used to estimate non-energy uses of
27 fuels are aggregated by fuel type, rather than disaggregated by both fuel type and particular industries. Also,
28 GHGRP-reported data on quantities of fuel consumed as feedstocks by petrochemical producers are unable to be
29 used due to the data failing GHGRP CBI aggregation criteria. EPA will continue to look for ways to incorporate this
30 data into future Inventories that will allow for easier data integration between the non-energy uses of fuels
31 category and the petrochemicals category presented in this chapter. This planned improvement is still under
32 development and has not been completed to report on progress in this current *Inventory*.

33 **4.14 HCFC-22 Production (CRT Source** 34 **Category 2B9a)**

35 Trifluoromethane (HFC-23 or CHF₃) is generated as a byproduct during the manufacture of chlorodifluoromethane
36 (HCFC-22), which is primarily employed in refrigeration and air conditioning systems and as a chemical feedstock
37 for manufacturing synthetic polymers. Between 1990 and 2000, U.S. production of HCFC-22 increased significantly
38 as HCFC-22 replaced chlorofluorocarbons (CFCs) in many applications. Between 2000 and 2007, U.S. production
39 fluctuated but generally remained above 1990 levels. In 2008 and 2009, U.S. production declined markedly and has
40 remained near 2009 levels since. Because HCFC-22 depletes stratospheric ozone, its production for non-feedstock

1 uses was phased out in 2020 under the U.S. Clean Air Act.⁵⁰ Feedstock production, however, is permitted to
2 continue indefinitely.

3 HCFC-22 is produced by the reaction of chloroform (CHCl₃) and hydrogen fluoride (HF) in the presence of a catalyst,
4 SbCl₅. The reaction of the catalyst and HF produces SbCl_xF_y, (where x + y = 5), which reacts with chlorinated
5 hydrocarbons to replace chlorine atoms with fluorine. The HF and chloroform are introduced by submerged piping
6 into a continuous-flow reactor that contains the catalyst in a hydrocarbon mixture of chloroform and partially
7 fluorinated intermediates. The vapors leaving the reactor contain HCFC-21 (CHCl₂F), HCFC-22 (CHClF₂), HFC-23
8 (CHF₃), HCl, chloroform, and HF. The under-fluorinated intermediates (HCFC-21) and chloroform are then
9 condensed and returned to the reactor, along with residual catalyst, to undergo further fluorination. The final
10 vapors leaving the condenser are primarily HCFC-22, HFC-23, HCl and residual HF. The HCl is recovered as a useful
11 byproduct, and the HF is removed. Once separated from HCFC-22, the HFC-23 may be released to the atmosphere,
12 recaptured for use in a limited number of applications, or destroyed.

13 Two facilities produced HCFC-22 in the United States in 2022. Emissions of HFC-23 from this activity in 2022 were
14 estimated to be 1.8 MMT CO₂ Eq. (0.1 kt) (see Table 4-57 and Table 4-58). This quantity represents an 18 percent
15 increase from 2021 emissions and a 95 percent decrease from 1990 emissions. The decrease from 1990 emissions
16 was caused primarily by changes in the HFC-23 emission rate (kg HFC-23 emitted/kg HCFC-22 produced). The
17 decrease from 2021 emissions was caused by both a decrease in the HFC-23 emission rate at one plant and a
18 decrease in the total quantity of HCFC-22 produced. The long-term decrease in the emission rate is primarily
19 attributable to six factors: (a) five plants that did not capture and destroy the HFC-23 generated have ceased
20 production of HCFC-22 since 1990; (b) one plant that captures and destroys the HFC-23 generated began to
21 produce HCFC-22; (c) one plant implemented and documented a process change that reduced the amount of HFC-
22 23 generated; (d) the same plant began recovering HFC-23, primarily for destruction and secondarily for sale; (e)
23 another plant began destroying HFC-23; and (f) the same plant, whose emission rate was higher than that of the
24 other two plants, ceased production of HCFC-22 in 2013.

25 Emissions from HCFC-22 production are reported under fluorochemical production (CRT category 2B9) in this
26 *Inventory*, which also includes the production of fluorochemicals other than HCFC-22 described further in section
27 4.15 of this chapter.

28 **Table 4-57: HFC-23 Emissions from HCFC-22 Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
HCFC-22 Production	38.6	16.8	2.7	3.1	1.8	2.2	1.8

29 **Table 4-58: HFC-23 Emissions from HCFC-22 Production (kt HFC-23)**

Year	1990	2005	2018	2019	2020	2021	2022
HCFC-22 Production	3	1	+	+	+	+	+

30 + Does not exceed 0.5 kt.

31 Methodology and Time-Series Consistency

32 To estimate HFC-23 emissions for five of the eight HCFC-22 plants that have operated in the United States since
33 1990, methods comparable to the Tier 3 methods in the *2006 IPCC Guidelines* (IPCC 2006) were used throughout
34 the time series. Emissions for 2010 through 2022 were obtained through reports submitted by U.S. HCFC-22
35 production facilities to EPA's Greenhouse Gas Reporting Program (GHGRP). EPA's GHGRP mandates that all HCFC-
36 22 production facilities report their annual emissions of HFC-23 from HCFC-22 production processes and HFC-23
37 destruction processes. Previously, data were obtained by EPA through collaboration with an industry association

⁵⁰ As construed, interpreted, and applied in the terms and conditions of the *Montreal Protocol on Substances that Deplete the Ozone Layer* [42 U.S.C. §7671m(b), CAA §614].

1 that received voluntarily reported HCFC-22 production and HFC-23 emissions annually from all U.S. HCFC-22
 2 producers from 1990 through 2009. These emissions were aggregated and reported to EPA on an annual basis.

3 For the other three plants, the last of which closed in 1993, methods comparable to the Tier 1 method in the 2006
 4 IPCC Guidelines were used. Emissions from these three plants have been calculated using the recommended
 5 emission factor for unoptimized plants operating before 1995 (0.04 kg HCFC-23/kg HCFC-22 produced).

6 The five plants that have operated since 1994 measure (or, for the plants that have since closed, measured)
 7 concentrations of HFC-23 as well as mass flow rates of process streams to estimate their generation of HFC-23.
 8 Plants using thermal oxidation to abate their HFC-23 emissions monitor the performance of their oxidizers to verify
 9 that the HFC-23 is almost completely destroyed. One plant that releases a small fraction of its byproduct HFC-23
 10 periodically measures HFC-23 concentrations at process vents using gas chromatography. This information is
 11 combined with information on quantities of products (e.g., HCFC-22) to estimate HFC-23 emissions.

12 To estimate 1990 through 2009 emissions, reports from an industry association were used that aggregated HCFC-
 13 22 production and HFC-23 emissions from all U.S. HCFC-22 producers and reported them to EPA (ARAP 1997, 1999,
 14 2000, 2001, 2002, 2003, 2004, 2005, 2006, 2007, 2008, 2009, and 2010). To estimate 2010 through 2022
 15 emissions, facility-level data (including both HCFC-22 production and HFC-23 emissions) reported through EPA's
 16 GHGRP were analyzed. In 1997 and 2008, comprehensive reviews of plant-level estimates of HFC-23 emissions and
 17 HCFC-22 production were performed (RTI 1997; RTI 2008). The 1997 and 2008 reviews enabled U.S. totals to be
 18 reviewed, updated, and where necessary, corrected. The reviews also allowed plant-level uncertainty analyses
 19 (Monte-Carlo simulations) to be performed for 1990, 1995, 2000, 2005, and 2006. Estimates of annual U.S. HCFC-
 20 22 production are presented in Table 4-59.

21 **Table 4-59: HCFC-22 Production (kt)**

Year	1990	2005	2012	2018	2019	2020	2021	2022
Production	139	156	96	C	C	C	C	C

C (CBI)

Note: HCFC-22 production in 2013 through 2022 is considered confidential business information (CBI) as there were only two producers of HCFC-22 in those years.

22 **Uncertainty**

23 The uncertainty analysis presented in this section was based on a plant-level Monte Carlo stochastic simulation for
 24 2006. The Monte Carlo analysis used estimates of the uncertainties in the individual variables in each plant's
 25 estimating procedure. This analysis was based on the generation of 10,000 random samples of model inputs from
 26 the probability density functions for each input. A normal probability density function was assumed for all
 27 measurements and biases except the equipment leak estimates for one plant; a log-normal probability density
 28 function was used for this plant's equipment leak estimates. The simulation for 2006 yielded a 95-percent
 29 confidence interval for U.S. emissions of 6.8 percent below to 9.6 percent above the reported total.

30 The relative errors yielded by the Monte Carlo stochastic simulation for 2006 were applied to the U.S. emission
 31 estimate for 2022. The resulting estimates of absolute uncertainty are likely to be reasonably accurate because (1)
 32 the methods used by the two remaining plants to estimate their emissions are not believed to have changed
 33 significantly since 2006, and (2) although the distribution of emissions among the plants has changed between
 34 2006 and 2022 (because one plant has closed), the plant that currently accounts for most emissions had a relative
 35 uncertainty in its 2006 (as well as 2005) emissions estimate that was similar to the relative uncertainty for total
 36 U.S. emissions. Thus, the closure of one plant is not likely to have a large impact on the uncertainty of the national
 37 emission estimate.

38 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-60. HFC-23 emissions
 39 from HCFC-22 production were estimated to be between 1.7 and 2.0 MMT CO₂ Eq. at the 95 percent confidence
 40 level. This indicates a range of approximately 7 percent below and 10 percent above the emission estimate of 1.8
 41 MMT CO₂ Eq.

1 **Table 4-60: Approach 2 Quantitative Uncertainty Estimates for HFC-23 Emissions from HCFC-**
 2 **22 Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
HCFC-22 Production	HFC-23	1.8	1.7	2.0	-7%	+10%

^a Range of emissions reflects a 95 percent confidence interval.

3 QA/QC and Verification

4 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
 5 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the
 6 introduction of the IPPU chapter (see Annex 8 for more details). Under the GHGRP, EPA verifies annual facility-level
 7 reports through a multi-step process (e.g., including a combination of pre-and post-submittal electronic checks and
 8 manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate,
 9 complete, and consistent (EPA 2015).⁵¹ Based on the results of the verification process, EPA follows up with
 10 facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of
 11 general and category-specific QC procedures, including: range checks, statistical checks, algorithm checks, and
 12 year-to-year checks of reported data and emissions.

13 The GHGRP also requires source-specific quality control measures for the HCFC-22 Production category. Under
 14 EPA's GHGRP, HCFC-22 producers are required to (1) measure concentrations of HFC-23 and HCFC-22 in the
 15 product stream at least weekly using equipment and methods (e.g., gas chromatography) with an accuracy and
 16 precision of 5 percent or better at the concentrations of the process samples, (2) measure mass flows of HFC-23
 17 and HCFC-22 at least weekly using measurement devices (e.g., flowmeters) with an accuracy and precision of 1
 18 percent of full scale or better, (3) calibrate mass measurement devices at the frequency recommended by the
 19 manufacturer using traceable standards and suitable methods published by a consensus standards organization,
 20 (4) calibrate gas chromatographs at least monthly through analysis of certified standards, and (5) document these
 21 calibrations.

22 Recalculations Discussion

23 No recalculations were performed for the 1990 through 2021 portion of the time series.

24 Planned Improvements

25 At this time, there are no specific planned improvements for estimating HFC-23 emissions from HCFC-22
 26 production.

⁵¹ EPA (2015). Greenhouse Gas Reporting Program Report Verification. Available online at:
https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

4.15 Production of Fluorochemicals Other Than HCFC-22 (CRT Source Category 2B9b)

Twenty facilities in the U.S. produced or transformed approximately 150 fluorinated gases other than HCFC-22 in 2022, including saturated and unsaturated hydrofluorocarbons (HFCs), saturated and unsaturated perfluorocarbons (PFCs), sulfur hexafluoride (SF₆), nitrogen trifluoride (NF₃), hydrofluoroethers (HFEs), perfluoroalkylamines, and dozens of others. Emissions from fluorochemical production may include emissions of the intentionally manufactured chemical as well as reactant and by-product emissions. The compounds emitted depend upon the production or transformation process, but may include, e.g., HFCs, PFCs, SF₆, nitrous oxide (N₂O), NF₃, and many others. Potential sources of fluorinated GHG emissions at fluorochemical production facilities include process vents, equipment leaks, and evacuating returned containers.⁵² Production-related emissions of fluorinated GHGs occur from both process vents and equipment leaks. Process vent emissions occur from manufacturing equipment such as reactors, distillation columns, and packaging equipment. Equipment leak emissions, or fugitive emissions, occur from valves, flanges, pump seals, compressor seals, pressure relief valves, connectors, open-ended lines, and sampling connections. In addition, users of fluorinated GHGs may return empty containers (e.g., cylinders) to the production facility for reuse; prior to reuse, the residual fluorinated GHGs (often termed “heels”) may be evacuated from the container and are a potential emission source. In many cases, these “heels” are contaminated and are exhausted to a treatment device for destruction. In other cases, however, they are released into the atmosphere.⁵³

Emissions of all HFCs, PFCs, NF₃, and SF₆ from production of fluorochemicals other than hydrochlorofluorocarbon (HCFC)-22 are presented in Table 4-61 below for the years 1990, 2005, and the period 2018 to 2022.

The fluorinated GHG emissions reported under the Greenhouse Gas Reporting Program (GHGRP) include emissions of HFCs, PFCs, SF₆, NF₃, and numerous “other” fluorinated GHGs, such as octafluorotetrahydrofuran (C₄F₈O), trifluoromethyl sulfur pentafluoride (SF₅CF₃), and hexafluoropropylene oxide. Because they are not included among the seven UNFCCC-reportable gases or gas groups, the “other” fluorinated GHGs are not included in *Inventory* totals. However, their emissions are presented below because they often have high GWPs and large GWP-weighted emissions.

Total emissions of HFCs, PFCs, SF₆, and NF₃ from fluorochemical production are estimated to have increased from 39 MMT CO₂ Eq. (4,400 MT) in 1990 to a peak of 44 MMT CO₂ Eq. (6,200 MT) in 2002, declining to 3.3 MMT CO₂ Eq. (730 MT) in 2022. These trends reflect estimated changes in fluorinated gas production and increasing use of control devices. Prior to 2002, only 2 facilities are known to have operated control devices to destroy fluorinated GHG emissions. After 2002, additional production facilities began to install and use control devices to destroy fluorinated GHG emissions,⁵⁴ and fluorinated GHG emissions declined sharply from 44 MMT CO₂ Eq. (6,200 MT) in 2002 to 9.8 MMT CO₂ Eq. (1,900 MT) in 2005. There was a small upward trend in emissions from 2006 to 2009. An additional 2 facilities installed controls in 2011 and 2012, resulting in a decline of emissions from 9.7 MMT CO₂ Eq. (2,500 MT) in 2010 to 6.7 MMT CO₂ Eq. (1,300 MT) in 2012. Another 2 facilities installed controls in 2015 and 2016. Total fluorinated GHG emissions have continued to trend downward from 2014 (4.7 MMT CO₂ Eq. [980 MT]) to 2022 (3.3 MMT CO₂ Eq. [730 MT]).

⁵² The totals presented below also include emissions from destruction of previously produced fluorinated GHGs that are shipped to production facilities for destruction, e.g., because they are found to be irretrievably contaminated.

⁵³ IPCC (2019) *2019 Refinement to the 2006 IPCC Guidelines for National Greenhouse Gas Inventories*, Calvo Buendia, E., Tanabe, K., Kranjc, A., Baasansuren, J., Fukuda, M., Ngarize S., Osako, A., Pyrozhenko, Y., Shermanau, P. and Federici, S. (eds). Published: IPCC, Switzerland.

⁵⁴ One facility installed controls in 2003, and four facilities are assumed to have installed controls in 2005.

1 Emissions from the production of fluorochemicals other than HCFC-22 are reported under fluorochemical
2 production (CRT category 2B9) in conjunction with emissions from HCFC-22 production described in Section 4.14 of
3 this chapter.

4 **HFC Emissions**

5 Estimated emissions of HFCs increased from 9.2 MMT CO₂ Eq. in 1990 to a peak of 15 MMT CO₂ Eq. in 2004 (1,200
6 to 3,200 MT), declining with some fluctuation to 1.4 MMT CO₂ Eq. in 2022. Emissions in 1990 were primarily from
7 facilities producing compounds other than saturated HFCs. The subsequent trends in emissions were driven by the
8 growth in production of saturated HFCs and the imposition of controls. Production of saturated HFCs is estimated
9 to have increased from around 0.3 MMT CO₂ Eq. (2,000 MT) in 1990 to over 300 MMT CO₂ Eq. (100,000 MT) by
10 2010 as HFCs replaced ozone-depleting substances, which were being phased out under the Montreal Protocol and
11 Clean Air Act (U.S. EPA 2023a; U.S. EPA 2023b). This increase in HFC production drove HFC emissions to their 2004
12 peak. However, estimated emissions declined sharply in 2005 to 4.8 MMT CO₂ Eq. (1,500 MT) due to the assumed
13 addition of controls in that year. Estimated emissions of HFCs resumed their increase from 2005 to 2010 at 6.9
14 MMT CO₂ Eq. (2,300 MT), but again declined sharply in 2011 to 4.2 MMT CO₂ Eq. (1,200 MT) based on addition of
15 controls. Since 2012, HFC emissions have continued to trend downward. With the phase-out of production of
16 saturated HFCs (Kigali Amendment, and U.S. AIM program), the downward trend of HFC emissions is expected to
17 continue, but the share of HFC emissions that are not associated with saturated HFC production (approximately 0.7
18 MMT CO₂ Eq. in 2022) is likely to persist in the absence of additional controls.

19 **PFC Emissions**

20 Overall emissions of PFCs were relatively steady from 1990 to 2002 but dropped sharply from 25 MMT CO₂ Eq.
21 (2,900 MT) in 2002 to 1.6 MMT CO₂ Eq. (180 MT) in 2005, reflecting the addition of controls at high-emitting
22 facilities. Overall PFC emissions from 2005 to 2022 have remained steady, oscillating around 1.5 MMT CO₂ Eq. The
23 quantities of fluorinated GHGs produced or transformed at facilities emitting PFCs are estimated to have remained
24 generally steady between 1990 and 2009 and therefore do not contribute to the emissions trend before 2010. For
25 most of the fluorinated GHGs produced at these facilities, there was no available industry information to inform
26 activity estimates or trends for 1990 to 2009. Therefore, as discussed in the Methodology section below, 2010
27 production values from EPA's GHGRP were assumed to have held constant for these compounds from 1990 to
28 2010.

29 **SF₆ Emissions**

30 Emissions of SF₆ are estimated to have remained steady from 1990 to 2002 at roughly 3.8 MMT CO₂ Eq. (160 MT),
31 declining to 3.0 MMT CO₂ Eq. in 2003 due to the imposition of controls at one facility. Emissions declined more
32 sharply between 2006 and 2011 (3.0 million to 0.030 MMT CO₂ Eq. [130 to 1.3 MT]) due to the phaseout of
33 production at the major U.S. SF₆-producing facility, which is estimated to have been the largest source of SF₆
34 emissions from fluorochemical production. SF₆ emissions have continued to decline from 2011 to 2022, with the
35 exception of 2013 and 2014, when emissions increased briefly. In this analysis, U.S. SF₆ production is assumed to
36 follow the trend of U.S. SF₆ consumption except where facility production capacity caps production at a lower
37 level. SF₆ is used in several industries, including for electric power transmission and distribution (T&D) equipment,
38 magnesium production, and semiconductor manufacturing. Total SF₆ consumption, i.e., considering the trend
39 based on all three industries combined, is estimated to have decreased from 1990 to 1999, fluctuated through
40 2006, and declined through 2010.

41 **NF₃ Emissions**

42 Estimated emissions of NF₃ had a slight upward trend from 1990 to 2002 (0.69 MMT CO₂ Eq. to 0.83 MMT CO₂ Eq.
43 [43 MT to 52 MT]), declined sharply in 2003 to 0.28 MMT CO₂ Eq. (17 MT) due to one facility installing controls,
44 and then resumed a steady climb through 2010 to 0.70 MMT CO₂ Eq. (43 MT). After 2010, NF₃ emissions decreased

1 through 2018 to 0.11 MMT CO₂ Eq. (6.7 MT), and then increased between 2018 and 2022 to 0.50 MMT CO₂ Eq. (31
 2 MT). NF₃ may be emitted both from the production of NF₃ and from the production of other fluorochemicals. The
 3 dominant source since 2010 has been production of NF₃. Trends after 2010 were driven by changes both in NF₃
 4 production and in the emission rate (kg NF₃ emitted/kg NF₃ produced) for NF₃ production, with both contributing
 5 to increased emissions since 2018. For 1990 through 2009, the NF₃ that is emitted from the production of NF₃ is
 6 assumed to be influenced by the trajectory of NF₃ production, which is generally assumed to follow production
 7 trends in the semiconductor industry except where NF₃ facility capacity limits production further. Semiconductor
 8 production increased steadily from 1995 to 2007 but is estimated to have declined from 2007 through 2010. The
 9 NF₃ that is emitted from production of other fluorochemicals is affected by the production trends of the
 10 fluorochemicals at the emitting facility, which are assumed to have been flat before 2009 in most cases.

11 Other Fluorinated GHG Emissions

12 Other fluorinated GHGs, i.e., those not included in the UNFCCC-reportable gases or gas groups, are also emitted in
 13 significant quantities from fluorinated gas production and transformation processes. Estimated emissions of these
 14 other fluorinated GHGs are provided in Table 4-63 for the years 1990, 2005, and the period 2018 to 2022. The
 15 other fluorinated GHGs with the highest estimated emissions in 2022 are presented separately, and the remaining
 16 other fluorinated GHGs are aggregated.

17 Estimated emissions of these other fluorinated GHGs have declined over the time series, primarily due to the
 18 installation of control devices. Emissions of other fluorinated GHGs were steady from 1990 to 2002, at roughly 9.6
 19 MMT CO₂ Eq. (800 MT). These emissions declined sharply in 2003 to 0.88 MMT CO₂ Eq. (120 MT) due to the
 20 installation of controls at a major emitting facility, and they continued to slightly decline through 2012 to 0.82
 21 MMT CO₂ Eq. (110 MT). As is the case at facilities emitting PFCs, the quantities of fluorinated GHGs produced or
 22 transformed at facilities emitting other fluorinated GHGs are estimated to have remained generally steady
 23 between 1990 and 2009 and therefore do not contribute to the emissions trend before 2010. From 2013 through
 24 2019, emissions of other fluorinated GHGs fluctuated. They declined sharply in 2020 to around 0.13 MMT CO₂ Eq.
 25 due to a decrease in the emission rate at one facility, and they remained near this value through 2022.

26 **Table 4-61: Emissions of HFCs, PFCs, SF₆, and NF₃ from Production of Fluorochemicals Other**
 27 **Than HCFC-22 (MMT CO₂ Eq.)**

Gas	1990	2005	2018	2019	2020	2021	2022
HFC-23	6.8	1.5	0.7	0.6	0.4	0.3	0.3
1H,4H-Perfluorobutane	0.0	0.0	0.0	+	+	+	0.2
HFC-125	0.1	1.9	0.3	0.3	0.3	0.2	0.2
1H,6H-Perfluorohexane	0.0	0.0	0.0	+	+	+	0.2
HFC-134a	+	0.4	0.2	0.2	0.2	0.2	0.1
HFC-143a	0.1	0.7	0.6	0.4	0.2	0.1	0.1
Other HFCs	2.0	0.4	0.2	0.2	0.2	0.2	0.3
Perfluorocyclobutane	11.0	0.5	0.7	0.9	0.7	0.9	0.8
PFC-14 (Perfluoromethane)	6.7	0.4	0.4	0.4	0.5	0.6	0.5
Other PFCs	7.4	0.7	0.2	0.3	0.1	0.1	0.1
Sulfur hexafluoride	3.8	3.0	+	+	+	+	+
Nitrogen trifluoride	0.7	0.4	0.1	0.6	0.7	0.5	0.5
Total HFCs, PFCs, SF₆, and NF₃	38.8	9.8	3.3	3.9	3.2	3.1	3.3

+ Does not exceed 0.05 MMT CO₂ Eq.

1 **Table 4-62: Emissions of HFCs, PFCs, SF₆, and NF₃ from Production of Fluorochemicals Other**
 2 **Than HCFC-22 (Metric Tons)**

Gas	1990	2005	2018	2019	2020	2021	2022
HFC-23	552	118	55	46	35	23	28
1H,4H-Perfluorobutane	0	0	0	1	1	1	53
HFC-125	43	584	82	92	80	78	56
1H,6H-Perfluorohexane	0	0	0	1	+	1	41
HFC-134a	13	311	117	146	119	123	113
HFC-143a	29	147	124	93	40	28	26
Other HFCs	596	380	207	205	189	210	211
Perfluorocyclobutane	1,157	53	73	95	76	95	82
PFC-14 (Perfluoromethane)	1,017	53	65	62	77	83	76
Other PFCs	785	77	24	36	10	14	13
Sulfur hexafluoride	163	127	+	+	+	+	+
Nitrogen trifluoride	43	26	7	35	45	31	31
Total HFCs, PFCs, SF₆, and NF₃	4,398	1,877	753	814	670	687	730

+ Does not exceed 0.5 MT.

3 **Table 4-63: Emissions of Other Fluorinated GHGs from Production of Fluorochemicals Other**
 4 **Than HCFC-22 (MMT CO₂ Eq.)**

Gas	1990	2005	2018	2019	2020	2021	2022
Octafluorotetrahydrofuran	5.5	0.3	+	0.1	+	+	+
1,1,1,2,2,3,3-Heptafluoro-3-(1,2,2,2-tetrafluoroethoxy)-propane	+	+	+	+	+	+	+
Trifluoromethyl sulfur pentafluoride	2.5	0.1	+	0.1	+	+	+
Hexafluoropropylene oxide	0.4	0.3	0.3	0.3	+	+	+
FC-3283/FC-8270 (Perfluorotripropylamine)	+	+	+	+	+	+	+
Others	1.2	0.1	0.3	0.2	0.1	0.1	0.1
Total Other Fluorinated GHGs	9.6	0.9	0.6	0.6	0.1	0.1	0.1

+ Does not exceed 0.05 MMT CO₂ Eq.

5 **Table 4-64: Emissions of Other Fluorinated GHGs from Production of Fluorochemicals Other**
 6 **Than HCFC-22 (Metric Tons)**

Gas	1990	2005	2018	2019	2020	2021	2022
Octafluorotetrahydrofuran	398	21	4	4	2	1	2
1,1,1,2,2,3,3-Heptafluoro-3-(1,2,2,2-tetrafluoroethoxy)-propane	6	4	5	6	3	6	3
Trifluoromethyl sulfur pentafluoride	145	8	3	4	1	1	1
Hexafluoropropylene oxide	35	33	32	32	2	2	2
FC-3283/FC-8270 (Perfluorotripropylamine)	+	+	+	+	1	1	1
Others	228	49	76	84	35	33	35
Total Other Fluorinated GHGs	812	115	120	130	43	43	45

+ Does not exceed 0.5 MT.

7 Methodology

1 The 2006 IPCC Guidelines as elaborated by the 2019 Refinement include Tier 1, Tier 2, and Tier 3 methods for
2 estimating fluorinated GHG emissions from production of fluorinated compounds. The Tier 1 method calculates
3 emissions by multiplying a default emission factor by total production. Specific default emission factors exist for
4 production of SF₆ and NF₃; a more general default emission factor covers production of all other fluorinated GHGs.
5 (The more general default emission factor was developed based on data from U.S. facilities collected under the
6 GHGRP between 2011 and 2016.) The Tier 2 method calculates emissions using a mass-balance approach. The Tier
7 3 method is based on the collection of plant-specific data on the types and quantities of fluorinated GHGs emitted
8 from vents, leaks, container venting, and other sources, considering any abatement technology. The Tier 3 method
9 is often implemented by developing and applying facility-specific emission factors indexed to production.

10 Based on available data on emissions and activity, EPA used a form of the IPCC Tier 3 method to estimate
11 fluorinated GHG emissions from most U.S. production of fluorinated compounds. Emissions from U.S. production
12 for which there are fewer data are based on the Tier 1 method.

13 **Overview of GHGRP Data for this Source Category**

14 As discussed further below, much of the data used to develop the estimates presented here come from the
15 GHGRP. The data were collected under two sections of the GHGRP regulation—Subpart L, Fluorinated Gas
16 Production; and Subpart OO, Suppliers of Industrial Greenhouse Gases. Under Subpart L, certain fluorinated gas
17 production facilities must report their emissions from a range of processes and sources, detailed further below.
18 Data collected under Subpart L include emissions data for calendar years 2011 through 2022. Under Subpart OO,
19 fluorinated GHG suppliers (including fluorinated GHG producers) must report the quantities of each fluorinated
20 GHG that they produce, transform, destroy, import, or export. Data collected under Subpart OO include
21 production and transformation data for calendar years 2010 through 2022. Facilities’ production and
22 transformation data are not shown here because they are considered confidential business information under the
23 GHGRP.

24 ***Emissions Reported Under Subpart L of the GHGRP***

25 Under Subpart L, facilities that produce a fluorinated gas must report their greenhouse gas emissions if the facility
26 emits 25,000 MT CO₂ Eq. or more per year in combined emissions from fluorinated gas production, stationary fuel
27 combustion units, miscellaneous uses of carbonate, and all other applicable source categories listed in the rule.
28 (For purposes of calculating emissions from fluorinated gas production for inclusion in the total that is compared
29 to the threshold, emissions are assumed to be uncontrolled.) Facilities must report their fluorinated GHG
30 emissions from the production and transformation of fluorinated gases, from venting of residual fluorinated GHGs
31 from containers, and from destruction of previously produced fluorinated GHGs. The emissions reported from
32 production and transformation include both emissions from process vents and emissions from equipment leaks.

33 Under the GHGRP, “fluorinated GHGs,” whose emissions must be reported, include SF₆, NF₃, and any fluorocarbon
34 except for substances with vapor pressures below 1 Torr at 25 degrees C and substances that are regulated as
35 “controlled substances” under EPA’s ozone-protection regulations at 40 CFR Part 82, Subpart A (e.g.,
36 chlorofluorocarbons [CFCs], hydrochlorofluorocarbons [HCFCs], and halons). In addition to SF₆ and NF₃, this
37 definition includes hydrofluorocarbons (HFCs), perfluorocarbons (PFCs), hydrofluoroethers (HFEs), fully fluorinated
38 tertiary amines, perfluoropolyethers (including PPFMIE), and hydrofluoropolyethers, and others. “Fluorinated
39 gases,” from whose production or transformation emissions must be reported, include the fluorinated GHGs
40 detailed above as well as CFCs and HCFCs.⁵⁵

41 Facilities calculate emissions from process vents using one of two methods. For vents that emit 10,000 MT CO₂ Eq.
42 or more (considering controls) of fluorinated GHGs from continuous processes, facilities must use emissions testing
43 to establish an emission factor at least every ten years, or sooner if the process changes in a way that will

⁵⁵ HCFC-22 is considered a fluorinated gas under the GHGRP, but emissions from HCFC-22 production are reported separately from emissions from production of other fluorinated gases.

1 significantly affect emissions from the vent. For other process vents, facilities may use emissions testing,
 2 engineering calculations, or engineering assessments to establish the emission factor. Facilities then calculate their
 3 annual emissions based on the measured or calculated emission factor and related activity data, considering the
 4 extent to which the process is controlled and any destruction device or process malfunctions.

5 To calculate emissions from equipment leaks, facilities that report under Subpart L are required to collect
 6 information on the number and type of pieces of equipment; service of each piece of equipment; concentration of
 7 each fluorinated GHG in the stream; and the time period each piece of equipment was in service. Facilities use one
 8 or more of the following methods to calculate emissions from leaks:

- 9 • Average Emission Factor Approach in EPA Protocol for Equipment Leak Estimates.
- 10 • Other Approaches in EPA Protocol for Equipment Leak Estimates in conjunction with EPA Method 21.
- 11 • Other Approaches in EPA Protocol for Equipment Leak Estimates in conjunction with site-specific leak
 12 detection methods.
- 13 • Site-specific leak detection methods.

14 Most emissions are reported by chemical; the exceptions are (1) fluorinated GHGs that are emitted in quantities of
 15 1,000 MT CO₂ Eq. or less across all production and transformation processes at a facility and (2) fluorinated GHGs
 16 that are emitted from facilities that produce only one fluorinated GHG, where the emitted fluorinated GHG is not
 17 the fluorinated gas produced. In these cases, the emissions are reported in CO₂ Eq. by fluorinated GHG group.
 18 There are 12 fluorinated GHG groups, each of which encompasses a set of GHGs with roughly similar atmospheric
 19 behavior, including similar GWPs and atmospheric lifetimes. These include, e.g., fully fluorinated GHGs such as
 20 PFCs and SF₆, saturated HFCs with two or fewer hydrogen-carbon bonds, saturated HFCs with more than two
 21 carbon-hydrogen bonds, unsaturated HFCs and PFCs, and others (see Table 4-65 for a full list).

22 **Table 4-65: Fluorinated GHG Groups Under Which Certain Emissions Are Reported Under**
 23 **Subpart L of the GHGRP and Associated GWPs**

Fluorinated GHG Group	GHGRP Default Global Warming Potential (100-yr.)
Fully fluorinated GHGs	10,000
Saturated hydrofluorocarbons (HFCs) with 2 or fewer carbon-hydrogen bonds	3,700
Saturated HFCs with 3 or more carbon-hydrogen bonds	930
Saturated hydrofluoroethers (HFEs) and hydrochlorofluoroethers (HCFEs) with 1 carbon-hydrogen bond	5,700
Saturated HFEs and HCFEs with 2 carbon-hydrogen bonds	2,600
Saturated HFEs and HCFEs with 3 or more carbon-hydrogen bonds	270
Fluorinated formates	350
Fluorinated acetates, carbonofluoridates, and fluorinated alcohols other than fluorotelomer alcohols	30
Unsaturated PFCs, unsaturated HFCs, unsaturated HCFCs, unsaturated halogenated ethers, unsaturated halogenated esters, fluorinated aldehydes, and fluorinated ketones	1
Fluorotelomer alcohols	1
Fluorinated GHGs with carbon-iodine bond(s)	1
Other fluorinated GHGs	2,000

24 Two other datasets reported under Subpart L are relevant to estimating uncontrolled emission factors. (As
 25 discussed further below, such uncontrolled emission factors are applied to years before Subpart L reporting began
 26 (for CY 2011) and before emission controls were put into place.) First, in addition to reporting emissions by

1 chemical at the facility level, facilities report emissions from each production and transformation process at the
2 facility in tons of CO₂ Eq. by fluorinated GHG group. To calculate CO₂ Eq. emissions, facilities use a chemical-specific
3 100-year GWP where one is available for the compound of interest. If no chemical-specific 100-year GWP is
4 available for the compound of interest, facilities use the GHGRP default GWP for the fluorinated GHG group of
5 which the compound is a member. These default GWPs are shown in Table 4-61.

6 Second, for each process, facilities also report the extent to which emissions are abated (the effective destruction
7 efficiency or EDE) as a range. The EDE is calculated as follows:

$$8 \quad EDE = 1 - \frac{CE_{PV}}{UE_{PV}}$$

9 where:

EDE = Effective destruction efficiency of the process

CE_{PV} = Actual GWP-weighted controlled emissions from all vents for the process, MT CO₂ Eq.

UE_{PV} = Hypothetical GWP-weighted uncontrolled emissions from all vents for the process, MT CO₂ Eq.
(CE_{PV} will equal UE_{PV} if the process is not controlled, resulting in a calculated EDE of 0).

10 Note that the EDE is based on the extent to which emissions from process vents are controlled. Emissions from
11 equipment leaks are not included in the EDE calculation. Table 4-66 provides the EDE ranges available for facilities
12 to report and the arithmetic means of each range. The use of these datasets to calculate uncontrolled emission
13 factors is discussed in more detail in the “1990-2010 Emissions Estimates” section below.

14 **Verification of GHGRP Reports**

15 Data reported under the GHGRP, including emissions and production, are electronically verified using range
16 checks, internal consistency checks, and time-series consistency checks. Where the data fail a preliminary check,
17 EPA contacts the facility to see whether there is an explanation for the issue or whether the data are indeed
18 erroneous. In the latter case, facilities are required to correct the data. Where one or more of the anomalous data
19 elements is not explained or corrected, the report for that facility for that year is considered unverified.

20 **1990-2010 Emissions Estimates**

21 For the 17 fluorinated gas production facilities that have reported their emissions under the GHGRP, 1990 through
22 2010 emissions are estimated using (1) facility- and chemical-specific emission factors based on the emissions data
23 discussed under “2011-2022 Emissions” below, (2) reported or estimated production and transformation of
24 fluorinated GHGs at each facility in each year, and (3) reported and estimated levels of emissions control at each
25 facility in each year.

26 **Facility- and Chemical-Specific Emission Factors Reflecting Emissions Controls**

27 Facility- and chemical-specific emission factors were developed based on the 2011 to 2015 emissions reported
28 under the GHGRP (discussed above) and the 2011 to 2015 production and transformation of fluorinated GHGs
29 reported under the GHGRP. (Production and transformation of CFCs and HCFCs are not reported under the
30 GHGRP.) For each emitted fluorinated GHG at each facility, emissions of the fluorinated GHG were summed over
31 the five-year period. This sum was then divided by the sum of the quantities of all fluorinated GHGs produced or
32 transformed at the facility over the five-year period.⁵⁶ As discussed further below in the Uncertainty section,

⁵⁶ Permit data for two facilities indicated that they began controlling emissions at some point between 2011 and 2015. However, the actual emissions reported by these facilities did not change substantially after the date when the permit indicated

1 emissions of any particular fluorinated GHG are likely to occur only from a subset of the production or
 2 transformation processes at each facility, but in the absence of information on chemical-specific emissions at the
 3 process level, it was assumed that all fluorinated GHG production and transformation processes at the facility emit
 4 all fluorinated GHGs at the facility. This yielded the emission factors for each fluorinated GHG at each facility. Both
 5 emissions and activity (production + transformation) totals were summed over the five-year period to account for
 6 the intermittent and variable nature of some emissions and production/transformation processes. Compounds
 7 that were not emitted or produced/transformed between 2011 and 2015 but that were emitted or
 8 produced/transformed later were assumed not to have been emitted or produced/transformed (as applicable)
 9 before 2011.

10 **Facility- and Chemical-Specific Emission Factors Reflecting No Emissions Controls**

11 The 2011 to 2015 emissions reported under the GHGRP reflect emissions controls to the extent those are
 12 implemented at each facility. Because facilities have not always controlled their fluorinated GHG emissions since
 13 1990, uncontrolled emission factors were developed for each facility to apply to years when the facility’s emissions
 14 were not believed to be controlled. To estimate uncontrolled emissions, GHGRP data were first used to assess the
 15 2011 to 2015 levels of control for each production or transformation process at each facility.

16 To calculate uncontrolled emissions from each process and fluorinated GHG group, a point estimate of the
 17 effective destruction efficiency (EDE, described above) was required and was estimated using the arithmetic mean
 18 of the lower and upper bounds of the EDE range reported for the process.⁵⁷ This was consistent with the approach
 19 taken in the *2019 Refinement* to develop the Tier 1 factor for fluorinated gas production facilities. The reported
 20 vented emissions for each process and fluorinated GHG group were divided by (1 - arithmetic mean EDE) to obtain
 21 the estimated uncontrolled emissions from process vents for that process and fluorinated GHG group. For each
 22 fluorinated GHG group, the controlled emissions across processes (including emissions from both vents and leaks)
 23 and the uncontrolled emissions across processes (including emissions from both vents and leaks) were then
 24 summed, and the first was divided by the second. This yielded an average level of control for each fluorinated GHG
 25 group at each facility. All fluorinated GHGs within each fluorinated GHG group at each facility were assumed to be
 26 controlled to the same level. To estimate the uncontrolled emissions of each fluorinated GHG within each group at
 27 each facility, the emissions of each fluorinated GHG were divided by the level of control estimated for its
 28 fluorinated GHG group at the facility. The same procedure was used to estimate uncontrolled emission factors as
 29 had been used to estimate controlled emission factors: the estimated uncontrolled 2011 to 2015 emissions of each
 30 fluorinated GHG were summed, and this sum was divided by the sum of the quantities of all fluorinated GHGs
 31 produced or transformed at the facility from 2011 to 2015.

32 **Table 4-66: Destruction Efficiency Range Values Used to Estimate Pre-Abatement Emissions**
 33 **for Production and Transformation Processes**

DE ranges	Lower Bound	Upper Bound	Arithmetic Mean of Bounds
>=0% to <75%	0.0	0.75	0.375
>=75% to <95%	0.75	0.95	0.85
>=95% to <99%	0.95	0.99	0.97
>=99%	0.99	0.9999	0.995

that controls were imposed. For this reason, the reported 2011-2015 emissions and emission factors are believed to be representative of emissions for these facilities before 2011.

⁵⁷ Note that facilities would report a range of 0% to 75% even if they do not abate emissions at all; thus, the assumption that emissions are 37.5 percent controlled may overestimate the hypothetical uncontrolled emissions of some facilities, e.g., those that do not abate any emissions.

1 **Estimated Levels of Emissions Controls**

2 As discussed above, both uncontrolled emission factors and controlled emission factors were developed for each
 3 facility and fluorinated GHG; these emission factors were developed for estimating emissions from production and
 4 transformation processes for years 1990 to 2010. The following information and assumptions were used to
 5 determine whether and when emissions from facilities were likely to have been controlled from 1990 to 2010.⁵⁸
 6 For the estimated status of emissions controls at each facility reporting under Subpart L, and, where relevant, the
 7 starting year for those controls, see Table 4-67.

- 8 • Facilities with publicly available information on the presence and use of control devices were assumed to
 9 control their emissions starting in the year specified in the publicly available information. Publicly
 10 available information included operating permits, news articles on facility modifications, company press
 11 releases, etc. Where the publicly available information documents that a control device was in place
 12 beginning in a certain year, the facility was assumed to control process emissions beginning in that year,
 13 and the controlled emission factor was used in estimating emissions for that year and the following years.
 14 The uncontrolled emission factor was used to estimate emissions in earlier years.
- 15 • In the absence of other control information, facilities that never reported DRE ranges other than “>=0% to
 16 <75%” for their production and transformation processes during reporting years 2011 and 2012 were
 17 assumed to have no control devices in place during the time period 1990 to 2012.
- 18 • Facilities that reported DRE ranges other than “>=0% to <75%” for at least one production or
 19 transformation process for 2011 or 2012 but for which other control information was not available were
 20 assumed to have begun controlling their emissions in 2005.

21 **Table 4-67: Estimated Starting Years for Emission Controls at Each Fluorinated Gas**
 22 **Production Facility Reporting under Subpart L of the GHGRP**

Facility Name	Estimated Start Year	Basis of Estimation
3M COMPANY	No controls	Never reported a DRE range other than “>=0% to <75%”
3M CORDOVA	2003	Climate News Article ⁵⁹
3M Cottage Grove Center - Site	2016	Reported a DRE range other than “>=0% to <75%” for the first time in 2016
Airgas Therapeutics LLC - Scott Medical Products	No controls	Never reported a DRE range other than “>=0% to <75%”
ANDERSON DEVELOPMENT COMPANY	No controls	Never reported a DRE range other than “>=0% to <75%”
ARKEMA, INC.	2005	Reported a DRE range other than “>=0% to <75%” in 2011
Chemours - Corpus Christi Plant	No controls	Never reported a DRE range other than “>=0% to <75%”
CHEMOURS CHAMBERS WORKS	2005	Reported a DRE range other than “>=0% to <75%” in 2011
CHEMOURS COMPANY - FAYETTEVILLE WORKS	2015	Reported a DRE range other than “>=0% to <75%” for the first time in 2015
CHEMOURS EL DORADO	2005	Reported a DRE range other than “>=0% to <75%” in 2011
CHEMOURS LOUISVILLE WORKS	No controls	Never reported a DRE range other than “>=0% to <75%”
CHEMOURS WASHINGTON WORKS	2005	Reported a DRE range other than “>=0% to <75%” in 2011

⁵⁸ For the estimated status of emissions controls at each facility reporting under Subpart L, and, where relevant, the starting year for those controls, see Table A-3.

⁵⁹ See <https://insideclimatenews.org/news/29122022/3m-cordova-illinois-pfas-cf4-pollution/>.

DAIKIN AMERICA INC.	1993	Title V operating permit ⁶⁰
HONEYWELL INTERNATIONAL INC - BATON ROUGE PLANT	2012	Title V operating permit ⁶¹
HONEYWELL INTERNATIONAL INC - GEISMAR COMPLEX	2011	Title V operating permit ⁶²
Honeywell Metropolis	No controls	Never reported a DRE range other than ">=0% to <75%" (did not report under Subpart L)
MEXICHEM FLUOR INC.	1993	Title V operating permit ⁶³
Versum Materials US, LLC	No controls	Never reported a DRE range other than ">=0% to <75%"

1 **Activity Data**

2 The activity data for production and transformation of fluorinated compounds for 1990 to 2010 are based on
3 production and transformation data reported to EPA by certain facilities for certain years, on production capacity
4 data, and on fluorinated GHG production and consumption trends estimated for the various fluorinated GHG-
5 consuming industries.

6 **Production and Production Capacity Data**

7 Production data are available from reporting to the U.S. GHGRP under Subpart OO, Suppliers of Industrial
8 Greenhouse Gases, and from an industry survey conducted by U.S. EPA in 2008 and 2009. Production and
9 transformation data were reported under Subpart OO for 2010 and later years. The responses to the industry
10 survey included production data for certain fluorinated gases at certain facilities for the years 2004, 2005, and
11 2006. 2004 to 2006 production data are available for 15 fluorinated compounds. Year 2006 production at an SF₆-
12 producing facility was estimated based on production capacity data as described below. Production of certain
13 compounds at one other facility was estimated based on 2003 production capacity estimates from SRI 2004.

14 **Estimated Production**

15 **Estimated production for facilities and fluorinated GHGs for which production or production capacity data were** 16 **available for some years before 2010.**

17 For facilities and fluorinated GHGs for which production or production capacity data were available for 2006 or
18 2003, production between 2006 or 2003 (as applicable) and 2010 (or 2011) was estimated by interpolating
19 between the 2006 production or 2003 production capacity value and the 2010 (or 2011) production value reported
20 under Subpart OO.

21 For the years before the earliest year with production or production capacity data (e.g., years 1990 to 2002 or
22 2003), production was estimated based on growth or consumption trends for the major industries using each
23 fluorinated GHG.

- 24 • For fluorinated compounds that are commonly emitted in the semiconductor industry, estimates of U.S.
25 layer-weighted semiconductor production (Total Manufactured Layer Area, or TMLA) were used to inform
26 the fluorinated compound production estimates. Fluorinated compound production values were assumed
27 to vary with TMLA from 1990 to 2002 or 2003. For example, 1998 production of PFC-14 at a particular
28 facility was estimated by multiplying the 2003 production of PFC-14 at that facility by the ratio between
29 the TMLA estimated for 1998 and the TMLA estimated for 2003. Fluorinated compounds for which TMLA
30 was used to estimate production include PFC-14, PFC-116, PFC-218, perfluorocyclobutane (c-C₄F₈), and

⁶⁰ See <http://lf.adem.alabama.gov/WebLink/DocView.aspx?id=29951882&dbid=0>.

⁶¹ See <https://edms.deq.louisiana.gov/app/doc/view?doc=8579001>.

⁶² See <https://edms.deq.louisiana.gov/app/doc/view?doc=7812895>.

⁶³ See <https://edms.deq.louisiana.gov/app/doc/view?doc=1309650>.

- 1 NF₃. (Note that the TMLA data were also extrapolated from year 1995 to 1990 based on the average
2 change per year from 1995 to 2009.)
- 3 • SF₆ is commonly used in electric power systems, magnesium production, and electronics manufacturing.
4 SF₆ consumption estimates across these three industries for 1990 to 2003 were used to inform the SF₆
5 production data; SF₆ production was assumed to vary with consumption totals from 1990 to 2003.
 - 6 • For HFCs commonly used as replacements for ozone-depleting substances (ODS), such as HFCs used as
7 substitutes for CFCs and HCFCs in air-conditioning and refrigeration equipment, HFC production data for
8 certain fluorinated compounds from the Vintaging Model (VM) were used to inform the HFC production
9 estimates (EPA 2023b). HFC production values were assumed to vary with the VM estimates of
10 production. The industry trend data were applied to the list of HFCs in Table 4-68.

11 **Table 4-68: List of Saturated HFCs, Unsaturated HFCs (Hydrofluoroolefins or HFOs), and**
12 **Unsaturated HCFCs (Hydrochlorofluoroolefins or HCFOs) whose 1990-2009 Production Was**
13 **Estimated Using Vintaging Model, Virgin Manufacturing by Chemical**

Fluorinated Gas
HFC-23
HFC-32
HFC-125
HFC-134a
HFC-143a
HFC-152a
HFC-236fa
HFC-245fa
HFC-365mfc
HCFO-1233zd(E)
HFO-1234yf
HFO-1234ze
HFO-1336mzz(Z)
HFC-4310mee

14 **Estimated production for facilities and fluorinated GHGs for which production data before 2010 were not**
15 **available.**

16 In the absence of production data for the period 1990 to 2009, the production data reported to the GHGRP under
17 Subpart OO were extrapolated backward based on the industry trends discussed above. For compounds for which
18 industry trend data were unavailable, production was assumed to have remained constant over the time series.

19 In both cases, 2009 production was estimated by conducting a trend analysis on the Subpart OO production data
20 for years 2010 to 2015. In instances where there did not appear to be a trend, the average of the production
21 values for years 2010 to 2015 was used as the estimated production for year 2009. In instances where there was a
22 trend, the year 2010 (or 2011) production value was used as the estimated production for year 2009.

23 If the industry trend information discussed above was applicable to a fluorinated compound, it was assumed that
24 production varied with the industry trend from 1990 to 2009. If no industry trend information was available, it was
25 assumed that production from 1990 to 2008 remained constant at the 2009 value.

26 For facilities and fluorinated compounds where information was available on annual production capacity, the
27 estimated activity data was reviewed and compared to the known production capacity. For instances where the
28 estimated activity data exceeded known production capacity for a certain year, the production estimate was set
29 equal to the capacity value.

2011-2022 Emissions Estimates

For the 17 fluorinated gas production facilities that have reported their emissions under the GHGRP, 2011 to 2022 emissions are estimated using the fluorinated GHG emissions reported under Subpart L of the GHGRP.

As discussed above, most emissions reported under Subpart L are reported by chemical, but some emissions are reported only by fluorinated GHG group in MT CO₂ Eq. Between 2011 and 2022, the share of total CO₂ Eq. emissions reported only by fluorinated GHG group has ranged between 1 and 2 percent. In this analysis, to ensure that all emissions are reported by species, emissions that are reported only by fluorinated GHG group are assumed to consist of the fluorinated GHGs in that group that are reported by chemical at the facility. As discussed further in the Uncertainty section, this is likely to result in incorrect speciation of some emissions, but the impact of this incorrect speciation is expected to be small.

For an SF₆ production facility that ceased production in 2010, the year before emissions from fluorinated gas production were required to be reported under the GHGRP, SF₆ emissions were estimated using historical production capacity, the global growth rate of SF₆ sales reported in RAND 2008, and the Tier 1 default emission factor for production of SF₆ in the 2019 Refinement. For this plant, a 1982 SF₆ production capacity of 1,200 short tons (Perkins 1982) was multiplied by the ratio between the RAND survey SF₆ sales totals for 2006 and 1982, 1.52 (RAND 2008), resulting in estimated production of 1,652 metric tons in 2006. This production was assumed to have declined linearly to zero in 2011.

Uncertainty

The estimates in this memo are subject to a number of uncertainties. These uncertainties are generally greater for years before 2011, when reporting of fluorinated GHG emissions from fluorinated gas production began under the GHGRP, than for 2011 and following years. However, the emissions estimated from 2011 to 2022 are also subject to various uncertainties. The uncertainties for both the 1990 to 2010 and 2011 to 2022 periods are discussed in more detail below.

1990-2010 Uncertainty

The uncertainty of emissions estimated for 1990 through 2010 is considerably greater than that for emissions for 2011 through 2022 because emissions were not reported under the GHGRP. EPA has estimated emissions using estimated emission rates, fluorochemical production and transformation activity, and levels of control, and each set of estimates is subject to uncertainty.

Uncertainty regarding activity data

Identity of emitting processes

In reality, emissions of particular fluorinated GHGs are linked to production and/or transformation of particular fluorinated gases at facilities. However, GHGRP information/data does not link emissions of specific fluorinated GHGs to production or transformation of specific fluorinated gases. For the estimates presented here, therefore, all emissions are indexed to total production across all fluorinated gases. This may not capture trends in emissions that are driven by trends in production or transformation of subsets of the fluorinated gases produced at a facility.

Produced and emitted gases change over time

The set of gases produced at a facility, and therefore the set of fluorinated GHGs that are emitted by that facility, may change over time. It is likely that certain production and transformation processes that existed from 2011 to 2015 (the basis of the emission factors used to back-cast emissions in this analysis) did not exist throughout the entire previous time series (1990 to 2010). In such cases, emissions of the fluorinated GHGs emitted from the new processes will be overestimated by this analysis for certain years before 2011. On the other hand, it is also likely that some production and transformation processes, and their associated fluorinated GHG emissions, occurred only during the 1990 to 2010 period and not later, meaning that their emissions are not represented in the

1 emission factors developed based on the 2011 to 2015 emissions and production data collected under the GHGRP.
2 Such emissions will therefore not be captured by this analysis. The most prominent example of the second
3 situation is probably production of CFCs and HCFCs other than HCFC-22 between 1990 and 2009, which has
4 declined steadily since 1990 as the production of CFCs and HCFCs for emissive uses has been phased out under the
5 Montreal Protocol and Clean Air Act. Production of CFCs and HCFCs can sometimes result in emissions of HFCs or
6 PFCs.

7 **Quantity of produced gases**

8 Where production or production capacity data were available for certain fluorinated gases, facilities, and years
9 before 2010, those data were incorporated into this analysis. However, even for facilities and compounds for
10 which data were available in certain years, there were several years for which data were not available. For multiple
11 produced compounds, data were available only in 2010. To estimate trends in production of compounds for years
12 before production or production capacity data were available, production of certain compounds was indexed to
13 known national production or consumption trends for those compounds. This is the case for most HFCs, several
14 PFCs, SF₆, and NF₃. National production estimates are available for HFCs, increasing confidence in country-level
15 production estimates, but the distribution of production among the various HFC-producing facilities is uncertain.
16 Where estimated production was indexed to consumption (for several PFCs, SF₆, and NF₃), the uncertainty is larger
17 than for HFCs because changes in net imports/exports (which are not known) may also affect the production trend.

18 For certain fluorinated gases, trend information was not available; therefore, production was back-cast by
19 assuming that it had remained constant at the 2010 level from 1990 through 2009. This is a highly uncertain
20 assumption.

21 **Some production and transformation activity is not reported under Subpart OO or modeled in back-casting**

22 Under Subpart OO, quantities of fluorinated GHGs that are produced and transformed at the same facility are not
23 reported to us, although any emissions from such processes are reported under Subpart L. Such unreported
24 production and transformation are therefore not captured in the 1990 to 2010 activity estimates used to estimate
25 1990 through 2010 emissions. To the extent that such unreported production and transformation drive emissions
26 and change over time, the trends will not be captured by this analysis.

27 **Facilities that no longer produce fluorinated gases or that started producing them after 1990**

28 Some facilities may have produced fluorinated gases at some point between 1990 and 2010 that no longer
29 produced those compounds after 2010. One SF₆ producer is known to fall into this category and its 1990 to 2010
30 emissions were estimated, but there may be other facilities that are not included in this analysis. On the other
31 hand, some facilities for which 1990 to 2010 emissions were estimated may not have produced them over the
32 entire time series, in which case emissions of the compounds those facilities are assumed to have emitted could be
33 overestimated.

34 ***Uncertainty regarding emission factors***

35 **Emission rates change over time**

36 The emission factors used to estimate 1990 to 2010 emissions are based on the emissions and production reported
37 from 2011 to 2015, reflecting emission rates during that period. For processes that have been used throughout the
38 timeseries, emission rates may have changed over time as the process was optimized to increase efficiency,
39 decreasing by-product emissions, or alternatively, as the process was optimized to maximize production, which
40 sometimes increases by-product emissions. Emission rates also depend on the extent to which emissions are
41 controlled at the facility, the uncertainties for which are discussed further below.

42 **Emissions from container venting and destruction may not scale with production**

43 In this analysis, emissions from container venting and destruction of previously produced fluorinated GHGs were
44 included in the emission factors used to estimate 1990 to 2010 emissions. This implicitly assumes that such
45 emissions scale with production and transformation. While this seems likely to be broadly true, there may be

1 exceptions. However, since emissions from container venting and destruction are generally a small share of facility
2 emissions (2 percent, on average), the impact of such exceptions is expected to be small.

3 **Uncertainty regarding levels of control**

4 In this analysis, the arithmetic mean of the DRE range reported by each facility for each process was used to
5 estimate the DRE for that process and the uncontrolled emissions for that process. Since the emissions implied by
6 the bounds of each DRE range span at least a factor of four,⁶⁴ this is an uncertain assumption. The uncertainty is
7 mitigated somewhat by the fact that there are generally several processes at each facility, meaning that
8 departures from the assumed mean average out to some extent. There is also uncertainty in the assumptions that
9 (1) all fluorinated GHGS within a particular fluorinated GHG group are abated to the same extent and (2) facilities
10 for which control device start dates are unavailable began to control emissions in 2005.

11 **Quantitative uncertainty estimate for uncontrolled emission factors from 2019 Refinement**

12 As noted above, 2011 to 2016 data from the GHGRP was used to develop the Tier 1 default uncontrolled emission
13 factor for the 2019 Refinement, using methods similar to those described here. A Monte Carlo analysis performed
14 to assess the uncertainty of the Tier 1 default factor indicated that the uncertainty for each facility's uncontrolled
15 emission factor was less than 50 percent. This uncertainty estimate considered the uncertainty regarding the levels
16 of control, but not the uncertainty of applying factors from one time period at the facility to much earlier time
17 period (although the variability of each facility's emission factor over the 6-year span of the *2019 Refinement*
18 analysis was found to be relatively low).

19 **2011-2022 Uncertainty**

20 Emissions from 2011 to 2022 reflect reporting by fluorinated gas production facilities under the GHGRP. As
21 discussed above, emissions reported under the GHGRP are based on facility- and process-specific measurements
22 or calculations and are therefore expected to be reasonably accurate for the reporting facilities. (Emissions from
23 the largest sources, process vents emitting 10,000 MT CO₂ Eq. or more annually, are estimated using Tier 3
24 methods.)

25 **Unverified reports**

26 Ninety-five percent (171/180) of the Subpart L reports submitted by fluorinated gas production facilities from 2011
27 to 2022 are considered to be fully verified; five percent (9/180) of the reports include one or more data elements
28 that are not verified. One facility accounts for two thirds (6/9) of the unverified reports. Many of the issues in the
29 unverified reports for this facility relate to time-series inconsistencies that have arisen as the facility updates
30 reports for recent years, but not previous years, to reflect refinements to estimated emission rates. This facility has
31 accounted for between 6 percent (in 2011) and 29 percent (in 2022) of the GWP-weighted emissions reported for
32 this source category. The uncertainties for this facility therefore have an appreciable impact on the uncertainty of
33 the estimates for the source category as a whole, particularly in years before 2022.

34 **Facilities that produce fluorinated gases but do not report their emissions to the GHGRP**

35 As noted above, EPA is still in the process of estimating emissions for fluorinated gas production facilities that do
36 not report their emissions under Subpart L of the GHGRP. The estimates presented here for 2011 to 2022 are
37 therefore incomplete. Based on aggregated production estimates and the Tier 1 default emission factor in the
38 *2019 Refinement*, it is expected that emissions from non-reporting facilities will account for less than twenty
39 percent of total U.S. emissions from fluorinated gas production and transformation.

⁶⁴ For example, the DRE range 0 to 75% implies emissions of (1-0) x uncontrolled emissions to (1-75%) x uncontrolled emissions, or, rearranging and calculating, 0.25 x uncontrolled emissions to 1 x uncontrolled emissions, a factor of four.

1 **Facilities that do not produce fluorinated gases but may emit fluorinated GHGs from other fluorochemical**
2 **production processes**

3 Under the GHGRP, EPA collects information from facilities that produce fluorinated gases. While this likely includes
4 most, and possibly all, U.S. facilities that produce fluorochemicals of any kind, it is possible that some
5 fluorochemical producers do not report either their production of fluorochemicals or their emissions of fluorinated
6 GHGs to EPA under the GHGRP. In this case, emissions estimates based only on GHGRP reporting would
7 underestimate actual emissions.

8 At fluorinated gas production facilities that currently report their emissions under the GHGRP, it is possible that
9 some processes that emit fluorinated GHGs neither produce nor transform a fluorinated gas, in which case their
10 emissions would not be reported under the GHGRP. In that case, emissions estimates based only on GHGRP
11 reporting would underestimate actual emissions.

12 **Exclusion of nitrous oxide**

13 The GHGRP does not currently require facilities to report emissions of nitrous oxide (N₂O) from fluorinated gas
14 production or transformation, but the IPCC 2019 Refinement includes a default emission factor for N₂O from
15 production of NF₃, implying such emissions may occur. The GHGRP data (and this analysis) may therefore
16 underestimate emissions of N₂O from fluorinated gas production. Because the GWP of N₂O is considerably lower
17 than that of saturated HFCs, PFCs, and other fluorinated GHGs, any underestimate is expected to be relatively
18 small.

19 **Identity of emitted compounds**

20 In this analysis, it is assumed that emissions that are reported only in MT CO₂ Eq. by fluorinated GHG group consist
21 of the compounds in that group that are reported by species by the facility. However, if that were actually the
22 case, emissions of those compounds would have been included in the speciated emissions rather than reported
23 separately in MT CO₂ Eq. This analysis therefore incorrectly speciates some emissions. However, as noted in the
24 Methodology section, the share of total CO₂ Eq. emissions reported only by fluorinated GHG group is small, ranging
25 between 1 and 2 percent. Moreover, while the emissions are not assigned to the exact species emitted, they are
26 assigned to a species that is closely related and likely to have similar atmospheric impacts (e.g., another saturated
27 HFC with two or fewer carbon-hydrogen bonds). The impact of this uncertainty is therefore limited.

28 **Quantities of Reactants Consumed or Fluorinated Gases Produced**

29 The emissions reported under Subpart L are required to be calculated using process activity data, such as the
30 quantity of reactants consumed or the quantity of the fluorinated gas product produced. In general, the
31 uncertainties in process activity levels are expected to be small. The 2019 Refinement places such uncertainties “in
32 the region of 1 percent.”

33 Because the uncertainties enumerated above are either small or difficult to quantify, EPA did not attempt to
34 include them in the 2022 quantitative uncertainty estimate for this source category. The 2022 quantitative
35 uncertainty estimate includes the following uncertainties:

36 **Process Vent Emission Factors**

37 Process vent emission factors that were developed based on stack testing (for continuous process vents emitting
38 10,000 MT CO₂ Eq. or more) were estimated to have an uncertainty (95-percent confidence interval) of ±35
39 percent based on Subpart L requirements.⁶⁵ Process vent emission factors that were developed based on
40 calculations (for batch process vents and for continuous process vents emitting less than 10,000 MT CO₂ Eq.) were

⁶⁵ *Technical Support Document for Emissions from Production of Fluorinated Gases*, Office of Air and Radiation, U.S. Environmental Protection Agency, November 5, 2010. Available online at: https://www.epa.gov/sites/default/files/2015-02/documents/subpart-l_techsuppdoc.pdf.

1 estimated to have a larger uncertainty of ±50 percent. Continuous processes were assumed to have two vents per
 2 process; batch processes were assumed to have five vents per process.⁶⁶

3 **Equipment Leak Estimates**

4 The estimated equipment leaks reported by each facility for each process were estimated to have an uncertainty
 5 of ± 90 percent. The uncertainty of leak estimates depends on the method used to estimate leaks; there are
 6 multiple methods. For simplicity, this analysis uses a conservatively high uncertainty estimate that is appropriate
 7 for the Average EF Approach.

8 **Venting of Residual Gas in Containers**

9 The reported emissions of fluorinated GHGs from venting of residual gas in returned containers were estimated to
 10 have an uncertainty of ± 30 percent for each facility. This estimate is based on the Subpart L requirement to either
 11 measure the contents of each container or to measure the contents of at least 30 representative containers for
 12 each compound and container size and type.

13 The three uncertainties listed immediately above were convolved using error propagation to arrive at an overall
 14 uncertainty estimate for 2022. The results of the Approach 1 quantitative uncertainty analysis are summarized in
 15 Table 4-69. Emissions of HFCs, PFCs, SF₆, and NF₃ from production of fluorochemicals other than HCFC-22 were
 16 estimated to fall between 2.93 and 3.75 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of
 17 approximately 12 percent below and 12 percent above the emission estimate of 3.34 MMT CO₂ Eq.

18 **Table 4-69: Approach 1 Quantitative Uncertainty Estimates for HFC, PFC, SF₆, and NF₃ from**
 19 **Production of Fluorochemicals other than HCFC-22 (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate (%)			
			Lower Bound ^a	Upper Bound ^a	Lower Bound	Upper Bound
Production of Fluorochemicals other than HCFC-22	HFCs, PFCs, SF ₆ , and NF ₃	3.34	2.93	3.75	-12.3%	+12.3%

^a Absolute lower and upper bounds were calculated using the corresponding lower and upper bounds in percentages.

20 **QA/QC and Verification**

21 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
 22 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the
 23 introduction of the IPPU chapter (see Annex 8 for more details). Under the GHGRP, EPA verifies annual facility-level
 24 reports through a multi-step process (e.g., including a combination of pre-and post-submittal electronic checks and
 25 manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate,
 26 complete, and consistent (EPA 2015).⁶⁷ Based on the results of the verification process, EPA follows up with
 27 facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of
 28 general and category-specific QC procedures, including: range checks, statistical checks, algorithm checks, and
 29 year-to-year checks of reported data and emissions.

⁶⁶ *Economic Impact Analysis for the Mandatory Reporting of Greenhouse Gas Emissions F-Gases: Subparts I, L, DD, QQ, SS*, U.S. Environmental Protection Agency, November 2010. Available online at: <https://www.regulations.gov/document/EPA-HQ-OAR-2009-0927-0179>.

⁶⁷ EPA (2015). Greenhouse Gas Reporting Program Report Verification. Available online at: https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 The GHGRP also requires source-specific quality control measures for the Fluorinated Gas Production category.
2 Under the GHGRP, fluorinated gas producers are required to (1) develop and periodically update process vent-
3 specific emission factors using either measurements or engineering calculations, depending on the nature of the
4 process (continuous vs. batch) and the magnitude of emissions from the vent, (2) take more measurements of vent
5 emissions where variability is high, (3) use methods for sampling, measuring volumetric flow rates, non-
6 fluorinated-GHG gas analysis, and measuring stack gas moisture that have been validated using a scientifically
7 sound validation protocol, (4) use a quality-assured analytical measurement technology capable of detecting the
8 analyte of interest at the concentration of interest and use a sampling and analytical procedure validated with the
9 analyte of interest at the concentration of interest, (5) periodically test the performance of destruction devices
10 used to control emissions, (6) account for any malfunctions in the process or destruction device, (6) account for
11 emissions from equipment leaks, (7) measure the quantities of residual gas that are vented from returned
12 containers (or develop an emission factor based on at least 30 measurements per gas and container size and type),
13 (8) calibrate mass measurement devices at the frequency recommended by the manufacturer using traceable
14 standards and suitable methods published by a consensus standards organization, (9) calibrate analytical
15 equipment used to determine the concentration of fluorinated GHGs, and (10) document all measurements and
16 calibrations.

17 Recalculations

18 This is a new category included for the current (i.e., 1990 to 2022) *Inventory*, thus, no recalculations were
19 performed.

20 Planned Improvements

21 EPA is still in the process of developing estimates for other fluorinated gas production facilities that do not report
22 their emissions under the GHGRP (e.g., because their uncontrolled emissions fall below the 25,000-MT CO₂ Eq.
23 threshold). Based on aggregated production estimates and the Tier 1 default emission factor in the 2019
24 Refinement, it is expected that emissions from these facilities will account for less than twenty percent of total U.S.
25 emissions from fluorinated gas production and transformation. In addition, EPA is continuing to seek datasets that
26 can be used to improve and/or QA/QC emissions estimates, particularly for the years 1990 to 2009. These datasets
27 may include, for example, real-time facility-specific estimates or global “top-down,” atmosphere-based emissions
28 estimates that could be used to establish an upper limit on emissions of certain compounds.

29 4.16 Carbon Dioxide Consumption (CRT 30 Source Category 2B10)

31 Carbon dioxide (CO₂) is used for a variety of commercial applications, including food processing, chemical
32 production, carbonated beverage production, and refrigeration, and is also used in petroleum production for
33 enhanced oil recovery (EOR). CO₂ used for EOR is injected underground to enable additional petroleum to be
34 produced. For the purposes of this analysis, CO₂ used in food and beverage applications is assumed to be emitted
35 to the atmosphere. A further discussion of CO₂ used in EOR is described in the Energy chapter in Box 3-6 titled
36 “Carbon Dioxide Transport, Injection, and Geological Storage” and is not included in this section.

37 Carbon dioxide is produced from naturally-occurring CO₂ reservoirs, as a byproduct from the energy and industrial
38 production processes (e.g., ammonia production, fossil fuel combustion, ethanol production), and as a byproduct
39 from the production of crude oil and natural gas, which contain naturally occurring CO₂ as a component.

1 In 2022, the amount of CO₂ produced and captured for commercial applications and subsequently emitted to the
 2 atmosphere was 5.0 MMT CO₂ Eq. (5,000 kt) (see Table 4-70 and Table 4-71). This is less than a 1 percent increase
 3 (10 kt) from 2021 levels and is an increase of approximately 240 percent (3,528 kt) since 1990.

4 **Table 4-70: CO₂ Emissions from CO₂ Consumption (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
CO ₂ Consumption	1.5	1.4	4.1	4.9	5.0	5.0	5.0

5 **Table 4-71: CO₂ Emissions from CO₂ Consumption (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
CO ₂ Consumption	1,472	1,375	4,130	4,870	4,970	4,990	5,000

6 Methodology and Time-Series Consistency

7 Carbon dioxide emission estimates for 1990 through 2022 utilize a country-specific method and were based on the
 8 quantity of CO₂ extracted and transferred for industrial applications (i.e., non-EOR end-uses). Some of the CO₂
 9 produced by these facilities is used for EOR, and some is used in other commercial applications (e.g., chemical
 10 manufacturing, food and beverage). The IPCC does not have specific methodological guidelines for CO₂
 11 consumption, but the country-specific methodology used is consistent with a Tier 3 approach since it relies on
 12 facility-specific information.

13 2010 through 2022

14 For 2010 through 2022, data from EPA’s GHGRP (Subpart PP) were aggregated from facility-level reports to
 15 develop a national-level estimate for use in the *Inventory* (EPA 2023). Facilities report CO₂ extracted or produced
 16 from natural reservoirs and industrial sites, and CO₂ captured from energy and industrial processes and transferred
 17 to various end-use applications to EPA’s GHGRP. This analysis includes only reported CO₂ transferred to food and
 18 beverage end-uses. EPA is continuing to analyze and assess integration of CO₂ transferred to other end-uses to
 19 enhance the completeness of estimates under this source category. Other end-uses include industrial applications,
 20 such as metal fabrication. EPA is analyzing the information reported to ensure that other end-use data excludes
 21 non-emissive applications and publication will not reveal CBI. Additionally, a small amount of CO₂ is used as a
 22 refrigerant; use and emissions from this application are reported under Section 4.24 Substitution of Ozone
 23 Depleting Substances (CRT Source Category 2F). Reporters subject to EPA’s GHGRP Subpart PP are also required to
 24 report the quantity of CO₂ that is imported and/or exported. Currently, these data are not publicly available
 25 through the GHGRP due to data confidentiality reasons and hence are excluded from this analysis.

26 Facilities subject to Subpart PP of EPA’s GHGRP are required to measure CO₂ extracted or produced. More details
 27 on the calculation and monitoring methods applicable to extraction and production facilities can be found under
 28 Subpart PP: Suppliers of Carbon Dioxide of the regulation, Part 98⁶⁸ The number of facilities that reported data to
 29 EPA’s GHGRP Subpart PP (Suppliers of Carbon Dioxide) for 2010 through 2022 is much higher (ranging from 44 to
 30 53) than the number of facilities included in the *Inventory* for the 1990 to 2009 time period prior to the availability
 31 of GHGRP data (4 facilities). The difference is largely due to the fact the 1990 to 2009 data includes only CO₂
 32 transferred to end-use applications from naturally occurring CO₂ reservoirs and excludes industrial sites.

33 1990 through 2009

34 For 1990 through 2009, data from EPA’s GHGRP are not available. For this time period, CO₂ production data from
 35 four naturally-occurring CO₂ reservoirs were used to estimate annual CO₂ emissions. These facilities were Jackson

⁶⁸ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

1 Dome in Mississippi, Bravo and West Bravo Domes in New Mexico, and McCallum Dome in Colorado. The facilities
 2 in Mississippi and New Mexico produced CO₂ for use in both EOR and in other commercial applications (e.g.,
 3 chemical manufacturing, food production). The fourth facility in Colorado (McCallum Dome) produced CO₂ for
 4 commercial applications only (New Mexico Bureau of Geology and Mineral Resources 2006).

5 Carbon dioxide production data and the percentage of production that was used for non-EOR applications for the
 6 Jackson Dome, Mississippi facility were obtained from Advanced Resources International (ARI 2006, 2007) for 1990
 7 to 2000, and from the Annual Reports of Denbury Resources (Denbury Resources 2002 through 2010) for 2001 to
 8 2009 (see Table 4-72). Denbury Resources reported the average CO₂ production in units of MMCF CO₂ per day for
 9 2001 through 2009 and reported the percentage of the total average annual production that was used for EOR.
 10 Production from 1990 to 1999 was set equal to 2000 production, due to lack of publicly available production data
 11 for 1990 through 1999. Carbon dioxide production data for the Bravo Dome and West Bravo Dome were obtained
 12 from ARI for 1990 through 2009 (ARI 1990 to 2010). Data for the West Bravo Dome facility were only available for
 13 2009. The percentage of total production that was used for non-EOR applications for the Bravo Dome and West
 14 Bravo Dome facilities for 1990 through 2009 were obtained from New Mexico Bureau of Geology and Mineral
 15 Resources (Broadhead 2003; New Mexico Bureau of Geology and Mineral Resources 2006). Production data for the
 16 McCallum Dome (Jackson County), Colorado facility were obtained from the Colorado Oil and Gas Conservation
 17 Commission (COGCC) for 1999 through 2009 (COGCC 2014). Production data for 1990 to 1998 and percentage of
 18 production used for EOR were assumed to be the same as for 1999, due to lack of publicly available data.

19 **Table 4-72: CO₂ Production (kt CO₂) and the Percent Used for Non-EOR Applications**

Year	Jackson Dome, MS CO ₂ Production (kt) (% Non-EOR)	Bravo Dome, NM CO ₂ Production (kt) (% Non-EOR)	West Bravo Dome, NM CO ₂ Production (kt) (% Non-EOR)	McCallum Dome, CO CO ₂ Production (kt) (% Non-EOR)	Total CO ₂ Production from Extraction and Capture Facilities (kt)	% Non- EOR ^a
1990	1,344 (100%)	63 (1%)	+	65 (100%)	NE	NE
2005	1,254 (27%)	58 (1%)	+	63 (100%)	NE	NE
2018	IE	IE	IE	IE	58,400 ^b	7%
2019	IE	IE	IE	IE	61,300 ^b	8%
2020	IE	IE	IE	IE	44,700 ^b	11%
2021	IE	IE	IE	IE	43,980 ^b	11%
2022	IE	IE	IE	IE	46,800 ^b	11%

+ Does not exceed 0.5 percent.

NE (Not Estimated)

IE (Included Elsewhere)

^a Includes only food and beverage applications.

^b For 2010 through 2022, the publicly available GHGRP data were aggregated at the national level based on GHGRP CBI criteria. The Dome-specific CO₂ production values are accounted for (i.e. included elsewhere) in the Total CO₂ Production from Extraction and Capture Facilities values starting in 2010 and are not able to be disaggregated.

20 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
 21 through 2022. The methodology for CO₂ consumption spliced activity data from two different sources: Industry
 22 data for 1990 through 2009 and GHGRP data starting in 2010. Consistent with the *2006 IPCC Guidelines*, the
 23 overlap technique was applied to compare the two data sets for years where there was overlap. The data sets
 24 were determined to be inconsistent; the GHGRP data include CO₂ from industrial sources while the industry data
 25 do not. No adjustments were made to the activity data for 1990 through 2009 because the *2006 IPCC Guidelines*
 26 indicate that it is not good practice to use the overlap technique when the data sets are inconsistent.

Uncertainty – TO BE UPDATED FOR FINAL REPORT

There is uncertainty associated with the data reported through EPA’s GHGRP. Specifically, there is uncertainty associated with the amount of CO₂ consumed for food and beverage applications, given the GHGRP does have provisions that Subpart PP reporters are not required to report to the GHGRP if their emissions fall below certain thresholds, in addition to the exclusion of the amount of CO₂ transferred to all other end-use categories. This latter category might include CO₂ quantities that are being used for non-EOR industrial applications such as firefighting. Second, uncertainty is associated with the exclusion of imports/exports data for CO₂ suppliers. Currently these data are not publicly available through EPA’s GHGRP and hence are excluded from this analysis. EPA verifies annual facility-level reports through a multi-step process (e.g., combination of electronic checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent. Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred.⁶⁹ Given the lack of specific uncertainty ranges available on the data used, EPA uses an uncertainty range of ±5 percent for CO₂ consumed for food and beverage applications. The uncertainty range is derived from the default range for solvent use in Section 5.5 of Chapter 3 of the *2006 IPCC Guidelines*. These values are representative of CO₂ used in food and beverage based on expert judgment (RTI 2023).

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-73. Carbon dioxide consumption CO₂ emissions for 2022 were estimated to be between 4.7 and 5.2 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 5 percent below to 5 percent above the emission estimate of 5.0 MMT CO₂ Eq.

Table 4-73: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from CO₂ Consumption (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.)			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
CO ₂ Consumption	CO ₂	5.0	4.7	5.2	-5%	+5%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the introduction of the IPPU chapter (see Annex 8 for more details). More details on the greenhouse gas calculation, monitoring and QA/QC methods applicable to CO₂ Consumption can be found under Subpart PP (Suppliers of Carbon Dioxide) of the regulation (40 CFR Part 98).⁷⁰ EPA verifies annual facility-level GHGRP reports through a multi-step process (e.g., combination of electronic checks and manual reviews) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).⁷¹ Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including range checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.

⁶⁹ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

⁷⁰ See http://www.ecfr.gov/cgi-bin/text-idx?tpl=/ecfrbrowse/Title40/40cfr98_main_02.tpl.

⁷¹ See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 Recalculations Discussion

2 No recalculations were performed for the 1990 through 2021 portion of the time series.

3 Planned Improvements

4 EPA will continue to evaluate the potential to include additional GHGRP data on other emissive end-uses to
5 improve the accuracy and completeness of estimates for this source category. Particular attention will be made to
6 ensuring time-series consistency of the emissions estimates presented in future *Inventory* reports, consistent with
7 IPCC and UNFCCC guidelines. This is required as the facility-level reporting data from EPA's GHGRP, with the
8 program's initial requirements for reporting of emissions in calendar year 2010, are not available for all inventory
9 years (i.e., 1990 through 2009) as required for this *Inventory*. In implementing improvements and integration of
10 data from EPA's GHGRP, EPA will rely on the latest guidance from the IPCC on the use of facility-level data in
11 national inventories.⁷²

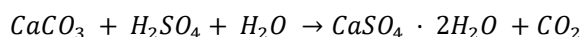
12 These improvements are still in process and will be incorporated into future *Inventory* reports. These are near-to
13 medium-term improvements.

14 4.17 Phosphoric Acid Production (CRT 15 Source Category 2B10)

16 Phosphoric acid (H₃PO₄) is a basic raw material used in the production of phosphate-based fertilizers. Phosphoric
17 acid production from natural phosphate rock is a source of carbon dioxide (CO₂) emissions, due to the chemical
18 reaction of the inorganic carbon (calcium carbonate) component of the phosphate rock.

19 Phosphate rock is mined in Florida and North Carolina, which account for more than 75 percent of total domestic
20 output, and in Idaho and Utah (USGS 2023). It is used primarily as a raw material for wet-process phosphoric acid
21 production. The composition of natural phosphate rock varies, depending on the location where it is mined.
22 Natural phosphate rock mined in the United States generally contains inorganic carbon in the form of calcium
23 carbonate (limestone) and may also contain organic carbon.

24 The phosphoric acid production process involves chemical reaction of the calcium phosphate (Ca₃(PO₄)₂)
25 component of the phosphate rock with sulfuric acid (H₂SO₄) and recirculated phosphoric acid (H₃PO₄) (EFMA 2000).
26 Phosphate rock also contains naturally occurring limestone (CaCO₃), ranging from 0.2 to 4.5 percent (as CO₂), with
27 domestic phosphate rock from Florida containing 3.1 percent limestone (as CO₂) (EFMA 2000). The generation of
28 CO₂ from limestone in the phosphate rock is from the associated limestone-sulfuric acid reaction, as shown below:



30 Total U.S. phosphate rock production in 2022 was an estimated 21 million metric tons (USGS 2023). Total imports
31 of phosphate rock to the United States in 2022 were 2.4 million metric tons (USGS 2023). Between 2018 and 2021,
32 most of the imported phosphate rock (95 percent) came from Peru, with 5 percent from Morocco (USGS 2023). All
33 phosphate rock mining companies in the United States are vertically integrated with fertilizer plants that produce
34 phosphoric acid located near the mines. The phosphoric acid production facilities that use imported phosphate
35 rock are located in Louisiana.

⁷² See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

1 Between 1990 and 2022, domestic phosphate rock production decreased by approximately 58 percent. Total CO₂
 2 emissions from phosphoric acid production were 0.8 MMT CO₂ Eq. (840 kt CO₂) in 2022 (see Table 4-74 and Table
 3 4-75). Domestic consumption of phosphate rock in 2022 was estimated to have decreased 3.9 percent relative to
 4 2021 levels. The COVID-19 pandemic did not impact the domestic phosphate rock market as both the fertilizer
 5 industry and related agricultural businesses were considered essential industries and were unaffected by
 6 pandemic “stay-at-home” orders issued in March 2020 (USGS 2021a).

7 **Table 4-74: CO₂ Emissions from Phosphoric Acid Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Phosphoric Acid Production	1.5	1.3	0.9	0.9	0.9	0.9	0.8

8 **Table 4-75: CO₂ Emissions from Phosphoric Acid Production (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Phosphoric Acid Production	1,529	1,342	937	909	901	874	840

9 Methodology and Time-Series Consistency

10 The United States uses a country-specific methodology consistent with and comparable to an IPCC Tier 1 approach
 11 to calculate emissions from production of phosphoric acid from phosphate rock based on the stoichiometry of the
 12 process reaction shown above. The *2006 IPCC Guidelines* do not provide a method for estimating process
 13 emissions (CO₂) from phosphoric acid production. Carbon dioxide emissions from production of phosphoric acid
 14 from phosphate rock are estimated by multiplying the average amount of inorganic carbon (expressed as CO₂)
 15 contained in the natural phosphate rock as calcium carbonate by the amount of phosphate rock that is used
 16 annually to produce phosphoric acid, accounting for domestic production and net imports for consumption. The
 17 estimation methodology is as follows:

18 Equation 4-10: CO₂ Emissions from Phosphoric Acid Production

$$19 E_{pa} = C_{pr} \times Q_{pr}$$

20 where,

E_{pa} = CO₂ emissions from phosphoric acid production, metric tons

C_{pr} = Average amount of carbon (expressed as CO₂) in natural phosphate rock, metric ton
 CO₂/ metric ton phosphate rock

Q_{pr} = Quantity of phosphate rock used to produce phosphoric acid

21 The CO₂ emissions calculation methodology assumes that all of the inorganic carbon (calcium carbonate) content
 22 of the phosphate rock reacts to produce CO₂ in the phosphoric acid production process and is emitted with the
 23 stack gas. The methodology also assumes that none of the organic carbon content of the phosphate rock is
 24 converted to CO₂ and that all of the organic carbon content remains in the phosphoric acid product.

25 From 1993 to 2004, the U.S. Geological Survey (USGS) *Mineral Yearbook: Phosphate Rock* disaggregated phosphate
 26 rock mined annually in Florida and North Carolina from phosphate rock mined annually in Idaho and Utah, and
 27 reported the annual amounts of phosphate rock exported and imported for consumption (see Table 4-76). For the
 28 years 1990 through 1992, and 2005 through 2022, only nationally aggregated mining data was reported by USGS.
 29 For the years 1990, 1991, and 1992, the breakdown of phosphate rock mined in Florida and North Carolina and the
 30 amount mined in Idaho and Utah are approximated using data reported by USGS for the average share of U.S.
 31 production in those states from 1993 to 2004. For the years 2005 through 2022, the same approximation method
 32 is used, but the share of U.S. production was assumed to be consistent with the ratio of production capacity in
 33 those states, which were obtained from the USGS commodity specialist for phosphate rock (USGS 2012; USGS
 34 2021b). For 1990 through 2022, data on U.S. domestic consumption of phosphate rock, consisting of domestic

1 reported sales and use of phosphate rock, exports of phosphate rock (primarily from Florida and North Carolina),
 2 and imports of phosphate rock for consumption, were obtained from USGS *Minerals Yearbook: Phosphate Rock*
 3 (USGS 1994 through 2015b) and from USGS *Minerals Commodity Summaries: Phosphate Rock* (USGS 2016 through
 4 2021a, 2022). From 2004 through 2022, the USGS reported no exports of phosphate rock from U.S. producers
 5 (USGS 2022).

6 The carbonate content of phosphate rock varies depending upon where the material is mined. Composition data
 7 for domestically mined and imported phosphate rock were provided by the Florida Institute of Phosphate
 8 Research, now known as the Florida Industrial and Phosphate Research Institute (FIPR 2003a). Phosphate rock
 9 mined in Florida contains approximately 1 percent inorganic C, and phosphate rock imported from Morocco
 10 contains approximately 1.46 percent inorganic C. Calcined phosphate rock mined in North Carolina and Idaho
 11 contains approximately 0.41 percent and 0.27 percent inorganic C, respectively (see Table 4-76). Similar to the
 12 phosphate rock mined in Morocco, phosphate rock mined in Peru contains approximately 5 percent CO₂ (Golder
 13 Associates and M3 Engineering 2016).

14 Carbonate content data for phosphate rock mined in Florida are used to calculate the CO₂ emissions from
 15 consumption of phosphate rock mined in Florida and North Carolina (more than 75 percent of domestic
 16 production), and carbonate content data for phosphate rock mined in Morocco and Peru are used to calculate CO₂
 17 emissions from consumption of imported phosphate rock. The CO₂ emissions calculation assumes that all of the
 18 domestic production of phosphate rock is used in uncalcined form. As of 2006, the USGS noted that one phosphate
 19 rock producer in Idaho produces calcined phosphate rock; however, no production data were available for this
 20 single producer (USGS 2006). The USGS confirmed that no significant quantity of domestic production of
 21 phosphate rock is in the calcined form (USGS 2012).

22 **Table 4-76: Phosphate Rock Domestic Consumption, Exports, and Imports (kt)**

Location/Year	1990	2005	2018	2019	2020	2021	2022
U.S. Domestic Consumption ^a	49,800	35,200	23,300	23,400	22,600	21,900	21,000
<i>FL and NC</i>	42,494	28,160	18,170	18,250	17,630	17,080	16,380
<i>ID and UT</i>	7,306	7,040	5,130	5,150	4,970	4,820	4,620
Exports—FL and NC	6,240	0	0	0	0	0	0
Imports	451	2,630	2,770	2,140	2,520	2,460	2,400
Total U.S. Consumption	44,011	37,830	26,070	25,540	25,120	24,360	23,400

^a U.S. domestic consumption values are based on reported phosphate rock sold or used by producers.

Note: Totals may not sum due to independent rounding.

23 **Table 4-77: Chemical Composition of Phosphate Rock (Percent by Weight)**

Composition	Central Florida	North Florida	North		Morocco	Peru
			Carolina (calcined)	Idaho (calcined)		
Total Carbon (as C)	1.60	1.76	0.76	0.60	1.56	NA
Inorganic Carbon (as C)	1.00	0.93	0.41	0.27	1.46	NA
Organic Carbon (as C)	0.60	0.83	0.35	0.00	0.10	NA
Inorganic Carbon (as CO ₂)	3.67	3.43	1.50	1.00	5.00	5.00

NA (Not Available)

Sources: FIPR (2003a), Golder Associates and M3 Engineering (2016)

24 Methodological approaches were applied to the entire time series to ensure consistency in emissions estimates
 25 from 1990 through 2022.

26 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

27 Phosphate rock production data used in the emission calculations were developed by the USGS through monthly
 28 and semiannual voluntary surveys of the active phosphate rock mines during 2021. Prior to 2006, USGS provided

the data disaggregated regionally; however, beginning in 2006, only total U.S. phosphate rock production was reported. Regional production for 2021 was estimated based on regional production data from 2017 to 2020 and multiplied by regionally-specific emission factors. There is uncertainty associated with the degree to which the estimated 2021 regional production data represents actual production in those regions. Total U.S. phosphate rock production data are not considered to be a significant source of uncertainty because all the domestic phosphate rock producers report their annual production to the USGS. Data for exports of phosphate rock used in the emission calculations are reported to the USGS by phosphate rock producers and are not considered to be a significant source of uncertainty. Data for imports for consumption are based on international trade data collected by the U.S. Census Bureau. These U.S. government economic data are not considered to be a significant source of uncertainty. Based on expert judgement of the USGS, EPA assigned an uncertainty range of ± 5 percent to the percentage of phosphate rock produced from Florida and North Carolina, and ± 5 percent to phosphoric acid production and imports, based on expert judgment (USGS 2012).

An additional source of uncertainty in the calculation of CO₂ emissions from phosphoric acid production is the carbonate composition of phosphate rock, as the composition of phosphate rock varies depending upon where the material is mined and may also vary over time. The *Inventory* relies on one study (FIPR 2003a) of chemical composition of the phosphate rock; limited data are available beyond this study. Another source of uncertainty is the disposition of the organic carbon content of the phosphate rock. A representative of FIPR indicated that in the phosphoric acid production process, the organic carbon content of the mined phosphate rock generally remains in the phosphoric acid product, which is what produces the color of the phosphoric acid product (FIPR 2003b). Organic carbon is therefore not included in the calculation of CO₂ emissions from phosphoric acid production.

A third source of uncertainty is the assumption that all domestically-produced phosphate rock is used in phosphoric acid production and used without first being calcined. Calcination of the phosphate rock would result in conversion of some of the organic carbon in the phosphate rock into CO₂; however, according to air permit information available to the public, at least one facility has calcining units permitted for operation (NCDENR 2013).

Finally, USGS indicated that in 2021 less than 5 percent of domestically-produced phosphate rock was used to manufacture elemental phosphorus and other phosphorus-based chemicals, rather than phosphoric acid (USGS 2022). According to USGS, there is only one domestic producer of elemental phosphorus, in Idaho, and no data were available concerning the annual production of this single producer. Elemental phosphorus is produced by reducing phosphate rock with coal coke, and it is therefore assumed that 100 percent of the carbonate content of the phosphate rock will be converted to CO₂ in the elemental phosphorus production process. The calculation for CO₂ emissions assumes that phosphate rock consumption, for purposes other than phosphoric acid production, results in CO₂ emissions from 100 percent of the inorganic carbon content in phosphate rock, but none from the organic carbon content.

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-78. 2021 phosphoric acid production CO₂ emissions were estimated to be between 0.8 and 1.1 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 18 percent below and 20 percent above the emission estimate of 0.9 MMT CO₂ Eq.

Table 4-78: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Phosphoric Acid Production (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Phosphoric Acid Production	CO ₂	0.9	0.8	1.1	-18%	+20%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

1 QA/QC and Verification

2 For more information on the general QA/QC process applied to this source category, consistent with the U.S.
3 Inventory QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the
4 introduction of the IPPU chapter (see Annex 8 for more details).

5 Recalculations Discussion

6 Recalculations were performed for 2021 to reflect updated USGS data on the total U.S. production of phosphate
7 rock. This update resulted in a decrease of 35 kt CO₂ in 2021.

8 Planned Improvements

9 EPA continues to evaluate potential improvements to the *Inventory* estimates for this source category, which
10 include direct integration of EPA's GHGRP data for 2010 through 2022 along with assessing applicability of
11 reported GHGRP data to update the inorganic carbon content of phosphate rock for prior years to ensure time-
12 series consistency. Specifically, EPA would need to assess that averaged inorganic carbon content data (by region
13 or other approaches) meets GHGRP confidential business information (CBI) screening criteria. EPA would then
14 need to assess the applicability of GHGRP data for the averaged inorganic carbon content (by region or other
15 approaches) from 2010 through 2022, along with other information to inform estimates in prior years in the
16 required time series (1990 through 2009) based on the sources of phosphate rock used in production of
17 phosphoric acid over time. In implementing improvements and integration of data from EPA's GHGRP, the latest
18 guidance from the IPCC on the use of facility-level data in national inventories will be relied upon.⁷³ These long-
19 term planned improvements are still in development by EPA and have not been implemented into the current
20 *Inventory* report.

21 4.18 Iron and Steel Production (CRT Source 22 Category 2C1) and Metallurgical Coke 23 Production

24 Iron and steel production is a multi-step process that generates process-related emissions of carbon dioxide (CO₂)
25 and methane (CH₄) as raw materials are refined into iron and then transformed into crude steel. Emissions from
26 conventional fuels (e.g., natural gas, fuel oil) consumed for energy purposes during the production of iron and steel
27 are accounted for in the Energy chapter.

28 Iron and steel production includes seven distinct production processes: metallurgical coke production, sinter
29 production, direct reduced iron (DRI) production, pellet production, pig iron.⁷⁴ production, electric arc furnace
30 (EAF) steel production, and basic oxygen furnace (BOF) steel production. The number of production processes at a

⁷³ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

⁷⁴ Pig iron is the common industry term to describe what should technically be called crude iron. Pig iron is a subset of crude iron that has lost popularity over time as industry trends have shifted. Throughout this report, pig iron will be used interchangeably with crude iron, but it should be noted that in other data sets or reports pig iron and crude iron may not be used interchangeably and may provide different values.

1 particular plant is dependent upon the specific plant configuration. Most process CO₂ generated from the iron and
2 steel industry is a result of the production of crude iron.

3 In addition to the production processes mentioned above, CO₂ is also generated at iron and steel mills through the
4 consumption of process byproducts (e.g., blast furnace gas, coke oven gas) used for various purposes including
5 heating, annealing, and electricity generation. Process byproducts sold off-site for use as synthetic natural gas are
6 also accounted for in these calculations. In general, CO₂ emissions are generated in these production processes
7 through the reduction and consumption of various carbon-containing inputs (e.g., ore, scrap, flux, coke
8 byproducts). Fugitive CH₄ emissions can also be generated from these processes, as well as from sinter, direct iron,
9 and pellet production.

10 In 2022, approximately eleven integrated iron and steel steelmaking facilities utilized BOFs to refine and produce
11 steel from iron, and raw steel was produced at 101 facilities across the United States. As of 2020, approximately 29
12 percent of steel production was attributed to BOFs and 71 percent to EAFs (AISI 2020). The trend in the United
13 States for integrated facilities has been a shift towards fewer BOFs and more EAFs. EAFs use scrap steel as their
14 main input and use significantly less energy than BOFs. There are also 14 cokemaking facilities, of which 3 facilities
15 are co-located with integrated iron and steel facilities (ACCCI 2021). In the United States, seven states account for
16 roughly 61 percent of total raw steel production: Indiana, Alabama, Tennessee, Kentucky, Mississippi, Arkansas,
17 and Ohio (AISI 2023).

18 Total annual production of crude steel in the United States was fairly constant between 2000 and 2008 and ranged
19 from a low of 99,320,000 tons to a high of 109,880,000 tons (2001 and 2004, respectively). Due to the decrease in
20 demand caused by the global economic downturn (particularly from the automotive industry), crude steel
21 production in the United States sharply decreased to 65,459,000 tons in 2009. Crude steel production was fairly
22 constant from 2011 through 2014, and after a dip in production from 2014 to 2015, crude steel production steadily
23 increased. Crude steel production dipped again in 2020 due to the COVID-19 pandemic and returned to pre-
24 pandemic levels in 2021. Production declined by approximately 6 percent in 2022 (AISI 2023). This decline may be
25 attributable to projections for decreased global end-use consumption due to multiple factors including the conflict
26 in Ukraine, continuing coronavirus disease 2019 (COVID-19) mitigation measures in China, rising energy costs and
27 interest rates, and global inflation (USGS 2023). The United States was the fourth largest producer of raw steel in
28 the world, behind China, India, and Japan, accounting for approximately 4.3 percent of world production in 2022
29 (AISI 2004 through 2023).

30 The majority of CO₂ emissions from the iron and steel production process come from the use of metallurgical coke
31 in the production of pig iron and from the consumption of other process byproducts, with lesser amounts emitted
32 from the use of carbon-containing flux and from the removal of carbon from pig iron used to produce steel.

33 According to the *2006 IPCC Guidelines*, the production of metallurgical coke from coking coal is considered to be an
34 energy use of fossil fuel, and the use of coke in iron and steel production is considered to be an industrial process
35 source. The *2006 IPCC Guidelines* suggest that emissions from the production of metallurgical coke should be
36 reported separately in the Energy sector, while emissions from coke consumption in iron and steel production
37 should be reported in the Industrial Processes and Product Use sector. The approaches and emission estimates for
38 both metallurgical coke production and iron and steel production, however, are presented here because much of
39 the relevant activity data is used to estimate emissions from both metallurgical coke production and iron and steel
40 production. For example, some byproducts (e.g., coke oven gas) of the metallurgical coke production process are
41 consumed during iron and steel production, and some byproducts of the iron and steel production process (e.g.,
42 blast furnace gas) are consumed during metallurgical coke production. Emissions associated with the consumption
43 of these byproducts are attributed at the point of consumption. Emissions associated with the use of conventional
44 fuels (e.g., natural gas, fuel oil) for electricity generation, heating and annealing, or other miscellaneous purposes
45 downstream of the iron and steelmaking furnaces are reported in the Energy chapter.

46 Metallurgical Coke Production

47 Emissions of CO₂ from metallurgical coke production in 2022 were 3.0 MMT CO₂ Eq. (2,954 kt CO₂) (see Table 4-79
48 and Table 4-80). Emissions decreased by 8 percent from 2021 to 2022 and have decreased by 47 percent since

1 1990. Coke production in 2022 was about 9 percent lower than in 2021 and 59 percent below 1990 (EIA 2023, AISI
2 2023).

3 Significant activity data for 2020 through 2022 were not available in time for publication of this report due to
4 industry consolidation that impacts the publication of data without revealing confidential business information.
5 Activity data for these years were estimated using 2019 values adjusted based on GHGRP emissions data, as
6 described in the Methodology and Time-Series Consistency section below.

7 **Table 4-79: CO₂ Emissions from Metallurgical Coke Production (MMT CO₂ Eq.)**

Gas	1990	2005	2018	2019	2020	2021	2022
CO ₂	5.6	3.9	1.3	3.0	2.3	3.2	3.0

8 **Table 4-80: CO₂ Emissions from Metallurgical Coke Production (kt CO₂)**

Gas	1990	2005	2018	2019	2020	2021	2022
CO ₂	5,608	3,921	1,282	3,006	2,325	3,224	2,954

9 Iron and Steel Production

10 Emissions of CO₂ and CH₄ from iron and steel production in 2022 were 37.7 MMT CO₂ Eq. (37,713 kt) and 0.0077
11 MMT CO₂ Eq. (0.3 kt CH₄), respectively (see Table 4-81 through Table 4-84). Emissions from iron and steel
12 production decreased by 2 percent from 2021 to 2022 and have decreased by 62 percent since 1990, due to
13 restructuring of the industry, technological improvements, and increased scrap steel utilization. Carbon dioxide
14 emission estimates include emissions from the consumption of carbonaceous materials in the blast furnace, EAF,
15 and BOF, as well as blast furnace gas and coke oven gas consumption for other activities at the steel mill.

16 Significant activity data for 2020 through 2022 were not available in time for publication of this report due to
17 industry consolidation that impacts the publication of data without revealing confidential business information.
18 Activity data for these years were estimated using 2019 values adjusted based on GHGRP emissions data, as
19 described in the Methodology and Time-Series Consistency section below.

20 In 2022, domestic production of pig iron decreased by 11 percent from 2020 levels. Overall, domestic pig iron
21 production has declined since the 1990s; pig iron production in 2022 was 59 percent lower than in 2000 and 60
22 percent below 1990. Carbon dioxide emissions from iron production have decreased by 81 percent (37.0 MMT CO₂
23 Eq.) since 1990. Carbon dioxide emissions from steel production have decreased by 16 percent (1.3 MMT CO₂ Eq.)
24 since 1990, while overall CO₂ emissions from iron and steel production have declined by 62 percent (61.4 MMT
25 CO₂ Eq.) from 1990 to 2022.

26 **Table 4-81: CO₂ Emissions from Iron and Steel Production (MMT CO₂ Eq.)**

Source/Activity Data	1990	2005	2018	2019	2020	2021	2022
Sinter Production	2.4	1.7	0.9	0.9	0.7	0.8	0.8
Iron Production	45.7	17.7	9.6	9.4	8.4	9.0	8.7
Pellet Production	1.8	1.5	0.9	0.9	0.8	0.8	0.8
Steel Production	8.0	9.4	6.0	5.8	5.7	5.8	6.7
Other Activities ^a	41.2	35.9	24.1	23.2	19.8	22.1	20.8
Total	99.1	66.2	41.6	40.1	35.4	38.6	37.7

^a Includes emissions from blast furnace gas and coke oven gas combustion for activities at the steel mill other than consumption in blast furnace, EAFs, or BOFs.

Note: Totals may not sum due to independent rounding.

1 **Table 4-82: CO₂ Emissions from Iron and Steel Production (kt)**

Source/Activity Data	1990	2005	2018	2019	2020	2021	2022
Sinter Production	2,448	1,663	937	876	749	836	787
Iron Production	45,709	17,666	9,589	9,365	8,420	9,038	8,673
Pellet Production	1,817	1,503	924	878	751	838	789
Steel Production	7,964	9,395	5,982	5,812	5,657	5,816	6,655
Other Activities ^a	41,194	35,934	24,149	23,158	19,820	22,119	20,814
Total	99,132	66,161	41,581	40,089	35,398	38,648	37,718

^a Includes emissions from blast furnace gas and coke oven gas combustion for activities at the steel mill other than consumption in blast furnace, EAFs, or BOFs.

Note: Totals may not sum due to independent rounding.

2 **Table 4-83: CH₄ Emissions from Iron and Steel Production (MMT CO₂ Eq.)**

Source/Activity Data	1990	2005	2018	2019	2020	2021	2022
Sinter Production	+	+	+	+	+	+	+

+ Does not exceed 0.05 MMT CO₂ Eq.

3 **Table 4-84: CH₄ Emissions from Iron and Steel Production (kt)**

Source/Activity Data	1990	2005	2018	2019	2020	2021	2022
Sinter Production	0.9	0.6	+	+	+	+	+

+ Does not exceed 0.5 kt.

4 Methodology and Time-Series Consistency

5 Emission estimates for metallurgical coke, EAF steel production, and BOF steel production presented in this
6 chapter utilize a country-specific approach based on Tier 2 methodologies provided by the *2006 IPCC Guidelines*, in
7 accordance with the IPCC methodological decision tree and available data. These Tier 2 methodologies call for a
8 mass balance accounting of the carbonaceous inputs and outputs during the iron and steel production process and
9 the metallurgical coke production process. Estimates for pig iron production apply Tier 2 methods consistent with
10 the *2006 IPCC Guidelines*, in accordance with the IPCC methodological decision tree and available data. Tier 1
11 methods are used for certain iron and steel production processes (i.e., sinter production, pellet production and DRI
12 production) for which available data are insufficient to apply a Tier 2 method (e.g., country-specific carbon
13 contents of inputs and outputs are not known). The majority of emissions are captured with higher tier methods,
14 as sinter production, pellet production, and DRI production only account for roughly 8 percent of total iron and
15 steel production emissions.

16 The Tier 2 methodology equation is as follows:

17 **Equation 4-11: CO₂ Emissions from Coke, Pig Iron, EAF Steel, and BOF Steel Production, based**
18 **on 2006 IPCC Guidelines Tier 2 Methodologies**

$$19 \quad E_{CO_2} = \left[\sum_a (Q_a \times C_a) - \sum_b (Q_b \times C_b) \right] \times \frac{44}{12}$$

20 where,

21 E_{CO_2} = Emissions from coke, pig iron, EAF steel, or BOF steel production, metric tons

22 a = Input material a

23 b = Output material b

- 1 Q_a = Quantity of input material a , metric tons
 2 C_a = Carbon content of input material a , metric tons C/metric ton material
 3 Q_b = Quantity of output material b , metric tons
 4 C_b = Carbon content of output material b , metric tons C/metric ton material
 5 $44/12$ = Stoichiometric ratio of CO₂ to C

6 The Tier 1 methodology equations are as follows:

7 **Equation 4-12: 2006 IPCC Guidelines Tier 1: Emissions from Sinter, Direct Reduced Iron, and**
 8 **Pellet Production (Equations 4.6, 4.7, and 4.8)**

9
$$E_{s,p} = Q_s \times EF_{s,p}$$

 10
$$E_{d,CO_2} = Q_d \times EF_{d,CO_2}$$

 11
$$E_{p,CO_2} = Q_p \times EF_{p,CO_2}$$

12 where,

- 13 $E_{s,p}$ = Emissions from sinter production process for pollutant p (CO₂ or CH₄), metric ton
 14 Q_s = Quantity of sinter produced, metric tons
 15 $EF_{s,p}$ = Emission factor for pollutant p (CO₂ or CH₄), metric ton p /metric ton sinter
 16 E_{d,CO_2} = Emissions from DRI production process for CO₂, metric ton
 17 Q_d = Quantity of DRI produced, metric tons
 18 EF_{d,CO_2} = Emission factor for CO₂, metric ton CO₂/metric ton DRI
 19 E_{p,CO_2} = Emissions from pellet production process for CO₂, metric ton
 20 Q_p = Quantity of pellets produced, metric tons
 21 EF_{p,CO_2} = Emission factor for CO₂, metric ton CO₂/metric ton pellets produced

22 A significant number of activity data that serve as inputs to emissions calculations were unavailable for 2020
 23 through 2022 at the time of publication and were estimated using 2019 values. To estimate annual emissions for
 24 these years, the EPA used process emissions data from the EPA's Greenhouse Gas Reporting Program (GHGRP)
 25 subpart Q for the iron and steel sector to adjust the estimated values for 2020 through 2022. GHGRP process
 26 emissions data decreased by approximately 14 percent from 2019 to 2020, increased by approximately 12 percent
 27 from 2020 to 2021, and decreased by approximately 6 percent from 2021 to 2022 (EPA 2023). These percentage
 28 changes were applied to 2019 activity data values to produce estimates for 2020 through 2022.

29 **Metallurgical Coke Production**

30 Coking coal is used to manufacture metallurgical coke which is used primarily as a reducing agent in the production
 31 of iron and steel but is also used in the production of other metals including zinc and lead (see Zinc Production and
 32 Lead Production sections of this chapter). Emissions associated with producing metallurgical coke from coking coal
 33 are estimated and reported separately from emissions that result from the iron and steel production process. To
 34 estimate emissions from metallurgical coke production, a Tier 2 method provided by the 2006 IPCC Guidelines was
 35 utilized. The amount of carbon contained in materials produced during the metallurgical coke production process
 36 (i.e., coke, coke breeze and coke oven gas) is deducted from the amount of carbon contained in materials
 37 consumed during the metallurgical coke production process (i.e., natural gas, blast furnace gas, and coking coal).
 38 For calculations, activity data for these inputs, including natural gas, blast furnace gas, and coking coke consumed
 39 for metallurgical coke production, are in units consistent with the carbon content values. Light oil, which is

1 produced during the metallurgical coke production process, is excluded from the deductions due to data
 2 limitations. The amount of carbon contained in these materials is calculated by multiplying the material-specific
 3 carbon content by the amount of material consumed or produced (see Table 4-85). The amount of coal tar
 4 produced was approximated using a production factor of 0.03 tons of coal tar per ton of coking coal consumed.
 5 The amount of coke breeze produced was approximated using a production factor of 0.075 tons of coke breeze per
 6 ton of coking coal consumed (Steiner 2008; DOE 2000). Data on the consumption of carbonaceous materials (other
 7 than coking coal) as well as coke oven gas production were available for integrated steel mills only (i.e., steel mills
 8 with co-located coke plants); therefore, carbonaceous material (other than coking coal) consumption and coke
 9 oven gas production were excluded from emission estimates for merchant coke plants. Carbon contained in coke
 10 oven gas used for coke-oven underfiring was not included in the deductions to avoid double-counting.

11 **Table 4-85: Material Carbon Contents for Metallurgical Coke Production**

Material	kg C/kg
Coal Tar ^a	0.62
Coke ^a	0.83
Coke Breeze ^a	0.83
Coking Coal ^b	0.75
Material	kg C/GJ
Coke Oven Gas ^c	12.1
Blast Furnace Gas ^c	70.8

^a Source: IPCC (2006), Vol. 3 Chapter 4, Table 4.3

^b Source: EIA (2017b)

^c Source: IPCC (2006), Vol. 2 Chapter 1, Table 1.3

12 Although the *2006 IPCC Guidelines* provide a Tier 1 CH₄ emission factor for metallurgical coke production (i.e., 0.1 g
 13 CH₄ per metric ton of coke production), it is not appropriate to use because CO₂ emissions were estimated using
 14 the Tier 2 mass balance methodology. The mass balance methodology makes a basic assumption that all carbon
 15 that enters the metallurgical coke production process either exits the process as part of a carbon-containing
 16 output or as CO₂ emissions. This is consistent with a preliminary assessment of aggregated facility-level
 17 greenhouse gas CH₄ emissions reported by coke production facilities under EPA's GHGRP. The assessment indicates
 18 that CH₄ emissions from coke production are insignificant and below 500 kt or 0.05 percent of total national
 19 emissions. Pending resources and significance, EPA continues to assess the possibility of including these emissions
 20 in future Inventories to enhance completeness but has not incorporated these emissions into this report.

21 Data relating to the mass of coking coal consumed at metallurgical coke plants and the mass of metallurgical coke
 22 produced at coke plants were taken from the Energy Information Administration (EIA) *Quarterly Coal Report:
 23 October through December* (EIA 1998 through 2019) and *EIA Quarterly Coal Report: January through March* (EIA
 24 2021 through 2023) (see Table 4-86). Data on the volume of natural gas consumption, blast furnace gas
 25 consumption, and coke oven gas production for metallurgical coke production at integrated steel mills were
 26 obtained from the American Iron and Steel Institute (AISI) *Annual Statistical Report* (AISI 2004 through 2023) and
 27 through personal communications with AISI (Steiner 2008) (see Table 4-87). These data from the *AISI Annual
 28 Statistical Report* were withheld for 2020 through 2022, so the 2019 values were used as estimated data for the
 29 missing 2020 through 2022 values and adjusted using GHGRP emissions data, as described earlier in this
 30 Methodology and Time-Series Consistency section.

31 The factor for the quantity of coal tar produced per ton of coking coal consumed was provided by AISI (Steiner
 32 2008). The factor for the quantity of coke breeze produced per ton of coking coal consumed was obtained through
 33 Table 2-1 of the report *Energy and Environmental Profile of the U.S. Iron and Steel Industry* (DOE 2000). Data on
 34 natural gas consumption and coke oven gas production at merchant coke plants were not available and were
 35 excluded from the emission estimate. Carbon contents for metallurgical coke, coal tar, coke oven gas, and blast
 36 furnace gas were provided by the *2006 IPCC Guidelines*. The carbon content for coke breeze was assumed to equal
 37 the carbon content of coke. Carbon contents for coking coal was from EIA.

1 **Table 4-86: Production and Consumption Data for the Calculation of CO₂ Emissions from**
 2 **Metallurgical Coke Production (Thousand Metric Tons)**

Source/Activity Data	1990	2005	2018	2019	2020	2021	2022
Metallurgical Coke Production							
Coking Coal Consumption at Coke Plants	35,269	21,259	16,635	16,261	13,076	15,957	14,523
Coke Production at Coke Plants	25,054	15,167	12,525	11,676	9,392	11,381	10,337
Coke Breeze Production	2,645	1,594	1,248	1,220	981	1,197	1,089
Coal Tar Production	1,058	638	499	488	392	479	436

3 **Table 4-87: Production and Consumption Data for the Calculation of CO₂ Emissions from**
 4 **Metallurgical Coke Production (Million ft³)**

Source/Activity Data	1990	2005	2018	2019	2020	2021	2022
Metallurgical Coke Production							
Coke Oven Gas Production	250,767	114,213	80,750	77,692	66,492	74,206	69,829
Natural Gas Consumption	599	2,996	2,275	2,189	1,873	2,091	1,967
Blast Furnace Gas Consumption	24,602	4,460	4,022	3,914	3,350	3,738	3,518

5 Iron and Steel Production

6 To estimate emissions from pig iron production in the blast furnace, the amount of carbon contained in the
 7 produced pig iron and blast furnace gas were deducted from the amount of carbon contained in inputs (i.e.,
 8 metallurgical coke, sinter, natural ore, pellets, natural gas, fuel oil, coke oven gas, carbonate fluxes or slagging
 9 materials, and direct coal injection). For calculations, activity data for these inputs, including coke consumed for
 10 pig iron production, are in units consistent with the carbon content values. The carbon contained in the pig iron,
 11 blast furnace gas, and blast furnace inputs was estimated by multiplying the material-specific carbon content by
 12 each material type (see Table 4-88). In the absence of a default carbon content value from the *2006 IPCC*
 13 *Guidelines* for pellet, sinter, or natural ore consumed for pig iron production, a country-specific approach based on
 14 Tier 2 methodology is used. Pellet, sinter, and natural ore used as an input for pig iron production is assumed to
 15 have the same carbon content as direct reduced iron (2 percent). Carbon in blast furnace gas used to pre-heat the
 16 blast furnace air is combusted to form CO₂ during this process. Carbon contained in blast furnace gas used as a
 17 blast furnace input was not included in the deductions to avoid double-counting.

18 Emissions from steel production in EAFs were estimated by deducting the carbon contained in the steel produced
 19 from the carbon contained in the EAF anode, charge carbon, and scrap steel added to the EAF. Small amounts of
 20 carbon from DRI and pig iron to the EAFs were also included in the EAF calculation. For BOFs, estimates of carbon
 21 contained in BOF steel were deducted from carbon contained in inputs such as natural gas, coke oven gas, fluxes
 22 (i.e., limestone and dolomite), and pig iron. In each case, the carbon was calculated by multiplying material-specific
 23 carbon contents by each material type (see Table 4-88). For EAFs, the amount of EAF anode consumed was
 24 approximated by multiplying total EAF steel production by the amount of EAF anode consumed per metric ton of
 25 steel produced (0.002 metric tons EAF anode per metric ton steel produced [Steiner 2008]). The amount of carbon-
 26 containing flux (i.e., limestone and dolomite) used in EAF and BOF steel production was deducted from the “Other
 27 Process Uses of Carbonates” source category (CRT Source Category 2A4) to avoid double-counting.

28 Carbon dioxide emissions from the consumption of blast furnace gas and coke oven gas for other activities
 29 occurring at the steel mill were estimated by multiplying the amount of these materials consumed for these
 30 purposes by the material-specific carbon content (see Table 4-88).

31 **Table 4-88: Material Carbon Contents for Iron and Steel Production**

Material	kg C/kg
Coke	0.83
Direct Reduced Iron	0.02
Dolomite	0.13

EAF Carbon Electrodes	0.82
EAF Charge Carbon	0.83
Limestone	0.12
Pig Iron	0.04
Steel	0.01
Material	kg C/GJ
Coke Oven Gas	12.1
Blast Furnace Gas	70.8

Source: IPCC (2006), Table 4.3. Coke Oven Gas and Blast Furnace Gas, Table 1.3.

1 Carbon dioxide emissions associated with sinter production, direct reduced iron production, pellet production, pig
2 iron production, steel production, and other steel mill activities were summed to calculate the total CO₂ emissions
3 from iron and steel production (see Table 4-81 and Table 4-82).

4 The sinter production process results in fugitive emissions of CH₄, which are emitted via leaks in the production
5 equipment, rather than through the emission stacks or vents of the production plants. The fugitive emissions were
6 calculated by applying Tier 1 emission factors taken from the *2006 IPCC Guidelines* for sinter production (see Table
7 4-89). Although the *2006 IPCC Guidelines* also provide a Tier 1 methodology for CH₄ emissions from pig iron
8 production, it is not appropriate to use because CO₂ emissions for pig iron production are estimated using the Tier
9 2 mass balance methodology. The mass balance methodology makes a basic assumption that all carbon that enters
10 the pig iron production process either exits the process as part of a carbon-containing output or as CO₂ emissions;
11 the estimation of CH₄ emissions is precluded. Annual analysis of facility-level emissions reported during iron
12 production further supports this assumption and indicates that CH₄ emissions are below 500 kt CO₂ Eq. and well
13 below 0.05 percent of total national emissions. The production of direct reduced iron could also result in emissions
14 of CH₄ through the consumption of fossil fuels (e.g., natural gas, etc.); however, these emission estimates are
15 excluded due to data limitations. Pending further analysis and resources, EPA may include these emissions in
16 future reports to enhance completeness. EPA is still assessing the possibility of including these emissions in future
17 reports and have not included this data in the current report.

18 **Table 4-89: CH₄ Emission Factors for Sinter and Pig Iron Production**

Material Produced	Factor	Unit
Sinter	0.07	kg CH ₄ /metric ton

Source: IPCC (2006), Table 4.2.

19 Emissions of CO₂ from sinter production, direct reduced iron production, and pellet production were estimated by
20 multiplying total national sinter production, total national direct reduced iron production, and total national pellet
21 production by Tier 1 CO₂ emission factors (see Table 4-90). Because estimates of sinter production, direct reduced
22 iron production, and pellet production were not available, production was assumed to equal consumption.

23 **Table 4-90: CO₂ Emission Factors for Sinter Production, Direct Reduced Iron Production, and 24 Pellet Production**

Material Produced	Metric Ton CO ₂ /Metric Ton
Sinter	0.2
Direct Reduced Iron	0.7
Pellet Production	0.03

Source: IPCC (2006), Table 4.1.

25 The consumption of coking coal, natural gas, distillate fuel, and coal used in iron and steel production are adjusted
26 for within the Energy chapter to avoid double-counting of emissions reported within the IPPU chapter as these
27 fuels were consumed during non-energy related activities. More information on this methodology and examples of

1 adjustments made between the IPPU and Energy chapters are described in Annex 2.1, Methodology for Estimating
2 Emissions of CO₂ from Fossil Fuel Combustion.

3 Sinter consumption and pellet consumption data for 1990 through 2020 were obtained from AISI's *Annual*
4 *Statistical Report* (AISI 2004 through 2022) and through personal communications with AISI (Steiner 2008) (see
5 Table 4-91). These data from the AISI *Annual Statistical Report* were withheld for 2020 through 2022, so the 2019
6 values were used as estimated data for the missing 2020 through 2022 values and adjusted using GHGRP emissions
7 data, as described earlier in this Methodology and Time-Series Consistency section.

8 In general, direct reduced iron (DRI) consumption data were obtained from the U.S. Geological Survey (USGS)
9 *Minerals Yearbook – Iron and Steel Scrap* (USGS 1991 through 2020) and personal communication with the USGS
10 Iron and Steel Commodity Specialist (Tuck 2023a). Data for DRI consumed in EAFs were not available for the years
11 1990 and 1991. EAF DRI consumption in 1990 and 1991 was calculated by multiplying the total DRI consumption
12 for all furnaces by the EAF share of total DRI consumption in 1992. Data for DRI consumed in BOFs were not
13 available for the years 1990 through 1993. BOF DRI consumption in 1990 through 1993 was calculated by
14 multiplying the total DRI consumption for all furnaces (excluding EAFs and cupola) by the BOF share of total DRI
15 consumption (excluding EAFs and cupola) in 1994.

16 The Tier 1 CO₂ emission factors for sinter production, direct reduced iron production and pellet production were
17 obtained through the *2006 IPCC Guidelines* (IPCC 2006). Time-series data for pig iron production, coke, natural gas,
18 fuel oil, sinter, and pellets consumed in the blast furnace; pig iron production; and blast furnace gas produced at
19 the iron and steel mill and used in the metallurgical coke ovens and other steel mill activities were obtained from
20 AISI's *Annual Statistical Report* (AISI 2004 through 2021) and through personal communications with AISI (Steiner
21 2008) (see Table 4-91 and Table 4-92). Data including blast furnace gas, coke oven gas, natural gas, limestone,
22 sinter, and natural ore consumption for blast furnaces, coke production, and steelmaking furnaces (EAFs and BOFs)
23 from the AISI *Annual Statistical Report* were withheld for 2020 through 2022, so the 2019 values were used as
24 estimated data for the missing 2020 through 2022 values and adjusted using GHGRP emissions data, as described
25 earlier in this Methodology and Time-Series Consistency section. Similarly, the percent of total steel production for
26 EAF and BOF steelmaking processes were withheld for 2021 and 2022, so the 2020 values were used as estimated
27 data for the missing 2021 values and adjusted using GHGRP emissions data, as described earlier in this
28 Methodology and Time-Series Consistency section.

29 Data for EAF steel production, carbon-containing flux, EAF charge carbon, and natural gas consumption were
30 obtained from AISI's *Annual Statistical Report* (AISI 2004 through 2022) and through personal communications
31 with AISI (AISI 2006 through 2016, Steiner 2008). The factor for the quantity of EAF anode consumed per ton of
32 EAF steel produced was provided by AISI (Steiner 2008). Data for BOF steel production, carbon-containing flux,
33 natural gas, natural ore, pellet, sinter consumption as well as BOF steel production were obtained from AISI's
34 *Annual Statistical Report* (AISI 2004 through 2023) and through personal communications with AISI (Steiner 2008).
35 Data for EAF consumption of natural gas and BOF consumption of coke oven gas, limestone, and natural ore from
36 the AISI *Annual Statistical Report* were not available for 2021 and 2022, so 2020 values were used as estimated
37 data for the missing 2021 and 2022 values and adjusted using GHGRP emissions data, as described earlier in this
38 Methodology and Time-Series Consistency section. Data for EAF and BOF scrap steel, pig iron, and DRI
39 consumption were obtained from the USGS *Minerals Yearbook – Iron and Steel Scrap* (USGS 1991 through 2021)
40 and personal communication with the USGS Iron and Steel Commodity Specialist (Tuck 2023a). Data on coke oven
41 gas and blast furnace gas consumed at the iron and steel mill (other than in the EAF, BOF, or blast furnace) were
42 obtained from AISI's *Annual Statistical Report* (AISI 2004 through 2021) and through personal communications
43 with AISI (Steiner 2008). These data were not available for 2021 and 2022, so 2020 values were used as estimated
44 data for the missing 2021 and 2022 values and adjusted using GHGRP emissions data, as described earlier in this
45 Methodology and Time-Series Consistency section. Some data from the AISI *Annual Statistical Report* on natural
46 gas consumption were withheld for 2020 through 2022, so the 2019 values were used as estimated data for the

1 missing 2020 through 2022 values and adjusted using GHGRP emissions data, as described earlier in this
 2 Methodology and Time-Series Consistency section.

3 Data on blast furnace gas and coke oven gas sold for use as synthetic natural gas were obtained from EIA's *Natural*
 4 *Gas Annual 2019* (EIA 2020). Carbon contents for direct reduced iron, EAF carbon electrodes, EAF charge carbon,
 5 limestone, dolomite, pig iron, and steel were provided by the *2006 IPCC Guidelines*. The carbon contents for
 6 natural gas, fuel oil, and direct injection coal were obtained from EIA (EIA 2017b) and EPA (EPA 2010). Heat
 7 contents for fuel oil and direct injection coal were obtained from EIA (EIA 1992, 2011); natural gas heat content
 8 was obtained from Table 37 of AISI's *Annual Statistical Report* (AISI 2004 through 2021). Heat contents for coke
 9 oven gas and blast furnace gas were provided in Table 37 of AISI's *Annual Statistical Report* (AISI 2004 through
 10 2021) and confirmed by AISI staff (Carroll 2016).

11 **Table 4-91: Production and Consumption Data for the Calculation of CO₂ and CH₄ Emissions**
 12 **from Iron and Steel Production (Thousand Metric Tons)**

Source/Activity Data	1990	2005	2018	2019	2020	2021	2022
Sinter Production	12,239	8,315	4,687	4,378	3,747	4,182	3,935
Direct Reduced Iron							
Production	517	1,303	C	C	C	C	C
Pellet Production	60,563	50,096	30,793	29,262	25,044	27,949	26,300
Pig Iron Production							
Coke Consumption	24,946	13,832	7,618	7,291	6,240	6,964	6,553
Pig Iron Production	49,669	37,222	24,058	22,302	18,320	22,246	19,791
Direct Injection Coal							
Consumption	1,485	2,573	2,569	2,465	2,110	2,354	2,216
EAF Steel Production							
EAF Anode and Charge							
Carbon Consumption	67	1,127	1,133	1,137	1,118	1,129	1,123
Scrap Steel Consumption	42,691	46,600	C	C	C	C	C
Flux Consumption	319	695	998	998	998	998	998
EAF Steel Production	33,511	52,194	58,904	61,172	51,349	57,307	53,926
BOF Steel Production							
Pig Iron Consumption	47,307	34,400	C	C	C	C	C
Scrap Steel Consumption	14,713	11,400	C	C	C	C	C
Flux Consumption	576	582	408	363	311	347	326
BOF Steel Production	43,973	42,705	27,704	26,591	21,384	23,865	22,457

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13 **Table 4-92: Production and Consumption Data for the Calculation of CO₂ Emissions from Iron**
 14 **and Steel Production (Million ft³ unless otherwise specified)**

Source/Activity Data	1990	2005	2018	2019	2020	2021	2022
Pig Iron Production							
Natural Gas Consumption	56,273	59,844	40,204	37,934	32,465	36,232	34,095
Fuel Oil Consumption							
(thousand gallons)	163,397	16,170	3,365	2,321	1,986	2,217	2,086
Coke Oven Gas							
Consumption	22,033	16,557	13,337	12,926	11,063	12,346	11,618
Blast Furnace Gas							
Production	1,439,380	1,299,980	871,860	836,033	715,509	798,522	751,418
EAF Steel Production							
Natural Gas Consumption	15,905	19,985	8,556	9,115	7,801	8,706	8,192
BOF Steel Production							
Coke Oven Gas							
Consumption	3,851	524	405	389	333	372	350

Other Activities							
Coke Oven Gas Consumption	224,883	97,132	67,008	64,377	55,096	61,489	57,861
Blast Furnace Gas Consumption	1,414,778	1,295,520	867,838	832,119	712,159	794,783	747,900

1 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
2 through 2022.

3 **Uncertainty– TO BE UPDATED FOR FINAL REPORT**

4 The estimates of CO₂ emissions from metallurgical coke production are based on assessing uncertainties in
5 material production and consumption data and average carbon contents. Uncertainty is associated with the total
6 U.S. coking coal consumption, total U.S. coke production, and materials consumed during this process. Data for
7 coking coal consumption and metallurgical coke production are from different data sources (EIA) than data for
8 other carbonaceous materials consumed at coke plants (AISI), which does not include data for merchant coke
9 plants. There is uncertainty associated with the fact that coal tar and coke breeze production were estimated
10 based on coke production because coal tar and coke breeze production data were not available. Since merchant
11 coke plant data is not included in the estimate of other carbonaceous materials consumed at coke plants, the mass
12 balance equation for CO₂ from metallurgical coke production cannot be reasonably completed; therefore, for the
13 purpose of this analysis, uncertainty parameters are applied to primary data inputs to the calculation (i.e., coking
14 coal consumption and metallurgical coke production) only.

15 The estimates of CO₂ emissions from iron and steel production are based on material production and consumption
16 data and average carbon contents. There is uncertainty associated with the assumption that pellet production,
17 direct reduced iron and sinter consumption are equal to production. There is uncertainty with the
18 representativeness of the associated IPCC default emission factors. There is uncertainty associated with the
19 assumption that all coal used for purposes other than coking coal is for direct injection coal. There is also
20 uncertainty associated with the carbon contents for pellets, sinter, and natural ore, which are assumed to equal
21 the carbon contents of direct reduced iron, when consumed in the blast furnace. There is uncertainty associated
22 with the consumption of natural ore under current industry practices. For EAF steel production, there is
23 uncertainty associated with the amount of EAF anode and charge carbon consumed due to inconsistent data
24 throughout the time series. Also for EAF steel production, there is uncertainty associated with the assumption that
25 100 percent of the natural gas attributed to “steelmaking furnaces” by AISI is process-related and nothing is
26 combusted for energy purposes. Uncertainty is also associated with the use of process gases such as blast furnace
27 gas and coke oven gas. Data are not available to differentiate between the use of these gases for processes at the
28 steel mill versus for energy generation (i.e., electricity and steam generation); therefore, all consumption is
29 attributed to iron and steel production. These data and carbon contents produce a relatively accurate estimate of
30 CO₂ emissions; however, there are uncertainties associated with each.

31 For calculating the emissions estimates from iron and steel and metallurgical coke production, EPA utilizes a
32 number of data points taken from the AISI *Annual Statistical Report* (ASR). This report serves as a benchmark for
33 information on steel companies in United States, regardless if they are a member of AISI, which represents
34 integrated producers (i.e., blast furnace and EAF). During the compilation of the 1990 through 2016 *Inventory*
35 report EPA initiated conversation with AISI to better understand and update the qualitative and quantitative
36 uncertainty metrics associated with AISI data elements. AISI estimates their data collection response rate to range
37 from 75 to 90 percent, with certain sectors of the iron and steel industry not being covered by the ASR; therefore,
38 there is some inherent uncertainty in the values provided in the AISI ASR, including material production and
39 consumption data. There is also some uncertainty to which materials produced are exported to Canada. As
40 indicated in the introduction to this section, the trend for integrated facilities has moved to more use of EAFs and
41 fewer BOFs. This trend may not be completely captured in the current data which also increases uncertainty. EPA
42 assigned an uncertainty range of ±10 percent for the primary data inputs (e.g., consumption and production values
43 for each production process, heat and carbon content values) to calculate overall uncertainty from iron and steel

production, and using this suggested uncertainty provided in Table 4.4 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023). During EPA’s discussion with AISI, AISI noted that an uncertainty range of ±5 percent would be a more appropriate approximation to reflect their coverage of integrated steel producers in the United States. EPA will continue to assess the best range of uncertainty for these values. EPA assigned an uncertainty range of ±25 percent for the Tier 1 CO₂ emission factors for the sinter, direct reduced iron, and pellet production processes, and using this suggested uncertainty provided in Table 4.4 of the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023).

The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-93 for metallurgical coke production and iron and steel production. Total CO₂ emissions from metallurgical coke production and iron and steel production for 2022 were estimated to be between 33.8 and 49.6 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 19 percent below and 19 percent above the emission estimate of 39.7 MMT CO₂ Eq. Total CH₄ emissions from metallurgical coke production and iron and steel production for 2021 were estimated to be between 0.007 and 0.010 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 20 percent below and 21 percent above the emission estimate of 0.008 MMT CO₂ Eq.

Table 4-93: Approach 2 Quantitative Uncertainty Estimates for CO₂ and CH₄ Emissions from Iron and Steel Production and Metallurgical Coke Production (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Metallurgical Coke & Iron and Steel Production	CO ₂	39.7	33.8	49.6	-19%	+19%
Metallurgical Coke & Iron and Steel Production	CH ₄	+	+	+	-20%	+21%

+ Does not exceed 0.05 MMT CO₂ Eq.

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

For more information on the general QA/QC process applied to this source category, consistent with Volume 1, Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in the introduction of the IPPU chapter.

Recalculations Discussion

Recalculations were performed for the year 2021 with updated USGS values for DRI, pig iron, and scrap steel consumption for both BOF and EAF steel production. Additionally, revisions to GHGRP data for 2020 and 2021 resulted in minor changes to activity data that were adjusted using GHGRP data, as described in the Methodology and Time-Series Consistency section. Compared to the previous *Inventory*, CO₂ emissions from steel production increased by less than 1 percent (7 kt CO₂) in 2020 and by less than 1 percent (211 kt CO₂) in 2021.

Planned Improvements

Significant activity data for 2020 through 2022 were not available for this report and were estimated using 2019 values and adjusted using GHGRP emissions data. EPA will continue to explore sources of 2020 through 2022 data and other estimation approaches. EPA will evaluate and analyze data reported under EPA’s GHGRP to improve the emission estimates for Iron and Steel Production process categories. Particular attention will be made to ensure time-series consistency of the emissions estimates presented in future *Inventory* reports, consistent with IPCC and

1 UNFCCC guidelines. This is required as the facility-level reporting data from EPA's GHGRP, with the program's initial
2 requirements for reporting of emissions in calendar year 2010, are not available for all inventory years (i.e., 1990
3 through 2009) as required for this *Inventory*. In implementing improvements and integration of data from EPA's
4 GHGRP, EPA will rely on the latest guidance from the IPCC on the use of facility-level data in national inventories.⁷⁵
5 This is a near to medium-term improvement, and per preliminary work, EPA estimates that the earliest this
6 improvement could be incorporated is the next (i.e., 2025) *Inventory* submission.

7 Additional improvements include accounting for emission estimates for the production of metallurgical coke in the
8 Energy chapter as well as identifying the amount of carbonaceous materials, other than coking coal, consumed at
9 merchant coke plants. Other potential improvements include identifying the amount of coal used for direct
10 injection and the amount of coke breeze, coal tar, and light oil produced during coke production. Efforts will also
11 be made to identify information to better characterize emissions from the use of process gases and fuels within
12 the Energy and IPPU chapters. Additional efforts will be made to improve the reporting between the IPPU and
13 Energy chapters, particularly the inclusion of a quantitative summary of the carbon balance in the United States.
14 This planned improvement is a long-term improvement and is still in development. It is not included in this current
15 *Inventory* report. EPA estimates that the earliest this improvement could be incorporated is the next (i.e., 2025)
16 *Inventory* submission.

17 4.19 Ferroalloy Production (CRT Source 18 Category 2C2)

19 Carbon dioxide (CO₂) and methane (CH₄) are emitted from the production of several ferroalloys. Ferroalloys are
20 composites of iron (Fe) and other elements such as silicon (Si), manganese (Mn), and chromium (Cr). Emissions
21 from fuels consumed for energy purposes during the production of ferroalloys are accounted for in the Energy
22 chapter. Emissions from the production of two types of ferrosilicon (25 to 55 percent and 56 to 95 percent silicon),
23 silicon metal (96 to 99 percent silicon), and miscellaneous alloys (32 to 65 percent silicon) have been calculated.

24 Emissions from the production of ferrochromium and ferromanganese are not included because of the small
25 number of manufacturers of these materials in the United States. Government information disclosure rules
26 prevent the publication of production data for these production facilities. Additionally, production of
27 ferrochromium in the United States ceased in 2009 (USGS 2013a).

28 Similar to emissions from the production of iron and steel, CO₂ is emitted when metallurgical coke is oxidized
29 during a high-temperature reaction with iron and the selected alloying element. Due to the strong reducing
30 environment, CO is initially produced and eventually oxidized to CO₂. A representative reaction equation for the
31 production of 50 percent ferrosilicon (FeSi) is given below:



33 While most of the carbon contained in the process materials is released to the atmosphere as CO₂, a percentage is
34 also released as CH₄ and other volatiles. The amount of CH₄ that is released is dependent on furnace efficiency,
35 operation technique, and control technology.

⁷⁵ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

1 Ferroalloys are used to alter the material properties of the steel. Ferroalloys are produced in conjunction with the
 2 iron and steel industry, often at co-located facilities, and production trends closely follow that of the iron and steel
 3 industry. As of 2020, 11 facilities in the United States produce ferroalloys (USGS 2022b).

4 Emissions of CO₂ from ferroalloy production in 2022 were 1.3 MMT CO₂ Eq. (1,327 kt CO₂) (see Table 4-94 and
 5 Table 4-95), which is a 15 percent reduction since 2021 and a 38 percent reduction since 1990. Emissions of CH₄
 6 from ferroalloy production in 2022 were 0.01 MMT CO₂ Eq. (0.4 kt CH₄), which is a 15 percent decrease since 2021
 7 and a 45 percent decrease since 1990. The decrease in emissions since 1990 can largely be attributed to the
 8 closure of two facilities in 2018.

9 **Table 4-94: CO₂ and CH₄ Emissions from Ferroalloy Production (MMT CO₂ Eq.)**

Gas	1990	2005	2018	2019	2020	2021	2022
CO ₂	2.2	1.4	2.1	1.6	1.4	1.6	1.3
CH ₄	+	+	+	+	+	+	+
Total	2.2	1.4	2.1	1.6	1.4	1.6	1.3

+ Does not exceed 0.05 MMT CO₂ Eq.

10 **Table 4-95: CO₂ and CH₄ Emissions from Ferroalloy Production (kt)**

Gas	1990	2005	2018	2019	2020	2021	2022
CO ₂	2,152	1,392	2,063	1,598	1,377	1,567	1,327
CH ₄	1	+	1	+	+	+	+

+ Does not exceed 0.5 kt

11 Methodology and Time-Series Consistency

12 Emissions of CO₂ and CH₄ from ferroalloy production are calculated⁷⁶ using a Tier 1 method from the *2006 IPCC*
 13 *Guidelines*, in accordance with the IPCC methodological decision tree and available data. Annual ferroalloy
 14 production is multiplied by material-specific default emission factors provided by IPCC (IPCC 2006). The Tier 1
 15 equations for CO₂ and CH₄ emissions are as follows:

16 **Equation 4-13: 2006 IPCC Guidelines Tier 1: CO₂ Emissions for Ferroalloy Production (Equation**
 17 **4.15)**

$$E_{CO_2} = \sum_i (MP_i \times EF_i)$$

19 where,

20 E_{CO₂} = CO₂ emissions, metric tons

21 MP_{*i*} = Production of ferroalloy type *i*, metric tons

22 EF_{*i*} = Generic emission factor for ferroalloy type *i*, metric tons CO₂/metric ton specific
 23 ferroalloy product

⁷⁶ EPA has not integrated aggregated facility-level GHGRP information to inform these estimates. The aggregated information (e.g., activity data and emissions) associated with production of ferroalloys did not meet criteria to shield underlying confidential business information (CBI) from public disclosure.

1 **Equation 4-14: 2006 IPCC Guidelines Tier 1: CH₄ Emissions for Ferroalloy Production (Equation**
2 **4.18)**

$$E_{CH_4} = \sum_i (MP_i \times EF_i)$$

4 where,

5 E_{CH_4} = CH₄ emissions, kg

6 MP_i = Production of ferroalloy type i , metric tons

7 EF_i = Generic emission factor for ferroalloy type i , kg CH₄/metric ton specific ferroalloy product

8 Default emission factors were used because country-specific emission factors are not currently available. The
9 following emission factors were used to develop annual CO₂ and CH₄ estimates:

- 10 • Ferrosilicon, 25 to 55 percent Si and Miscellaneous Alloys, 32 to 65 percent Si: 2.5 metric tons CO₂/metric
11 ton of alloy produced, 1.0 kg CH₄/metric ton of alloy produced.
- 12 • Ferrosilicon, 56 to 95 percent Si: 4.0 metric tons CO₂/metric ton alloy produced, 1.0 kg CH₄/metric ton of
13 alloy produced.
- 14 • Silicon Metal: 5.0 metric tons CO₂/metric ton metal produced, 1.2 kg CH₄/metric ton metal produced.

15 It was assumed that 100 percent of the ferroalloy production was produced using petroleum coke in an electric arc
16 furnace process (IPCC 2006), although some ferroalloys may have been produced with coking coal, wood, other
17 biomass, or graphite carbon inputs. The amount of petroleum coke consumed in ferroalloy production was
18 calculated assuming that the petroleum coke used is 90 percent carbon (C) and 10 percent inert material (Onder
19 and Bagdoyan 1993).

20 The use of petroleum coke for ferroalloy production is adjusted for within the Energy chapter as this fuel was
21 consumed during non-energy related activities. Additional information on the adjustments made within the Energy
22 sector for Non-Energy Use of Fuels is described in both the Methodology section of CO₂ from Fossil Fuel
23 Combustion (3.1 Fossil Fuel Combustion [CRT Source Category 1A]) and Annex 2.1, Methodology for Estimating
24 Emissions of CO₂ from Fossil Fuel Combustion.

25 Ferroalloy production data for 1990 through 2022 (see Table 4-96) were obtained from the U.S. Geological Survey
26 (USGS) through the *Minerals Yearbook: Silicon* (USGS 1996 through 2022) and the *Minerals Industry Survey: Silicon*
27 (USGS 2023b). The following data were available from the USGS publications for the time series:

- 28 • Ferrosilicon, 25 to 55 percent Si: Annual production data were available from 1990 through 2010.
- 29 • Ferrosilicon, 56 to 95 percent Si: Annual production data were available from 1990 through 2010.
- 30 • Silicon Metal: Annual production data were available from 1990 through 2005. Production data for 2005
31 were used as estimates for 2006 through 2010 because data for these years were not available due to
32 government information disclosure rules.
- 33 • Miscellaneous Alloys, 32 to 65 percent Si: Annual production data were available from 1990 through
34 1998. Starting 1999, USGS reported miscellaneous alloys and ferrosilicon containing 25 to 55 percent
35 silicon as a single category.

36 Starting with the 2011 publication, USGS ceased publication of production quantity by ferroalloy product and
37 began reporting all the ferroalloy production data as a single category (i.e., Total Silicon Materials Production). This
38 is due to the small number of ferroalloy manufacturers in the United States and government information
39 disclosure rules. Ferroalloy product shares developed from the 2010 production data (i.e., ferroalloy product
40 production divided by total ferroalloy production) were used with the total silicon materials production quantity to
41 estimate the production quantity by ferroalloy product type for 2011 through 2022 (USGS 2017 through 2022).

1 **Table 4-96: Production of Ferroalloys (Metric Tons)**

Year	1990	2005	2018	2019	2020	2021	2022
Ferrosilicon 25%-55%	321,385	123,000	189,846	147,034	126,681	144,227	122,119
Ferrosilicon 56%-95%	109,566	86,100	167,511	129,736	111,778	127,259	107,752
Silicon Metal	145,744	148,000	183,642	142,229	122,541	139,514	118,128
Misc. Alloys 32-65%	72,442	NA	NA	NA	NA	NA	NA

NA (Not Available) for product type, aggregated with ferrosilicon (25-55% Si)

2 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
3 through 2022.

4 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

5 Annual ferroalloy production was reported by the USGS in three broad categories until the 2010 publication:
6 ferroalloys containing 25 to 55 percent silicon (including miscellaneous alloys), ferroalloys containing 56 to 95
7 percent silicon, and silicon metal (through 2005 only, 2005 value used as an estimate for 2006 through 2010).
8 Starting with the *2011 Minerals Yearbook: Silicon*, USGS started reporting all the ferroalloy production under a
9 single category: total silicon materials production. The total silicon materials quantity was allocated across the
10 three categories, based on the 2010 production shares for the three categories. Refer to the Methodology section
11 for further details. Additionally, production data for silvery pig iron (alloys containing less than 25 percent silicon)
12 are not reported by the USGS to avoid disclosing proprietary company data. Emissions from this production
13 category, therefore, were not estimated.

14 Some ferroalloys may be produced using wood or other biomass as a primary or secondary carbon source
15 (carbonaceous reductants); however, information and data regarding these practices were not available. Emissions
16 from ferroalloys produced with wood or other biomass would not be counted under this source because wood-
17 based carbon is of biogenic origin.⁷⁷ Even though emissions from ferroalloys produced with coking coal or graphite
18 inputs would be counted in national trends, they may be generated with varying amounts of CO₂ per unit of
19 ferroalloy produced. The most accurate method for these estimates would be to base calculations on the amount
20 of reducing agent used in the process, rather than the amount of ferroalloys produced. These data, however, were
21 not available, and are also often considered confidential business information.

22 Emissions of CH₄ from ferroalloy production will vary depending on furnace specifics, such as type, operation
23 technique, and control technology. Higher heating temperatures and techniques such as sprinkle charging would
24 reduce CH₄ emissions; however, specific furnace information was not available or included in the CH₄ emission
25 estimates.

26 EPA assigned a uncertainty range of ±25 percent for the primary emission factors (i.e., ferrosilicon 25-55% Si,
27 ferrosilicon 56-95% Si, and silicon metal), and an uncertainty range of ±5 percent for the 2010 production values
28 for ferrosilicon 25-55% Si, ferrosilicon 56-95% Si, and silicon metal production and the 2021 total silicon materials
29 production value used to calculate emissions from overall ferroalloy production. Using these suggested
30 uncertainties provided in in Table 4.9 of Section 4.3.3.2 of the *2006 IPCC Guidelines* is appropriate based on expert
31 judgment (RTI 2023).

32 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-97. Ferroalloy
33 production CO₂ emissions from 2022 were estimated to be between 1.4 and 1.8 MMT CO₂ Eq. at the 95 percent
34 confidence level. This indicates a range of approximately 13 percent below and 13 percent above the emission
35 estimate of 1.6 MMT CO₂ Eq. Ferroalloy production CH₄ emissions were estimated to be between a range of
36 approximately 13 percent below and 13 percent above the emission estimate of 0.01 MMT CO₂ Eq.

⁷⁷ Emissions and sinks of biogenic carbon are accounted for in the Land Use, Land-Use Change, and Forestry chapter.

Table 4-97: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Ferroalloy Production (MMT CO₂ Eq. and Percent)

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (%)			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Ferroalloy Production	CO ₂	1.6	1.4	1.8	-13%	+13%
Ferroalloy Production	CH ₄	+	+	+	-13%	+13%

+ Does not exceed 0.05 MMT CO₂ Eq.
^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

For more information on the general QA/QC process applied to this source category, consistent with Volume 1, Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in the introduction of the IPPU chapter and Annex 8.

Recalculations Discussion

No recalculations were performed for the 1990 to 2021 portion of the time series.

Planned Improvements

Pending available resources and prioritization of improvements for more significant sources, EPA will continue to evaluate and analyze data reported under EPA’s GHGRP that would be useful to improve the emission estimates and category-specific QC procedures for the Ferroalloy Production source category. Given the small number of facilities and reporting thresholds, particular attention will be made to ensure completeness and time-series consistency of the emissions estimates presented in future *Inventory* reports, consistent with IPCC and UNFCCC guidelines. This is required as the facility-level reporting data from EPA’s GHGRP, with the program’s initial requirements for reporting of emissions in calendar year 2010, are not available for all inventory years (i.e., 1990 through 2009) as required for this *Inventory*. In implementing improvements and integration of data from EPA’s GHGRP, the latest guidance from the IPCC on the use of facility-level data in national inventories will be relied upon.⁷⁸ This is a long-term planned improvement, and EPA is still assessing the possibility of incorporating this improvement into the *Inventory*. This improvement has not been included in the current *Inventory* report.

4.20 Aluminum Production (CRT Source Category 2C3)

Aluminum is a lightweight, malleable, and corrosion-resistant metal that is used in many manufactured products, including aircraft, automobiles, bicycles, and kitchen utensils. As of recent reporting, the United States was the

⁷⁸ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

1 ninth⁷⁹ largest producer of primary aluminum with an estimated aluminum production of 860 thousand metric
 2 tons, with approximately 1.2 percent of the world total production (USGS 2022). The United States was also a
 3 major importer of primary aluminum. The production of primary aluminum—in addition to consuming large
 4 quantities of electricity—results in process-related emissions of carbon dioxide (CO₂) and two perfluorocarbons
 5 (PFCs): perfluoromethane (CF₄) and perfluoroethane (C₂F₆).

6 Carbon dioxide is emitted during the aluminum smelting process when alumina (aluminum oxide, Al₂O₃) is reduced
 7 to aluminum using the Hall-Héroult reduction process. The reduction of the alumina occurs through electrolysis in
 8 a molten bath of natural or synthetic cryolite (Na₃AlF₆). The reduction cells contain a carbon (C) lining that serves
 9 as the cathode. Carbon is also contained in the anode, which can be a carbon mass of paste, coke briquettes, or
 10 prebaked carbon blocks from petroleum coke. During reduction, most of this carbon is oxidized and released to the
 11 atmosphere as CO₂.

12 Process emissions of CO₂ from aluminum production were estimated to be 1.4 MMT CO₂ Eq. (1,446 kt) in 2022 (see
 13 Table 4-98). The carbon anodes consumed during aluminum production consist of petroleum coke and, to a minor
 14 extent, coal tar pitch. The petroleum coke portion of the total CO₂ process emissions from aluminum production is
 15 considered to be a non-energy use of petroleum coke and is accounted for here and not under the CO₂ from fossil
 16 fuel combustion source category of the Energy sector. Similarly, the coal tar pitch portion of these CO₂ process
 17 emissions is accounted for here.

18 **Table 4-98: CO₂ Emissions from Aluminum Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Aluminum Production	6.8	4.1	1.5	1.9	1.7	1.5	1.4

19 **Table 4-99: CO₂ Emissions from Aluminum Production (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Aluminum Production	6,831	4,142	1,455	1,880	1,748	1,541	1,446

20 In addition to CO₂ emissions, the aluminum production industry is also a source of PFC emissions. During the
 21 smelting process, when the alumina ore content of the electrolytic bath falls below critical levels required for
 22 electrolysis, rapid voltage increases occur, which are termed High Voltage Anode Effects (HVAEs). HVAEs cause
 23 carbon from the anode and fluorine from the dissociated molten cryolite bath to combine, thereby producing
 24 fugitive emissions of CF₄ and C₂F₆. In general, the magnitude of emissions for a given smelter and level of
 25 production depends on the frequency and duration of these anode effects. As the frequency and duration of the
 26 anode effects increase, emissions increase. Another type of anode effect, Low Voltage Anode Effects (LVAEs),
 27 became a concern in the early 2010s as the aluminum industry increasingly began to use cell technologies with
 28 higher amperage and additional anodes (IPCC 2019). LVAEs emit CF₄ and are included in PFC emission totals from
 29 2006 forward.

30 Since 1990, emissions of CF₄ and C₂F₆ have both declined by 96 and 97 percent respectively, to 0.62 MMT CO₂ Eq.
 31 of CF₄ (0.1 kt) and 0.08 MMT CO₂ Eq. of C₂F₆ (0.01 kt) in 2022, respectively, as shown in Table 4-100 and Table
 32 4-101. This decline is due both to reductions in domestic aluminum production and to actions taken by aluminum
 33 smelting companies to reduce the frequency and duration of anode effects. These actions include technology and
 34 operational changes such as employee training, use of computer monitoring, and changes in alumina feeding
 35 techniques. Since 1990, aluminum production has declined by 78 percent, while the combined CF₄ and C₂F₆
 36 emission rate (per metric ton of aluminum produced) has been reduced by 78 percent. PFC emissions decreased by
 37 approximately 18 percent between 2021 and 2022. Aluminum production also decreased in 2022, down 3 percent
 38 from 2021.

⁷⁹ Based on the U.S. USGS (2022) Aluminum factsheet, assuming all countries grouped under the “other countries” categories
 all have lower production than the U.S. Available at: <https://pubs.usgs.gov/periodicals/mcs2023/mcs2023-aluminum.pdf>.

1 **Table 4-100: PFC Emissions from Aluminum Production (MMT CO₂ Eq.)**

Gas	1990	2005	2018	2019	2020	2021	2022
CF ₄	16.1	2.6	1.0	1.1	1.2	0.8	0.7
C ₂ F ₆	3.2	0.5	0.4	0.3	0.2	0.1	0.1
Total	19.3	3.1	1.4	1.4	1.4	0.9	0.8

Note: Totals may not sum due to independent rounding.

2 **Table 4-101: PFC Emissions from Aluminum Production (kt)**

Gas	1990	2005	2018	2019	2020	2021	2022
CF ₄	2.4	0.4	0.2	0.2	0.2	0.1	0.1
C ₂ F ₆	0.29	0.05	0.03	0.03	0.02	0.01	0.01

3 In 2022, U.S. primary aluminum production totaled approximately 0.86 million metric tons, a 3 percent decrease
 4 from 2021 production levels (USGS 2023). In 2022, three companies managed production at six operational
 5 primary aluminum smelters in five states. Two smelters operated at full capacity during 2022. The other four
 6 smelters operated at reduced capacity and one of these four smelters began a temporary shutdown in June (USGS
 7 2023). Domestic smelters were operating at about 52 percent of capacity of 1.64 million tons per year at year end
 8 2022 (USGS 2023).

9 Methodology and Time-Series Consistency

10 Process CO₂ and PFC (i.e., CF₄ and C₂F₆) emission estimates from primary aluminum production for 2010 through
 11 2022 are available from EPA’s GHGRP Subpart F (Aluminum Production) (EPA 2023). Under EPA’s GHGRP, facilities
 12 began reporting primary aluminum production process emissions (for 2010) in 2011; as a result, GHGRP data (for
 13 2010 through 2022) are available to be incorporated into the *Inventory*. EPA’s GHGRP mandates that all facilities
 14 that contain an aluminum production process must report: CF₄ and C₂F₆ emissions from anode effects in all
 15 prebake and Søderberg electrolysis cells, CO₂ emissions from anode consumption during electrolysis in all prebake
 16 and Søderberg cells, and all CO₂ emissions from onsite anode baking. To estimate the process emissions, EPA’s
 17 GHGRP uses the process-specific equations detailed in Subpart F (aluminum production).⁸⁰ These equations are
 18 based on the Tier 2/Tier 3 IPCC (2006) methods for primary aluminum production, and Tier 1 methods when
 19 estimating missing data elements. It should be noted that the same methods (i.e., *2006 IPCC Guidelines*) were used
 20 for estimating the emissions prior to the availability of the reported GHGRP data in the *Inventory*. Prior to 2010,
 21 aluminum production data were provided through EPA’s Voluntary Aluminum Industrial Partnership (VAIP).

22 As previously noted, the use of petroleum coke for aluminum production is adjusted for within the Energy chapter
 23 to avoid double counting emissions as this fuel was consumed during non-energy related activities. Additional
 24 information on the adjustments made within the Energy sector for Non-Energy Use of Fuels is described in both
 25 the Methodology section of CO₂ from Fossil Fuel Combustion (3.2 Carbon Emitted from Non-Energy Uses of Fossil
 26 Fuels [CRT Source Category 1A]) and Annex 2.3, Methodology for Estimating Carbon Emitted from Non-Energy
 27 Uses of Fossil Fuels.

28 Process CO₂ Emissions from Anode Consumption and Anode Baking

29 Carbon dioxide emission estimates for the years prior to the introduction of EPA’s GHGRP in 2010 were estimated
 30 using *2006 IPCC Guidelines* methods, but individual facility reported data were combined with process-specific
 31 emissions modeling. These estimates were based on information previously gathered from EPA’s Voluntary
 32 Aluminum Industrial Partnership (VAIP) program, U.S. Geological Survey (USGS) Mineral Commodity reviews, and

⁸⁰ Code of Federal Regulations, Title 40: Protection of Environment, Part 98: Mandatory Greenhouse Gas Reporting, Subpart F—Aluminum Production. See <https://www.ecfr.gov/cgi-bin/text-idx?SID=24a41781dfe4218b339e914de03e8727&mc=true&node=pt40.23.98&rgn=div5#sp40.23.98.f>.

1 The Aluminum Association (USAA) statistics, among other sources. Since pre- and post-GHGRP estimates use the
2 same methodology, emission estimates are comparable across the time series.

3 Most of the CO₂ emissions released during aluminum production occur during the electrolysis reaction of the
4 carbon anode, as described by the following reaction:



6 For prebake smelter technologies, CO₂ is also emitted during the anode baking process. These emissions can
7 account for approximately 10 percent of total process CO₂ emissions from prebake smelters.

8 Depending on the availability of smelter-specific data, the CO₂ emitted from electrolysis at each smelter was
9 estimated from: (1) the smelter's annual anode consumption, (2) the smelter's annual aluminum production and
10 rate of anode consumption (per ton of aluminum produced) for previous and/or following years, or (3) the
11 smelter's annual aluminum production and IPCC default CO₂ emission factors. The first approach tracks the
12 consumption and carbon content of the anode, assuming that all carbon in the anode is converted to CO₂. Sulfur,
13 ash, and other impurities in the anode are subtracted from the anode consumption to arrive at a carbon
14 consumption figure. This approach corresponds to either the IPCC Tier 2 or Tier 3 method, depending on whether
15 smelter-specific data on anode impurities are used. The second approach interpolates smelter-specific anode
16 consumption rates to estimate emissions during years for which anode consumption data are not available. This
17 approach avoids substantial errors and discontinuities that could be introduced by reverting to Tier 1 methods for
18 those years. The last approach corresponds to the IPCC Tier 1 method (IPCC 2006) and is used in the absence of
19 present or historic anode consumption data.

20 The equations used to estimate CO₂ emissions in the Tier 2 and 3 methods vary depending on smelter type (IPCC
21 2006). For Prebake cells, the process formula accounts for various parameters, including net anode consumption,
22 and the sulfur, ash, and impurity content of the baked anode. For anode baking emissions, the formula accounts
23 for packing coke consumption, the sulfur and ash content of the packing coke, as well as the pitch content and
24 weight of baked anodes produced. For Söderberg cells, the process formula accounts for the weight of paste
25 consumed per metric ton of aluminum produced, and pitch properties, including sulfur, hydrogen, and ash
26 content.

27 Through the VAIP, anode consumption (and some anode impurity) data have been reported for 1990, 2000, 2003,
28 2004, 2005, 2006, 2007, 2008, and 2009. Where available, smelter-specific process data reported under the VAIP
29 were used; however, if the data were incomplete or unavailable, information was supplemented using industry
30 average values recommended by IPCC (2006). Smelter-specific CO₂ process data were provided by 18 of the 23
31 operating smelters in 1990 and 2000, by 14 out of 16 operating smelters in 2003 and 2004, 14 out of 15 operating
32 smelters in 2005, 13 out of 14 operating smelters in 2006, 5 out of 14 operating smelters in 2007 and 2008, and 3
33 out of 13 operating smelters in 2009. For years where CO₂ emissions data or CO₂ process data were not reported
34 by these companies, estimates were developed through linear interpolation, and/or assuming representative (e.g.,
35 previously reported or industry default) values.

36 In the absence of any previous historical smelter-specific process data (i.e., 1 out of 13 smelters in 2009; 1 out of
37 14 smelters in 2006, 2007, and 2008; 1 out of 15 smelters in 2005; and 5 out of 23 smelters between 1990 and
38 2003), CO₂ emission estimates were estimated using Tier 1 Söderberg and/or Prebake emission factors (metric ton
39 of CO₂ per metric ton of aluminum produced) from IPCC (2006).

40 **Process PFC Emissions from Anode Effects**

41 **High Voltage Anode Effects**

42 Smelter-specific PFC emissions from aluminum production for 2010 through 2022 were reported to EPA under its
43 GHGRP. To estimate their PFC emissions from HVAEs and report them under EPA's GHGRP, smelters use an
44 approach identical to the Tier 3 approach in the *2006 IPCC Guidelines* (IPCC 2006). Specifically, they use a smelter-
45 specific slope coefficient as well as smelter-specific operating data to estimate an emission factor using the
46 following equation:

$$PFC = S \times AE$$

$$AE = F \times D$$

where,

PFC = CF₄ or C₂F₆, kg/MT aluminum

S = Slope coefficient, PFC/AE

AE = Anode effect, minutes/cell-day

F = Anode effect frequency per cell-day

D = Anode effect duration, minutes

They then multiply this emission factor by aluminum production to estimate PFC emissions from HVAEs. All U.S. aluminum smelters are required to report their emissions under EPA's GHGRP.

Perfluorocarbon emissions for the years prior to 2010 were estimated using the same equation, but the slope-factor used for some smelters was technology-specific rather than smelter-specific, making the method a Tier 2 rather than a Tier 3 approach for those smelters. Emissions and background data were reported to EPA under the VAIP. For 1990 through 2009, smelter-specific slope coefficients were available and were used for smelters representing between 30 and 94 percent of U.S. primary aluminum production. The percentage changed from year to year as some smelters closed or changed hands and as the production at remaining smelters fluctuated. For smelters that did not report smelter-specific slope coefficients, IPCC technology-specific slope coefficients were applied (IPCC 2006). The slope coefficients were combined with smelter-specific anode effect data collected by aluminum companies and reported under the VAIP to estimate emission factors over time. For 1990 through 2009, smelter-specific anode effect data were available for smelters representing between 80 and 100 percent of U.S. primary aluminum production. Where smelter-specific anode effect data were not available, representative values (e.g., previously reported or industry averages) were used.

For all smelters, emission factors were multiplied by annual production to estimate annual emissions at the smelter level. For 1990 through 2009, smelter-specific production data were available for smelters representing between 30 and 100 percent of U.S. primary aluminum production. (For the years after 2000, this percentage was near the high end of the range.) Production at non-reporting smelters was estimated by calculating the difference between the production reported under VAIP and the total U.S. production supplied by USGS, and then allocating this difference to non-reporting smelters in proportion to their production capacity. Emissions were then aggregated across smelters to estimate national emissions.

Table 4-102: Summary of HVAE Emissions (MMT CO₂ Eq.)

Year	1990	2005	2018	2019	2020	2021	2022
HVAE Emissions	19.3	3.1	1.4	1.4	1.4	0.9	0.7

Low Voltage Anode Effects

LVAE emissions of CF₄ were estimated for 2006 through 2022 based on the Tier 1 (technology-specific, production-based) method in the *2019 Refinement to the 2006 IPCC Guidelines for National Greenhouse Gas Inventories* (IPCC 2019). Prior to 2006, LVAE emissions are believed to have been negligible.⁸¹ The Tier 1 method is used in the LVAE emissions calculations from aluminum production in the absence of smelter-specific data available to quantify the LVAE-specific process emissions. National aluminum production estimates (allocated to smelters as described

⁸¹ The *2019 Refinement* states, "Since 2006, the global aluminum industry has undergone changes in technology and operating conditions that make LVAE emissions much more prevalent¹²; these changes have occurred not only through uptake of newer technologies (e.g., PFPB_L to PFPB_M) but also during upgrades within the same technology in order to maximize productivity and reduce energy use" (IPCC 2019). Footnote #12 uses the example of PFPB_L, which is prevalent in the United States, as an older technology that has been upgraded.

1 below) and the technology used in individual smelters were the best available data to perform the emissions
 2 calculations, as smelter-specific production data is not publicly available.

3 The following equation was used to estimate LVAE PFC emissions:

4 **Equation 4-15: CF₄ Emissions Resulting from Low Voltage Anode Effects**

5
$$LVAE E_{CF_4} = LVAE EF_{CF_4} \times MP$$

6 where,

7 LVAE E_{CF₄} = LVAE emissions of CF₄ from aluminum production, kg CF₄

8 LVAE EF_{CF₄} = LVAE emission factor for CF₄ (default by cell technology type)

9 MP = metal production by cell technology type, tons Al.

10 In the LVAE emissions calculations, the Metal Production (MP) factor is calculated differently for the years 2006
 11 through 2009 than for 2010 and beyond. For years prior to GHGRP reporting (2006 through 2009), the MP factor is
 12 calculated by dividing the annual production reported by USGS with the total U.S. capacity reported for this
 13 specific year, based on the USGS yearbook and applying this national utilization factor to each facility’s production
 14 capacity to obtain an estimated facility production value. For GHGRP reporting years (2010+), the methodology to
 15 calculate the MP value was changed to allocate the total annual production reported by USAA, based on the
 16 distribution of CO₂ emissions amongst the operating smelters in a specific year. The latter improves the accuracy of
 17 the LVAE emissions estimates over assuming capacity utilization is the same at all smelters. The main drawback of
 18 using this methodology to calculate the MP factor is that, in some instances, it led to production estimates that are
 19 slightly larger (less than six percent) than the production capacity reported that year. In practice, this is most likely
 20 explained by the differences in process efficiencies at each facility and to a lesser extent, differences in
 21 measurements and methods used by each facility to obtain their CO₂ estimates and the degree of uncertainty in
 22 the USGS annual production reporting.

23 Once LVAE emissions were estimated, they were then combined with HVAE emissions estimates to calculate total
 24 PFC emissions from aluminum production.

25 **Table 4-103: Summary of LVAE Emissions (MMT CO₂ Eq.)**

Year	2006	2018	2019	2020	2021	2022
LVAE Emissions	0.13	0.05	0.07	0.06	0.05	0.05

26 **Production Data**

27 Between 1990 and 2009, production data were provided under the VAIP by 21 of the 23 U.S. smelters that
 28 operated during at least part of that period. For the non-reporting smelters, production was estimated based on
 29 the difference between reporting smelters and national aluminum production levels as reported to USGS, with
 30 allocation to specific smelters based on reported production capacities (USGS 1990 through 2009).

31 National primary aluminum production data for 2010 through 2022 were compiled using USGS Mineral Industry
 32 Surveys, and the USGS Mineral Commodity Summaries.

33 **Table 4-104: Production of Primary Aluminum (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
Production (kt)	4,048	2,481	891	1093	1,012	889	860

34 Methodological approaches were applied to the entire time-series to ensure time-series consistency from 1990
 35 through 2022.

1 Uncertainty

2 Uncertainty was estimated for the CO₂, CF₄, and C₂F₆ emission values reported by each individual facility to EPA’s
 3 GHGRP, taking into consideration the uncertainties associated with aluminum production, anode effect minutes,
 4 and slope factors. The uncertainty bounds used for these parameters were established based on information
 5 collected under the VAIP and held constant through 2022. Uncertainty surrounding the reported CO₂, CF₄, and C₂F₆
 6 emission values were determined to have a normal distribution with uncertainty ranges of approximately 6
 7 percent below to 6 percent above, 16 percent below to 16 percent above, and 20 percent below to 20 percent
 8 above their 2022 emission estimates, respectively.

9 For LVAE, since emission values were not reported through EPA’s GHGRP but estimated instead through a Tier 1
 10 methodology, the uncertainty analysis examined uncertainty associated with primary capacity data as well as
 11 technology-specific emission factors. Uncertainty for each facility’s primary capacity, reported in the USGS
 12 Yearbook, was estimated to have a Pert Beta distribution with an uncertainty range of 7 percent below to 7
 13 percent above the capacity estimates based on the uncertainty of reported capacity data, the number of years
 14 since the facility reported new capacity data, and uncertainty in capacity utilization. Uncertainty was applied to
 15 LVAE emission factors according to technology using the uncertainty ranges provided in the *2019 Refinement to*
 16 *the 2006 IPCC Guidelines*. An uncertainty range for Horizontal Stud Sjøderberg (HSS) technology was not provided
 17 in the *2019 Refinement to the 2006 IPCC Guidelines* due to insufficient data, so a normal distribution and
 18 uncertainty range of ±99 percent was applied for that technology based on expert judgment. A Monte Carlo
 19 analysis was applied to estimate the overall uncertainty of the CO₂, CF₄, and C₂F₆ emission estimates for the U.S.
 20 aluminum industry as a whole, and the results are provided below.

21 The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-85. Aluminum
 22 production-related CO₂ emissions were estimated to be between 1.41 and 1.48 MMT CO₂ Eq. at the 95 percent
 23 confidence level. This indicates a range of approximately 2 percent below to 2 percent above the emission
 24 estimate of 1.446 MMT CO₂ Eq. Also, production-related CF₄ emissions were estimated to be between 0.62 and
 25 0.73 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of approximately 8 percent below to 8
 26 percent above the emission estimate of 0.676 MMT CO₂ Eq. Aluminum production-related C₂F₆ emissions were
 27 estimated to be between 0.078 and 0.09 MMT CO₂ Eq. at the 95 percent confidence level. This indicates a range of
 28 approximately 9 percent below to 9 percent above the emission estimate of 0.083 MMT CO₂ Eq. Finally, Aluminum
 29 production-related aggregated PFCs emissions were estimated to be between 0.71 and 0.82 MMT CO₂ Eq. at the
 30 95 percent confidence level. This indicates a range of approximately 7 percent below to 7 percent above the
 31 emission estimate of 0.760 MMT CO₂ Eq.

32 **Table 4-105: Approach 2 Quantitative Uncertainty Estimates for CO₂ and PFC Emissions from**
 33 **Aluminum Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
Aluminum Production	CO ₂	1.446	1.41	1.48	-2%	+2%
Aluminum Production	CF ₄	0.676	0.62	0.73	-8%	+8%
Aluminum Production	C ₂ F ₆	0.083	0.078	0.09	-9%	+9%
Aluminum Production	PFCs	0.76	0.71	0.82	-7%	+7%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

34 QA/QC and Verification

35 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
 36 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the
 37 introduction of the IPPU chapter (see Annex 8 for more details). For the GHGRP data, EPA verifies annual facility-

1 level reports through a multi-step process (e.g., including a combination of pre-and post-submittal electronic
 2 checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are
 3 accurate, complete, and consistent (EPA 2015).⁸² Based on the results of the verification process, EPA follows up
 4 with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a
 5 number of general and category-specific QC procedures, including: range checks, statistical checks, algorithm
 6 checks, and year-to-year checks of reported data and emissions.

7 Recalculations Discussion

8 No recalculations were performed for the 1990 through 2022 portion of the time series.

9 Planned Improvements

10 No recalculations were performed for the current *Inventory*.

11 4.21 Magnesium Production and Processing 12 (CRT Source Category 2C4)

13 The magnesium metal production and casting industry uses sulfur hexafluoride (SF₆) as a cover gas to prevent the
 14 rapid oxidation of molten magnesium in the presence of air. Sulfur hexafluoride has been used in this application
 15 around the world for more than 30 years. A dilute gaseous mixture of SF₆ with dry air and/or carbon dioxide (CO₂)
 16 is blown over molten magnesium metal to induce and stabilize the formation of a protective crust. A small portion
 17 of the SF₆ reacts with the magnesium to form a thin molecular film of mostly magnesium oxide and magnesium
 18 fluoride. The amount of SF₆ reacting in magnesium production and processing is considered to be negligible and
 19 thus all SF₆ used is assumed to be emitted into the atmosphere. Alternative cover gases, such as AM-cover™
 20 (containing HFC-134a), Novec™ 612 (FK-5-1-12) and dilute sulfur dioxide (SO₂) systems can and are being used by
 21 some facilities in the United States. However, many facilities in the United States are still using traditional SF₆ cover
 22 gas systems. Carbon dioxide is also released during primary magnesium production if carbonate based raw
 23 materials, such as dolomite, are used. During the processing of these raw materials to produce magnesium,
 24 calcination occurs which results in a release of CO₂ emissions.

25 The magnesium industry emitted 1.1 MMT CO₂ Eq. (0.05 kt) of SF₆, 0.03 MMT CO₂ Eq. (0.02 kt) of HFC-134a, and
 26 0.003 MMT CO₂ Eq. (2.9 kt) of CO₂ in 2022. This represents an decrease of approximately 4 percent from total 2021
 27 emissions (see Table 4-106 and Table 4-107) and a decrease in SF₆ emissions by 3 percent. In 2022, total HFC-134a
 28 emissions decreased from 0.040 MMT CO₂ Eq. to 0.029 MMT CO₂ Eq., or a 28 percent decrease as compared to
 29 2021 emissions. FK 5-1-12 emissions in 2022 were consistent with 2021. The emissions of the carrier gas, CO₂,
 30 increased from 2.91 kt in 2021 to 2.94 kt in 2022, or 1 percent.

31 **Table 4-106: SF₆, HFC-134a, FK 5-1-12 and CO₂ Emissions from Magnesium Production and**
 32 **Processing (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
SF ₆	5.6	3.0	1.1	0.9	0.9	1.2	1.1
HFC-134a	0.0	0.0	0.1	0.1	0.1	+	+

⁸² GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

CO ₂	0.1	+	+	+	+	+	+
FK 5-1-12 ^a	0.0	0.0	+	+	+	+	+
Total	5.7	3.0	1.1	1.0	0.9	1.2	1.2

+ Does not exceed 0.05 MMT CO₂ Eq.

^aEmissions of FK 5-1-12 are not included in totals.

Note: Totals may not sum due to independent rounding.

1 **Table 4-107: SF₆, HFC-134a, FK 5-1-12 and CO₂ Emissions from Magnesium Production and**
 2 **Processing (kt)**

Year	1990	2005	2018	2019	2020	2021	2022
SF ₆	0.2	0.1	+	+	+	+	+
HFC-134a	0.0	0.0	0.1	+	+	+	+
CO ₂	129.0	3.6	1.6	2.4	3.0	2.9	2.9
FK 5-1-12 ^a	0.0	0.0	+	+	+	+	+

+ Does not exceed 0.5 kt

^aEmissions of FK 5-1-12 are not included in totals.

3 Methodology and Time-Series Consistency

4 Emission estimates for the magnesium industry incorporate information provided by industry participants in EPA's
 5 SF₆ Emission Reduction Partnership for the Magnesium Industry as well as emissions data reported through
 6 Subpart T (Magnesium Production and Processing) of EPA's GHGRP. The Partnership started in 1999 and, in 2010,
 7 participating companies represented 100 percent of U.S. primary and secondary production and 16 percent of the
 8 casting sector production (i.e., die, sand, permanent mold, wrought, and anode casting). SF₆ emissions for 1999
 9 through 2010 from primary production, secondary production (i.e., recycling), and die casting were generally
 10 reported by Partnership participants. Partners reported their SF₆ consumption, which is assumed to be equivalent
 11 to emissions. Along with SF₆, some Partners reported their HFC-134a and FK 5-1-12 consumed, which is also
 12 assumed to be equal to emissions. The last reporting year under the Partnership was 2010. Emissions data for
 13 2011 through 2020 are obtained through EPA's GHGRP. Under the program, owners or operators of facilities that
 14 have a magnesium production or casting process must report emissions from use of cover or carrier gases, which
 15 include SF₆, HFC-134a, FK 5-1-12 and CO₂. Consequently, cover and carrier gas emissions from magnesium
 16 production and processing were estimated for three time periods, depending on the source of the emissions data:
 17 1990 through 1998 (pre-EPA Partnership), 1999 through 2010 (EPA Partnership), and 2011 through 2022 (EPA
 18 GHGRP). The methodologies described below also make use of magnesium production data published by the U.S.
 19 Geological Survey (USGS) as available.

20 1990 through 1998

21 To estimate emissions for 1990 through 1998, industry SF₆ emission factors were multiplied by the corresponding
 22 metal production and consumption (casting) statistics from USGS. For this period, it was assumed that there was
 23 no use of HFC-134a or FK 5-1-12 cover gases, and hence emissions were not estimated for these alternatives.

24 Sulfur hexafluoride emission factors from 1990 through 1998 were based on a number of sources and
 25 assumptions. Emission factors for primary production were available from U.S. primary producers for 1994 and
 26 1995. The primary production emission factors were 1.2 kg SF₆ per metric ton for 1990 through 1993, and 1.1 kg
 27 SF₆ per metric ton for 1994 through 1997. The emission factor for secondary production from 1990 through 1998
 28 was assumed to be constant at the 1999 average Partner value. An emission factor for die casting of 4.1 kg SF₆ per
 29 metric ton, which was available for the mid-1990s from an international survey (Gjestland and Magers 1996), was
 30 used for years 1990 through 1996. For 1996 through 1998, the emission factor for die casting was assumed to
 31 decline linearly to the level estimated based on Partner reports in 1999. This assumption is consistent with the
 32 trend in SF₆ sales to the magnesium sector that was reported in the RAND survey of major SF₆ manufacturers,
 33 which showed a decline of 70 percent from 1996 to 1999 (RAND 2002). Sand casting emission factors for 1990

1 through 2001 were assumed to be the same as the 2002 emission factor for all but one facility, which used an
2 emission factor derived from 2011 GHGRP data and held constant to back cast emissions for 1990-1998. The
3 emission factors for the other processes (i.e., permanent mold, wrought, and anode casting), about which less is
4 known, were assumed to remain constant at levels defined in Table 4-107. The emission factors for the other
5 processes (i.e., permanent mold, wrought, and anode casting) were based on discussions with industry
6 representatives.

7 The quantities of CO₂ carrier gas used for each production type have been estimated using the 1999 estimated CO₂
8 emissions data and the annual calculated rate of change of SF₆ use in the 1990 through 1999 time period. For each
9 year and production type, the rate of change of SF₆ use between the current year and the subsequent year was
10 first estimated. This rate of change was then applied to the CO₂ emissions of the subsequent year to determine the
11 CO₂ emission of the current year.

12 Carbon dioxide emissions from the calcination of dolomite in the primary production of magnesium were
13 calculated based on the *2006 IPCC Guidelines* Tier 2 method by multiplying the estimated primary production of
14 magnesium by an emissions factor of 3.62 kilogram of CO₂ per kilogram of magnesium produced.⁸³ For 1990
15 through 1998, production was estimated to be equal to the production capacity of the facility.

16 **1999 through 2010**

17 The 1999 through 2010 emissions from primary and secondary production were based on information provided by
18 EPA's industry Partners. In some instances, there were years of missing Partner data, including SF₆ consumption
19 and metal processed. For these situations, emissions were estimated through interpolation where possible, or by
20 holding company-reported emissions (as well as production) constant from the previous year. For alternative cover
21 gases, including HFC-134a and FK 5-1-12, mainly reported data was relied upon. That is, unless a Partner reported
22 using an alternative cover gas, it was not assumed it was used. Emissions of alternate gases were also estimated
23 through linear interpolation where possible.

24 The die casting emission estimates for 1999 through 2010 were also based on information supplied by industry
25 Partners. When a Partner was determined to be no longer in production, its metal production and usage rates
26 were set to zero. Missing data on emissions or metal input was either interpolated or held constant at the last
27 available reported value. In 1999 through 2010, Partners were assumed to account for all die casting tracked by
28 USGS. For 1999, die casters who were not Partners were assumed to be similar to Partners who cast small parts.
29 Due to process requirements, these casters consume larger quantities of SF₆ per metric ton of processed
30 magnesium than casters that process large parts. Consequently, emission estimates from this group of die casters
31 were developed using an average emission factor of 5.2 kg SF₆ per metric ton of magnesium. This emission factor
32 was developed using magnesium production and SF₆ usage data for the year 1999. In 2008, the derived emission
33 factor for die casting began to increase after many years of largely decreasing emission factors. As determined
34 through an analysis of activity data reported from the USGS, this increase is due to a temporary decrease in
35 production at many facilities between 2008 and 2010, which reflects the change in production that occurred
36 during the recession.

37 The emissions from other casting operations were estimated by multiplying emission factors (kg SF₆ per metric ton
38 of metal produced or processed) by the amount of metal produced or consumed from USGS, with the exception of
39 some years for which Partner sand casting emissions data are available. The emission factors for sand casting
40 activities were acquired through the data reported by the Partnership for 2002 to 2006. For 1999 through 2001,
41 the sandcasting emission factor was held constant at the 2002 Partner-reported level. For 2007 through 2010, the
42 sandcasting Partner did not report and the reported emission factor from 2005 was applied to the Partner and to a
43 non GHGRP sand casters. Activity data for 2005 was obtained from USGS (USGS 2005b). One non partner sand

⁸³ See https://www.ipcc-nggip.iges.or.jp/public/2006gl/pdf/3_Volume3/V3_4_Ch4_Metal_Industry.pdf.

1 casting facility reported to GHGRP in 2011 and had an emission factor derived for 2011, this factor was used to
2 back cast emissions for this facility from 1999-2010.

3 The emission factors for primary production, secondary production and sand casting for the 1999 to 2010 are not
4 published to protect company-specific production information. However, the emission factor for primary
5 production has not risen above the average 1995 Partner value of 1.1 kg SF₆ per metric ton. The emission factors
6 for the other industry sectors (i.e., permanent mold, wrought, and anode casting) were based on discussions with
7 industry representatives. The emission factors for casting activities are provided below in Table 4-108.

8 The emissions of HFC-134a and FK-5-1-12 were included in the estimates for only instances where Partners
9 reported that information to the Partnership. Emissions of these alternative cover gases were not estimated for
10 instances where emissions were not reported.

11 Carbon dioxide carrier gas emissions were estimated using the emission factors developed based on GHGRP-
12 reported carrier gas and cover gas data, by production type. It was assumed that the use of carrier gas, by
13 production type, is proportional to the use of cover gases. Therefore, an emission factor, in kg CO₂ per kg cover gas
14 and weighted by the cover gases used, was developed for each of the production types. GHGRP data, on which
15 these emissions factors are based, was available for primary, secondary, die casting and sand casting. The emission
16 factors were applied to the quantity of all cover gases used (SF₆, HFC-134a, and FK-5-1-12) by production type in
17 this time period for producers that reported CO₂ emissions from 2011-2022 through the GHGP. Carrier gas
18 emissions for the 1999 through 2010 time period were only estimated for those Partner companies that reported
19 using CO₂ as a carrier gas through the GHGRP. Using this approach helped ensure time-series consistency.
20 Emissions of carrier gases for permanent mold, wrought, and anode processes were estimated using the ratio of
21 total CO₂ emissions to total cover gas emissions for primary, secondary, die and sand in a given year and the total
22 SF₆ emissions from each permanent mold, wrought, and anodes processes respectively in that same year. CO₂
23 emissions from the calcination of dolomite were estimated using the same approach as described above. At the
24 end of 2001, the sole magnesium production plant operating in the United States that produced magnesium metal
25 using a dolomitic process that resulted in the release of CO₂ emissions ceased its operations (USGS 1995b through
26 2023).

27 **Table 4-108: SF₆ Emission Factors (kg SF₆ per metric ton of magnesium)**

Year	Die Casting ^a	Permanent Mold	Wrought	Anodes
1999	1.75 ^b	2	1	1
2000	0.72	2	1	1
2001	0.72	2	1	1
2002	0.71	2	1	1
2003	0.81	2	1	1
2004	0.79	2	1	1
2005	0.77	2	1	1
2006	0.88	2	1	1
2007	0.64	2	1	1
2008	0.97	2	1	1
2009	1.41	2	1	1
2010	1.43	2	1	1

^a Weighted average includes all die casters, Partners and non-Partners. For the majority of the time series (2000 through 2010), Partners made up 100 percent of die casters in the United States.

^b Weighted average that includes an estimated emission factor of 5.2 kg SF₆ per metric ton of magnesium for die casters that do not participate in the Partnership.

2011 through 2022

For 2011 through 2022, for the primary and secondary producers, GHGRP-reported cover and carrier gases emissions data were used. For sand and die casting, some emissions data was obtained through EPA's GHGRP. Additionally, in 2018 a new GHGRP reporter began reporting permanent mold emissions. The balance of the emissions for this industry segment was estimated based on previous Partner reporting (i.e., for Partners that did not report emissions through EPA's GHGRP) or were estimated by multiplying emission factors by the amount of metal produced or consumed. Partners who did not report through EPA's GHGRP were assumed to have continued to emit SF₆ at the last reported level, which was from 2010 in most cases, unless publicly available sources indicated that these facilities have closed or otherwise eliminated SF₆ emissions from magnesium production (ARB 2015). Many Partners that did report through the GHGRP showed increases in SF₆ emissions driven by increased production related to a continued economic recovery after the 2008 recession. One Partner in particular reported an anonymously large increase in SF₆ emissions from 2010 to 2011, further driving increases in emissions between the two time periods of inventory estimates. All Partners were assumed to have continued to consume magnesium at the last reported level. Where the total metal consumption estimated for the Partners fell below the U.S. total reported by USGS, the difference was multiplied by the emission factors discussed in the section above, i.e., non-partner emission factors. For the other types of production and processing (i.e., permanent mold, wrought, and anode casting), emissions were estimated by multiplying the industry emission factors with the metal production or consumption statistics obtained from USGS (USGS 1995b-2023). USGS data for 2022 were not yet available at the time of the analysis, so the 2021 values were held constant through 2022 as an estimate.

Emissions of carrier gases for permanent mold, wrought, and anode processes were estimated using an approach consistent with the 1999 through 2010 time series.

Methodological approaches were applied to the entire time series to ensure time-series consistency from 1990 through 2022. 2006 IPCC Guidance methodologies were used throughout the timeseries, mainly either a Tier 2 or Tier 3 approach depending on available data.

Uncertainty

Uncertainty surrounding the total estimated emissions in 2022 is attributed to the uncertainties around SF₆, HFC-134a, and CO₂ emission estimates. To estimate the uncertainty surrounding the estimated 2022 SF₆ emissions from magnesium production and processing, the uncertainties associated with three variables were estimated: (1) emissions reported by magnesium producers and processors for 2022 through EPA's GHGRP, (2) emissions estimated for magnesium producers and processors that reported via the Partnership in prior years but did not report 2022 emissions through EPA's GHGRP, and (3) emissions estimated for magnesium producers and processors that did not participate in the Partnership or report through EPA's GHGRP. An uncertainty of 5 percent was assigned to the emissions (usage) data reported by each GHGRP reporter for all the cover and carrier gases (per the 2006 IPCC Guidelines). If facilities did not report emissions data during the current reporting year through EPA's GHGRP, SF₆ emissions data were held constant at the most recent available value reported through the Partnership. The uncertainty associated with these values was estimated to be 30 percent for each year of extrapolation (per the 2006 IPCC Guidelines). The uncertainty of the total inventory estimate remained relatively constant between 2021 and 2022.

Alternate cover gas and carrier gases data was set equal to zero if the facilities did not report via the GHGRP. For those industry processes that are not represented in the Partnership, such as permanent mold and wrought casting, SF₆ emissions were estimated using production and consumption statistics reported by USGS and estimated process-specific emission factors (see Table 4-108). The uncertainties associated with the emission factors and USGS-reported statistics were assumed to be 75 percent and 25 percent, respectively. Emissions associated with die casting and sand casting activities utilized emission factors based on Partner reported data with an uncertainty of 75 percent. In general, where precise quantitative information was not available on the uncertainty of a parameter, a conservative (upper-bound) value was used.

1 Additional uncertainties exist in these estimates that are not addressed in this methodology, such as the basic
 2 assumption that SF₆ neither reacts nor decomposes during use. The melt surface reactions and high temperatures
 3 associated with molten magnesium could potentially cause some gas degradation. Previous measurement studies
 4 have identified SF₆ cover gas degradation in die casting applications on the order of 20 percent (Bartos et al. 2007).
 5 Sulfur hexafluoride may also be used as a cover gas for the casting of molten aluminum with high magnesium
 6 content; however, the extent to which this technique is used in the United States is unknown.

7 The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-109. Total emissions
 8 associated with magnesium production and processing were estimated to be between 1.06 and 1.24 MMT CO₂ Eq.
 9 at the 95 percent confidence level. This indicates a range of approximately 7.7 percent below to 8.0 percent above
 10 the 2022 emission estimate of 1.52 MMT CO₂ Eq. The uncertainty estimates for 2022 are slightly higher to the
 11 uncertainty reported for 2021 in the previous *Inventory*. This increase in uncertainty is attributed to the increased
 12 number of facilities with interpolated emissions and the increasing number of years for facilities with emissions
 13 held constant.

14 **Table 4-109: Approach 2 Quantitative Uncertainty Estimates for SF₆, HFC-134a and CO₂**
 15 **Emissions from Magnesium Production and Processing (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Magnesium Production	SF ₆ , HFC- 134a, CO ₂	1.5	1.1	1.2	-7.7%	+8.0%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

16 QA/QC and Verification

17 General quality assurance/quality control (QA/QC) procedures were applied consistent with the U.S. Inventory
 18 QA/QC plan, which is in accordance with Volume 1, Chapter 6 of *2006 IPCC Guidelines* as described in the
 19 introduction of the IPPU chapter (see Annex 8 for more details). For the GHGRP data, EPA verifies annual facility-
 20 level reports through a multi-step process (e.g., including a combination of pre-and post-submittal electronic
 21 checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are
 22 accurate, complete, and consistent (EPA 2015).⁸⁴ Based on the results of the verification process, EPA follows up
 23 with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a
 24 number of general and category-specific QC procedures, including: range checks, statistical checks, algorithm
 25 checks, and year-to-year checks of reported data and emissions.

26 Recalculations Discussion

27 One Die Casting facility which had previously had emissions back cast at a constant level had its back casting
 28 methodology updated using linear growth from 0 to reported emissions levels between 2001 and 2014, resulting in
 29 decreases in SF₆ emissions across 2001 to 2013.

30 Sand Casting Emissions for 2021 were updated based on 2021 specific data available in the 2021 data tables
 31 release from USGS's Mineral Yearbook. 2021 data was previously held constant at 2020 levels due to USGS Mineral

⁸⁴ GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 Yearbook data only going through 2020. The updated production of sand cast magnesium was larger than what
2 was estimated for 2021 in the previous *Inventory* cycle leading to an increase in SF₆ emissions in 2021.

3 One sand casting facility, which had previously only been estimated from 2011 onward, was confirmed to have
4 emissions across the time series, an updated emission factor for 2011 was calculated and used to back cast
5 emissions from 1990 to 2010.

6 Review of facility responses indicate that changes over time in the emission factors for this industry have occurred
7 as facilities switch to using systems with cover gases other than SF₆ (e.g. SO₂) and also during time-periods where
8 back-up SF₆-based systems are used due to the failure of the primary (non-SF₆) system have occurred, leading to
9 the periodic spike in SF₆ usage rates.

10 **Planned Improvements**

11 Cover gas research conducted over the last decade has found that SF₆ used for magnesium melt protection can
12 have degradation rates on the order of 20 percent in die casting applications (Bartos et al. 2007). Current emission
13 estimates assume (per the *2006 IPCC Guidelines*) that all SF₆ utilized is emitted to the atmosphere. Additional
14 research may lead to a revision of the *2006 IPCC Guidelines* to reflect this phenomenon and until such time,
15 developments in this sector will be monitored for possible application to the *Inventory* methodology.

16 Additional emissions are generated as byproducts from the use of alternate cover gases, which are not currently
17 accounted for. Research on this topic is developing, and as reliable emission factors become available, these
18 emissions will be incorporated into the *Inventory*.

19 **4.22 Lead Production (CRT Source Category** 20 **2C5)**

21 In 2022, lead was produced in the United States using only secondary production processes. Until 2014, lead
22 production in the United States involved both primary and secondary processes—both of which emit carbon
23 dioxide (CO₂) (Sjardin 2003). Emissions from fuels consumed for energy purposes during the production of lead are
24 accounted for in the Energy chapter.

25 Primary production of lead through the direct smelting of lead concentrate produces CO₂ emissions as the lead
26 concentrates are reduced in a furnace using metallurgical coke (Sjardin 2003). Primary lead production, in the form
27 of direct smelting, previously occurred at a single smelter in Missouri. This primary lead smelter was closed at the
28 end of 2013, and a small amount of residual lead was processed during demolition of the facility in 2014 (USGS
29 2015). Beginning in 2015, primary lead production no longer occurred in the United States.

30 Similar to primary lead production, CO₂ emissions from secondary lead production result when a reducing agent,
31 usually metallurgical coke, is added to the smelter to aid in the reduction process. Carbon dioxide emissions from
32 secondary production also occur through the treatment of secondary raw materials (Sjardin 2003). Secondary
33 production primarily involves the recycling of lead acid batteries and post-consumer scrap at secondary smelters.
34 Secondary lead production in the United States has fluctuated over the past 20 years, reaching a high of 1,180,000
35 metric tons in 2007 and again in 2019. In 2022, secondary lead production accounted for 100 percent of total U.S.
36 lead production. The lead-acid battery industry accounted for about 92 percent of the reported U.S. lead
37 consumption in 2022 (USGS 2023a).

38 In 2022, secondary lead production in the United States decreased by approximately 3 percent compared to 2021
39 (USGS 2023a). Secondary lead production in 2022 is 3 percent higher than in 1990 (USGS 1994 and 2023). The
40 United States has become more reliant on imported refined lead, owing to the closure of the last primary lead
41 smelter in 2013. Exports of spent starting-lighting-ignition (SLI) batteries decreased between 2014 and 2017, and

1 subsequently recovered beginning in 2018. Exports were 10 percent higher in the first 9 months of 2021 compared
 2 to the same time period in 2014 (USGS 2015 through 2023b). In the first 9 months of 2022, 24.6 million spent SLI
 3 lead-acid batteries were exported, 4 percent less than that in the same time period in 2021 (USGS 2023c).

4 Emissions of CO₂ from lead production in 2022 were 0.4 MMT CO₂ Eq. (428 kt), which is a 3 percent decrease
 5 compared to 2021 and a 17 percent decrease compared to 1990 (see Table 4-110 and Table 4-111) (USGS 2023a,
 6 USGS 2023b).

7 The United States was the third largest mine producer of lead in the world, behind China and Australia, and
 8 accounted for approximately 6 percent of world production in 2022 (USGS 2023a).

9 **Table 4-110: CO₂ Emissions from Lead Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Lead Production	0.5	0.6	0.5	0.5	0.5	0.4	0.4

10 **Table 4-111: CO₂ Emissions from Lead Production (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Lead Production	516	553	527	531	450	439	428

11 Methodology and Time-Series Consistency

12 Carbon dioxide emissions from lead production⁸⁵ are calculated based on Sjardin’s work (Sjardin 2003) for lead
 13 production emissions and use Tier 1 methods from the *2006 IPCC Guidelines*, in accordance with the IPCC
 14 methodological decision tree and available data. The Tier 1 equation is as follows:

15 **Equation 4-16: 2006 IPCC Guidelines Tier 1: CO₂ Emissions From Lead Production (Equation**
 16 **4.32)**

$$17 \quad CO_2 \text{ Emissions} = (DS \times EF_{DS}) + (S \times EF_S)$$

18 where,

19 DS = Lead produced by direct smelting, metric ton

20 S = Lead produced from secondary materials

21 EF_{DS} = Emission factor for direct smelting, metric tons CO₂/metric ton lead product

22 EF_S = Emission factor for secondary materials, metric tons CO₂/metric ton lead product

23 For primary lead production using direct smelting, Sjardin (2003) and the *2006 IPCC Guidelines* provide an emission
 24 factor of 0.25 metric tons CO₂/metric ton lead. For secondary lead production, Sjardin (2003) and the *2006 IPCC*
 25 *Guidelines* provide an emission factor of 0.25 metric tons CO₂/metric ton lead for direct smelting, as well as an
 26 emission factor of 0.2 metric tons CO₂/metric ton lead produced for the treatment of secondary raw materials (i.e.,
 27 pretreatment of lead acid batteries). Since the secondary production of lead involves both the use of the direct
 28 smelting process and the treatment of secondary raw materials, Sjardin recommends an additive emission factor
 29 to be used in conjunction with the secondary lead production quantity. The direct smelting factor (0.25) and the
 30 sum of the direct smelting and pretreatment emission factors (0.45) are multiplied by total U.S. primary and
 31 secondary lead production, respectively, to estimate CO₂ emissions.

⁸⁵ EPA has not integrated aggregated facility-level Greenhouse Gas Reporting Program (GHGRP) information to inform these estimates. The aggregated information (e.g., activity data and emissions) associated with Lead Production did not meet criteria to shield underlying confidential business information (CBI) from public disclosure.

1 The production and use of coking coal for lead production is adjusted for within the Energy chapter as this fuel was
 2 consumed during non-energy related activities. Additional information on the adjustments made within the Energy
 3 sector for Non-Energy Use of Fuels is described in both the Methodology section of CO₂ from Fossil Fuel
 4 Combustion (Section 3.1 Fossil Fuel Combustion (CRT Source Category 1A)) and Annex 2.1, Methodology for
 5 Estimating Emissions of CO₂ from Fossil Fuel Combustion.

6 The 1990 through 2022 activity data for primary and secondary lead production (see Table 4-112) were obtained
 7 from the U.S. Geological Survey (USGS 1995 through 2023a).

8 **Table 4-112: Lead Production (Metric Tons)**

Year	1990	2005	2018	2019	2020	2021	2022
Primary	404,000	143,000	0	0	0	0	0
Secondary	922,000	1,150,000	1,170,000	1,180,000	1,000,000	975,000	950,000

9 Methodological approaches discussed below were applied to applicable years to ensure time-series consistency in
 10 emissions from 1990 through 2022.

11 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

12 Uncertainty associated with lead production relates to the emission factors and activity data used. The direct
 13 smelting emission factor used in primary production is taken from Sjardin (2003) who averaged the values
 14 provided by three other studies (Dutrizac et al. 2000; Morris et al. 1983; Ullman 1997). For secondary production,
 15 Sjardin (2003) added a CO₂ emission factor associated with battery treatment. The applicability of these emission
 16 factors to plants in the United States is uncertain. EPA assigned an uncertainty range of ±20 percent for these
 17 emission factors, and using this suggested uncertainty provided in Table 4.23 of the *2006 IPCC Guidelines* for a Tier
 18 1 emission factor by process type is appropriate based on expert judgment (RTI 2023).

19 There is also a smaller level of uncertainty associated with the accuracy of primary and secondary production data
 20 provided by the USGS which is collected via voluntary surveys; the uncertainty of the activity data is a function of
 21 the reliability of reported plant-level production data and the completeness of the survey response. EPA currently
 22 uses an uncertainty range of ±10 percent for primary and secondary lead production, and using this suggested
 23 uncertainty provided in Table 4.23 of the *2006 IPCC Guidelines* for Tier 1 national production data is appropriate
 24 based on expert judgment (RTI 2023).

25 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-113. Lead production
 26 CO₂ emissions in 2022 were estimated to be between 0.4 and 0.5 MMT CO₂ Eq. at the 95 percent confidence level.
 27 This indicates a range of approximately 15 percent below and 15 percent above the emission estimate of 0.4 MMT
 28 CO₂ Eq.

29 **Table 4-113: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Lead
 30 Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.)			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
Lead Production	CO ₂	0.4	0.4	0.5	-15%	+15%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

1 QA/QC and Verification

2 For more information on the general QA/QC process applied to this source category, consistent with Volume 1,
3 Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in the introduction of
4 the IPPU chapter.

5 Initial review of activity data show that EPA's GHGRP Subpart R lead production data and resulting emissions are
6 fairly consistent with those reported by USGS. EPA is still reviewing available GHGRP data, reviewing QC analysis to
7 understand differences in data reporting (i.e., threshold implications), and assessing the possibility of including this
8 planned improvement in future *Inventory* reports (see Planned Improvements section below). Currently, GHGRP
9 data are used for QA purposes only.

10 Recalculations Discussion

11 Recalculations were implemented for 2020 and 2021 based on revised USGS data for secondary lead production.
12 Compared to the previous *Inventory*, emissions decreased by 3 percent (14 kt CO₂) for 2020 and by 2 percent (7 kt
13 CO₂) for 2021 (USGS 2023b).

14 Planned Improvements

15 Pending resources and prioritization of improvements for more significant sources, EPA will continue to evaluate
16 and analyze data reported under EPA's GHGRP that would be useful to improve the emission estimates and
17 category-specific QC for the Lead Production source category, in particular considering completeness of reported
18 lead production given the reporting threshold. Particular attention will be made to ensuring time-series
19 consistency of the emissions estimates presented in future *Inventory* reports, consistent with IPCC and UNFCCC
20 guidelines. This is required as the facility-level reporting data from EPA's GHGRP, with the program's initial
21 requirements for reporting of emissions in calendar year 2010, are not available for all inventory years (i.e., 1990
22 through 2009) as required for this *Inventory*. In implementing improvements and integration of data from EPA's
23 GHGRP, the latest guidance from the IPCC on the use of facility-level data in national inventories will be relied
24 upon.⁸⁶

25 4.23 Zinc Production (CRT Source Category 26 2C6)

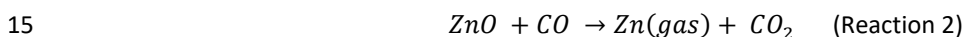
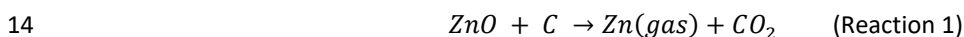
27 Zinc production in the United States consists of both primary and secondary processes. Of the primary and
28 secondary processes currently used in the United States, only the electrothermic and Waelz kiln secondary
29 processes result in non-energy carbon dioxide (CO₂) emissions (Viklund-White 2000). Emissions from fuels
30 consumed for energy purposes during the production of zinc are accounted for in the Energy chapter.

31 The majority of zinc produced in the United States is used for galvanizing. Galvanizing is a process where zinc
32 coating is applied to steel in order to prevent corrosion. Zinc is used extensively for galvanizing operations in the
33 automotive and construction industry. Zinc is also used in the production of zinc alloys and brass and bronze alloys
34 (e.g., brass mills, copper foundries, and copper ingot manufacturing). Zinc compounds and dust are also used, to a
35 lesser extent, by the agriculture, chemicals, paint, and rubber industries.

⁸⁶ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

1 Production of zinc can be conducted with a range of pyrometallurgical (e.g., electrothermic furnace, Waelz kiln,
2 flame reactor, batch retorts, Pinto process, and PIZO process) and hydrometallurgical (e.g., hydrometallurgical
3 recovery, solvent recovery, solvent extraction-electrowinning, and electrolytic) processes. Hydrometallurgical
4 production processes are assumed to be non-emissive since no carbon is used in these processes (Sjardin 2003).
5 Primary production in the United States is conducted through the non-emissive electrolytic process, while
6 secondary techniques include the electrothermic and Waelz kiln processes, as well as a range of other processes.
7 Worldwide primary zinc production also employs a pyrometallurgical process using an Imperial Smelting Furnace;
8 however, this process is not used in the United States (Sjardin 2003).

9 In the electrothermic process, roasted zinc concentrate and secondary zinc products enter a sinter feed where
10 they are burned to remove impurities before entering an electric retort furnace. Metallurgical coke is added to the
11 electric retort furnace as a carbon-containing reductant. This concentration step, using metallurgical coke and high
12 temperatures, reduces the zinc oxides and produces vaporized zinc, which is then captured in a vacuum
13 condenser. This reduction process also generates non-energy CO₂ emissions.



16 In the Waelz kiln process, electric arc furnace (EAF) dust, which is captured during the recycling of galvanized steel,
17 enters a kiln along with a reducing agent (typically carbon-containing metallurgical coke). When kiln temperatures
18 reach approximately 1,100 to 1,200 degrees Celsius, zinc fumes are produced, which are combusted with air
19 entering the kiln. This combustion forms zinc oxide, which is collected in a baghouse or electrostatic precipitator,
20 and is then leached to remove chloride and fluoride. The use of carbon-containing metallurgical coke in a high-
21 temperature fuming process results in non-energy CO₂ emissions. Through this process, approximately 0.33 metric
22 tons of zinc is produced for every metric ton of EAF dust treated (Viklund-White 2000).

23 In the flame reactor process, a waste feed stream, which can include EAF dust, is processed in a high-temperature
24 environment (greater than 2,000 °C) created by the combustion of natural gas or coal and oxygen-enriched air.
25 Volatile metals, including zinc, are forced into the gas phase and drawn into a combustion chamber, where air is
26 introduced and oxidation occurs. The metal oxide product is then collected in a dust collection system (EPA 1992).

27 In 2022, the only companies in the United States that used emissive technology to produce secondary zinc
28 products were Befesa Holding US Inc (Befesa) and Steel Dust Recycling (SDR). The secondary zinc facilities operated
29 by Befesa were acquired from American Zinc Recycling (AZR) (formerly "Horsehead Corporation") in 2021. PIZO
30 Operating Company, LLC (PIZO) operated a secondary zinc production facility that processed EAF dust in
31 Blytheville, AR from 2009 to 2012.

32 For Befesa, EAF dust is recycled in Waelz kilns at their Calumet, IL; Palmerton, PA; Rockwood, TN; and Barnwell, SC
33 facilities. The former AZR facility in Beaumont, TX processed EAF dust via flame reactor from 1993 through 2009
34 (AZR 2021, Horsehead 2014). These Waelz kiln and flame reactor facilities produce intermediate zinc products
35 (crude zinc oxide or calcine). Prior to 2014, most of output from these facilities were transported to their Monaca,
36 PA facility where the products were smelted into refined zinc using electrothermic technology. In April 2014, the
37 Monaca smelter was permanently closed and replaced by a new facility in Mooresboro, NC in 2014.

38 The Mooresboro facility uses a hydrometallurgical process (i.e., solvent extraction with electrowinning technology)
39 to produce zinc products, which is assumed to be non-emissive as described above. Production at the Mooresboro
40 facility was idled in April 2016 and re-started in March 2020 (Recycling Today 2020). Direct consumption of coal,
41 coke, and natural gas were replaced with electricity consumption (Horsehead 2012b). The Mooresboro facility uses
42 leaching and solvent extraction (SX) technology combined with electrowinning, melting, and casting technology. In
43 this process, Waelz Oxide (WOX) is first washed in water to remove soluble elements such as chlorine, potassium,
44 and sodium, and then is leached in a sulfuric acid solution to dissolve the contained zinc creating a pregnant liquor
45 solution (PLS). The PLS is then processed in a solvent extraction step in which zinc is selectively extracted from the
46 PLS using an organic solvent creating a purified zinc-loaded electrolyte solution. The loaded electrolyte solution is
47 then fed into the electrowinning process in which electrical energy is applied across a series of anodes and

1 cathodes submerged in the electrolyte solution causing the zinc to deposit on the surfaces of the cathodes. As the
 2 zinc metal builds up on these surfaces, the cathodes are periodically harvested in order to strip the zinc from their
 3 surfaces (Horsehead 2015).

4 SDR recycles EAF dust into intermediate zinc products using Waelz kilns and sells the intermediate products to
 5 companies who smelt it into refined products.

6 Emissions of CO₂ from zinc production in 2022 were estimated to be 0.9 MMT CO₂ Eq. (947 kt CO₂) (see Table
 7 4-114). All 2022 CO₂ emissions resulted from secondary zinc production processes. Emissions from zinc production
 8 in the United States have increased overall since 1990 due to a gradual shift from non-emissive primary production
 9 to emissive secondary production. In 2022, emissions were estimated to be 50 percent higher than they were in
 10 1990. Emissions decreased 6 percent from 2021 levels.

11 **Table 4-114: CO₂ Emissions from Zinc Production (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
Zinc Production	0.6	1.0	1.0	1.0	1.0	1.0	0.9

12 **Table 4-115: CO₂ Emissions from Zinc Production (kt CO₂)**

Year	1990	2005	2018	2019	2020	2021	2022
Zinc Production	632	1,030	999	1,026	977	1,007	947

13 U.S. zinc mine production increased by 9 percent in 2022 compared to 2021, due in part to higher mill throughput
 14 and zinc ore grades at the Red Dog Mine in Alaska, the largest zinc mine in the United States. In 2022, United
 15 States primary and secondary refined zinc production were estimated to total 220,000 metric tons (USGS 2023)
 16 (see Table 4-116), remaining at approximately the same production level as in 2021. Secondary zinc production fell
 17 to its lowest point in the time series in 2019, following the closure of the Monaca, PA smelter in 2014 and issues
 18 with the AZR secondary zinc refinery in Mooresboro, NC. Secondary zinc production has increased significantly
 19 since the reopening of the idled Mooresboro facility in March 2020 (USGS 2021; AZP 2021).

20 **Table 4-116: Zinc Production (Metric Tons)**

Year	1990	2005	2018	2019	2020	2021	2022
Primary	262,704	191,120	101,000	101,000	101,000	101,000	101,000
Secondary	95,708	156,000	15,000	14,000	79,000	119,000	119,000
Total	358,412	347,120	116,000	115,000	180,000	220,000	220,000

21 Methodology and Time-Series Consistency

22 Emissions of CO₂ emissions from zinc production⁸⁷ using the electrothermic primary production and Waelz kiln
 23 secondary production processes are calculated using a Tier 1 method from the *2006 IPCC Guidelines*, in accordance
 24 with the IPCC methodological decision tree and available data (IPCC 2006). The Tier 1 equation used to estimate
 25 emissions from zinc production is as follows:

26 **Equation 4-17: 2006 IPCC Guidelines Tier 1: CO₂ Emissions from Zinc Production (Equation**
 27 **4.33)**

28
$$E_{CO_2} = Zn \times EF_{default}$$

⁸⁷ EPA has not integrated aggregated facility-level Greenhouse Gas Reporting Program (GHGRP) information to inform these estimates. The aggregated information (e.g., activity data and emissions) associated with Zinc Production did not meet criteria to shield underlying confidential business information (CBI) from public disclosure.

1 where,

2 E_{CO_2} = CO₂ emissions from zinc production, metric tons

3 Zn = Quantity of zinc produced, metric tons

4 $EF_{default}$ = Default emission factor, metric tons CO₂/metric ton zinc produced

5 The Tier 1 emission factors provided by IPCC for Waelz kiln-based secondary production were derived from
6 metallurgical coke consumption factors and other data presented in Viklund-White (2000). These coke
7 consumption factors as well as other inputs used to develop the Waelz kiln emission factors are shown below. IPCC
8 does not provide an emission factor for electrothermic processes due to limited information; therefore, the Waelz
9 kiln-specific emission factors were also applied to zinc produced from electrothermic processes. Starting in 2014,
10 refined zinc produced in the United States used hydrometallurgical processes and is assumed to be non-emissive.

11 For Waelz kiln-based production, IPCC recommends the use of emission factors based on EAF dust consumption, if
12 possible, rather than the amount of zinc produced since the amount of reduction materials used is more directly
13 dependent on the amount of EAF dust consumed. Since only a portion of emissive zinc production facilities
14 consume EAF dust, the emission factor based on zinc production is applied to the non-EAF dust consuming
15 facilities, while the emission factor based on EAF dust consumption is applied to EAF dust consuming facilities.

16 The Waelz kiln emission factor based on the amount of zinc produced was developed based on the amount of
17 metallurgical coke consumed for non-energy purposes per ton of zinc produced (i.e., 1.19 metric tons coke/metric
18 ton zinc produced) (Viklund-White 2000), and the following equation:

19 **Equation 4-18: Waelz Kiln CO₂ Emission Factor for Zinc Produced**

$$20 \quad EF_{Waelz \ Kiln} = \frac{1.19 \text{ metric tons coke}}{\text{metric tons zinc}} \times \frac{0.85 \text{ metric tons C}}{\text{metric tons coke}} \times \frac{3.67 \text{ metric tons CO}_2}{\text{metric tons C}} = \frac{3.70 \text{ metric tons CO}_2}{\text{metric tons zinc}}$$

21 Refined zinc production levels for AZR's Monaca, PA facility (utilizing electrothermic technology) were available
22 from the company for years 2005 through 2013 (Horsehead 2008, 2011, 2012, 2013, and 2014). The Monaca
23 facility was permanently shut down in April 2014 and replaced by AZR's new facility in Mooresboro, NC. The new
24 facility uses hydrometallurgical process to produce refined zinc products. Hydrometallurgical production processes
25 are assumed to be non-emissive since no carbon is used in these processes (Sjardin 2003).

26 Metallurgical coke consumption for non-EAF dust consuming facilities for 1990 through 2004 were extrapolated
27 using the percentage change in annual refined zinc production at secondary smelters in the United States, as
28 provided by the U.S. Geological Survey (USGS) *Minerals Yearbook: Zinc* (USGS 1995 through 2006). Metallurgical
29 coke consumption for 2005 through 2013 were based on the secondary zinc production values obtained from the
30 Horsehead Corporation Annual Report Form 10-K: 2005 through 2008 from the 2008 10-K (Horsehead Corp 2009);
31 2009 and 2010 from the 2010 10-K (Horsehead Corp. 2011); and 2011 through 2013 from the associated 10-K
32 (Horsehead Corp. 2012a, 2013, 2014). Metallurgical coke consumption levels for 2014 and later were zero due to
33 the closure of the AZR (formerly "Horsehead Corporation") electrothermic furnace facility in Monaca, PA. The
34 secondary zinc produced values for each year were then multiplied by the 3.70 metric tons CO₂/metric ton zinc
35 produced emission factor to develop CO₂ emission estimates for the AZR electrothermic furnace facility.

36 The Waelz kiln emission factor based on the amount of EAF dust consumed was developed based on the amount
37 of metallurgical coke consumed per ton of EAF dust consumed (i.e., 0.4 metric tons coke/metric ton EAF dust
38 consumed) (Viklund-White 2000), and the following equation:

39 **Equation 4-19: Waelz Kiln CO₂ Emission Factor for EAF Dust Consumed**

$$40 \quad EF_{EAF \ Dust} = \frac{0.4 \text{ metric tons coke}}{\text{metric tons EAF Dust}} \times \frac{0.85 \text{ metric tons C}}{\text{metric tons coke}} \times \frac{3.67 \text{ metric tons CO}_2}{\text{metric tons C}} = \frac{1.24 \text{ metric tons CO}_2}{\text{metric tons EAF Dust}}$$

1 Metallurgical coke consumption for EAF dust consuming facilities for 1990 through 2022 were calculated based on
2 the values of EAF dust consumed. The total amount of EAF dust consumed by the Waelz kilns currently operated
3 by Befesa was available from AZR (formerly “Horsehead Corporation”) in financial reports for years 2006 through
4 2015 (Horsehead 2007, 2008, 2010a, 2011, 2012a, 2013, 2014, 2015, and 2016), from correspondence with AZR for
5 2016 through 2019 (AZR 2020), and from correspondence with Befesa for 2020 through 2022 (Befesa 2022, 2023).
6 The EAF dust consumption values for each year were then multiplied by the 1.24 metric tons CO₂/metric ton EAF
7 dust consumed emission factor to develop CO₂ emission estimates for Befesa’s Waelz kiln facilities.

8 The amount of EAF dust consumed by SDR and their total production capacity were obtained from SDR’s facility in
9 Alabama for the years 2011 through 2022 (SDR 2012, 2014, 2015, 2017, 2018, 2021, 2022, 2023). The SDR facility
10 has been operational since 2008, underwent expansion in 2011 to include a second unit (operational since early- to
11 mid-2012), and expanded its capacity again in 2017 (SDR 2018). Annual consumption data for SDR was not publicly
12 available for the years 2008, 2009, and 2010. These data were estimated using data for AZR’s Waelz kilns for 2008
13 through 2010 (Horsehead 2007, 2008, 2010a, 2010b, 2011). Annual capacity utilization ratios were calculated using
14 AZR’s annual consumption and total capacity for the years 2008 through 2010. AZR’s annual capacity utilization
15 ratios were multiplied with SDR’s total capacity to estimate SDR’s consumption for each of the years, 2008 through
16 2010 (SDR 2013). The 1.24 metric tons CO₂/metric ton EAF dust consumed emission factor was then applied to
17 SDR’s estimated EAF dust consumption to develop CO₂ emission estimates for those Waelz kiln facilities.

18 PIZO’s facility in Arkansas was operational from 2009 to 2012 (PIZO 2021). The amount of EAF dust consumed by
19 PIZO’s facility for 2009 through 2012 was not publicly available. EAF dust consumption for PIZO’s facility for 2009
20 and 2010 were estimated by calculating annual capacity utilization of AZR’s Waelz kilns and multiplying this
21 utilization ratio by PIZO’s total capacity (PIZO 2012). EAF dust consumption for PIZO’s facility for 2011 through
22 2012 were estimated by applying the average annual capacity utilization rates for AZR and SDR (Grupo PROMAX)
23 to PIZO’s annual capacity (Horsehead 2012; SDR 2012; PIZO 2012). The 1.24 metric tons CO₂/metric ton EAF dust
24 consumed emission factor was then applied to PIZO’s estimated EAF dust consumption to develop CO₂ emission
25 estimates for those Waelz kiln facilities.

26 The production and use of coking coal for zinc production is adjusted for within the Energy chapter as this fuel was
27 consumed during non-energy related activities. Additional information on the adjustments made within the Energy
28 sector for Non-Energy Use of Fuels is described in both the Methodology section of CO₂ from Fossil Fuel
29 Combustion (3.1 Fossil Fuel Combustion (CRT Source Category 1A)) and Annex 2.1, Methodology for Estimating
30 Emissions of CO₂ from Fossil Fuel Combustion.

31 Beginning with the 2017 USGS *Minerals Commodity Summary: Zinc*, United States primary and secondary refined
32 zinc production were reported as one value, total refined zinc production. Prior to this publication, primary and
33 secondary refined zinc production statistics were reported separately. For years 2016 through 2022, only one
34 facility produced primary zinc. Primary zinc produced from this facility was subtracted from the USGS 2016 to 2022
35 total zinc production statistic to estimate secondary zinc production for these years.

36 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
37 through 2022.

38 **Uncertainty – TO BE UPDATED FOR FINAL INVENTORY REPORT**

39 The uncertainty associated with these estimates is two-fold, relating to activity data and emission factors used.

40 First, there is uncertainty associated with the amount of EAF dust consumed in the United States to produce
41 secondary zinc using emission-intensive Waelz kilns. The estimate for the total amount of EAF dust consumed in
42 Waelz kilns is based on combining the totals for (1) the EAF dust consumption value obtained for the kilns currently
43 operated by Befesa (and formerly operated by AZR or Horsehead Corporation) and (2) an EAF dust consumption
44 value obtained from the Waelz kiln facility operated by SDR. For the 1990 through 2015 estimates, EAF dust
45 consumption values for the kilns currently operated by Befesa were obtained from annual financial reports to the
46 Securities and Exchange Commission (SEC) by AZR. In 2016, AZR reorganized as a private company and ceased

1 providing annual reports to the SEC (Recycling Today 2017). EAF dust consumption values for subsequent years
 2 from the Befesa kilns and SDR have been obtained from personal communication with facility representatives.
 3 Since actual EAF dust consumption information is not available for PIZO’s facility (2009 through 2010) and SDR’s
 4 facility (2008 through 2010), the amount is estimated by multiplying the EAF dust recycling capacity of the facility
 5 (available from the company’s website) by the capacity utilization factor for AZR (which was available from
 6 Horsehead Corporation financial reports). The EAF dust consumption for PIZO’s facility for 2011 through 2012 was
 7 estimated by multiplying the average capacity utilization factor developed from AZR and SDR’s annual capacity
 8 utilization rates by PIZO’s EAF dust recycling capacity. Therefore, there is uncertainty associated with the
 9 assumption used to estimate PIZO’s annual EAF dust consumption values for 2009 through 2012 and SDR’s annual
 10 EAF dust consumption values for 2008 through 2010. EPA uses an uncertainty range of ± 5 percent for these EAF
 11 dust consumption data inputs, based upon expert elicitation from the USGS commodity specialist.

12 Second, there is uncertainty associated with the emission factors used to estimate CO₂ emissions from secondary
 13 zinc production processes. The Waelz kiln emission factors are based on materials balances for metallurgical coke
 14 and EAF dust consumed as provided by Viklund-White (2000). Therefore, the accuracy of these emission factors
 15 depends upon the accuracy of these materials balances. Data limitations prevented the development of emission
 16 factors for the electrothermic process. Therefore, emission factors for the Waelz kiln process were applied to both
 17 electrothermic and Waelz kiln production processes. Consistent with the ranges in Table 4.25 of the *2006 IPCC*
 18 *Guidelines*, EPA assigned an uncertainty range of ± 20 percent for the Tier 1 Waelz kiln emission factors, which are
 19 provided by Viklund-White in the form of metric tons of coke per metric ton of EAF dust consumed and metric tons
 20 of coke per metric ton of zinc produced. In order to convert coke consumption rates to CO₂ emission rates, values
 21 for the heat and carbon content of coke were obtained from Table 4.2 – Tier 2 of the *2006 IPCC Guidelines*. An
 22 uncertainty range of ± 10 percent was assigned to these coke data elements based upon Table 4.25, Tier 2 –
 23 National Reducing Agent & Process Materials Data.

24 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-117. Zinc production
 25 CO₂ emissions from 2022 were estimated to be between 0.8 and 1.2 MMT CO₂ Eq. at the 95 percent confidence
 26 level. This indicates a range of approximately 19 percent below and 20 percent above the emission estimate of 1.0
 27 MMT CO₂ Eq.

28 **Table 4-117: Approach 2 Quantitative Uncertainty Estimates for CO₂ Emissions from Zinc**
 29 **Production (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Zinc Production	CO ₂	1.0	0.8	1.2	-19%	+20%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

30 QA/QC and Verification

31 For more information on the general QA/QC process applied to this source category, consistent with Volume 1,
 32 Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in the introduction of
 33 the IPPU chapter.

34 Recalculations Discussion

35 Recalculations were performed for the year 2021 based on updated EAF dust consumption data. Compared to the
 36 previous *Inventory*, 2021 emissions from zinc production increased by 4 percent (38 kt CO₂).

1 Planned Improvements

2 Pending resources and prioritization of improvements for more significant sources, EPA will continue to evaluate
3 and analyze data reported under EPA's GHGRP that would be useful to improve the emission estimates and
4 category-specific QC for the Zinc Production source category, in particular considering completeness of reported
5 zinc production given the reporting threshold. Given the small number of facilities in the United States, particular
6 attention will be made to risks for disclosing CBI and ensuring time-series consistency of the emissions estimates
7 presented in future *Inventory* reports, consistent with IPCC and UNFCCC guidelines. This is required as the facility-
8 level reporting data from EPA's GHGRP, with the program's initial requirements for reporting of emissions in
9 calendar year 2010, are not available for all inventory years (i.e., 1990 through 2009) as required for this *Inventory*.
10 In implementing improvements and integration of data from EPA's GHGRP, the latest guidance from the IPCC on
11 the use of facility-level data in national inventories will be relied upon.⁸⁸ This is a long-term planned improvement,
12 and EPA is still assessing the possibility of including this improvement in future *Inventory* reports.

13 4.24 Electronics Industry (CRT Source 14 Category 2E)

15 The electronics industry uses multiple greenhouse gases in its manufacturing processes. In semiconductor
16 manufacturing, these include long-lived fluorinated greenhouse gases used for plasma etching and chamber
17 cleaning (CRT Source Category 2E1), fluorinated heat transfer fluids used for temperature control and other
18 applications (CRT Source Category 2E4), and nitrous oxide (N₂O) used to produce thin films through chemical vapor
19 deposition and in other applications (reported under CRT Source Category 2H3). Similar to semiconductor
20 manufacturing, the manufacturing of micro-electro-mechanical systems (MEMS) devices (reported under CRT
21 Source Category 2E5 Other) and photovoltaic (PV) cells (CRT Source Category 2E3) requires the use of multiple
22 long-lived fluorinated greenhouse gases for various processes.

23 The gases most commonly employed in the electronics industry are trifluoromethane (hydrofluorocarbon (HFC)-23
24 or CHF₃), perfluoromethane (CF₄), perfluoroethane (C₂F₆), nitrogen trifluoride (NF₃), and sulfur hexafluoride (SF₆),
25 although other fluorinated compounds such as perfluoropropane (C₃F₈) and perfluorocyclobutane (c-C₄F₈) are also
26 used. The exact combination of compounds is specific to the process employed.

27 In addition to emission estimates for these seven commonly used fluorinated gases, this *Inventory* contains
28 emissions estimates for N₂O and other HFCs and unsaturated, low-GWP PFCs including C₅F₈, C₄F₆, HFC-32, HFC-41,
29 and HFC-134a. These additional HFCs and PFCs are emitted from etching and chamber cleaning processes in much
30 smaller amounts, accounting for 0.02 percent of emissions (in CO₂ Eq.) from these processes.

31 For semiconductors, a single 300 mm silicon wafer that yields between 400 to 600 semiconductor products
32 (devices or chips) may require more than 100 distinct fluorinated-gas-using process steps, principally to deposit
33 and pattern dielectric films. Plasma etching (or patterning) of dielectric films, such as silicon dioxide and silicon
34 nitride, is performed to provide pathways for conducting material to connect individual circuit components in each
35 device. The patterning process uses plasma-generated fluorine atoms, which chemically react with exposed
36 dielectric film to selectively remove the desired portions of the film. The material removed as well as undissociated
37 fluorinated gases flow into waste streams and, unless emission abatement systems are employed, into the
38 atmosphere. Plasma enhanced chemical vapor deposition (PECVD) chambers, used for depositing dielectric films,
39 are cleaned periodically using fluorinated and other gases. During the cleaning cycle the gas is converted to

⁸⁸ See http://www.ipcc-nggip.iges.or.jp/public/tb/TFI_Technical_Bulletin_1.pdf.

1 fluorine atoms in plasma, which etches away residual material from chamber walls, electrodes, and chamber
2 hardware. Undissociated fluorinated gases and other products pass from the chamber to waste streams and,
3 unless abatement systems are employed, into the atmosphere.

4 In addition to emissions of unreacted gases, some fluorinated compounds can also be transformed in the plasma
5 processes into different fluorinated compounds which are then exhausted, unless abated, into the atmosphere.
6 For example, when C₂F₆ is used in cleaning or etching, CF₄ is typically generated and emitted as a process
7 byproduct. In some cases, emissions of the byproduct gas can rival or even exceed emissions of the input gas, as is
8 the case for NF₃ used in remote plasma chamber cleaning, which often generates CF₄ as a byproduct.

9 Besides dielectric film etching and PECVD chamber cleaning, much smaller quantities of fluorinated gases are used
10 to etch polysilicon films and refractory metal films like tungsten.

11 Nitrous oxide is used in manufacturing semiconductor devices to produce thin films by CVD and nitridation
12 processes as well as for N-doping of compound semiconductors and reaction chamber conditioning (Doering
13 2000).

14 Liquid perfluorinated compounds are also used as heat transfer fluids (F-HTFs) for temperature control, device
15 testing, cleaning substrate surfaces and other parts, and soldering in certain types of semiconductor
16 manufacturing production processes. Leakage and evaporation of these fluids during use is a source of fluorinated
17 gas emissions (EPA 2006). Unweighted F-HTF emissions consist primarily of perfluorinated amines,
18 hydrofluoroethers, perfluoropolyethers (specifically, PFPMEs), and perfluoroalkylmorpholines. Three percent or
19 less consist of HFCs, PFCs, and SF₆ (where PFCs are defined as compounds including only carbon and fluorine). With
20 the exceptions of the hydrofluoroethers and most of the HFCs, all of these compounds are very long-lived in the
21 atmosphere and have global warming potentials (GWPs) near 10,000.⁸⁹

22 MEMS and photovoltaic cell manufacturing require thin film deposition and etching of material with a thickness of
23 one micron or more, so the process is less intricate and complex than semiconductor manufacturing. The
24 manufacturing process is different than semiconductors, but generally employs similar techniques. Like
25 semiconductors, MEMS and photovoltaic cell manufacturers use fluorinated compounds for etching, cleaning
26 reactor chambers, and temperature control. CF₄, SF₆, and the Bosch process (which consists of alternating steps of
27 SF₆ and C₄F₈) are used to manufacture MEMS (EPA 2010). Photovoltaic cell manufacturing predominately uses CF₄,
28 to etch crystalline silicon wafers, and C₂F₆ or NF₃ during chamber cleaning after deposition of SiN_x films (IPCC
29 2006), although other F-GHGs may be used. Similar to semiconductor manufacturing, both MEMS and photovoltaic
30 cell manufacturing use N₂O in depositing films and other manufacturing processes. MEMS and photovoltaic
31 manufacturing may also employ HTFs for cooling process equipment (EPA 2010).

32 Emissions from all fluorinated greenhouse gases (including F-HTFs) and N₂O for semiconductors, MEMS and
33 photovoltaic cells manufacturing are presented in Table 4-118 below for the years 1990, 2005, and the period
34 2018 to 2022. The rapid growth of the electronics industry and the increasing complexity (growing number of
35 layers and functions)⁹⁰ of electronic products led to an increase in emissions of 152 percent between 1990 and
36 1999, when emissions peaked at 8.4 MMT CO₂ Eq. Emissions began to decline after 1999, reaching a low point in

⁸⁹ The GWP of PFPME, a perfluoropolyether used as an F-HTF, is included in the IPCC *Fourth Assessment Report* with a value of 10,300. The GWPs of the perfluorinated amines and perfluoroalkylmorpholines that are used as F-HTFs have not been evaluated in the peer-reviewed literature. However, evaluations by the manufacturer indicate that their GWPs are near 10,000 (78 FR 20632), which is expected given that these compounds are both saturated and fully fluorinated. EPA assigns a default GWP of 10,000 to compounds that are both saturated and fully fluorinated and that do not have chemical-specific GWPs in either the Fourth or the Fifth Assessment Reports.

⁹⁰ Complexity is a term denoting the circuit required to connect the active circuit elements (transistors) on a chip. Increasing miniaturization, for the same chip size, leads to increasing transistor density, which, in turn, requires more complex interconnections between those transistors. This increasing complexity is manifested by increasing the levels (i.e., layers) of wiring, with each wiring layer requiring fluorinated gas usage for its manufacture.

1 2009 before rebounding to 2006 emission levels and more or less plateauing at the current level, which represents
2 a 44 percent decline from 1999 to 2022. Together, industrial growth, adoption of emissions reduction technologies
3 (including but not limited to abatement technologies) and shifts in gas usages resulted in a net increase in
4 emissions of approximately 41 percent between 1990 and 2022. Total emissions from semiconductor manufacture
5 in 2022 were lower than 2021 emissions, decreasing by 7.5 percent, largely due to a large decrease in SF₆
6 emissions. The decrease in SF₆ are seen in facilities that manufacture 200 mm wafer size that do not have
7 abatement systems installed as well as 300 mm wafer size that have abatement systems installed.

8 For U.S. semiconductor manufacturing in 2022, total CO₂-equivalent emissions of all fluorinated greenhouse gases
9 and N₂O from deposition, etching, and chamber cleaning processes were estimated to be 4.6 MMT CO₂ Eq. This is a
10 decrease in emissions from 1999 of 45 percent, and an increase in emissions from 1990 of 40 percent. These
11 trends are driven by the above stated reasons.

12 Photovoltaic cell and MEMS manufacturing emissions of all fluorinated greenhouse gases are in Table 4-118. While
13 EPA has developed a simple methodology to estimate emissions from non-reporters and to back-cast emissions
14 from these sources for the entire time series, there is very high uncertainty associated with these emission
15 estimates.

16 The emissions reported by facilities manufacturing MEMS included emissions of C₂F₆, C₃F₈, c-C₄F₈, CF₄, HFC-23, NF₃,
17 N₂O and SF₆,⁹¹ and were equivalent to only 0.110 percent to 0.264 percent of the total reported emissions from
18 electronics manufacturing in 2011 to 2022. F-GHG emissions, the primary type of emissions for MEMS, ranged
19 from 0.0003 to 0.012 MMT CO₂ Eq. from 1991 to 2022. Based upon information in the World Fab Forecast (WFF), it
20 appears that some GHGRP reporters that manufacture both semiconductors and MEMS are reporting their
21 emissions as only from semiconductor manufacturing (GHGRP reporters must choose a single classification per
22 fab). Emissions from non-reporters have not been estimated.

23 Total CO₂-equivalent emissions from manufacturing of photovoltaic cells were estimated to range from 0.0003
24 MMT CO₂ Eq. to 0.0330 MMT CO₂ Eq. from 1998 to 2022 and were equivalent to between 0.003 percent to 0.77
25 percent of the total reported emissions from electronics manufacturing. F-GHG emissions, the primary type of
26 emissions for photovoltaic cells, ranged from 0.0003 to 0.032 MMT CO₂ Eq. from 1998 to 2022. Emissions from
27 manufacturing of photovoltaic cells were estimated using an emission factor developed from reported data from a
28 single manufacturer between 2015 and 2016. This emission factor was then applied to production capacity
29 estimates from non-reporting facilities. Reported emissions from photovoltaic cell manufacturing consisted of CF₄,
30 C₂F₆, c-C₄F₈, CHF₃, NF₃, and N₂O.⁹²

31 Emissions of F-HTFs, grouped by HFCs, PFCs or SF₆ are presented in Table 4-118. Emissions of F-HTFs that are not
32 HFCs, PFCs or SF₆ are not included in inventory totals and are included for informational purposes only.

33 Since reporting of F-HTF emissions began under EPA's GHGRP in 2011, total F-HTF emissions (reported and
34 estimated non-reported) have fluctuated between 0.4 MMT CO₂ Eq. and 0.9 MMT CO₂ Eq., with an overall
35 declining trend between 2011 to 2022. An analysis of the data reported to EPA's GHGRP indicates that F-HTF
36 emissions account for anywhere between 9 percent and 17 percent of total annual emissions (F-GHG, N₂O and F-
37 HTFs) from semiconductor manufacturing.⁹³ Table 4-120 shows F-HTF emissions in tons by compound group based

⁹¹ Gases not reported by MEMS manufacturers to the GHGRP are currently listed as "NE" in the CRTs. Since no facilities report using these gases, emissions of these gases are not estimated for this sub-sector. However, there is insufficient data to definitively conclude that they are not used by non-reporting facilities.

⁹² Gases not reported by PV manufacturers to the GHGRP are currently listed as "NE" in the CRTs. Since no facilities report using these gases, emissions of these gases are not estimated for this sub-sector. However, there is insufficient data to definitively conclude that they are not used by non-reporting facilities.

⁹³ Emissions data for HTFs (in tons of gas) from the semiconductor industry from 2011 through 2022 were obtained from the EPA GHGRP annual facility emissions reports.

1 on reporting to EPA’s GHGRP and the interpolated share of F-HTF emissions to F-GHG emissions for select years
 2 prior to reporting.⁹⁴

3 **Table 4-118: PFC, HFC, SF₆, NF₃, and N₂O Emissions from Electronics Industry (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
CF ₄	0.8	1.0	1.6	1.5	1.6	1.7	1.6
C ₂ F ₆	1.8	1.8	1.1	0.9	0.8	0.9	0.9
C ₃ F ₈	+	0.1	0.1	0.1	0.1	0.1	0.1
C ₄ F ₈	0.0	0.1	0.1	0.1	0.1	0.1	0.1
HFC-23	0.2	0.2	0.3	0.3	0.3	0.4	0.3
SF ₆	0.5	0.8	0.8	0.8	0.8	0.9	0.7
NF ₃	+	0.4	0.6	0.5	0.6	0.7	0.6
C ₄ F ₆	+	+	+	+	+	+	+
C ₅ F ₈	+	+	+	+	+	+	+
CH ₂ F ₂	+	+	+	+	+	+	+
CH ₃ F	+	+	+	+	+	+	+
CH ₂ FCF ₃	+	+	+	+	+	+	0
Total Semiconductors	3.3	4.3	4.5	4.2	4.3	4.7	4.3
CF ₄	0.0	+	+	+	+	+	+
C ₂ F ₆	0.0	+	+	+	+	+	+
C ₃ F ₈	0.0	+	0.0	0.0	0.0	0.0	0.0
C ₄ F ₈	0.0	+	+	+	+	+	+
HFC-23	0.0	+	+	+	+	+	+
SF ₆	0.0	+	+	+	+	+	+
NF ₃	0.0	0.0	+	+	+	+	+
Total MEMS	0.0	+	+	+	+	+	+
CF ₄	0.0	+	+	+	+	+	+
C ₂ F ₆	0.0	+	+	+	+	+	+
C ₄ F ₈	0.0	+	+	+	+	+	+
HFC-23	0.0	+	+	+	+	+	+
SF ₆	0.0	0.0	0.0	0.0	0.0	0.0	0.0
NF ₃	0.0	0.0	0.0	0.0	0.0	0.0	0.0
Total PV	0.0	+	+	+	+	+	+
N ₂ O (Semiconductors)	+	0.1	0.2	0.2	0.3	0.3	0.3
N ₂ O (MEMS)	0.0	+	+	+	+	+	+
N ₂ O (PV)	0.0	+	+	+	+	+	+
Total N₂O	+	0.1	0.2	0.2	0.3	0.3	0.3
HFC, PFC and SF ₆ F-HTFs	0.0	+	+	+	+	+	+
Total Electronics Industry	3.3	4.5	4.8	4.4	4.5	4.8	4.7

+ Does not exceed 0.05 MMT CO₂ Eq.

4 **Table 4-119: PFC, HFC, SF₆, NF₃, and N₂O Emissions from Semiconductor Manufacture (Metric**
 5 **Tons)**

Year	1990	2005	2018	2019	2020	2021	2022
CF ₄	114.8	145.3	237.0	221.1	237.4	252.8	240.9
C ₂ F ₆	160.0	163.4	98.7	83.2	76.0	80.0	80.8
C ₃ F ₈	0.4	7.3	12.2	9.9	9.4	11.0	13.4

⁹⁴ Many fluorinated heat transfer fluids consist of perfluoropolymethylisopropyl ethers (PFPMIEs) of different molecular weights and boiling points that are distilled from a mixture. “BP 200 °C” (and similar terms below) indicate the boiling point of the fluid in degrees Celsius. For more information, see <https://www.regulations.gov/document?D=EPA-HQ-OAR-2009-0927-0276>.

C ₄ F ₈	0.0	10.9	6.2	5.7	6.1	6.6	5.9
HFC-23	14.6	14.1	26.4	25.6	26.5	32.6	24.9
SF ₆	21.7	33.4	33.8	32.8	33.6	39.8	31.1
NF ₃	2.8	26.2	34.4	33.5	37.2	40.9	39.1
C ₄ F ₆	0.7	0.9	0.8	0.9	0.8	1.2	0.8
C ₅ F ₈	0.5	0.6	0.5	0.4	0.4	0.4	0.4
CH ₂ F ₂	0.6	0.8	0.9	1.0	1.1	1.1	1.0
CH ₃ F	1.4	1.8	2.6	2.5	3.1	3.2	2.4
CH ₂ FCF ₃	+	+	+	+	+	+	0.0
N ₂ O	135.9	463.3	881.5	822.2	1,023.0	1,083.6	1,111.2

+ Does not exceed 0.05 MT.

1 **Table 4-120: F-HTF Emissions from Electronics Manufacture by Compound Group (kt CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
HFCs	0.0	0.9	2.7	1.1	0.9	1.2	1.5
PFCs	0.0	3.8	10.0	8.4	8.0	5.7	7.3
SF ₆	0.0	5.6	13.2	6.0	13.2	9.4	4.1
HFEs	0.0	39.4	4.6	1.3	5.5	4.0	14.2
PFPMIes	0.0	110.5	183.3	172.6	154.8	155.4	162.8
Perfluoroalkylmorpholines	0.0	66.4	58.5	56.4	62.8	55.9	19.4
Perfluorotrialkylamines	0.0	209.9	414.7	363.8	391.5	382.2	233.5
Total F-HTFs	0.0	436.5	687.0	608.4	636.6	613.6	443.0

Note: Emissions of F-HTFs that are not HFCs, PFCs or SF₆ are not included in inventory totals and are included for informational purposes only. Emissions presented for informational purposes include HFEs, PFPMIes, perfluoroalkylmorpholines, and perfluorotrialkylamines.

2 Methodology and Time-Series Consistency

3 Emissions are based on data reported through Subpart I, Electronics Manufacture, of EPA's GHGRP, semiconductor
4 manufacturing Partner-reported emissions data received through EPA's PFC⁹⁵ Reduction/Climate Partnership,
5 EPA's PFC Emissions Vintage Model (PEVM)—a model that estimates industry emissions from etching and chamber
6 cleaning processes in the absence of emission control strategies (Burton and Beizaie 2001)⁹⁶—and estimates of
7 industry activity (i.e., total manufactured layer area and manufacturing capacity). The availability and applicability
8 of reported emissions data from the EPA Partnership and EPA's GHGRP and activity data differ across the 1990
9 through 2022 time series. Consequently, fluorinated greenhouse gas (F-GHG) emissions from etching and chamber
10 cleaning processes for semiconductors were estimated using seven distinct methods, one each for the periods
11 1990 through 1994, 1995 through 1999, 2000 through 2006, 2007 through 2010, 2011 and 2012, 2013 and 2014,
12 and 2015 through 2022. Nitrous oxide emissions were estimated using five distinct methods, one each for the
13 period 1990 through 1994, 1995 through 2010, 2011 and 2012, 2013 and 2014, and 2015 through 2022. The
14 methodology discussion below for these time periods focuses on semiconductor emissions from etching, chamber

⁹⁵ In the context of the EPA Partnership and PEVM, PFC refers to perfluorocompounds, not perfluorocarbons.

⁹⁶ A Partner refers to a participant in the U.S. EPA PFC Reduction/Climate Partnership for the Semiconductor Industry. Through a Memorandum of Understanding (MoU) with the EPA, Partners voluntarily reported their PFC emissions to the EPA by way of a third party, which aggregated the emissions through 2010.

1 cleaning, and uses of N₂O. Other emissions for MEMS, photovoltaic cells, and HTFs were estimated using the
2 approaches described immediately below.

3 **MEMS**

4 GHGRP-reported emissions (F-GHG and N₂O) from the manufacturing of MEMS are available for the years 2011 to
5 2022. Emissions from manufacturing of MEMS for years prior to 2011 were calculated by linearly interpolating
6 emissions between 1990 (at zero MMT CO₂ Eq.) and 2011, the first year where emissions from manufacturing of
7 MEMS was reported to the GHGRP. Based upon information in the World Fab Forecast (WFF), it appears that some
8 GHGRP reporters that manufacture both semiconductors and MEMS are reporting their emissions as only from
9 semiconductor manufacturing; however, emissions from MEMS manufacturing are likely being included in
10 semiconductor totals. Emissions were not estimated for non-reporters.

11 **Photovoltaic Cells**

12 GHGRP-reported emissions (F-GHG and N₂O) from the manufacturing of photovoltaic cells are available for 2011,
13 2012, 2015, and 2016 from two manufacturers. EPA estimates the emissions from manufacturing of PVs from non-
14 reporting facilities by multiplying the estimated capacity of non-reporters by a calculated F-GHG emission factor
15 and N₂O emission factor based on GHGRP reported emissions from the manufacturer (in MMT CO₂ Eq. per
16 megawatt) that reported emissions in 2015 and 2016. This manufacturer's emissions are expected to be more
17 representative of emissions from the sector, as their emissions were consistent with consuming only CF₄ for
18 etching processes and are a large-scale manufacturer, representing 28 percent of the U.S. production capacity in
19 2016. The second photovoltaic manufacturer only produced a small fraction of U.S. production (<4 percent). They
20 also reported the use of NF₃ in remote plasma cleaning processes, which does not have an emission factor in Part
21 98 for PV manufacturing, requiring them to report emissions equal to consumption. The total F-GHG emissions
22 from non-reporters are then disaggregated into individual gases using the gas distribution from the 2015 to 2016
23 manufacturer. Manufacturing capacities in megawatts were drawn from DisplaySearch, a 2015 Congressional
24 Research Service Report on U.S. Solar Photovoltaic Manufacturing, and self-reported capacity by GHGRP reporters.
25 EPA estimated that during the 2015 to 2016 period, 28 percent of manufacturing capacity in the United States was
26 represented through reported GHGRP emissions. Capacities are estimated for the full time series by linearly scaling
27 the total U.S. capacity between zero in 1997 to the total capacity reported of crystalline silicon (c-Si) PV
28 manufacturing in 2000 in DisplaySearch and then linearly scaling between the total capacity of c-Si PV
29 manufacturing in DisplaySearch in 2009 to the total capacity of c-Si PV manufacturing reported in the
30 Congressional Research Service report in 2012. Capacities were held constant for non-reporters for 2012 to 2019.
31 In 2020, non-reporter capacity declined due to the closure of several PV manufacturing plants. This capacity was
32 held constant for 2021 and 2022. Average emissions per MW from the GHGRP reporter in 2015 and 2016 were
33 then applied to the total capacity prior to 2015. Emissions for 2014 from the GHGRP reporter that reported in 2015
34 and 2016 were scaled to the number of months open in 2014. For 1998 through 2022, emissions per MW
35 (capacity) from the GHGRP reporter were applied to the non-reporters. For 2017 through 2022, there are no
36 reported PV emissions. Therefore, emissions were estimated using the EPA-derived emission factor and estimated
37 manufacturing capacity from non-reporters only.

38 **HTFs**

39 Facility emissions of F-HTFs from semiconductor manufacturing are reported to EPA under its GHGRP and are
40 available for the years 2011 through 2022. EPA estimates the emissions of F-HTFs from non-reporting
41 semiconductor facilities by calculating the ratio of GHGRP-reported fluorinated HTF emissions to GHGRP reported
42 F-GHG emissions from etching and chamber cleaning processes, and then multiplying this ratio by the F-GHG
43 emissions from etching and chamber cleaning processes estimated for non-reporting facilities. Fluorinated HTF use
44 in semiconductor manufacturing is assumed to have begun in the early 2000s and to have gradually displaced
45 other HTFs (e.g., de-ionized water and glycol) in semiconductor manufacturing (EPA 2006). For time-series
46 consistency, EPA interpolated the share of F-HTF emissions to F-GHG emissions between 2000 (at 0 percent) and

1 2011 (at 17 percent) and applied these shares to the unadjusted F-GHG emissions during those years to estimate
2 the 4-143 missions.

3 **Semiconductors**

4 *1990 through 1994*

5 From 1990 through 1994, Partnership data were unavailable, and emissions were modeled using PEVM (Burton
6 and Beizaie 2001).⁹⁷ The 1990 to 1994 emissions are assumed to be uncontrolled, since reduction strategies such
7 as chemical substitution and abatement were yet to be developed.

8 PEVM is based on the recognition that fluorinated greenhouse gas emissions from semiconductor manufacturing
9 vary with: (1) the number of layers that comprise different kinds of semiconductor devices, including both silicon
10 wafer and metal interconnect layers, and (2) silicon consumption (i.e., the area of semiconductors produced) for
11 each kind of device. The product of these two quantities, Total Manufactured Layer Area (TMLA), constitutes the
12 activity data for semiconductor manufacturing. PEVM also incorporates an emission factor that expresses
13 emissions per unit of manufactured layer-area. Emissions are estimated by multiplying TMLA by this emission
14 factor.

15 PEVM incorporates information on the two attributes of semiconductor devices that affect the number of layers:
16 (1) linewidth technology (the smallest manufactured feature size),⁹⁸ and (2) product type (discrete, memory or
17 logic).⁹⁹ For each linewidth technology, a weighted average number of layers is estimated using VLSI product-
18 specific worldwide silicon demand data in conjunction with complexity factors (i.e., the number of layers per
19 Integrated Circuit (IC) specific to product type (Burton and Beizaie 2001; ITRS 2007). PEVM derives historical
20 consumption of silicon (i.e., square inches) by linewidth technology from published data on annual wafer starts
21 and average wafer size (VLSI Research, Inc. 2012).

22 The emission factor in PEVM is the average of four historical emission factors, each derived by dividing the total
23 annual emissions reported by the Partners for each of the four years between 1996 and 1999 by the total TMLA
24 estimated for the Partners in each of those years. Over this period, the emission factors varied relatively little (i.e.,
25 the relative standard deviation for the average was 5 percent). Since Partners are believed not to have applied
26 significant emission reduction measures before 2000, the resulting average emission factor reflects uncontrolled
27 emissions and hence may be used here to estimate 1990 through 1994 emissions. The emission factor is used to
28 estimate U.S. uncontrolled emissions using publicly available data on world (including U.S.) silicon consumption.

29 As it was assumed for this time period that there was no consequential adoption of fluorinated-gas-reducing
30 measures, a fixed distribution of fluorinated-gas use was assumed to apply to the entire U.S. industry to estimate
31 gas-specific emissions. This distribution was based upon the average fluorinated-gas purchases made by

⁹⁷ Various versions of the PEVM exist to reflect changing industrial practices. From 1990 to 1994 emissions estimates are from PEVM v1.0, completed in September 1998. The emission factor used to estimate 1990 to 1994 emissions is an average of the 1995 and 1996 emissions factors, which were derived from Partner reported data for those years.

⁹⁸ By decreasing features of Integrated Circuit components, more components can be manufactured per device, which increases its functionality. However, as those individual components shrink it requires more layers to interconnect them to achieve the functionality. For example, a microprocessor manufactured with 65 nm feature sizes might contain as many as 1 billion transistors and require as many as 11 layers of component interconnects to achieve functionality, while a device manufactured with 130 nm feature size might contain a few hundred million transistors and require 8 layers of component interconnects (ITRS 2007).

⁹⁹ Memory devices manufactured with the same feature sizes as microprocessors (a logic device) require approximately one-half the number of interconnect layers, whereas discrete devices require only a silicon base layer and no interconnect layers (ITRS 2007). Since discrete devices did not start using PFCs appreciably until 2004, they are only accounted for in the PEVM emissions estimates from 2004 onwards.

1 semiconductor manufacturers during this period and the application of IPCC default emission factors for each gas
2 (Burton and Beizaie 2001).

3 PEVM only addressed the seven main F-GHGs (CF₄, C₂F₆, C₃F₈, c-C₄F₈, HFC-23, SF₆, and NF₃) used in semiconductor
4 manufacturing. Through reporting under Subpart I of EPA's GHGRP, data on other F-GHGs (C₄F₆, C₅F₈, HFC-32, HFC-
5 41, HFC-134a) used in semiconductor manufacturing became available and EPA was therefore able to extrapolate
6 this data across the entire 1990 to 2022 timeseries. To estimate emissions for these "other F-GHGs", emissions
7 data from Subpart I between 2014 to 2016 were used to estimate the average share or percentage contribution of
8 these gases as compared to total F-GHG emissions. Subpart I emission factors were updated for 2014 by EPA as a
9 result of a larger set of emission factor data becoming available, so reported data from 2011 through 2013 was not
10 utilized for the average. To estimate non-reporter emissions from 2011-2022, the average emissions data from
11 Subpart I of 2011 to 2022 was used.

12 To estimate N₂O emissions, it was assumed the proportion of N₂O emissions estimated for 1995 (discussed below)
13 remained constant for the period of 1990 through 1994.

14 *1995 through 1999*

15 For 1995 through 1999, total U.S. emissions were extrapolated from the total annual emissions reported by the
16 Partners (1995 through 1999). Partner-reported emissions are considered more representative (e.g., in terms of
17 capacity utilization in a given year) than PEVM-estimated emissions and are used to generate total U.S. emissions
18 when applicable. The emissions reported by the Partners were divided by the ratio of the total capacity of the
19 plants operated by the Partners and the total capacity of all of the semiconductor plants in the United States; this
20 ratio represents the share of capacity attributable to the Partnership. This method assumes that Partners and non-
21 Partners have identical capacity utilizations and distributions of manufacturing technologies. Plant capacity data is
22 contained in the World Fab Forecast (WFF) database and its predecessors, which is updated quarterly. Gas-specific
23 emissions were estimated using the same method as for 1990 through 1994.

24 For this time period emissions of other F-GHGs (C₄F₆, C₅F₈, HFC-32, HFC-41, HFC-134a) were estimated using the
25 method described above for 1990 to 1994.

26 For this time period, the N₂O emissions were estimated using an emission factor that was applied to the annual,
27 total U.S. TMLA manufactured. The emission factor was developed using a regression-through-the-origin (RTO)
28 model: GHGRP reported N₂O emissions were regressed against the corresponding TMLA of facilities that reported
29 no use of abatement systems. Details on EPA's GHGRP reported emissions and development of emission factor
30 using the RTO model are presented in the 2011 through 2012 section. The total U.S. TMLA for 1995 through 1999
31 was estimated using PEVM.

32 *2000 through 2006*

33 Emissions for the years 2000 through 2006—the period during which Partners began the consequential application
34 of fluorinated greenhouse gas-reduction measures—were estimated using a combination of Partner-reported
35 emissions and adjusted PEVM modeled emissions. The emissions reported by Partners for each year were
36 accepted as the quantity emitted from the share of the industry represented by those Partners. Remaining
37 emissions, those from non-Partners, were estimated using PEVM, with one change. To ensure time-series
38 consistency and to reflect the increasing use of remote clean technology (which increases the efficiency of the
39 production process while lowering emissions of fluorinated greenhouse gases), the average non-Partner emission
40 factor (PEVM emission factor) was assumed to begin declining gradually during this period. Specifically, the non-
41 Partner emission factor for each year was determined by linear interpolation, using the end points of 1999 (the
42 original PEVM emission factor) and 2011 (a new emission factor determined for the non-Partner population based
43 on GHGRP-reported data, described below).

44 The portion of the U.S. total emissions attributed to non-Partners is obtained by multiplying PEVM's total U.S.

1 emissions figure by the non-Partner share of U.S. total silicon capacity for each year as described above.¹⁰⁰ Gas-
2 specific emissions from non-Partners were estimated using linear interpolation between the gas-specific emissions
3 distributions of 1999 (assumed to be the same as that of the total U.S. Industry in 1994) and 2011 (calculated from
4 a subset of non-Partners that reported through the GHGRP as a result of emitting more than 25,000 MT CO₂ Eq.
5 per year). Annual updates to PEVM reflect published figures for actual silicon consumption from VLSI Research,
6 Inc., revisions and additions to the world population of semiconductor manufacturing plants, and changes in IC
7 fabrication practices within the semiconductor industry (see ITRS 2008 and Semiconductor Equipment and
8 Materials Industry 2011).^{101, 102, 103} For this time period emissions of other F-GHGs (C₄F₆, C₅F₈, HFC-32, HFC-41,
9 HFC-134a) were estimated using the method described above for 1990 to 1994.
10 Nitrous oxide emissions were estimated using the same methodology as the 1995 through 1999 methodology.

11 *2007 through 2010*

12 For the years 2007 through 2010, emissions were also estimated using a combination of Partner reported
13 emissions and adjusted PEVM modeled emissions to provide estimates for non-Partners; however, two
14 improvements were made to the estimation method employed for the previous years in the time series. First, the
15 2007 through 2010 emission estimates account for the fact that Partners and non-Partners employ different
16 distributions of manufacturing technologies, with the Partners using manufacturing technologies with greater
17 transistor densities and therefore greater numbers of layers.¹⁰⁴ Second, the scope of the 2007 through 2010
18 estimates was expanded relative to the estimates for the years 2000 through 2006 to include emissions from
19 research and development (R&D) fabs. This additional enhancement was feasible through the use of more detailed
20 data published in the WFF. PEVM databases were updated annually as described above. The published world
21 average capacity utilization for 2007 through 2010 was used for production fabs, while for R&D fabs a 20 percent

¹⁰⁰ This approach assumes that the distribution of linewidth technologies is the same between Partners and non-Partners. As discussed in the description of the method used to estimate 2007 emissions, this is not always the case.

¹⁰¹ Special attention was given to the manufacturing capacity of plants that use wafers with 300 mm diameters because the actual capacity of these plants is ramped up to design capacity, typically over a 2 to 3 year period. To prevent overstating estimates of partner-capacity shares from plants using 300 mm wafers, *design* capacities contained in WFF were replaced with estimates of *actual installed* capacities for 2004 published by Citigroup Smith Barney (2005). Without this correction, the partner share of capacity would be overstated, by approximately 5 percent. For perspective, approximately 95 percent of all new capacity additions in 2004 used 300 mm wafers, and by year-end those plants, on average, could operate at approximately 70 percent of the design capacity. For 2005, actual installed capacities were estimated using an entry in the World Fab Watch database (April 2006 Edition) called "wafers/month, 8-inch equivalent," which denoted the actual installed capacity instead of the fully-ramped capacity. For 2006, actual installed capacities of new fabs were estimated using an average monthly ramp rate of 1100 wafer starts per month (wspm) derived from various sources such as semiconductor fabtech, industry analysts, and articles in the trade press. The monthly ramp rate was applied from the first-quarter of silicon volume (FQSV) to determine the average design capacity over the 2006 period.

¹⁰² In 2006, the industry trend in co-ownership of manufacturing facilities continued. Several manufacturers, who are Partners, now operate fabs with other manufacturers, who in some cases are also Partners and in other cases are not Partners. Special attention was given to this occurrence when estimating the Partner and non-Partner shares of U.S. manufacturing capacity.

¹⁰³ Two versions of PEVM are used to model non-Partner emissions during this period. For the years 2000 to 2003 PEVM v3.2.0506.0507 was used to estimate non-Partner emissions. During this time, discrete devices did not use PFCs during manufacturing and therefore only memory and logic devices were modeled in the PEVM v3.2.0506.0507. From 2004 onwards, discrete device fabrication started to use PFCs, hence PEVM v4.0.0701.0701, the first version of PEVM to account for PFC emissions from discrete devices, was used to estimate non-Partner emissions for this time period.

¹⁰⁴ EPA considered applying this change to years before 2007 but found that it would be difficult due to the large amount of data (i.e., technology-specific global and non-Partner TMLA) that would have to be examined and manipulated for each year. This effort did not appear to be justified given the relatively small impact of the improvement on the total estimate for 2007 and the fact that the impact of the improvement would likely be lower for earlier years because the estimated share of emissions accounted for by non-Partners is growing as Partners continue to implement emission-reduction efforts.

1 figure was assumed (SIA 2009).

2 In addition, publicly available utilization data was used to account for differences in fab utilization for
3 manufacturers of discrete and IC products for 2010 emissions for non-Partners. The Semiconductor Capacity
4 Utilization (SICAS) Reports from SIA provides the global semiconductor industry capacity and utilization,
5 differentiated by discrete and IC products (SIA 2009 through 2011). PEVM estimates were adjusted using
6 technology-weighted capacity shares that reflect the relative influence of different utilization. Gas-specific
7 emissions for non-Partners were estimated using the same method as for 2000 through 2006.

8 For this time period emissions of other F-GHGs (C₅F₈, CH₂F₂, CH₃F, CH₂FCF₃, C₂H₂F₄) were estimated using the
9 method described above for 1990 to 1994. Nitrous oxide emissions were estimated using the same methodology
10 as the 1995 through 1999 methodology.

11 *2011 through 2012*

12 The fifth method for estimating emissions from semiconductor manufacturing covers the period 2011 through
13 2012. This methodology differs from previous years because the EPA's Partnership with the semiconductor
14 industry ended (in 2010) and reporting under EPA's GHGRP began. Manufacturers whose estimated uncontrolled
15 emissions equal or exceed 25,000 MT CO₂ Eq. per year (based on default F-GHG-specific emission factors and total
16 capacity in terms of substrate area) are required to report their emissions to EPA. This population of reporters to
17 EPA's GHGRP included both historical Partners of EPA's PFC Reduction/Climate Partnership as well as non-Partners
18 some of which use gallium arsenide (GaAs) technology in addition to Si technology.¹⁰⁵ Emissions from the
19 population of manufacturers that were below the reporting threshold were also estimated for this time period
20 using EPA-developed emission factors and estimates of facility-specific production obtained from WFF. Inventory
21 totals reflect the emissions from both reporting and non-reporting populations.

22 Under EPA's GHGRP, semiconductor manufacturing facilities report emissions of F-GHGs (for all types of F-GHGs)
23 used in etch and clean processes as well as emissions of fluorinated heat transfer fluids. (Fluorinated heat transfer
24 fluids are used to control process temperatures, thermally test devices, and clean substrate surfaces, among other
25 applications.) They also report N₂O emissions from CVD and other processes. The F-GHGs and N₂O were
26 aggregated, by gas, across all semiconductor manufacturing GHGRP reporters to calculate gas-specific emissions
27 for the GHGRP-reporting segment of the U.S. industry. At this time, emissions that result from heat transfer fluid
28 use that are HFC, PFC and SF₆ are included in the total emission estimates from semiconductor manufacturing, and
29 these GHGRP-reported emissions have been compiled and presented in Table 4-118. F-HTF emissions resulting
30 from other types of gases (e.g., HFEs) are not presented in semiconductor manufacturing totals in Table 4-118 and
31 Table 4-119 but are shown in Table 4-120 for informational purposes.

32 Changes to the default emission factors and default destruction or removal efficiencies (DREs) used for GHGRP
33 reporting affected the emissions trend between 2013 and 2014. These changes did not reflect actual emission rate
34 changes but data improvements. Therefore, for the current *Inventory*, EPA adjusted the time series of GHGRP-
35 reported data for 2011 through 2013 to ensure time-series consistency using a series of calculations that took into
36 account the characteristics of a facility (e.g., wafer size and abatement use). To adjust emissions for facilities that
37 did not report abatement in 2011 through 2013, EPA simply applied the revised emission factors to each facility's
38 estimated gas consumption by gas, process type and wafer size. In 2014, EPA also started collecting information on
39 fab-wide DREs and the gases abated by process type, which were used in calculations for adjusting emissions from
40 facilities that abated F-GHGs in 2011 through 2013.

- 41 • To adjust emissions for facilities that abated emissions in 2011 through 2013, EPA first calculated the
42 quantity of gas abated in 2014 using reported F-GHG emissions, the revised default DREs (or the

¹⁰⁵ GaAs and Si technologies refer to the wafer on which devices are manufactured, which use the same PFCs but in different ways.

1 estimated site-specific DRE,¹⁰⁶ if a site-specific DRE was indicated), and the fab-wide DREs reported in
2 2014.¹⁰⁷ To adjust emissions for facilities that abated emissions in 2011 through 2013, EPA first estimated
3 the percentage of gas passing through abatement systems for remote plasma clean in 2014 using the ratio
4 of emissions reported for CF₄ and NF₃.

- 5 • EPA then estimated the quantity of NF₃ abated for remote plasma clean in 2014 using the ratio of
6 emissions reported for CF₄ (which is not abated) and NF₃. This abated quantity was then subtracted from
7 the total abated quantity calculated as described in the bullet above.
- 8 • To account for the resulting remaining abated quantity, EPA assumed that the percentage of gas passing
9 through abatement systems was the same across all remaining gas and process type combinations where
10 abatement was reported for 2014.
- 11 • The percentage of gas abated was then assumed to be the same in 2011 through 2013 (if the facility
12 claimed abatement that year) as in 2014 for each gas abated in 2014.

13 The revised emission factors and DREs were then applied to the estimated gas consumption for each facility by gas,
14 process type and wafer size.¹⁰⁸

15 For the segment of the semiconductor industry that is below EPA's GHGRP reporting threshold, and for R&D
16 facilities, which are not covered by EPA's GHGRP, emission estimates are based on EPA-developed emission factors
17 for the F-GHGs and N₂O and estimates of manufacturing activity. The new emission factors (in units of mass of CO₂
18 Eq./TMLA [million square inches (MSI)]) are based on the emissions reported under EPA's GHGRP by facilities
19 without abatement and on the TMLA estimates for these facilities based on the WFF (SEMI 2012, 2013).¹⁰⁹ In a
20 refinement of the method used to estimate emissions for the non-Partner population for prior years, different
21 emission factors were developed for different subpopulations of fabs, disaggregated by wafer size (200 mm and
22 300 mm). For each of these groups, a subpopulation-specific emission factor was obtained using a regression-
23 through-the-origin (RTO) model: facility-reported aggregate emissions of seven F-GHGs (CF₄, C₂F₆, C₃F₈, C-C₄F₈,
24 CHF₃, SF₆ and NF₃)¹¹⁰ were regressed against the corresponding TMLA to estimate an aggregate F-GHG emissions
25 factor (CO₂ Eq./MSI TMLA), and facility-reported N₂O emissions were regressed against the corresponding TMLA to
26 estimate a N₂O emissions factor (CO₂ Eq./MSI TMLA). For each subpopulation, the slope of the RTO model is the
27 emission factor for that subpopulation. Information on the use of point-of-use abatement by non-reporting fabs
28 was not available; thus, EPA conservatively assumed that non-reporting facilities did not use point-of-use
29 abatement.

30 For 2011 and 2012, estimates of TMLA relied on the capacity utilization of the fabs published by the U.S. Census
31 Bureau's Historical Data Quarterly Survey of Plant Capacity Utilization (USCB 2011, 2012). Similar to the

¹⁰⁶ EPA generally assumed site-specific DREs were as follows: CF₄, Etch (90 percent); all other gases, Etch (98 percent); NF₃,
Clean (95 percent); CF₄, Clean (80 percent), and all other gases, Clean (80 percent). There were a few exceptions where a higher
DRE was assumed to ensure the calculations operated correctly when there was 100 percent abatement.

¹⁰⁷ If abatement information was not available for 2014 or the reported incorrectly in 2014, data from 2015 or 2016 was
substituted.

¹⁰⁸ Since facilities did not report by fab before 2014, fab-wide DREs were averaged if a facility had more than one fab. For
facilities that reported more than one wafer size per facility, the percentages of a facility's emissions per wafer size were
estimated in 2014 and applied to earlier years, if possible. If the percentage of emissions per wafer size were unknown, a 50/50
split was used.

¹⁰⁹ EPA does not have information on fab-wide DREs for this time period, so it is not possible to estimate uncontrolled
emissions from fabs that reported point-of-use abatement. These fabs were therefore excluded from the regression analysis.
(They are still included in the national totals.)

¹¹⁰ Only seven gases were aggregated because inclusion of F-GHGs that are not reported in the Inventory results in
overestimation of emission factor that is applied to the various non-reporting subpopulations.

1 assumption for 2007 through 2010, facilities with only R&D activities were assumed to utilize only 20 percent of
2 their manufacturing capacity. All other facilities in the United States are assumed to utilize the average percent of
3 the manufacturing capacity without distinguishing whether fabs produce discrete products or logic products.

4 Non-reporting fabs were then broken out into subpopulations by wafer size (200 mm and 300 mm). using
5 information available through the WFF. The appropriate emission factor was applied to the total TMLA of each
6 subpopulation of non-reporting facilities to estimate the CO₂-equivalent emissions of that subpopulation.

7 Gas-specific, CO₂-equivalent emissions for each subpopulation of non-reporting facilities were estimated using the
8 corresponding reported distribution of gas-specific, CO₂-equivalent emissions from which the aggregate emission
9 factors, based on GHGRP-reported data, were developed. Estimated in this manner, the non-reporting population
10 accounted for 4.9 and 5.0 percent of U.S. emissions in 2011 and 2012, respectively. The GHGRP-reported emissions
11 and the calculated non-reporting population emissions are summed to estimate the total emissions from
12 semiconductor manufacturing.

13 *2013 and 2014*

14 For 2013 and 2014, as for 2011 and 2012, F-GHG and N₂O emissions data received through EPA's GHGRP were
15 aggregated, by gas, across all semiconductor-manufacturing GHGRP reporters to calculate gas-specific emissions
16 for the GHGRP-reporting segment of the U.S. industry. However, for these years WFF data was not available.
17 Therefore, an updated methodology that does not depend on the WFF derived activity data was used to estimate
18 emissions for the segment of the industry that are not covered by EPA's GHGRP. For the facilities that did not
19 report to the GHGRP (i.e., which are below EPA's GHGRP reporting threshold or are R&D facilities), emissions were
20 estimated based on the proportion of total U.S. emissions attributed to non-reporters for 2011 and 2012. EPA used
21 a simple averaging method by first estimating this proportion for both F-GHGs and N₂O for 2011, 2012, and 2015
22 and 2016, resulting in one set of proportions for F-GHGs and one set for N₂O, and then applied the average of each
23 set to the 2013 and 2014 GHGRP reported emissions to estimate the non-reporters' emissions. Fluorinated gas-
24 specific, CO₂-equivalent emissions for non-reporters were estimated using the corresponding reported distribution
25 of gas-specific, CO₂-equivalent emissions reported through EPA's GHGRP for 2013 and 2014.

26 GHGRP-reported emissions in 2013 were adjusted to capture changes to the default emission factors and default
27 destruction or removal efficiencies used for GHGRP reporting, affecting the emissions trend between 2013 and
28 2014. EPA used the same method to make these adjustments as described above for 2011 and 2012 GHGRP data.

29 *2015 through 2022*

30 Similar to the methods described above for 2011 and 2012, and 2013 and 2014, EPA relied upon emissions data
31 reported directly through the GHGRP. For 2015 through 2022, EPA took an approach similar to the one used for
32 2011 and 2012 to estimate emissions for the segment of the semiconductor industry that is below EPA's GHGRP
33 reporting threshold, and for R&D facilities, which are not covered by EPA's GHGRP. However, in a change from
34 previous years, EPA was able to develop new annual emission factors for 2015 through 2022 using TMLA from WFF
35 and a more comprehensive set of emissions, i.e., fabs with as well as without abatement control, as new
36 information about the use of abatement in GHGRP fabs and fab-wide were available. Fab-wide DREs represent
37 total fab CO₂ Eq.-weighted controlled F-GHG and N₂O emissions (emissions after the use of abatement) divided by
38 total fab CO₂ Eq.-weighted uncontrolled F-GHG and N₂O emissions (emission prior to the use of abatement).

39 Using information about reported emissions and the use of abatement and fab-wide DREs, EPA was able to
40 calculate uncontrolled emissions (each total F-GHG and N₂O) for every GHGRP reporting fab. Using this, coupled
41 with TMLA estimated using methods described above (see 2011 through 2012), EPA derived emission factors by
42 year, gas type (F-GHG or N₂O), and wafer size (200 mm and less or 300 mm) by dividing the total annual emissions
43 reported by GHGRP reporters by the total TMLA estimated for those reporters. These emission factors were
44 multiplied by estimates of non-reporter TMLA to arrive at estimates of total F-GHG and N₂O emissions for non-
45 reporters for each year. For each wafer size, the total F-GHG emissions were disaggregated into individual gases

1 using the shares of total emissions represented by those gases in the emissions reported to the GHGRP by
2 unabated fabs producing that wafer size.

3 **Data Sources**

4 GHGRP reporters, which consist of former EPA Partners and non-Partners, estimated their emissions using a
5 default emission factor method established by EPA. Like the Tier 2c Method in the *2019 Refinement to the 2006*
6 *IPCC Guidelines*, this method uses different emission and byproduct generation factors for different F-GHGs and
7 process types and uses factors for different wafer sizes (i.e., 300mm vs. 150 and 200mm) and CVD clean subtypes
8 (in situ thermal, in situ plasma, and remote plasma). Starting with 2014 reported emissions, EPA's GHGRP required
9 semiconductor manufacturers to apply updated emission factors to estimate their F-GHG emissions. For the years
10 2011 through 2013 reported emissions, semiconductor manufacturers used older emission factors to estimate
11 their F-GHG emissions (Federal Register / Vol. 75, No. 230 /December 1, 2010, 74829). Subpart I emission factors
12 were updated for 2014 by EPA as a result of a larger set of emission factor data becoming available as part of the
13 Subpart I petition process, which took place from 2011 through 2013. In addition to semiconductor manufacturing,
14 GHGRP also includes reported emissions from MEMS and PV producers.

15 Historically, semiconductor industry partners estimated and reported their emissions using a range of methods
16 and uneven documentation. It is assumed that most Partners used a method at least as accurate as the IPCC's Tier
17 2a Methodology, recommended in the *2006 IPCC Guidelines*. Partners are estimated to have accounted for
18 between 56 and 79 percent of F-GHG emissions from U.S. semiconductor manufacturing between 1995 and 2010,
19 with the percentage declining in recent years as Partners increasingly implemented abatement measures.

20 Estimates of operating plant capacities and characteristics for Partners and non-Partners were derived from the
21 Semiconductor Equipment and Materials Industry (SEMI) WFF (formerly World Fab Watch) database (1996 through
22 2012, 2013, 2016, 2018, 2021, and 2023) (e.g., Semiconductor Materials and Equipment Industry 2021). Actual
23 worldwide capacity utilizations for 2008 through 2010 were obtained from Semiconductor International Capacity
24 Statistics (SICAS) (SIA 2009 through 2011). Estimates of the number of layers for each linewidth was obtained from
25 International Technology Roadmap for Semiconductors: 2013 Edition (Burton and Beizaie 2001; ITRS 2007; ITRS
26 2008; ITRS 2011; ITRS 2013). PEVM utilized the WFF, SICAS, and ITRS, as well as historical silicon consumption
27 estimates published by VLSI. Actual quarterly U.S. capacity utilizations for 2011, 2012, 2014 to 2022 were obtained
28 from the U.S. Census Bureau's Historical Data Quarterly Survey of Plant Capacity Utilization (USCB 2011, 2012,
29 2015, 2016, 2017, 2018, 2019, 2020, 2021, 2022).

30 Estimates of PV manufacturing capacity, which are used to calculate emissions from non-reporting facilities, are
31 based on data from two sources. A historical market analysis from DisplaySearch provided estimates of U.S.
32 manufacturing capacity from 2000 to 2009 (DisplaySearch 2010). Domestic PV cell production for 2012 was
33 obtained from a Congressional Research Service report titled *U.S. Solar Photovoltaic Manufacturing: Industry*
34 *Trends, Global Competition, Federal Support* (Platzer 2015).

35 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

36 A quantitative uncertainty analysis of this source category was performed using the IPCC-recommended Approach
37 2 uncertainty estimation methodology, the Monte Carlo stochastic simulation technique. The Monte Carlo
38 stochastic simulation was performed on the total emissions estimate from the electronics industry, represented in
39 equation form as:

40 **Equation 4-20: Total Emissions from Electronics Industry**

$$\begin{aligned} 41 \quad & \text{Total Emissions (E}_T\text{)} \\ 42 \quad & = \text{Semiconductors F – GHG and N}_2\text{O Emissions (E}_{\text{Semi}}\text{)} + \text{MEMS F} \\ 43 \quad & \text{– GHG and N}_2\text{O Emissions (E}_{\text{MEMS}}\text{)} + \text{PV F – GHG and N}_2\text{O Emissions (E}_{\text{PV}}\text{)} \\ 44 \quad & + \text{HFC, PFC and SF}_6 \text{ F – HTFs Emissions (E}_{\text{HTF}}\text{)} \end{aligned}$$

1 The uncertainty in the total emissions for the electronics industry, presented in Table 4-121 below, results from
2 the convolution of four distributions of emissions, namely from semiconductors manufacturing, MEMS
3 manufacturing, PV manufacturing and emissions of heat transfer fluids. The approaches for estimating uncertainty
4 in each of the sources are described below:

5 **Semiconductors Manufacture Emission Uncertainty**

6 The Monte Carlo stochastic simulation was performed on the emissions estimate from semiconductor
7 manufacturing, represented in equation form as:

8 **Equation 4-21: Total Emissions from Semiconductor Manufacturing**

$$\begin{aligned} & \text{Semiconductors F – GHG and } N_2\text{O Emissions (} E_{Semi} \text{)} \\ & = \text{GHGRP Reported F – GHG Emissions (} E_{R,F-GHG,Semi} \text{) + Non} \\ & \text{– Reporters' Estimated F – GHG Emissions (} E_{NR,F-GHG,Semi} \text{)} \\ & + \text{GHGRP Reported } N_2\text{O Emissions (} E_{R,N_2O,Semi} \text{) + Non} \\ & \text{– Reporters' Estimated } N_2\text{O Emissions (} E_{NR,N_2O,Semi} \text{)} \end{aligned}$$

9
10
11
12
13
14 The uncertainty in E_{Semi} results from the convolution of four distributions of emissions, $E_{R,F-GHG,Semi}$, $E_{R,N_2O,Semi}$, $E_{NR,F-GHG,Semi}$
15 and $E_{NR,N_2O,Semi}$. The approaches for estimating each distribution and combining them to arrive at the
16 reported 95 percent confidence interval (CI) for E_{Semi} are described in the remainder of this section.

17 The uncertainty estimate of $E_{R,F-GHG,Semi}$, or GHGRP-reported F-GHG emissions, is developed based on gas-specific
18 uncertainty estimates of emissions for two industry segments, one processing 200 mm or less wafers and one
19 processing 300 mm wafers. Uncertainties in emissions for each gas and industry segment are based on an
20 uncertainty analysis conducted during the assessment of emission estimation methods for the Subpart I
21 rulemaking in 2012 (see Technical Support for Modifications to the Fluorinated Greenhouse Gas Emission
22 Estimation Method Option for Semiconductor Facilities under Subpart I, docket EPA–HQ–OAR–2011–0028).¹¹¹ This
23 assessment relied on facility-specific gas information by gas and wafer size, and incorporated uncertainty
24 associated with both emission factors and gas consumption quantities. The 2012 analysis did not consider the use
25 of abatement.

26 For the industry segment that manufactured 200 mm wafers, estimates of uncertainty at a 95 percent CI ranged
27 from ± 29 percent for C_3F_8 to ± 10 percent for CF_4 . For the corresponding 300 mm industry segment, estimates of
28 uncertainty at the 95 percent CI ranged from ± 36 percent for C_4F_8 to ± 16 percent for CF_4 . For gases for which
29 uncertainty was not analyzed in the 2012 assessment (e.g., CH_2F_2), EPA applied the 95 percent CI range equivalent
30 to the range for the gas and industry segment with the highest uncertainty from the 2012 assessment. These gas
31 and wafer-specific uncertainty estimates were developed to represent uncertainty at a facility-level, but they are
32 applied to the total emissions across all the facilities that did not abate emissions as reported under EPA's GHGRP
33 at a national-level. Hence, it is noted that the uncertainty estimates used may be overestimating the uncertainties
34 at a national-level.

¹¹¹ On November 13, 2013, EPA published a final rule revising Subpart I (Electronics Manufacturing) of the GHGRP (78 FR 68162). The revised rule includes updated default emission factors and updated default destruction and removal efficiencies that are slightly different from those that semiconductor manufacturers were required to use to report their 2012 emissions. The uncertainty analyses that were performed during the development of the revised rule focused on these updated defaults but are expected to be reasonably representative of the uncertainties associated with the older defaults, particularly for estimates at the country level. (They may somewhat underestimate the uncertainties associated with the older defaults at the facility level.) For simplicity, the 2012 estimates are assumed to be unbiased although in some cases, the updated (and therefore more representative) defaults are higher or lower than the older defaults. Multiple models and sensitivity scenarios were run for the Subpart I analysis. The uncertainty analysis presented here made use of the Input gas and wafer size model (Model 1) under the following conditions: Year = 2010, f = 20, n = SIA3.

1 For those facilities reporting abatement of emissions under EPA's GHGRP, estimates of uncertainties for the no
2 abatement industry segments are modified to reflect the use of full abatement (abatement of all gases from all
3 cleaning and etching equipment) and partial abatement. These assumptions used to develop uncertainties for the
4 partial and full abatement facilities are identical for 200 mm and 300 mm wafer processing facilities. For all
5 facilities reporting gas abatement, a triangular distribution of destruction or removal efficiency is assumed for each
6 gas. The triangular distributions range from an asymmetric and highly uncertain distribution of zero percent
7 minimum to 90 percent maximum with 70 percent most likely value for CF_4 to a symmetric and less uncertain
8 distribution of 85 percent minimum to 95 percent maximum with 90 percent most likely value for C_4F_8 , NF_3 , and
9 SF_6 . For facilities reporting partial abatement, the distribution of fraction of the gas fed through the abatement
10 device, for each gas, is assumed to be triangularly distributed as well. It is assumed that no more than 50 percent
11 of the gases are abated (i.e., the maximum value) and that 50 percent is the most likely value, and the minimum is
12 zero percent. Consideration of abatement then resulted in four additional industry segments, two 200-mm wafer-
13 processing segments (one fully and one partially abating each gas) and two 300-mm wafer-processing segment
14 (one fully and the other partially abating each gas). Gas-specific emission uncertainties were estimated by
15 convolving the distributions of unabated emissions with the appropriate distribution of abatement efficiency for
16 fully and partially abated facilities using a Monte Carlo simulation.

17 The uncertainty in $E_{R,F-GHG,Semi}$ is obtained by allocating the estimates of uncertainties to the total GHGRP-reported
18 emissions from each of the six industry segments, and then running a Monte Carlo simulation which results in the
19 95 percent CI for emissions from GHGRP-reporting facilities ($E_{R,F-GHG,Semi}$).

20 The uncertainty in $E_{R,N_2O,Semi}$ is obtained by assuming that the uncertainty in the emissions reported by each of the
21 GHGRP reporting facilities results from the uncertainty in quantity of N_2O consumed and the N_2O emission factor
22 (or utilization). Similar to analyses completed for Subpart I (see Technical Support for Modifications to the
23 Fluorinated Greenhouse Gas Emission Estimation Method Option for Semiconductor Facilities under Subpart I,
24 docket EPA-HQ-OAR-2011-0028), the uncertainty of N_2O consumed was assumed to be 20 percent. Consumption
25 of N_2O for GHGRP reporting facilities was estimated by back-calculating from emissions reported and assuming no
26 abatement. The quantity of N_2O utilized (the complement of the emission factor) was assumed to have a triangular
27 distribution with a minimum value of zero percent, mode of 20 percent and maximum value of 84 percent. The
28 minimum was selected based on physical limitations, the mode was set equivalent to the Subpart I default N_2O
29 utilization rate for chemical vapor deposition, and the maximum was set equal to the maximum utilization rate
30 found in ISMI Analysis of Nitrous Oxide Survey Data (ISMI 2009). The inputs were used to simulate emissions for
31 each of the GHGRP reporting, N_2O -emitting facilities. The uncertainty for the total reported N_2O emissions was
32 then estimated by combining the uncertainties of each facilities' reported emissions using Monte Carlo simulation.

33 The estimate of uncertainty in $E_{NR,F-GHG,Semi}$ and $E_{NR,N_2O,Semi}$ entailed developing estimates of uncertainties for the
34 emissions factors and the corresponding estimates of TMLA.

35 The uncertainty in TMLA depends on the uncertainty of two variables—an estimate of the uncertainty in the
36 average annual capacity utilization for each level of production of fabs (e.g., full scale or R&D production) and a
37 corresponding estimate of the uncertainty in the number of layers manufactured. For both variables, the
38 distributions of capacity utilizations and number of manufactured layers are assumed triangular for all categories
39 of non-reporting fabs. The most probable utilization is assumed to be 82 percent, with the highest and lowest
40 utilization assumed to be 89 percent, and 70 percent, respectively. For the triangular distributions that govern the
41 number of possible layers manufactured, it is assumed the most probable value is one layer less than reported in
42 the ITRS; the smallest number varied by technology generation between one and two layers less than given in the
43 ITRS and largest number of layers corresponded to the figure given in the ITRS.

44 The uncertainty bounds for the average capacity utilization and the number of layers manufactured are used as
45 inputs in a separate Monte Carlo simulation to estimate the uncertainty around the TMLA of both individual
46 facilities as well as the total non-reporting TMLA of each sub-population.

47 The uncertainty around the emission factors for non-reporting facilities is dependent on the uncertainty of the
48 total emissions (MMT CO_2 Eq. units) and the TMLA of each reporting facility in that category. For each wafer size

1 for reporting facilities, total emissions were regressed on TMLA (with an intercept forced to zero) for 10,000
2 emission and 10,000 TMLA values in a Monte Carlo simulation, which results in 10,000 total regression coefficients
3 (emission factors). The 2.5th and the 97.5th percentile of these emission factors are determined, and the bounds
4 are assigned as the percent difference from the estimated emission factor.

5 The next step in estimating the uncertainty in emissions of reporting and non-reporting facilities in semiconductor
6 manufacture is convolving the distribution of reported emissions, emission factors, and TMLA using Monte Carlo
7 simulation. For this Monte Carlo simulation, the distributions of the reported F-GHG gas- and wafer size-specific
8 emissions are assumed to be normally distributed, and the uncertainty bounds are assigned at 1.96 standard
9 deviations around the estimated mean. There were some instances, though, where departures from normality were
10 observed for variables, including for the distributions of the gas- and wafer size-specific N₂O emissions, TMLA, and
11 non-reporter emission factors, both for F-GHGs and N₂O. As a result, the distributions for these parameters were
12 assumed to follow a pert beta distribution.

13 MEMS Manufacture Emission Uncertainty

14 The Monte Carlo stochastic simulation was performed on the emissions estimate from MEMS manufacturing,
15 represented in equation form as:

16 Equation 4-22: Total Emissions from MEMS Manufacturing

$$17 \quad \text{MEMS F-GHG and N}_2\text{O Emissions (E}_{\text{MEMS}}) = \text{GHGRP Reported F-GHG Emissions (E}_{\text{R,F-GHG, MEMS}}) + \text{GHGRP} \\ 18 \quad \quad \quad \text{Reported N}_2\text{O Emissions (E}_{\text{R,N}_2\text{O, MEMS}})$$

$$19 \quad \quad \quad \text{MEMS F – GHG and N}_2\text{O Emissions (E}_{\text{MEMS}}) \\ 20 \quad \quad \quad = \text{GHGRP Reported F – GHG Emissions (E}_{\text{R,F-GHG, MEMS}}) \\ 21 \quad \quad \quad + \text{GHGRP Reported N}_2\text{O Emissions (E}_{\text{R,N}_2\text{O, MEMS}})$$

22 Emissions from MEMS manufacturing are only quantified for GHGRP reporters. MEMS manufacturers that report
23 to the GHGRP all report the use of 200 mm wafers. Some MEMS manufacturers report using abatement
24 equipment. Therefore, the estimates of uncertainty at the 95 percent CI for each gas emitted by MEMS
25 manufacturers are set equal to the gas-specific uncertainties for manufacture of 200mm semiconductor wafers
26 with partial abatement. The same assumption is applied for uncertainty levels for GHGRP reported MEMS N₂O
27 emissions (E_{R,N₂O, MEMS}).

28 PV Manufacture Emission Uncertainty

29 The Monte Carlo stochastic simulation was performed on the emissions estimate from PV manufacturing,
30 represented in equation form as:

31 Equation 4-23: Total Emissions from PV Manufacturing

$$32 \quad \text{PV F-GHG and N}_2\text{O Emissions (E}_{\text{PV}}) = \text{Non-Reporters' Estimated F-GHG Emissions (E}_{\text{NR,F-GHG,PV}}) + \text{Non-} \\ 33 \quad \quad \quad \text{Reporters' Estimated N}_2\text{O Emissions (E}_{\text{NR,N}_2\text{O,PV}})$$

$$34 \quad \quad \quad \text{PV F – GHG and N}_2\text{O Emissions (E}_{\text{PV}}) \\ 35 \quad \quad \quad = \text{Non – Reporters' Estimated F – GHG Emissions (E}_{\text{NR,F-GHG,PV}}) + \text{Non} \\ 36 \quad \quad \quad \text{– Reporters' Estimated N}_2\text{O Emissions (E}_{\text{NR,N}_2\text{O,PV}})$$

37 Emissions from PV manufacturing are only estimated for non-GHGRP reporters. There were no reported emissions
38 from PV manufacturing in GHGRP in 2021. The “Non-Reporters’ Estimated F-GHG Emissions” term in Equation 4-23
39 was estimated using an emission factor developed using emissions from reported data in 2015 and 2016 and total
40 non-reporters’ capacity. Due to a lack of information and data and because they represent similar physical and
41 chemical processes, the uncertainty at the 95 percent CI level for non-reporter PV capacity is assumed to be the
42 same as the uncertainty in non-reporter TMLA for semiconductor manufacturing. Similarly, the uncertainty for the

1 PV manufacture emission factors are assumed to be the same as the uncertainties in emission factors used for
 2 non-reporters in semiconductor manufacture.

3 **Heat Transfer Fluids Emission Uncertainty**

4 There is a lack of data related to the uncertainty of emission estimates of heat transfer fluids used for electronics
 5 manufacture. Therefore, per the *2006 IPCC Guidelines* (IPCC 2006, Volume 3, Chapter 6), uncertainty bounds of 20
 6 percent were applied to estimate uncertainty associated with the various types of heat transfer fluids, including
 7 PFCs, HFC, and SF₆, at the national level.

8 The results of the Approach 2 quantitative uncertainty analysis for electronics manufacturing are summarized in
 9 Table 4-121. These results were obtained by convolving—using Monte Carlo simulation—the distributions of
 10 emissions for each reporting and non-reporting facility that manufactures semiconductors, MEMS, or PVs and use
 11 heat transfer fluids. The emissions estimate for total U.S. F-GHG, N₂O, and HTF emissions from electronics
 12 manufacturing were estimated to be between 4.88 and 5.50 MMT CO₂ Eq. at a 95 percent CI level. This range
 13 represents 6 percent below to 6 percent above the 2021 emission estimate of 5.19 MMT CO₂ Eq. for all emissions
 14 from electronics manufacture. This range and the associated percentages apply to the estimate of total emissions
 15 rather than those of individual gases. Uncertainties associated with individual gases will be somewhat higher than
 16 the aggregate but were not explicitly modeled.

17 **Table 4-121: Approach 2 Quantitative Uncertainty Estimates for HFC, PFC, SF₆, NF₃ and N₂O**
 18 **Emissions from Electronics Manufacture (MMT CO₂ Eq. and Percent)**

Source	Gas	2021 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			Lower Bound ^b	Upper Bound ^b	Lower Bound (%)	Upper Bound (%)
Electronics Industry	HFC, PFC, SF ₆ , NF ₃ , and N ₂ O	4.8	4.5	5.1	-6%	+6%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

^b Absolute lower and upper bounds were calculated using the corresponding lower and upper bounds in percentages.

19 **QA/QC and Verification**

20 For its GHGRP data, EPA verifies annual facility-level reports through a multi-step process (e.g., including a
 21 combination of pre-and post-submittal electronic checks and manual reviews by staff) to identify potential errors
 22 and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).¹¹² Based on the results
 23 of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-
 24 submittals checks are consistent with a number of general and category-specific QC procedures including range
 25 checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.

26 For more information on the general QA/QC process applied to this source category, consistent with Volume 1,
 27 Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in the introduction of
 28 the IPPU chapter and Annex 8 for more details.

¹¹² GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 Recalculations Discussion

2 Any resubmitted emissions data reported to EPA’s GHGRP from all prior years were updated in this *Inventory*.
3 Additionally, EPA made the following changes:

- 4 • To estimate non-reporter F-GHG and N₂O emissions, EPA relies on data reported through Subpart I and
5 the World Fab Forecast. This process requires EPA to map facilities that report through Subpart I and
6 which are also represented in the World Fab Forecast. For this *Inventory* update, EPA identified and made
7 corrections to a few instances of this mapping based on new information and additional reviews of the
8 data. This had minimal effects on emission estimates.
- 9 • EPA re-ran regression analyses for years 2010 to 2021 to reflect updates to Subpart I and the World Fab
10 Forecast. These changes had minor effects on the emission factors, standard error, and R² values for all
11 years. This resulted in the recalculation of non-reporter’s F-GHG and N₂O estimates for all years.
- 12 • EPA recalculated HTF emissions for years 1990 to 2021 using IPCC *Fifth Assessment Report* (AR5) GWP
13 values (IPCC 2013). Emission estimates were incorrectly calculated using the IPCC *Fourth Assessment*
14 *Report* (AR4) GWP values (IPCC 2007) in the previous *Inventory* (EPA 2023). Overall, the impact of these
15 recalculations led to an average decrease of 0.009 MMT CO₂ Eq. (0.19 percent) across the time series
16 (1990 through 2021).
- 17 • EPA recalculated fluorinated GHG consumption for 2014 to 2021 using the following methodology.
18 Fluorinated GHG consumption estimates for unabated fabs were calculated using reported GHGRP
19 emissions data and default emission factors for Subpart I. Because certain fluorinated GHGs are
20 generated as by-products as well as used as input gases, both input gas emission factors and by-product
21 gas emission factors were factored into this calculation. For abated fabs, a “consumption factor” was
22 developed by dividing the reported emissions of each fluorinated GHG from unabated fabs by the
23 estimated consumption of each fluorinated GHG for each wafer size. Fluorinated GHG consumption for
24 2021 was estimated based on GDP growth of the 2020 consumption estimate. The consumption estimate
25 will be updated with reported 2021 GHGRP emissions data.
- 26 • EPA refined the non-reporting population for 2015 to 2022 by conducting an analysis into the criteria
27 being used to determine which fabs should be included and excluded from this population. Overall, the
28 impact of this refinement led to an average increase in semiconductor emissions by 0.02 MMT CO₂ Eq.
29 (0.45 percent) for the time series 2015 to 2022.
- 30 • EPA recalculated non-reporter emissions for 2015 to 2022 by developing emission factors for individual
31 gases and calculated on an MT basis. Overall, the impact of this refinement led to an average increase in
32 semiconductor emissions by 0.04 MMT CO₂ Eq. (0.84 percent) for the time series 2014 to 2022.

33 Planned Improvements

34 The *Inventory* methodology uses data reported through the EPA Partnership (for earlier years) and EPA’s GHGRP
35 (for later years) to extrapolate the emissions of the non-reporting population. While these techniques are well
36 developed, the accuracy of the emissions estimates for the non-reporting population could be further increased
37 through EPA’s further investigation of and improvement upon the accuracy of estimated activity in the form of
38 TMLA.

39 The *Inventory* uses utilization from two different sources for various time periods—SEMI to develop PEVM and to
40 estimate non-Partner emissions for the period 1995 to 2010 and U.S. Census Bureau for 2011 through 2022. SEMI
41 reported global capacity utilization for manufacturers through 2011. U.S. Census Bureau capacity utilization
42 include U.S. semiconductor manufacturers as well as assemblers. Further analysis on the impacts of using a new
43 and different source of utilization data could prove to be useful in better understanding of industry trends and
44 impacts of utilization data sources on historical emission estimates.

1 Estimates of semiconductor non-reporter and non-Partner emissions are based on EPA-developed emission factors
 2 for the time periods pre-2010, 2011 through 2012, and 2015 through 2022. Based on the data available for these
 3 time periods, the methods used to develop emission factors for non-reporters and non-Partners are slightly
 4 inconsistent for semiconductors (e.g., how data representing emissions and TMLA from the manufacture of various
 5 wafer sizes are aggregated or disaggregated for purposes of calculating emission factors). Further analyses to
 6 support potentially adjusting the methods for developing these emission factors could be done to better ensure
 7 consistency across the time series.

8 The methodology for estimating semiconductor emissions from non-reporters uses data from the International
 9 Technology Roadmap for Semiconductors (ITRS) on the number of layers associated with various technology node
 10 sizes. The ITRS has now been replaced by the International Roadmap for Devices and Systems (IRDS), which has
 11 published updated data on the number of layers used in each device type and node size (in nanometers).
 12 Incorporating this updated dataset will improve the accuracy of emissions estimates from non-reporting
 13 semiconductor fabs.

14 4.25 Substitution of Ozone Depleting 15 Substances (CRT Source Category 2F)

16 Hydrofluorocarbons (HFCs), perfluorocarbons (PFCs), and carbon dioxide (CO₂) are used as alternatives to several
 17 classes of ozone-depleting substances (ODSs) that are being phased out under the terms of the *Montreal Protocol*
 18 and the Clean Air Act Amendments of 1990.¹¹³ Ozone-depleting substances—chlorofluorocarbons (CFCs), halons,
 19 carbon tetrachloride, methyl chloroform, and hydrochlorofluorocarbons (HCFCs)—are used in a variety of
 20 industrial applications including refrigeration and air conditioning equipment, solvent cleaning, foam production,
 21 sterilization, fire extinguishing, and aerosols. Although HFCs and PFCs are not harmful to the stratospheric ozone
 22 layer, they are potent greenhouse gases. On December 27, 2020, the American Innovation and Manufacturing
 23 (AIM) Act was enacted by Congress and directs EPA to address HFCs by phasing down production and consumption
 24 (i.e., production plus import minus export), maximizing reclamation and minimizing releases from equipment, and
 25 facilitating the transition to next-generation technologies through sector-based restrictions. Emission estimates for
 26 HFCs, PFCs, and CO₂ used as substitutes for ODSs are provided in Table 4-122 and Table 4-123.¹¹⁴

27 **Table 4-122: Emissions of HFCs, PFCs, and CO₂ from ODS Substitutes (MMT CO₂ Eq.)**

Gas	1990	2005	2018	2019	2020	2021	2022
HFC-23	0.0	+	+	+	+	+	+
HFC-32	0.0	0.3	6.1	6.9	7.8	9.4	10.5
HFC-125	+	8.2	48.8	53.1	57.8	66.3	72.0
HFC-134a	+	72.8	56.4	55.3	54.1	50.0	48.3
HFC-143a	+	10.0	29.7	29.9	29.9	30.0	29.8
HFC-236fa	0.0	1.0	0.8	0.7	0.7	0.6	0.6
CF ₄	0.0	+	+	+	+	+	0.1
CO ₂	+	+	+	+	+	+	+
Other Saturated HFCs ^a	0.3	6.9	15.9	16.0	15.9	16.3	16.8
Other PFCs and HFOs ^b	+	0.1	+	+	+	+	+

¹¹³ [42 U.S.C § 7671, CAA Title VI].

¹¹⁴ Emissions of ODSs are not included here consistent with UNFCCC reporting guidelines for national inventories noted in Box 4-1. See Annex 6.2 for more details on emissions of ODSs. Emissions from CO₂ used in the food and beverage industry are separately reported in Chapter 4.15 Carbon Dioxide Consumption but does not include CO₂ in ODS substitute use sectors as a refrigerant, foam blowing agent, or fire extinguishing agent.

Total	0.3	99.5	157.9	162.1	166.2	172.7	178.1
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+ Does not exceed 0.05 MMT CO₂ Eq.

^a Other Saturated HFCs represents an unspecified mix of saturated HFCs, which includes HFC-152a, HFC-227ea, HFC-245fa, HFC-365mfc, and HFC-43-10mee.

^b Other PFCs and HFOs represents an unspecified mix of PFCs and HFOs, which includes HCFO-1233zd(E), HFO-1234yf, HFO-1234ze(E), HFO-1336mzz(Z), C₄F₁₀, and PFC/PFPEs, the latter being a proxy for a diverse collection of PFCs and perfluoropolyethers (PFPEs) employed for solvent applications. For estimating purposes, the GWP value used for PFC/PFPEs was based upon n-C₆F₁₄.

Note: Totals may not sum due to independent rounding.

1 **Table 4-123: Emissions of HFCs, PFCs, and CO₂ from ODS Substitution (Metric Tons)**

Gas	1990	2005	2018	2019	2020	2021	2022
HFC-23	0	1	2	2	2	2	3
HFC-32	0	397	9,008	10,156	11,461	13,958	15,582
HFC-125	+	2,580	15,406	16,761	18,240	20,909	22,704
HFC-134a	+	56,029	43,419	42,558	41,590	38,447	37,167
HFC-143a	+	2,093	6,188	6,230	6,234	6,240	6,203
HFC-236fa	0	127	99	91	84	78	72
CF ₄	0	3	5	5	4	4	4
CO ₂	14	1,325	3,093	3,303	3,516	3,734	3,969
Other Saturated HFCs ^a	M	M	M	M	M	M	M
Other PFCs and HFOs ^b	M	M	M	M	M	M	M

+ Does not exceed 0.5 MT.

M (Mixture of Gases).

^a Other Saturated HFCs represents an unspecified mix of saturated HFCs, which includes HFC-152a, HFC-227ea, HFC-245fa, HFC-365mfc, and HFC-43-10mee.

^b Other PFCs and HFOs represents an unspecified mix of PFCs and HFOs, which includes HCFO-1233zd(E), HFO-1234yf, HFO-1234ze(E), HFO-1336mzz(Z), C₄F₁₀, and PFC/PFPEs, the latter being a proxy for a diverse collection of PFCs and perfluoropolyethers (PFPEs) employed for solvent applications. For estimating purposes, the GWP value used for PFC/PFPEs was based upon n-C₆F₁₄.

2 In 1990 and 1991, the only significant emissions of HFCs and PFCs as substitutes to ODSs were relatively small
3 amounts of HFC-152a—used as an aerosol propellant and also a component of the refrigerant blend R-500 used in
4 chillers. Beginning in 1992, HFC-134a was used in growing amounts as a refrigerant in motor vehicle air-
5 conditioners and in refrigerant blends such as R-404A.¹¹⁵ In 1993, the use of HFCs in foam production began, and
6 in 1994 ODS substitutes for halons entered widespread use in the United States as halon production was phased
7 out. In 1995, these compounds also found applications as solvents. Non-fluorinated ODS substitutes, such as CO₂,
8 have been used in place of ODS in certain foam production and fire extinguishing uses since the 1990s.

9 The use and subsequent emissions of HFCs, PFCs, and CO₂ as ODS substitutes has been increasing from small
10 amounts in 1990 to 178.1 MMT CO₂ Eq. emitted in 2022. This increase was in large part the result of efforts to
11 phase out CFCs, HCFCs, and other ODSs in the United States. Use and emissions of HFCs are expected to start
12 decreasing in the next few years and continue downward as production and consumption of HFCs are phased
13 down to 15 percent of their baseline levels by 2036 through an allowance allocation and trading program
14 established by EPA. Improvements in recovery practices and the use of alternative gases and technologies, through
15 voluntary actions and in response to existing and potential future regulations under the AIM Act, will also
16 contribute to a reduction in HFC use and emissions.

¹¹⁵ R-404A contains HFC-125, HFC-143a, and HFC-134a.

1 Table 4-124 presents emissions of HFCs, PFCs, and CO₂ as ODS substitutes by end-use sector for 1990 through
 2 2022. The refrigeration and air-conditioning sector is further broken down by sub-sector. The end-use sectors that
 3 contributed the most toward emissions of HFCs, PFCs, and CO₂ as ODS substitutes in 2022 include refrigeration and
 4 air-conditioning (144.6 MMT CO₂ Eq., or approximately 81 percent), aerosols (17.0 MMT CO₂ Eq., or approximately
 5 10 percent), and foams (11.7 MMT CO₂ Eq., or approximately 7 percent). Within the refrigeration and air-
 6 conditioning end-use sector residential unitary AC, part of the Residential Stationary Air-conditioning subsector
 7 shown below, was the highest emitting end-use (40.8 MMT CO₂ Eq.), followed by large retail food, which is part of
 8 the Commercial Refrigeration subsector. Each of the end-use sectors is described in more detail below.

9 **Table 4-124: Emissions of HFCs, PFCs, and CO₂ from ODS Substitutes (MMT CO₂ Eq.) by Sector**

Sector	1990	2005	2018	2019	2020	2021	2022
Refrigeration/Air							
Conditioning	+	83.0	122.7	126.5	130.6	139.5	144.6
Commercial Refrigeration	+	14.9	39.6	40.2	40.6	41.0	41.4
Domestic Refrigeration	+	0.2	1.2	1.2	1.2	1.1	1.0
Industrial Process							
Refrigeration	+	1.8	13.8	15.0	16.2	17.4	18.6
Transport Refrigeration	+	1.6	6.9	7.4	7.9	8.4	8.8
Mobile Air Conditioning	+	61.5	28.7	26.6	24.6	22.9	20.8
Residential Stationary Air							
Conditioning	+	1.2	26.2	29.4	33.2	41.5	46.4
Commercial Stationary Air							
Conditioning	+	1.7	6.2	6.6	6.9	7.3	7.6
Aerosols	0.2	10.2	16.7	17.0	17.3	17.7	17.0
Foams	+	3.5	14.2	14.1	13.7	10.8	11.7
Solvents	+	1.6	2.0	2.0	2.0	2.1	2.1
Fire Protection	+	1.2	2.4	2.5	2.5	2.6	2.6
Total	0.3	99.5	157.9	162.1	166.2	172.7	178.1

10 + Does not exceed 0.05 MMT CO₂ Eq.

11 Note: Totals may not sum due to independent rounding.

12 Refrigeration/Air Conditioning

13 The refrigeration and air-conditioning sector includes a wide variety of equipment types that have historically used
 14 CFCs or HCFCs. End-uses within this sector include motor vehicle air-conditioning, retail food refrigeration,
 15 refrigerated transport (e.g., ship holds, truck trailers, railway freight cars), household refrigeration, residential and
 16 small commercial air-conditioning and heat pumps, chillers (large comfort cooling), cold storage facilities, and
 17 industrial process refrigeration (e.g., systems used in food processing, chemical, petrochemical, pharmaceutical, oil
 18 and gas, metallurgical, and other industries). As the ODS phaseout has taken effect, most equipment has been
 19 retrofitted or replaced to use HFC-based substitutes. Common HFCs in use today in refrigeration/air-conditioning
 20 equipment are HFC-134a, R-410A,¹¹⁶ R-404A, and R-507A.¹¹⁷ Lower-GWP options such as hydrofluoroolefin (HFO)-
 21 1234yf in motor vehicle air-conditioning, R-717 (ammonia) in cold storage and industrial applications, and R-744
 22 (carbon dioxide) and HFC/HFO blends in retail food refrigeration, are also being used. Manufacturers of residential
 23 and commercial air conditioning have announced their plans to use HFC-32 and R-454B¹¹⁸ in the future, and at
 24 least one manufacturer has announced the availability of chillers operating on HFC-32 as of 2023 (Carrier, 2023).
 25 These refrigerants are emitted to the atmosphere during equipment operation (as a result of component failure,

¹¹⁶ R-410A contains HFC-32 and HFC-125.

¹¹⁷ R-507A, also called R-507, contains HFC-125 and HFC-143a.

¹¹⁸ R-454B contains HFC-32 and HFO-1234yf.

1 leaks, and purges), as well as at manufacturing (if charged at the factory), installation, servicing, and disposal
2 events.

3 **Aerosols**

4 Aerosol propellants are used in metered dose inhalers (MDIs) and a variety of personal care products and
5 technical/specialty products (e.g., duster sprays and safety horns). Pharmaceutical companies that produce MDIs—
6 a type of inhaled therapy used to treat asthma and chronic obstructive pulmonary disease—have replaced the use
7 of CFCs with HFC-propellant alternatives. The earliest ozone-friendly MDIs were produced with HFC-134a, but the
8 industry is using HFC-227ea as well. Conversely, since the use of CFC propellants in other types of aerosols was
9 banned in the United States in 1978, most non-medical consumer aerosol products have not transitioned to HFCs,
10 but to “not-in-kind” technologies, such as solid or roll-on deodorants and finger-pump sprays. The transition away
11 from ODSs in specialty aerosol products has also led to the introduction of non-fluorocarbon alternatives (e.g.,
12 hydrocarbon propellants) in certain applications, in addition to HFC-134a or HFC-152a. Other low-GWP options
13 such as HFO-1234ze(E) are being used as well. These propellants are released into the atmosphere as the aerosol
14 products are used.

15 **Foams**

16 Chlorofluorocarbons and HCFCs have traditionally been used as foam blowing agents to produce polyurethane
17 (PU), polystyrene, polyolefin, and phenolic foams, which are used in a wide variety of products and applications.
18 Since the *Montreal Protocol*, flexible PU foams as well as other types of foam, such as polystyrene sheet,
19 polyolefin, and phenolic foam, have transitioned almost completely away from fluorocompounds into alternatives
20 such as CO₂ and hydrocarbons. The majority of rigid PU foams have transitioned to HFCs—primarily HFC-134a and
21 HFC-245fa. Today, these HFCs are used to produce PU appliance, PU commercial refrigeration, PU spray, and PU
22 panel foams—used in refrigerators, vending machines, roofing, wall insulation, garage doors, and cold storage
23 applications. In addition, HFC-152a, HFC-134a, and CO₂ are used to produce polystyrene sheet/board foam, which
24 is used in food packaging and building insulation. Low-GWP fluorinated foam blowing agents in use include HFO-
25 1234ze(E) and HCFO-1233zd(E). Emissions of blowing agents occur when the foam is manufactured as well as
26 during the foam lifetime and at foam disposal, depending on the particular foam type.

27 **Solvents**

28 Chlorofluorocarbons, methyl chloroform (1,1,1-trichloroethane or TCA), and to a lesser extent carbon tetrachloride
29 (CCl₄) were historically used as solvents in a wide range of cleaning applications, including precision, electronics,
30 and metal cleaning. Since their phaseout, metal cleaning end-use applications have primarily transitioned to non-
31 fluorocarbon solvents and not-in-kind processes. The precision and electronics cleaning end-uses have transitioned
32 in part to high-GWP gases, due to their high reliability, excellent compatibility, good stability, low toxicity, and
33 selective solvency. These applications rely on HFC-43-10mee, HFC-365mfc, HFC-245fa, and to a lesser extent, PFCs.
34 Electronics cleaning involves removing flux residue that remains after a soldering operation for printed circuit
35 boards and other contamination-sensitive electronics applications. Precision cleaning may apply to either
36 electronic components or to metal surfaces, and is characterized by products, such as disk drives, gyroscopes, and
37 optical components, that require a high level of cleanliness and generally have complex shapes, small clearances,
38 and other cleaning challenges. The use of these solvents yields fugitive emissions of these HFCs and PFCs.

39 **Fire Protection**

40 Fire protection applications include portable fire extinguishers (“streaming” applications) that originally used halon
41 1211, and total flooding applications that originally used halon 1301, as well as some halon 2402. Since the
42 production and import of virgin halons were banned in the United States in 1994, the halon replacement agent of
43 choice in the streaming sector has been dry chemical, although HFC-236fa is also used to a limited extent. In the
44 total flooding sector, HFC-227ea has emerged as the primary replacement for halon 1301 in applications that

1 require clean agents. Other HFCs, such as HFC-23 and HFC-125, are used in smaller amounts. The majority of HFC-
2 227ea in total flooding systems is used to protect essential electronics, as well as in civil aviation, military mobile
3 weapons systems, oil/gas/other process industries, and merchant shipping. Fluoroketone FK-5-1-12 is also used as
4 a low-GWP option and 2-BTP is being use in niche applications. As fire protection equipment is tested or deployed,
5 emissions of these fire protection agents occur.

6 Methodology and Time-Series Consistency

7 Using a Tier 2 method in accordance with the IPCC methodological decision tree, a detailed Vintaging Model of
8 ODS-containing equipment and products was used to estimate the actual—versus potential—emissions of various
9 ODS substitutes, including HFCs, PFCs, and CO₂. The name of the model refers to the fact that it tracks the use and
10 emissions of various compounds for the annual “vintages” of new equipment that enter service in each end-use.
11 The Vintaging Model predicts ODS and ODS substitute use in the United States based on modeled estimates of the
12 quantity of equipment or products sold each year containing these chemicals and the amount of the chemical
13 required to manufacture and/or maintain equipment and products over time. Emissions for each end-use were
14 estimated by applying annual leak rates and release profiles, which account for the lag in emissions from
15 equipment as they leak over time. By aggregating the data for 80 different end-uses, the model produces
16 estimates of annual use and emissions of each compound. Further information on the Vintaging Model is
17 contained in Annex 3.9.

18 Methodological approaches were applied to the entire time series to ensure time-series consistency from 1990
19 through 2022.

20 Uncertainty – TO BE UPDATED FOR FINAL REPORT

21 Given that emissions of ODS substitutes occur from thousands of different kinds of equipment and from millions of
22 point and mobile sources throughout the United States, emission estimates must be made using analytical tools
23 such as the Vintaging Model or the methods outlined in IPCC (2006). Though the model is more comprehensive
24 than the IPCC default methodology, significant uncertainties still exist with regard to the levels of equipment sales,
25 equipment characteristics, and end-use emissions profiles that were used to estimate annual emissions for the
26 various compounds.

27 The uncertainty analysis quantifies the level of uncertainty associated with the aggregate emissions across the 80
28 end-uses in the Vintaging Model. In order to calculate uncertainty, functional forms were developed to simplify
29 some of the complex “vintaging” aspects of some end-use sectors, especially with respect to refrigeration and air-
30 conditioning, and to a lesser degree, fire extinguishing. These sectors calculate emissions based on the entire
31 lifetime of equipment, not just equipment put into commission in the current year, thereby necessitating
32 simplifying equations. The functional forms used variables that included growth rates, emission factors, transition
33 from ODSs, change in charge size as a result of the transition, disposal quantities, disposal emission rates, and
34 either stock (e.g., number of air conditioning units in operation) for the current year or ODS consumption before
35 transition to alternatives began (e.g., in 1985 for most end-uses). Uncertainty was estimated around each variable
36 within the functional forms based on expert judgment, and a Monte Carlo analysis was performed.

37 The most significant sources of uncertainty for the ODS Substitutes source category include the total stock of
38 refrigerant installed in industrial process refrigeration and cold storage equipment, as well as the charge size for
39 technical aerosols using HFC-134a.

40 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 4-125. Substitution of
41 ozone depleting substances HFC and PFC emissions were estimated to be between 165.2 and 197.8 MMT CO₂ Eq.
42 at the 95 percent confidence level. This indicates a range of approximately 4.2 percent below to 14.7 percent
43 above the emission estimate of 172.5 MMT CO₂ Eq.

Table 4-125: Approach 2 Quantitative Uncertainty Estimates for HFC and PFC Emissions from ODS Substitutes (MMT CO₂ Eq. and Percent)

Source	Gases	2021 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower	Upper	Lower	Upper
			Bound	Bound	Bound	Bound
Substitution of Ozone Depleting Substances	HFCs and PFCs	172.5	165.2	197.8	-4.2%	+14.7%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

QA/QC and Verification

For more information on the general QA/QC process applied to this source category, consistent with Volume 1, Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in the introduction of the IPPU chapter. Category specific QA/QC findings are described below.

The QA and verification process for individual gases and sources in the Vintaging Model includes review against up-to-date market information, including equipment stock estimates, leak rates, and sector transitions to new chemicals and technologies. In addition, comparisons against published emission and consumption sources by gas and by source are performed when available as described further below. Independent peer reviews of the Vintaging Model are periodically performed, including one conducted in 2017 (EPA 2018), to confirm Vintaging Model estimates and identify updates. For the purposes of reporting emissions to protect Confidential Business Information (CBI), some HFCs and PFCs are grouped into two unspecified mixes of saturated HFCs and other PFCs and HFOs. The HFCs and PFCs within the unspecified mix of HFCs and PFCs are modelled and verified individually in the same process as all other gases and sources in the Vintaging Model.

Data from EPA’s Greenhouse Gas Reporting Program (GHGRP)¹¹⁹ and emissions of some fluorinated greenhouse gases estimated for the contiguous United States by scientists at the National Oceanic and Atmospheric Administration (NOAA) were used to perform additional quality control as specified in *2006 IPCC Guidelines for National Greenhouse Gas Inventories* and the *2019 Refinement to the 2006 IPCC Guidelines for National Greenhouse Gas Inventories (IPCC 2019)*. These comparisons are detailed further in Annex 3.9.

Recalculations Discussion

For the current *Inventory*, updates to the Vintaging Model included updating 2022 growth rates for residential and commercial unitary air-conditioning to align with annual sales estimates published by AHRI. Projected growth rates were updated for residential unitary air-conditioning to align with projected residential housing available from the Energy Information Administration (EIA) and commercial unitary air-conditioning growth rates were updated

¹¹⁹ For the GHGRP data, EPA verifies annual facility-level and company-level reports through a multi-step process (e.g., including a combination of pre-and post-submittal electronic checks and manual reviews by staff) to identify potential errors and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015). Based on the results of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-submittals checks are consistent with a number of general and category-specific QC procedures, including range checks, statistical checks, algorithm checks, and year-to-year checks of reported data.

1 based on new commercial floorspace growth projections from EIA (EPA 2023a). Growth rates for window units
2 were updated to align with sales data for Energy Star- and non-Energy Star-certified units (EPA 2023b).

3 The Vintaging Model was also updated to include the addition of two end-uses representing multi-split air-
4 conditioning units: small ductless mini-split and multi-split air-conditioning and large ductless mini-split and multi-
5 split air-conditioning. These end-uses were split from the existing residential unitary air-conditioning end-use.
6 Stock for ductless mini-split and multi-split air-conditioning systems were estimated based on the EIA Residential
7 Energy and Consumption Survey (RECS) and growth projected based on annual sales of split systems under 33,000
8 BTU/h (small mini/multi-splits) and split systems between 33,000 BTU/h and 65,000 BTU/h (large mini/multi-splits)
9 published by AHRI (EPA 2023c).

10 In addition, the market size and growth rates of the streaming agents end-use in the Fire Suppression sector was
11 updated, improving alignment of halon 1211 consumption with the 2022 Fire Suppression Technical Options
12 Committee (FSTOC) estimates, the lifetime was adjusted to reflect internal inspection timelines, rather than
13 physical extinguisher lifetimes, and refilling of leaks throughout the lifetime of extinguishers was modeled (EPA
14 2023d).

15 Together, these updates increased ODS substitute emissions on average by 0.06 MMT CO₂ Eq. (0.6 percent)
16 between 1990 and 2021, compared to the previous (i.e., 2023) *Inventory* submission.

17 **Planned Improvements**

18 Future improvements to the Vintaging Model are planned for the Refrigeration and Air-conditioning, Fire
19 Suppression, and Aerosols sectors. Specifically, refrigerated storage space estimates published biannually from the
20 United States Department of Agriculture (USDA) are being compared to cold storage warehouse space currently
21 estimated in the Vintaging Model. Flooding agent fire suppression market transitions are under review to align
22 more closely with real world activities. In addition, further refinement of HFC consumption in MDIs is expected
23 from review of data collected on HFC use for MDI production, imports, and exports in response to requests
24 pursuant to AIM Act regulations for application-specific allowances for MDIs. EPA expects these revisions to be
25 prepared for the 2024 or 2025 *Inventory* submission.

26 As discussed above, future reporting under the AIM Act may provide useful information for verification purposes
27 and possible improvements to the Vintaging Model, such as information on HFC stockpiling behaviors. EPA expects
28 this reporting by late 2023 and incorporation into the 2025 or 2026 report. Should the data suggest structural
29 changes to the model, such as the handling of stockpiles before use, EPA expects to introduce the revised model
30 for the 2025 or 2026 *Inventory* submission.

31 Several potential improvements to the *Inventory* were identified in the 2022 *Inventory* submission based on the
32 comparisons mentioned above and discussed in Annex 3.9—net supply values from the GHGRP and emission
33 estimates derived from atmospheric measurements—and remain valid. To estimate HFC emissions for just the
34 contiguous United States, matching the coverage by the atmospheric measurements, EPA will investigate the
35 availability of data from Alaska, Hawaii, and U.S. territories. This is planned by the next (i.e., 2025) *Inventory*
36 submission. To improve estimates of HFC-125 and HFC-143a, further research into the refrigeration market can be
37 made. Research in this industry on the shift away from blends such as R-404A or success in lowering emission rates
38 could be used to improve the *Inventory* estimate. This is planned for the 2025 *Inventory*. That said, for the years
39 where both the atmospheric measurements and the model display a roughly constant emission of HFC-143a at
40 similar levels, the new results suggest robust estimates for the refrigeration market. Uncertainty estimates by
41 species would aid in comparisons to atmospheric data. EPA continues to explore the possibility of revising the
42 Monte Carlo analysis to differentiate between species, starting with the higher-emitted HFCs identified above, in a
43 future (i.e., 2025) *Inventory* submission. Reclamation reports and, when available, information gathered under the
44 AIM Act, could be used to improve the understanding of how chemical moves through the economy and could
45 resolve some of the temporal effects discussed in Annex 3.9. This would likely require revisions to the basic model
46 structure and could be introduced for the 2026 or 2027 *Inventory* submission. The additional data from the
47 atmospheric measurements suggests additional items to investigate. The faster uptick in HFC-32 and HFC-125

1 emissions suggests additional emissions of R-410A compared to the model’s estimation. Further investigation into
2 the average emission rate, the variability over time of the emission rate, stocks, lifetimes, and other factors will be
3 investigated for the next (i.e., 2025) *Inventory* submission.

4 4.26 Electrical Equipment (CRT Source 5 Category 2G1)

6 The largest use of sulfur hexafluoride (SF₆), both in the United States and internationally, is as an electrical
7 insulator and interrupter in equipment that transmits and distributes electricity (RAND 2004). The gas has been
8 employed by the electric power industry in the United States since the 1950s because of its dielectric strength and
9 arc-quenching characteristics. It is used in gas-insulated substations, circuit breakers, and other switchgear. SF₆ has
10 replaced flammable insulating oils in many applications and allows for more compact substations in dense urban
11 areas. Another greenhouse gas emitted in much smaller amounts by the electric power industry is
12 tetrafluoromethane (CF₄), which is mixed with SF₆ to avoid liquefaction at low temperatures (Middleton 2000).
13 While mixed gas circuit breakers are more common in extremely cold climates in geographies outside of the
14 United States, some U.S. manufacturers of electrical equipment are emitting CF₄ during the manufacturing of
15 equipment designed to hold the SF₆/CF₄ gas mixture. However, no electrical equipment facilities in the United
16 States have reported emissions of or equipment using CF₄. SF₆ emissions exceed PFC emissions from electric power
17 systems on both a GWP-unweighted and GWP-weighted basis.

18 Fugitive emissions of SF₆ and CF₄ can escape from gas-insulated substations and switchgear through seals,
19 especially from older equipment. The gas can also be released during equipment manufacturing, installation,
20 servicing, and disposal. Emissions of SF₆ and CF₄ from equipment manufacturing and from electrical equipment
21 systems were estimated to be 5.1 MMT CO₂ Eq. (0.3 kt) in 2022. This quantity represents a 79 percent decrease
22 from the estimate for 1990 (see Table 4-126 and Table 4-127). There are a few potential causes for this decrease: a
23 sharp increase in the price of SF₆ during the 1990s and a growing awareness of the environmental impact of SF₆
24 emissions through programs such as EPA’s voluntary SF₆ Emission Reduction Partnership for Electric Power
25 Systems (Partnership) and EPA’s GHGRP, regulatory drivers at the state and local levels, and research and
26 development of alternative gases to SF₆ that can be used in gas-insulated substations. Utilities participating in the
27 Partnership have lowered their emission factor from 13 percent in 1999 (kg SF₆ emitted per kg of nameplate
28 capacity) to 0.9 percent in 2022. SF₆ emissions reported by electric power systems to EPA’s GHGRP have decreased
29 by 56 percent from 2011 to 2022,¹²⁰ with much of the reduction seen from utilities that are not participants in the
30 Partnership. These utilities may be making relatively large reductions in emissions as they take advantage of
31 relatively large and/or inexpensive emission reduction opportunities (i.e., “low hanging fruit,” such as replacing
32 major leaking circuit breakers) that Partners have already taken advantage of under the voluntary program
33 (Ottinger et al. 2014). Total emissions from electrical equipment in 2022 were lower than 2021 emissions,
34 decreasing by 14.8 percent.

¹²⁰ Analysis of emission trends from facilities reporting to EPA’s GHGRP is imperfect due to an inconsistent group of reporters year to year. A facility that has reported total non-biogenic greenhouse gas emissions below 15,000 metric tons of carbon dioxide equivalent (MT CO₂ Eq.) for three consecutive years or below 25,000 MT CO₂ Eq. for five consecutive years to EPA’s GHGRP can discontinue reporting for all direct emitter subparts. For this sector, most of the variability in the group of reporters is due to facilities exiting the GHGRP due to being below one of these thresholds; however, facilities must re-enter the program if their emissions at a later date are above 25,000 MT CO₂ Eq., which may occur for a variety of reasons, including changes in facility size and changes in emission rates.

1 **Table 4-126: SF₆ and CF₄ Emissions from Electric Power Systems and Electrical Equipment**
 2 **Manufacturers (MMT CO₂ Eq.)**

	1990	2005	2018	2019	2020	2021	2022
Electric Power Systems	24.3	11.2	4.7	5.7	5.3	5.6	4.8
Electrical Equipment Manufacturers	0.3	0.7	0.3	0.4	0.5	0.4	0.3
Total	24.7	11.9	5.0	6.1	5.9	6.0	5.1

Note: Totals may not sum due to independent rounding.

3 **Table 4-127: SF₆ and CF₄ Emissions from Electric Power Systems and Electrical Equipment**
 4 **Manufacturers (kt)**

	1990	2005	2018	2019	2020	2021	2022
SF ₆ Emissions	1	0.5	0.2	0.3	0.3	0.3	0.2
CF ₄ Emissions	NO	0.00029	NO	0.0001	0.00002	0.00002	+

+ Does not exceed 0.000005 kt.

NO (Not Occurring)

5 Methodology and Time-Series Consistency

6 The estimates of emissions from electrical equipment are comprised of emissions from electric power systems and
 7 emissions from the manufacture of electrical equipment. The methodologies for estimating both sets of emissions
 8 are described below.

9 1990 through 1998 Emissions from Electric Power Systems

10 Emissions from electric power systems from 1990 through 1998 were estimated based on (1) the emissions
 11 estimated for this source category in 1999, which, as discussed in the next section, were based on the emissions
 12 reported during the first year of EPA's SF₆ Emission Reduction Partnership for Electric Power Systems (Partnership),
 13 and (2) the RAND survey of global SF₆ emissions. Because most utilities participating in the Partnership reported
 14 emissions only for 1999 through 2011, modeling was used to estimate SF₆ emissions from electric power systems
 15 for the years 1990 through 1998. To perform this modeling, U.S. emissions were assumed to follow the same
 16 trajectory as global emissions from this source during the 1990 through 1999 period. To estimate global emissions,
 17 the RAND survey of global SF₆ sales was used, together with the following equation for estimating emissions, which
 18 is derived from the mass-balance equation for chemical emissions (Volume 3, Equation 7.3) in the *2006 IPCC*
 19 *Guidelines*.¹²¹ (Although Equation 7.3 of the *2006 IPCC Guidelines* appears in the discussion of substitutes for
 20 ozone-depleting substances, it is applicable to emissions from any long-lived pressurized equipment that is
 21 periodically serviced during its lifetime.)

22 Equation 4-24: Estimation for SF₆ Emissions from Electric Power Systems

23 Emissions (kilograms SF₆) = SF₆ purchased to refill existing equipment (kilograms) + nameplate capacity of retiring
 24 equipment (kilograms)¹²²

¹²¹ Ideally, sales to utilities in the United States between 1990 and 1999 would be used as a model. However, this information was not available. There were only two U.S. manufacturers of SF₆ during this time period, so it would not have been possible to conceal sensitive sales information by aggregation.

¹²² Nameplate capacity is defined as the amount of SF₆ within fully charged electrical equipment.

1 Note that the above equation holds whether the gas from retiring equipment is released or recaptured; if the gas
2 is recaptured, it is used to refill existing equipment, thereby lowering the amount of SF₆ purchased by utilities for
3 this purpose.

4 Gas purchases by utilities and equipment manufacturers from 1961 through 2003 are available from the RAND
5 (2004) survey. To estimate the quantity of SF₆ released or recovered from retiring equipment, the nameplate
6 capacity of retiring equipment in a given year was assumed to equal 81.2 percent of the amount of gas purchased
7 by electrical equipment manufacturers 40 years previous (e.g., in 2000, the nameplate capacity of retiring
8 equipment was assumed to equal 81.2 percent of the gas purchased in 1960). The remaining 18.8 percent was
9 assumed to have been emitted at the time of manufacture. The 18.8 percent emission factor is an average of IPCC
10 default SF₆ emission rates for Europe and Japan for 1995 (IPCC 2006). The 40-year lifetime for electrical equipment
11 is also based on IPCC (2006). The results of the two components of the above equation were then summed to yield
12 estimates of global SF₆ emissions from 1990 through 1999.

13 U.S. emissions between 1990 and 1999 are assumed to follow the same trajectory as global emissions during this
14 period. To estimate U.S. emissions, global emissions for each year from 1990 through 1998 were divided by the
15 estimated global emissions from 1999. The result was a time series of factors that express each year's global
16 emissions as a multiple of 1999 global emissions. Historical U.S. emissions were estimated by multiplying the factor
17 for each respective year by the estimated U.S. emissions of SF₆ from electric power systems in 1999 (estimated to
18 be MMT CO₂ Eq.).

19 Two factors may affect the relationship between the RAND sales trends and actual global emission trends. One is
20 utilities' inventories of SF₆ in storage containers. When SF₆ prices rise, utilities are likely to deplete internal
21 inventories before purchasing new SF₆ at the higher price, in which case SF₆ sales will fall more quickly than
22 emissions. On the other hand, when SF₆ prices fall, utilities are likely to purchase more SF₆ to rebuild inventories, in
23 which case sales will rise more quickly than emissions. This effect was accounted for by applying 3-year smoothing
24 to utility SF₆ sales data. The other factor that may affect the relationship between the RAND sales trends and
25 actual global emissions is the level of imports from and exports to Russia and China. SF₆ production in these
26 countries is not included in the RAND survey and is not accounted for in any another manner by RAND. However,
27 atmospheric studies confirm that the downward trend in estimated global emissions between 1995 and 1998 was
28 real (see the Uncertainty discussion below).

29 **1999 through 2022 Emissions from Electric Power Systems**

30 Emissions from electric power systems from 1999 to 2022 were estimated based on: (1) reporting from utilities
31 participating in EPA's SF₆ Emission Reduction Partnership for Electric Power Systems (Partners), which began in
32 1999; (2) reporting from utilities covered by EPA's GHGRP, which began in 2012 for emissions occurring in 2011
33 (GHGRP-Only Reporters); (3) SF₆ emissions from California estimated by the California Air Resources Board (CARB)
34 and (4) the relationship between utilities' reported emissions and their transmission miles as reported in the 2001,
35 2004, 2007, 2010, 2013, and 2016 Utility Data Institute (UDI) Directories of Electric Power Producers and
36 Distributors (UDI 2001, 2004, 2007, 2010, 2013, and 2017), and 2019, 2020, and 2021 Homeland Infrastructure
37 Foundation-Level Data (HIFLD) (HIFLD 2019, 2020, and 2021), which was applied to the electric power systems that
38 do not report to EPA (Non-Reporters). Total U.S. transmission mileage was interpolated between 2016 and 2019 to
39 estimate transmission mileage of electric power systems in 2017 and 2018. (Transmission miles are defined as the
40 miles of lines carrying voltages above 34.5 kV).

1 **Partners**

2 Over the period from 1999 to 2022, Partner utilities, which for inventory purposes are defined as utilities that
3 either currently are or previously have been part of the Partnership,¹²³ represented 49 percent, on average, of
4 total U.S. transmission miles. Partner utilities estimated their emissions using a Tier 3 utility-level mass balance
5 approach (IPCC 2006). If a Partner utility did not provide data for a particular year, emissions were interpolated
6 between years for which data were available or extrapolated based on Partner-specific transmission mile growth
7 rates. In 2012, many Partners began reporting their emissions (for 2011 and later years) through EPA's GHGRP
8 (discussed further below) rather than through the Partnership. In 2022, less than 1 percent of the total emissions
9 attributed to Partner utilities were reported through Partnership reports. Approximately 99.6 percent of the total
10 emissions attributed to Partner utilities were reported and verified through EPA's GHGRP.¹²⁴ Overall, the emission
11 rates reported by Partners have decreased significantly throughout the time series.

12 **Non-Partners**

13 Non-Partners consist of two groups: Utilities that have reported to the GHGRP beginning in 2012 (reporting 2011
14 emissions) or later years (GHGRP-only Reporters) and utilities that have never reported to the GHGRP (Non-
15 Reporters). EPA's GHGRP requires users of SF₆ in electric power systems to report emissions if the facility has a
16 total SF₆ nameplate capacity that exceeds 17,820 pounds. (This quantity is the nameplate capacity that would
17 result in annual SF₆ emissions equal to 25,000 metric tons of CO₂ equivalent at the historical emission rate reported
18 under the Partnership). As under the Partnership, electric power systems that report their SF₆ emissions under
19 EPA's GHGRP are required to use the Tier 3 utility-level mass-balance approach. GHGRP-Only Reporters accounted
20 for 16 percent of U.S. transmission miles and 14 percent of estimated U.S. emissions from electric power system in
21 2022.¹²⁵

22 From 1999 through 2018, emissions from both GHGRP-only Reporters and Non-Reporters were estimated in the
23 same way. From 1999 through 2008, emissions were estimated using the results of a regression analysis that
24 correlated the 1999 emissions from Partner utilities with their 1999 transmission miles.¹²⁶ The 1999 regression
25 coefficient (emission factor) was held constant through 2008 and multiplied by the transmission miles estimated
26 for the non-Partners for each year.

27 The 1999 regression equation for Non-Partners was developed based on the emissions reported by a subset of
28 Partner utilities who reported non-zero emissions and non-zero transmission miles (representing approximately 50
29 percent of total U.S. transmission miles). The regression equation for 1999 is displayed in the equation below.

¹²³ Starting in the 1990 to 2015 Inventory, partners who had reported three years or less of data prior to 2006 were removed. Most of these Partners had been removed from the list of current Partners but remained in the Inventory due to the extrapolation methodology for non-reporting partners.

¹²⁴ Only data reported as of August 21, 2023 are used in the emission estimates for the prior year of reporting. Emissions for Partners that did not report to the Partnership or GHGRP are extrapolated for three years using a utility-specific transmission mile growth rate. After four consecutive years of non-reporting they are included in the 'non-reporting Partners' category. It should be noted that data reported through EPA's GHGRP must go through a verification process. For electric power systems, verification involved a series of electronic range, completeness, and algorithm checks for each report submitted.

¹²⁵ GHGRP-reported and Partner transmission miles from a number of facilities were equal to zero with non-zero emissions. These facilities emissions were added to the emissions totals for their respective parent companies when identifiable and not included in the regression equation when not identifiable or applicable. Other facilities reported non-zero transmission miles with zero emissions, or zero transmission miles and zero emissions. These facilities were not included in the development of the regression equations (discussed further below). These emissions are already implicitly accounted for in the relationship between transmission miles and emissions.

¹²⁶ In the United States, SF₆ is contained primarily in transmission equipment rated above 34.5 kV.

Equation 4-25: Regression Equation for Estimating SF₆ Emissions of Non-Reporting Facilities in 1999

$$\text{Emissions (kg)} = 0.771 \times \text{Transmission Miles}$$

The 1999 emission factor (0.77 SF₆ emissions/Transmission Miles) for the non-Partners was held constant to estimate non-Partner emissions from 2000-2008. Non-partner emissions were assumed to decrease beginning in 2009, trending toward the regression coefficient (emission factor) calculated for the GHGRP-only reporters based on their reported 2011 emissions and transmission miles. Emission factors for 2009 and 2010 were linearly interpolated between the 1999 and 2011 emission factors. For 2009, the emissions of non-Partners were estimated by multiplying their transmission miles by the interpolated 2009 emission factor (0.65 kg/transmission mile).

The 2011 regression equation was developed based on the emissions reported by GHGRP-Only Reporters who reported non-zero emissions and non-zero transmission miles (representing approximately 23 percent of total U.S. transmission miles). The regression equation for 2011 is displayed below.

Equation 4-26: Regression Equation for Estimating SF₆ Emissions of GHGRP-Only Reporters in 2011

$$\text{Emissions (kg)} = 0.397 \times \text{Transmission Miles}$$

For 2011 and later years, the emissions of GHGRP-only reporters were generally equated to their reported emissions, unless they did not report. The emissions of GHGRP-only reporters that have years of non-reporting between reporting years are gap filled by interpolating between reported values.

For 2010 and later years, the emissions of non-Reporters were estimated by multiplying their transmission miles by the estimated 2010 emission factor (0.52 kg/transmission mile), which was held constant from 2010 through 2022.

Off-ramping GHGRP Facilities

The GHGRP program has an “off-ramp” provision (40 CFR Part 98.2(i)) that exempts facilities from reporting under certain conditions. If reported total greenhouse gas emissions are below 15,000 metric tons of carbon dioxide equivalent (MT CO₂ Eq.) for three consecutive years or below 25,000 MT CO₂ Eq. for five consecutive years, the facility may elect to discontinue reporting. Emissions of GHGRP reporters that have off-ramped are extrapolated for three years of non-reporting using a utility-specific transmission mile growth rate, unless the utility has transmission mileage in California. After three consecutive years of non-reporting, emissions for facilities (except those in California) that off-ramped from GHGRP were estimated using an emissions rate derived from the reported emissions and transmission miles of GHGRP-only reporters in the respective year. For facilities in California, a California-specific emissions rate is used as described in the following section.

Table 4-128: GHGRP-only Average Emission Rate (kg per mile)

	2011	2018	2019	2020	2021	2022
Average emission rate	0.43	0.22	0.29	0.27	0.25	0.22

Table 4-129: Categorization of Utilities and Timeseries for Application of Corresponding Emission Estimation Methodologies

Categorization of Utilities	Timeseries
Partners	1999 - 2021
Non-Partners (GHGRP-Only)	2011 – 2021
Non-Partners (Remaining Non-Reporting Utilities)	1999 – 2021
Off-ramping GHGRP Facilities	2017 – 2021

1 **California**

2 CARB reports the total SF₆ emissions from electrical equipment within the state of California (CARB 2022). Because
 3 California utilities are required to report their SF₆ emissions to CARB even when they are not required to report to
 4 the GHGRP, CARB’s estimates of California SF₆ emissions are expected to be more accurate for the California
 5 utilities that do not report to GHGRP than the methodology described above. As a result, the CARB SF₆ emissions
 6 estimates are used as California’s contribution to the national total for 2011-2022, except in years where CARB’s
 7 estimate is smaller than the California estimates reported to EPA or years for which CARB has not published
 8 estimates. Since CARB’s emissions estimates include emissions from facilities that do not report to GHGRP,
 9 emissions for California GHGRP reporters that have off-ramped are not extrapolated. Specifically, CARB estimates
 10 are used for 2011, 2012, 2013, 2014, 2017, 2018, 2019, and 2020.

11 For each utility with transmission mileage in California, the GHGRP or voluntarily reported emissions attributed to
 12 California for that utility were determined using the percentage of that utility’s transmission mileage within
 13 California based on data from HIFLD. These emissions across all California utilities were summed to find the
 14 California emissions that were reported through GHGRP or voluntarily to the EPA. Then, if CARB’s emissions
 15 estimates for the reporting year were larger than the those from GHGRP and voluntary reporting, CARB’s
 16 emissions replaced the California emissions from GHGRP and voluntary reporting.

17 If CARB’s emissions estimates were lower than the California emissions from GHGRP and voluntary reporting, it is
 18 assumed there is likely an error, as this would imply negative emissions by GHGRP non-reporters. This was the case
 19 in 2015 and 2016. For these years, the GHGRP and voluntarily reported emissions from California are retained, and
 20 emissions from non-reporting utilities are estimated using a California-specific SF₆ emissions rate, which is based
 21 on CARB emission data. The California SF₆ emissions rate of 0.46 lbs SF₆ per transmission mile is found by taking
 22 the average of CARB emissions divided by the total California transmission mileage in years where CARB estimates
 23 are larger. Emissions from California non-reporting utilities are then found by multiplying the California SF₆
 24 emissions rate by the California transmission mileage from non-reporting utilities. This methodology is also used if
 25 CARB has not published emissions estimates for a particular year. CARB has not yet published estimates for 2021
 26 or 2022.

27 **Table 4-130: California GHGRP and Voluntarily Reported SF₆ Emissions Compared to CARB’s**
 28 **SF₆ Emissions (MMT CO₂ Eq.)**

	2011	2015	2016	2017	2018	2019	2020	2021	2022
CA GHGRP and Voluntary	0.19	0.16	0.24	0.12	0.11	0.14	0.20	0.14	0.15
CARB (CARB 2022)	0.24	0.14	0.10	0.18	0.14	0.17	0.24	NE	NE
Final CA	0.24	0.20	0.28	0.18	0.14	0.17	0.24	0.19	0.20

29 NE (Not Estimated)

30 **Total Industry Emissions** Total electric power system emissions from 1999 through 2022 were determined for each
 31 year by summing the Partner reported and estimated emissions (reported data was available through the EPA’s SF₆
 32 Emission Reduction Partnership for Electric Power Systems), the GHGRP-only reported emissions, off-ramping
 33 GHGRP Facilities (non-reporters), non-reporters who eventually report to GHGRP, and the non-reporting utilities’
 34 emissions (except California). Then, the California GHGRP and voluntarily reported emissions are subtracted from
 35 the total and replaced with CARB’s emissions (or GHGRP and voluntarily reported emissions plus California non-
 36 reporting utilities’ emissions).

37 **Non-Partner Transmission Miles**

38 Data on transmission miles for each Non-Reporter for the years 2000, 2003, 2006, and 2009, 2012, and 2016 were
 39 obtained from the 2001, 2004, 2007, 2010, 2013, and 2017 UDI Directories of Electric Power Producers and
 40 Distributors, respectively (UDI 2001, 2004, 2007, 2010, 2013, and 2017). For 2019 to 2022 non-reporter
 41 transmission mileage was derived by subtracting reported transmission mileage data from the total U.S.

1 transmission mileage from 2019 to 2022 HIFLD Data (HIFLD 2019, 2020, 2021, and 2022). The following trends in
2 transmission miles have been observed over the time series:

- 3 • The U.S. transmission system grew by over 22,000 miles between 2000 and 2003 yet declined by almost
4 4,000 miles between 2003 and 2006. Given these fluctuations, periodic increases are assumed to occur
5 gradually. Therefore, transmission mileage was assumed to increase at an annual rate of 1.2 percent
6 between 2000 and 2003 and decrease by 0.20 percent between 2003 and 2006.
- 7 • The U.S. transmission system’s annual growth rate grew to 1.7 percent from 2006 to 2009 as transmission
8 miles increased by more than 33,000 miles.
- 9 • The annual growth rate for 2009 through 2012 was calculated to be 1.4 percent as transmission miles
10 grew yet again by almost 28,000 miles during this time period.
- 11 • The annual transmission mile growth rate for 2012 through 2016 was calculated to be 0.2 percent, as
12 transmission miles increased by approximately 6,600 miles.
- 13 • The annual transmission mile growth rate for 2016 through 2020 was calculated to be 0.9 percent, as
14 transmission miles increased by approximately 26,000 miles.
- 15 • The annual transmission mile growth rate for 2020 through 2021 was calculated to be 2.2 percent, as
16 transmission miles increased by approximately 16,000 miles.
- 17 • The annual transmission mile growth rate for 2021 through 2022 was calculated to be 0.7 percent, as
18 transmission miles increased by approximately 5,500 miles.

19 Transmission miles for each year for non-reporters were calculated by interpolating between UDI reported values
20 obtained from the 2001, 2004, 2007, 2010, 2013 and 2017 UDI directories and HIFLD data for 2019 and
21 subsequent years. In cases where a non-reporter previously reported the GHGRP or the Partnership, transmission
22 miles were interpolated between the most recently reported value and the next available UDI value.

23 **1990 through 2022 Emissions from Manufacture of Electrical Equipment**

24 Three different methods were used to estimate 1990 to 2022 emissions from original electrical equipment
25 manufacturers (OEMs).

- 26 • OEM SF₆ emissions from 1990 through 2000 were derived by assuming that manufacturing emissions
27 equaled 10 percent of the quantity of SF₆ provided with new equipment. The 10 percent emission rate is
28 the average of the “ideal” and “realistic” manufacturing emission rates (4 percent and 17 percent,
29 respectively) identified in a paper prepared under the auspices of the International Council on Large
30 Electric Systems (CIGRE) in February 2002 (O’Connell et al. 2002). The quantity of SF₆ provided with new
31 equipment was estimated based on statistics compiled by the National Electrical Manufacturers
32 Association (NEMA). These statistics were provided for 1990 to 2000.
- 33 • OEM SF₆ emissions from 2000 through 2010 were estimated by (1) interpolating between the emission
34 rate estimated for 2000 (10 percent) and an emission rate estimated for 2011 based on reporting by
35 OEMs through the GHGRP (5.7 percent), and (2) estimating the quantities of SF₆ provided with new
36 equipment for 2001 to 2010. The quantities of SF₆ provided with new equipment were estimated using
37 Partner reported data and the total industry SF₆ nameplate capacity estimate (156.5 MMT CO₂ Eq. in
38 2010). Specifically, the ratio of new nameplate capacity to total nameplate capacity of a subset of
39 Partners for which new nameplate capacity data was available from 1999 to 2010 was calculated. These
40 ratios were then multiplied by the total industry nameplate capacity estimate for each year to derive the
41 amount of SF₆ provided with new equipment for the entire industry. Additionally, to obtain the 2011
42 emission rate (necessary for estimating 2001 through 2010 emissions), the estimated 2011 emissions
43 (estimated using the third methodology listed below) were divided by the estimated total quantity of SF₆

1 provided with new equipment in 2011. The 2011 quantity of SF₆ provided with new equipment was
2 estimated in the same way as the 2001 through 2010 quantities.

- 3 • OEM CF₄ emissions from 1991 through 2010 were estimated by using an average ratio of reported SF₆ and
4 CF₄ emissions from 2011 through 2013. This ratio was applied to the estimated SF₆ emissions for 1991
5 through 2010 to arrive at CF₄ emissions. CF₄ emissions are estimated starting in 1991 and assumed zero
6 prior to 1991 based on the entry of the CF₄/SF₆ gas mixture into the market (Middleton 2000).
- 7 • OEM emissions from 2011 through 2022 were estimated using the SF₆ and CF₄ emissions from OEMs
8 reporting to the GHGRP, and an assumption that these reported emissions account for a conservatively
9 low estimate of 50 percent of the total emissions from all U.S. OEMs (those that report and those that do
10 not).
- 11 • OEM SF₆ emissions from facilities off-ramping from the GHGRP were determined by extrapolation. First,
12 emission growth rates were calculated for each reporting year for each OEM reporting facility as well as
13 an average emissions growth rate (2011 through 2022). Averages of reported emissions from last three
14 consecutive reporting years were multiplied by the average growth rate for each off-ramping OEM to
15 estimate emissions for the non-reporting year(s).

16 Methodological approaches were applied to the entire time series to ensure time-series consistency from 1990
17 through 2022.

18 Uncertainty

19 To estimate the uncertainty associated with emissions of SF₆ and CF₄ from electrical equipment, uncertainties
20 associated with four quantities were estimated: (1) emissions from Partners, (2) emissions from GHGRP-Only
21 Reporters, (3) emissions from Non-Reporters, and (4) emissions from manufacturers of electrical equipment. A
22 Monte Carlo analysis was then applied to estimate the overall uncertainty of the emissions estimate.

23 Total emissions from the SF₆ Emission Reduction Partnership include emissions from both reporting (through the
24 Partnership or EPA's GHGRP) and non-reporting Partners. For reporting Partners, individual Partner-reported SF₆
25 data was assumed to have an uncertainty of +/- 10 percent. Based on a Monte Carlo analysis, the cumulative
26 uncertainty of all Partner-reported data was estimated to be 4.5 percent. The uncertainty associated with
27 extrapolated or interpolated emissions from non-reporting Partners was assumed to be 20 percent.

28 For GHGRP-Only Reporters, reported SF₆ data was assumed to have an uncertainty of 10 percent. Based on a
29 Monte Carlo analysis, the cumulative uncertainty of all GHGRP-Only reported data was estimated to be 7.4
30 percent.

31 As discussed below, EPA has substantially revised its method for estimating emissions from non-Reporters,
32 assuming that the average emission rate of non-Reporters has declined much more slowly than the average
33 emission rate of reporting facilities rather than declining at the same rate. This assumption brings the U.S. SF₆
34 emissions estimated in this *Inventory* into better agreement with the U.S. SF₆ emissions inferred from atmospheric
35 observations. However, it must be emphasized that the actual emission rates of non-Reporters remain unknown. It
36 is possible that they are lower or even higher than estimated here. One possibility is that SF₆ sources other than
37 electric power systems are contributing to the emissions inferred from atmospheric observations, implying that
38 the emissions from non-Reporters are lower than estimated here. Another is that the emissions inferred from
39 atmospheric measurements are over- (or under-) estimated, implying that emissions from non-Reporters could be
40 either lower or higher than estimated here. These uncertainties are difficult to quantify and are not reflected in the
41 estimated uncertainty below. The estimated uncertainty below accounts only for the two sources of uncertainty
42 associated with the regression equations used to estimate emissions in 2019 from Non-Reporters: (1) uncertainty
43 in the coefficients (as defined by the regression standard error estimate), and (2) the uncertainty in total
44 transmission miles for Non-Reporters. Uncertainties were also estimated regarding (1) estimates of SF₆ and CF₄
45 emissions from OEMs reporting to EPA's GHGRP, and (2) the assumption on the percent share of OEM emissions
46 from OEMs reporting to EPA's GHGRP.

1 The results of the Approach 2 quantitative uncertainty analysis are summarized in Table 1-57. electrical equipment
 2 emissions were estimated to be between 3.8 and 6.4 MMT CO₂ Eq. at the 95 percent confidence level, a range of
 3 approximately 25 percent below and 26 percent above the emission estimate of 5.1 MMT CO₂ Eq. CF₄ emissions
 4 were estimated to be between 0.0000059 and 0.0000088 MMT CO₂ Eq. at the 95 percent confidence level, a range
 5 of approximately 20 percent below and 20 percent above the emission estimate of 0.0000074 MMT CO₂ Eq.

6 **Table 4-131: Approach 2 Quantitative Uncertainty Estimates for SF₆ and CF₄ Emissions from**
 7 **Electrical Equipment (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to 2022 Emission Estimate ^a			
			(MMT CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
Electrical Equipment	SF ₆	5.1	3.8	6.4	-25%	+26%
Electrical Equipment	CF ₄	0.0000074	0.0000059	0.0000088	-20%	+20%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

8 In addition to the uncertainty quantified above for the 2022 estimate, there is uncertainty associated with the
 9 emission rates of GHGRP-only facilities before 2011 and of non-Reporters throughout the time series. As noted
 10 above in the discussion of the uncertainty of non-Reporters for 2022, these uncertainties are difficult to quantify.

11 There is also uncertainty associated with using global SF₆ sales data to estimate U.S. emission trends from 1990
 12 through 1999. However, the trend in global emissions implied by sales of SF₆ appears to reflect the trend in global
 13 emissions implied by changing SF₆ concentrations in the atmosphere. That is, emissions based on global sales
 14 declined by 29 percent between 1995 and 1998 (RAND 2004), and emissions based on atmospheric measurements
 15 declined by 17 percent over the same period (Levin et al. 2010).

16 Several pieces of evidence indicate that U.S. SF₆ emissions were reduced as global emissions were reduced. First,
 17 the decreases in sales and emissions coincided with a sharp increase in the price of SF₆ that occurred in the mid-
 18 1990s and that affected the United States as well as the rest of the world. A representative from DILO, a major
 19 manufacturer of SF₆ recycling equipment, stated that most U.S. utilities began recycling rather than venting SF₆
 20 within two years of the price rise. Finally, the emissions reported by the one U.S. utility that reported its emissions
 21 for all the years from 1990 through 1999 under the Partnership showed a downward trend beginning in the mid-
 22 1990s.

23 QA/QC and Verification

24 For more information on the general QA/QC process applied to this source category, consistent with Volume 1,
 25 Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in the introduction of
 26 the IPPU chapter and Annex 8 for more details. Category specific QC findings are described below.

27 For the GHGRP data, EPA verifies annual facility-level reports through a multi-step process (e.g., including a
 28 combination of pre-and post-submittal electronic checks and manual reviews by staff) to identify potential errors
 29 and ensure that data submitted to EPA are accurate, complete, and consistent (EPA 2015).¹²⁷ Based on the results
 30 of the verification process, EPA follows up with facilities to resolve mistakes that may have occurred. The post-
 31 submittals checks are consistent with a number of general and category-specific QC procedures including: range
 32 checks, statistical checks, algorithm checks, and year-to-year checks of reported data and emissions.

¹²⁷ GHGRP Report Verification Factsheet. See https://www.epa.gov/sites/production/files/2015-07/documents/ghgrp_verification_factsheet.pdf.

1 Additionally, EPA provides additional quality control for the SF₆ emissions estimates using atmospheric derived
2 estimates for comparison. The *2019 Refinement to the 2006 IPCC Guidelines for National Greenhouse Gas*
3 *Inventories* (IPCC 2019) Volume 1: General Guidance and Reporting, Chapter 6: Quality Assurance, Quality Control
4 and Verification notes that atmospheric concentration measurements can provide independent data sets as a basis
5 for comparison with inventory estimates. Further, it identifies fluorinated gases as particularly suited for such
6 comparisons. The *2019 Refinement* makes this conclusion for fluorinated gases based on their lack of significant
7 natural sources,¹²⁸ their generally long atmospheric lifetimes, their well-known loss mechanisms, and the potential
8 uncertainties in bottom-up inventory methods for some of their sources. Unlike non-fluorinated greenhouse gases
9 (CO₂, CH₄, and N₂O), SF₆ has no significant natural sources; therefore, the SF₆ estimates derived from atmospheric
10 measurements are driven overwhelmingly by anthropogenic emissions. The *2019 Refinement* provides guidance on
11 conducting such comparisons (as summarized in Table 6.2 of IPCC 2019 Volume 1, Chapter 6) and provides
12 guidance on using such comparisons to identify areas of improvement in national inventories (as summarized in
13 Box 6.5 of IPCC 2019 Volume 1, Chapter 6). Emission estimates derived from atmospheric measurements of SF₆
14 made at NOAA and described in Hu et al. (2022) were used to perform a comparison to the inventory estimates.
15 This comparison resulted in changes to historical emission estimates, as more thoroughly described in the previous
16 *Inventory* cycle (EPA 2022). No further changes were made to the electrical equipment estimates for the current
17 (i.e., 1990 through 2022) *Inventory* based on this comparison.

18 Recalculations Discussion

19 Several updates to activity data led to recalculations of previous *Inventory* results. The major updates are as
20 follows:

- 21 • As discussed in the methodology above, CARB estimates of SF₆ emissions from electrical equipment in
22 California were used as California's contribution to the national total starting in 2011, except in cases
23 where CARB emissions were lower than GHGRP and voluntarily reported emissions from California, or in
24 years where CARB has not published SF₆ emissions estimates.
- 25 • Updates were made to reporter emissions where facilities had resubmitted data.
- 26 • A correction was made to estimate 2021 nameplate capacities for two off-ramping utilities, which off-
27 ramped in 2021; these estimations were inadvertently omitted in the previous *Inventory*.
- 28 • SF₆ emissions from electrical equipment manufacturing was corrected due to an erroneous data pull in
29 the previous *Inventory*. This caused emissions to increase in years 2011 through 2019.
- 30 • Partner transmission mileage used for calculating average share of Partner utilities across the time series
31 and for estimating nameplate capacity for non-reporting utilities was corrected. The calculation was
32 previously referencing partner transmission mileage for the prior year for 2013 through 2021.

33 Planned Improvements

34 EPA plans to revisit the methodology for determining emissions from the manufacture of electrical equipment, in
35 particular, the assumption that emissions reported by OEMs account for a conservatively low estimate of 50
36 percent of the total emissions from all U.S. OEMs. Additional market research will be required to confirm or modify
37 the assumptions regarding the portion of industry not reporting to the GHGRP program.

4.27 SF₆ and PFCs from Other Product Use (CRT Source Category 2G.2)

There are a variety of other products and processes that use fluorinated greenhouse gases. This section estimates emissions of sulphur hexafluoride (SF₆) and perfluorocarbons (PFCs) from other product use (CRT Source Category 2G2), including military and scientific applications. Many of these applications utilize SF₆ or PFCs to exploit their unique chemical properties, such as the high dielectric strength of SF₆ and the stability of PFCs. Emission profiles from these processes may vary greatly, ranging from immediate and unavoidable release of all of the chemical to largely avoidable, delayed release from leak-tight products after decades of use.

Military applications employ SF₆ and PFCs in many processes, such as the use of SF₆ in the radar systems of military reconnaissance planes of the Boeing E-3A type, commonly known as Airborne Warning and Control Systems (AWACS). These systems use SF₆ to prevent electric flashovers in the hollow conductors of the antenna, where voltages can reach up to 135 kilovolts (kV). During ascent of the planes, SF₆ is automatically released from the AWACS to maintain appropriate pressure difference between the system and the outside air. During descent, the system is automatically charged with SF₆ from an SF₆ container on board. Most emissions occur during ascent but may also occur from system leakage during other phases of flight or during time on the ground. Emissions from AWACS are largely dependent on the number of active planes and sorties (take-offs) per year.

Other uses of SF₆ in military applications include the oxidation of lithium in navel torpedoes and infrared decoys. SF₆ has also been documented for use in the quieting of torpedo propellers, as well as a by-product of the processing of nuclear material for the production of fuel and nuclear warheads.

Military electronics are believed to be a key application for PFC heat transfer fluids, particularly in areas such as ground and airborne radar avionics, missile guidance systems, and sonar. PFCs may also be used to cool electric motors, especially for equipment where noise reduction is a priority (e.g., submarines). The specific PFCs used in military applications are similar to heat transfer fluids identified in the electronics industry (see Section 4.23). PFCs are typically contained in a closed system, so the emissions are most likely to occur during the manufacture, maintenance, and disposal of equipment.

SF₆ and PFCs are also employed in several scientific applications, such as for use in particle accelerators. Particle accelerators can be found in university and research settings, as well as in industrial and medical applications. SF₆ is typically used as an insulating gas and is operated in a vessel exceeding atmospheric pressure. The amount of SF₆ used in particle accelerators is largely dependent on the terminal voltage of the unit. Emissions of SF₆ typically occur when SF₆ is transferred to storage tanks while maintenance is occurring, when pressure relief valves are actuated, and through slow leaks. The emission and charge assumptions for industrial and medical particle accelerators differ from those of university and research accelerators, as discussed in the methodology below. PFCs (particularly PFC-14) may also be used in particle accelerators as particle detectors or counters (Workman 2022).

SF₆ may also be employed in other high-voltage scientific equipment, including lasers, x-rays, and electron microscopes. SF₆ emission estimates for this equipment were not disaggregated from particle accelerators for this *Inventory*.

There is a range of unidentified processes that also use SF₆ and PFCs, such as R&D activities. PFCs are likely used primarily as heat transfer fluids (HTFs). Emissions reported for these unknown activities group under "Other Scientific Applications".

Emissions of SF₆ and PFCs from the applications outlined above are presented in Table 4-132.

1 **Table 4-132: SF₆ and PFC Emissions from Other Product Use (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
SF ₆	0.6	0.6	0.4	0.4	0.3	+	0.3
Total AWACs	0.6	0.6	0.4	0.4	0.3	+	0.3
SF ₆	0.3	0.3	0.0	0.0	0.0	0.0	0.0
PFCs	0.1	0.1	0.2	0.2	0.2	0.1	0.2
Total Other Military Applications	0.4	0.4	0.2	0.2	0.2	0.1	0.2
SF ₆	0.4	0.5	0.4	0.2	0.1	0.2	0.2
PFC-14	+	+	+	+	+	+	+
Total Particle Accelerators	0.4	0.5	0.4	0.2	0.1	0.2	0.2
SF ₆	+	+	+	+	0.1	0.2	0.1
PFCs	+	+	+	+	+	+	+
Total Other Scientific Applications	+	+	+	+	0.1	0.2	0.1
Total Other Product Use	1.4	1.5	0.9	0.8	0.7	0.5	0.8

+ Does not exceed 0.05 MMT CO₂ Eq.

Note: PFC subtotals include estimates for HFEs.

2 Methodology and Time-Series Consistency

3 Emissions are based primarily on data reported through the Federal Energy Management Program (FEMP).
 4 However, the availability of data from FEMP differs across the 1990 through 2022 time series. Consequently,
 5 additional emission estimates were made through utilizing methodologies from the Intergovernmental Panel on
 6 Climate Change (IPCC). Emissions from military applications and scientific applications were estimated separately,
 7 and the approaches are described immediately below.

8 Military Applications

9 *1990 through 2007*

10 FEMP data was not readily available for the 1990 to 2007 period as the first reporting year was in 2008. Thus for
 11 this period, estimated SF₆ emissions from AWACS were calculated based using the IPCC Tier 1 methodology (IPCC
 12 2006). IPCC provided a default emissions factor of 740kg of SF₆ per plane per year. It was assumed that the U.S.
 13 AWAC fleet between 1990 through 2022 remained constant at 33 planes. The IPCC methodology was utilized for all
 14 years from 1990 to 2007.

15 Emissions for other military applications were estimated by taking the average of the emissions estimated for
 16 other applications as described in the next section for first five FEMP reporting years (i.e., 2008 through 2012) and
 17 held constant between 1990 through 2007. Emissions from AWACS were not specifically reported by FEMP; the
 18 estimates developed for AWACS using the IPCC methodology above were employed to determine emissions from
 19 remaining unidentified military applications.

20 *2008 through 2021*

21 For the period 2008 through 2021, the Department of Defense (DOD) reported emission data through FEMP which
 22 were used to develop estimates for SF₆ and PFCs from other military applications. SF₆ emission estimates
 23 developed for AWACS using the IPCC Tier 1 methodology (see 1990 through 2007) were compared against SF₆
 24 emissions reported by DOD between 2008 and 2021. In years where SF₆ emissions reported by DOD was smaller
 25 than those estimated using the IPCC Tier 1 methodology, DOD-reported emissions were assumed to account for
 26 total AWAC emissions; in years where DOD emissions were greater than the calculated AWAC emissions, the
 27 remainder is assumed to be from other SF₆ applications.

28 Emissions from PFCs, HFEs, and other perfluoro compounds are directly reported by DOD. In years where there are
 29 data gaps from FEMP between two reporting years, expected emissions were interpolated.

1 **2022**

2 Estimates for emissions from AWACS and other military uses for 2022 were determined by taking an average of
3 the previous five reporting years (i.e., 2017 through 2021).

4 **Scientific Applications**

5 *1990 through 2007*

6 For the period 1990 through 2007, where no reported data is available from the Department of Energy (DOE),
7 estimates for emissions of SF₆ and PFCs from other product use at Department of Energy Laboratories were
8 determined by taking an average of the first five reporting years (i.e., 2008 through 2012) and held constant from
9 1990 to 2007.

10 SF₆ emissions from other (non-DOE) research and industrial particle accelerators in the United States was
11 calculated based on the IPCC Tier 1 methodology for estimating emissions from industrial and university/research
12 particle accelerators. Default emission factors, charge sizes, and usage rates are provided by size and type of
13 accelerator in the IPCC methodology. These default assumptions were multiplied against the number of particle
14 accelerators estimated to be active in the United States by year. This methodology remained the same from 1990
15 to 2007.

16 *2008 through 2021*

17 For the period 2008 through 2021, SF₆ and PFC emissions from government particle accelerators and other
18 scientific equipment were developed using DOE-reported emissions. SF₆ and PFC emissions from particle
19 accelerators were directly reported by DOE. Other fugitive emissions reported by DOE for SF₆ were assumed to
20 represent emissions from particle accelerators and other scientific equipment, as well as two DOE-managed power
21 facilities (WAPA and BPA).¹²⁹ Emissions from these two facilities were subtracted out to present only SF₆ emissions
22 from scientific equipment. Reported fugitive emissions for PFC-14 were assumed to wholly represent particle
23 accelerator applications. SF₆ emissions from non-government particle accelerators were estimated using the IPCC
24 Tier 1 methodology used for 1990 through 2007.

25 Process emissions from other applications for SF₆ and PFCs were reported by DOE for activities such as R&D, and
26 these emissions were summed by gas. However, the estimates presented here do not include emissions reported
27 for semiconductor research and manufacture, or from refrigeration and air conditioning. Emissions from additional
28 PFCs, HFEs, and other perfluoro compounds are directly reported by DOE and are reported as “Other
29 Applications.” Emissions reported to FEMP were generally calculated based on consumption data. In a number of
30 years, negative values for emissions were reported due to more gas being returned to supply than purchased in a
31 given year. When negative values were reported, EPA took the average of that year and the proceeding and
32 following year and applied that value to all three years. This 3-year average was assumed to be more
33 representative of actual emissions.

34 In years where there are data gaps between two reporting years, emissions were interpolated.

35 **2022**

36 For emission estimates developed using DOE-reported emissions, estimates for 2022 were determined by taking
37 an average of the previous five reporting years (i.e., 2017 through 2021). SF₆ emissions from non-government
38 particle accelerators were estimated using the same IPCC Tier 1 methodology used for 2008 through 2021.

¹²⁹ DOE-reported fugitive emissions for SF₆ and PFCs includes emissions from high-voltage scientific equipment such as lasers, x-rays, and electron microscopes. Emissions from this equipment is included in the particle accelerators total.

Uncertainty – TO BE UPDATED FOR FINAL REPORT

QA/QC and Verification

For more information on the general QA/QC process applied to this source category, consistent with Volume 1, Chapter 6 of the 2006 IPCC Guidelines, see the QA/QC and Verification Procedures section in the introduction of the IPPU chapter and Annex 8 for more details.

Recalculations Discussion

This is a new category included for the current (i.e., 1990 to 2022) Inventory, thus, no recalculations were performed.

4.28 Nitrous Oxide from Product Uses (CRT Source Category 2G3)

Nitrous oxide (N₂O) is a clear, colorless, oxidizing liquefied gas with a slightly sweet odor which is used in a wide variety of specialized product uses and applications. The amount of N₂O that is actually emitted depends upon the specific product use or application.

There are a total of three N₂O production facilities currently operating in the United States (Ottinger 2021). Nitrous oxide is primarily used in carrier gases with oxygen to administer more potent inhalation anesthetics for general anesthesia, and as an anesthetic in various dental and veterinary applications. The second main use of N₂O is as a propellant in pressure and aerosol products, the largest application being pressure-packaged whipped cream. Small quantities of N₂O also are used in the following applications:

- Oxidizing agent and etchant used in semiconductor manufacturing;
- Oxidizing agent used, with acetylene, in atomic absorption spectrometry;
- Production of sodium azide, which is used to inflate airbags;
- Fuel oxidant in auto racing; and
- Oxidizing agent in blowtorches used by jewelers and others (Heydorn 1997).

Production of N₂O in 2022 was approximately 15 kt (see Table 4-133).

Table 4-133: N₂O Production (kt)

Year	1990	2005	2018	2019	2020	2021	2022
Production (kt)	16	15	15	15	15	15	15

Nitrous oxide emissions were 3.8 MMT CO₂ Eq. (14 kt N₂O) in 2022 (see Table 4-134). Production of N₂O stabilized during the 1990s because medical markets had found other substitutes for anesthetics, and more medical procedures were being performed on an outpatient basis using local anesthetics that do not require N₂O. The use of N₂O as a propellant for whipped cream has also stabilized due to the increased popularity of cream products packaged in reusable plastic tubs (Heydorn 1997).

1 **Table 4-134: N₂O Emissions from N₂O Product Usage (MMT CO₂ Eq.)**

Year	1990	2005	2018	2019	2020	2021	2022
N ₂ O Product Usage	3.8	3.8	3.8	3.8	3.8	3.8	3.8

2 **Table 4-135: N₂O Emissions from N₂O Product Usage (kt N₂O)**

Year	1990	2005	2018	2019	2020	2021	2022
N ₂ O Product Usage	14	14	14	14	14	14	14

3 **Methodology and Time-Series Consistency**

4 Emissions from N₂O product uses are calculated using a country-specific methodology that is consistent with 2006
 5 *IPCC Guidelines* and based on available data. The 2006 *IPCC Guidelines* do not define tier methodologies for this
 6 source category. The following equation is adapted from Equation 8.24 of the 2006 *IPCC Guidelines*:

7 **Equation 4-27: N₂O Emissions from Product Use**

8
$$E_{pu} = \sum_a (P \times S_a \times ER_a)$$

9 where,

- 10 E_{pu} = N₂O emissions from product uses, metric tons
- 11 P = Total U.S. production of N₂O, metric tons
- 12 a = specific application
- 13 S_a = Share of N₂O usage by application *a*
- 14 ER_a = Emission rate for application *a*, percent

15 The share of total quantity of N₂O usage by end-use represents the share of national N₂O produced that is used by
 16 the specific subcategory (e.g., anesthesia, food processing). In 2020, the medical/dental industry used an
 17 estimated 89.5 percent of total N₂O produced, followed by food processing propellants at 6.5 percent. All other
 18 subcategories, including semiconductor manufacturing, atomic absorption spectrometry, sodium azide production,
 19 auto racing, and blowtorches, used the remainder of the N₂O produced. This subcategory breakdown changed
 20 slightly in the mid-1990s. For instance, the small share of N₂O usage in the production of sodium azide declined
 21 significantly during the 1990s. Due to the lack of information on the specific time period of the phase-out in this
 22 market subcategory, most of the N₂O usage for sodium azide production is assumed to have ceased after 1996,
 23 with the majority of its small share of the market assigned to the larger medical/dental consumption subcategory
 24 (Heydorn 1997). For 1990 through 1996, N₂O usage was allocated across the following subcategories: medical
 25 applications, food processing propellant, and sodium azide production. A usage emissions rate was then applied
 26 for each subcategory to estimate the amount of N₂O emitted.

27 Only the medical/dental and food propellant subcategories were assumed to release emissions into the
 28 atmosphere that are not captured under another source category, and therefore these subcategories were the
 29 only usage subcategories with emission rates. Emissions of N₂O from semiconductor manufacturing are described
 30 in Section 4.23 Electronics Industry (CRT Source Category 2E) and reported under CRT Source Category 2H3. For
 31 the medical/dental subcategory, due to the poor solubility of N₂O in blood and other tissues, none of the N₂O is
 32 assumed to be metabolized during anesthesia and quickly leaves the body in exhaled breath. Therefore, an
 33 emission factor of 100 percent was used for this subcategory (IPCC 2006). For N₂O used as a propellant in
 34 pressurized and aerosol food products, none of the N₂O is reacted during the process and all of the N₂O is emitted
 35 to the atmosphere, resulting in an emission factor of 100 percent for this subcategory (IPCC 2006). For the
 36 remaining subcategories, all of the N₂O is consumed or reacted during the process, and therefore the emission rate
 37 was considered to be zero percent (Tupman 2002).

1 The 1990 through 1992 N₂O production data were obtained from SRI Consulting's *Nitrous Oxide, North America*
2 (Heydorn 1997). Nitrous oxide production data for 1993 through 1995 were not available. Production data for
3 1996 was specified as a range in two data sources (Heydorn 1997; Tupman 2002). In particular, for 1996, Heydorn
4 (1997) estimates N₂O production to range between 13.6 and 18.1 thousand metric tons. Tupman (2002) provided a
5 narrower range (15.9 to 18.1 thousand metric tons) for 1996 that falls within the production bounds described by
6 Heydorn (1997). Tupman (2002) data are considered more industry-specific and current; therefore, the midpoint
7 of the narrower production range was used to estimate N₂O emissions for years 1993 through 2001 (Tupman
8 2002). The 2002 and 2003 N₂O production data were obtained from the Compressed Gas Association Nitrous
9 Oxide Fact Sheet and Nitrous Oxide Abuse Hotline (CGA 2002, 2003). These data were also provided as a range. For
10 example, in 2003, CGA (2003) estimates N₂O production to range between 13.6 and 15.9 thousand metric tons.
11 Due to the lack of publicly available data, production estimates for years 2004 through 2022 were held constant at
12 the 2003 value.

13 The 1996 share of the total quantity of N₂O used by each subcategory was obtained from SRI Consulting's *Nitrous*
14 *Oxide, North America* (Heydorn 1997). The 1990 through 1995 share of total quantity of N₂O used by each
15 subcategory was kept the same as the 1996 number provided by SRI Consulting. The 1997 through 2001 share of
16 total quantity of N₂O usage by sector was obtained from communication with a N₂O industry expert (Tupman
17 2002). The 2002 and 2003 share of total quantity of N₂O usage by sector was obtained from CGA (2002, 2003). Due
18 to the lack of publicly available data, the share of total quantity of N₂O usage data for years 2004 through 2021
19 was assumed to equal the 2003 value. The emission factor for the food processing propellant industry was
20 obtained from SRI Consulting's *Nitrous Oxide, North America* (Heydorn 1997) and confirmed by a N₂O industry
21 expert (Tupman 2002). The emission factor for all other subcategories was obtained from communication with a
22 N₂O industry expert (Tupman 2002). The emission factor for the medical/dental subcategory was obtained from
23 the *2006 IPCC Guidelines*.

24 Methodological approaches were applied to the entire time series to ensure consistency in emissions from 1990
25 through 2022.

26 **Uncertainty – TO BE UPDATED FOR FINAL REPORT**

27 The overall uncertainty associated with the 2022 N₂O emission estimate from N₂O product usage was calculated
28 using the *2006 IPCC Guidelines* (2006) Approach 2 methodology. Uncertainty associated with the parameters used
29 to estimate N₂O emissions include production data, total market share of each end use, and the emission factors
30 applied to each end use, respectively. The uncertainty associated with N₂O production data is ±25 percent, based
31 on expert judgment. The uncertainty associated with the market share for the medical/dental subcategory is ±0.56
32 percent, and uncertainty for the market share of food propellant subcategory is ±25 percent, both based on expert
33 judgment. Uncertainty for emission factors was assumed to be zero, and using this suggested uncertainty provided
34 in the *2006 IPCC Guidelines* is appropriate based on expert judgment (RTI 2023).

35 The results of this Approach 2 quantitative uncertainty analysis are summarized in Table 4-136. Nitrous oxide
36 emissions from N₂O product usage were estimated to be between 2.9 and 4.6 MMT CO₂ Eq. at the 95 percent
37 confidence level. This indicates a range of approximately 24 percent below to 24 percent above the emission
38 estimate of 3.8 MMT CO₂ Eq.

1 **Table 4-136: Approach 2 Quantitative Uncertainty Estimates for N₂O Emissions from N₂O**
 2 **Product Usage (MMT CO₂ Eq. and Percent)**

Source	Gas	2022 Emission Estimate (MMT CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a (MMT CO ₂ Eq.)			
			Lower Bound	Upper Bound	Lower Bound (%)	Upper Bound (%)
N ₂ O from Product Uses	N ₂ O	3.8	2.9	4.6	-24%	+24%

^a Range of emission estimates predicted by Monte Carlo stochastic simulation for a 95 percent confidence interval.

3 QA/QC and Verification

4 For more information on the general QA/QC process applied to this source category, consistent with Volume 1,
 5 Chapter 6 of the *2006 IPCC Guidelines*, see the QA/QC and Verification Procedures section in the introduction of
 6 the IPPU chapter.

7 Recalculations Discussion

8 No recalculations were performed for the time series compared to the previous *Inventory*.

9 Planned Improvements

10 EPA recently initiated an evaluation of alternative production statistics for cross-verification and updating time-
 11 series activity data, emission factors, assumptions, etc., and a reassessment of N₂O product use subcategories that
 12 accurately represent trends. This evaluation includes conducting a literature review of publications and research
 13 that may provide additional details on the industry. This work remains ongoing, and thus far no additional sources
 14 of data have been found to update this category.

15 Pending additional resources and planned improvement prioritization, EPA may also evaluate production and use
 16 cycles, and the potential need to incorporate a time lag between production and ultimate product use and
 17 resulting release of N₂O. Additionally, planned improvements include considering imports and exports of N₂O for
 18 product uses.

19 Finally, for future Inventories, EPA will examine data from EPA's GHGRP to improve the emission estimates for the
 20 N₂O product use subcategory. Particular attention will be made to ensure aggregated information can be published
 21 without disclosing CBI and time-series consistency, as the facility-level reporting data from EPA's GHGRP are not
 22 available for all inventory years as required in this *Inventory*. This is a lower priority improvement, and EPA is still
 23 assessing the possibility of incorporating aggregated GHGRP CBI data to estimate emissions; therefore, this
 24 planned improvement is still in development and not incorporated in the current *Inventory* report.

4.29 Industrial Processes and Product Use Sources of Precursor Gases

In addition to the main greenhouse gases addressed above, many industrial processes can result in emissions of various greenhouse gas precursors. The reporting requirements of the Paris Agreement and the UNFCCC¹³⁰ request that information should be provided on precursor emissions, which include carbon monoxide (CO), nitrogen oxides (NO_x), non-methane volatile organic compounds (NMVOCs), and sulfur dioxide (SO₂). These gases are not direct greenhouse gases, but indirectly impact Earth's radiative balance by altering the concentrations of greenhouse gases (e.g., ozone) and atmospheric aerosol (e.g., particulate sulfate). Combustion byproducts such as CO and NO_x are emitted from industrial applications that employ thermal incineration as a control technology. NMVOCs, commonly referred to as "hydrocarbons," are the primary gases emitted from most processes employing organic or petroleum-based products, and can also result from the product storage and handling.

Accidental releases of precursors associated with product use and handling can constitute major emissions in this category. In the United States, emissions from product use are primarily the result of solvent evaporation, whereby the lighter hydrocarbon molecules in the solvents escape into the atmosphere. The major categories of product uses include: degreasing, graphic arts, surface coating, other industrial uses of solvents (e.g., electronics), dry cleaning, and non-industrial uses (e.g., uses of paint thinner). Product usage in the United States also results in the emission of hydrofluorocarbons (HFCs) and small amounts of hydrofluoroethers (HFEs), which are included under Substitution of Ozone Depleting Substances and the Electronics Industry in this chapter.

Total emissions of NO_x, CO, NMVOCs, and SO₂ from non-energy industrial processes and product use from 1990 to 2022 are reported in Table 4-137.

Table 4-137: NO_x, CO, NMVOC, and SO₂ Emissions from Industrial Processes and Product Use (kt)

Gas/Source	1990	2005	2018	2019	2020	2021	2022
NO_x	517	429	314	304	274	275	274
Mineral Industry	160	200	118	114	101	101	101
Other Industrial Processes ^a	69	91	72	70	68	69	68
Metal Industry	96	58	63	60	52	52	52
Chemical Industry	192	80	61	59	54	54	54
CO	3,783	1,417	876	864	739	739	739
Metal Industry	2,261	707	447	448	340	340	340
Other Industrial Processes ^a	248	378	186	184	177	177	177
Mineral Industry	182	120	111	106	96	96	96
Chemical Industry	1,093	211	132	126	125	125	125
NMVOCs	6,733	3,418	2,943	2,814	3,191	3,191	3,191
Other Industrial Processes ^a	6,021	3,147	2,827	2,700	3,087	3,087	3,087
Chemical Industry	601	221	88	86	81	81	81
Mineral Industry	9	10	7	7	6	6	6
Metal Industry	102	40	21	20	17	17	17
SO₂	1,112	577	213	195	164	164	164
Other Industrial Processes ^a	97	57	23	19	19	19	19
Chemical Industry	283	242	106	97	83	83	83
Mineral Industry	166	138	25	25	26	26	26
Metal Industry	566	140	58	53	37	37	37

¹³⁰ See paragraph 51 of Annex to 18/CMA.1 available online at: https://unfccc.int/sites/default/files/resource/CMA2018_03a02E.pdf.

^a Other Industrial Processes includes storage and transport, other industrial processes (manufacturing of agriculture, food, and kindred products; wood, pulp, paper, and publishing products; rubber and miscellaneous plastic products; machinery products; construction; transportation equipment; and textiles, leather, and apparel products), and miscellaneous sources (catastrophic/accidental release, other combustion (structural fires), health services, repair shops, and fugitive dust). It does not include agricultural fires or slash/prescribed burning, which are accounted for under the Field Burning of Agricultural Residues source.

Note: Totals by gas may not sum due to independent rounding.

Source: (EPA 2023a). Emission categories from EPA (2023a) are aggregated into sectors and categories reported under the Paris Agreement and the UNFCCC as shown in Table ES-3.

1 Methodology and Time-Series Consistency

2 Emission estimates for 1990 through 2020 were obtained from data published on the National Emissions Inventory
3 (NEI) Air Pollutant Emissions Trends Data website (EPA 2023a). For Table 4-137, NEI reported emissions of CO, NO_x,
4 SO₂, and NMVOCs were recategorized from NEI Emissions Inventory System (EIS) sectors to source categories more
5 closely aligned with reporting sectors and categories under the Paris Agreement and the UNFCCC based on
6 discussions between the EPA GHG Inventory and NEI staff (see crosswalk documented in Annex 6.3).¹³¹ EIS sectors
7 mapped to the IPPU sector categories in this report include: chemical and allied product manufacturing, metals
8 processing, storage and transport, solvent utilization, other industrial processes, and miscellaneous sources. As
9 described in the NEI Technical Support Documentation (TSD) (EPA 2023b), NEI emissions are estimated through a
10 combination of emissions data submitted directly to the EPA by state, local, and tribal air agencies, as well as
11 additional information added by the Agency from EPA emissions programs, such as the emission trading program,
12 Toxics Release Inventory (TRI), and data collected during rule development or compliance testing.

13 Methodological approaches were applied to the entire time series to ensure time-series consistency from 1990
14 through 2021, which are described in detail in the NEI's TSD and on EPA's Air Pollutant Emission Trends web site
15 (EPA 2023a; EPA 2023b). A quantitative uncertainty analysis was not performed.

¹³¹ The NEI estimates and reports emissions from six criteria air pollutants (CAPs) and 187 hazardous air pollutants (HAPs) in support of National Ambient Air Quality Standards. EPA reported CAP emission trends are grouped into 60 sectors and 15 Tier 1 source categories, which broadly cover similar source categories to those presented in this chapter. For reporting precursor emissions in the common reporting tables (CRTs), EPA has mapped and regrouped emissions of greenhouse gas precursors (CO, NO_x, SO₂, and NMVOCs) from NEI's EIS sectors to better align with NIR source categories, and to ensure consistency and completeness to the extent possible. See Annex 6.3 for more information on this mapping.